

## Supporting Information

### Dual Tuning of Nickel Sulfide Nanoflake Array Electrocatalyst through Nitrogen Doping and Carbon Coating for Efficient and Stable Water Splitting

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Chengchun Tang<sup>a,b</sup>

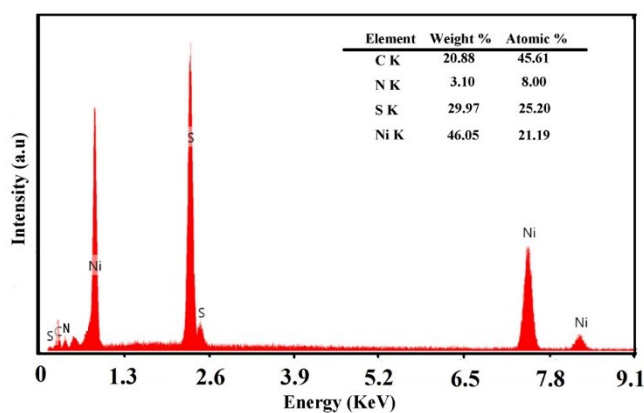
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**Figure S1.** The EDS of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF

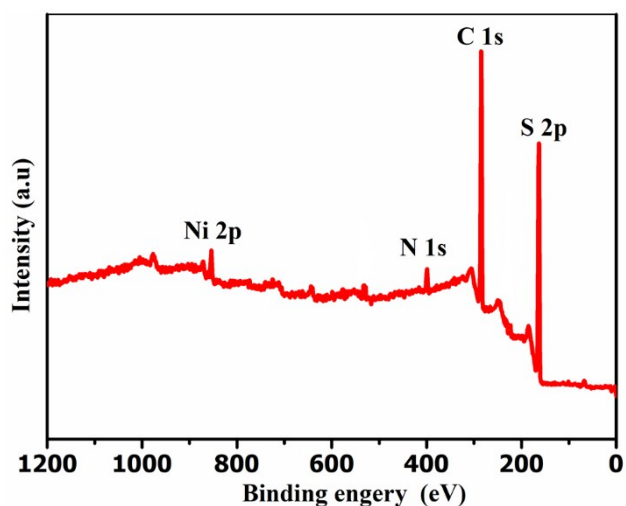


Figure S2. Survey XPS spectra of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF

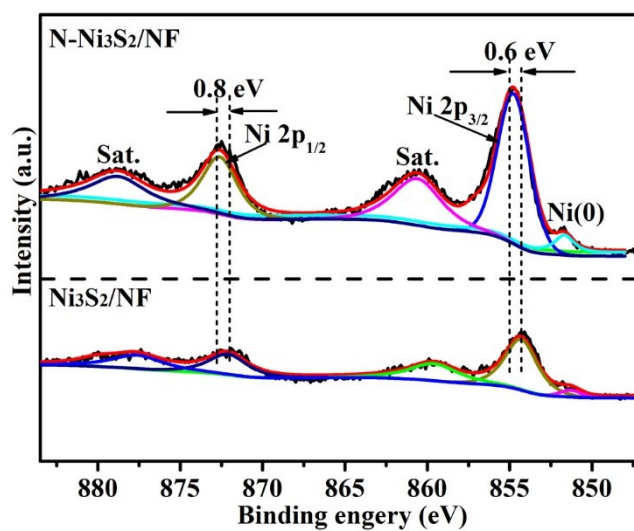


Figure S3. The XPS spectra of Ni 2p for Ni<sub>3</sub>S<sub>2</sub>/NF and N-Ni<sub>3</sub>S<sub>2</sub>/NF

