

Supporting information: First principles study on
hydrogen doping induced metal-to-insulator transition in rare
earth nickelates RNiO_3 ($\text{R} = \text{Pr}, \text{Nd}, \text{Sm}, \text{Eu}, \text{Gd}, \text{Tb}, \text{Dy}, \text{Yb}$)

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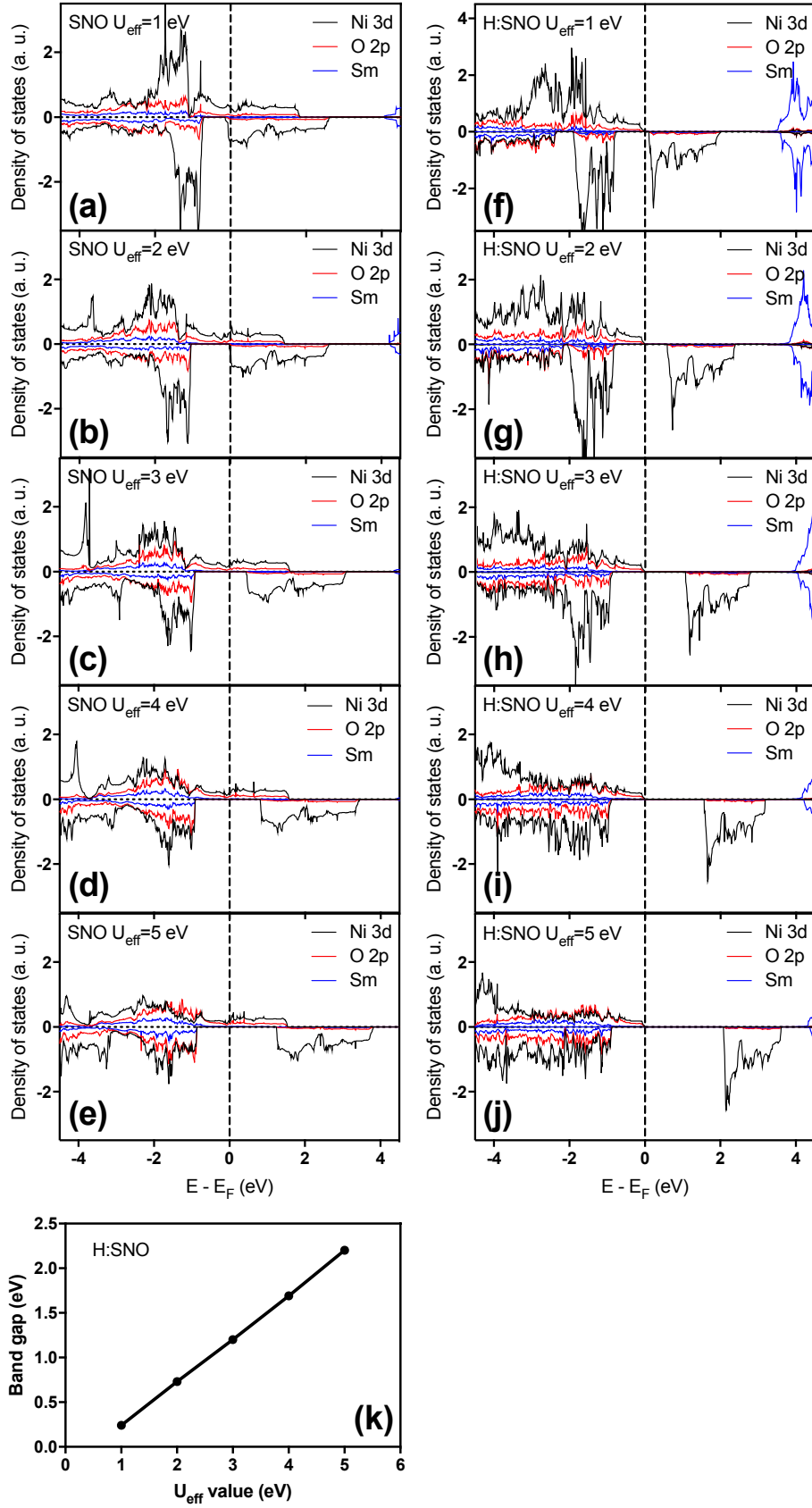


Figure S1. Projected density of states for pure SmNiO₃ (a)-(e) and H:SmNiO₃ (f)-(j) using the different U_{eff} values (1~5 eV). The predicted band gap values vs U_{eff} values for H:SmNiO₃ (k).

