

Supplementary Information

Switching Charge Transfer Characteristics of Quaterthiophene from p-type to n-type via Interactions with Carbon Nanotube

Ankita Joshi and C.N. Ramachandran*

Department of Chemistry, Indian Institute of Technology Roorkee, Roorkee, Uttarakhand,
India-247667

*e-mail: ramcnfcy@iitr.ac.in

Calculation of Charge Transport Parameters

The transfer integral for an electron (t_-) can be calculated from the energies of LUMO and LUMO+1,

$$t_- = \frac{E_{LUMO+1} - E_{LUMO}}{2} \quad (1)$$

and that for a hole (t_+) can be determined from the energies of HOMO and HOMO-1,

$$t_+ = \frac{E_{HOMO} - E_{HOMO-1}}{2} \quad (2)$$

The reorganization energy for any charged species (hole/electron) is given by,

$(E(\text{charged species}) - E^0(\text{charged species})) + (E(\text{neutral geometry obtained from the charged species}) - E^0(\text{neutral geometry}))$, where E is the energy of unoptimized and E^0 is the energy of optimized species.

The rate of charge transfer (k) is calculated based on Marcus theory,

$$k = \frac{4\pi^2}{h} \frac{1}{\sqrt{4\pi k_B T}} t^2 e^{-\lambda/k_B T} \quad (3)$$

where, h is the Planck's constant, T is the room temperature (298 K) and k_B is the Boltzmann constant.

The diffusion coefficient D is given by

$$D = \frac{kL^2}{2} \quad (4)$$

where L is the nearest center-to-center distance

The Einstein relation for the mobility of charge carriers (μ) is given by,

$$\mu = \frac{eD}{k_B T} \quad (5)$$

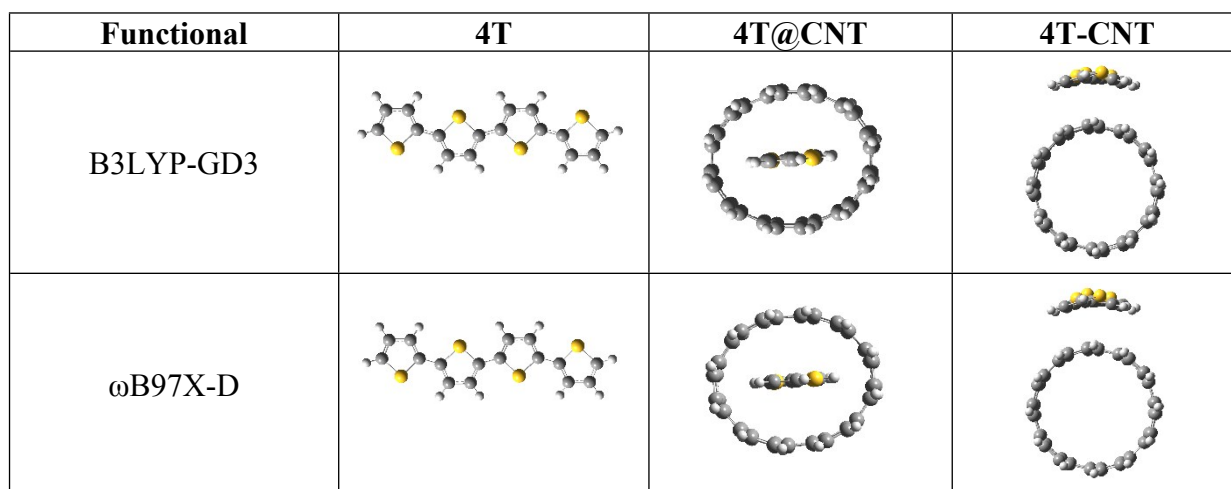


Figure S1. Optimized geometries of 4T, 4T@CNT and 4T-CNT obtained at B3LYP-GD3 and ω B97X-D levels.

