

**Supplementary information for**  
**“Quantifying the rigidity of 2D carbide (MXenes)”**

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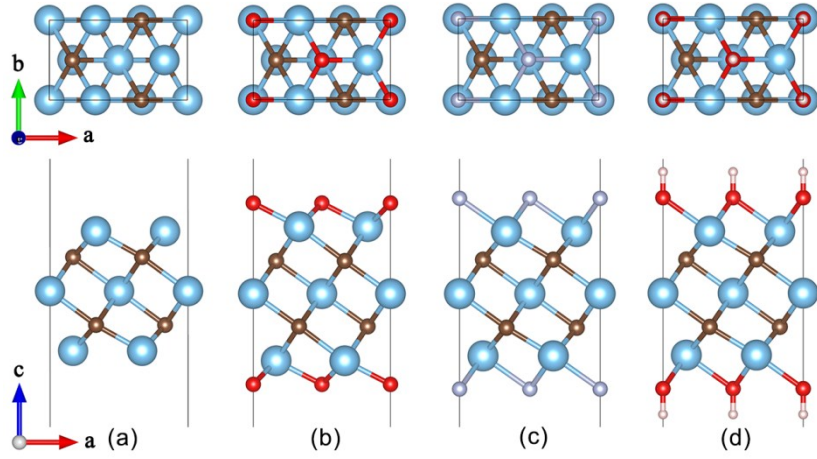
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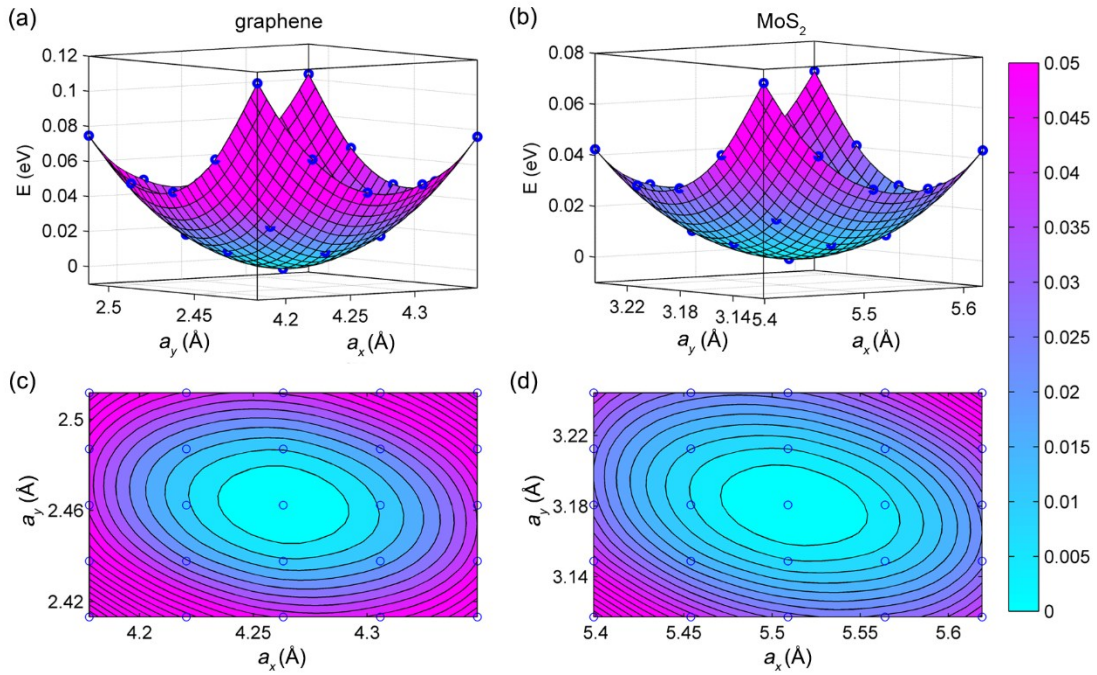
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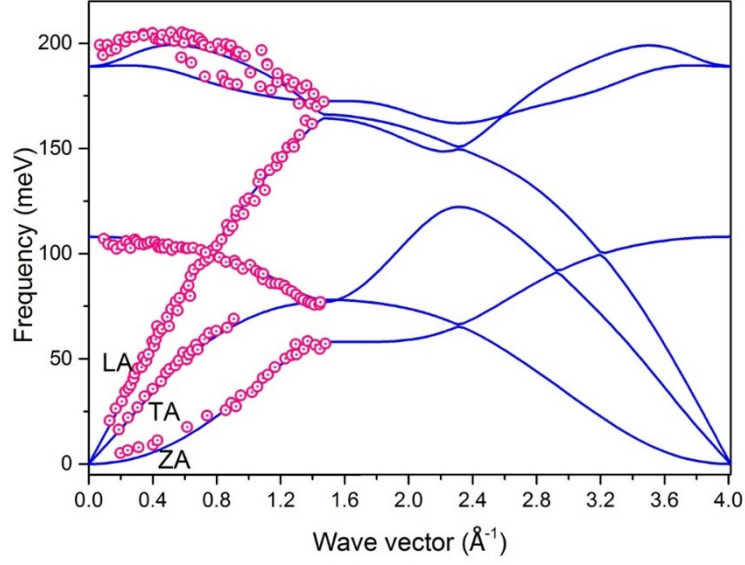
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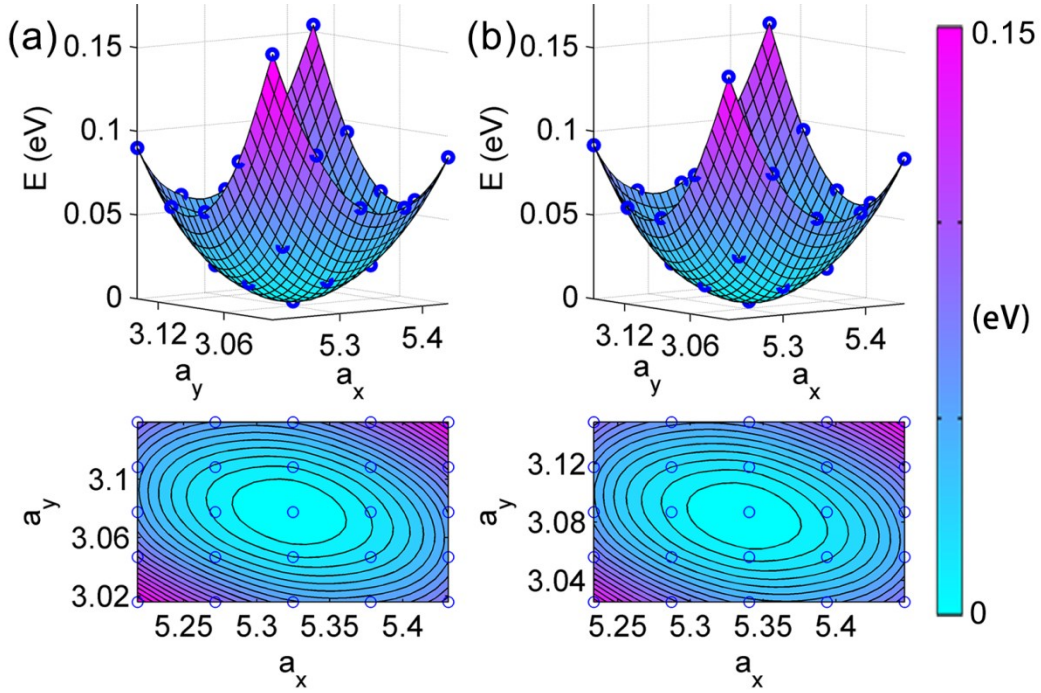
**Fig. S1** Atomic geometries of rectangular cell of  $\text{Ti}_3\text{C}_2\text{T}_2$  MXenes. (a)  $\text{Ti}_3\text{C}_2$  (b)  $\text{Ti}_3\text{C}_2\text{O}_2$  (c)  $\text{Ti}_3\text{C}_2\text{F}_2$  and (d)  $\text{Ti}_3\text{C}_2(\text{OH})_2$