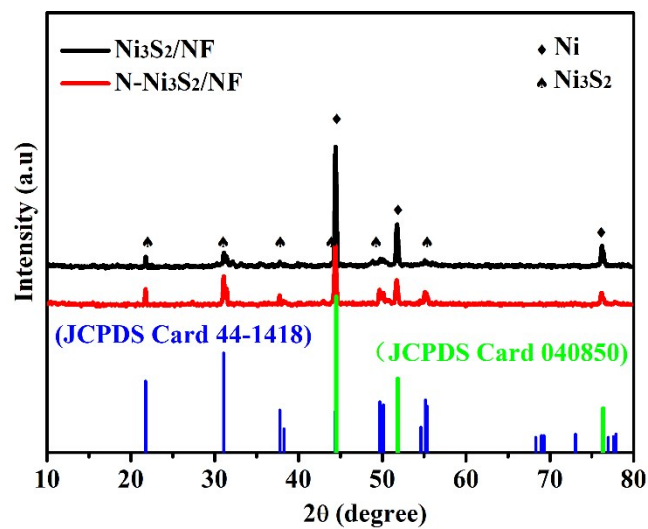
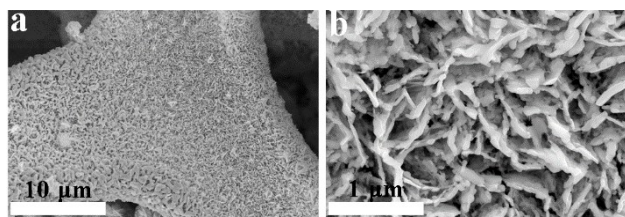
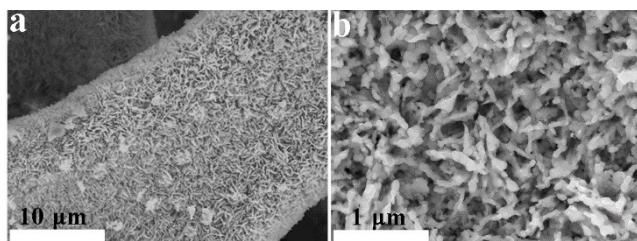


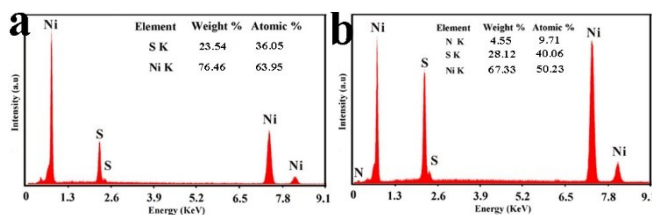
Figure S4. The XRD pattern of Ni<sub>3</sub>S<sub>2</sub>/NF and N-Ni<sub>3</sub>S<sub>2</sub>/NF



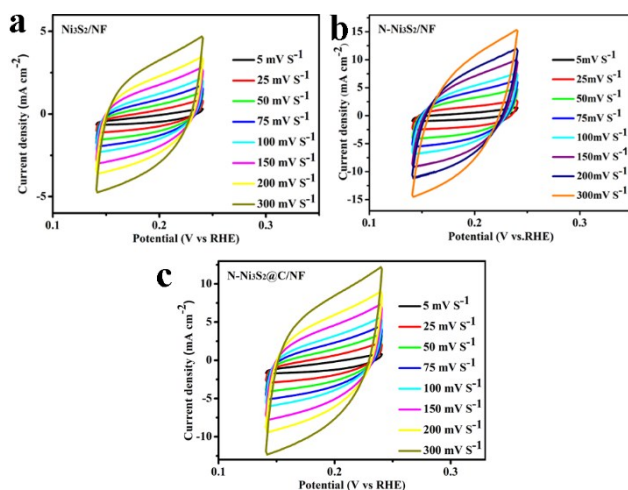
**Figure S5.** The SEM image of  $\text{Ni}_3\text{S}_2/\text{NF}$  with different magnifications



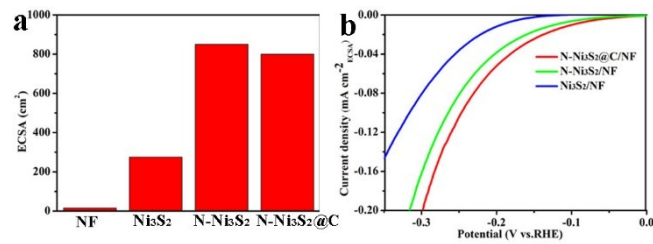
**Figure S6.** The SEM image of  $\text{N-Ni}_3\text{S}_2/\text{NF}$  with different magnifications



