

Electronic Supplementary Information (ESI)

# Bimetallic NiCo/CNF Encapsulated in N-doped Carbon Shell as Electrocatalyst for Zn-Air Battery and Water Splitting

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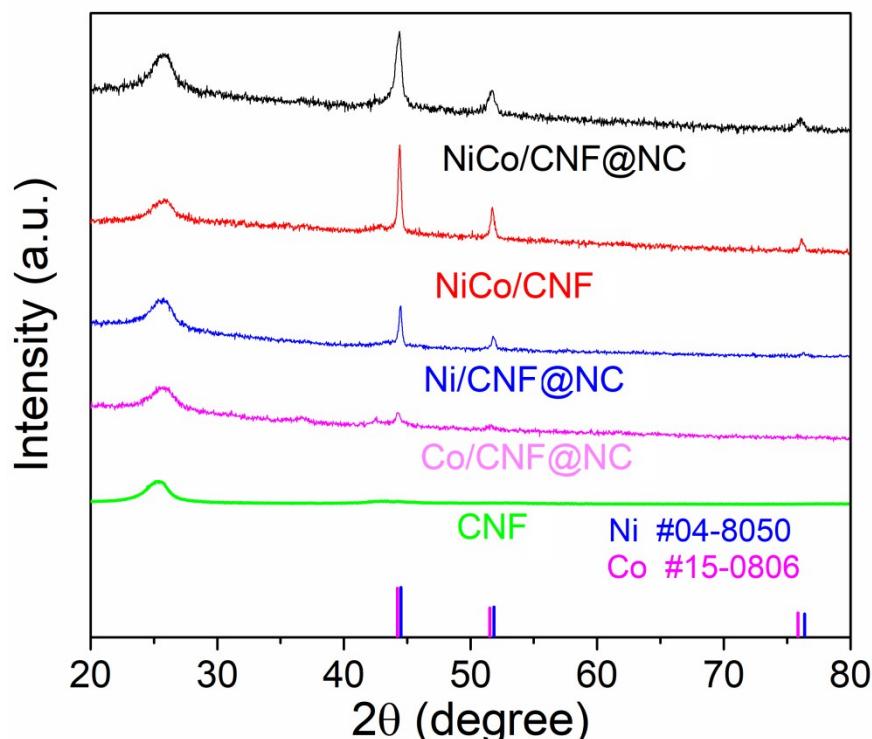
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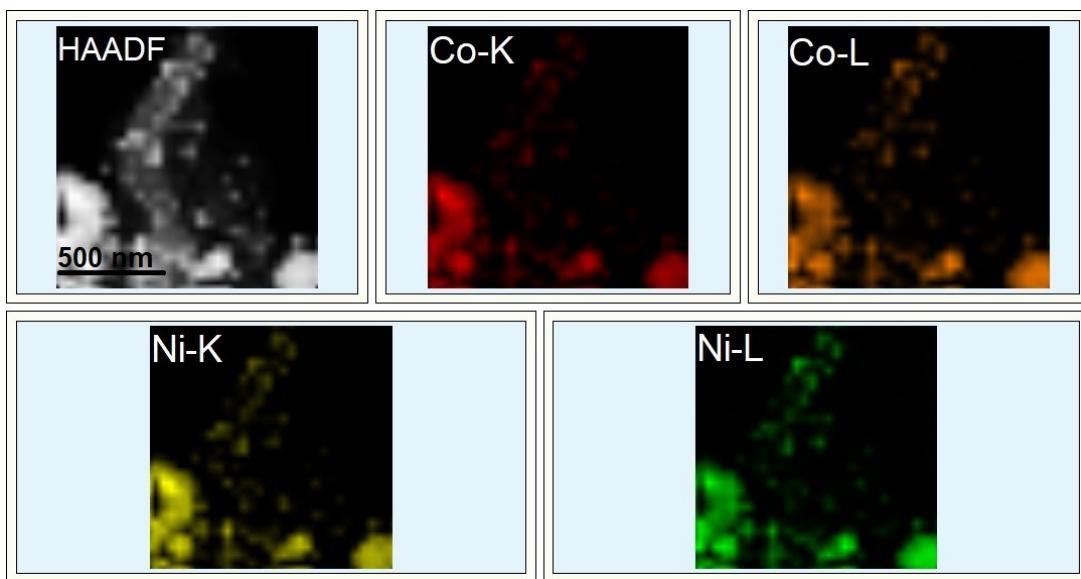
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### Materials used in this work

