## Coordination compounds in lithium storage and lithium-ion transport

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| Electrodes | Materials  | Reaction mechanism  | Voltage <sup>a</sup> | Capacity <sup>b</sup> | Advantages        | Disadvantages           | Ref. |
|------------|--|---|----------------------|-----------------------|-------------------|-------------------------|------|
|            |  |   | (V)                  | $(mA h g^{-1})$       |                   |                         |      |
| Cathode    | Organic materials (1,4-                                      | $C_6H_4O_2+2Li^++2e^-\leftrightarrow Li_2C_6H_4O_2$                             | 2.7                  | 495                   | High capacity     | High solubility and low | 13   |
|            | benzoquinone)  |   |                      |                       | and low cost      | conductivity            |      |
|            | Lithium-rich layered oxides                                  | LiCoO <sub>2</sub> ↔CoO <sub>2</sub> +Li <sup>+</sup> +e <sup>-</sup>           | 3.7                  | 274                   | High volumetric   | High cost, poor cycling | 15   |
|            | $(Li_{1+x}M_{1-x}O_2, M = Mn, Ni, Co)$                       |   |                      |                       | energy density    | stability               |      |
|            | Polyanionic compounds  | LiFePO₄↔FePO₄+Li <sup>+</sup> +e <sup>-</sup>                                   | 3.45                 | 170                   | High stability    | Low conductivity        | 16   |
|            | (LiFePO <sub>4</sub> )                                       |   |                      |                       | and cyclability   |                         |      |
|            | Spinel oxides  | $LiMn_2O_4 \leftrightarrow 2MnO_2 + Li^+ + e^-$                                 | 4.0                  | 150                   | Fast kinetics     | High solubility in the  | 15   |
|            | (LiMn <sub>2</sub> O <sub>4</sub> )                          |   |                      |                       |                   | electrolyte             |      |
|            | Nickel-rich layered oxides (LiNi1-                           | $LiNi_{0.5}Mn_{1.5}O_2 \leftrightarrow Ni_{0.5}Mn_{1.5}O_2 + Li^+ + e^-$        | 4.7                  | 146                   | High operating    | Poor cycling and        | 16   |
|            | $_{x}M_{x}O_{2}$ , M = Co, Mn and Al)                        |   |                      |                       | voltage           | thermal stability       |      |
| Anode      | Conversion-type  | $Fe_2O_3+6Li^++6e^-\leftrightarrow 2Fe+3Li_2O$                                  | 0.9                  | 1005                  | High capacity     | Large structural change | 15   |
|            | (Transition metal oxides)                                    |   |                      |                       |                   | and low conductivity    |      |
|            | Alloying-type  | Sn+4.4Li <sup>+</sup> +4.4e <sup>-</sup> ↔Li <sub>4.4</sub> Sn                  | 0.5                  | 990                   | High capacity     | Massive volume          | 17   |
|            | (Si, Sn, P)  |   |                      |                       | and low potential | variation               |      |
|            | Insertion-type   | 6C+Li <sup>+</sup> +e <sup>-</sup> ↔LiC <sub>6</sub>                            | 0.1                  | 370                   | Fast kinetics and | Low appaits             | 3    |
|            | (Graphite, Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub> ) |   |                      |                       | high abundance    | Low capacity            |      |
|            | Organic materials  | $Li_2C_8H_4O_4+2Li^++2e^-\leftrightarrow Li_4C_6H_4O_2$<br>(insertion reaction) | 0.85                 | 301                   | Fast kinetics     | High solubility and low | 13   |
|            | (Conjugated carboxylates)                                    |   |                      |                       |                   | conductivity            |      |

**Table S1** Electrochemical reaction mechanisms, potentials, and theoretical capacities of representative electrode materials for LIBs

<sup>a</sup> Average discharge voltage (vs. Li<sup>+</sup>/Li); <sup>b</sup> theoretical capacity.