

Electronic Supplementary Material (ESI)

**Powerful CO₂ electroreduction performance with N-carbon doped
by Ni single atoms**

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Supporting Figures

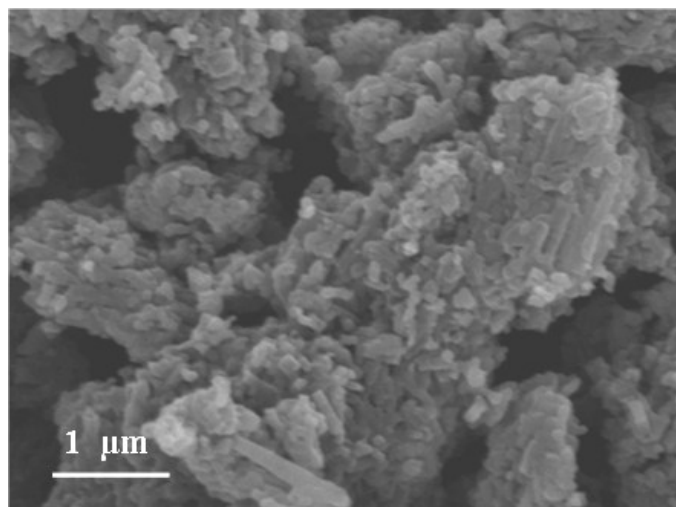


Fig. S1 The SEM image of the obtained N-C sample.

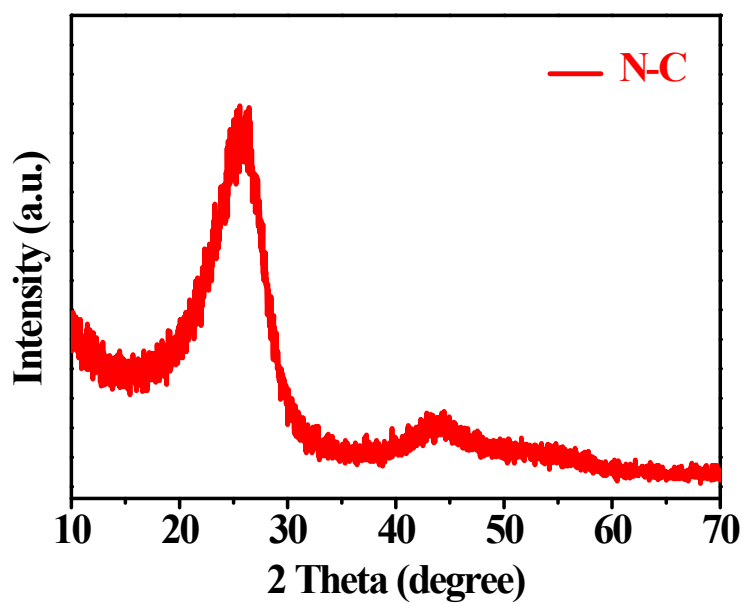


Fig. S2 The XRD pattern of the obtained N-C sample.

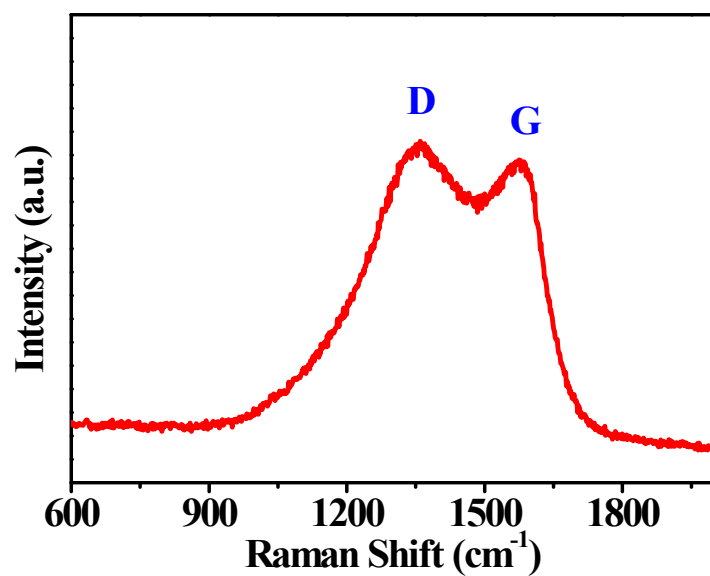


Fig. S3 The Raman spectrum of the obtained Ni-N-C sample.

