

## Supporting Information

### **Sulfur Vacancies-Tailored NiCo<sub>2</sub>S<sub>4</sub> Nanosheet Arrays for Hydrogen Evolution Reaction at All pH Values**

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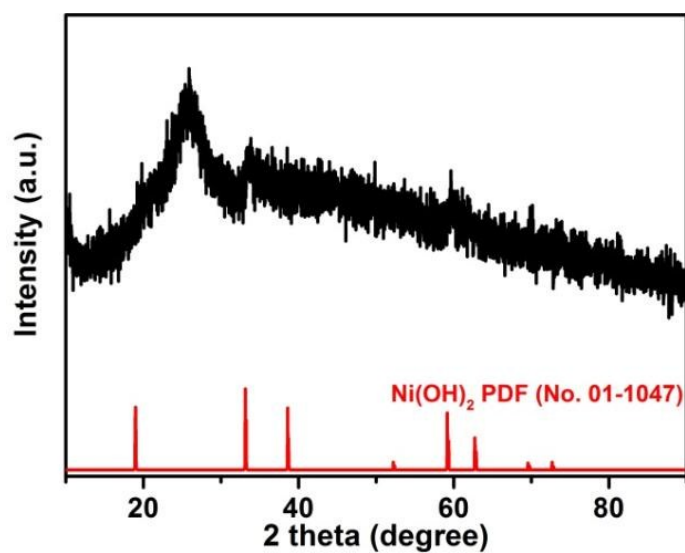
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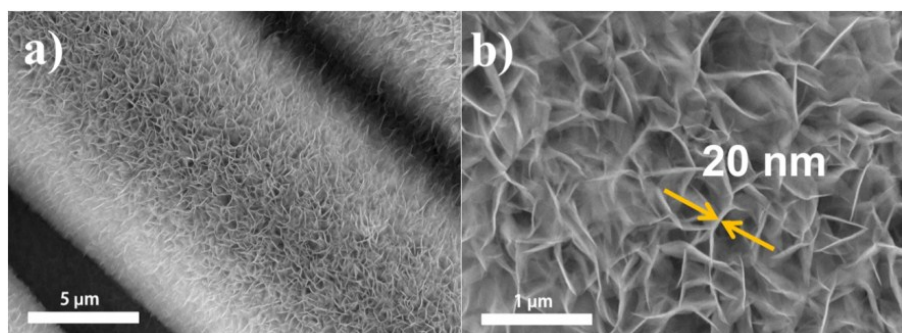
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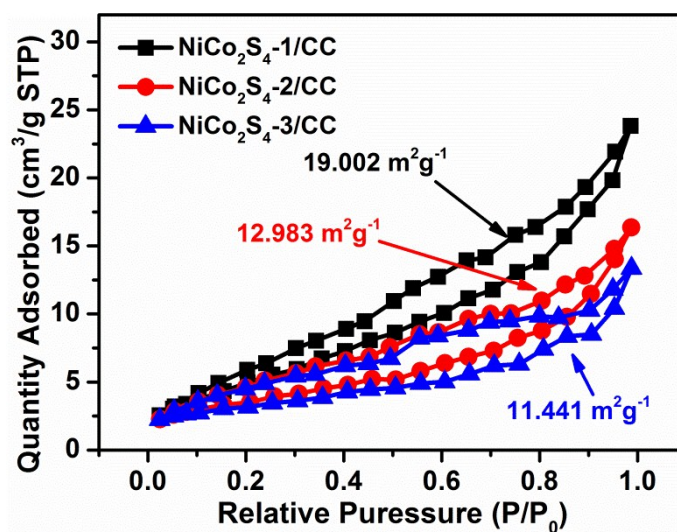
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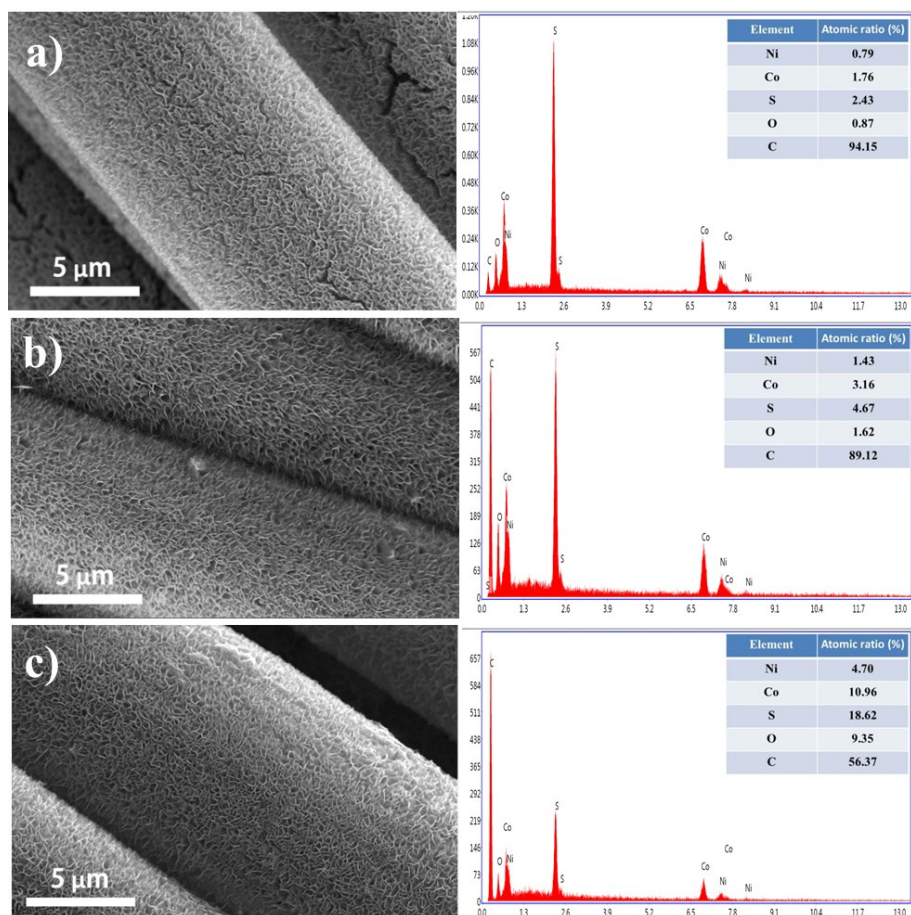
**Figure S1.** XRD pattern of NiCo hydroxide nanosheet arrays precursors on carbon cloth.



