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## Supporting Information

## Effects of different exchanging ions on the band structure and photocatalytic activity of defect pyrochlore oxide: A case study in KNbTeO<sub>6</sub>

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Fig. S1 The spectrum of halogen lamp.



**Fig. S2.** Schematic structure of a) pristine  $\text{KNbTeO}_6$  b) Ag doped- $\text{KNbTeO}_6$  c) Cu doped-KNbTeO<sub>6</sub> d) Sn doped- $\text{KNbTeO}_6$ . Color legend: K in purple, Nb in yellow, Te in



Fig. S3 Rietveld refinement of KNTO, AgKNTO, CuKNTO, and SnKNTO before heating at 300 °C for 5h

Table S1 Refined parameters and reliability factor of KNTO, AgKNTO, CuKNTO, and SnKNTO after heating at 300 °C for 5h. according to synchrotron XRD patterns (using 12 keV monochromatic X-ray).

Atom	Site	Х	У	Z	Occupancy	Beq
KNTO; a=	10.2368(1)	Å, Rexp : 5.70	Rwp : 11.96	GOF : 2.10		
Κ	32e	0.1042(5)	0.1042(5)	0.1042(5)	0.25	4.3(2)
Nb	16c	0.5000	0.5000	0.5000	0.5	2.1(1)
Te	16c	0.5000	0.5000	0.5000	0.5	2.1(1)
0	48f	0.4326(2)	0.1250	0.1250	1	2.6(1)
AgKNTO; a=10.2526(1)Å, Rexp : 5.39 Rwp : 10.36 GOF : 1.92						
Κ	32e	0.1008(2)	0.1008(2)	0.1008(2)	0.2	2.4(1)
Ag	16d	0.0000	0.0000	0.0000	0.1	2.4(1)
Nb	16c	0.5000	0.5000	0.5000	0.5	1.1(1)
Te	16c	0.5000	0.5000	0.5000	0.5	1.1(1)
0	48f	0.4259(2)	0.1250	0.1250	1	1.5(1)
CuKNTO; a= 10.2484(1) Å, Rexp : 5.90 Rwp : 12.63 GOF : 2.14						
Κ	32e	0.1010(4)	0.1010(4)	0.1010(4)	0.2	4.9(2)
Cu	32e	0.1010(4)	0.1010(4)	0.1010(4)	0.025	4.9(2)
Nb	16c	0.5000	0.5000	0.5000	0.5	2.8(2)
Te	16c	0.5000	0.5000	0.5000	0.5	2.8(2)
0	48f	0.4311(2)	0.1250	0.1250	1	3.4(8)
SnKNTO; a= 10.2673(1) Å, Rexp : 5.34 Rwp : 10.48 GOF : 1.96						
Κ	32e	0.0897(1)	0.0897(1)	0.0897(1)	0.15	7.2(2)
Sn	32e	0.0897(1)	0.0897(1)	0.0897(1)	0.05	7.2(2)
Nb	16c	0.5000	0.5000	0.5000	0.5	0.5(1)
Te	16c	0.5000	0.5000	0.5000	0.5	0.5(1)
0	48f	0.4292(2)	0.1250	0.1250	1	0.6(1)



Fig. S4 The energy dispersive X-ray spectroscopy (EDS) elemental mapping images of KNTO,

AgKNTO, CuKNTO, and SnKNTO.



Fig. S5 N<sub>2</sub> adsorption-desorption isotherms of KNTO and SnKNTO



Fig. S6 Nb 3d (a) and Te 3d (b) XPS spectra of KNTO, AgKNTO, CuKNTO, and SNKNTO.



Fig. S7 Projected density of states of Sn 5s (blue line) which mainly contribute as bonding interactions (-5.5 eV) and antibonding interactions (-0.5 eV) in the valence band. The grey area illustrates the total DOS.



Fig. S8 The UV-Vis spectra of MB solution degraded with SnKNTO at various times