

Supporting Information:

Temperature-Dependence of the Dielectric Relaxation of Water using Non-Polarizable Water Models

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Table S1: Calculated Kirkwood factors, rate constant for the hydrogen-bond breaking, and HB-bond energetics of water at 298K (19 non-polarizable water models).

Water Model	Kirkwood factors		Hydrogen-bond network dynamics			
	g_K	G_K	$\tau_{HB}[\text{ps}]$	$k_b[\text{ps}^{-1}]$	$n_b^{cycle}(\nu_D)$	$\Delta G[\text{kcal/mol}]$
three-point models						
SPC	2.515	3.744	6.003	0.167	4.94	2.144
SPC/E	2.499	3.722	6.888	0.145	5.17	2.226
SPC/E _B	2.468	3.676	8.742	0.114	5.36	2.367
SPC/FW	2.794	4.166	7.417	0.135	5.66	2.270
SPC-DC	2.693	4.015	6.306	0.159	4.97	2.173
TIP3P	3.605	5.381	3.658	0.273	5.88	1.851
TIP3PF	3.278	4.891	4.959	0.202	6.17	2.031
TIP3P-FB	2.678	3.993	7.670	0.130	5.93	2.289
H ₂ O-DC	2.670	3.980	7.340	0.136	5.37	2.263
OPC3	2.655	3.958	7.539	0.133	4.99	2.279
four-point models						
OPC	2.519	3.755	7.354	0.136	4.87	2.264
TIP4P	2.127	3.160	5.353	0.187	3.91	2.076
TIP4PEW	2.345	3.491	7.160	0.140	4.76	2.249
TIP4P-FB	2.545	3.793	8.493	0.118	4.59	2.350
TIP4P2005	2.121	3.154	7.966	0.126	4.31	2.312
TIP4Q	2.657	3.961	7.630	0.131	5.58	2.286
TIP4P/ ϵ	2.618	3.902	7.986	0.125	5.32	2.313
five-point models						
TIP5P	3.485	5.199	6.490	0.154	6.93	2.190
TIP5PEW	3.646	5.441	6.173	0.162	6.76	2.161