Table S1. Crystal structures of SmNiO₃ and their percentage error from the experimental structure of SmNiO₃ (*Phys. Rev B.*, 1998, **57**, R3189-R3192) with different U_{eff} values applied.

U _{eff} value	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	Volume (Å ³)	<i>a</i> error (%)	<i>b</i> error (%)	<i>c</i> error (%)	Volume error (%)
1 eV	5.270	5.381	7.459	211.52	-1.16	-0.79	-1.28	-3.21
2 eV	5.238	5.417	7.483	212.32	-1.76	-0.13	-0.97	-2.84
3 eV	5.268	5.369	7.481	211.59	-1.20	-1.01	-0.99	-3.17
4 eV	5.270	5.356	7.482	211.19	-1.16	-1.25	-0.98	-3.36
5 eV	5.271	5.349	7.484	211.01	-1.14	-1.38	-0.95	-3.44

Table S2. Crystal structures of H:SmNiO₃ with different U_{eff} values applied.

U _{eff} value	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	Volume (Å ³)
1 eV	5.169	6.063	7.584	237.68
2 eV	5.290	5.979	7.537	238.39
3 eV	5.240	6.011	7.553	237.90
4 eV	5.257	5.992	7.547	237.73
5 eV	5.258	5.980	7.549	237.36

Table S3. Energy differences between various AFM states in the monoclinic lattice ($P2_1/n$) and the FM state in the orthorhombic lattice ($Pbnm$) for three different $RNiO_3$. The ground states are marked with bold texts. These oxide are in the insulating phase in their ground states.

	T-AFM ($E_{AFM}-E_{FM}$)	E-AFM ($E_{AFM}-E_{FM}$)	A-AFM ($E_{AFM}-E_{FM}$)	S-AFM ($E_{AFM}-E_{FM}$)
PrNiO ₃	-0.46 eV	-0.34 eV	0.17 eV	-0.45 eV
SmNiO ₃	-0.21 eV	-0.12 eV	0.16 eV	-0.20 eV
DyNiO ₃	0.83 eV	-0.05 eV	0.14 eV	-0.07 eV

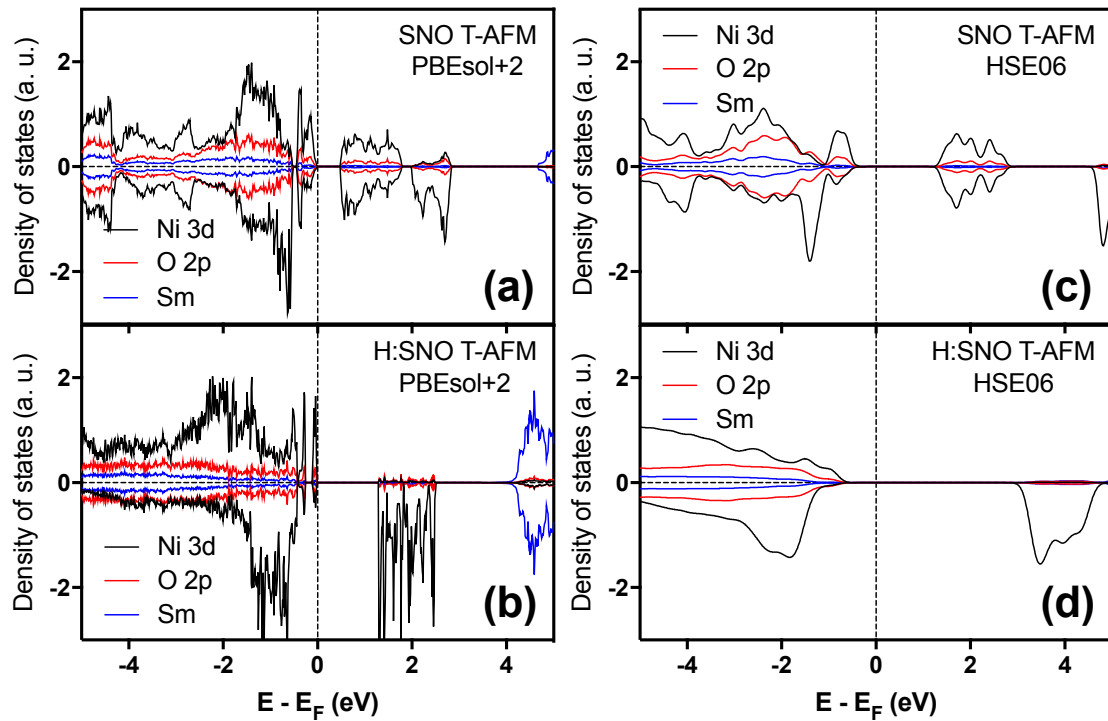


Figure S2. Projected density of states for pristine and H doped $SmNiO_3$ in T-AFM states for PBEsol+2 (a) and (b), and for HSE06 (c) and (d). The band gap values were predicted to increase upon H doping by both PBEsol+2 (from 0.48 eV to 1.30 eV) and HSE06 (from 1.37 eV to 2.96 eV) along with changes in the electron occupation of Ni 3d orbitals.

