

Crystal Phase Induced Band Gap Energy Enhancing Photo-Catalytic Properties of Zn-Fe₂O₄/Au NPs: Experimental and Theoretical studies

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Supplementary supporting information

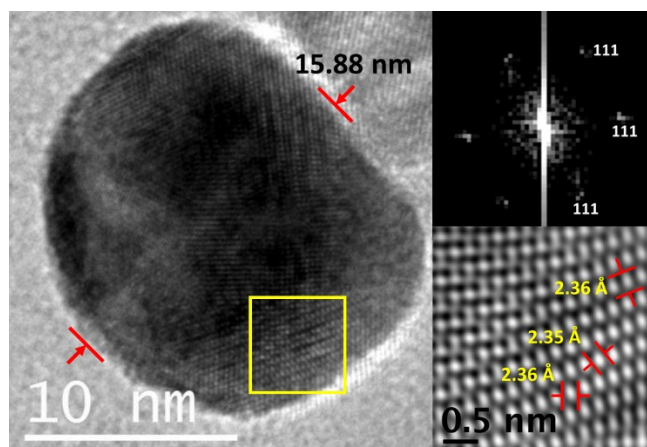


Figure S1. HRTEM micrograph for indexation Au nanoparticles

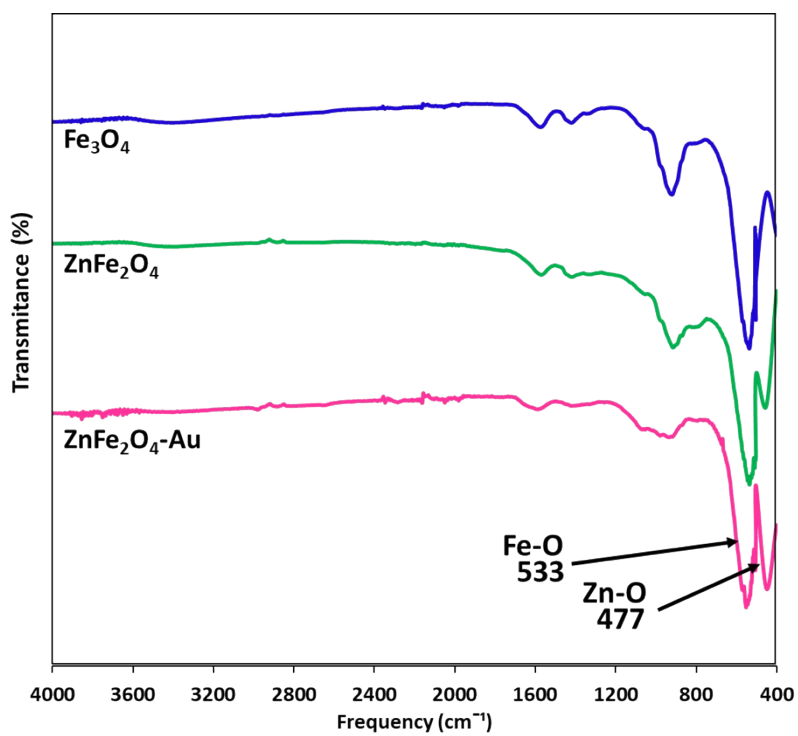


Figure S 1a. FTIR patterns of prepared iron oxide nanoparticles.