**Figure S2.** SEM of NiCo hydroxide nanosheet arrays precursors on carbon cloth.



**Figure S3.** Nitrogen adsorption-desorption isotherms curve of NiCo<sub>2</sub>S<sub>4</sub>-1/CC, NiCo<sub>2</sub>S<sub>4</sub>-2/CC and NiCo<sub>2</sub>S<sub>4</sub>-3/CC.



**Figure S4.** SEM and corresponding EDS for a) NiCo<sub>2</sub>S<sub>4</sub>-1/CC, b) NiCo<sub>2</sub>S<sub>4</sub>-2/CC and c) NiCo<sub>2</sub>S<sub>4</sub>-3/CC.

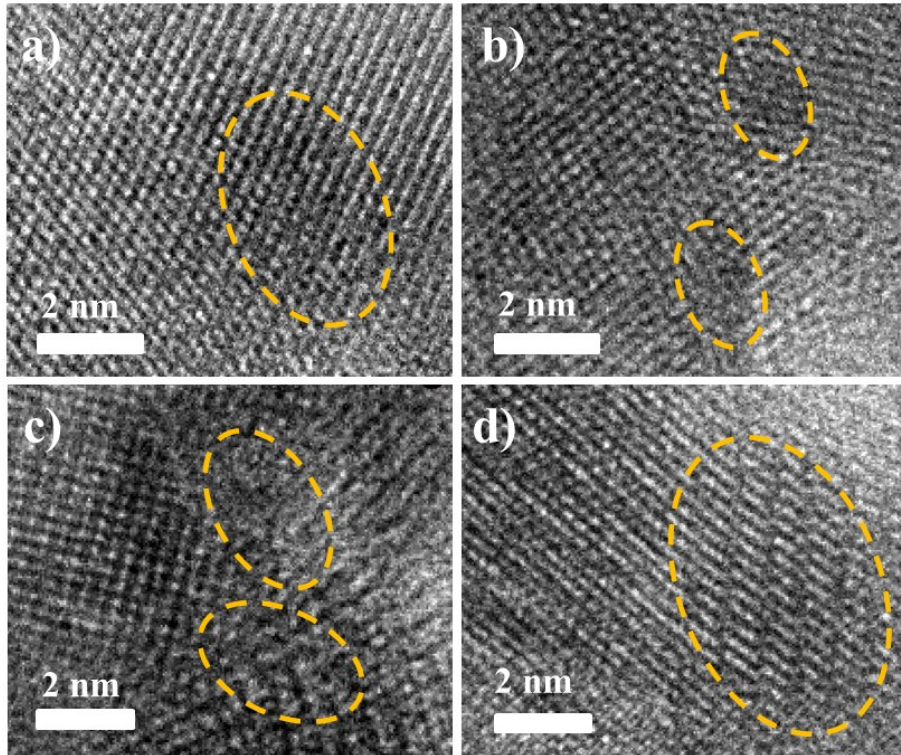


Figure S5. HETEM images of NiCo<sub>2</sub>S<sub>4</sub>-2/CC (a-d)

