

**Supporting Information for**

**Chemical Pressure-Chemical Knowledge:  
Squeezing Bonds and Lone Pairs within  
the Valence Shell Electron Pair Repulsion  
Model**

A. Lobato,<sup>\*a</sup> H. H. Osman,<sup>b</sup> M. A. Salvadó,<sup>b</sup> M. Taravillo,<sup>a</sup>  
V. G. Baonza<sup>a,c</sup> and J. M. Recio<sup>\*b</sup>

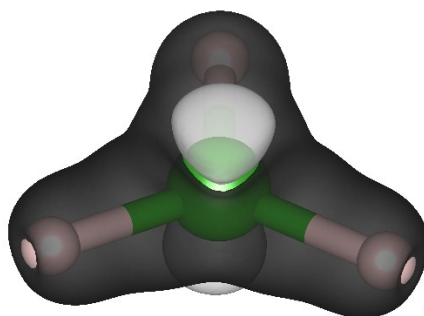
- a. Malta-Consolider Team, Dpto. de Química Física, Facultad de Ciencias Químicas, Universidad Complutense de Madrid, 28040 Madrid, Spain.
- b. MALTA-Consolider Team, Dpto de Química Física y Analítica, Universidad de Oviedo, E-33006 Oviedo, Spain.
- c. Instituto de Geociencias IGEO (CSIC-UCM), 28040 Madrid, Spain.

## Computational details of studied molecules

Table S1: Calculation details and geometrical parameters of the molecules studied.

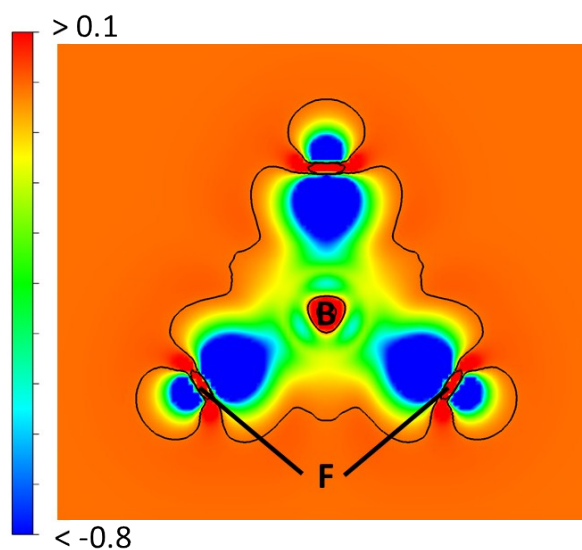
Molecule	ecut (ha)	k-points	nfft grid	Bond Length (Å)	Bond Angle (deg)
H <sub>2</sub> O	270	1x1x1	288x288x288	d(O-H) = 0.97120	$\alpha$ (H-O-H) = 104.92
NH <sub>3</sub>	220	1x1x1	256x256x256	d(N-H) = 1.0217	$\alpha$ (H-N-H) = 107.26
NF <sub>3</sub>	300	1x1x1	300x300x300	d(N-F) = 1.37301	$\alpha$ (F-N-F) = 101.70
AsH <sub>3</sub>	280	1x1x1	288x288x288	d(As-H) = 1.52132	$\alpha$ (H-As-H) = 90.62
PH <sub>3</sub>	240	1x1x1	270x270x270	d(P-H) = 1.42719	$\alpha$ (H-P-H) = 91.83
PF <sub>3</sub>	290	1x1x1	300x300x300	d(P-F) = 1.56403	$\alpha$ (F-P-F) = 97.44
PCl <sub>3</sub>	270	1x1x1	290x290x290	d(P-Cl) = 2.03959	$\alpha$ (Cl-P-Cl) = 100.36
SO <sub>2</sub>	260	1x1x1	288x288x288	d(S-O) = 1.4281	$\alpha$ (O-S-O) = 119.54
SF <sub>4</sub>	280	1x1x1	290x290x290	d(S-F) <sub>ax</sub> = 1.6482 d(S-F) <sub>eq</sub> = 1.5541	$\alpha$ (F-S-F) <sub>ax</sub> = 173.07 $\alpha$ (F-S-F) <sub>eq</sub> = 100.61
ClF <sub>3</sub>	360	1x1x1	324x324x324	d(Cl-F) <sub>ax</sub> = 1.6921 d(Cl-F) <sub>eq</sub> = 1.6026	$\alpha$ (F-Cl-F) <sub>ax</sub> = 176.9 $\alpha$ (F-Cl-F) <sub>eq</sub> = 88.19
XeF <sub>2</sub>	380	1x1x1	340x340x340	d(Xe-F) = 2.0084	$\alpha$ (F-Xe-F) = 179.99
BeH <sub>2</sub>	290	1x1x1	290x290x290	d(Be-H) = 1.3167	$\alpha$ (H-Be-H) = 179.99
BH <sub>3</sub>	280	1x1x1	280x280x280	d(B-H) = 1.1983	$\alpha$ (H-B-H) = 120.00
BF <sub>3</sub>	320	1x1x1	320x320x320	d(B-F) = 1.3104	$\alpha$ (F-B-F) = 119.99
Ethylene	300	1x1x1	300x300x300	d(C-C) = 1.3309 d(C-H) = 1.0852	$\alpha$ (H-C-H) = 119.49

### Chemical Pressure distribution in $\text{BH}_3$ Molecule



**Figure S1.** 3D isosurfaces of chemical pressure (CP) distributions within the  $\text{BH}_3$  molecule. Isosurface values:  $\text{CP}=+0.013$  (white) and  $-0.013$  (black). Green and white spheres indicate boron and hydrogen atoms, respectively.

### Chemical Pressure distribution in $\text{BF}_3$ Molecule



**Figure S2.** Chemical pressure heat-maps of the  $\text{BF}_3$  molecule along the molecular plane. Black curves:  $\text{CP} = 0$  contour.