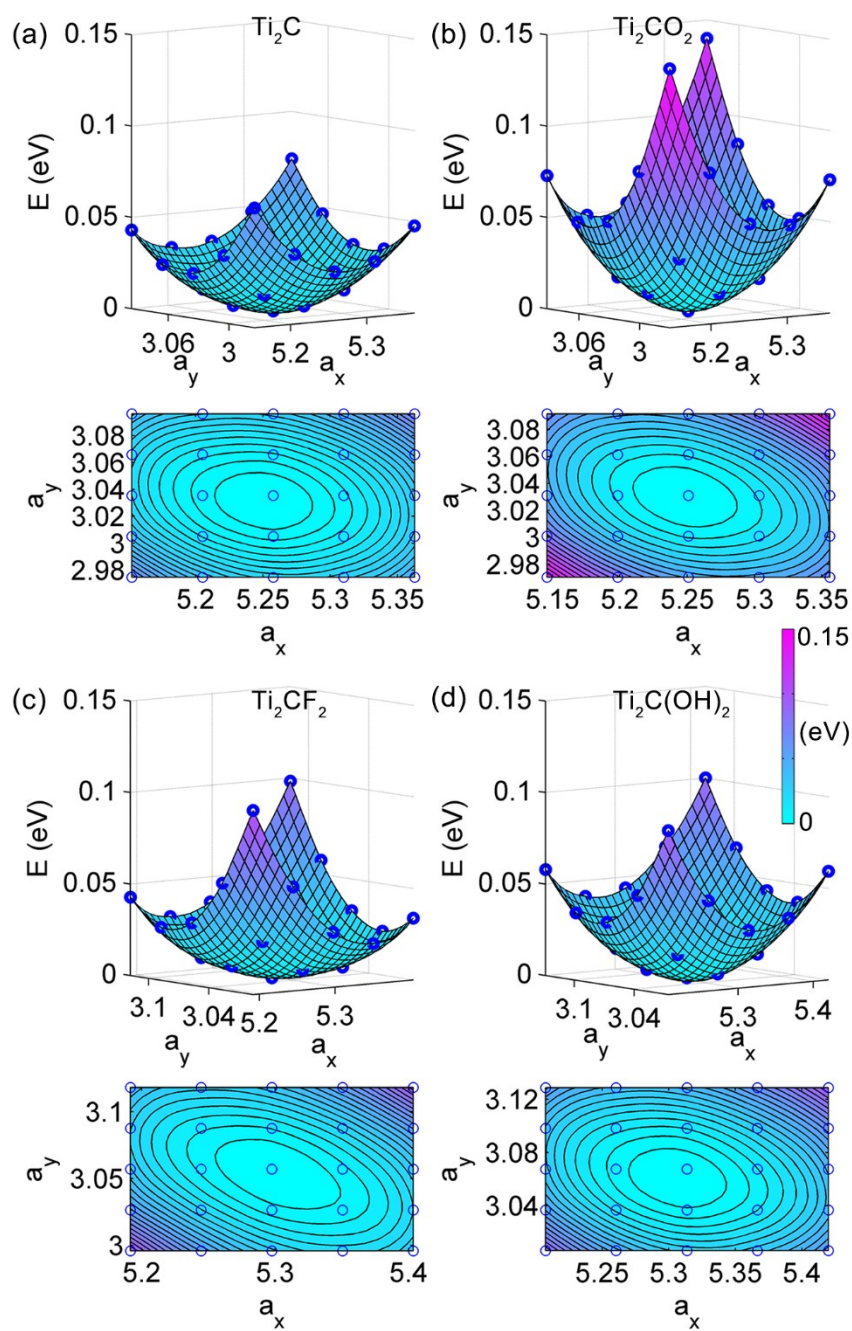
**Fig. S2** The three-dimensional plot of  $a_x$ ,  $a_y$  and corresponding strain energies of (a) graphene and (b)  $\text{MoS}_2$ . Contour of energy in the  $a_x$ - $a_y$  plane of (c) graphene and (d)  $\text{MoS}_2$ . For graphene,  $C = 337.8744$  N/m. This value is in a good agreement with earlier experimental and theoretical study. Our calculated value of the in-plane stiffness of graphene is in good agreement with the experimental value of  $340 \pm 50$  N/m and justifies the reliability of our method. For  $\text{MoS}_2$ ,  $C = 122.4$  N/m, agrees well with reported value of 123 N/m



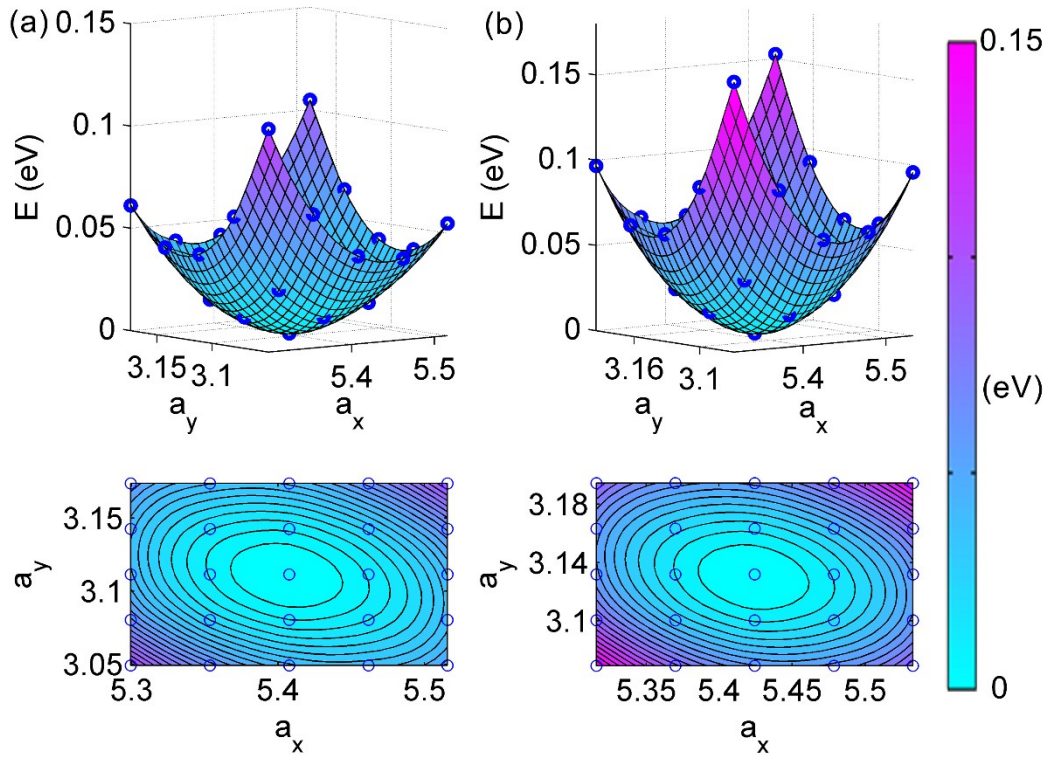
**Fig. S3** Calculated phonon dispersion in this work and experimental phonon dispersion of monolayer graphene by high-resolution electron energy loss spectroscopy<sup>1</sup>



