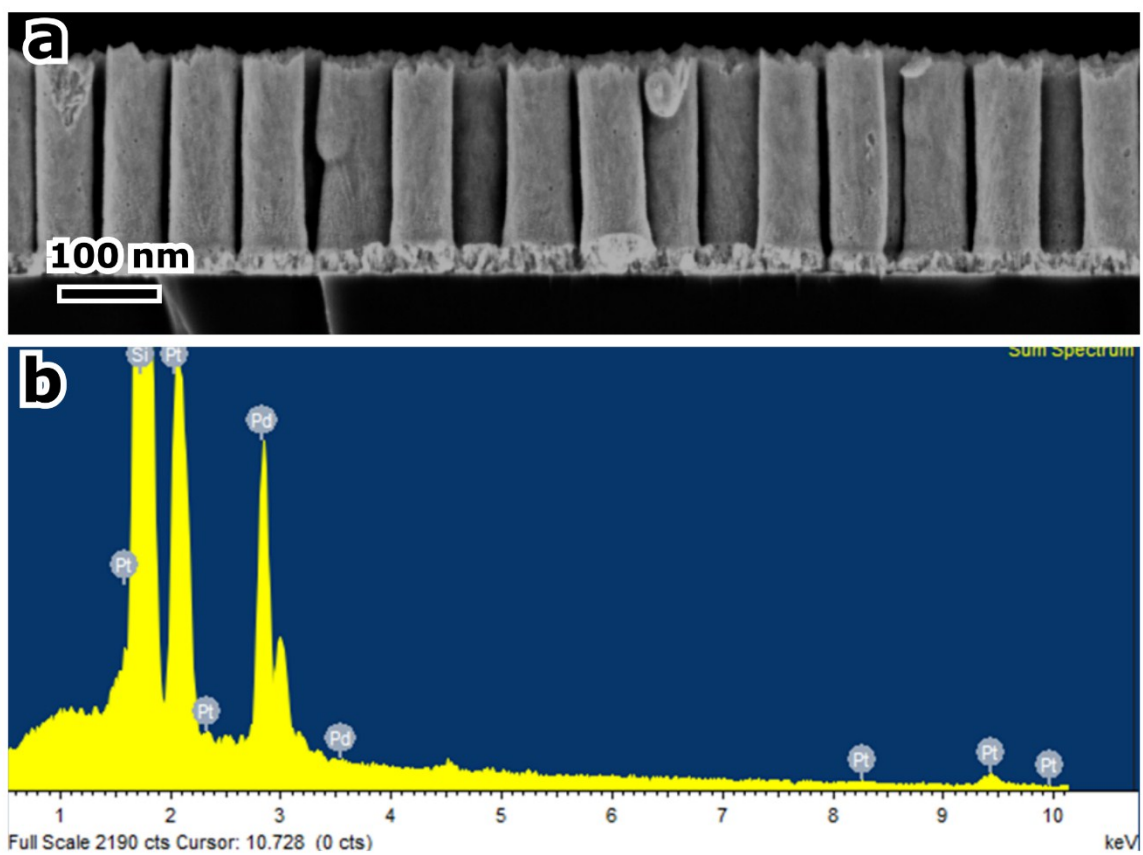


## Supplementary Information

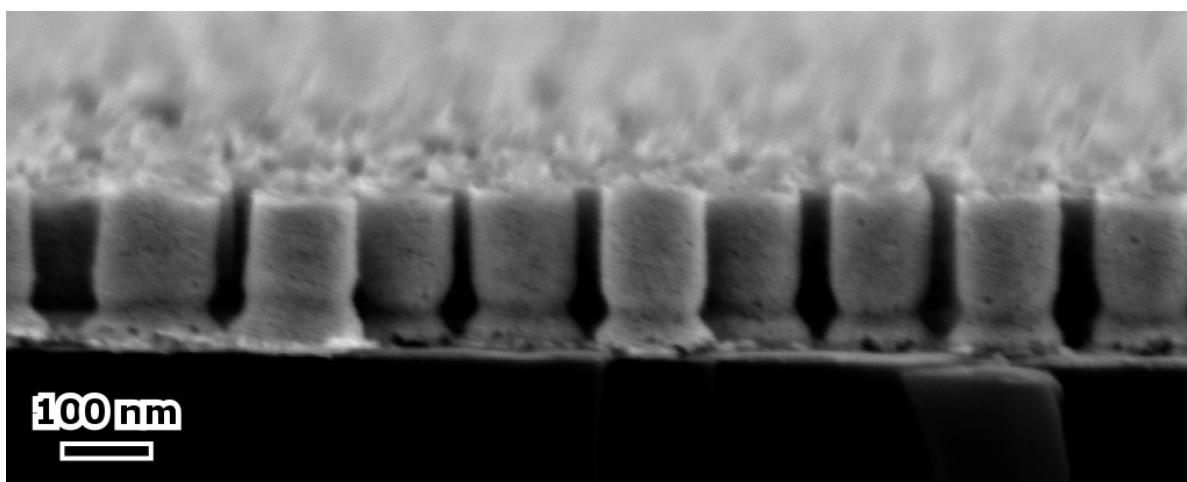
### **Porous PtPd alloy nanotubes. Towards high performance electrocatalysts with low Pt-loading**