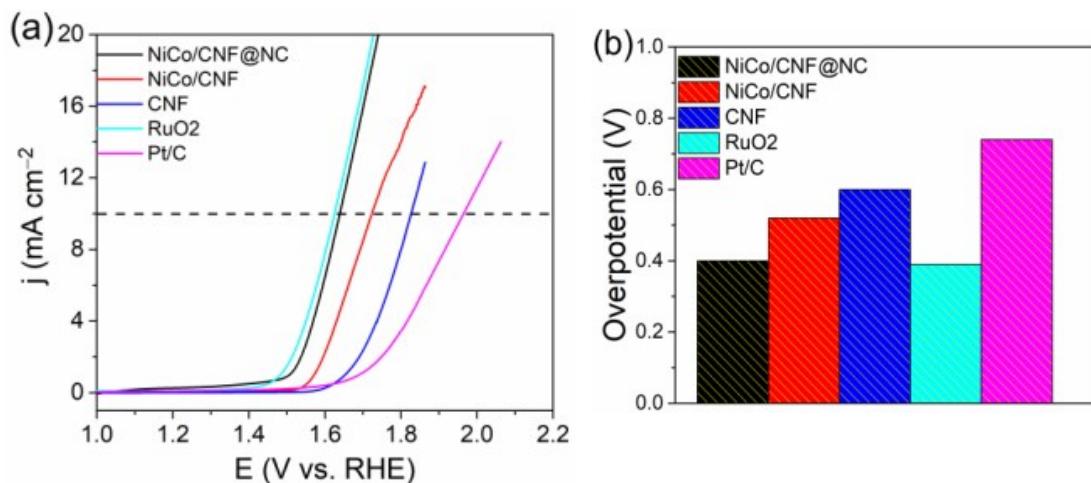
All of the chemical reagents were of analytical grade and directly used as received without further purification. Cobalt (II) acetate tetrahydrate ( $\text{Co}(\text{CH}_3\text{COO})_2 \cdot 4\text{H}_2\text{O}$ , > 99.5%) and potassium hydroxide (KOH) were supplied by Guangdong chemicals. Nickel acetate tetrahydrate ( $\text{Ni}(\text{CH}_3\text{COO})_2 \cdot 4\text{H}_2\text{O}$ , > 99 %), urea ( $\text{H}_2\text{NCONH}_2$ , > 99.0 %) and ethanol ( $\text{C}_2\text{H}_5\text{OH}$ , >99.7 %) were supplied by Tianjin Fuchen. Platinum on carbon (Pt/C, 20 wt.%) was provided by Johnson Matthey fuel cells. Nafion ionomer solution (5 %) was obtained from Dupont. Carbon nanofibers (CNF) was supplied by Aladdin Co. Ltd. Trizma base and dopamine hydrochloride were bought from Sigma-Aldrich. Argon (Ar), nitrogen ( $\text{N}_2$ ) and oxygen ( $\text{O}_2$ ) (99.99 %) gases were provided by Xi'an Taida Chemical Reagent Co. Ltd. Ultra-pure distilled water ( $18.25 \text{ M}\Omega \cdot \text{cm}^{-1}$ ) was used to prepare all electrolyte solutions.



