

Supporting Information

DFT calculations of the synergistic effect of λ - MnO_2 /Graphene composites for electrochemical adsorption of lithium ions

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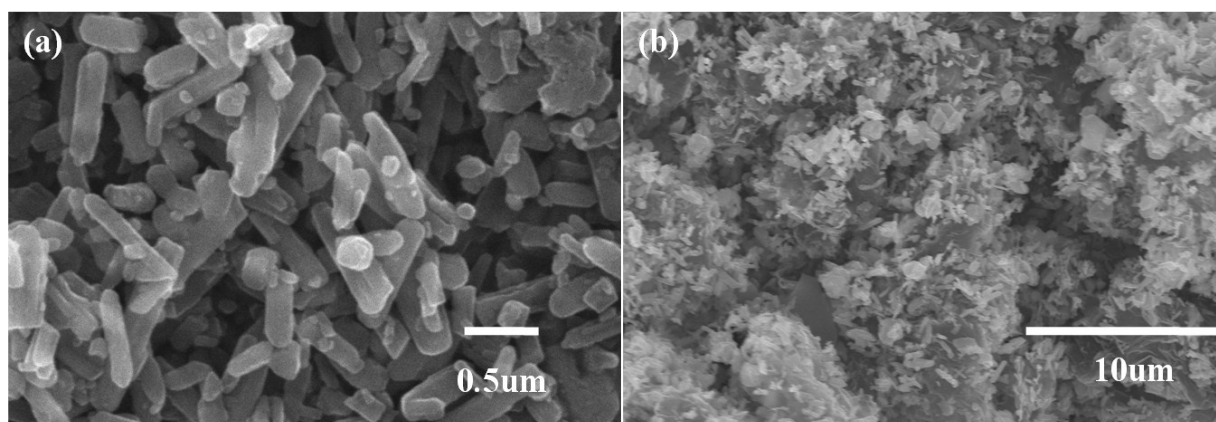


Fig. S1 SEM images of (a) λ - MnO_2 and (b) λ - MnO_2 /Graphene.

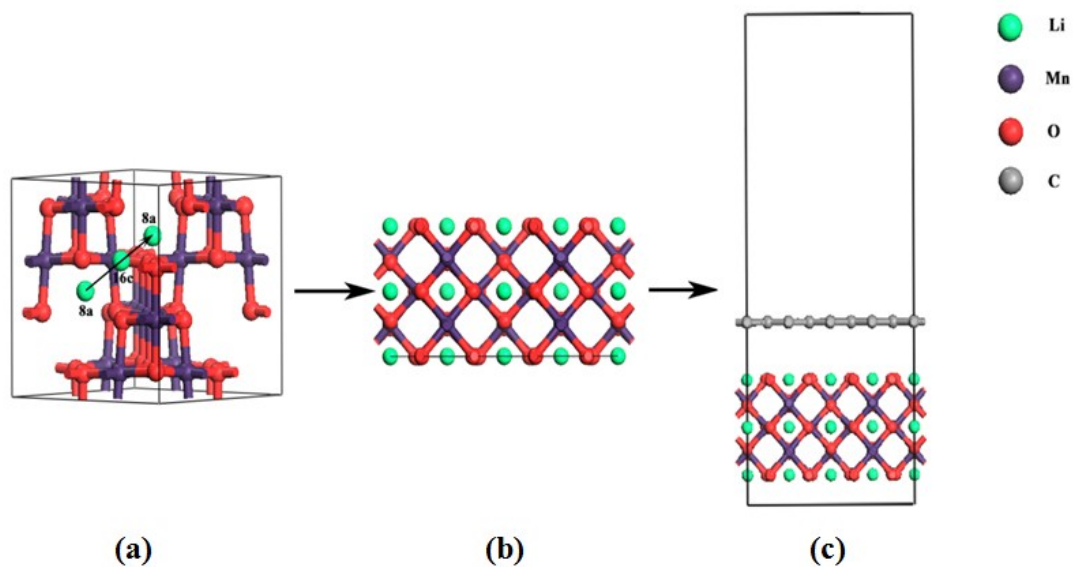


Fig. S2 The building process of λ -MnO₂/Graphene with Li model: (a) the diffusion path of Li in λ -MnO₂; (b) the (1 0 1) crystal plane of LiMn₂O₄; (c) the building model of λ -MnO₂/Graphene with Li.