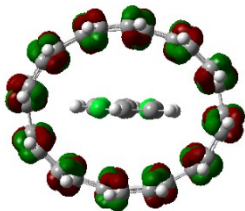
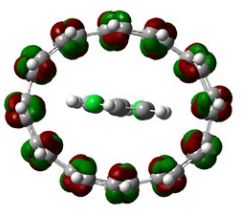
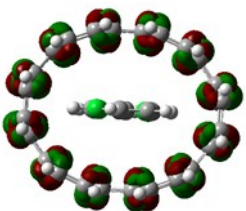
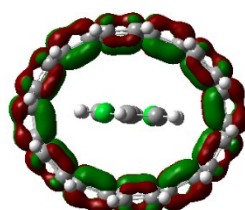
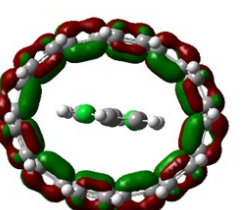
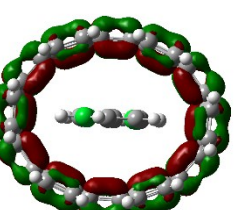
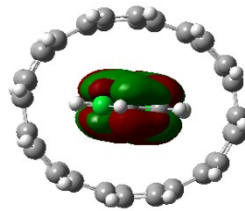
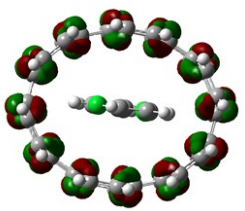
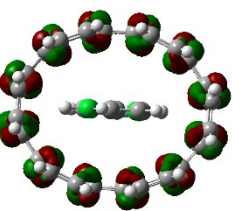
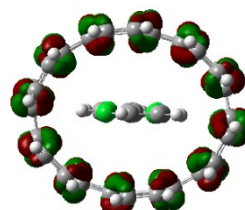
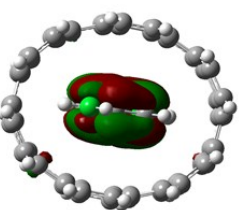
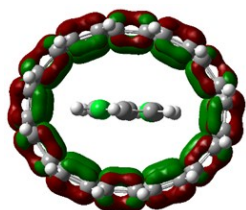
B97-D	B3LYP-GD3	ω B97X-D
		
$E_{\text{LUMO}+1} = -2.93 \text{ eV}$	$E_{\text{LUMO}+1} = -2.60 \text{ eV}$	$E_{\text{LUMO}+1} = -1.42 \text{ eV}$
		
$E_{\text{LUMO}} = -3.11 \text{ eV}$	$E_{\text{LUMO}} = -2.79 \text{ eV}$	$E_{\text{LUMO}} = -1.66 \text{ eV}$
		
$E_{\text{HOMO}} = -3.76 \text{ eV}$	$E_{\text{HOMO}} = -4.29 \text{ eV}$	$E_{\text{HOMO}} = -5.64 \text{ eV}$
		
$E_{\text{HOMO}-1} = -3.91 \text{ eV}$	$E_{\text{HOMO}-1} = -4.44 \text{ eV}$	$E_{\text{HOMO}-1} = -5.86 \text{ eV}$

Figure S2. Selected molecular orbitals and the corresponding energy for 4T@CNT obtained at different levels.

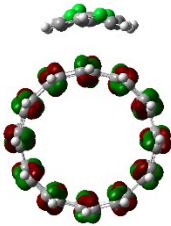
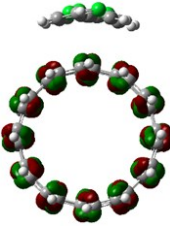
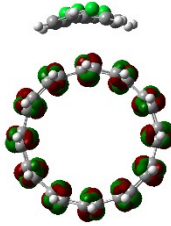


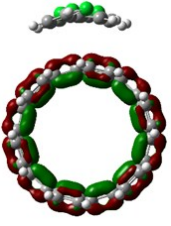

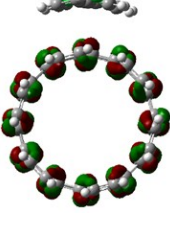
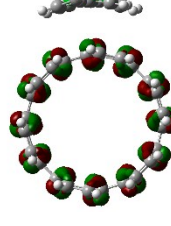



B97-D	B3LYP-GD3	ω B97X-D
		
$E_{\text{LUMO}+1} = -2.90 \text{ eV}$	$E_{\text{LUMO}+1} = -2.56 \text{ eV}$	$E_{\text{LUMO}+1} = -1.40 \text{ eV}$
		