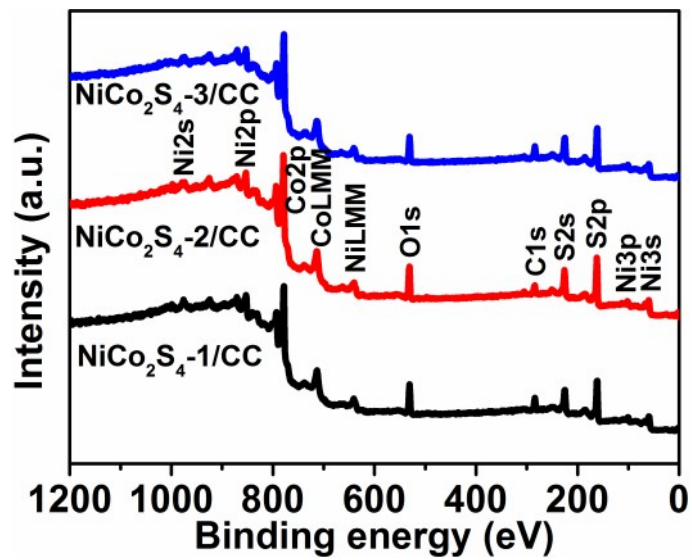
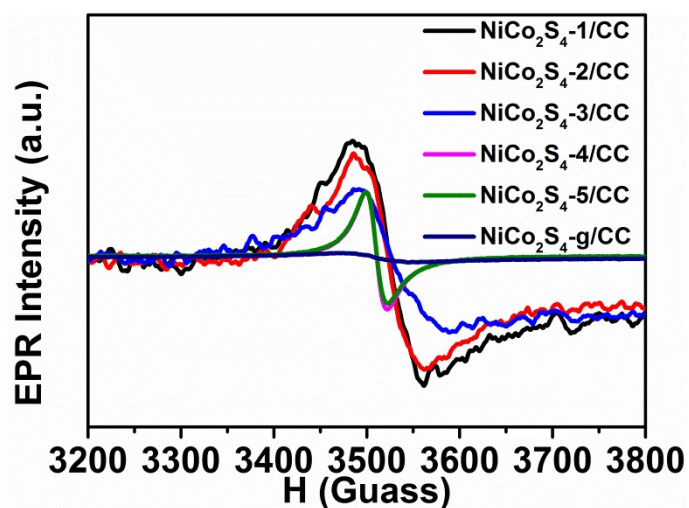
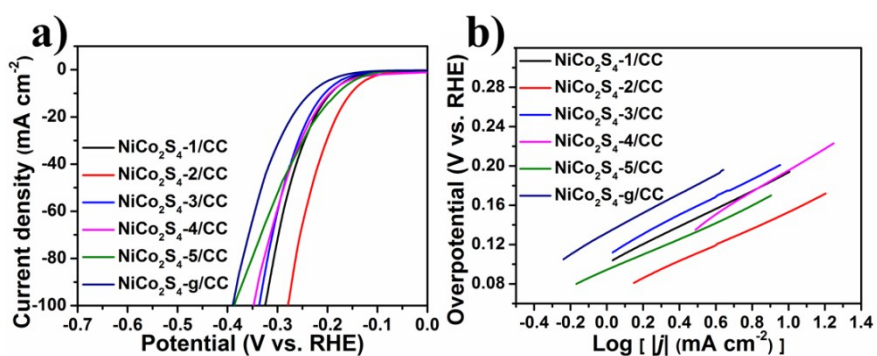


Figure S6. XPS spectra of three samples.



**Figure S7.** EPR curves of NiCo<sub>2</sub>S<sub>4</sub>-1/CC, NiCo<sub>2</sub>S<sub>4</sub>-2/CC, NiCo<sub>2</sub>S<sub>4</sub>-3/CC, NiCo<sub>2</sub>S<sub>4</sub>-4/CC, NiCo<sub>2</sub>S<sub>4</sub>-5/CC and NiCo<sub>2</sub>S<sub>4</sub>-g/CC.



**Figure S8.** a) LSV and Tafel curves b) of NiCo<sub>2</sub>S<sub>4</sub>-1/CC, NiCo<sub>2</sub>S<sub>4</sub>-2/CC, NiCo<sub>2</sub>S<sub>4</sub>-3/CC, NiCo<sub>2</sub>S<sub>4</sub>-4/CC, NiCo<sub>2</sub>S<sub>4</sub>-5/CC and NiCo<sub>2</sub>S<sub>4</sub>-g/CC.