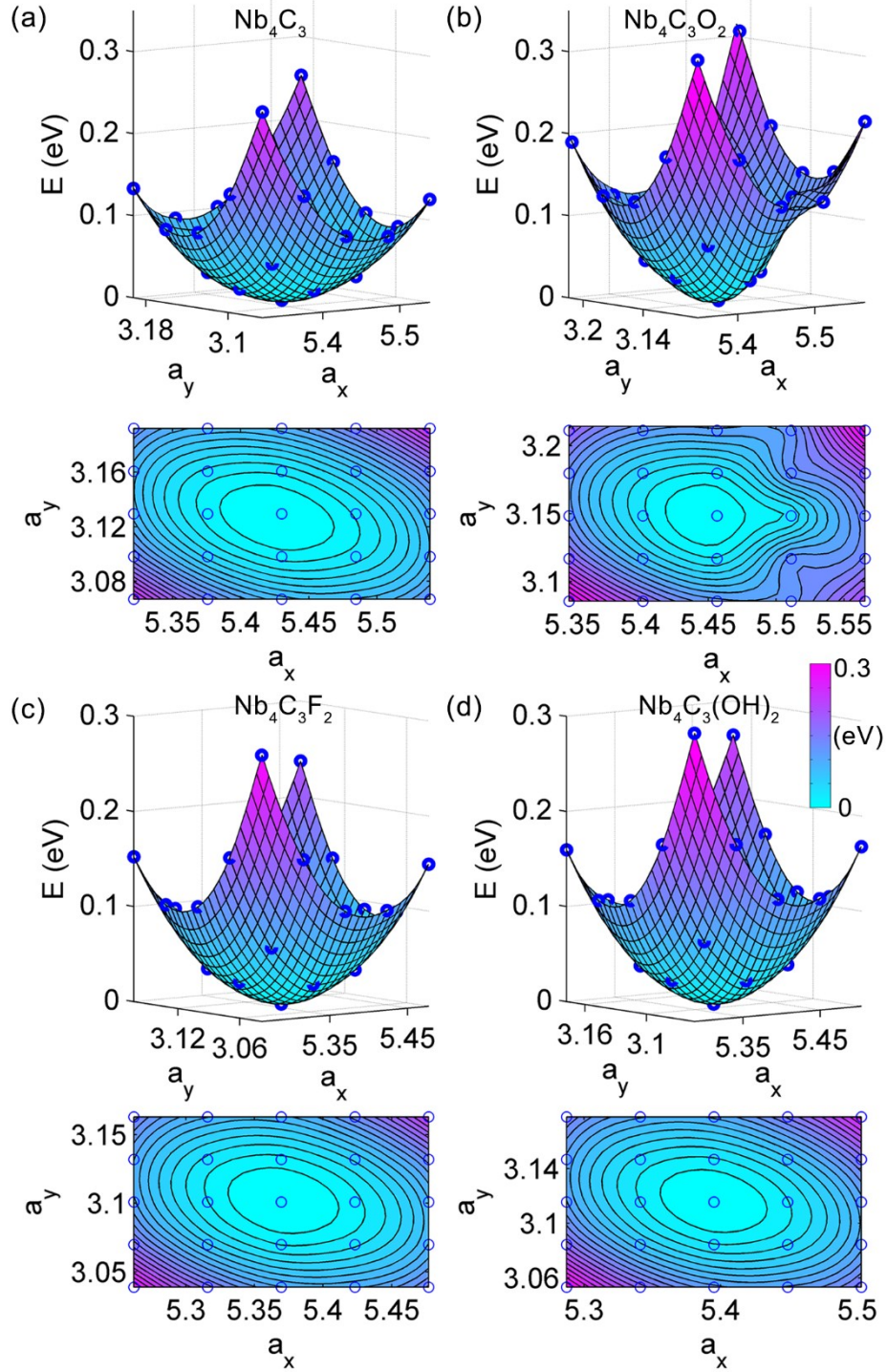
**Fig. S4** Energy surface of strained monolayer  $\text{Ti}_3\text{C}_2\text{T}_2$  MXenes for the calculation of in-plane stiffness. The three-dimensional plot of  $a_x$ ,  $a_y$  and corresponding strain energies and contour of energy in the  $a_x$ - $a_y$  plane of (a)  $\text{Ti}_3\text{C}_2\text{F}_2$  and (b)  $\text{Ti}_3\text{C}_2(\text{OH})_2$ . The blue balls are based on calculation and the lines are the fitted formula. Note that the stiffness increases by functionalization in  $\text{Ti}_3\text{C}_2$  MXenes



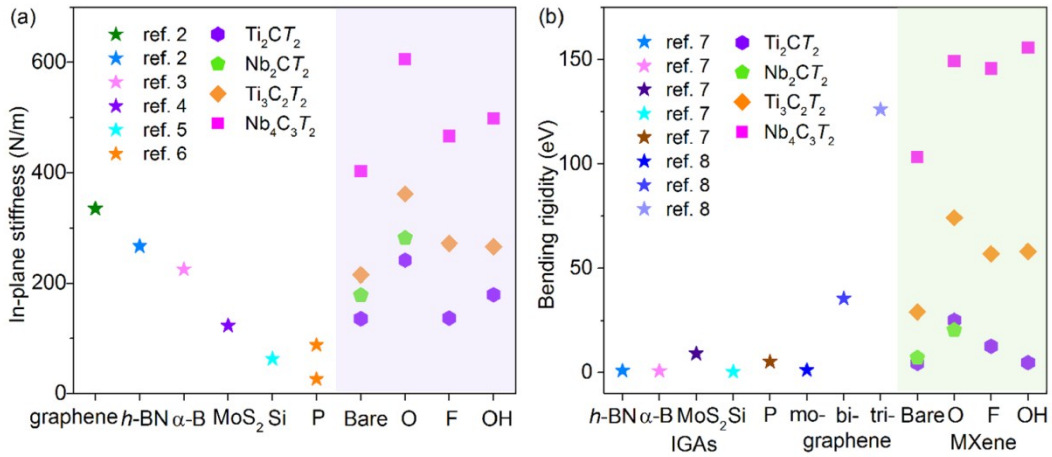
**Fig. S5** Stiffness of monolayer  $\text{Ti}_2\text{CT}_2$  MXenes. The three-dimensional plot of  $a_x$ ,  $a_y$  and corresponding strain energies and contour of energy in the  $a_x - a_y$  plane of (a)  $\text{Ti}_2\text{C}$ , (b)  $\text{Ti}_2\text{CO}_2$ , (c)  $\text{Ti}_2\text{CF}_2$ , and (d)  $\text{Ti}_2\text{C}(\text{OH})_2$ . The blue balls are actual points and the lines are the fitted formula. Note that the stiffness increases by functionalization in  $\text{Ti}_2\text{C}$  MXenes



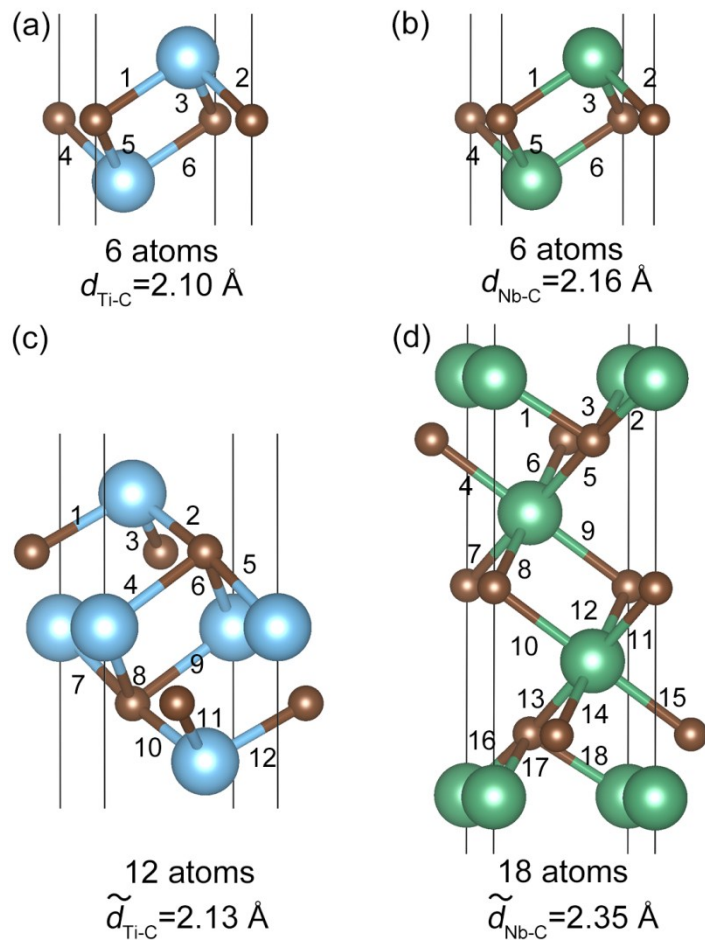
**Fig. S6** Stiffness of monolayer  $\text{Nb}_2\text{C}$  and  $\text{Nb}_2\text{CO}_2$  MXenes. The three-dimensional plot of  $a_x$ ,  $a_y$  and corresponding strain energies and contour of energy in the  $a_x - a_y$  plane of (a)  $\text{Nb}_2\text{C}$ , and (b)  $\text{Nb}_2\text{CO}_2$ . The blue balls are actual points and the lines are the fitted formula. Note that the stiffness increases by functionalization in  $\text{Nb}_2\text{C}$  MXenes