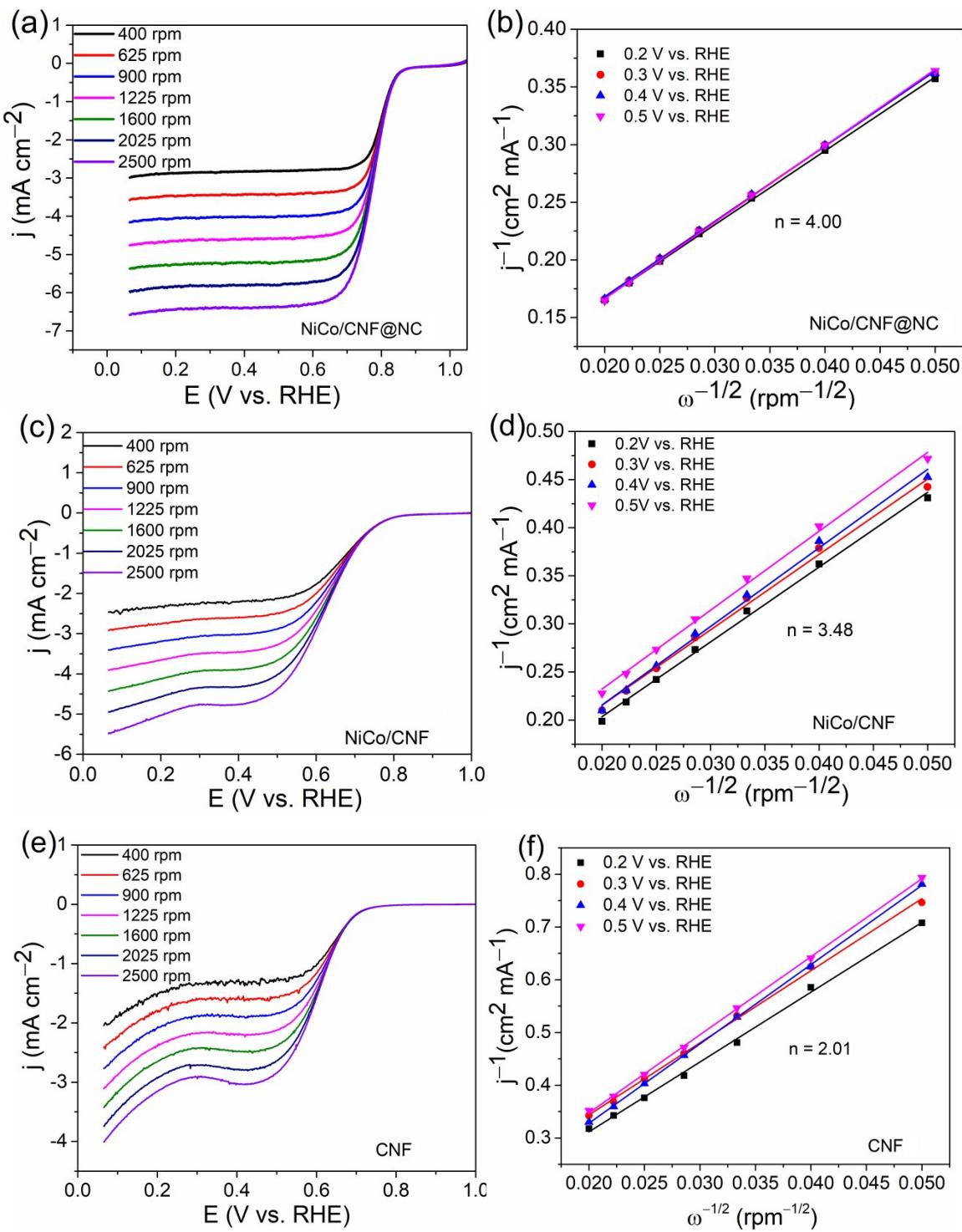
**Figure S1.** XRD patterns of NiCo/CNF@NC, NiCo/CNF, Ni/CNF@NC, Co/CNF@NC and CNF