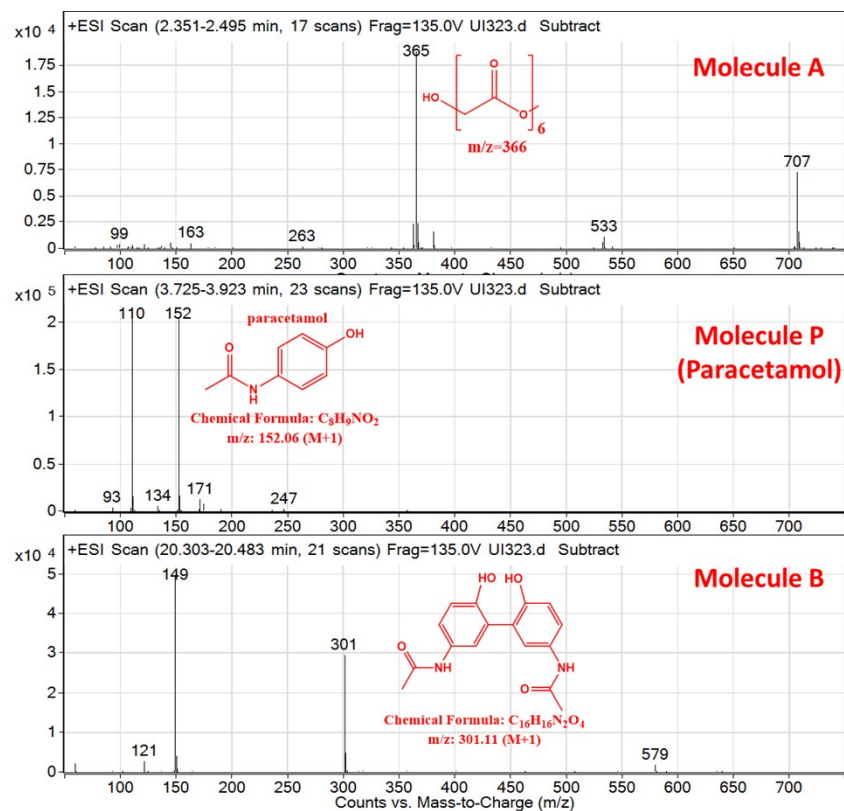


Figure S2. Mass spectrometry for HPLC separation peaks of commercial paracetamol tablets

