

Supporting information

Ultrathin Carbon Coated CoO Nanosheet Arrays as Efficient Electrocatalysts for Hydrogen Evolution Reaction

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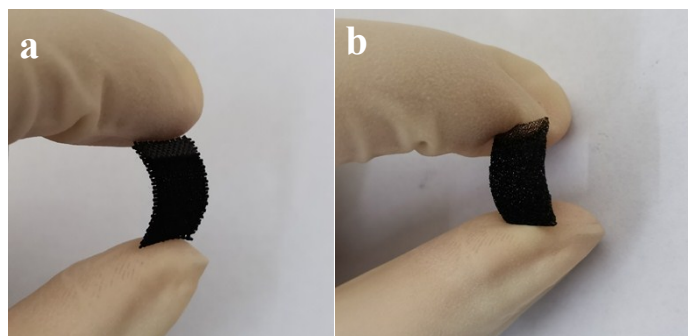


Figure. S1 The photographs of C@CoO/CC and C@CoO/NF under bending.

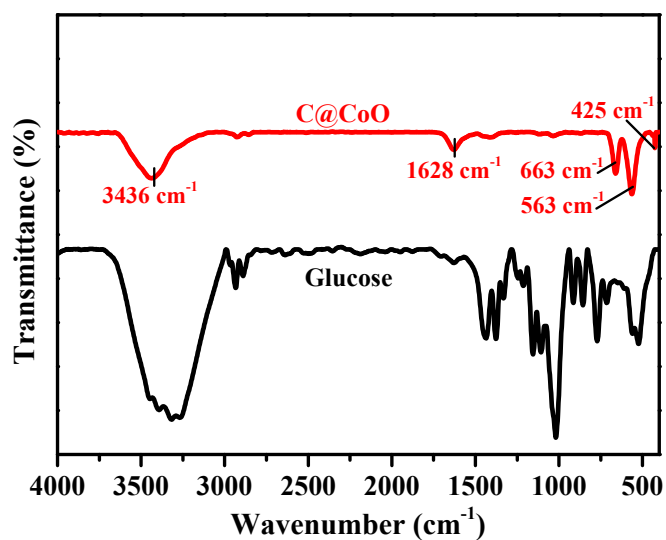


Figure. S2 The FTIR spectra of C@CoO and glucose.

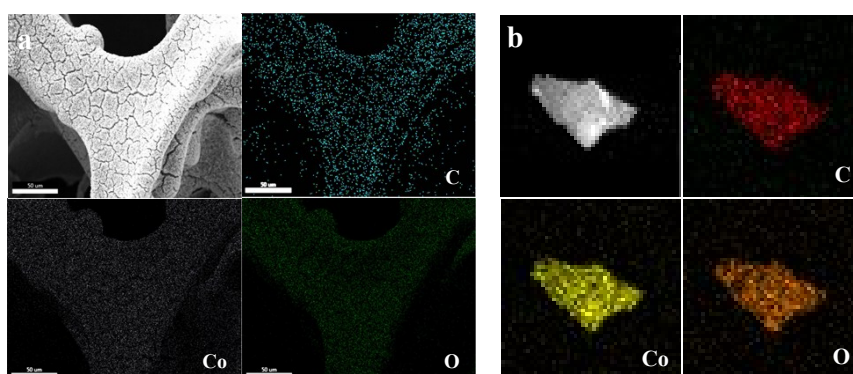


Figure. S3 (a) SEM image of C@CoO/NF and corresponding elemental mappings for C, Co and O atoms. (b) TEM image of C@CoO and corresponding elemental mappings for C, Co and O atoms.

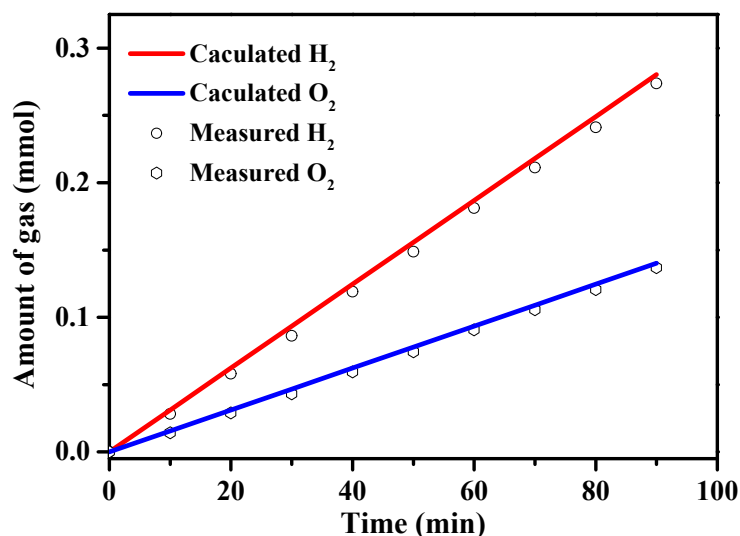


Figure. S4 The amount of gas theoretically calculated and experimentally measured versus time on cathode (C@CoO/CC) and anode (Carbon rod).