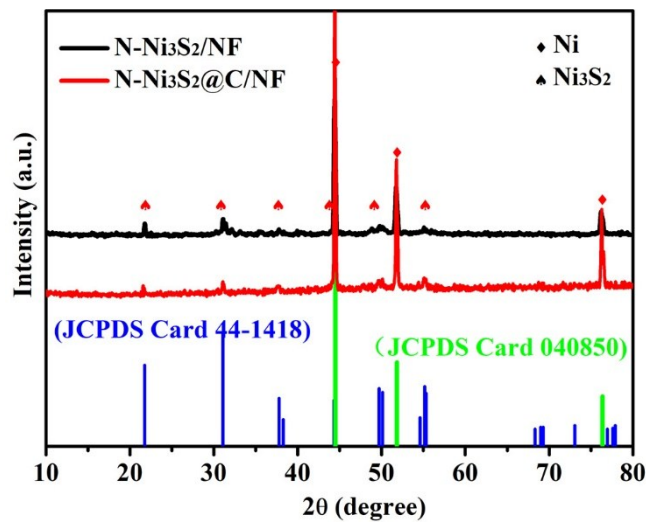
**Figure S7.** a) The EDS of  $\text{Ni}_3\text{S}_2/\text{NF}$  and b)  $\text{N-Ni}_3\text{S}_2/\text{NF}$



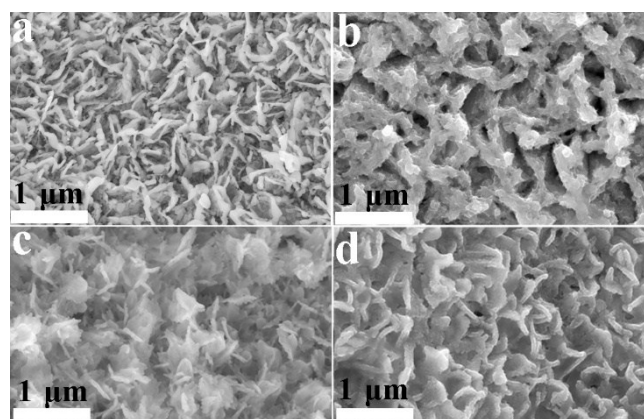
**Figure S8.** a) Typical cyclic voltammetry curves of  $\text{N-Ni}_3\text{S}_2@\text{C}/\text{NF}$ , b)  $\text{N-Ni}_3\text{S}_2/\text{NF}$  and c)  $\text{Ni}_3\text{S}_2/\text{NF}$  in 1M KOH with different scan rates



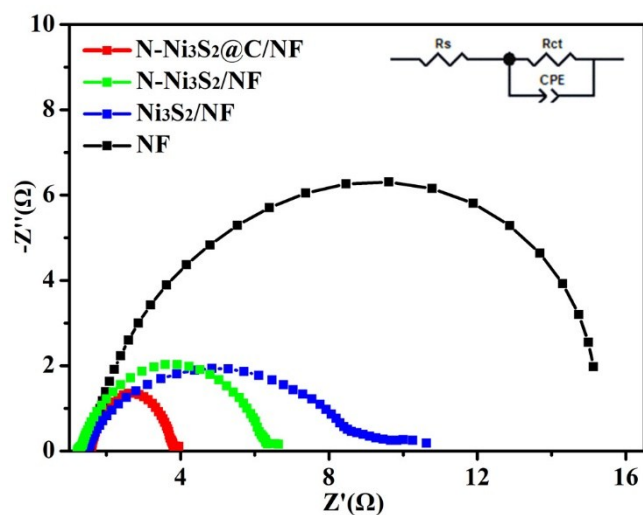
**Figure S9.** a) The ECSA of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF, N-Ni<sub>3</sub>S<sub>2</sub>/NF, Ni<sub>3</sub>S<sub>2</sub>/NF and NF. b) Polarization curves from normalized to the electrochemical active surface area (ECSA) for HER.