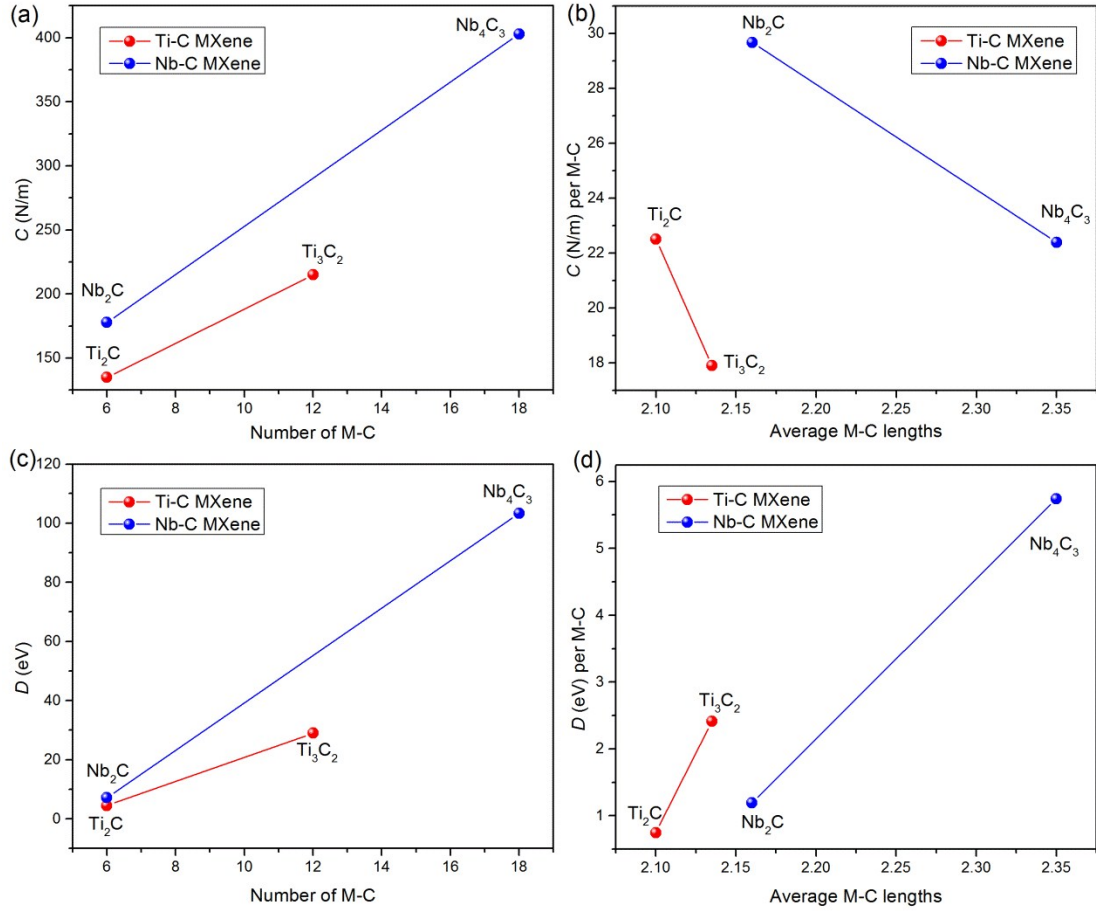
**Fig. S7** In-plane stiffness of monolayer  $\text{Nb}_4\text{C}_3\text{T}_2$  MXenes. The three-dimensional plot of  $a_x$ ,  $a_y$  and corresponding strain energies, and contour of energy in the  $a_x - a_y$  plane of (a)  $\text{Nb}_4\text{C}_3$ , (b)  $\text{Nb}_4\text{C}_3\text{O}_2$ , (c)  $\text{Nb}_4\text{C}_3\text{F}_2$ , and (d)  $\text{Nb}_4\text{C}_3(\text{OH})_2$ . The lengths of  $a_x$ ,  $a_y$  are in Å



**Fig. S8** (a) In-plane stiffness and (b) bending rigidity of MXenes and other typical 2D materials. In-plane stiffness of graphene <sup>2</sup>, *h*-BN <sup>2</sup>, Borophene <sup>3</sup>, MoS<sub>2</sub> <sup>4</sup>, Silicene <sup>5</sup>, phosphorene <sup>6</sup> and out-of-plane bending rigidities <sup>7,8</sup> are collected from literature

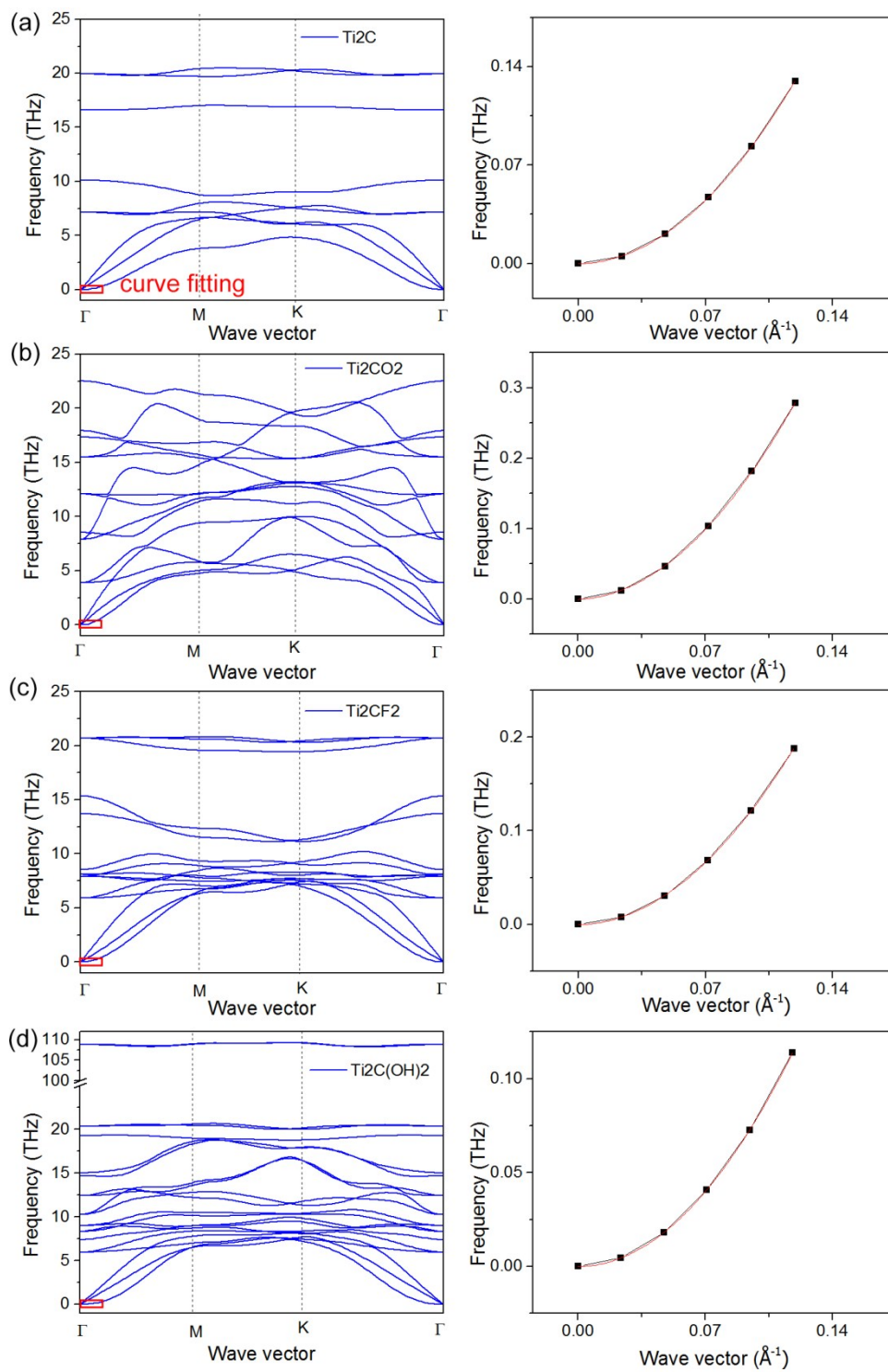


**Fig. S9** Schematic of number of M-C bonds and average M-C bond lengths in (a) Ti<sub>2</sub>C, (b) Nb<sub>2</sub>C, (c) Ti<sub>3</sub>C<sub>2</sub>, and (d) Nb<sub>4</sub>C<sub>3</sub> MXenes.