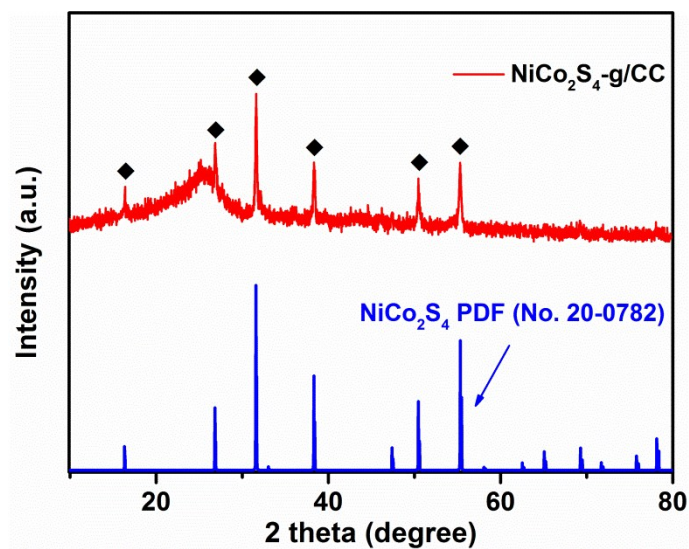


Figure S9. XRD pattern of NiCo<sub>2</sub>S<sub>4</sub>-g/CC.

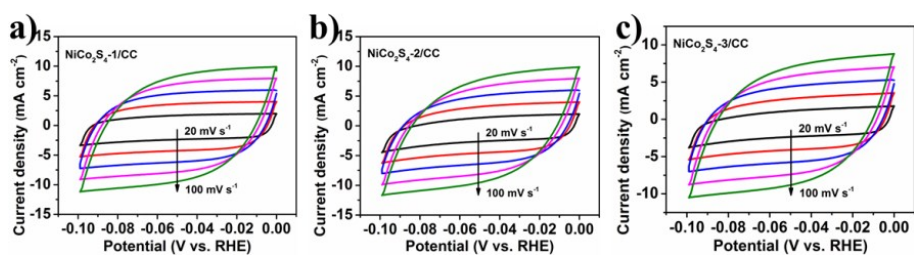


Figure S10. Voltammograms of a) NiCo<sub>2</sub>S<sub>4</sub>-1/CC, b) NiCo<sub>2</sub>S<sub>4</sub>-2/CC and c) NiCo<sub>2</sub>S<sub>4</sub>-3/CC recording at the scan rates of 20, 40, 60, 80, and 100 mV s<sup>-1</sup> in 1M KOH.

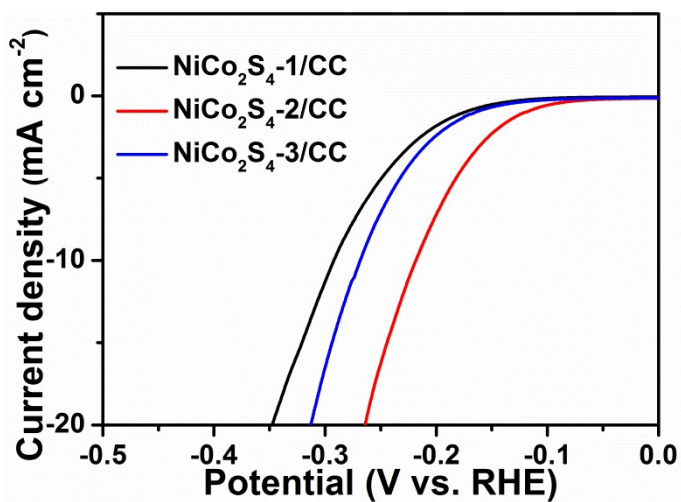
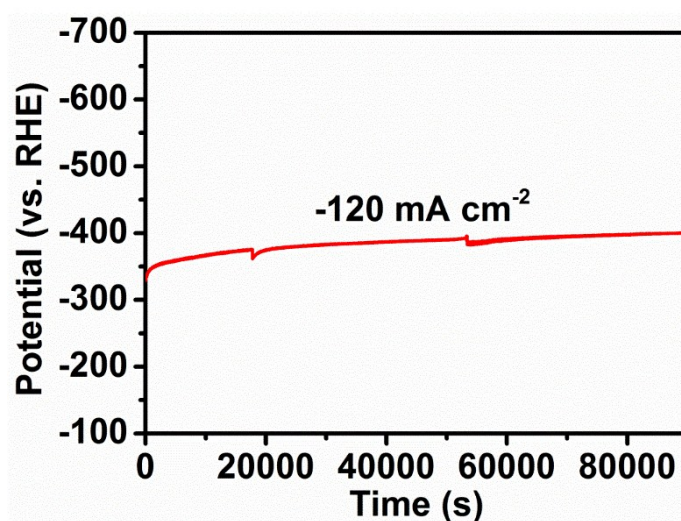
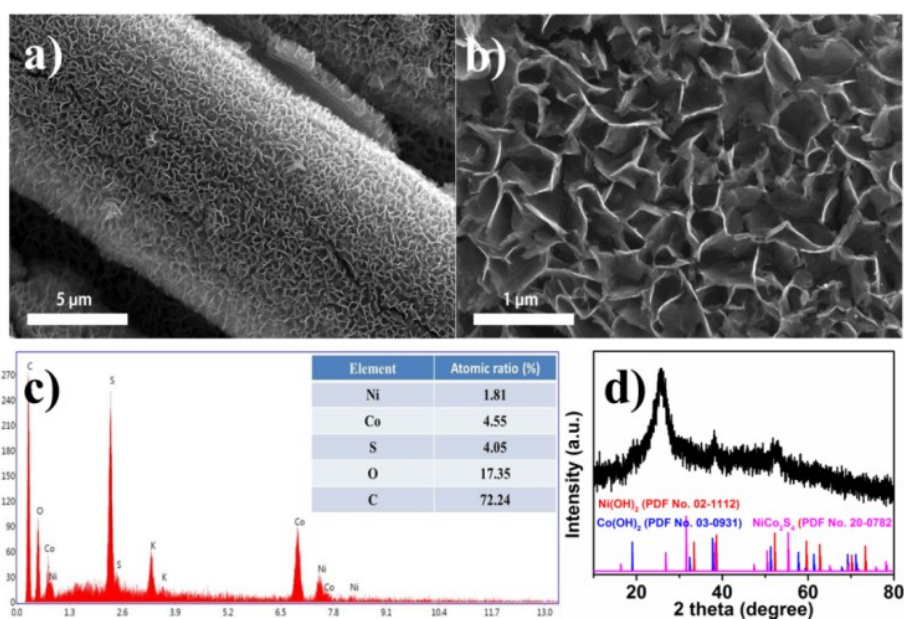


