

Effect of anion reorientation on proton mobility in the solid acids family  
 $\text{CsH}_y\text{XO}_4$  ( $X = \text{S}, \text{P}, \text{Se}, y = 1, 2$ ) from ab initio molecular dynamics  
simulations

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# 1 Derivation of the equation for extrapolation of experimental diffusion coefficients at 500 K for CsHSO<sub>4</sub> and CsHSeO<sub>4</sub>

A linear correlation between the logarithm of the diffusion coefficient and the inverse temperature is immediately apparent from Figure 3 of [Blinč, Dolinsek, Lahajnar, Zupancic, Shuvalov, Baranov, *phys. stat. sol. (b)*, **1984**, 123: K83-K87]. Thus, we obtained the following equations for the calculation of the diffusion coefficient  $D$  with respect to different temperatures  $T$  by careful linear parametrization of the curves shown in Figure 3 reported by Blinč et al.:

$$\log(D) = m \cdot \frac{1000}{T} + n \quad (1)$$

For CsHSO<sub>4</sub>, we obtain  $m = -4.36$  and  $n = 10.37$ . For CsHSeO<sub>4</sub>, we obtain  $m = -4.00$  and  $n = 10.25$ .

## 2 Intermolecular hydrogen-oxygen RDFs

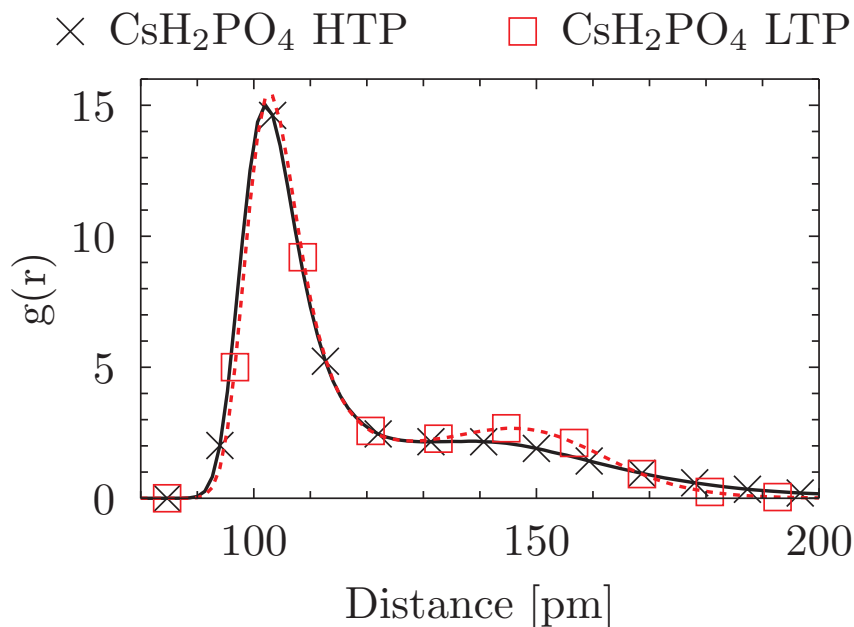


Figure 1: RDF of the intermolecular oxygen-hydrogen distances for the solid acid CsH<sub>2</sub>PO<sub>4</sub>.

The intermolecular hydrogen-oxygen RDFs for the solid acids are depicted in Figures 1 - 3. The RDFs show no significant difference for the comparison of the HTP and LTP of a given compound.

CsHSO<sub>4</sub> and CsHSeO<sub>4</sub> show almost the same distribution of the H-O distances, whereas the RDF of CsH<sub>2</sub>PO<sub>4</sub> is significantly changed. For CsH<sub>2</sub>PO<sub>4</sub> smaller O-O distances can be observed, indicating again shorter and stronger hydrogen bonds.

In summary, there are no (significant) differences between the oxygen-oxygen and oxygen-hydrogen RDFs of the HTPs and LTPs. HTPs and LTPs form hydrogen bonds in a similar amount and strength.