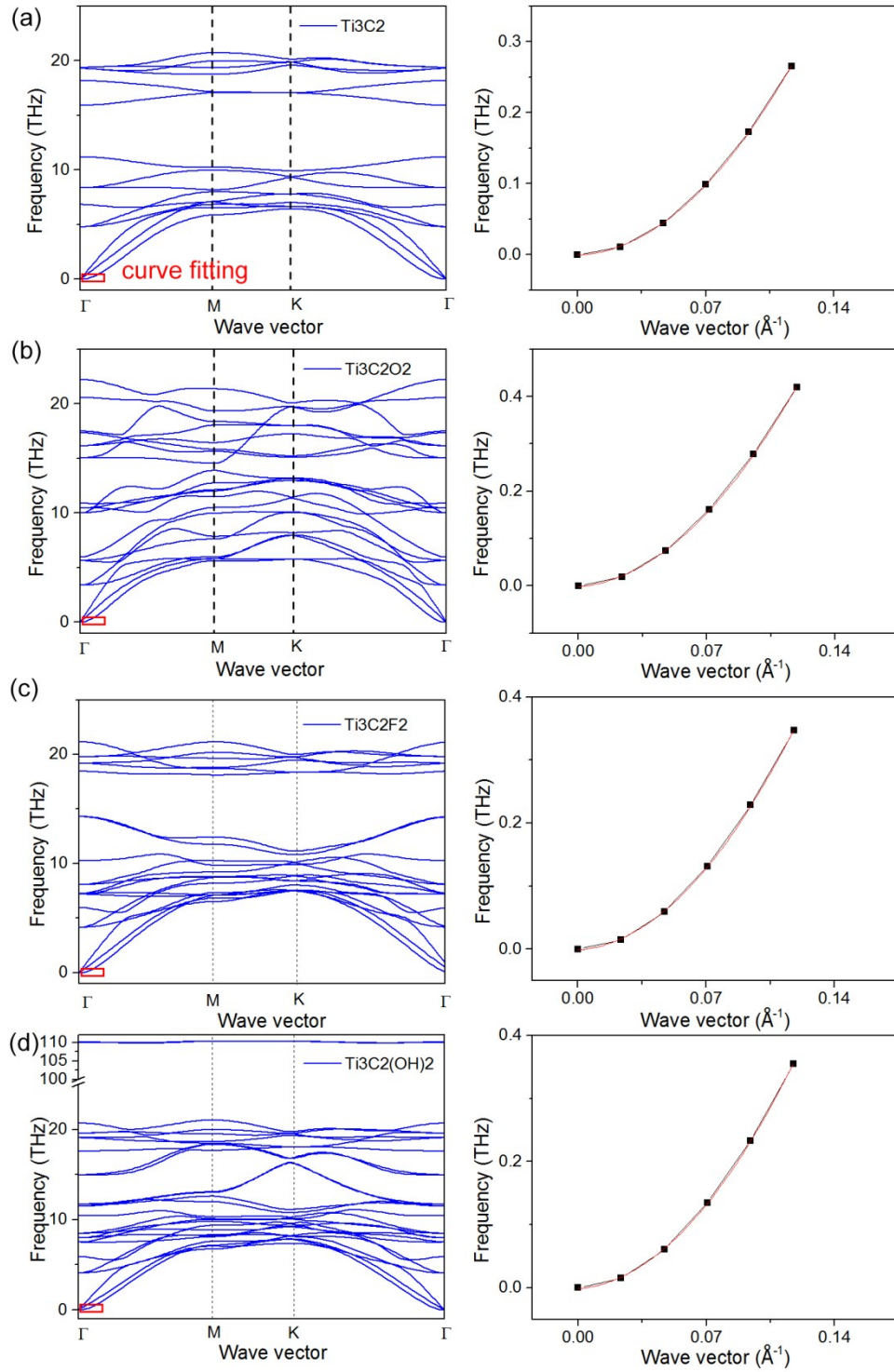


**Fig. S10** In-plane stiffness  $C$  and bending rigidity  $D$  versus number of M-C and average M-C bond lengths. The thickness-dependence of in-plane stiffness in MXenes can also be easily understood at microscopic scale if the chemical bonds are considered. The covalent/ionic mixing bonds M-C constitute the basic framework of MXenes considering the nearest neighbors. Increasing  $[M_{n+1}X_n]$  layer thickness means increasing the number of M-C bonds, and thus increasing the in-plane stiffness. The  $[M_{n+1}X_n]$  layer as the substrate provides the basic stiffness. Then the functional groups strengthen to different extents.

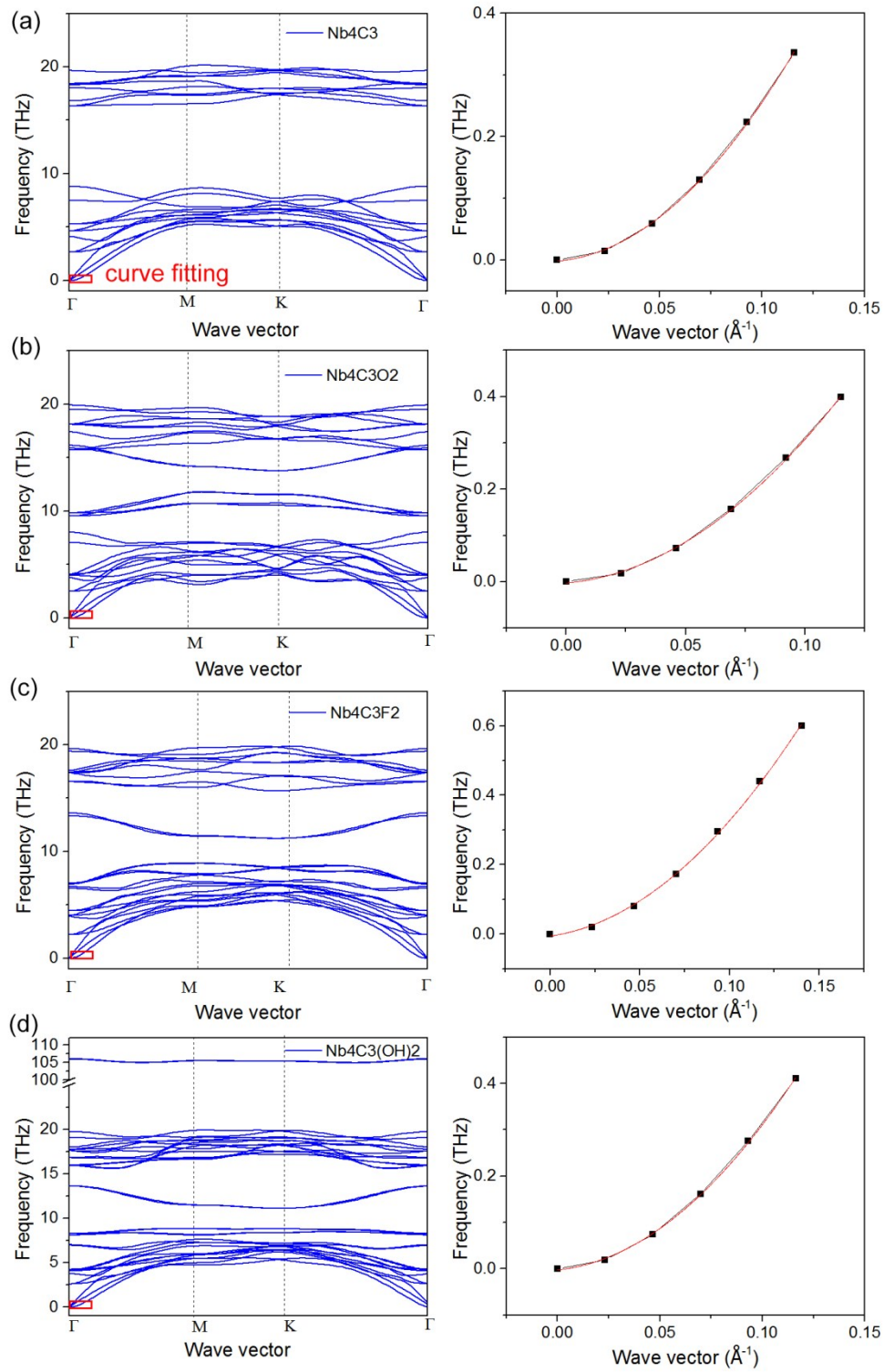




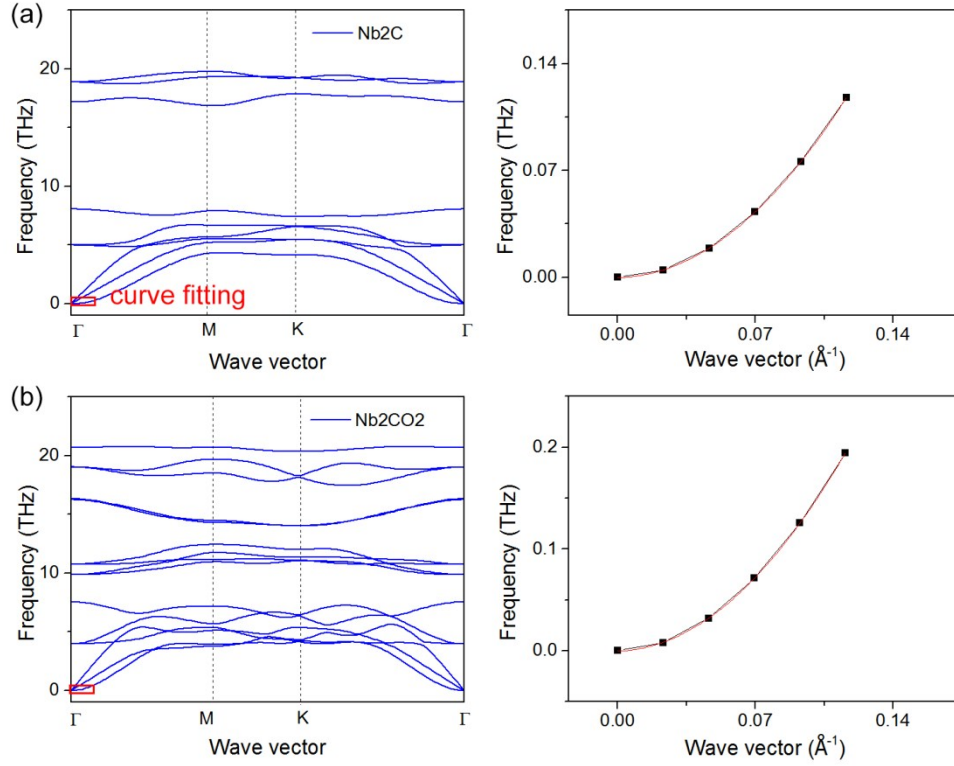
**Fig. S11** phonon dispersion and curve fitting details of  $\text{Ti}_2\text{CT}_2$  MXenes.



