

Electronic Supplementary Information

Long-distance proton transfer induced by a single ammonia molecule: ion mobility mass spectrometry of protonated benzocaine reacted with NH₃

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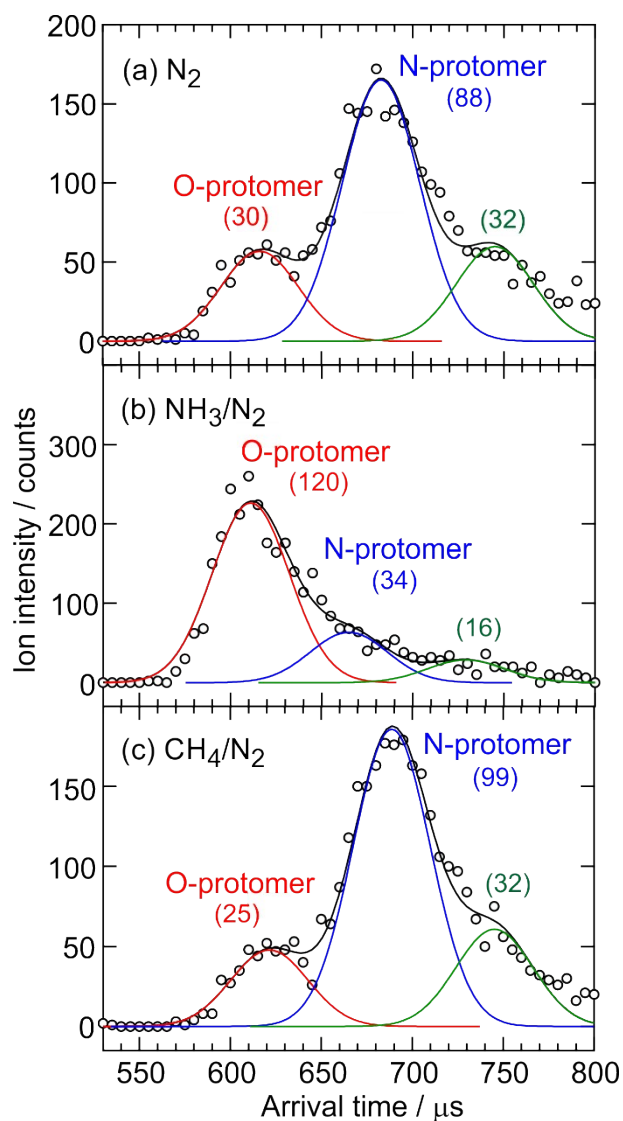


Fig. S1 Arrival time distributions of $\text{BC}\cdot\text{H}^+$. (a) N_2 buffer gas in the ion drift tube (Pressure in the drift tube $P_{\text{DT}} = 0.11$ Torr), (b) 0.5% NH_3/N_2 buffer gas ($P_{\text{DT}} = 0.10$ Torr), (c) 1% CH_4/N_2 buffer gas ($P_{\text{DT}} = 0.11$ Torr). Red, blue and green curves are Gaussian functions for fitting the experimental plots (black circles). Red, blue and green curves represent the O-protomer, N-protomer, and $\text{BC}\cdot\text{H}^+$ fragments caused by the dissociation of acetonitrile-solvated N-protomers. Each area of bands with red, blue and green curves is shown in parenthesis.