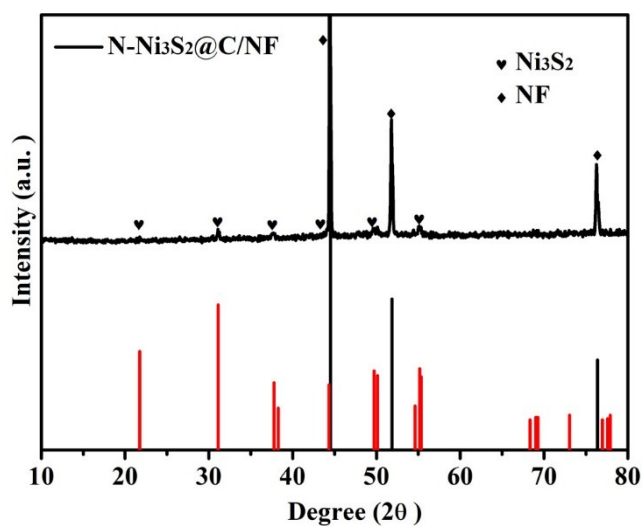
**Figure S10.** The XRD of N-Ni<sub>3</sub>S<sub>2</sub>/NF and N-Ni<sub>3</sub>S<sub>2</sub>@C/NF after HER



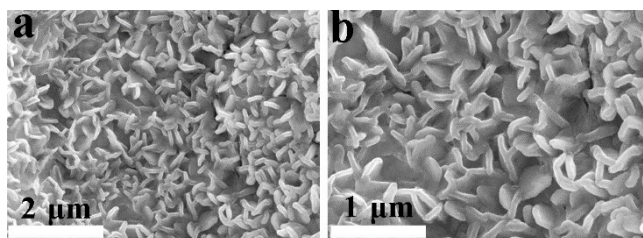
**Figure S11.** SEM images of a-b) N-Ni<sub>3</sub>S<sub>2</sub>/NF and c-d) N-Ni<sub>3</sub>S<sub>2</sub>@C/NF before and after HER in 1 M KOH



**Figure S12.** Electrochemical impedance spectroscopy (EIS) Nyquist plots at 1.53 V vs. RHE for all the synthesized materials.



**Figure S13.** The XRD of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF after OER



**Figure S14.** a-b) The SEM image of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF after OER in 1 M KOH with different magnifications

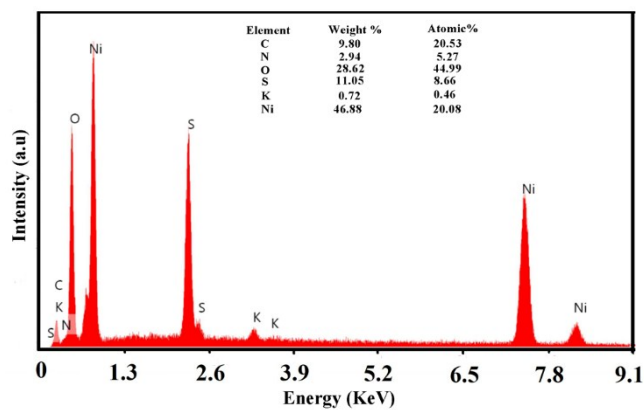


Figure S15. The EDS of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF after OER