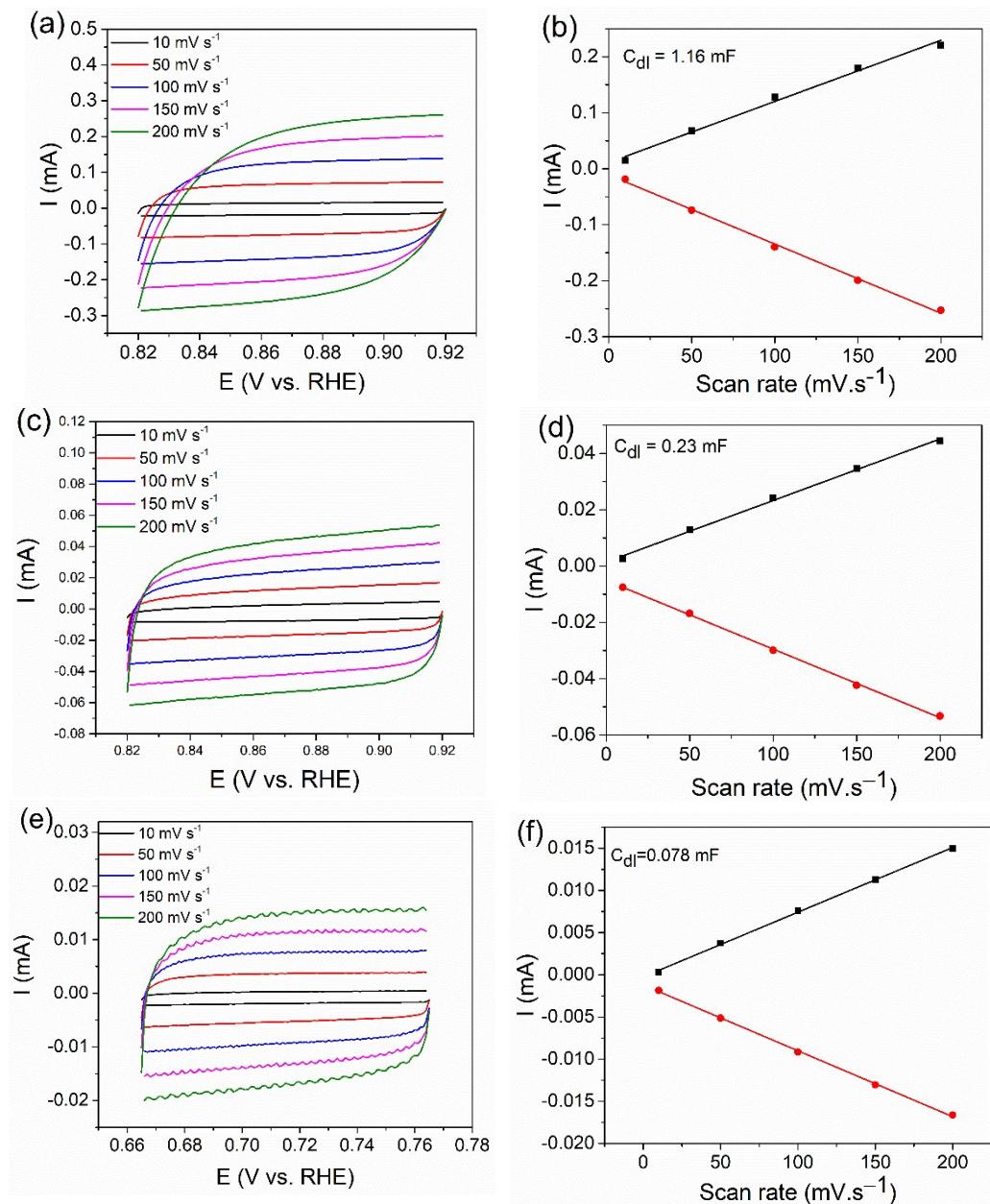
**Figure S2.** The typical high-angle annular dark-field scanning TEM (HAADF-STEM) image and the corresponding EDS mapping of NiCo/CNF@NC showing the distribution of Co and Ni elements.