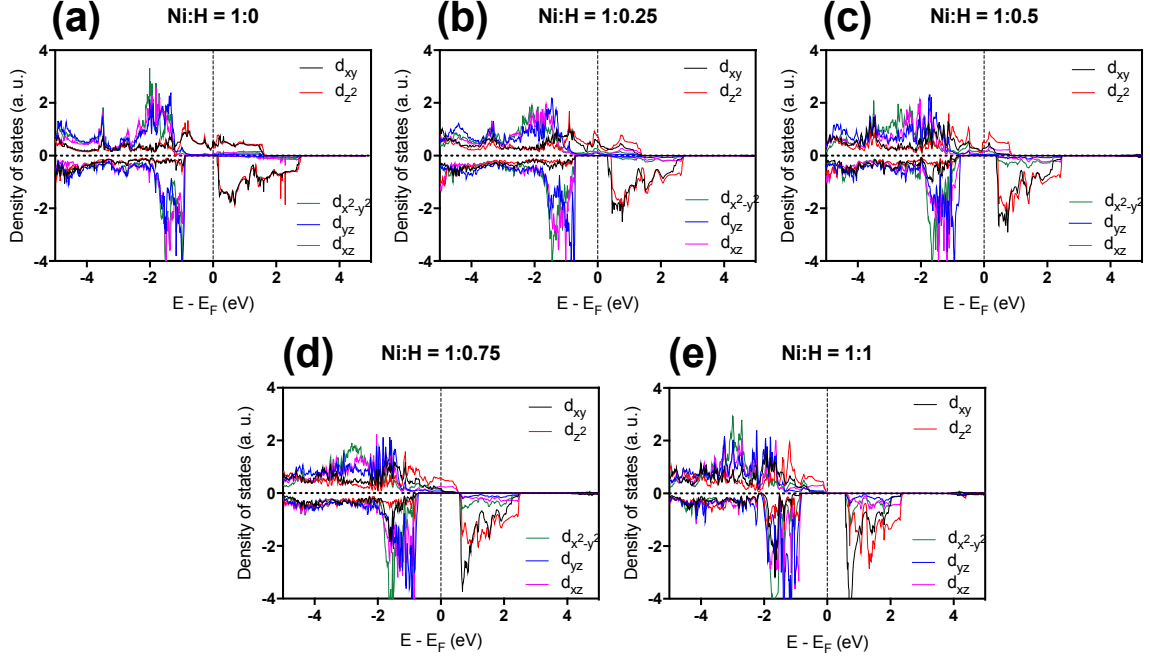


Figure S3. Projected Ni d-orbitals (d_{xy} , d_{z^2} , $d_{x^2-y^2}$, d_{yz} and d_{xz}) density of states of SmNiO_3 at different H doping concentrations. Due to the $\sqrt{2} \times \sqrt{2} \times 2$ supercell of standard perovskite used in the calculations, the d_{xy} and d_{z^2} correspond to conventional e_g orbitals, while $d_{x^2-y^2}$, d_{yz} and d_{xz} correspond to conventional t_{2g} orbitals. As the H doping concentration increases, the spin up e_g states shift from unoccupied states to the valence bands.

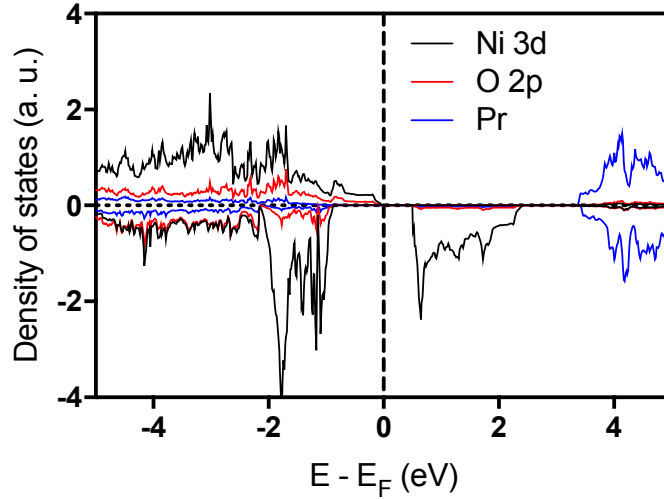


Figure S4. Projected density of states for hydrogen doped PrNiO_3 at the level of PBEsol+3.

