

Supporting Information for
**Surface engineering of phosphorene nanoribbons by transition metal
heteroatoms for spintronics**

Mi-Mi Dong^a, Zi-Qun Wang^a, Guang-Ping Zhang^a, Chuan-Kui Wang^{a*}, Xiao-Xiao Fu^{a*}

^a Shandong Province Key Laboratory of Medical Physics and Image Processing
Technology, School of Physics and Electronics, Shandong Normal University, Jinan
250358, China

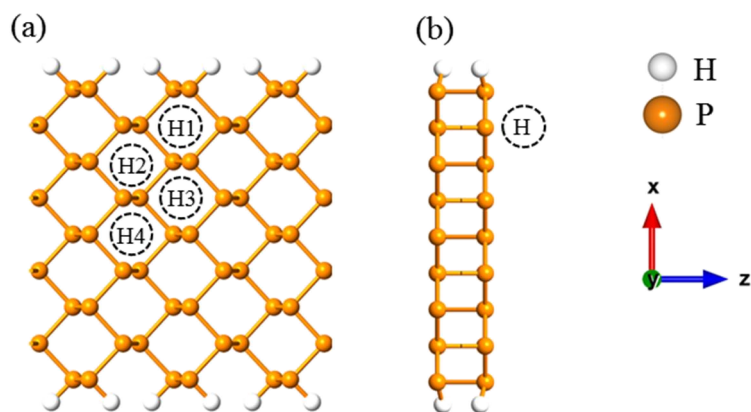


Fig. S1 (a) Top view and (b) side view for structures of H-APNR with 3 unit cells. Here, H1~H4 represent different hollow sites of adsorption for TM atoms.

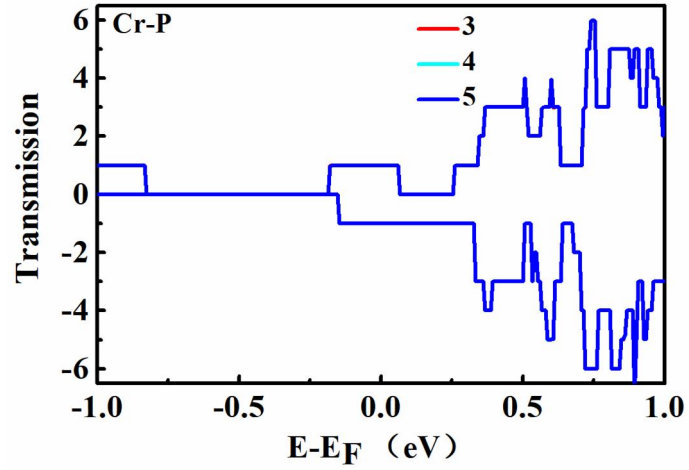


Fig. S2 The transmission spectra of Cr adsorbed H-APNRs for central region containing 3, 4, and 5 units with the zero bias in parallel spin configurations.

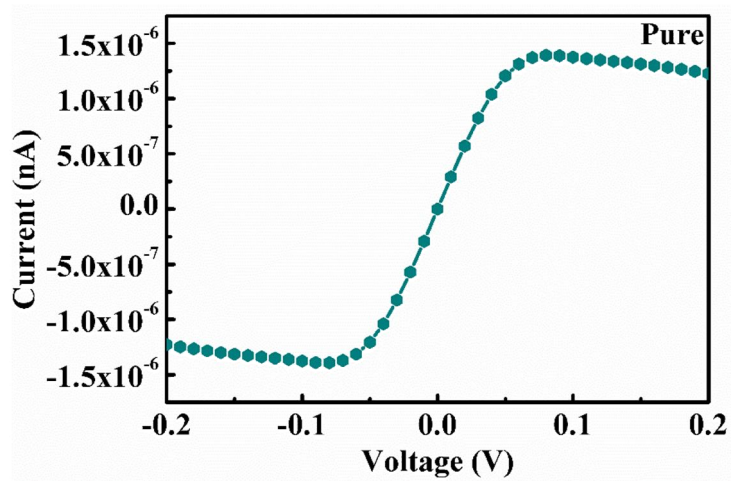


Fig. S3 Current-voltage characteristic curve of pure H-APNR.