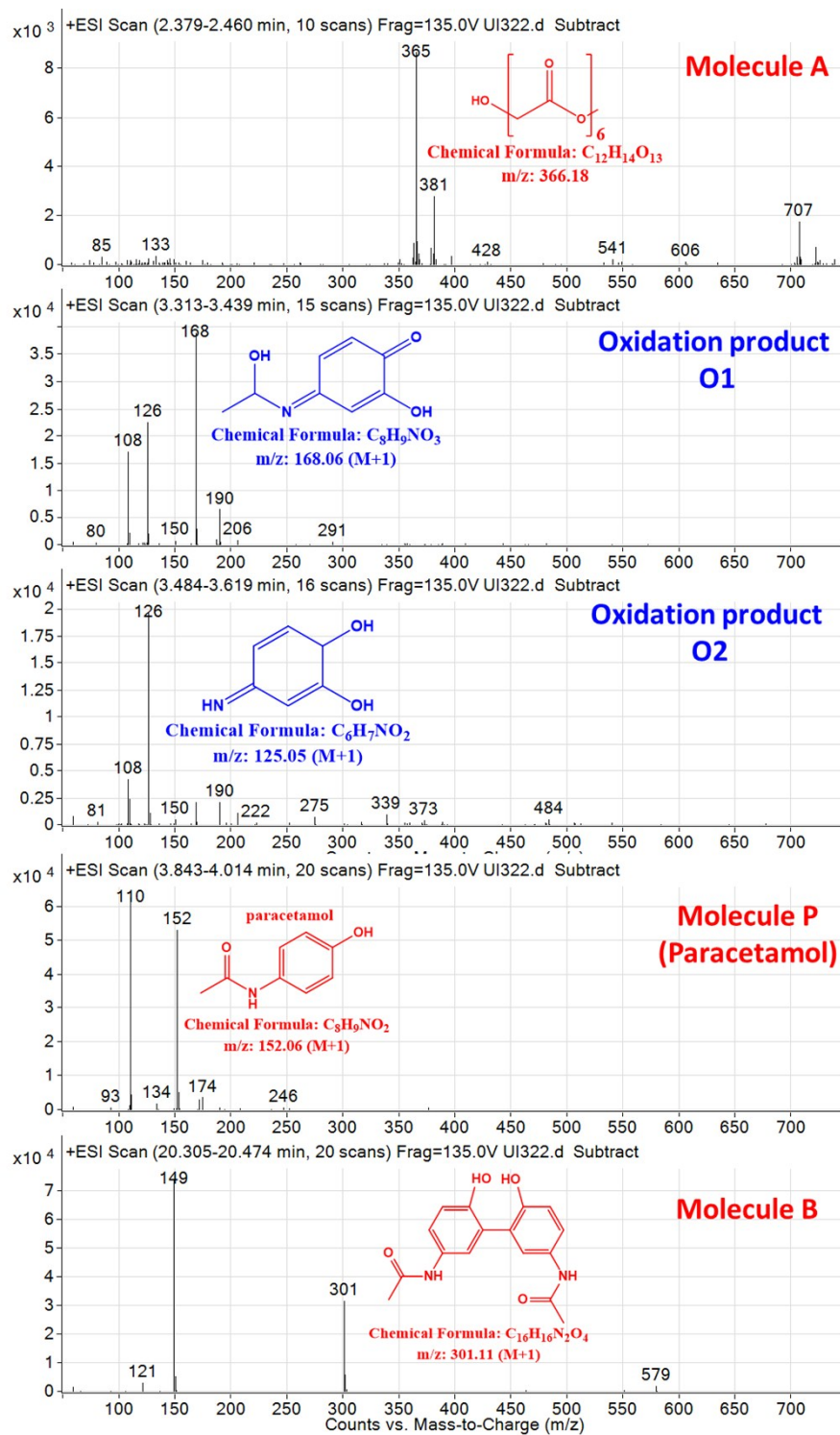


Figure S3. Mass spectrometry for HPLC separation peaks of oxidized paracetamol

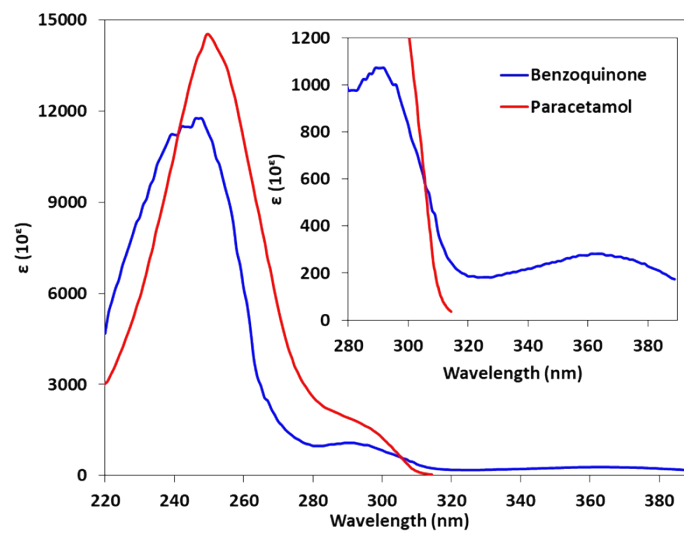


Figure S4. UV-Vis absorption profile of paracetamol and benzoquinone.

