SUPPLEMENTARY INFORMATION

Measuring Multiple ¹⁷O-¹³C *J*-couplings in Naphthalaldehydic Acid: A Combined Solid State NMR and Density Functional Theory Approach

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Table S1. The first principles GIPAW DFT calculated ${}^{1}\text{H}/{}^{2}\text{D}$ and ${}^{13}\text{C}$ NMR parameters δ_{iso} and C_{Q} derived using the CASTEP code applied to the geometry optimised structure of naphthalaldehydic acid **1**. The DFT determined shifts are given by $\delta_{iso} = -[\sigma - \sigma_{ref}]$, where $\sigma_{ref} = 30$ ppm for ${}^{1}\text{H}$ and $\sigma_{ref} = 164$ ppm for ${}^{13}\text{C}$. The C_{Q} (kHz) values are given for all possible ${}^{2}\text{H}$ environments in **1** even though only one position (H8) was deuterated.

Site	$oldsymbol{\delta}_{ ext{iso}}$ calc. (ppm)	C _Q calc. (kHz)	Assignment
H1	1.7	678	-C ₁₂ ² D-OH
H2	4.6	416	aromatic
H3	5.1	421	aromatic
H4	5.0	423	aromatic
H5	4.6	453	aromatic
H6	4.2	423	aromatic
H7	4.7	425	aromatic
H8	6.6	305	$-C_{12}H-O^{2}D$
C1	111.3		aromatic
C2	119.2		aromatic
C3	116.3		aromatic
C4	117.4		aromatic
C5	122.2		aromatic
C6	113.9		aromatic
C7	111.2		aromatic
C8	112.5		aromatic
C9	123.2		aromatic
C10	118.2		aromatic
C11	155.7		C=O
C12	91.3		C-OH

Table S2. The computed ${}^{n}J({}^{13}C, {}^{17}O)$, ${}^{n}J({}^{13}C, {}^{13}C)$ and ${}^{n}J({}^{13}C, {}^{1}H)$ couplings to the C₁₁ and C₁₂ positions in solid naphthalaldehydic acid **1** as derived from the first principles GIPAW DFT calculations using the CASTEP code applied to the geometry optimised structure of naphthalaldehydic acid **1**.

Bond	Length (Å)	J _{iso} calc. (Hz)	J _{ansio} calc. (Hz)
C11 01	1.2219	27.42	27.47
C11 O2	1.3417	36.47	13.08
C11 C1	1.4801	57.76	4.42
C11 C2	2.4729	5.66	2.71
C11 C10	2.4966	5.19	2.25
C11 C12	2.5096	-1.8	-0.88
C11 H2	2.6309	3.68	1.78
C11 H32	2.7993	-0.36	-1.22
C11 H8	2.8600	1.79	0.85
C11 C9	2.8806	3.97	0.48
C11 H23	3.2628	0.17	0.85
C11 H25	3.2931	-0.22	-0.77
C11 O3	3.4671	-0.1	1.16
C11 O10	3.5656	-0.02	-0.25
C11 C35	3.7695	0.49	0.22
C11 C3	3.7722	3.75	0.48
C11 C37	3.7845	-0.05	-0.3
C11 C5	3.7967	2.77	0.31
C 11 H1	3.8540	-0.01	0.68
C11 H22	3.8662	0.09	0.41
C11 O11	3.9499	0.01	0.23
C11 C34	4.1073	0.04	0.15
C11 C8	4.1688	0.16	0.31
C11 C4	4.2633	-0.2	-0.14
C11 H7	4.4951	0.04	0.42
C11 H3	4.5279	1.28	0.48
C11 C36	4.6898	0.07	0.1
C11 O9	4.7727	-0.09	-0.15
C11 C7	4.8405	0.14	0.18
C12 H8	1.0002	149.83	30.38
C12 O3	1.3858	20.48	16.11
C12 O2	1.4766	31.65	16.78
C12 C9	1.5082	39.47	9.21
C12 H1	1.8441	0.94	9.21
C12 C8	2.4901	5.52	1.39
C12 C11	2.5096	-1.75	-0.6
C12 C10	2.5374	4.24	1.3
C12 H7	2.6356	3.18	1.61
C12 C1	2.9244	1.74	0.43
C12 011	3.1820	-1.13	-0.44
C12 012	3.4251	0	0.3
C12 H25	3.4980	0.01	-0.61
C12 O1	3.5884	0.19	1.87
C12 C38	3.7845	-0.05	-0.32