Figure S11. LSV curves of NiCo<sub>2</sub>S<sub>4</sub>-1/CC, NiCo<sub>2</sub>S<sub>4</sub>-1/CC and NiCo<sub>2</sub>S<sub>4</sub>-3/CC normalized by special surface area.



**Figure S12.** Durability measurement of NiCo<sub>2</sub>S<sub>4</sub>-2/CC with a continuous current density of -120 mA cm<sup>-2</sup>.



**Figure S13.** a-b) SEM images of NiCo<sub>2</sub>S<sub>4</sub>-2/CC electrode after 36000 s durability measurement. c) Corresponding EDS spectrum. d) XRD pattern of NiCo<sub>2</sub>S<sub>4</sub>-2/CC electrocatalyst after 36000 s stable test.

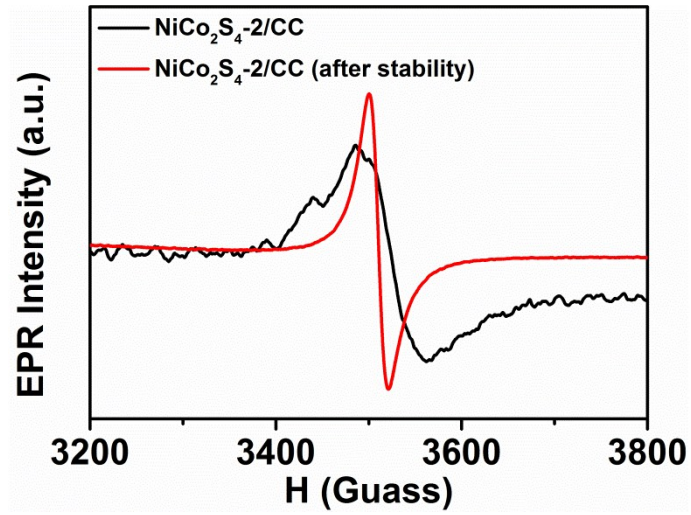


Figure S14. EPR curve after 10 h of continuous test.