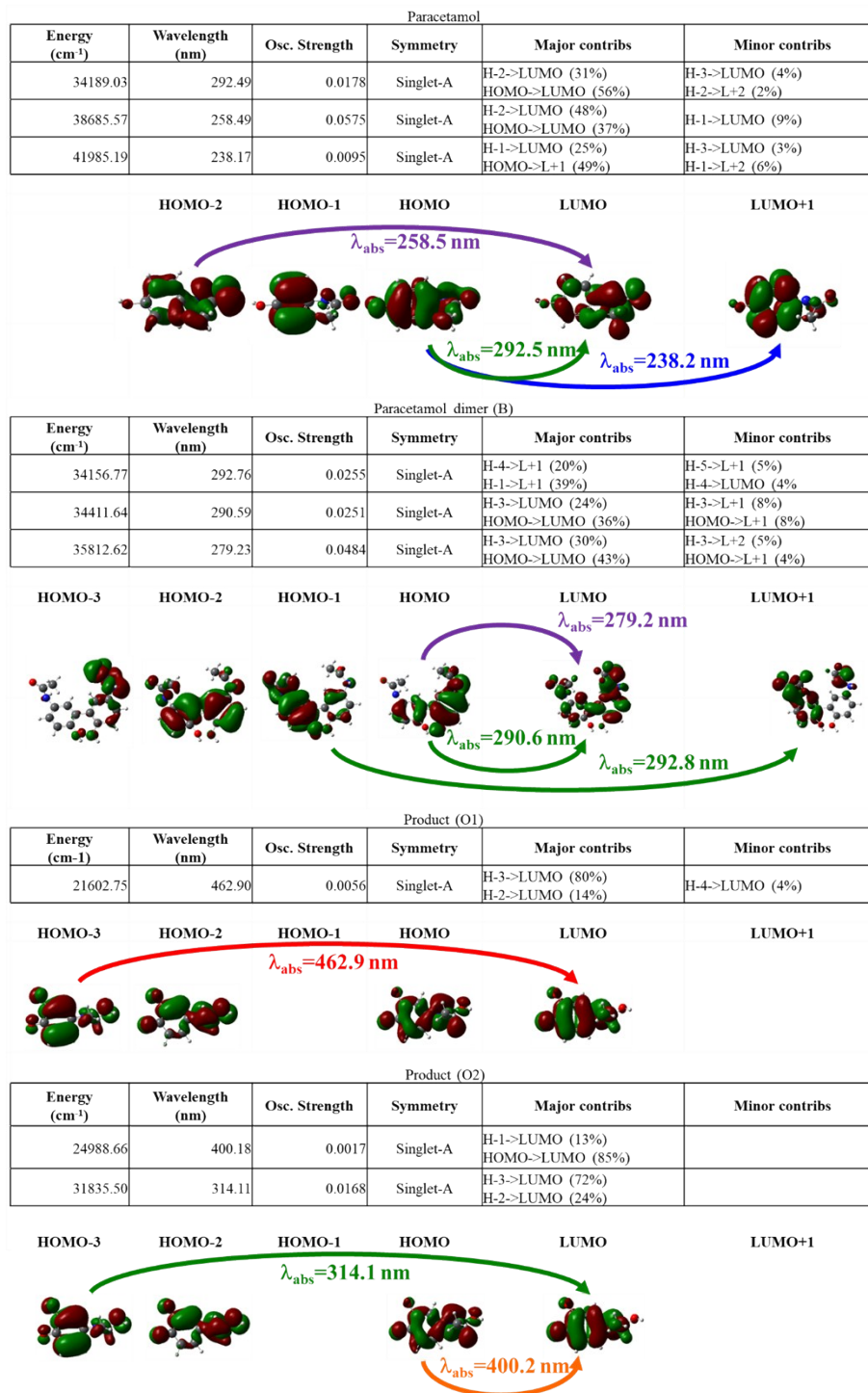


Figure S5. Theoretical parameters for UV-Vis spectra of HPLC identified products through (B3LYP/6-311G+)