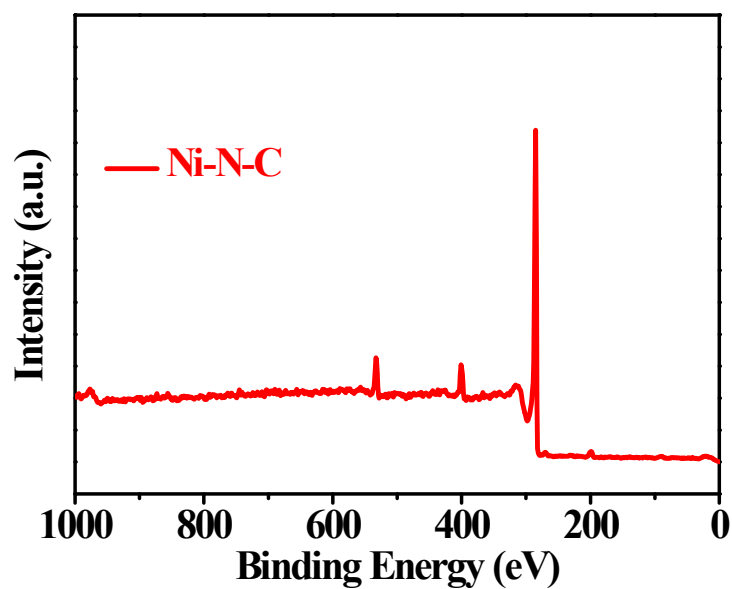


Fig. S4 The survey XPS spectrum of the Ni-N-C sample.

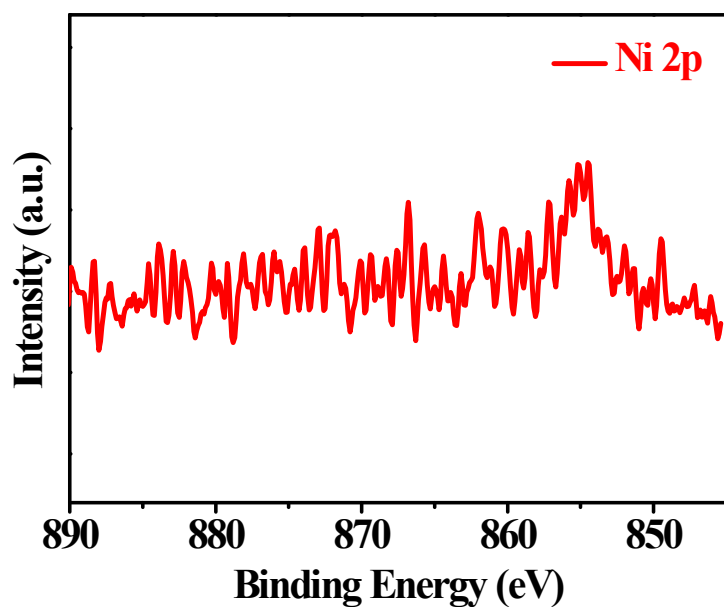


Fig. S5 The high-resolution Ni 2p XPS spectrum of the Ni-N-C catalyst.