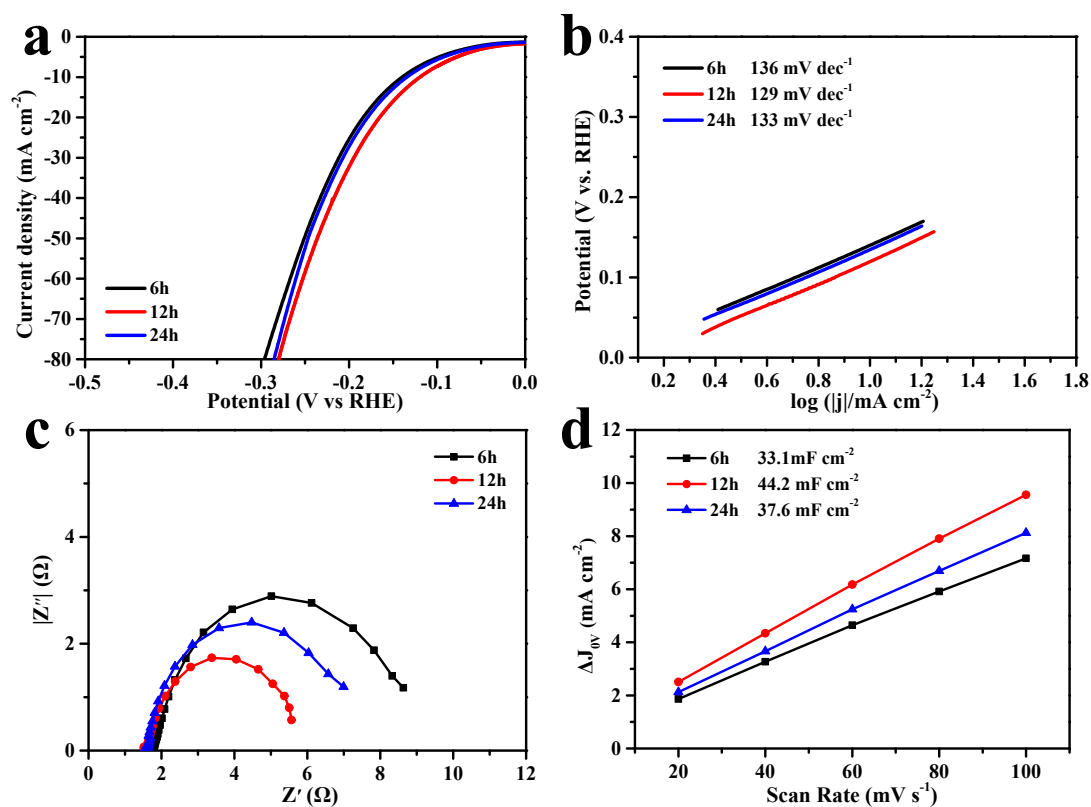


Figure. S5 (a) LSV curves of the C@CoO/CC prepared under different conditions with a scan rate of 2 mV s⁻¹ for HER in 1.0 M KOH. (b) The corresponding Tafel plots derived from (a). (c) Nyquist plots for HER tested at -0.15 V (vs. RHE). (d) The double layer capacity C_{dl} of electrocatalysts.

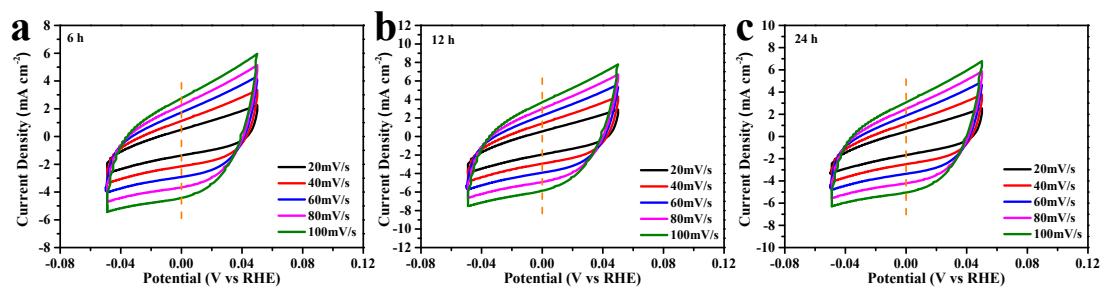


Figure S6. Different scan rates of CVs of C@CoO/CC prepared under different conditions.