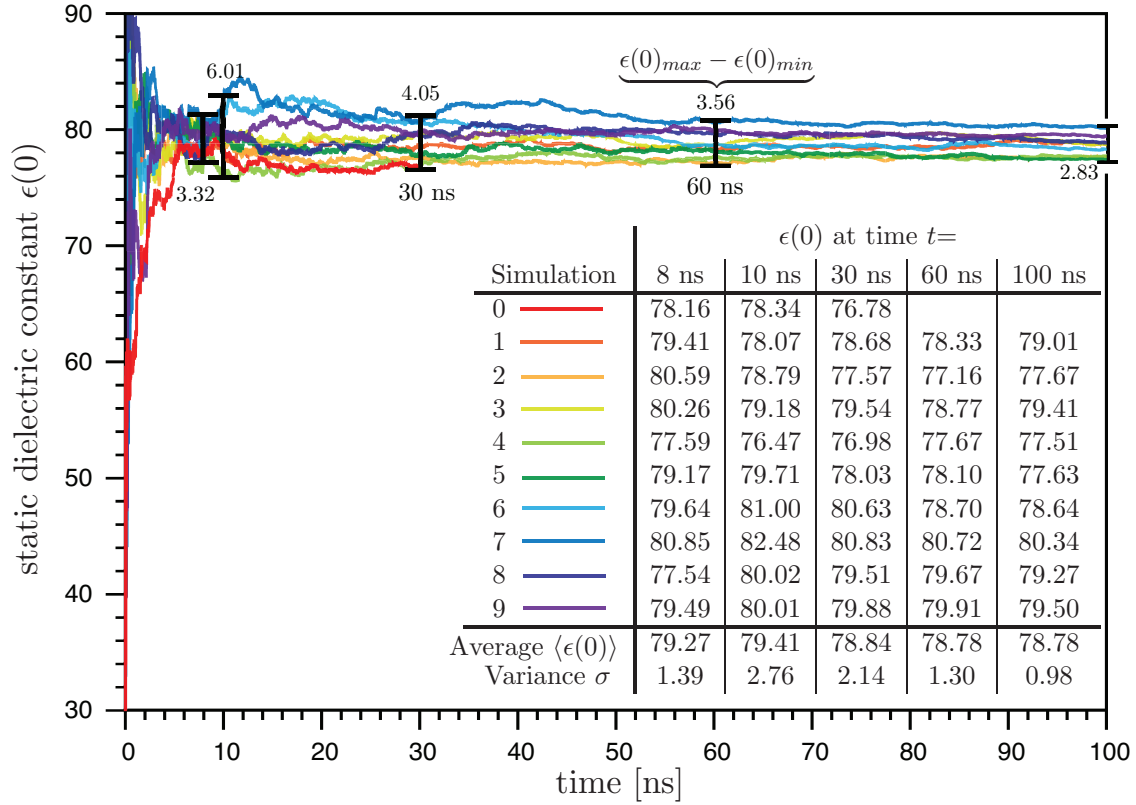


Figure S1: Slow convergence of the static dielectric constant (illustrated for the TIP4P-FB water model at 298 K). The results presented in tables are averages of the dielectric properties over last nanosecond of 30 ns simulations.

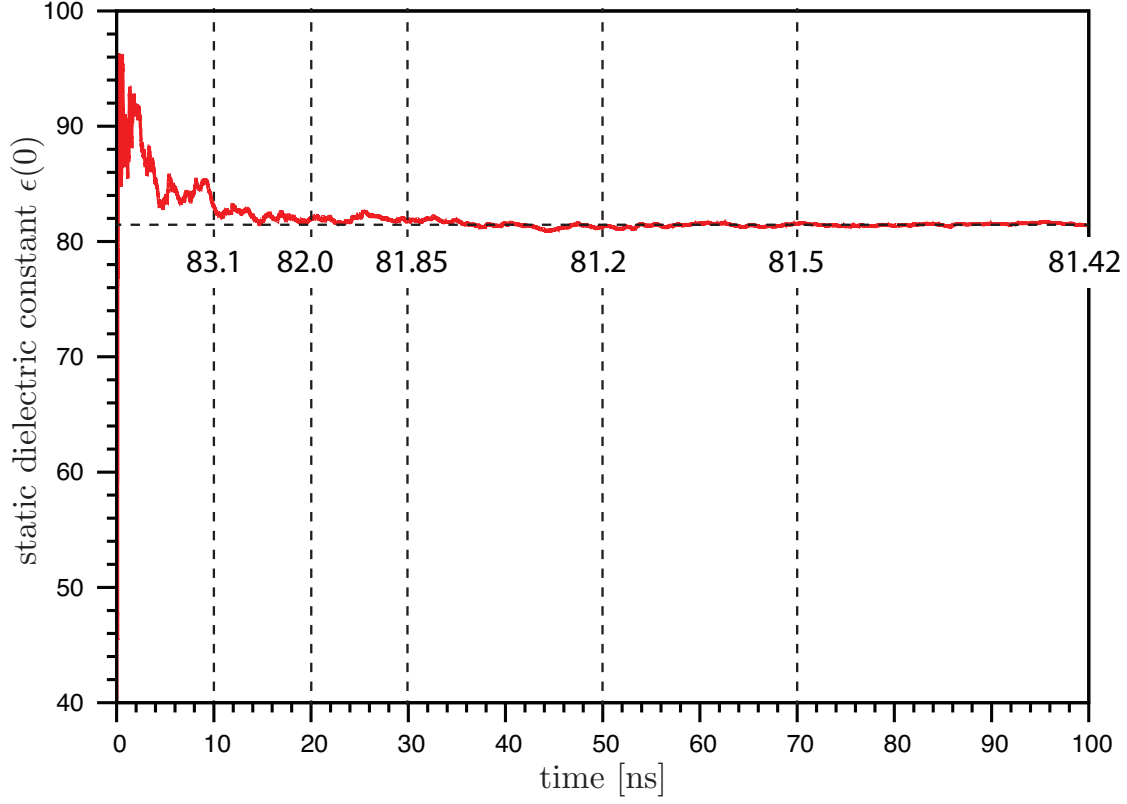


Figure S2: Slow convergence of the static dielectric constant (illustrated for the TIP3P-FB water model at 298 K).

