

Supplementary for

Mechanistic Insight into Methane Dry Reforming over Cobalt: A Density
Functional Theory Study

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Table S1. The elementary processes and corresponding kinetic parameters for DRM over Co(0001)

Elementary processes	$E_{a,f}$ (eV)	$E_{a,r}$ (eV)	$\nu_{0,f}$ (s ⁻¹)	$\nu_{0,r}$ (s ⁻¹)
$\text{CH}_4(\text{g}) + 2^* \rightarrow \text{CH}_3^* + \text{H}^*$	0.95	1.07	2.62E+9	5.50E+13
$\text{CH}_3^* + ^* \rightarrow \text{CH}_2^* + \text{H}^*$	0.78	0.65	3.94E+13	2.69E+13
$\text{CH}_2^* + \text{H}^* \rightarrow \text{CH}^* + ^*$	0.28	0.67	1.62E+13	1.03E+13
$\text{CH}^* + ^* \rightarrow \text{C}^* + \text{H}^*$	1.15	0.82	2.29E+13	3.46E+13
$\text{CO}_2(\text{g}) + ^* \rightarrow \text{CO}^* + \text{O}^*$	0.26	1.47	9.83E+8	6.98E+13
$\text{CO}_2^* + \text{H}^* = \text{trans-COOH}^* + ^*$	1.15	1.16	1.97E+13	8.56E+13
$\text{trans-COOH}^* = \text{cis-COOH}^*$	0.79	0.50	2.17E+13	6.04E+13
$\text{cis-COOH}^* + ^* = \text{CO}^* + \text{OH}^*$	0.29	1.66	2.71E+13	6.53E+13
$\text{CH}_3^* + \text{O}^* = \text{CH}_3\text{O}^* + ^*$	1.99	1.50	4.39E+13	5.37E+13
$\text{CH}_2^* + \text{O}^* = \text{CH}_2\text{O}^* + ^*$	1.97	1.32	8.49E+12	4.38E+12
$\text{CH}^* + \text{O}^* = \text{CHO}^* + ^*$	1.34	0.68	8.77E+12	1.43E+13
$\text{C}^* + \text{O}^* = \text{CO}^* + ^*$	1.64	2.53	7.29E+13	1.12E+13
$\text{CH}_3^* + \text{OH}^* = \text{CH}_3\text{OH}^* + ^*$	2.37	1.60	6.38E+12	9.38E+12
$\text{CH}_2^* + \text{OH}^* = \text{CH}_2\text{OH}^*$	1.57	0.84	9.64E+12	1.08E+12
$\text{CH}^* + \text{OH}^* = \text{CHOH}^* + ^*$	1.88	1.07	4.36E+12	7.86E+12
$\text{C}^* + \text{OH}^* = \text{COH}^* + ^*$	1.27	1.52	6.72E+12	9.82E+12
$\text{CH}_3\text{OH}^* + ^* = \text{CH}_2\text{OH}^* + \text{H}^*$	1.16	1.04	3.82E+13	1.92E+13
$\text{CH}_2\text{OH}^* + ^* = \text{CHOH}^* + \text{H}^*$	0.62	0.94	9.81E+12	8.92E+12
$\text{CHOH}^* + ^* = \text{COH}^* + \text{H}^*$	0.07	0.83	1.07E+13	2.45E+13
$\text{COH}^* + ^* = \text{CO}^* + \text{H}^*$	0.95	1.20	2.38E+13	6.74E+12
$\text{CH}_3\text{OH}^* + ^* = \text{CH}_3\text{O}^* + \text{H}^*$	0.93	1.64	2.32E+13	4.31E+13
$\text{CH}_2\text{OH}^* + ^* = \text{CH}_2\text{O}^* + \text{H}^*$	0.57	0.89	1.92E+12	9.28E+12
$\text{CHOH}^* + ^* = \text{CHO}^* + \text{H}^*$	1.05	1.42	4.35E+13	2.85E+13
$\text{CH}_3\text{O}^* + ^* = \text{CH}_2\text{O}^* + \text{H}^*$	0.98	0.54	1.76E+13	1.27E+13
$\text{CH}_2\text{O}^* + ^* = \text{CHO}^* + \text{H}^*$	0.42	0.74	3.87E+13	2.81E+13
$\text{CHO}^* + ^* = \text{CO}^* + \text{H}^*$	0.16	1.34	2.21E+13	8.81E+12
$\text{O}^* + \text{H}^* = ^*\text{OH}^* + ^*$	1.37	1.05	1.31E+13	1.42E+14
$\text{OH}^* + \text{H}^* = \text{H}_2\text{O}^* + ^*$	1.50	0.85	2.68E+14	1.95E+14
$\text{OH}^* + \text{OH}^* = \text{H}_2\text{O}^* + \text{O}^*$	0.98	0.70	2.29E+13	2.03E+13
$2\text{H}^* = \text{H}_2^* + ^*$	1.00	1.14	8.78E+12	1.82E+13

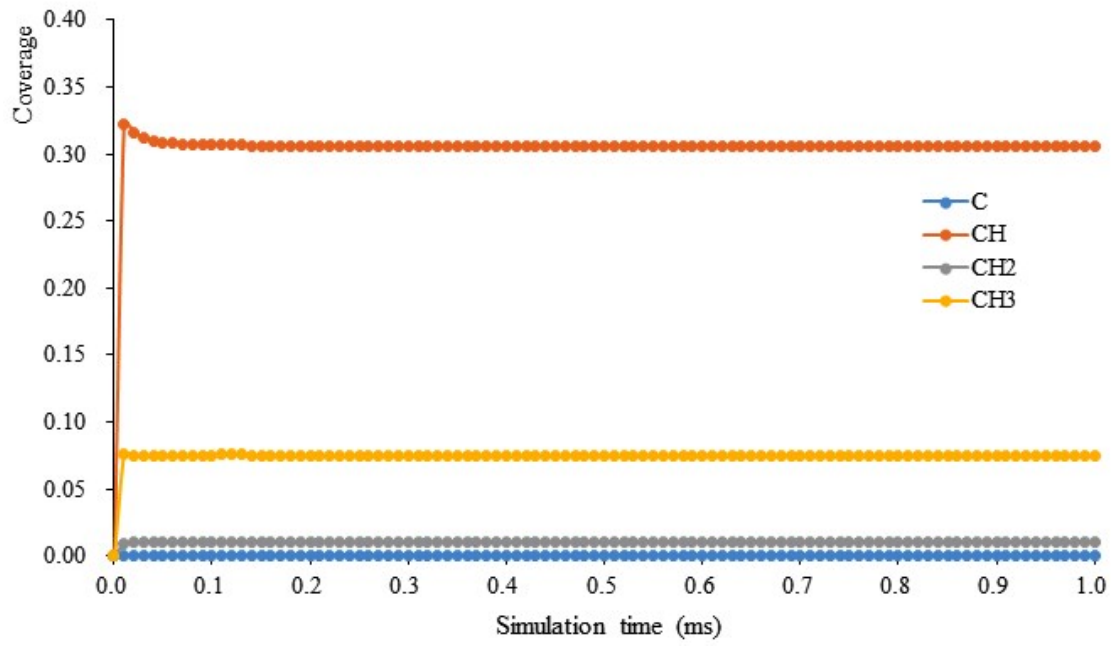


Fig. S1. The change of coverage of CH_x(x=0-3) over time within 1.0 ms simulation time on Co(0001)