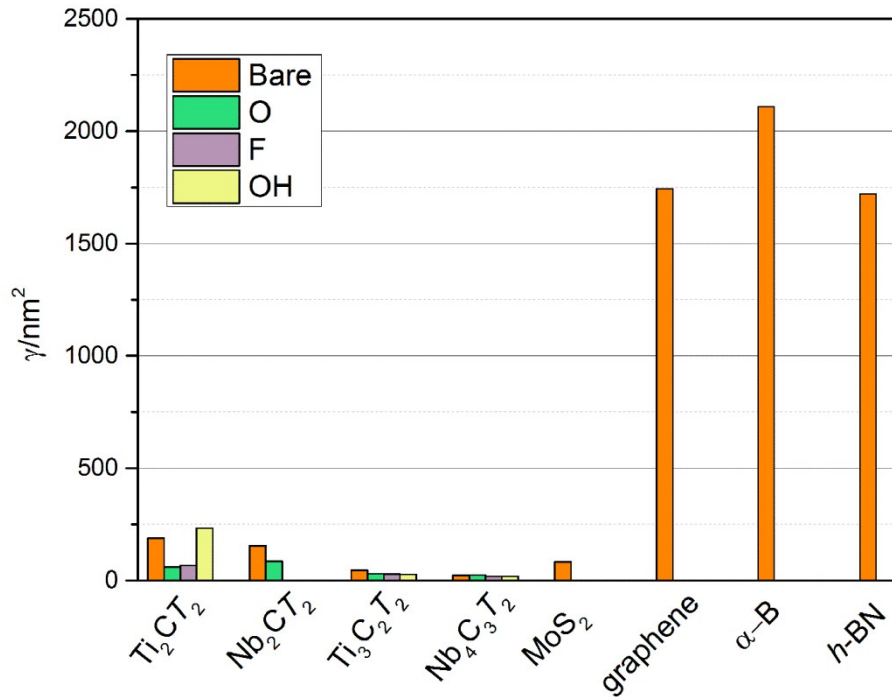
**Fig. S12** phonon dispersion and curve fitting details of  $Ti_3C_2T_2$  MXenes.



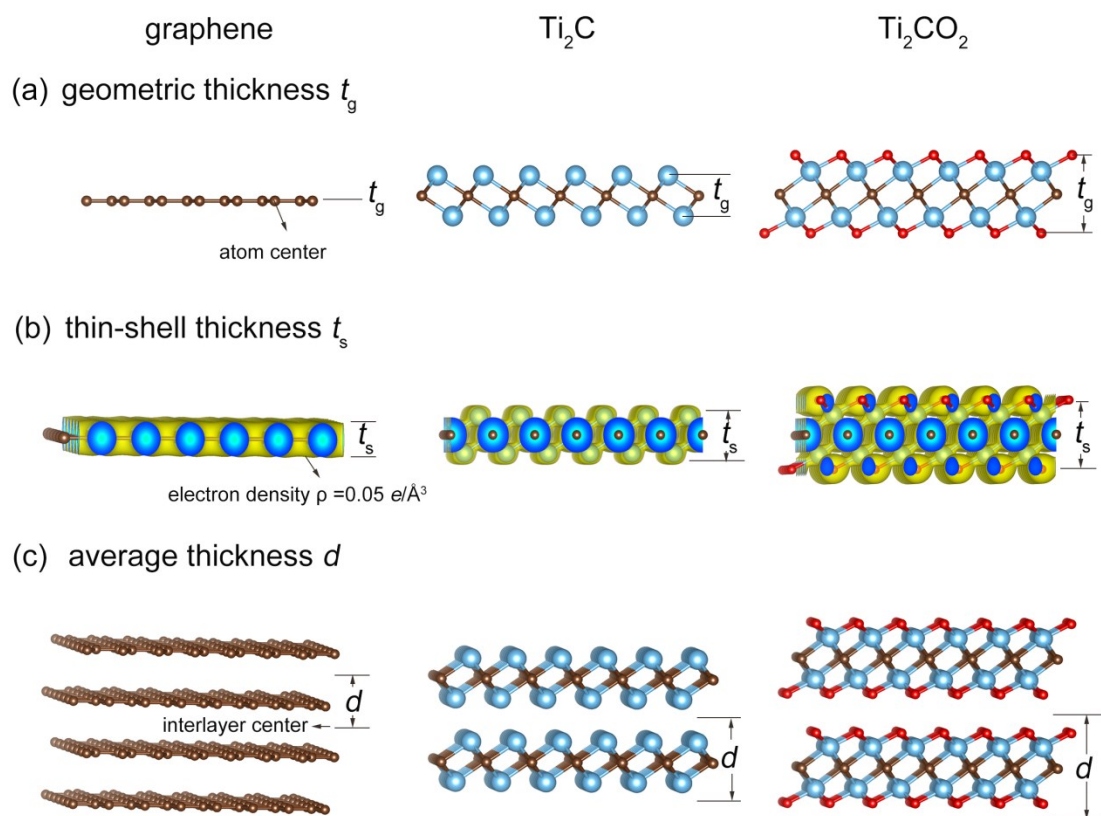
**Fig. S13** phonon dispersion and curve fitting details of  $\text{Nb}_4\text{C}_3\text{T}_2$  MXenes.



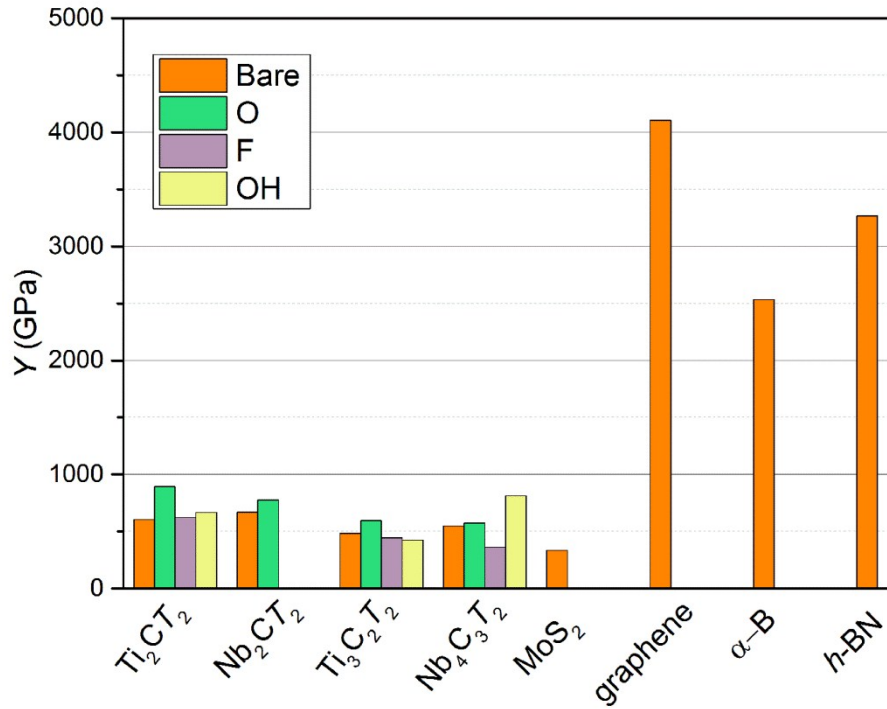
**Fig. S14** phonon dispersion and curve fitting details of Nb<sub>2</sub>C and Nb<sub>2</sub>CO<sub>2</sub> MXenes.