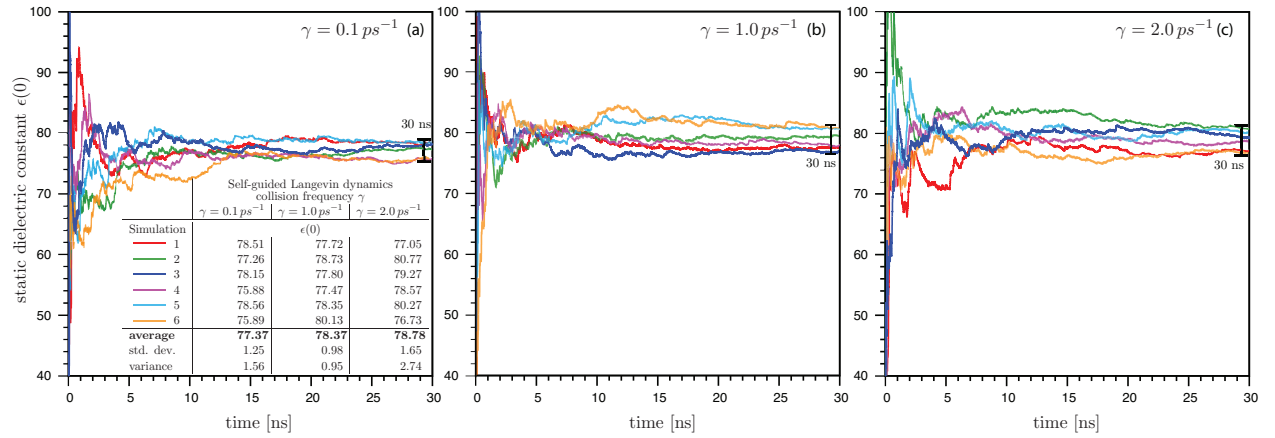


Figure S3: Stochastic variation in ϵ in Langevin dynamics for varying collision frequency γ in Langevin thermostat (TIP4P-FB water model): (a) 0.1 ps^{-1} , (b) 1.0 ps^{-1} , (c) 2.0 ps^{-1} .