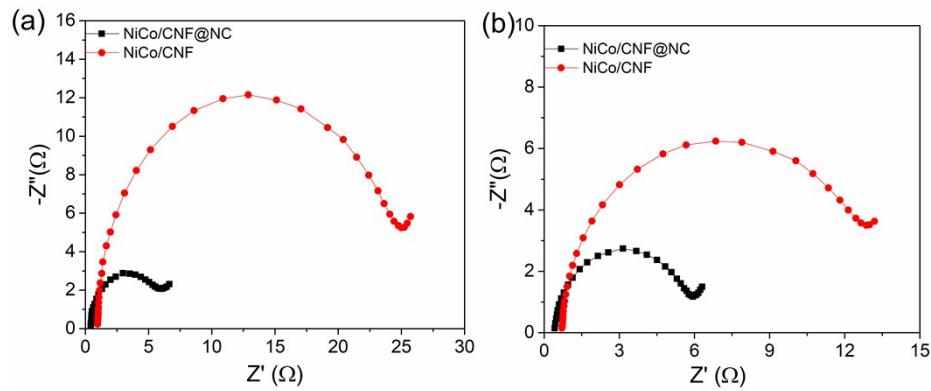
**Figure S3** a) OER polarization curves for NiCo/CNF@NC, NiCo/CNF, CNF, RuO<sub>2</sub>, and Pt/C catalysts in O<sub>2</sub> saturated 0.1 M KOH solution. (b) overpotentials derived from OER polarization curves at  $10 \text{ mA cm}^{-2}$ .



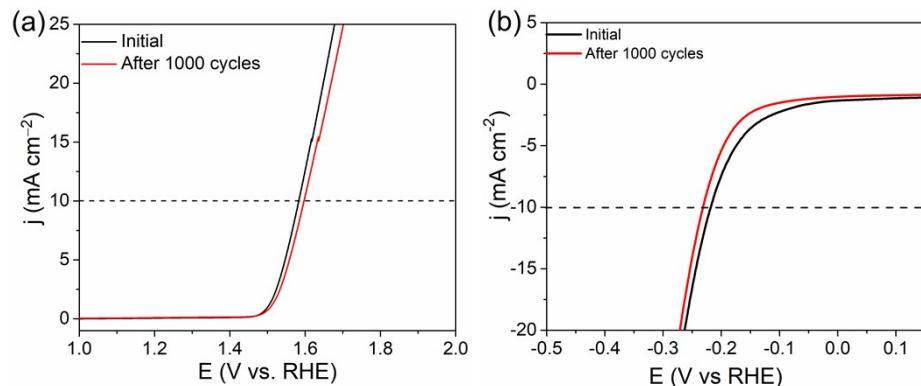
**Figure S4.** LSV curves of (a) NiCo/CNF@NC (c) NiCo/CNF, (e) CNF catalysts at different scanning rates and (b, d, f) the corresponding Koutecky–Levich (K–L) plots, respectively.



**Figure S5.** CV in  $\text{N}_2$  saturated 0.1 M KOH within non-Faradaic regions (a, c and e) and the corresponding current vs. scan rates respectively (b, d and f) for NiCo/CNF@NC, NiCo/CNF and CNF catalysts, respectively.



**Figure S6** Nyquist plots of NiCo/CNF@NC and NiCo/CNF (a)At OER overpotential of 0.35 V (b) At HER overpotential of 0.22 V.



**Figure S7** LSV of NiCo/CNF@NC catalyst before and after ADT test (a) OER by cycling between 1.2 and 1.8 V vs. RHE at scan rate of 100 mV s<sup>-1</sup>. (b) HER by cycling between -0.3 and 0.2 V vs. RHE at scan rate of 100 mV s<sup>-1</sup>.

**Table S1.** Comparison of the electrochemical performances of the electrocatalysts in this work and other related literature.  $E_{onset}$ , ORR onset potential;  $E_{1/2}$ , the half-wave potential;  $j$ , diffusion limiting current density at 1600 rpm;  $\eta_{OER}$ , OER overpotential at 10 mA cm<sup>-2</sup>;  $\Delta E$  the potential gap between OER potential and  $E_{1/2}$ . All the potentials reported are V vs. RHE.