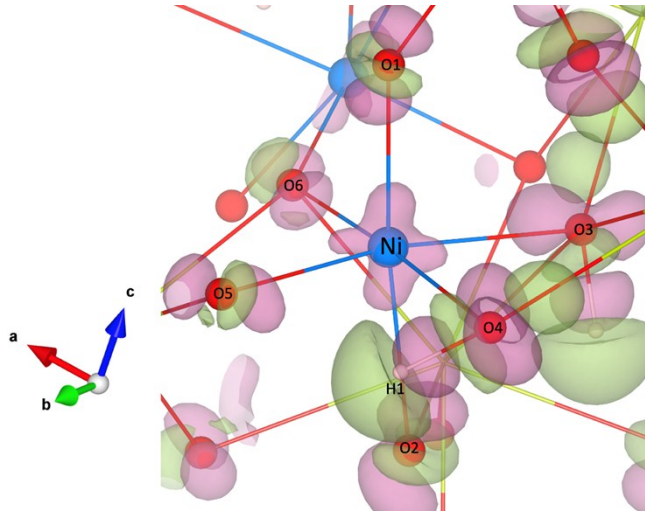


Figure S5. Electron density difference plot obtained by subtracting the electron density of pure SmNiO_3 and the electron density of single H atom from the electron density of H doped SmNiO_3 . The purple contour has positive sign, indicating gaining electrons upon H doping, while the light blue contour has negative sign, indicating losing electrons with hydrogen doping on SmNiO_3 . The O1 and O2 are located at top and bottom apex of NiO_6 octahedron, and O3 to O6 are oxygens in the middle plane of the octahedron. The charge difference shows that Ni is gaining electrons in the form of d_{z^2} orbital from hydrogen atom H1, which is bonded to O4.

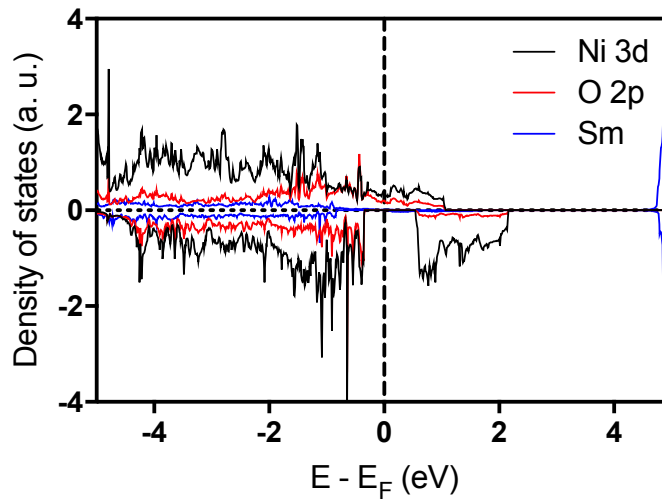


Figure S6. Projected density of states at the PBEsol+2 level for the distorted H: SmNiO_3 structure obtained by removing all hydrogens from H: SmNiO_3 . The metallic character was retained with the distortion, suggesting that the distortion alone cannot cause the metal-to-insulator transition.

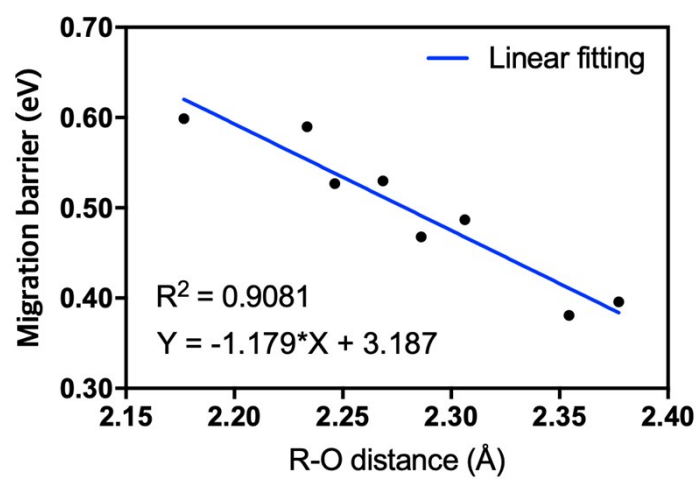


Figure S7. Migration barrier of Route 1 vs R-O distance.