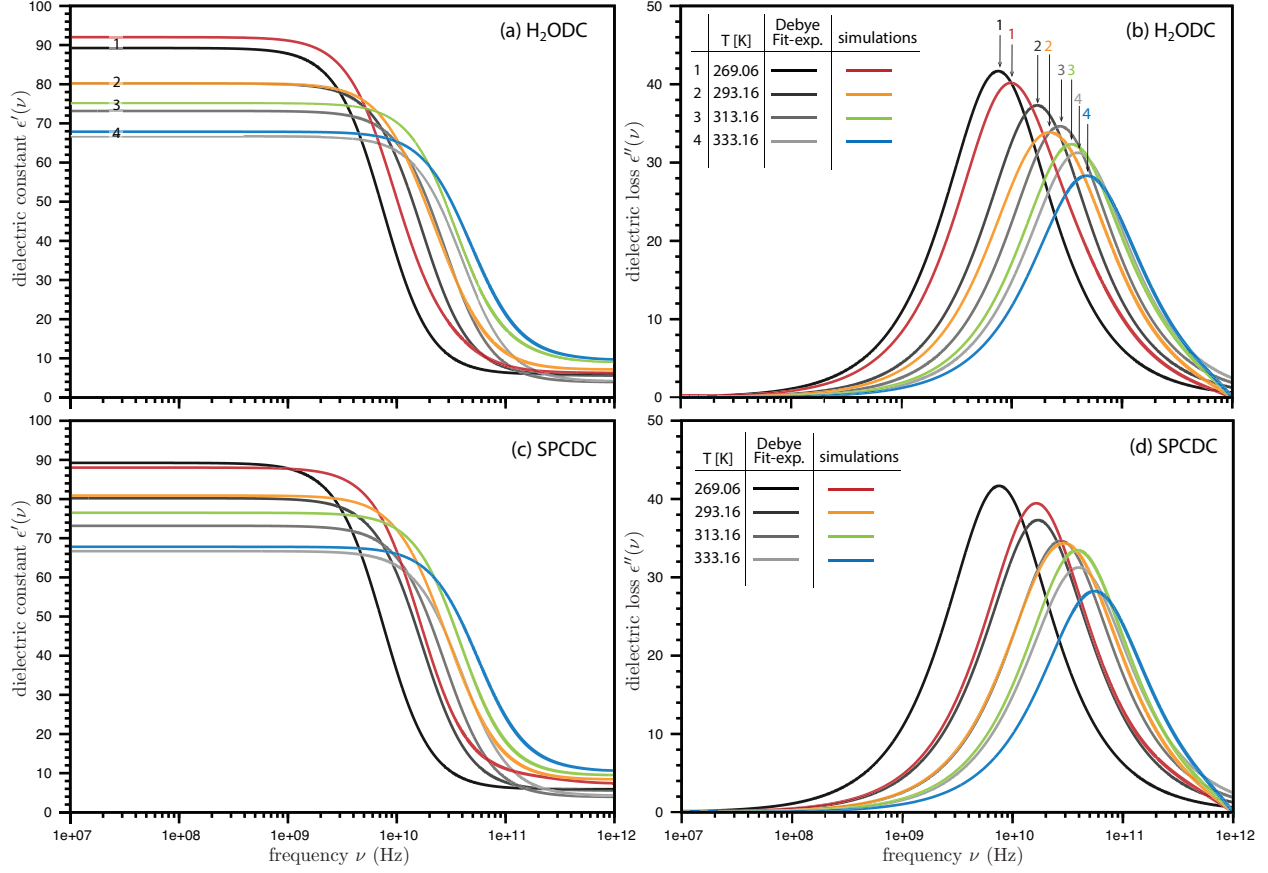


Figure S4: Comparison of the Debye function fit to the experimental data and the simulated dielectric spectra obtained for H₂ODC and SPCDC water models.^{S1} The difference between experimental and simulated static dielectric constants are within the simulation error (2-3%). The simulations predict the water relaxation is simulated at higher frequencies and shorter relaxation times (τ_D) than the Debye-function fitted to the experimental data.