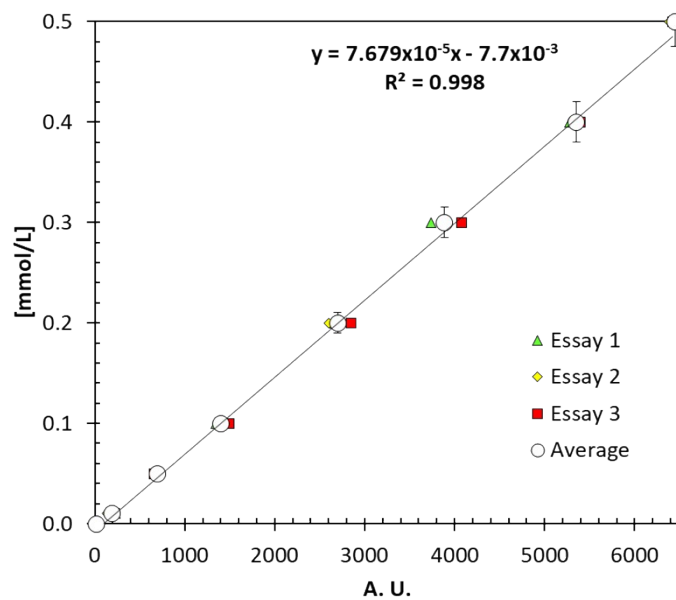


Figure S6. Calibration curve for identification of paracetamol by HPLC

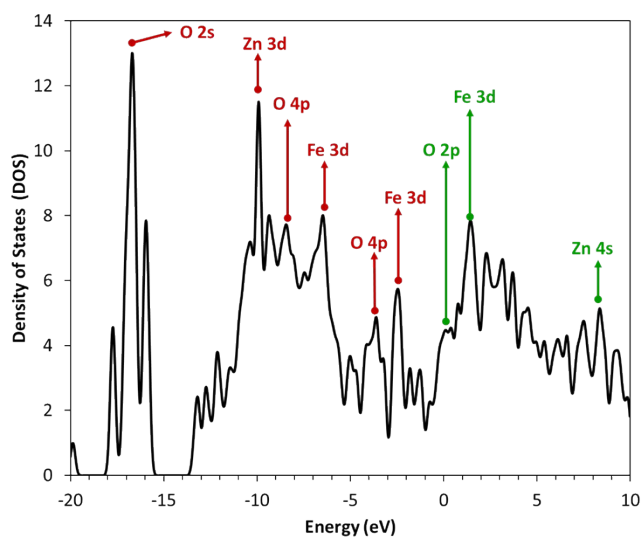


Figure S7. Total Density of States (DOS) for spinel ZnFe_2O_4 .

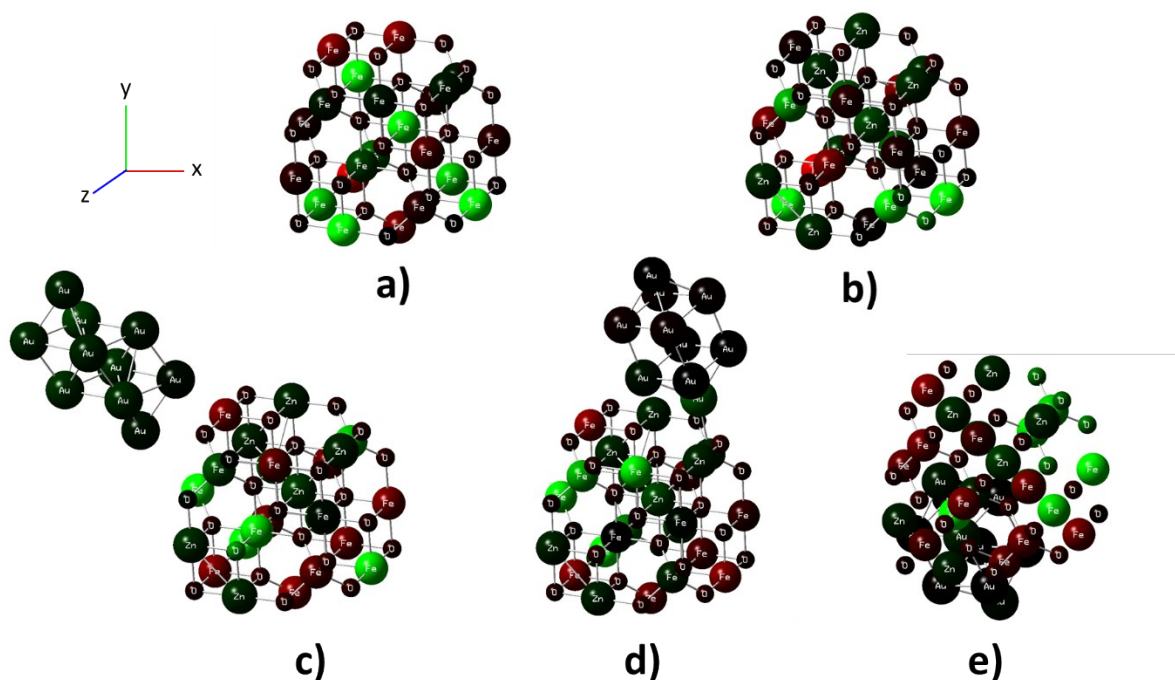


Figure S8. Mulliken charge for spinel ferrites with general formula MFe_2O_4 ($\text{M}=\text{Fe}^{2+}$, Zn^{2+}); a) Fe_3O_4 , b) ZnFe_2O_4 , c) $\text{ZnFe}_2\text{O}_4[0T0]\text{-Au}_{10}$, d) $\text{ZnFe}_2\text{O}_4[001]\text{-Au}_{10}$, e) $\text{ZnFe}_2\text{O}_4[1T1]\text{-Au}_{10}$.