$E_{\text{LUMO}} = -3.09 \text{ eV}$	$E_{\text{LUMO}} = -2.78 \text{ eV}$	$E_{\text{LUMO}} = -1.65 \text{ eV}$
		
$E_{\text{HOMO}} = -3.89 \text{ eV}$	$E_{\text{HOMO}} = -4.26 \text{ eV}$	$E_{\text{HOMO}} = -5.61 \text{ eV}$
		
$E_{\text{HOMO}-1} = -4.08 \text{ eV}$	$E_{\text{HOMO}-1} = -4.47 \text{ eV}$	$E_{\text{HOMO}-1} = -5.84 \text{ eV}$

Figure S3. Selected molecular orbitals and the corresponding energy for **4T-CNT** obtained at different levels.

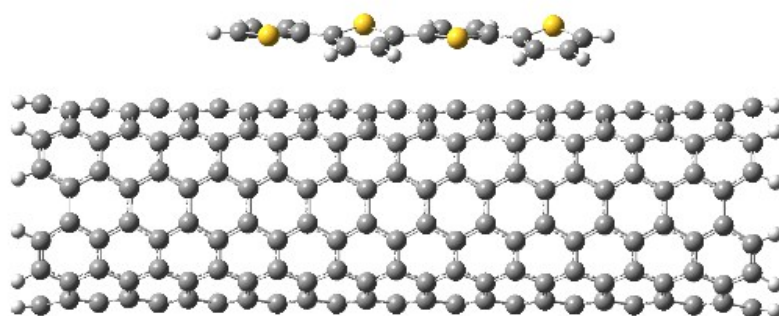


Figure S4. Optimized geometry of the exohedral complex of **long CNT** with **4T** obtained at B3LYP-GD3 level.

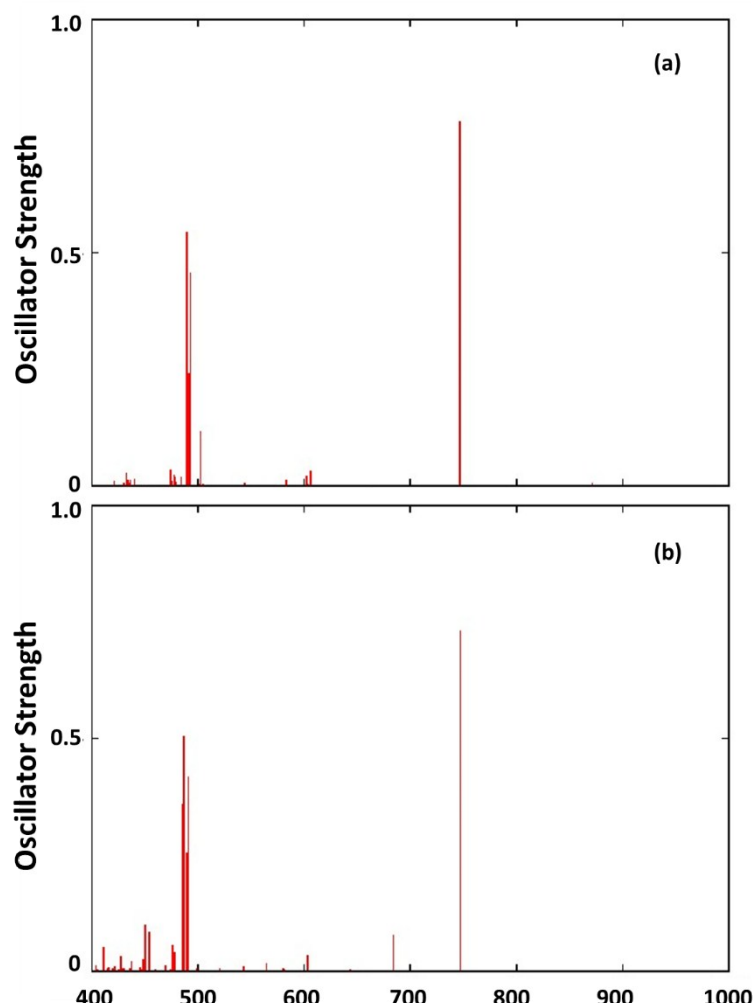


Figure S5. Simulated absorption spectra of (a) 4T@(7,7)CNT and (b) 4T-(7,7)CNT at B3LYP-GD3/6-31G(d,p) level.