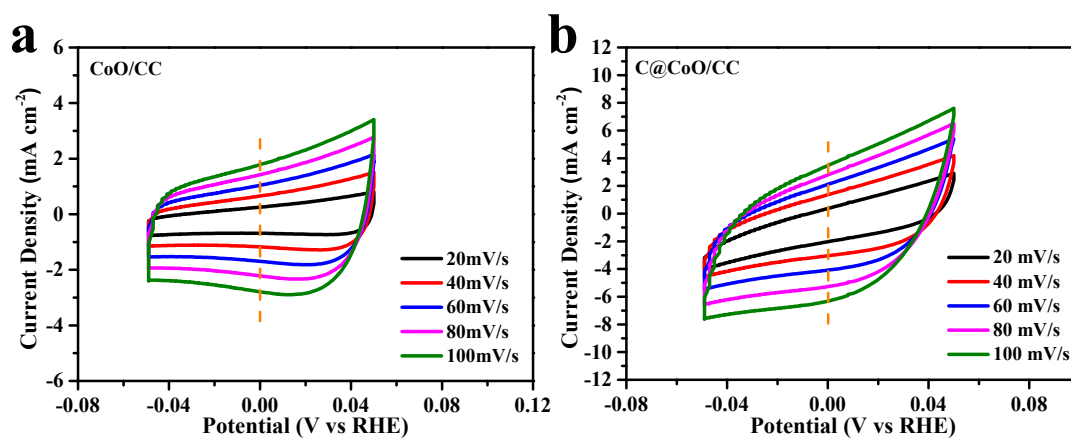


Figure S7. Different scan rates of CVs of (a) CoO/CC and (b) C@CoO/CC.