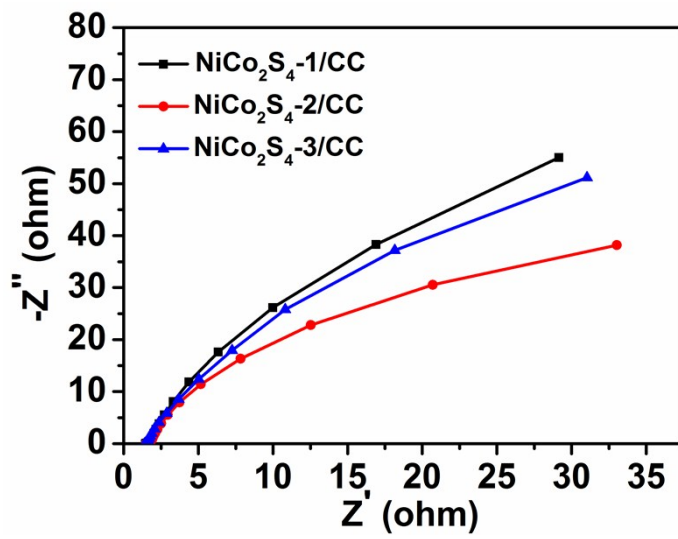
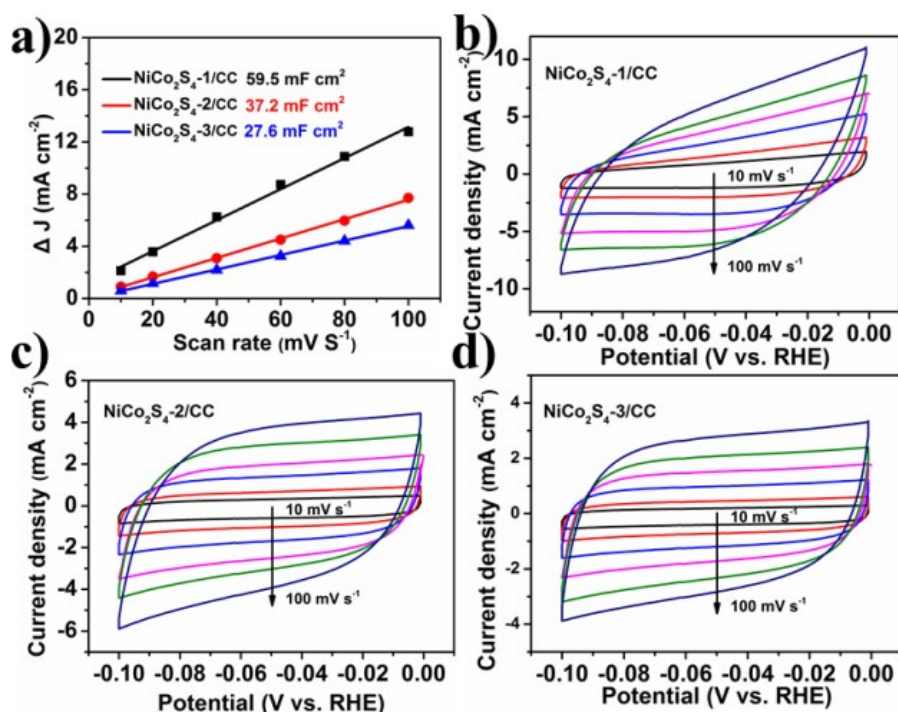
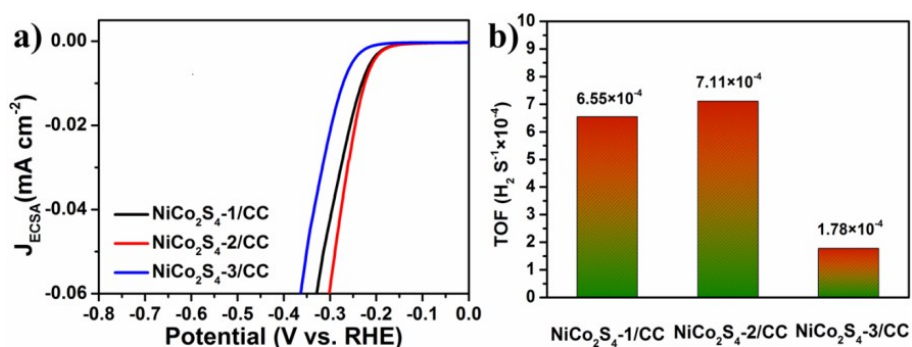


Figure S15. Nyquist plots of NiCo<sub>2</sub>S<sub>4</sub>-1/CC, NiCo<sub>2</sub>S<sub>4</sub>-2/CC, and NiCo<sub>2</sub>S<sub>4</sub>-3/CC in 0.5 M H<sub>2</sub>SO<sub>4</sub>.