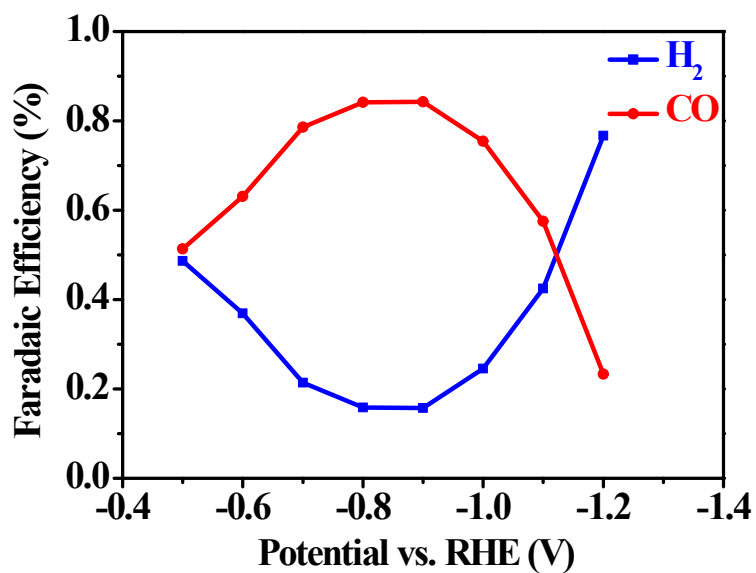


Fig. S6 The CO₂RR performance of the controlled sample prepared using pure Zn-bidppz without the adding of Ni as precursor.

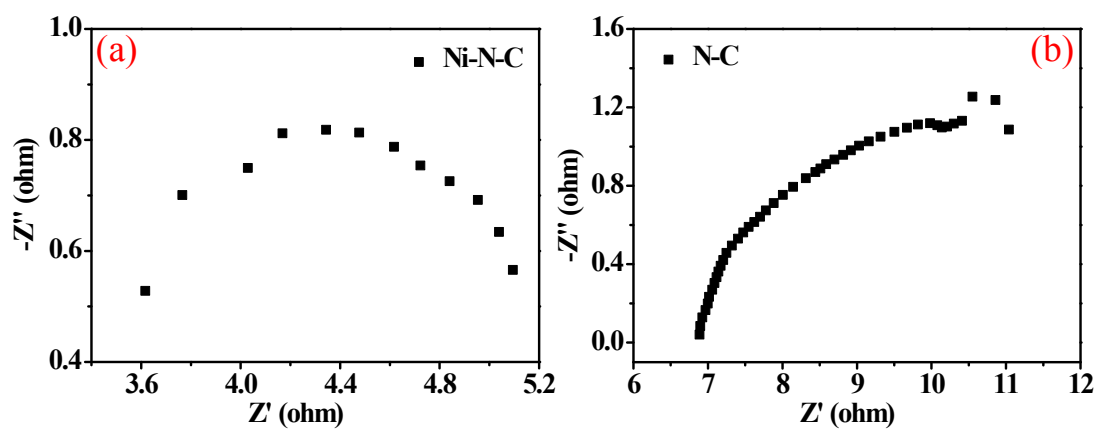


Fig. S7 The Nyquist plots of Ni-N-C (a) and N-C (b) catalysts.

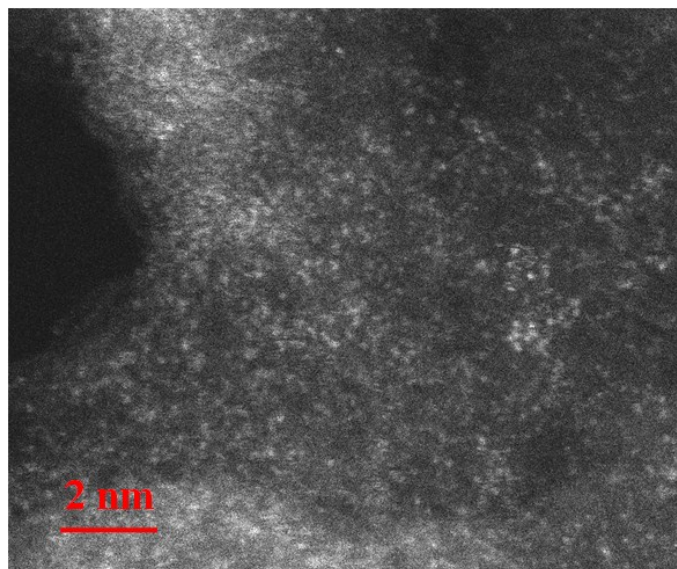


Fig. S8 The HAADF-STEM image of Ni-N-C catalyst after electrocatalysis.