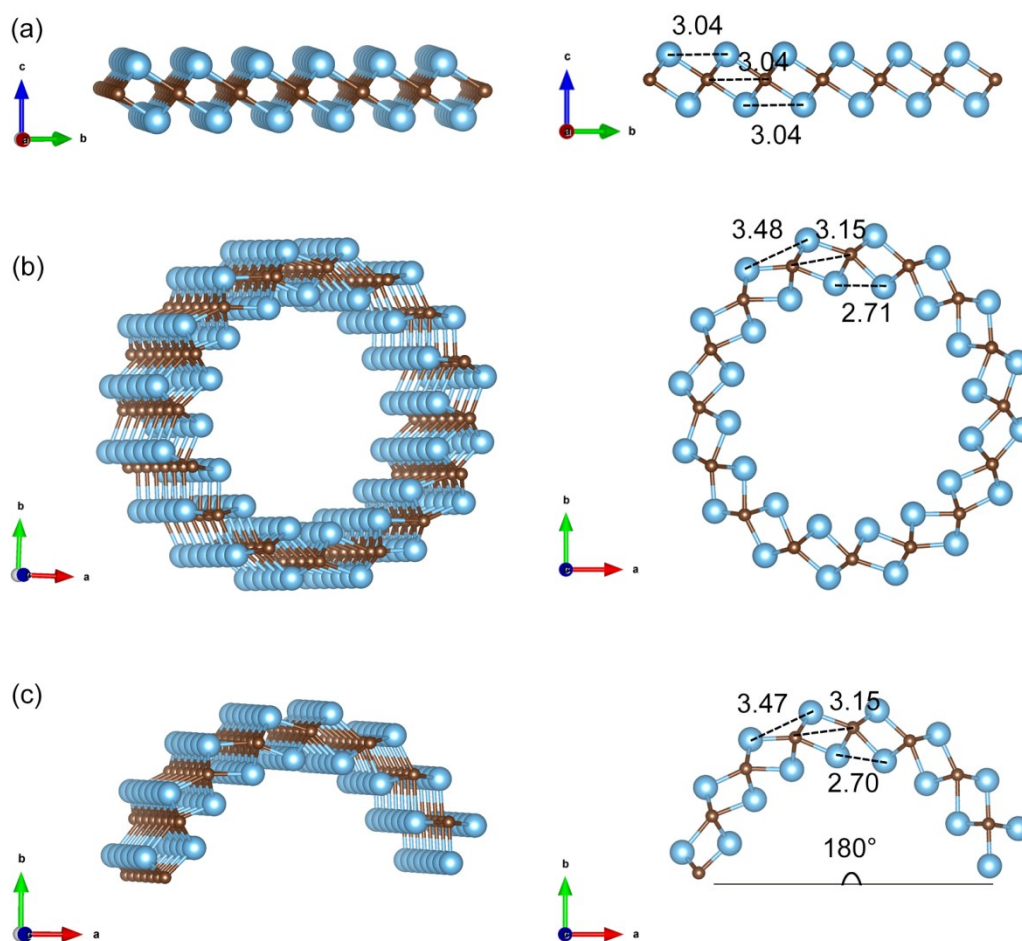
**Fig. S15** Foppl-von Karman number per unit area of MXenes and typical 2D materials.  $\gamma$  of MXenes is much lower than that of single-atom-thick graphene. Available data in the literature demonstrates the reliability of this study



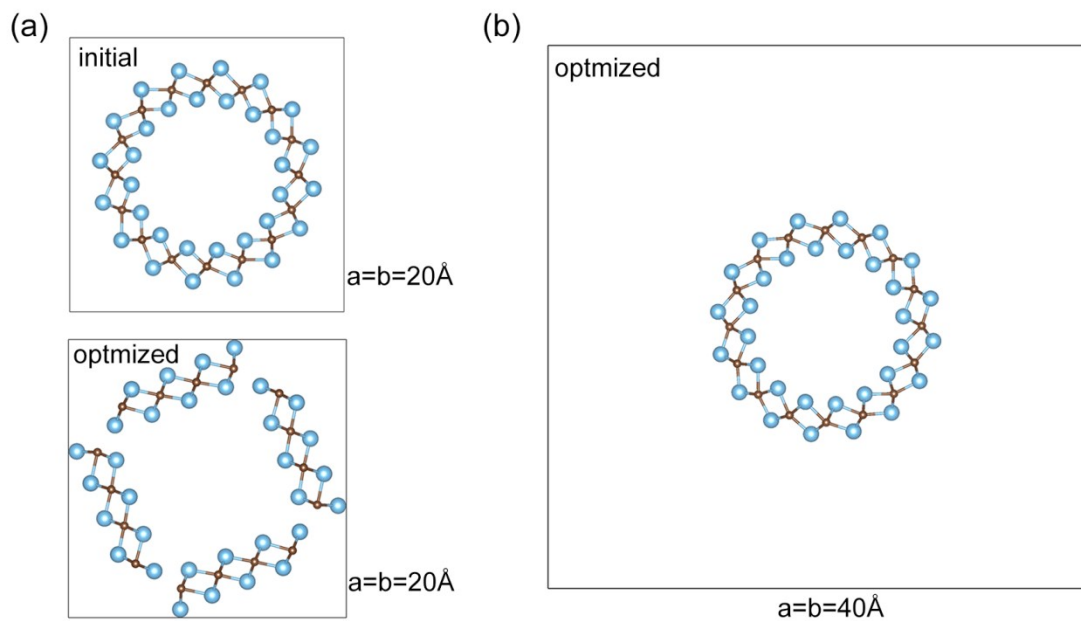
**Fig. S16** Schematic of (a) geometric thickness  $t_g$ , (b) thin-shell thickness  $t_s$ , and (c) average thickness  $d$ . Compared with geometric thickness, electrons contribute to the thickness in the thin-shell thickness like ‘glue’. Isosurface of electron density of  $0.05 \text{ e}/\text{\AA}^3$  is presented in (b)



**Fig. S17** Young's moduli  $Y$  of MXenes and typical 2D materials. We note that the Young's modulus of graphene calculated by  $Y = C/t_s$  is 4.1 TPa, about four-fold of 1.0 TPa, which was obtained assuming effective thickness equal to average layer thickness of graphite 3.35 Å.<sup>9</sup> By more accurate micro Raman spectroscopy measurement, Young's modulus of graphene was estimated to be more than 2 TPa.<sup>10</sup> The ultrahigh Young's modulus of graphene is attributed to the lack of  $\sigma$  bond participation under bending.<sup>11</sup> And the authors pointed out that the plate phenomenology is filled in multilayer (number of layer  $N=3$  with only 6% error). As the ambiguous nature of thickness of atomic thick materials, in-plane stiffness is more appropriate parameter than the Young's modulus.



**Fig. S18** Optimized structures of (a)  $\text{Ti}_2\text{C}$  MXene monolayer, a  $6 \times 6$  supercell (b) A  $\text{Ti}_2\text{C}$  (8, 8) nanotube with 48 atoms/unit cell (c) A  $\text{Ti}_2\text{C}$  sheet bend to 180 semi-tube, with C atoms fixed while relaxing Ti atoms. Note that the bent MXenes behaviors like a plate, with uniform deformation, involving extension on the convex while compression on the concave side. The result shows that the shell generally satisfy the  $D \sim N^3$  (N represents number of atomic layers) relationship.<sup>12</sup> For MXene, we did not observe the lubricated sliding. Instead, the uniform deformation is observed in MXene. The outer diameter and inner diameter of the optimized  $\text{Ti}_2\text{C}$  (8, 8) nanotube are 16.0 Å and 11.5 Å respectively, in good agreement with earlier work.<sup>13</sup>



**Fig. S19** Structural optimization of  $\text{Ti}_2\text{C}$  (8, 8) nanotube with 48 atoms/unit cell with (a) a  $20.00 \text{ \AA} \times 20.00 \text{ \AA} \times 3.03 \text{ \AA}$  box, and (b)  $40.00 \text{ \AA} \times 40.00 \text{ \AA} \times 3.03 \text{ \AA}$  box. By using sufficient large box like  $40.00 \text{ \AA} \times 40.00 \text{ \AA} \times 3.03 \text{ \AA}$  box, the disintegration of tubular structure into a bundle of nanostripes, which is also observed in  $\text{Ti}_2\text{C}$  (ref. <sup>14</sup>) and  $\text{Sc}_2\text{C}$  (ref. <sup>15</sup>) did not occur.