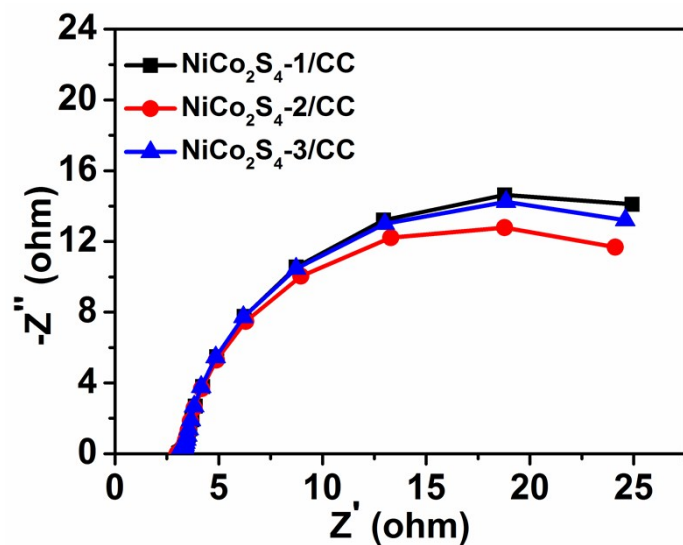




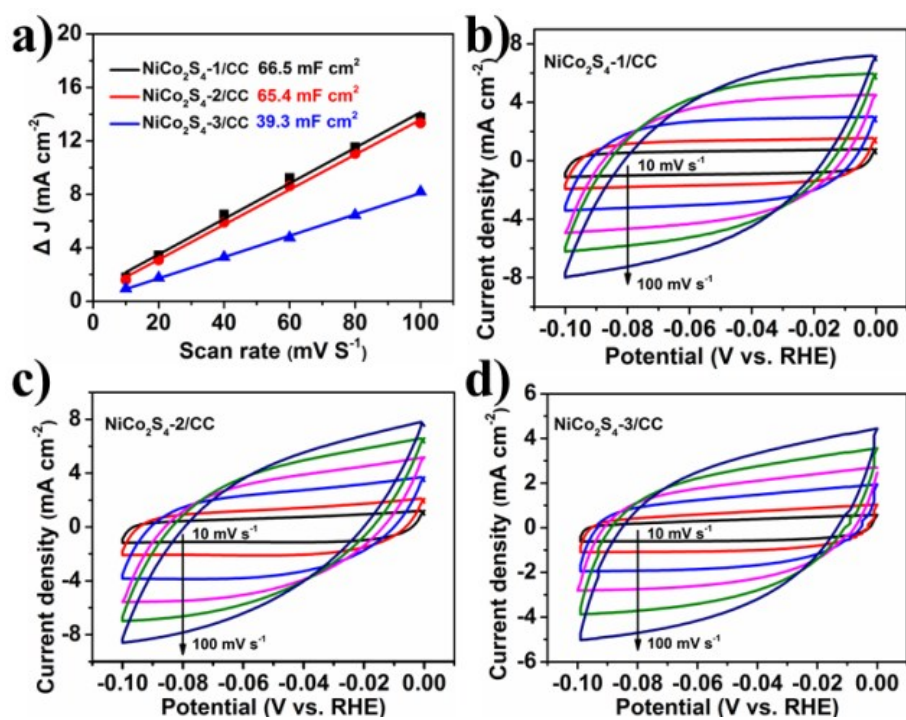
**Figure S16.**  $C_{dl}$  (a) and different cyclic voltammograms of NiCo<sub>2</sub>S<sub>4</sub>-1/CC (b), NiCo<sub>2</sub>S<sub>4</sub>-2/CC (c) and NiCo<sub>2</sub>S<sub>4</sub>-3/CC (d) in 0.5 M H<sub>2</sub>SO<sub>4</sub>



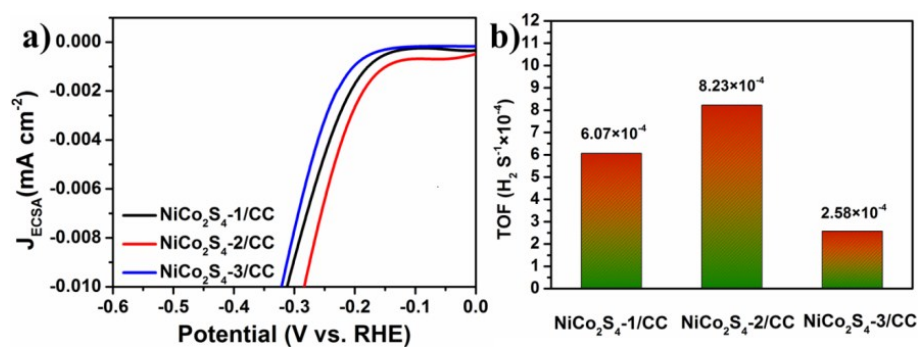
**Figure S17.** a) Polarization curves of as-prepared samples normalized by ECSA in 0.5 M H<sub>2</sub>SO<sub>4</sub>, b) Calculated TOF value at overpotential of -200 mV in 0.5 M H<sub>2</sub>SO<sub>4</sub>.