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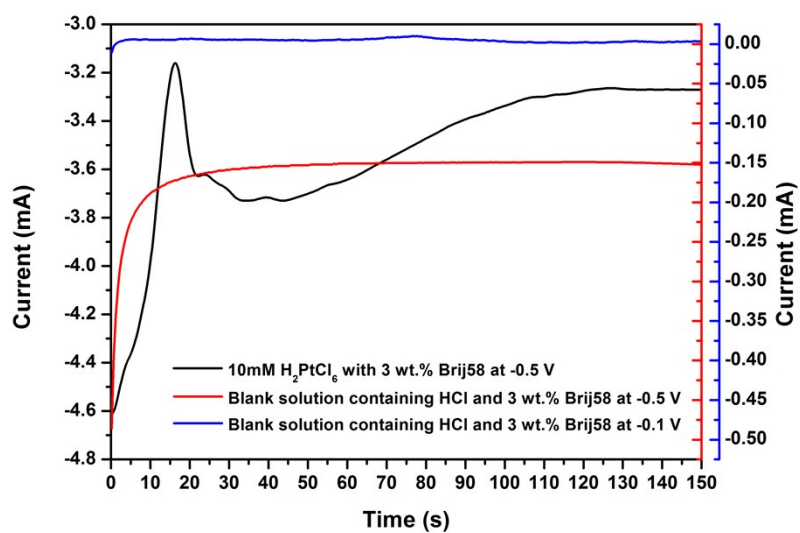
<sup>a</sup> Institute for Materials & Surface Technology, University of Applied Sciences, Kiel, Germany



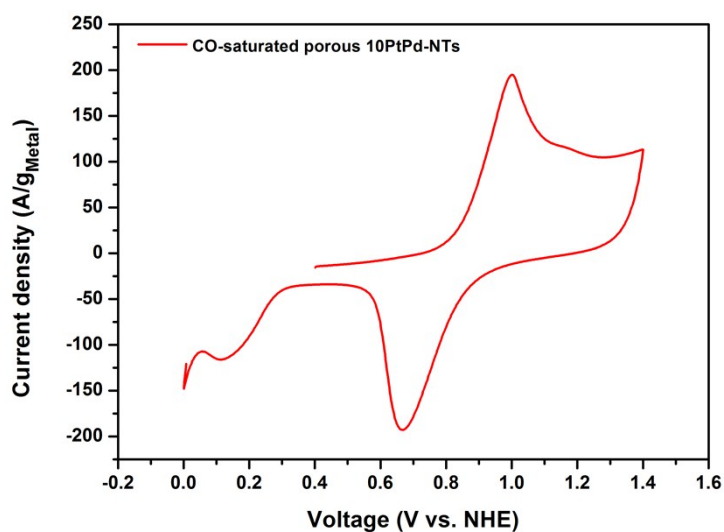
**Fig. S1** SEM micrograph showing a cross section of a PtPd alloy NRs sample (a) , and corresponding EDS spectrum (b).



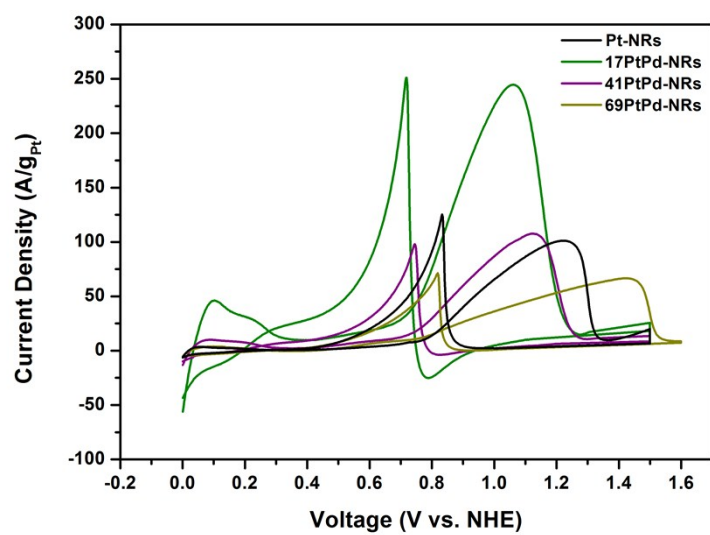
**Fig. S2** Cross-section of Pt array electrodeposited at -0.5 V with no Brij58 in the solution.