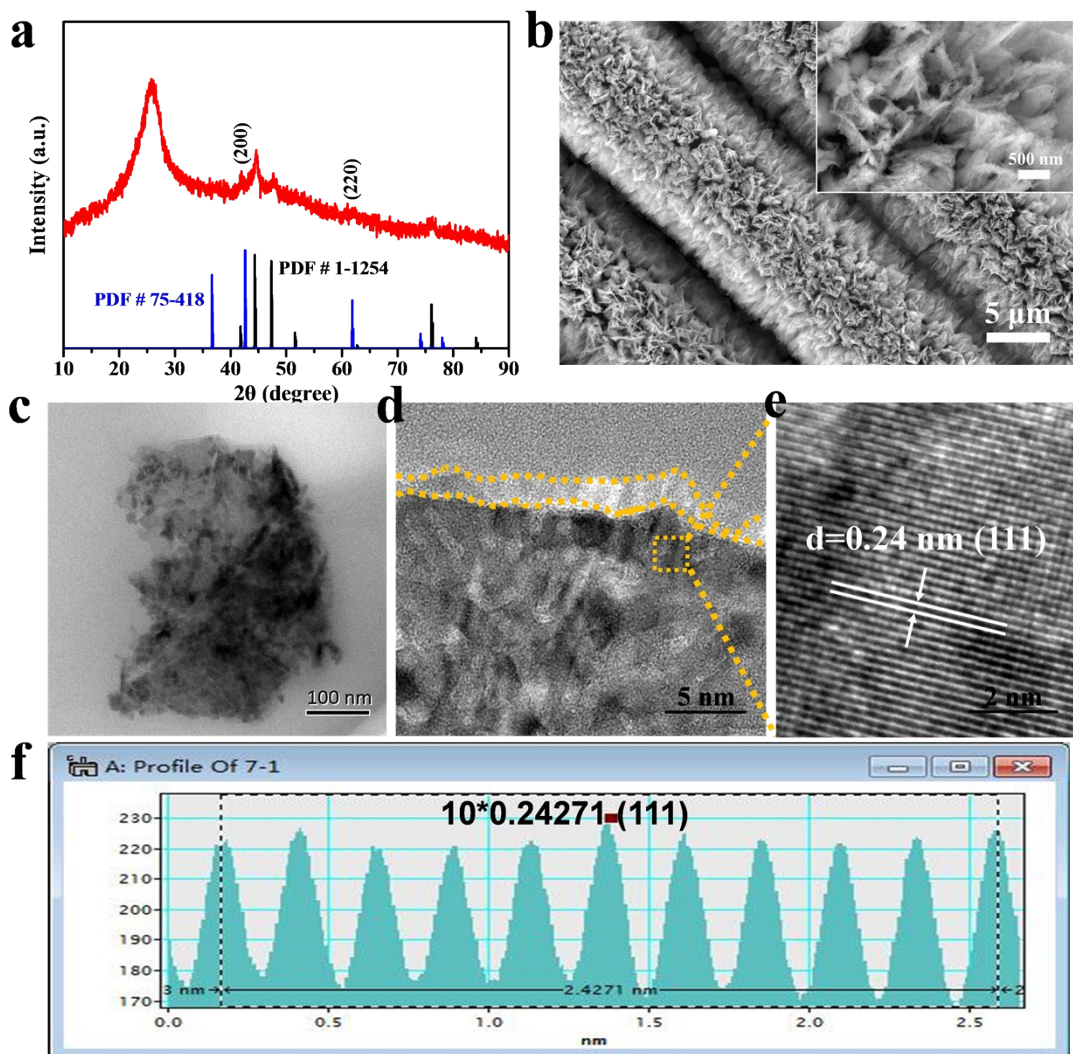


Figure S8. (a) The XRD pattern, (b) SEM, (c) TEM and (d, e) HRTEM images of C@CoO/CC after long-term durability test. (f) the screenshot of ten times for lattice spacing.

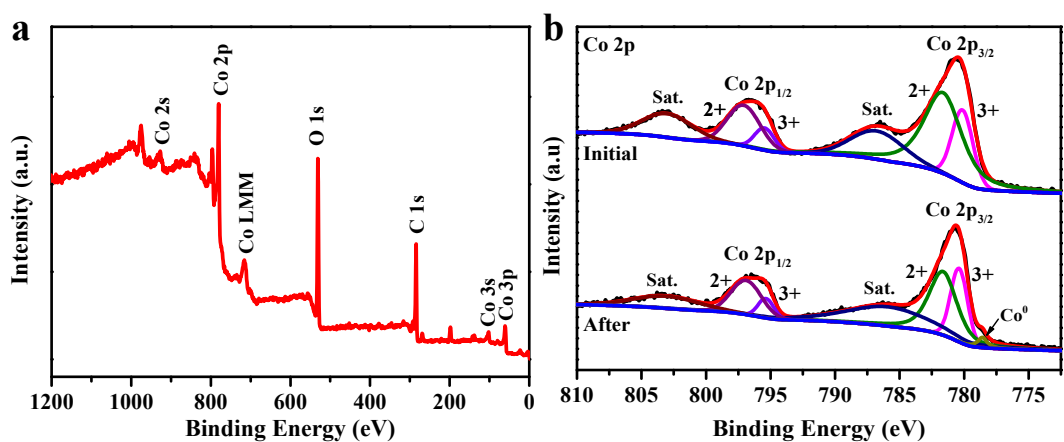


Figure S9. (a) XPS survey spectra and (b) Co 2p for C@CoO after long-term durability test.

Table S1. Comparison of HER performances of C@CoO/CC with previously reported transition metal oxide-based HER electrocatalysts.

Catalyst	Substrate	Electrolyte	$\eta_{@-10mA\ cm^{-2}}$ (mV vs RHE)	Tafel slope (mV dec ⁻¹)	Ref.
C@CoO	CC ^a	1M KOH	-120	129	This work
Ni/NiO-CNT	GCE ^b	1M KOH	~ -100	82	1
NiCo ₂ O ₄ @NiO@Ni	NF ^c	1M KOH	-124	58	2
Ni-NiO/N-rGO	NF	1M KOH	-135	46	3
Co ₃ O ₄ -MTA	NF	1M KOH	-190 (20 mA cm ⁻²)	98	4
CoO/MoS ₂	CC	1M KOH	-173	83	5
Cu ₂ O/Co ₃ O ₄ /DC	GCE	0.5M H ₂ SO ₄	-160	73	6
FeCoO	NF	1M KOH	-205	118	7
Co ₃ O ₄ @BNC	GCE	1M KOH	-178	100.3	8
NiFe LDH	NF	1M NaOH	-210	-	9
CoO/MoO _x	NF	1M KOH	-163	44	10
Co/CoO/BC	GCE	1M KOH	-210	93.3	11
Co ₃ O ₄	NF	1M KOH	-225	53	12
NiO/Co ₃ O ₄	GCE	1M KOH	> -600	61	13
Co ₃ O ₄ /MoS ₂	NF	1M KOH	-205	128	14
C-Co ₃ O ₄	TM ^d	1M KOH	-163	89	15
Co ₃ O ₄ /Co ₄ N	CC	1M KOH	-90	58	16

CC^a: Carbon cloth; GCE^b: Glassy carbon electrode; NF^c: Ni foam; TM^d: Ti mesh

Reference

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