Catalysts	Loading mg/cm <sup>2</sup>	KOH Molarity	ORR			$\eta_{OER}$	$\Delta E$ (V)	Ref.
			$E_{onset}$	$E_{1/2}$	$j$ (mA cm <sup>-2</sup> )			
NiCo/CNF@NC	0.26	0.1 M	1.00	0.78	-5.37	0.40	0.85	This work
NiCo/CNF	0.26	0.1 M	0.84	0.62	-4.43	0.52	1.13	This work
CNF	0.26	0.1 M	0.74	0.58	-3.43	0.60	1.25	This work
NiCo@NCNT-700	0.212	0.1 M	0.93	0.82	~5.0	---	---	1
NiCo@NC-2	0.40	0.1 M	0.96	0.81	-6.54	0.53	0.95	2
NCNT/CoONiO-NiCo	0.21	1 M	0.97	0.83	~4.4	0.27	0.67	3
NiCo@NCNTs	0.464	0.1 M	---	---	---	0.41	---	4
Ni <sub>1</sub> Co <sub>2</sub>	0.28	0.1 M	---	---	---	0.4	---	5
NiCo <sub>2</sub> O <sub>4</sub> /G	0.4	0.1 M	0.87	0.54	~4.2	0.44	1.13	6
NiCo <sub>2</sub> O <sub>4</sub>	---	0.1 M	0.84	0.75	-5.13	0.49	0.97	7
1D NiCo <sub>2</sub> O <sub>4</sub>	0.9	0.1 M	1.03	0.78	~6.0	0.39	0.84	8
Ni <sub>x</sub> O <sub>y</sub> /NC	0.21	0.1 M	---	0.71	---	0.41	0.93	9
Co-N/C 800	0.24	0.1 M	---	0.78	~6.2	0.51	0.96	10
Pt/C	0.16	0.1 M	1.03	0.80	-5.40	0.74	1.17	This work

**Table S2.** Comparison of HER and OER performance of NiCo/CNF@NC catalyst with reference catalyst prepared in this work and other related literatures.  $\eta_{HER}$  (HER overpotential)  $\eta_{OER}$  (OER overpotential). All the potential reported are V vs. RHE.

Catalysts	Loading mg/cm <sup>2</sup>	KOH Molarity	HER		OER		Ref.
			Current density (mA cm <sup>-2</sup> )	$\eta_{HER}$	Current density (mA cm <sup>-2</sup> )	$\eta_{OER}$	
NiCo/CNF@NC	0.26	1 M	10	0.22	10	0.35	This work
NiCo/CNF	0.26	1 M	10	0.42	10	0.49	This work
Ni/CNF@NC	0.26	1 M	10	0.37	10	0.45	This work
Co/CNF@NC	0.26	1 M	10	0.36	10	0.39	This work
Ni–Co–Fe (NCF)-MOF	0.20	0.1 M	10	0.27	10	0.32	11
Ni-NiO/N-rGO	0.21	0.1 M	10	0.21	10	0.22	12
NGO/Ni <sub>7</sub> S <sub>6</sub>	0.21	0.1 M	10	0.38	10	0.38	13
Ni <sub>0.33</sub> Co <sub>0.67</sub> S <sub>2</sub>	0.30	1 M	---	---	10	0.33	14
NixCo <sub>3-x</sub> O <sub>4</sub>	3.0	1 M	---	---	10	0.37	15
CoOx@CN	0.12	1 M	10	0.232	10	0.26	16

**Table S3.** Comparison of NiCo/CNF@NC based Zn-air battery performances with the other related works in the reported literatureres. Where: OCP, open circuit potential.