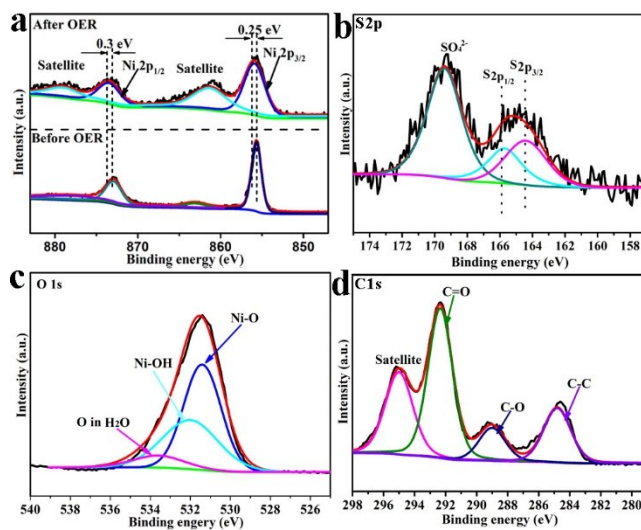
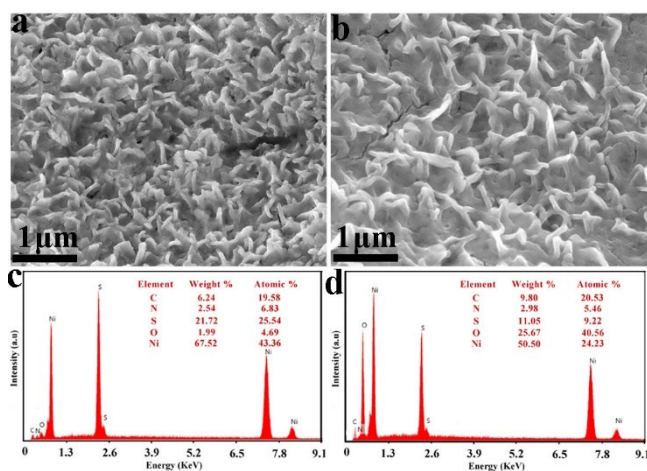
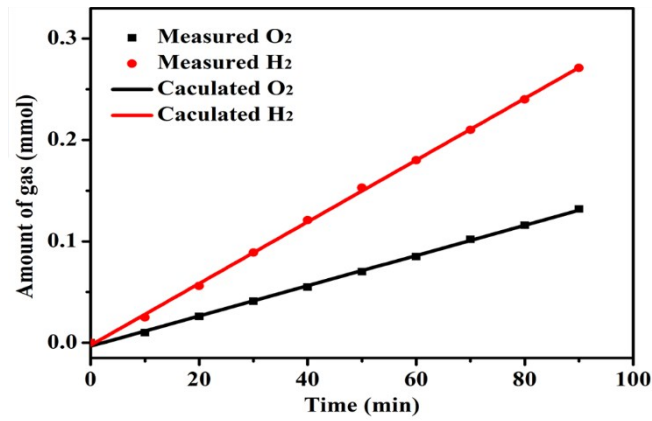


Figure S16. The XPS spectra of a) Ni 2p, b) S 2p, c) O 1s and d) C 1s for N-

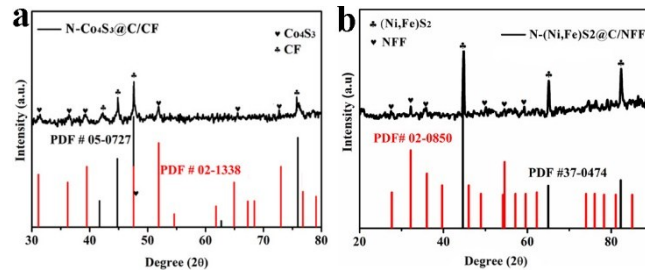
Ni<sub>3</sub>S<sub>2</sub>@C/NF after OER



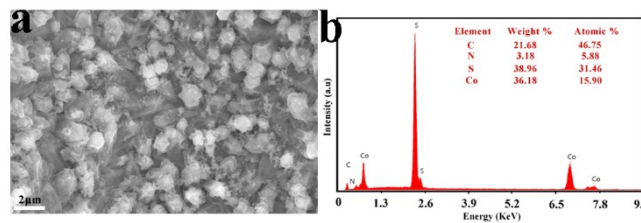
**Figure S17.** SEM images of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF after overall water splitting. a) cathode  
 b) anode and EDS results of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF after overall water splitting. c) cathode  
 d) anode



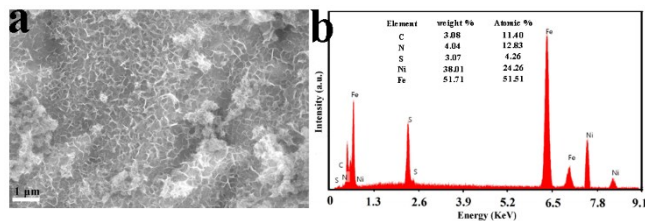
**Figure S18.** The amount of gas theoretically calculated and experimentally measured versus time for overall water splitting.