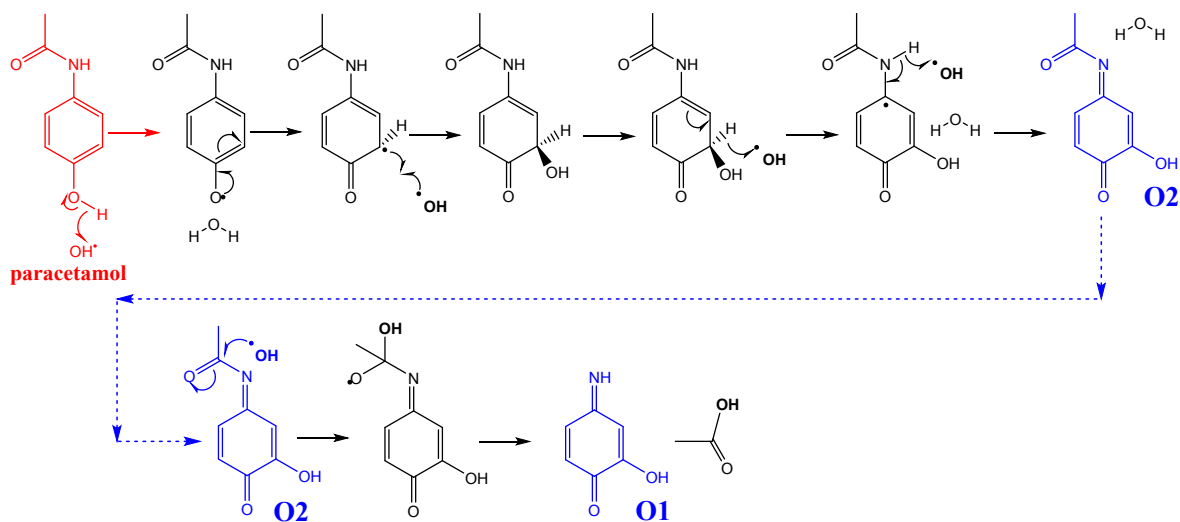


Figure S9. Detailed mechanism proposal for reaction pathway 1

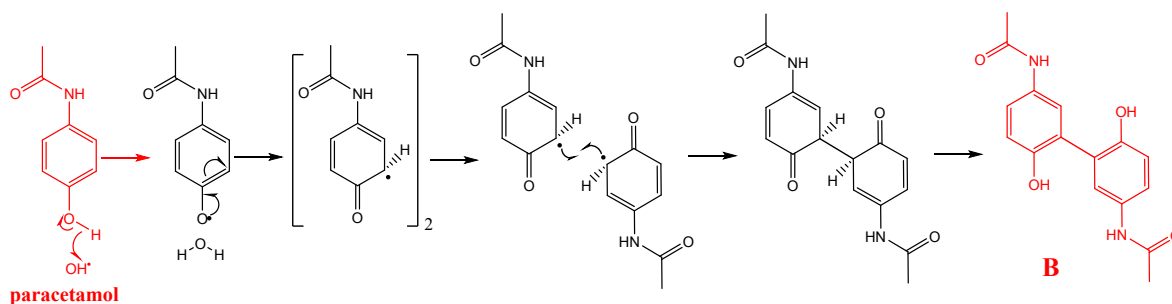


Figure S10. Detailed mechanism proposal for reaction pathway 2

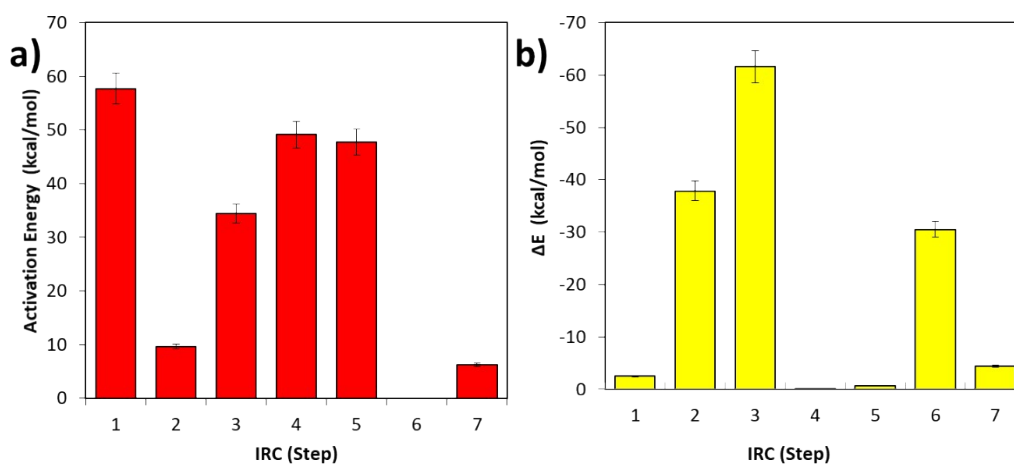


Figure 11. Thermodynamical features of reaction pathway 1 proposed for paracetamol oxidation by OH radicals; a) Activation Energy (E_A); b) Enthalpies of reaction (ΔH)