**Figure S18.** Nyquist plots of NiCo<sub>2</sub>S<sub>4</sub>-1/CC, NiCo<sub>2</sub>S<sub>4</sub>-2/CC, and NiCo<sub>2</sub>S<sub>4</sub>-3/CC in 1 M PBS.



**Figure S19.**  $C_{dl}$  (a) and different cyclic voltammograms of NiCo<sub>2</sub>S<sub>4</sub>-1/CC (b), NiCo<sub>2</sub>S<sub>4</sub>-2/CC (c) and NiCo<sub>2</sub>S<sub>4</sub>-3/CC (d) in 1 M PBS.



**Figure S20.** a) Polarization curves of as-prepared samples normalized by ECSA in 0.5 M H<sub>2</sub>SO<sub>4</sub>, b) Calculated TOF value at overpotential of -200 mV in 1 M PBS.

**Table S1.** EDS and ICP-OES results of Ni, Co and S in NiCo<sub>2</sub>S<sub>4</sub>.

	EDS (Ni: Co: S) molar ratio	ICP (Ni: Co: S) molar ratio	NiCo <sub>2</sub> S <sub>4</sub> (Ni/Co: S by ICP) molar ratio
NiCo <sub>2</sub> S <sub>4</sub> -1/CC	0.79:1.76:2.43	0.82:1.72:2.46	1:2.09:3.0
NiCo <sub>2</sub> S <sub>4</sub> -2/CC	1.43:3.16:4.67	0.61:1.40:2.13	1:2.29:3.5
NiCo <sub>2</sub> S <sub>4</sub> -3/CC	4.70:10.96:18.62	1.03:2.11:3.82	1:2.05:3.71

**Table S2.** Comparison of HERR performances of NiCo<sub>2</sub>S<sub>4</sub>-2/CC with previously reported non-precious metal HER electrocatalysts.

Catalyst	Substrate	Electrolyte	J (mA cm <sup>-2</sup> )	$\eta$ (mV vs RHE)	Ref.
NiCo <sub>2</sub> S <sub>4</sub> /CC	CC <sup>a</sup>	1M KOH	10	150	This work
NiCo <sub>2</sub> S <sub>4</sub> /NF	NF <sup>a</sup>	1M KOH	10	210	1
NiCo <sub>2</sub> S <sub>4</sub> @NiCo <sub>2</sub> O <sub>4</sub> /NF	NF <sup>c</sup>	1M KOH	10	190	2
NiCo <sub>2</sub> S <sub>4</sub>	GCE <sup>d</sup>	1M KOH	10	148	3
NiCo <sub>2</sub> S <sub>4</sub> /NF	NF	1M KOH	10	169	4
NiCo <sub>2</sub> S <sub>4</sub> /NF	NF	1M KOH	10	191	5
NiCo <sub>2</sub> S <sub>4</sub> @Pd	GCE	1M KOH	10	87	6
N-NiCo <sub>2</sub> S <sub>4</sub>	NF	1M KOH	10	41	7
Ni-Co-S-P/Graphene	GCE	1M KOH	10	117	8
Co <sub>9</sub> S <sub>8</sub> -Ni <sub>x</sub> S <sub>y</sub> /NF	NF	1M KOH	10	163	9
MoS <sub>2</sub> -Ni <sub>x</sub> S <sub>y</sub> /NF	CFP <sup>e</sup>	1M KOH	10	139	10
Co <sub>9</sub> S <sub>8</sub> -NiCo <sub>2</sub> S <sub>4</sub>	GCE	1M KOH	10	172	11
NiCoS@C-dot	GCE	1M KOH	10	232	12
CoS@Ni <sub>3</sub> S <sub>2</sub>	NF	1M KOH	10	204	13
Ni <sub>2</sub> P/Ni <sub>3</sub> S <sub>2</sub>	NF	1M KOH	10	130	14
MoS <sub>2</sub>	RDE <sup>f</sup>	1M KOH	10	154	15
CoS <sub>2</sub>	GCE	1M KOH	10	193	16
N-Ni <sub>3</sub> S <sub>2</sub> /NF	NF	1M KOH	10	155	17
Se-(NiCo)S <sub>x</sub> /(OH) <sub>x</sub>	NF	1M KOH	10	103	18
CoP	GCE	1M KOH	10	154	19
rGO@NiMnCo	GCE	1M KOH	10	151	20

CC<sup>a</sup>: Carbon Cloth

NF<sup>b</sup>: Ni Foam

GCE<sup>c</sup>: Glassy Carbon Electrode

CFP<sup>d</sup>: Carbon Fiber Paper

RDE<sup>f</sup> : Rotating Disk Electrode



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