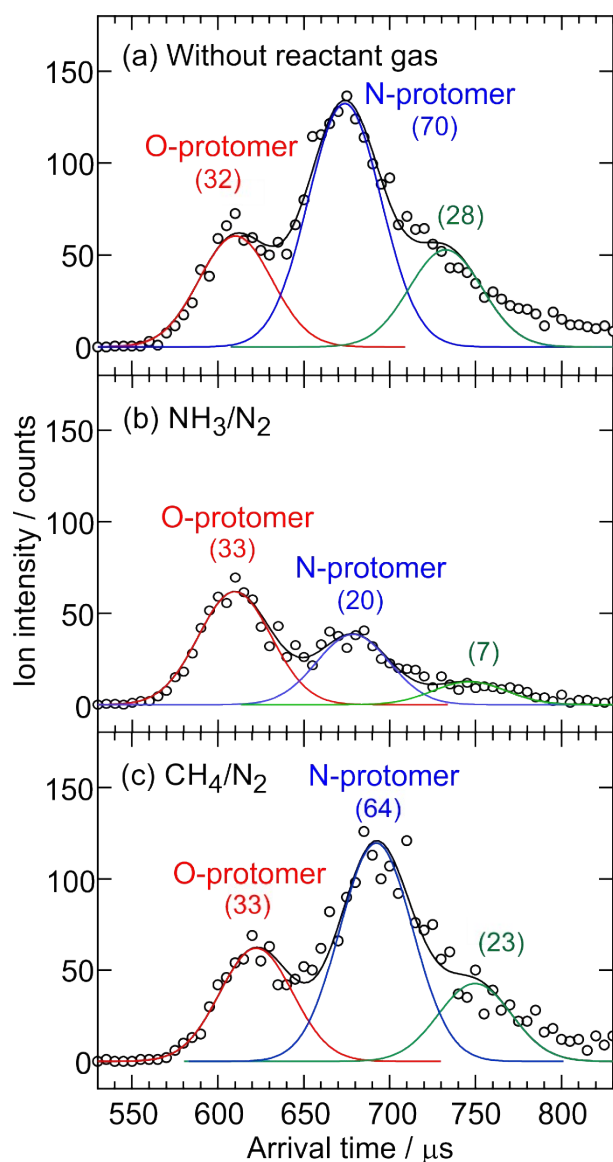


Fig. S2 Arrival time distributions of $\text{BC}\cdot\text{H}^+$. (a) Without reactant gas in the ion guide, (b) 20% NH_3/N_2 gas in the ion guide ($\Delta P = 3.1 \times 10^{-3}$ Pa), (c) 20% CH_4/N_2 gas in the ion guide ($\Delta P = 3.0 \times 10^{-3}$ Pa). The pressure of N_2 buffer gas in the ion drift tube was $P_{\text{DT}} = 0.11$ Torr. Red, blue and green curves are Gaussian functions for fitting the experimental plots (black circles). Red, blue and green curves represent the O-protomer, N-protomer, and $\text{BC}\cdot\text{H}^+$ fragments caused by the dissociation of acetonitrile-solvated N-protomers. Each area of bands with red, blue and green curves is shown in parenthesis.

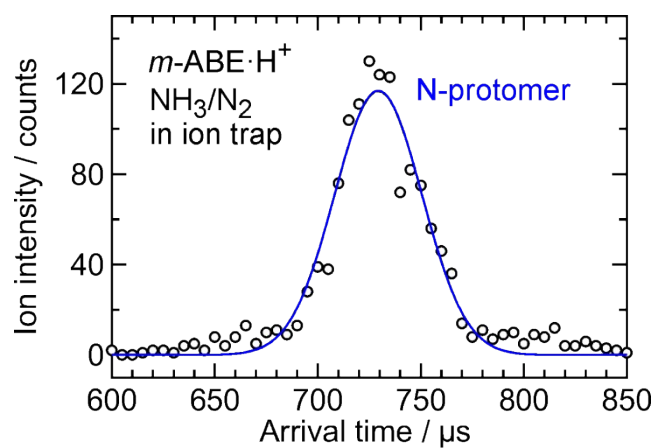


Fig. S3 Arrival time distributions of $m\text{-ABE}\cdot\text{H}^+$ with 1% NH_3/N_2 gas in the ion trap. The partial pressure of NH_3/N_2 gas ($\Delta P = 2.7 \times 10^{-3}$ Pa) was monitored by the vacuum gauge of the chamber. The pressure of N_2 buffer gas in the ion drift tube was $P_{\text{DT}} = 0.20$ Torr. The blue curve is Gaussian functions for fitting the experimental plots (black circles).