Catalysts	OCP (V)	Specific capacity		Energy density		Cyclability	Ref.
		mA h g <sup>-1</sup>	Wh kg <sup>-1</sup>	mA h cm <sup>-2</sup>	Wh cm <sup>-2</sup>		
NiCo/CNFs@NC	1.46	764 @20 mA cm <sup>-2</sup>	871 @20 mA cm <sup>-2</sup>	1800 s/cycle (at 5 mA cm <sup>-2</sup> ) for 130 cycles with negligible loss in voltage at the end.		This work	
NCNT/CoONiO-NiCo	---	504 @20 mA cm <sup>-2</sup>	615 @20 mA cm <sup>-2</sup>	400 s/cycle (at 20 mA cm <sup>-2</sup> ) for 100 cycles with negligible change of voltage at the end.		3	
NiCo <sub>2</sub> S <sub>4</sub> /N-CNT	1.49	431.1 @10 mA cm <sup>-2</sup>	554.6 @10 mA cm <sup>-2</sup>	400 s/cycle (10 mA cm <sup>-2</sup> ) for 150 cycles.		17	
Co <sub>3</sub> O <sub>4</sub> -NP/N-rGO	1.51	786 @5 mA cm <sup>-2</sup>	997 @5 mA cm <sup>-2</sup>	400 s/cycle (at 5 mA cm <sup>-2</sup> ) 1600 cycles no obvious voltage change.		18	
NiCo <sub>2</sub> O <sub>4</sub>	1.45	580 @20 mA cm <sup>-2</sup>	725@20 mA cm <sup>-2</sup>	1200s/cycle (at 20 mA cm <sup>-2</sup> ) for 50 cycles with a small overpotential of 0.14 V.		8	
NCO/N-rGO	---	712 @5 A g <sup>-2</sup>	---	600s/cycle (10 at 5 A g <sup>-2</sup> ) for 50 cycles with only 0.07 V overpotential gap increase.		19	
NiCo <sub>2</sub> O <sub>4</sub> -CNTs-400	1.38	670 @10 mA cm <sup>-2</sup>	700 @10 mA cm <sup>-2</sup>	600s/cycle (10 mA cm <sup>-2</sup> ) for 240 cycles.		20	

**Table S4.** Comparison of the water splitting performances of NiCo/CNF@NC catalyst and electrocatalysts based on other previously reported literatures.

Catalysts	Loading mg/cm <sup>2</sup>	KOH Molarity	Support	η@10mA cm <sup>-2</sup>	Ref.
NiCo/CNF@NC	4.6	1 M	Ni foam	0.31	This work
Ni/NiP	11	1 M	Ni foam	0.38	21
Ni <sub>0.9</sub> Fe <sub>0.1</sub> /NC	2.0	1 M	Ni foam	0.35	22
Ni <sub>5</sub> P <sub>4</sub>	---	1 M	Ni foam	0.47	23
NiSe NWs	2.8	1 M	Ni foam	0.40	24
NiCo <sub>2</sub> S <sub>4</sub> @NiFe LDH/NF	---	1 M	Ni foam	0.37	25
NiCo <sub>2</sub> S <sub>4</sub> NW	---	1 M	Ni foam	0.40	26
NiFe/NiCo <sub>2</sub> O <sub>4</sub> /NF	---	1 M	Ni foam	0.44	27
Pt-CoS <sub>2</sub> /CC	~ 0.5	1 M	Carbon cloth	0.32	28