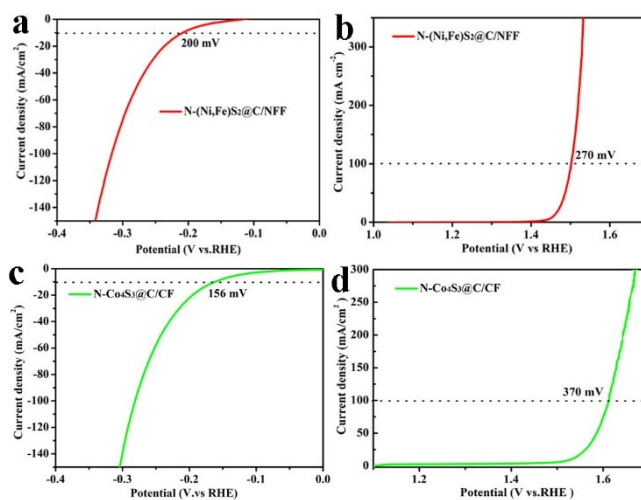
**Figure S19.** The XRD of a) N-Co<sub>4</sub>S<sub>3</sub>@C/CF and b) N-(Ni,Fe)S<sub>2</sub>@C/NFF



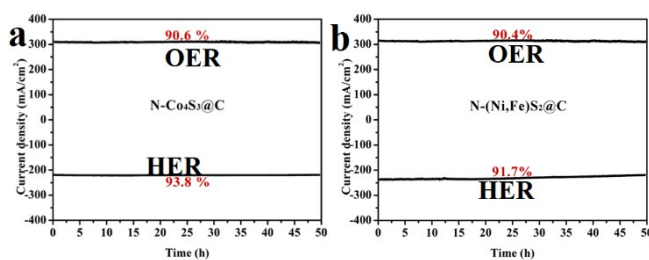
**Figure S20.** a) The SEM image and b) EDS of N-Co<sub>4</sub>S<sub>3</sub>@C/CF



**Figure S21.** a) The SEM image and b) EDS of N-(Ni,Fe)S<sub>2</sub>@C/NFF



**Figure S22.** a) *iR*-corrected linear sweep voltammetry curves of N-(Ni,Fe)S<sub>2</sub>@C/NFF for HER and b)OER in 1 M KOH.c) *iR*-corrected linear sweep voltammetry curves of N-Co<sub>4</sub>S<sub>3</sub>@C/CF for HER and d)OER in1 M KOH.



**Figure S23.** a) The *i-t* curves of N-Co<sub>4</sub>S<sub>3</sub>@C, b) N-(Ni,Fe)S<sub>2</sub>@C for HER and OER in 1 M KOH .



**Table S1** Comparison of HER performance of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF product with various well-developed electrocatalysts in the literature.

| Catalyst                                    | Electrolyte | J<br>(mA cm <sup>-2</sup> ) | $\eta$<br>(mV) | Tafel<br>(mV dec <sup>-1</sup> ) | Ref.      |
|---|-------------|-----------------------------|----------------|----------------------------------|-----------|
| N-Ni <sub>3</sub> S <sub>2</sub> @C/NF      | 1 M KOH     | 10                          | 113            | 90                               | This work |
| N-Ni <sub>3</sub> S <sub>2</sub> /NF        | 1 M KOH     | 10                          | 110            | -                                | 1         |
| Ni <sub>3</sub> S <sub>2</sub> nanorod      | 1 M KOH     | 10                          | 200            | 107                              | 2         |
| Ni <sub>3</sub> S <sub>2</sub> nanosheet    | 1 M KOH     | 10                          | 223            | -                                | 3         |
| NiS microsphere                             | 1 M KOH     | 20                          | 158            | 83                               | 4         |
| Fe <sub>2</sub> Ni <sub>2</sub> N           | 1 M KOH     | 10                          | 110            | -                                | 5         |
| NiCo <sub>2</sub> S <sub>4</sub> nanowire   | 1 M KOH     | 10                          | 210            | 59                               | 6         |
| Ni <sub>3</sub> S <sub>2</sub> needle array | 1 M KOH     | 10                          | 117            | 130                              | 7         |
| NiFe/NiCo <sub>2</sub> O <sub>4</sub> /NF   | 1 M KOH     | 10                          | 105            | 88                               | 8         |
| NiFe@NC                                     | 1 M KOH     | 10                          | 200            | -                                | 9         |
| CoP   | 1 M KOH     | 10                          | 209            | 51                               | 10        |
| Co <sub>3</sub> Se <sub>4</sub> /CF         | 1 M KOH     | 10                          | 179            | -                                | 11        |