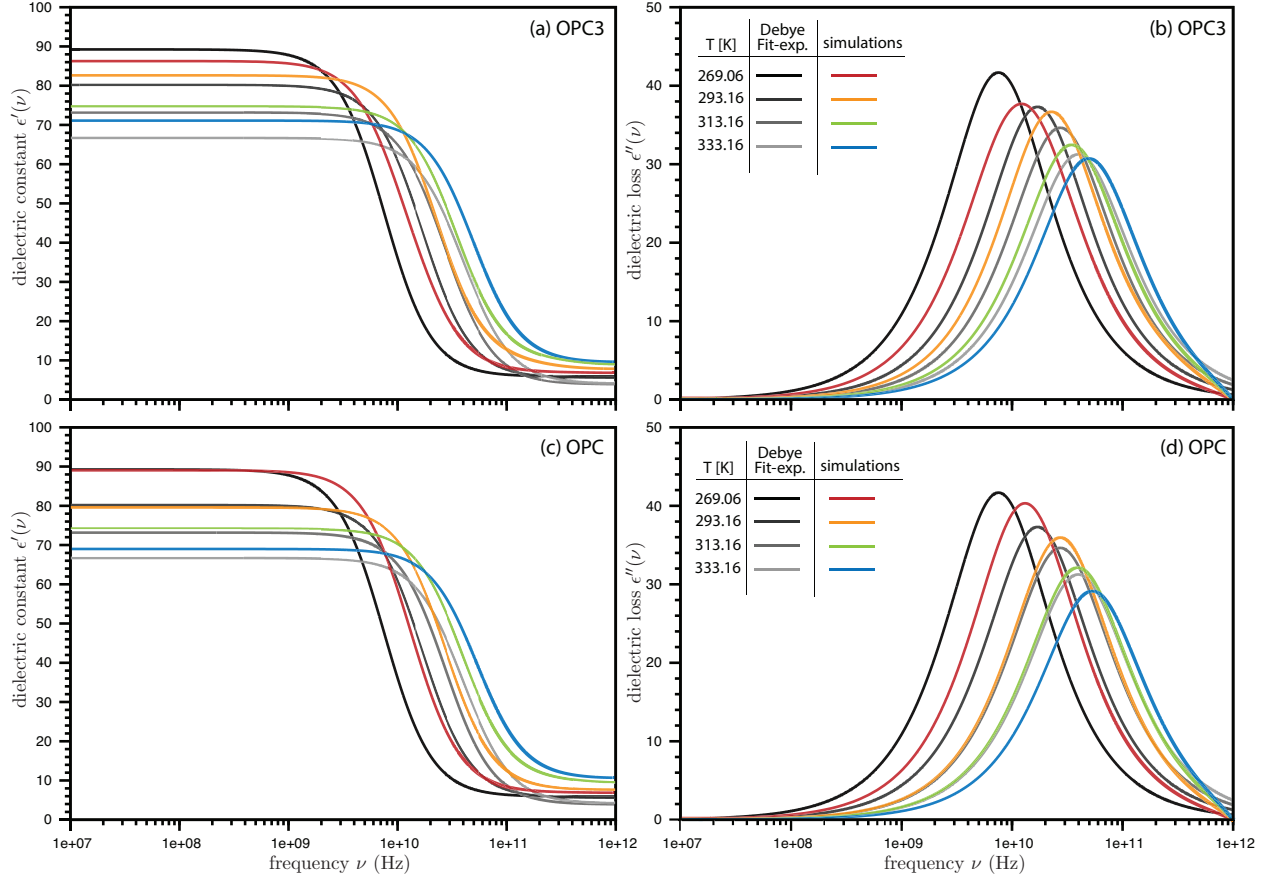


Figure S5: Comparison of the Debye function fit to the experimental data and the simulated dielectric spectra obtained for OPC3 and OPC water models.^{S2,S3} The difference between experimental and simulated static dielectric constants are within the simulation error (2-3%). The simulations predict the water relaxation is simulated at higher frequencies and shorter relaxation times (τ_D) than the Debye-function fitted to the experimental data.

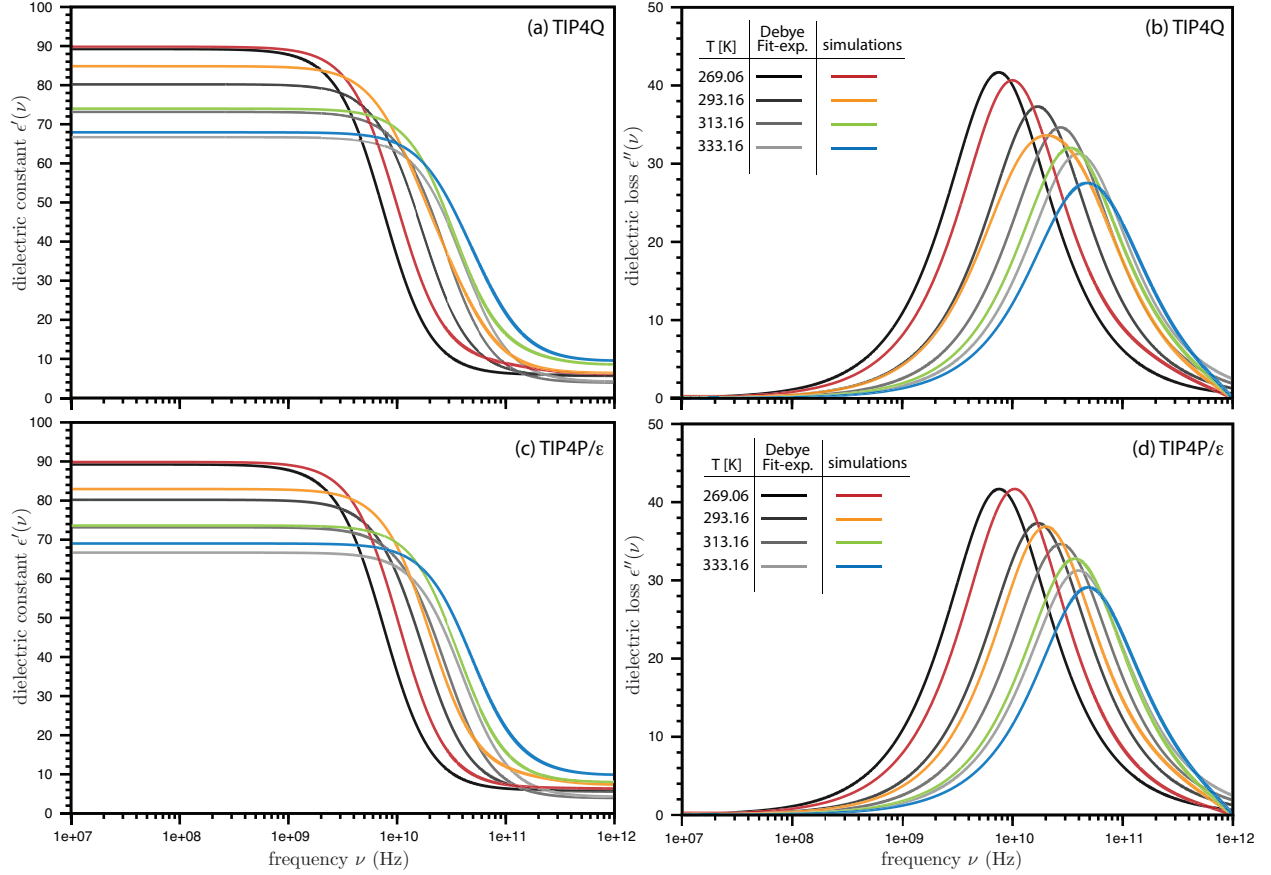


Figure S6: Comparison of the Debye function fit to the experimental data and the simulated dielectric spectra obtained for TIP3P-FB and TIP4P-FB water models.^{S4} The difference between experimental and simulated static dielectric constants are within the simulation error (2-3%). The simulations predict the water relaxation is simulated at higher frequencies and shorter relaxation times (τ_D) than the Debye-function fitted to the experimental data.