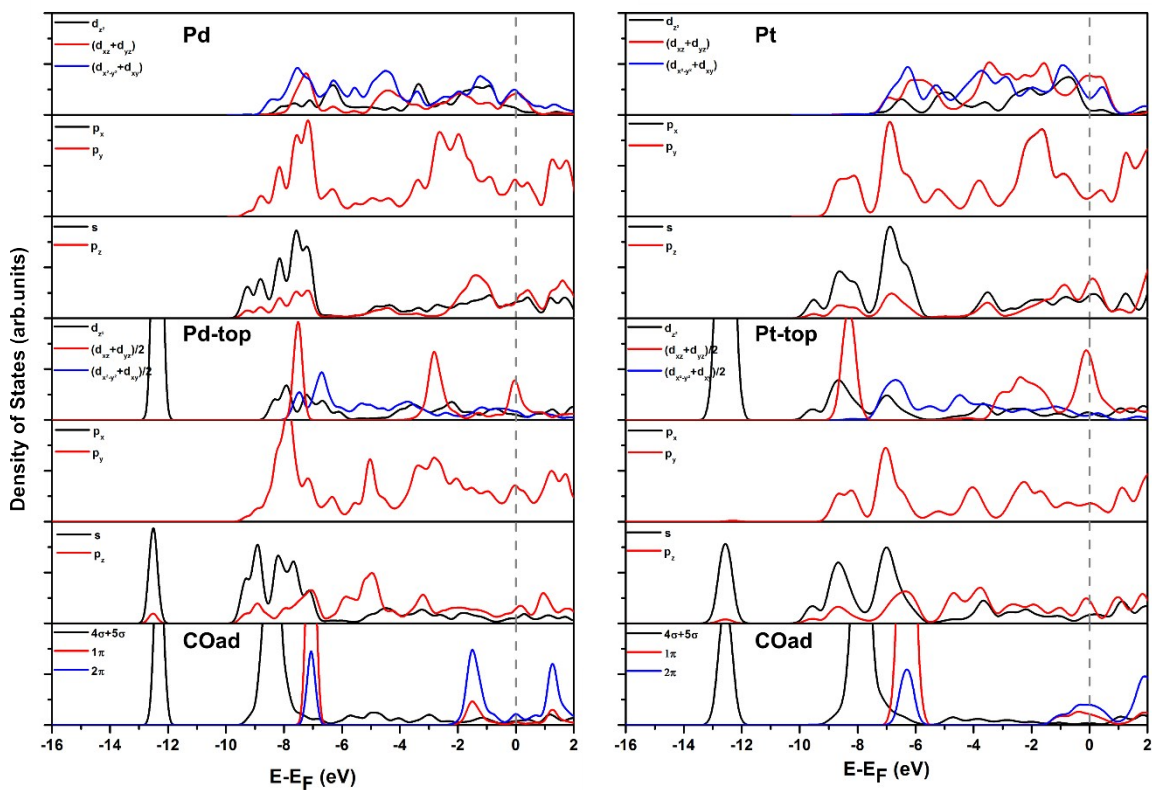
**Fig. S3** The chronoamperometric curves corresponding to the potentiostatic electrodeposition of the porous Pt NTs at -0.5 V, and hydrogen evolution test done in a blank solution containing HCl and 3 wt.% Brij58 ( pH=2.5) at -0.1 V and -0.5 V.



**Fig. S4** CO oxidation voltammogram of a CO-saturated porous 10PtPd-NTs in CO-free 0.5 M  $H_2SO_4$  electrolyte. Scan rate:  $50\text{ mV s}^{-1}$ .



**Fig. S5** Current density (current normalized by the Pt-mass) vs. voltage of the plain PtPd-NRs containing different Pt-contents in 0.5 M H<sub>2</sub>SO<sub>4</sub> + 0.5 M CH<sub>3</sub>OH. Scan rate: 50 mV s<sup>-1</sup>.



**Fig. S6** Orbital electronic density of states for top sites of pure Pt, and Pd atoms, and adsorbed CO. This Figure shows s, p, and d –states for pure Pd and Pt. The d band center is -2.41 eV for Pt and -1.94 eV for Pd, the s- and p-band are covering a large energy range.  $p_x$  and  $p_y$  degenerate, and the  $p_z$  show different behavior than later ones. Considering the effect of CO molecule on the metal states Pd and Pt, one can observe different qualitative behaviors, the d-band is extended to a large energy range; more important new states were a rise in  $d_{z^2}$  and  $sp_z$  from the interaction with  $\sigma$ . For CO the  $4\sigma+5\sigma$  peaks positions for Pt (-8.5 eV) and Pd (-8 eV), the  $2\pi$  get partially occupied at the same energy level with  $1\pi$ .

**Table S1** Volume fractions of 10 mM H<sub>2</sub>PtCl<sub>6</sub> and 10 mM K<sub>2</sub>PdCl<sub>4</sub> aqueous electrolytes, and their corresponding Pt-content obtained from EDS analysis, and from XRD (the calculated lattice parameters and Vegard's law).

Volume Fraction	Pt-Content from EDS (at%)	Pt-Content from XRD (at%)
Porous PtPd NTs		
1:19	5.5	5
2:18	10.54	10
3:16	17.44	18
4:16	22.40	23
7:12	49.04	43
Plain PtPd NRs		
1:4	10.78	17
4:5	42.24	41
7:4	68.51	69