## References

1. L. M. Zeng, X. Z. Cui, L. S. Chen, T. Ye, W. M. Huang, R. G. Ma, X. H. Zhang and J. L. Shi, *Carbon*, 2017, **114**, 347-355.
2. Y. Fu, H. Y. Yu, C. Jiang, T. H. Zhang, R. Zhan, X. W. Li, J. F. Li, J. H. Tian and R. Z. Yang, *Adv Funct Mater*, 2018, **28**, 201705094.
3. X. Liu, M. Park, M. G. Kim, S. Gupta, G. Wu and J. Cho, *Angew Chem Int Edit*, 2015, **54**, 9654-9658.
4. J. Yu, Y. J. Zhong, W. Zhou and Z. P. Shao, *J Power Sources*, 2017, **338**, 26-33.
5. T. V. Vineesh, S. Mubarak, M. G. Hahm, V. Prabu, S. Alwarappan and T. N. Narayanan, *Sci Rep-Uk*, 2016, **6**.
6. D. U. Lee, B. J. Kim and Z. W. Chen, *J Mater Chem A*, 2013, **1**, 4754-4762.
7. C. Jin, F. L. Lu, X. C. Cao, Z. R. Yang and R. Z. Yang, *J Mater Chem A*, 2013, **1**, 12170-12177.
8. M. Prabu, K. Ketpang and S. Shanmugam, *Nanoscale*, 2014, **6**, 3173-3181.
9. J. Masa, W. Xia, I. Sinev, A. Q. Zhao, Z. Y. Sun, S. Grutzke, P. Weide, M. Muhler and W. Schuhmann, *Angew Chem Int Edit*, 2014, **53**, 8508-8512.
10. W. H. Hu, Q. Wang, S. S. Wu and Y. M. Huang, *J Mater Chem A*, 2016, **4**, 16920-16927.
11. W. Ahn, M. G. Park, D. U. Lee, M. H. Seo, G. P. Jiang, Z. P. Cano, F. M. Hassan and Z. W. Chen, *Adv Funct Mater*, 2018, **28**, 201802129.
12. X. Liu, W. Liu, M. Ko, M. Park, M. G. Kim, P. Oh, S. Chae, S. Park, A. Casimir, G. Wu and J. Cho, *Adv Funct Mater*, 2015, **25**, 5799-5808.
13. K. Jayaramulu, J. Masa, O. Tomanec, D. Peeters, V. Ranc, A. Schneemann, R. Zboril, W. Schuhmann and R. A. Fischer, *Adv Funct Mater*, 2017, **27**, 1700451.
14. Z. Peng, D. S. Jia, A. M. Al-Enizi, A. A. Elzatahry and G. F. Zheng, *Adv Energy Mater*, 2015, **5**.
15. Y. G. Li, P. Hasin and Y. Y. Wu, *Advanced materials*, 2010, **22**, 1926-1929.
16. H. Y. Jin, J. Wang, D. F. Su, Z. Z. Wei, Z. F. Pang and Y. Wang, *J Am Chem Soc*, 2015, **137**, 2688-2694.
17. X. P. Han, X. Y. Wu, C. Zhong, Y. D. Deng, N. Q. Zhao and W. B. Hu, *Nano Energy*, 2017, **31**, 541-550.
18. X. P. Han, G. W. He, Y. He, J. F. Zhang, X. R. Zheng, L. L. Li, C. Zhong, W. B. Hu, Y. D. Deng and T. Y. Ma, *Adv Energy Mater*, 2018, **8**, 1702222.
19. P. Moni, S. Hyun, A. Vignesh and S. Shanmugam, *Chem Commun*, 2017, **53**, 7836-7839.
20. C. Y. Ma, N. N. Xu, J. L. Qiao, S. S. Jian and J. J. Zhang, *Int J Hydrogen Energ*, 2016, **41**, 9211-9218.
21. G. F. Chen, T. Y. Ma, Z. Q. Liu, N. Li, Y. Z. Su, K. Davey and S. Z. Qiao, *Adv Funct Mater*, 2016, **26**, 3314-3323.
22. X. Zhang, H. M. Xu, X. X. Li, Y. Y. Li, T. B. Yang and Y. Y. Liang, *Acs Catal*, 2016, **6**, 580-588.
23. M. Ledendecker, S. K. Calderon, C. Papp, H. P. Steinruck, M. Antonietti and M. Shalom, *Angew Chem Int Edit*, 2015, **54**, 12361-12365.
24. C. Tang, N. Y. Cheng, Z. H. Pu, W. Xing and X. P. Sun, *Angew Chem Int Edit*, 2015, **54**, 9351-9355.
25. J. Liu, J. S. Wang, B. Zhang, Y. J. Ruan, L. Lv, X. Ji, K. Xu, L. Miao and J. J. Jiang, *ACS applied materials & interfaces*, 2017, **9**, 15364-15372.
26. A. Sivanantham, P. Ganesan and S. Shanmugam, *Adv Funct Mater*, 2016, **26**, 4661-4672.
27. C. L. Xiao, Y. B. Li, X. Y. Lu and C. Zhao, *Adv Funct Mater*, 2016, **26**, 3515-3523.
28. X. P. Han, X. Y. Wu, Y. D. Deng, J. Liu, J. Lu, C. Zhong and W. B. Hu, *Adv Energy Mater*, 2018, **8**, 1800935.