Table S1 The maximum displacements of phosphorus atoms in adsorption configurations from the initial locations in phosphene nanoribbons in the y direction (ΔY_{Max}), shortest distances between TM atoms and phosphorus atom ($d_{\text{TM-P}}$), adsorption energy (E_a and E_a^*), cohesive energy (E_c), magnetic moment (M), and the charge transfer (ΔQ) of TM atoms.

Adatom	Site	ΔY_{Max} (Å)	$d_{\text{TM-P}}$ (Å)	E_a (eV)	E_a^* (eV)	E_c (eV)	M (μ_B)	ΔQ (e)
Sc	H1	0.31	2.55	-3.46	–		0.71	
	H2	0.26	2.57	-3.45	–		0.76	
	H3	0.29	2.59	-3.46	–		0.74	
	H4	0.28	2.59	-3.46	–	-3.90	0.73	-0.15
Ti	H1	0.38	2.42	-4.10	–		1.82	
	H2	0.29	2.43	-4.08	–		1.89	
	H3	0.30	2.43	-4.15	–		1.89	
	H4	0.30	2.44	-4.16	–	-4.85	1.89	-0.19
V	H1	0.31	2.33	-2.82	–		2.86	
	H2	0.37	2.34	-2.85	–		2.85	
	H3	0.30	2.34	-2.86	–		2.85	
	H4	0.38	2.34	-2.86	–	-5.31	2.84	-0.10
Cr	H1	0.06	2.45	-2.92	–		4.83	
	H2	0.09	2.45	-3.02	–		4.82	
	H3	0.09	2.45	-3.05	–		4.82	
	H4	0.04	2.45	-3.05	–	-4.10	4.82	-0.14
Mn	H1	0.72	2.33	-2.13	-1.20		4.72	
	H2	0.23	2.06	-2.27	-1.57		4.67	
	H3	0.17	2.33	-2.26	-1.57		4.65	
	H4	0.18	2.33	-2.27	-1.59	-2.92	4.65	-0.08
Fe	H1	0.07	2.13	-4.10	-3.06		2.29	
	H2	0.15	2.13	-4.04	-2.61		2.33	
	H3	0.18	2.13	-4.06	-2.94		2.33	
	H4	0.21	2.14	-4.07	-2.81	-4.28	2.34	-0.09
Co	H1	0.09	2.11	-4.99	-3.83		1.09	
	H2	0.24	2.12	-4.95	-3.85		1.04	
	H3	0.16	2.12	-4.98	-3.87		1.04	
	H4	0.16	2.44	-4.99	-3.88	-4.39	1.04	-0.07
Ni	H1	0.31	2.12	-5.30	-4.28		0.00	
	H2	0.26	2.12	-5.29	-4.26		0.00	
	H3	0.11	2.12	-5.32	-4.28		0.00	
	H4	0.08	2.12	-5.33	-4.29	-4.44	0.00	-0.06

E_a^* represents the adsorption energy by the other group¹.

Table S2 The comparisons for the maximum displacements of phosphorus atoms in adsorption configurations from the initial locations in phosphorene nanoribbons in the y direction (ΔY_{Max}), the shortest distances between TM atoms and phosphorus atom ($d_{\text{TM-P}}$), and the adsorption energy (E_a) without (no vdW) and with vdW consideration (vdW).

Adatom	ΔY_{Max} (Å)		$d_{\text{TM-P}}$ (Å)		E_a (eV)	
	no vdW	vdW	no vdW	vdW	no vdW	vdW
Sc	0.28	0.25	2.59	2.58	-3.46	-3.26
Ti	0.30	0.28	2.44	2.44	-4.16	-4.16
V	0.38	0.37	2.34	2.34	-2.86	-2.86
Cr	0.04	0.05	2.45	2.45	-3.05	-3.05
Mn	0.18	0.21	2.33	2.33	-2.27	-2.26
Fe	0.21	0.21	2.14	2.13	-4.07	-4.06
Co	0.16	0.15	2.44	2.43	-4.99	-5.01
Ni	0.08	0.11	2.12	2.12	-5.53	-5.53