Table S1. Calculated values of ionization energy (IE), electron affinity (EA), energy gap between HOMO and LUMO (ΔE_{H-L}) for 4T, CNT and their complexes at different levels. All values are in eV.

Complex	Functional	VIE	AIE	VEA	AEA	ΔE_{H-L}
4T	B97-D	6.01	5.89	-0.72	-0.82	1.87
	B3LYP-GD3	6.33	6.13	-0.59	-0.77	3.15 (3.13 ^a)
	ω B97X-D	6.86	6.49	-0.21	-0.56	6.93
(6,6)CNT ^b	B97-D	4.93	4.90	-2.15	-2.17	0.80
	B3LYP-GD3	5.06	5.03	-2.06	-2.09	1.47
	ω B97X-D	5.50	5.43	-1.83	-1.90	3.97
4T@CNT	B97-D	— ^c	— ^c	-2.13	-2.15	0.64
	B3LYP-GD3	5.03	5.01	-2.04	-2.07	1.50
	ω B97X-D	5.48	5.43	-1.81	-1.87	3.98
4T-CNT	B97-D	4.86	4.83	-2.13	-2.15	0.80
	B3LYP-GD3	5.01	4.97	-2.04	-2.07	1.45
	ω B97X-D	5.45	5.37	-1.81	-1.88	3.96

^aRef. [48]

^bRef. [42-44]


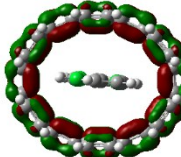
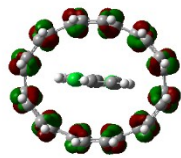
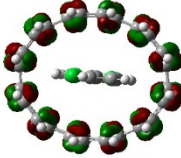
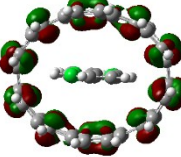

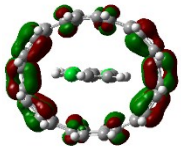
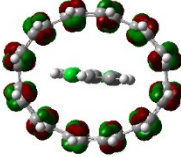
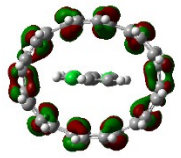
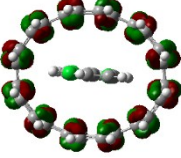
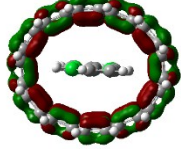
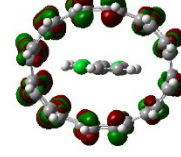
^cConvergence criteria not met.


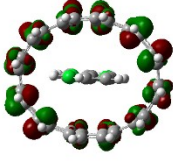
Table S2. Calculated values of maximum absorption wavelength (λ_{max}), oscillator strength (f) and light-harvesting efficiency (LHE) for 4T at different levels.

Functional	λ_{\max} (nm)	f	LHE (%)
B97-D	490	1.14	93
B3LYP-GD3	423 (436) ^a	1.19	94
ω B97X-D	344	1.20	94

^aRef. [59]

Table S3. The absorption wavelength, oscillator strength, orbital contribution and molecular orbitals involved in the transition of the complex 4T@CNT obtained at B3LYP-GD3/6-31G(d,p) level.

Wavelength (nm)	Oscillator strength	Orbital contribution (%)	Molecular orbitals involved	
738	0.76	48	 HOMO-2	 LUMO
		47	 HOMO	 LUMO+1
473	0.27	22	 HOMO-7	 LUMO
		20	 HOMO-5	 LUMO+1
468	0.05	67	 HOMO-8	 LUMO+1
		35	 HOMO-2	 LUMO

463	0.05		HOMO-2	LUMO+9
		42	 HOMO-2	 LUMO+10

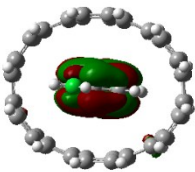
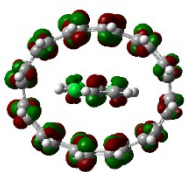
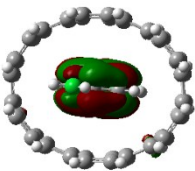
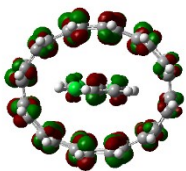
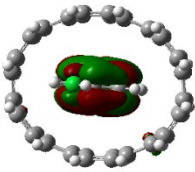
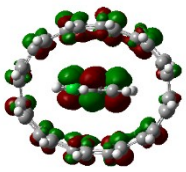