**Table S1.** A summary of parameters fitted through the energy–strain curved surface of monolayer MXenes

MXenes	$c_1$	$c_2$	$c_3$
Ti <sub>2</sub> C	70.32	73.63	30.26
Ti <sub>2</sub> CO <sub>2</sub>	133.70	46.12	85.15
Ti <sub>2</sub> CF <sub>2</sub>	86.48	46.12	77.58
Ti <sub>2</sub> C(OH) <sub>2</sub>	97.17	94.45	46.12
Ti <sub>3</sub> C <sub>2</sub>	114.90	123.10	114.90
Ti <sub>3</sub> C <sub>2</sub> O <sub>2</sub>	193.50	189.1	103.8
Ti <sub>3</sub> C <sub>2</sub> F <sub>2</sub>	151.00	156.00	84.87
Ti <sub>3</sub> C <sub>2</sub> (OH) <sub>2</sub>	147.00	152.80	77.12
Nb <sub>4</sub> C <sub>3</sub>	238.10	241.40	153.00
Nb <sub>4</sub> C <sub>3</sub> O <sub>2</sub>	335.80	269.40	120.90
Nb <sub>4</sub> C <sub>3</sub> F <sub>2</sub>	259.60	255.10	133.80
Nb <sub>4</sub> C <sub>3</sub> (OH) <sub>2</sub>	281.70	279.00	151.80
Nb <sub>2</sub> C	102.30	103.90	59.86
Nb <sub>2</sub> CO <sub>2</sub>	157.90	157.00	73.55
graphene	114.50	114.90	40.95
MoS <sub>2</sub>	71.35	72.31	35.57

**Table S2.** Theoretical in-plane stiffness  $C$ , out-of-plane bending rigidity  $D$ , Foppl-von Karman number per unit area,  $\gamma$ , effective shell thickness  $t_s$ , average layer thickness  $d$ , and geometric layer thickness  $t_g$  of MXenes and typical 2D materials. Note that the Termination-Termination interaction within one layer is quite weak, that they contribute little to the thickness of the shell in functionalized MXenes. This is in agreement with the  $[M_{n+1}X_n]$  provide the framework and basic in-plane stiffness.

formula	$C$ (N/m)	$D$ (eV)	$\gamma$ (nm <sup>-2</sup> )	$\nu$	$t_s$ (Å)	$d$ (Å)	$t_g$ (Å)
Ti <sub>2</sub> C	135.53	4.47	189.50	0.21	2.46	4.86	2.30
Ti <sub>2</sub> CO <sub>2</sub>	241.86	25.02	60.42	0.32	4.22	6.77	4.42
Ti <sub>2</sub> CF <sub>2</sub>	136.67	12.65	67.52	0.45	3.76	6.95	4.78
Ti <sub>2</sub> C(OH) <sub>2</sub>	179.12	4.81	232.74	0.24	2.20	8.84	6.73
Nb <sub>2</sub> C	178.13	7.17	155.27	0.29	2.66	4.93	2.36
Nb <sub>2</sub> CO <sub>2</sub>	281.65	20.45	86.08	0.23	3.63	6.89	4.64
Ti <sub>3</sub> C <sub>2</sub>	214.85	28.99	46.32	0.50	4.41	7.35	4.63
Ti <sub>3</sub> C <sub>2</sub> O <sub>2</sub>	361.42	74.29	30.41	0.27	6.05	9.29	8.94
Ti <sub>3</sub> C <sub>2</sub> F <sub>2</sub>	271.93	56.95	29.84	0.28	6.09	9.35	7.21
Ti <sub>3</sub> C <sub>2</sub> (OH) <sub>2</sub>	266.05	58.08	28.63	0.26	6.25	9.64	9.22
Nb <sub>4</sub> C <sub>3</sub>	402.59	103.34	24.35	0.32	6.65	10.26	7.59
Nb <sub>4</sub> C <sub>3</sub> O <sub>2</sub>	605.99	149.15	25.39	0.18	6.76	12.60	9.82
Nb <sub>4</sub> C <sub>3</sub> F <sub>2</sub>	466.38	145.65	20.01	0.26	7.48	13.31	10.65
Nb <sub>4</sub> C <sub>3</sub> (OH) <sub>2</sub>	497.97	155.70	19.99	0.27	7.46	14.25	12.52
MoS <sub>2</sub>	123	9.14	84.11	0.25	3.66	6.15	3.12
graphene	335	1.20	1744.79	0.18	0.82	3.34	0

**Table S3** Optimized multilayer structures by PW91-OBS scheme in the determination of average layer thickness  $d$ . For  $\text{Ti}_2\text{CT}_2$  and  $\text{Nb}_2\text{CT}_2$ , one unit cell includes only one layer, while for  $\text{Ti}_3\text{C}_2\text{T}_2$  and  $\text{Nb}_4\text{C}_3\text{T}_2$ , one unit cell includes 2 layers.

formula	Stacking type	$c$ (Å)	$d$ (Å)
$\text{Ti}_2\text{C}$	Bernal	4.86	4.86
$\text{Ti}_2\text{CO}_2$	Bernal	6.77	6.77
$\text{Ti}_2\text{CF}_2$	Bernal	6.95	6.95
$\text{Ti}_2\text{C}(\text{OH})_2$	Bernal	8.84	8.84
$\text{Nb}_2\text{C}$	Bernal	4.93	4.93
$\text{Nb}_2\text{CO}_2$	Bernal	6.89	6.89
$\text{Ti}_3\text{C}_2$	Bernal	14.70	7.35
$\text{Ti}_3\text{C}_2\text{O}_2$	Bernal	18.58	9.29
$\text{Ti}_3\text{C}_2\text{F}_2$	Bernal	18.70	9.35
$\text{Ti}_3\text{C}_2(\text{OH})_2$	Bernal	19.28	9.64
$\text{Nb}_4\text{C}_3$	Bernal	20.52	10.26
$\text{Nb}_4\text{C}_3\text{O}_2$	Bernal	25.20	12.60
$\text{Nb}_4\text{C}_3\text{F}_2$	Bernal	26.62	13.31
$\text{Nb}_4\text{C}_3(\text{OH})_2$	Bernal	28.50	14.25
$\text{MoS}_2$	Bernal	12.30	6.15
graphene	Bernal	6.68	3.34

Average layer thickness is calculated based on the most stable multilayer stacking structures of MXenes. Like graphene stack into graphite in Bernal stacking configuration, homogeneously terminated MXenes energetically favor Bernal stacking, as demonstrated in the earlier work<sup>16</sup>.

### Derivation of bending rigidity from ZA phonon branches:

The bending rigidity of 2D materials deriving from ZA branches of phonon dispersion is described in an easy to understand in the review paper of graphene.<sup>17</sup> In part III of this paper (Page 132-133), authors talked about flexural phonons, elasticity, and crumpling of graphene. We make extracts below. They started from the elastic energy:

$$E_0 = \frac{\kappa}{2} \int d^2r (\nabla \cdot N)^2 \approx \frac{\kappa}{2} \int d^2r (\nabla^2 h)^2, \quad (S1)$$

Where  $\kappa$  is the bending rigidity and  $h$  is the height variable.  $r$  is the in-plane vector and  $N$  is the unit vector normal to the surface.

Then rewrite in momentum space as:

$$E_0 = \frac{\kappa}{2} \sum_k k^4 h_{-k} h_k. \quad (S2)$$

Then canonically quantize the problem by introducing a momentum operator  $P_k$  that has the following commutator with  $h_k$ :

$$[h_k, P_k] = i\delta_{k,k}, \quad (S3)$$

And Hamiltonian as

$$H = \sum_k \left\{ \frac{P_{-k} P_k}{2\sigma} + \frac{\kappa k^4}{2} h_{-k} h_k \right\}, \quad (S4)$$

Where  $\sigma$  is 2D mass density. From the Heisenberg equations of motion for the operators, it is trivial to find that  $h_k$  oscillates harmonically with a frequency given by

$$\omega_{flex}(k) = \left(\frac{\kappa}{\sigma}\right)^{1/2} k^2. \quad (S5)$$

Which is the long-wavelength dispersion of flexural modes. The bending rigidity  $\kappa$  is

related with flexural modes frequencies by the above equation.

### Details of data processing.

The fitted function from data in **Fig. 4a** is as following:

MXenes:

$$D = 0.3741C - 52.8394 \quad (\text{S6})$$

Effective thickness < 1 angstrom, graphene, h-BN,  $\alpha$ -B

$$D = 0.0044C - 0.2514 \quad (\text{S7})$$

The fitted function from data in **Fig. 4b** is as following:

$$\gamma = 8.6679 - 74.0043t_s^{-1} + 1227.3120t_s^{-2} \quad (\text{S8})$$

$$D = -6.2497 + 12.0250t_s - 4.7626t_s^2 + 0.8122t_s^3 \quad (\text{S9})$$

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