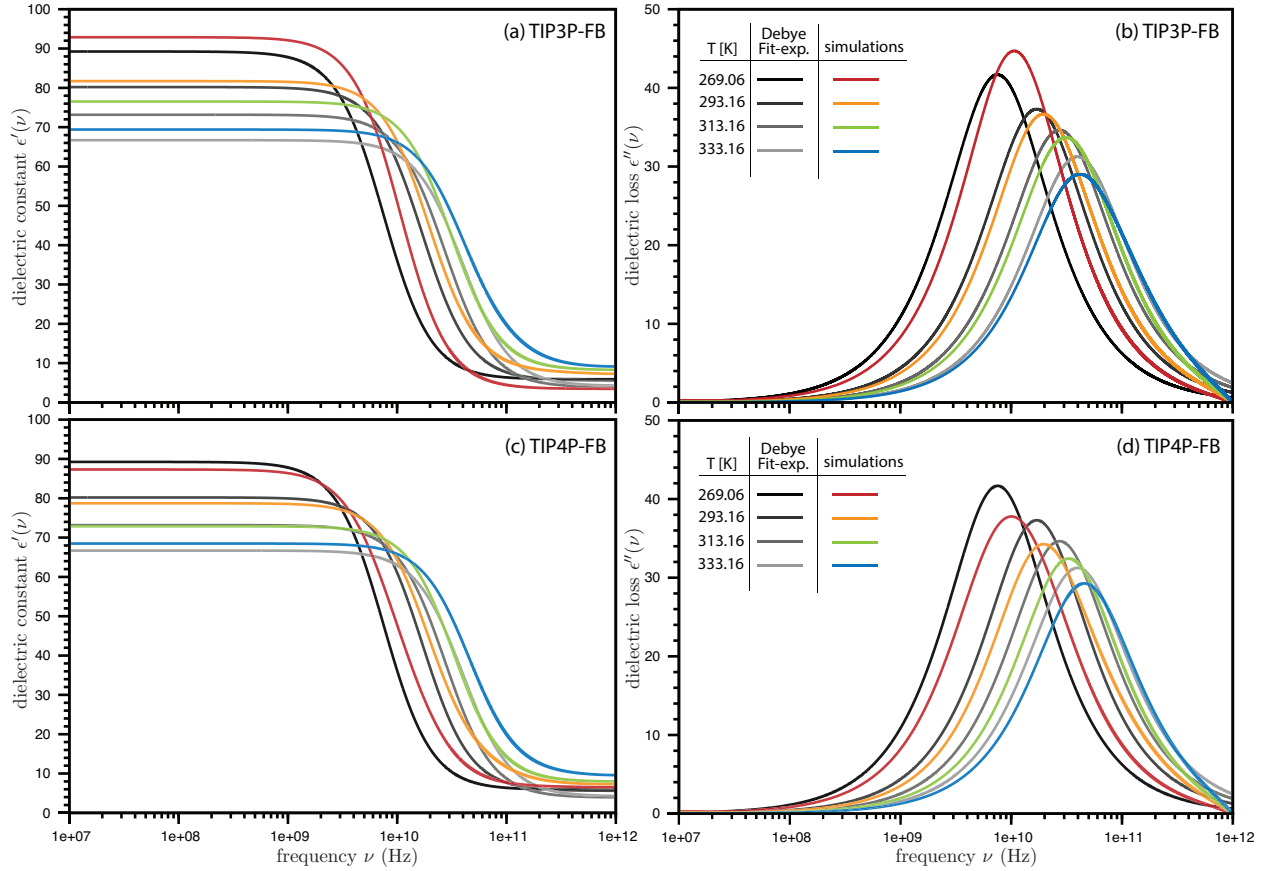


Figure S7: Comparison of the Debye function fit to the experimental data and the simulated dielectric spectra obtained for TIP4Q and TIP4P/ ϵ water models.^{S5,S6} The difference between experimental and simulated static dielectric constants are within the simulation error (2-3%). The simulations predict the water relaxation is simulated at higher frequencies and shorter relaxation times (τ_D) than the Debye-function fitted to the experimental data.

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