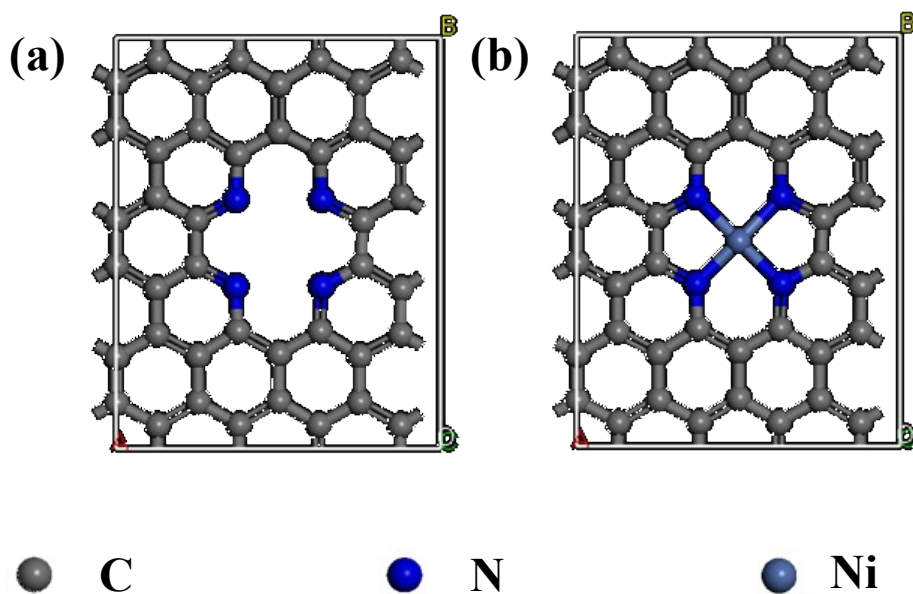


Fig. S9 The optimized structures of N-C (a) and Ni-N-C (b).

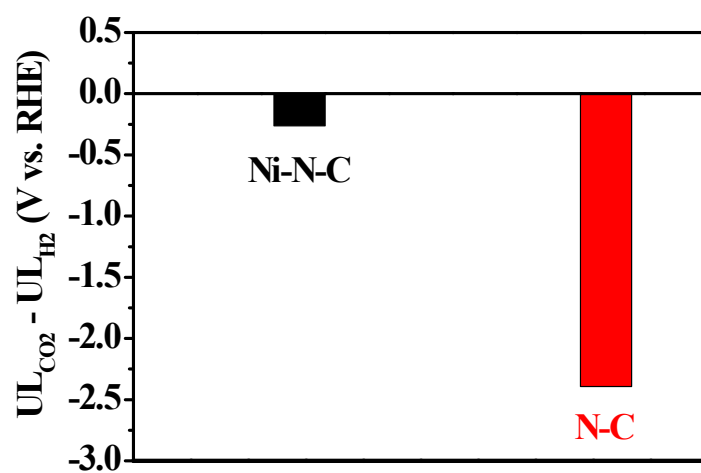


Fig. S10 The difference in limiting potentials for CO₂ reduction and H₂ evolution of N-C and Ni-N-C.

Table S1 Structural parameters of Ni-N-C extracted from the EXAFS fitting ($S_0^2=0.85$).

Sample	Scattering pair	CN	R($\text{\AA}$)	$\sigma^2(10^{-3}\text{\AA}^2)$	$\Delta E_0(\text{eV})$	R factor
Ni-N/C	Ni -N	3.9 $\pm$ 1.1	1.90 $\pm$ 0.02	6.6 $\pm$ 1.7	5.6 $\pm$ 1.4	0.0073

S_0^2 is the amplitude reduction factor; CN is the coordination number; R is interatomic distance (the bond length between central atoms and surrounding coordination atoms); σ^2 is Debye-Waller factor (a measure of thermal and static disorder in absorber-scatterer distances); ΔE_0 is edge-energy shift (the difference between the zero kinetic energy value of the sample and that of the theoretical model). R factor is used to value the goodness of the fitting.

Table S2 The Zero point energy (ZPE) corrections and entropic contributions to the free energies.

		ZPE/eV	TS/eV
CO ₂		0.2673	0.6606
COOH *	C-N	0.5725	8.839X10 ⁻³
	Ni-C-N	0.5281	0.0128
CO *	C-N	0.1291	1.23X10 ⁻⁵
	Ni-C-N	0.1227	1.93X10 ⁻⁵
H ₂ O (l)		0.5681	0.67
CO		0.1317	0.6108
H*	C-N	0.307	2.121X10 ⁻³
	Ni-C-N	0.3112	2.557x10 ⁻³
H ₂		0.2689	0.4038