

The role of osmolytes in the temperature-triggered conformational transition of poly(N-vinylcaprolactam): an experimental and computational study

Payal Narang¹, Tiago E. de Oliveira^{2*}, Pannuru Venkatesu^{1*} and Paulo A. Netz²

¹ Department of Chemistry, University of Delhi, Delhi – 110007, India

² Universidade Federal do Rio Grande do Sul, Porto Alegre, Brazil

Table S1: The table presents the derived partial charges on the atoms of PVCL monomer using ATB 3.0.

Region	Group and atom		Derived charges
Backbone	CH1	C	0.229
		H	0.070
	CH2	C	-0.387
		H	0.122
Amide	CAM	C	0.531
	NAM	N	-0.213
	OAM	O	-0.616
Cycle	CC1	C	-0.170
		H	0.065
	CC2	C	0.033
		H	0.016
	CC3	C	-0.070
		H	0.026
	CC4	C	0.025
		H	0.022
	CC5	C	-0.124
		H	0.095

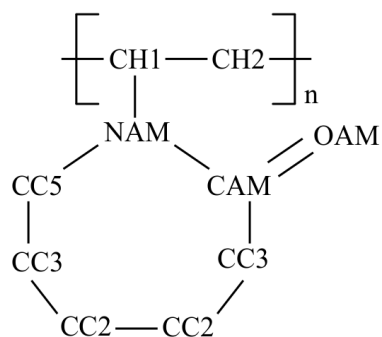


Figure S1: Atom types for the PVCL monomer

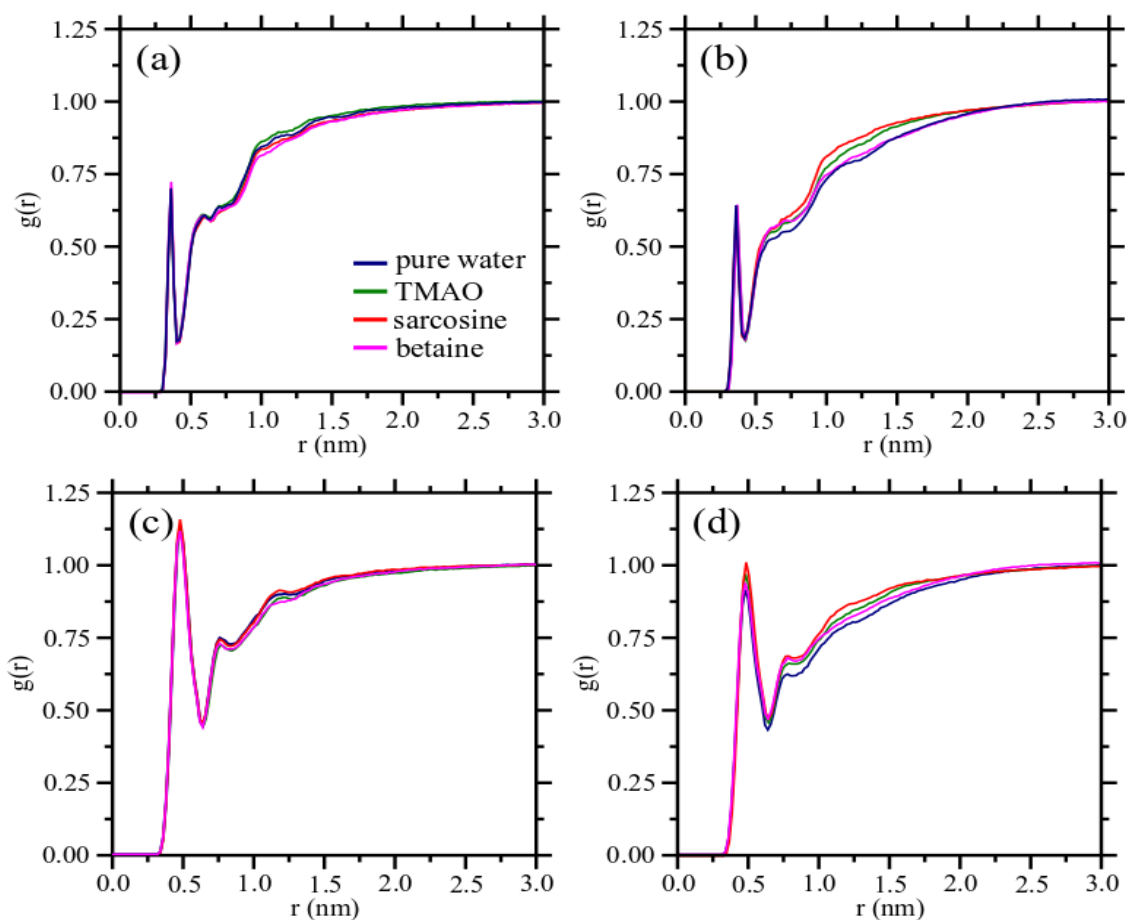


Figure S2: Radial distribution function (all) polymer-water at 280K (a) and 340 K (b), and between the atoms belonging to the side cycle and water at 280 K (c) and 340 K (d).

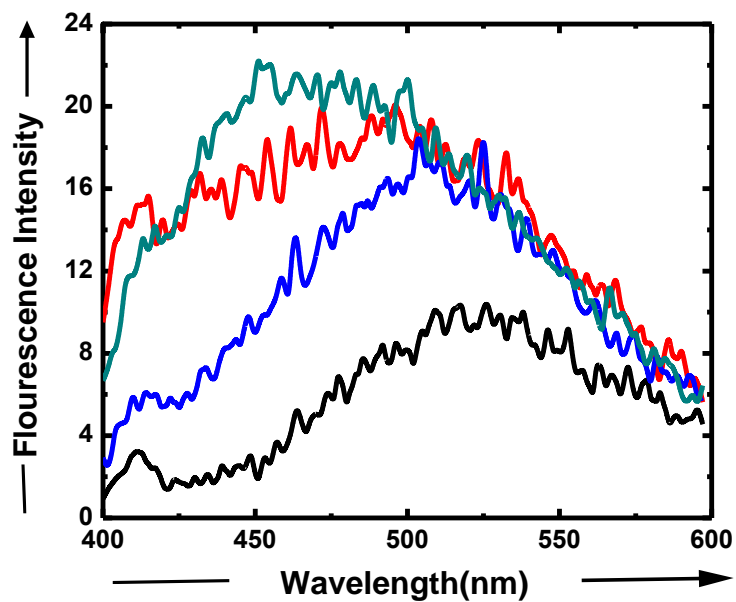


Figure S3: The fluorescence spectra of ANS in (a) water (black) (b) 0.5 M TMAO (red) (c) 0.5 M betaine (blue) and (d) 0.5 M sarcosine (dark cyan).