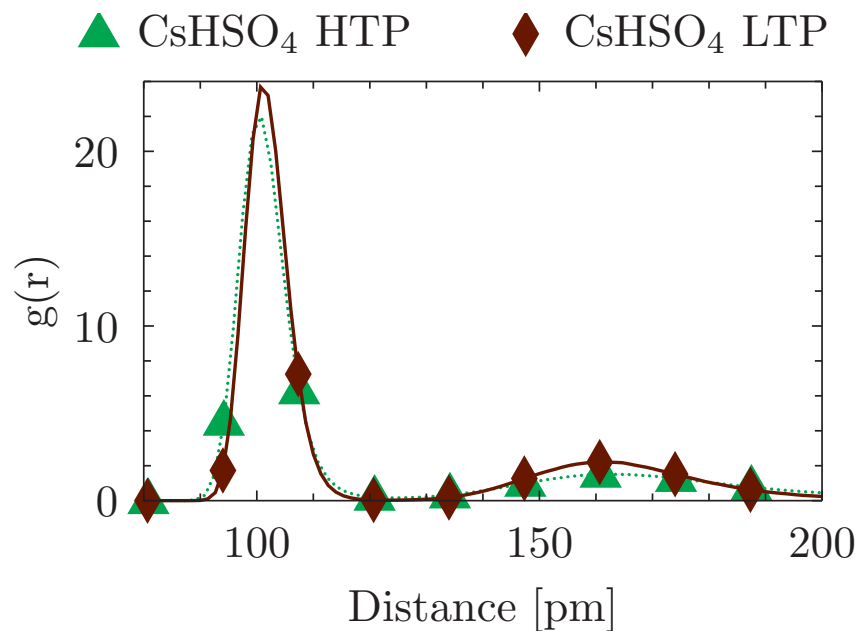


Figure 2: RDF of the intermolecular oxygen-hydrogen distances for the solid acid  $\text{CsHSO}_4$ .

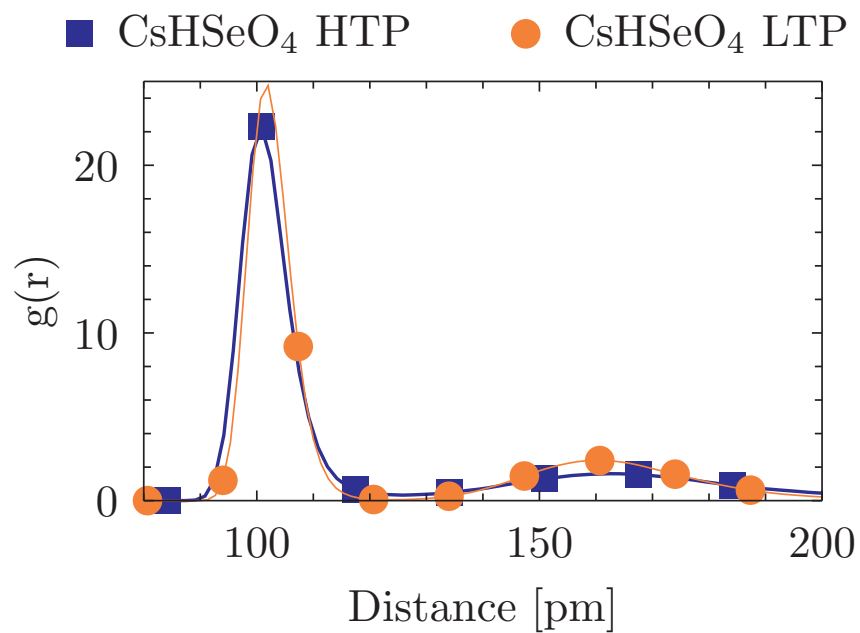
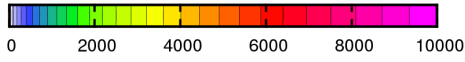
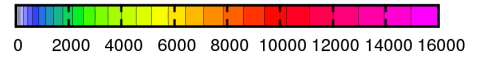


Figure 3: RDF of the intermolecular oxygen-hydrogen distances for the solid acid  $\text{CsHSeO}_4$ .

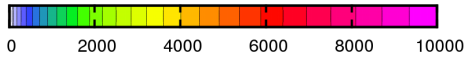
### 3 Distinction between proton conductors and insulators: Combined Distribution Functions (CDF)



(a) Color bar of the CDF of the HTP of  $\text{CsH}_2\text{PO}_4$ .



(b) Color bar of the CDF of the LTP of  $\text{CsH}_2\text{PO}_4$ .



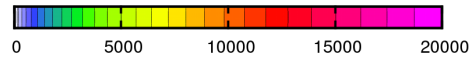
(c) Color bar of the CDF of the HTP of  $\text{CsHSO}_4$



(d) Color bar of the CDF of the LTP of  $\text{CsHSO}_4$



(e) Color bar of the CDF of the HTP of  $\text{CsHSeO}_4$



(f) Color bar of the CDF of the LTP of  $\text{CsHSeO}_4$ .

Figure 4: Color bars associated with the Combined Distribution Function (CDF) of the HTPs and LTPs of  $\text{CsH}_2\text{PO}_4$ ,  $\text{CsHSeO}_4$  and  $\text{CsHSO}_4$  from the main article.