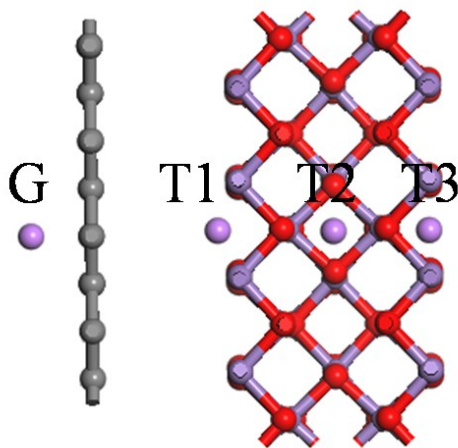


Fig. S3 The adsorption sites of the Li atoms in λ -MnO₂/Graphene.

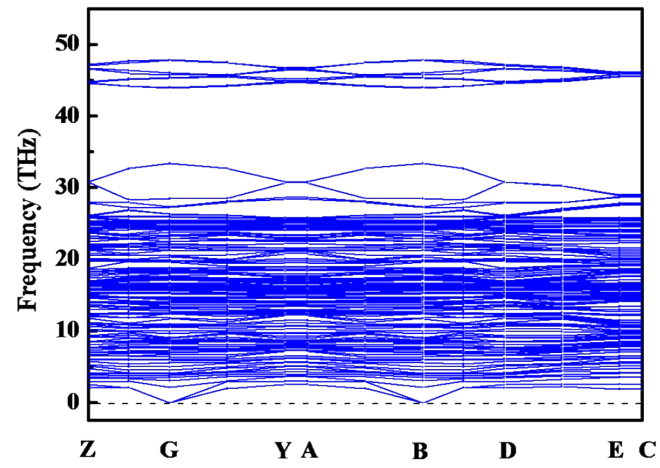


Fig. S4 Calculated phonon dispersion curves of λ -MnO₂/Graphene.

Table S1 Distance and energy before and after optimized λ -MnO₂/Graphene

Distance of λ -MnO ₂ /Graphene (Å)	Optimized distance of λ -MnO ₂ /Graphene (Å)	Energy of λ -MnO ₂ /Graphene (eV)
5	4.845	-35396.419
4.5	4.306	-35396.435
4	3.895	-35396.677
3.5	3.406	-35396.948
3	3.008	-35396.627
2.5	2.609	-35395.077
2	2.558	-35392.794

Table S2 The binding energies of Li at different positions of λ -MnO₂/Graphene

Position of Li	Bonding energy (eV)
G	-2.0234728
T1	-5.6050328
T2	-4.9601237
T3	-5.0477628
T1	-5.6017268 (With dipole correction)

Table S3 The values of $L_{\text{Mn-O}}$ and $d_{\text{Li-O}}$ in LiMn₂O₄ bulk and LiMn₂O₄/Graphene interface

	$L_{\text{Mn-O}}$ (Å)	$d_{\text{Li-O}}$ (Å)
Inside of LiMn ₂ O ₄	1.967	1.911
Interface of LiMn ₂ O ₄ /Graphene	1.943	1.934

Table S4 Mulliken charges population and bond population for MMn₂O₄ (M = Li, Na,

Mg)

M		Mulliken population analysis
Li	Li (e)	1.06
	Li-O	0
Na	Na (e)	0.84
	Na-O	0.13
Mg	Mg (e)	1.79
	Mg-O	-1.38

Table S5 The electrode values of graphene, λ -MnO₂ and λ -MnO₂/Graphene in EIS

Electrode	Graphene	λ -MnO ₂	λ -MnO ₂ /Graphene
R _e (Ω)	35.42	29.96	27.88
C _L (F)	2.999*10 ⁻⁵	8.046*10 ⁻⁵	2.119*10 ⁻⁵
R _{ct} (Ω)	43.12	21.03	16.25
Z _w (Ω)	2.164*10 ⁻³	7.624*10 ⁻³	5.255*10 ⁻²
C _{dl} (F)	1.878*10 ⁻³	2.477*10 ⁻²	1.810*10 ⁻¹