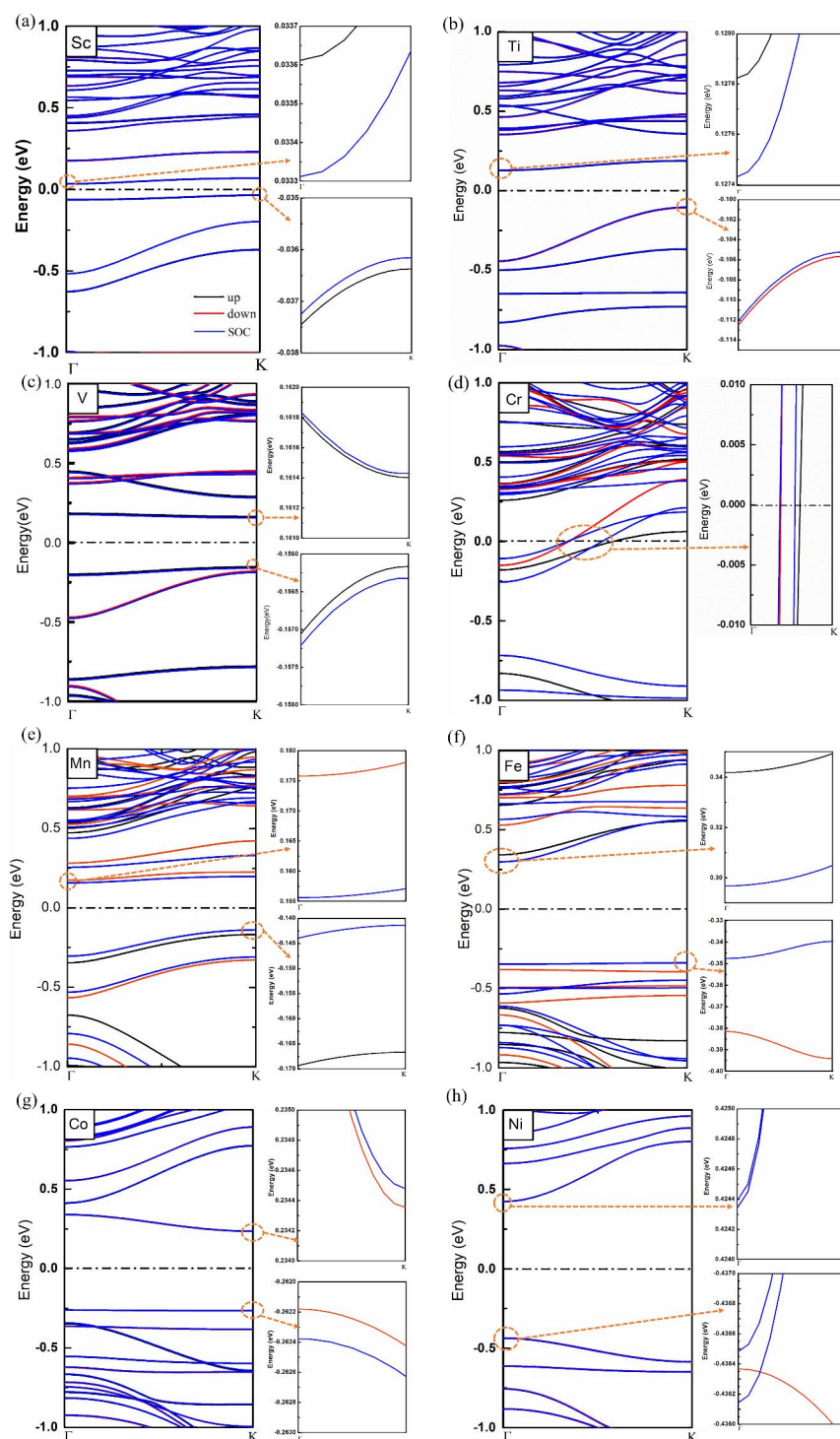


Fig. S4 The band structures of (a) Sc, (b) Ti, (c) V, (d) Cr, (e) Mn, (f) Fe, (g) Co and (h) Ni adsorbed H-APNRs. The black, red and blue colors show the band structure of spin up, down and the spin-orbit coupling, respectively..