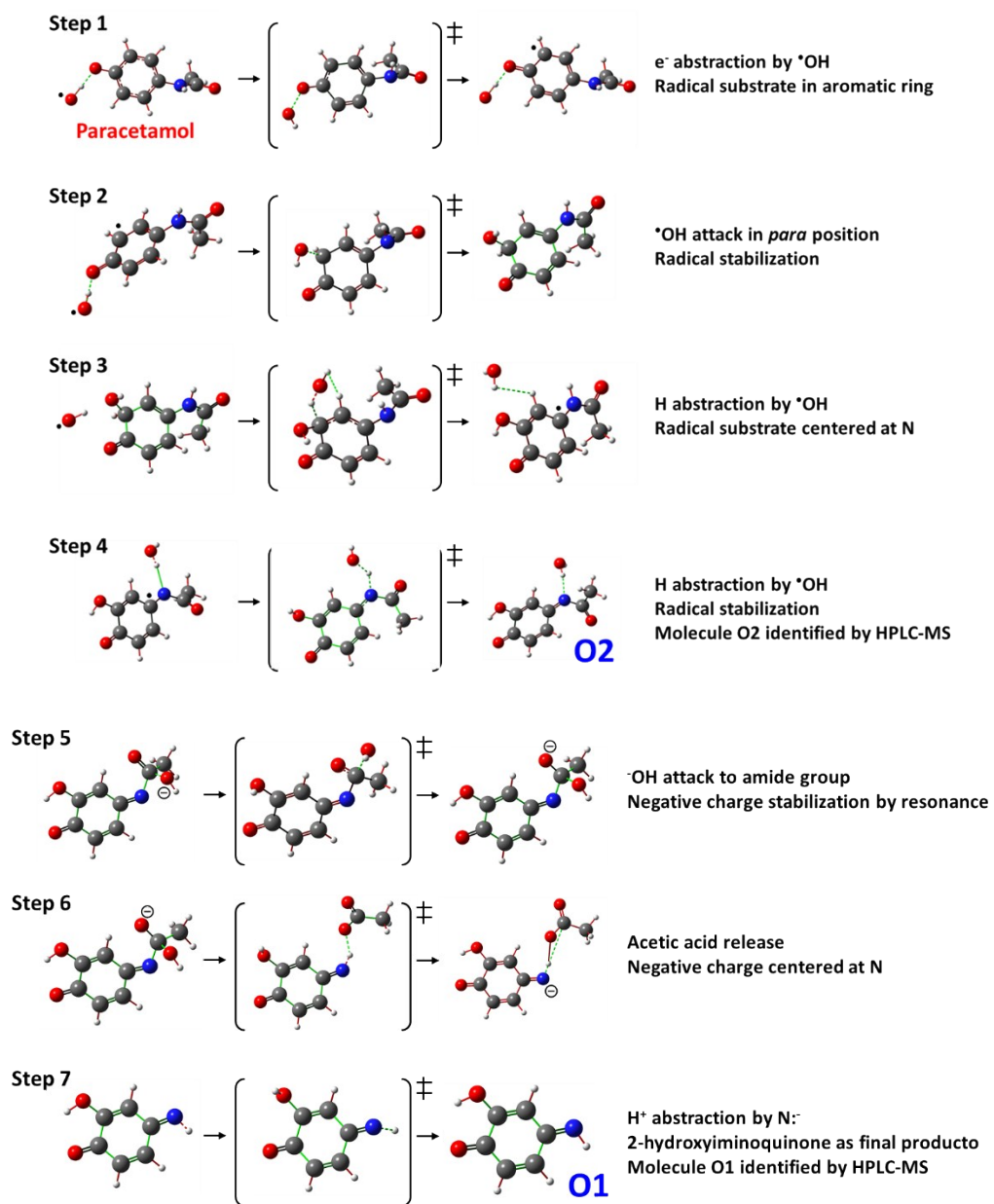


Figure S12. Reaction pathway summary

Table S1. Structure parameters calculated with HSEh1PBE functional compared with experimental values from spinel ZnFe_2O_4 .