**Table S2** Comparison of OER performance of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF product with various well-developed electrocatalysts in the literature

| Catalyst                                 | Electrolyte | J<br>(mA cm <sup>-2</sup> ) | $\eta$<br>(mV) | Tafel<br>(mV dec <sup>-1</sup> ) | Ref.      |
|--|-------------|-----------------------------|----------------|----------------------------------|-----------|
| N-Ni <sub>3</sub> S <sub>2</sub> @C/NF   | 1M KOH      | 100                         | 310            | 75                               | This work |
| N-Ni <sub>3</sub> S <sub>2</sub> /NF     | 1M KOH      | 170                         | 350            | 70                               | 1         |
| Ni <sub>3</sub> S <sub>2</sub> nanorod   | 1M KOH      | 100                         | 370            | 140                              | 12        |
| Ni <sub>3</sub> S <sub>2</sub> nanosheet | 1M KOH      | 100                         | 350            | -                                | 3         |
| NiS microsphere                          | 1M KOH      | 50                          | 335            | 120                              | 4         |
| NiFe-LDH                                 | 1M KOH      | 100                         | 300            | 47                               | 13        |
| NiCoFe LDH                               | 1M KOH      | 30                          | 233            | 53                               | 14        |
| NiCo LDH                                 | 1M KOH      | 100                         | 340            | 57                               | 15        |
| FeCoW                                    | 1M KOH      | 10                          | 191            | -                                | 16        |

**Table S3** Comparison of water splitting cell voltage of N-Ni<sub>3</sub>S<sub>2</sub>@C/NF product with various well-developed electrocatalysts in the literature

| Catalyst   | Electrolyte | J<br>(mA cm <sup>-2</sup> ) | Overall<br>voltage (V) | Durability<br>(h) | Ref.      |
|--|-------------|-----------------------------|------------------------|-------------------|-----------|
| N-Ni <sub>3</sub> S <sub>2</sub> @C/NF                               | 1 M KOH     | 10                          | 1.57                   | 140               | This work |
| N-Ni <sub>3</sub> S <sub>2</sub> /NF                                 | 1 M KOH     | 10                          | 1.48                   | 8                 | 1         |
| NiS microsphere  | 1 M KOH     | 10                          | 1.64                   | 35                | 4         |
| NiSe nanowire  | 1 M KOH     | 10                          | 1.63                   | 20                | 17        |
| Fe <sub>2</sub> Ni <sub>2</sub> N nanoarrays                         | 1 M KOH     | 10                          | 1.65                   | 10                | 5         |
| Ni@Co-Ni-P   | 1 M KOH     | 100                         | 1.61                   | 528               | 18        |
| NiCo <sub>2</sub> S <sub>4</sub> nanowire                            | 1 M KOH     | 10                          | 1.63                   | 50                | 6         |
| MoS <sub>2</sub> /Ni <sub>3</sub> S <sub>2</sub><br>Heterostructures | 1 M KOH     | 10                          | 1.56                   | 10                | 19        |
| CoS-Co(OH) <sub>2</sub>  | 1 M KOH     | 10                          | 1.58                   | 28                | 20        |
| Ni/Ni <sub>8</sub> P <sub>3</sub>                                    | 1 M KOH     | 10                          | 1.61                   | 10                | 21        |
| NiCoP  | 1 M KOH     | 10                          | 1.58                   | 12                | 22        |

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