C12 C7	3.7985	3.06	0.42
C12 C5	3.8431	2.21	0.41
C12 H14	3.9169	0.09	0.4
C12 H32	3.9290	0.05	0.43
C12 C37	4.2499	0.08	0.2
C12 O10	4.30	0.03	0.15
C12 C2	4.3026	0.3	0.12
C12 C6	4.3097	-0.13	-0.04
C12 H27	4.4585	-0.03	-0.26
C12 H6	4.5468	1.05	0.3
C12 C22	4.6480	0.14	0.11
C12 H2	4.8362	0.04	0.21
C12 H23	4.9427	0.08	0.11
C12 C4	4.9808	0.43	0.13
C12 H15	5.0121	0	0.13
C12 H13	5.1148	0.02	0.05
C12 C3	5.1686	0.08	0.04
C12 C48	5.2560	0.04	0.06
C12 H5	5.2596	0.19	0.09
C12 C21	5.2923	0.03	0.04
C12 H22	5 3466	0.06	0.13
C12 C42	5.3587	0	-0.09
C12 C23	5 4309	0.03	0.07
C12 C40	5 4519	0.05	0.07
C12 C35	5 6624	0.05	0.05
C12 H28	5 6631	0.01	0.03
C12 H4	5 8277	0.36	0.1
C12 C34	5 8853	0.02	0.07
C12 C39	5 9710	0.02	0.08
C12 H31	5 9830	0.02	0.00
C12 C43	5 9957	0	-0.07
C12 H3	6.0922	0.06	0.11
C12 H26	6.0934	-0.05	-0.08
C12 C47	6 1487	0.01	0.04
C12 C41	6 2139	0.05	0.09
C12 09	6.4811	-0.02	-0.05
C12 C20	6.5280	0.02	0.02
C12 C24	6 6034	0.02	0.04
C12 H21	6.7665	0.06	0.06
C12 C36	6.7738	0.02	0.03
C12 C26	7.0582	0.02	0.03
C12 C15	7 1017	0.01	0.03
C12 C33	7,1369	0.03	0.03
C12 C44	7,3228	0.03	0.05
C12 06	7.3299	-0.01	-0.04
C12 H20	7.3431	0.05	0.04
C12 H12	7,3803	0.04	0.05
C12 C19	7,4689	0.01	0.02
			5.52

C12 C46	7.5194	0.01	0.02
C12 H16	7.5198	0.05	0.08
C12 C14	7.6195	0.02	0.03
C12 C27	7.9281	0.02	0.02
C12 C45	8.030	0.03	0.02
C12 H29	8.0581	0.03	0.05
C12 C32	8.0987	0.03	0.02
C12 O4	8.1239	0	0.02
C12 H30	8.1947	0.05	0.04
C12 C31	8.2290	0.03	0.01
C12 08	8.2807	0	-0.01
C12 C16	8.4147	0.01	0.02
C12 C18	8.6911	0.01	0.02
C12 05	8.7988	0	0.02
C12 H24	8.8875	0.04	0.04
C12 H10	8.9694	0.06	0.06
C12 H9	9.0007	0.04	0.05
C12 H19	9.0582	0.06	0.07
C12 C28	9.1136	0.02	0.02
C12 C17	9.1137	0.02	0.02
C12 C30	9.2124	0.02	0.01
C12 C13	9.2250	0.03	0.04
C12 C25	9.2935	0.02	0.02
C12 H11	9.3703	0.06	0.05
C12 H17	9.4421	0.05	0.06
C12 07	9.6422	0.01	0.01



Figure S1: The DFT predicted (C_{11} red, C_{12} blue, Simpson simulated) ${}^{13}C/{}^{17}O$ heteronuclear spin echo dephasing behaviour for (a) C_{11} and (b) C_{12} acquired at 11.7 T ($v_r = 11$ kHz) as a function of evolution time τ (0 to 32 ms), the 0-16 ms experimentally scaled version is given in table 6. These show that the cosine modulated feature appears at large τ delays.



Figure S2: The SIMPSON simulated DFT-derived ${}^{13}C/{}^{17}O$ or ${}^{13}C$ heteronuclear spin echo dephasing behaviour for each ${}^{1}J$ contribution for (a) C₁₁ and (b) C₁₂. Red = C₁₁=O, Green O-C_{11/12}-O, Blue C₁₂-OH, Purple C_{11/12}-C_{1/9} and grey T_2 '. Each decay has a normalised initial point ($\tau = 0$) to 1 and these do not consider the individual sites natural abundance.

Table S3: An example of the SIMPSON input used to simulate the ¹J-couplings.

spinsys { channels 13C 17O nuclei 13C 17O jcoupling 1 2 31.65 0 0 0 0 0 quadrupole 2 2 8.9e6 0.25 0 0 0 } par { proton_frequency 500e6 spin_rate 11000 sw 1000/2 np 1024 crystal_file rep30 gamma_angles 10 start_operator I1x detect_operator I1p verbose 1101 variable rf 83333 variable n 1 } proc pulseq {} { global par maxdt 1.0 set rf \$par(rf) set rf2 [expr \$rf/3] set t180 [expr 0.5e6/\$rf] set tr2 [expr 0.5e6/\$par(spin_rate)-\$t180] set tr [expr 1e6/\$par(spin_rate)] reset delay \$tr2 pulse \$t180 \$rf x \$rf2 x delay \$tr2 store 1 reset delay \$tr store 2 acq for {set i 2} {\$i < \$par(np)} {incr i} { reset prop 2 prop 1 delay \$tr store 1 acq } } proc main {} { global par fsave [fsimpson] \$par(name).fid }