<i>Parameter</i>	Fe_3O_4	ZnFe_2O_4	$\text{ZnFe}_2\text{O}_4[0\text{T}0]\text{-Au}_{10}$	$\text{ZnFe}_2\text{O}_4[001]\text{-Au}_{10}$	$\text{ZnFe}_2\text{O}_4[1\text{T}1]\text{-Au}_{10}$	ZnFe_2O_4 (exp.)
Lattice parameter (Å)	7.97	7.98	7.98	7.99	7.99	8.27
Volume (Å ³)	506.26	508.16	508.59	510.08	510.12	565.60
Relative deviation (%)	3.76	3.63	3.63	3.52	3.52	-
$\Delta E_{\text{HOMO-LUMO}}$ (eV)	0.08	0.1	0.43	0.13	0.09	-
Bandgap (eV)	2.73	2.07	1.44	2.79	2.13	1.9
Relative deviation (%)	2.56	-8.21	31.94	-31.8	-10.79	-
Fe-O (Å)	2.08	2.08	2.09	2.09	2.09	2.04
Zn-O (Å)	-	2.02	2.02	2.03	2.02	1.99

Table S2. Thermodynamical parameters for spinel ferrites with general formula MFe_2O_4 (M= Fe^{2+} , Zn^{2+}); a) Fe_3O_4 , b) ZnFe_2O_4 , c) $\text{ZnFe}_2\text{O}_4[0\text{T}0]\text{-Au}_{10}$, d) $\text{ZnFe}_2\text{O}_4[001]\text{-Au}_{10}$, e) $\text{ZnFe}_2\text{O}_4[1\text{T}1]\text{-Au}_{10}$.

<i>Parameter</i>	Fe_3O_4	ZnFe_2O_4	$\text{ZnFe}_2\text{O}_4[0\text{T}0]\text{-Au}_{10}$	$\text{ZnFe}_2\text{O}_4[001]\text{-Au}_{10}$	$\text{ZnFe}_2\text{O}_4[1\text{T}1]\text{-Au}_{10}$
Total Energy (kJ/mol)	-13.8x10 ⁶	-12.7x10 ⁶	-16.1x10 ⁶	-16.1x10 ⁶	-15.9x10 ⁶
Dipole moment (Dy)	299.7	380.5	749.2	364.7	951.7

Table S3. Atomic Mulliken charge analysis of studied ferrites

<i>Atom</i>	<i>Fe₃O₄</i>	<i>ZnFe₂O₄</i>	<i>ZnFe₂O₄[0T0]-Au₁₀</i>	<i>ZnFe₂O₄[001]-Au₁₀</i>	<i>ZnFe₂O₄[1T1]-Au₁₀</i>
<i>Mulliken charge</i>					
Fe ²⁺	-0.69-(-2.98)	-0.21-(-2.61)	-2.86-(-3.56)	-0.06-(-3.55)	-1.88-(-3.24)
Zn ²⁺	-	0.94-1.43	0.90-1.20	0.88-1.74	0.81-1.48
Fe ³⁺	0.50-6.67	0.94-6.35	0.34-7.30	0.24-6.03	1.00-8.33
O ²⁻	-0.44-(-0.94)	-0.49-(-1.10)	-0.12-(-1.4)	-0.44-(-1.18)	-0.37-(-1.41)
Au ⁰	-	-	0.51-1.23	-0.40-2.27	0.34-0.91

Table S4. Thermodynamic values for calculated reaction pathway 1 in paracetamol degradation

IRC	Total Charge	Spin	Activation Energy (EA)	Enthalpy (ΔE)	Total enthalpy (ΣΔE)
(Step)			(kcal/mol)		
1	-1	doublet	57.724	-2.508	-2.508
2	-1	doublet	9.594	-37.893	-40.401
3	0	doublet	34.458	-61.575	-101.977
4	0	singlet	49.151	-0.010	-101.987
5	-1	singlet	47.759	-0.674	-102.662
6	-1	singlet	-14.478	-30.535	-133.198
7	0	Singlet	6.196	-4.439	-137.637