## Supplementary information for

## Physical Origin Underlying the Prenucleation Cluster Mediated Nonclassical Nucleation Pathways for Calcium Phosphate

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Table S1. Summary of the simulations designed in the present study for investigating the PNC-mediated nonclassical nucleation mechanisms for calcium phosphate. The total cumulative simulation time is about  $5.2 \ \mu s$ .

Simulation methods	Simulation systems	Simulation setup	Total time
	Ca-P <sub>1</sub>	6 ns (NPT)+2 ns (NVT)+34 330 ns <sup>3</sup>	
US1	CaP <sub>1</sub> -P <sub>2</sub>	6 ns (NPT)+2 ns (NVT)+34 \arrow30 ns <sup>3</sup>	2.994 µs
	$CaP_1P_2-P_3$	6 ns (NPT)+2 ns (NVT)+31 \arrow330 ns <sup>3</sup>	
$MD^1$	$CaP_1P_2P_3$	6 <b>¤</b> 10 ns (NVT) <sup>4</sup>	0.060 µs
MTD <sup>1</sup>	Ca-P <sub>1</sub>	100 ns (NVT)	
	$Ca-P_1P_2$	300 ns (NVT)	0.700 µs
	$Ca-P_1P_2P_3$	300 ns (NVT)	
MD <sup>2</sup>	Ca/P=1:3	40 ns (NPT)	
	Ca/P=1:1	2 <b>°</b> 40 ns (NPT) <sup>5</sup>	0.200 µs
	Ca/P=2:1	2 <b>\\$</b> 40 ns (NPT) <sup>5</sup>	
US <sup>2</sup>	PNC-PNC	6 ns (NPT)+2 ns (NVT)+35 <b>℃</b> 6 ns	
	PNC-Ca	6 ns (NPT)+2 ns (NVT)+44 <b>R</b> 6 ns	1.290 µs
	PNC_Ca-PNC 6 ns (NPT)+2 ns (NVT)+33 224 ns		

<sup>1</sup>The simulation methodologies are designed to investigate the PNC formation pathways and the thermodynamic stability of the obtained intermediate states. <sup>2</sup>These simulations are used to unravel the mechanisms along which the highly charged PNCs aggregate to form the new phase, such as the polymetric aggregates and the ACP. <sup>3</sup>A total number of about 34 window simulations for each PMF profile (Table S1) are carried out. <sup>4</sup>Six PNC configurations corresponding to the free energy minimum are exacted from the US trajectories to prepare the initial simulation systems. <sup>5</sup>Two simulation systems are prepared at different ion concentrations in solution to explore the dependence of the obtained aggregation trends on the experimental conditions.



Fig. S1. The pathway-specified strategies for free energy calculations to investigate the thermodynamics for the PNC formations. (a) the phosphate ions are proposed to progressively bind to one calcium ion in a multi-step fashion (Path 1). US together with WHAM is used to obtain PMF profiles; (b) two or three phosphate ions are simultaneously sequestered to one calcium ion (Path 2 and Path 3). The Well-tempered MTD simulations are employed to calculate the multi-dimensional free energy surface.



Fig. S2. A comparison of the PMF profiles for the  $Ca^{2+}$ -HPO<sub>4</sub><sup>2-</sup> pairing interaction computed by the US and MTD simulations.



Fig. S3. The mean forces exerting on the ions collected as an average over a series of configurations from the US simulations as a function of simulation time for the interaction between  $Ca(HPO_4)_2^{2-}$  and  $HPO_4^{2-}$  at two selected separation distances (3.2 and 10.0 Å). A well converged result for the force averaging is obtained after 20 ns.



Fig. S4. The pathway-specified strategies for the free energy calculations to investigate the molecular mechanism underlying the aggregations of the highly charged PNCs. (a) the PNC-PNC interaction (b) the coordination of the  $Ca(HPO_4)_3^{4-}$  to a free  $Ca^{2+}$  in solution, leading to the formation of  $Ca_2(HPO_4)_3^{2-}$ . (c) The self-assembly of  $Ca(HPO_4)_3^{4-}$  and  $Ca_2(HPO_4)_3^{2-}$  to form  $Ca_3(HPO_4)_6^{6-}$ , representing the aggregation process of the PNCs.



Fig. S5. The representative snapshots from the initial and the final stages of the PNC aggregation (Ca/P = 1.0) at the low ion concentration. The free PNCs successfully aggregate into small clusters.