Electronic Supplementary Information

## Experimental and theoretical study of multinuclear indium-oxo clusters in CHA zeolite for CH<sub>4</sub> activation at room temperature

Zen Maeno\*a, Shunsaku Yasumuraa, Chong Liua, Takashi Toyaoa, Kenichi Kona,

Akira Nakayama<sup>a,c</sup>, Jun-ya Hasegawa<sup>a</sup>, Ken-ichi Shimizu\*<sup>a,b</sup>

<sup>a</sup> Institute for Catalysis, Hokkaido University, N-21, W-10, Sapporo 001-0021, Japan

<sup>b</sup> Elements Strategy Initiative for Catalysts and Batteries, Kyoto University, Katsura,

Kyoto 615-8520, Japan

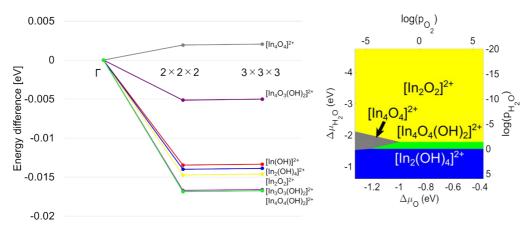
<sup>c</sup>JST, PRESTO, 4-1-8 Honcho Kawaguchi, Saitama 332-0012, Japan

\*Corresponding authors

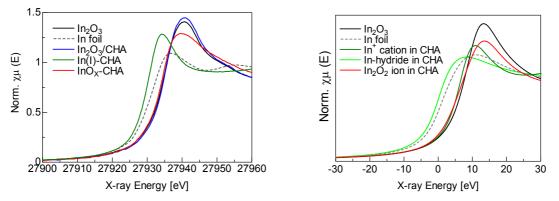
Zen Maeno, Ken-ichi Shimizu

E-mail: kshimizu@cat.hokudai.ac.jp, TEL: +81-11-706-9164

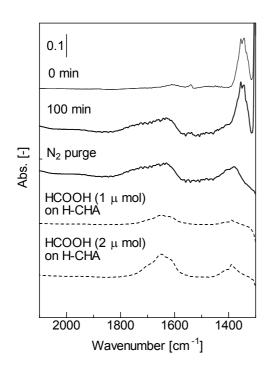
## **Figures and Tables**



**Figure S1.** Difference of total energy using  $2 \times 2 \times 2$  and  $3 \times 3 \times 3$  k point meshes and that using  $\Gamma$  point for each In-oxo ions on 8MR(4NN) (left). The phase diagram for 8MR(4NN) obtained using  $2 \times 2 \times 2$  k point mesh (right).



**Figure S2.** Comparison of experimental (left) and simulated (right) XANES spectra of a series of In-CHA and reference In compounds (In foil and  $In_2O_3$ ).



**Figure S3.** FT-IR spectra during reaction of  $CH_4$  on  $InO_X$ -CHA at room temperature and formic acid on H-CHA (as reference).

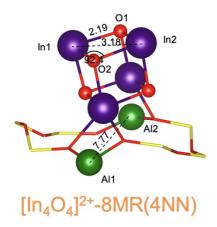
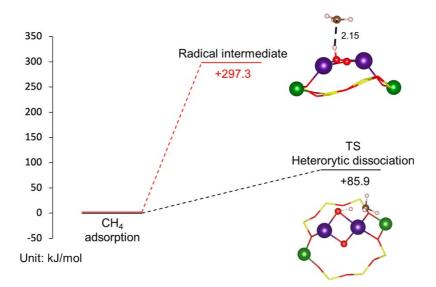


Figure S4. Optimized structure of [In<sub>4</sub>O<sub>4</sub>]<sup>2+</sup>-8MR 4NN



**Figure S5.** Comparison of energy diagrams in C–H cleavage of  $CH_4$  on  $[In_2O_2]^{2+}-8MR 4NN$  through heterolytic path (lower) and homolytic path (upper).

Sample	BET surface area (m <sup>2</sup> g <sup>-1</sup> )	Pore volume (mL g <sup>-1</sup> )
H-CHA	833	0.402
In <sub>2</sub> O <sub>3</sub> /CHA	785	0.378
InO <sub>x</sub> -CHA	710	0.344

Table S1. Specific surface areas and pore volumes of zeolites

Table S2. Total energy (eV) of In-oxo ions on 6MR(3NN), 8MR(3NN), and 8MR(4NN)

In species	6MR(3NN)	8MR(3NN)	8MR(4NN)
[In(OH)] <sup>2+</sup>	-873.3	-871.9	-871.8
[ln <sub>2</sub> O <sub>2</sub> ] <sup>2+</sup>	-876.5	-877.0	-877.2
[ln <sub>2</sub> (OH) <sub>4</sub> ] <sup>2+</sup>	-909.2	-909.4	-909.0
[ln <sub>3</sub> O <sub>3</sub> (OH) <sub>2</sub> ] <sup>2+</sup>	-897.0	-898.3	-898.0
[ln <sub>4</sub> O <sub>4</sub> ] <sup>2+</sup>	-897.4	-897.9	-898.1
[ln <sub>4</sub> O <sub>3</sub> (OH) <sub>2</sub> ] <sup>2+</sup>	-913.2	-913.9	-913.5
[ln <sub>4</sub> O <sub>4</sub> (OH) <sub>2</sub> ] <sup>2+</sup>	-919.3	-920.2	-920.0