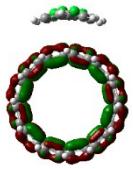
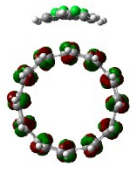
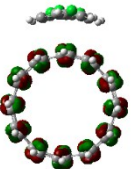
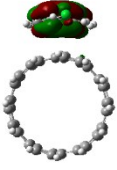
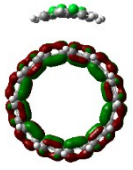

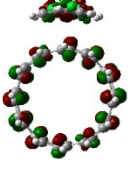



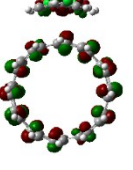

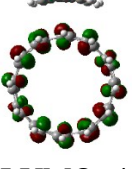
458	0.19	25	 <p>HOMO-1</p>	 <p>LUMO+14</p>
453	0.16	46	 <p>HOMO-1</p>	 <p>LUMO+14</p>
445	0.10	54	 <p>HOMO-1</p>	 <p>LUMO+15</p>

Table S4. The absorption wavelength, oscillator strength, orbital contribution and molecular orbitals involved in the transition of the complex **4T-CNT** obtained at B3LYP-GD3/6-31G(d,p) level.

Wavelength (nm)	Oscillator strength	Orbital contribution (%)	Molecular orbitals involved	
731	0.68	44	 HOMO-1	 LUMO
		42	 HOMO	 LUMO+1
685	0.17	87	 HOMO-2	 LUMO
464	0.24	61	 HOMO-2	 LUMO+3
462	0.26	50	 HOMO-2	 LUMO+4
460	0.32	26	 HOMO-2	 LUMO+3
459	0.31	34	 HOMO-2	 LUMO+4

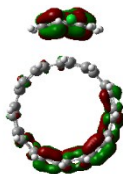
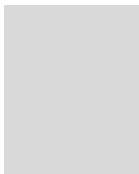
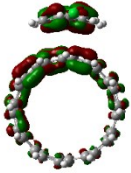
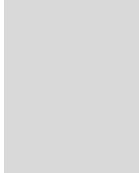
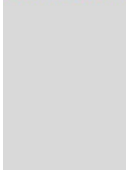
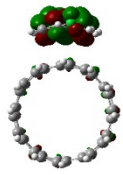
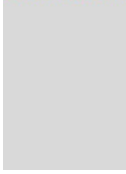
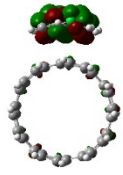
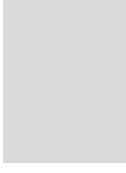
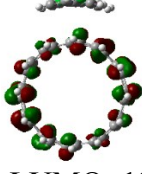
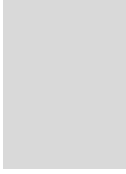

448	0.07	30	 HOMO-10	 LUMO
		19	 HOMO-9	 LUMO
446	0.14	35	 HOMO-2	 LUMO+5
445	0.09	24	 HOMO-2	 LUMO+5
		52	 HOMO	 LUMO+12
439	0.05	19	 HOMO-1	 LUMO+14

Table S5. Calculated values of transfer integral (t), internal reorganization energy (λ), rate constant (k) and carrier mobility (μ) for the complexes at different distance between centres of **4T** and **CNT** (d_{c-c}) obtained at B3LYP-GD3/6-31G(d,p) level.

Complex	d_{c-c} (Å)	λ^+ (meV)	λ^- (meV)	t^+ (meV)	t^- (meV)	k^+ (s⁻¹)	k^- (s⁻¹)	μ^+ (cm² V⁻¹s⁻¹)	μ^- (cm² V⁻¹s⁻¹)
4T@(7,7)CNT	4.29	59	52	60	67	0.24×10^{14}	0.51×10^{14}	0.86	1.83
4T-(7,7)CNT	3.31	63	58	57	63	0.20×10^{14}	0.29×10^{14}	0.42	0.62