

Supporting Information for

Porous Lantern-Like MFI Zeolite Composed of 2D Nanosheets for Highly Efficient Visible-Light-Driven Photocatalysis

Hui Liu ^a, Dang-guo Cheng ^{a,b,*}, Fengqiu Chen ^{a,b}, Xiaoli Zhan ^{a,b}

^a Zhejiang Provincial Key Laboratory of Advanced Chemical Engineering Manufacture Technology, College of Chemical and Biological Engineering, Zhejiang University, 38 Zheda Road, Hangzhou 310027, China

^b Institute of Zhejiang University-Quzhou, 78 Jiu Hua Boulevard North, Quzhou 324000, China

Corresponding Author (D. -G. Cheng)

Email: dgcheng@zju.edu.cn, Tel.: +86 571 87953382; Fax: +86-571-87951227.

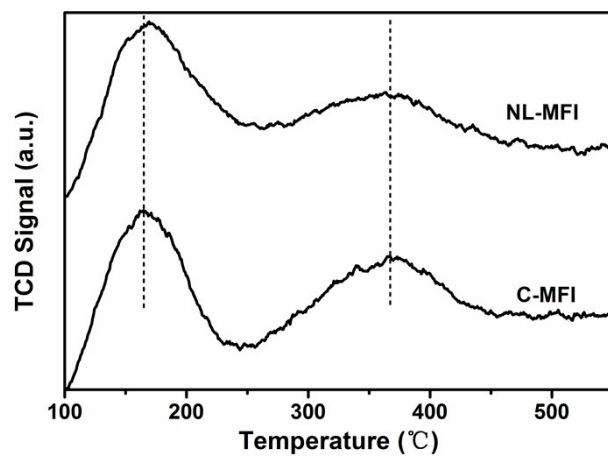


Figure S1. NH₃-TPD curves of C-MFI and NL-MFI.

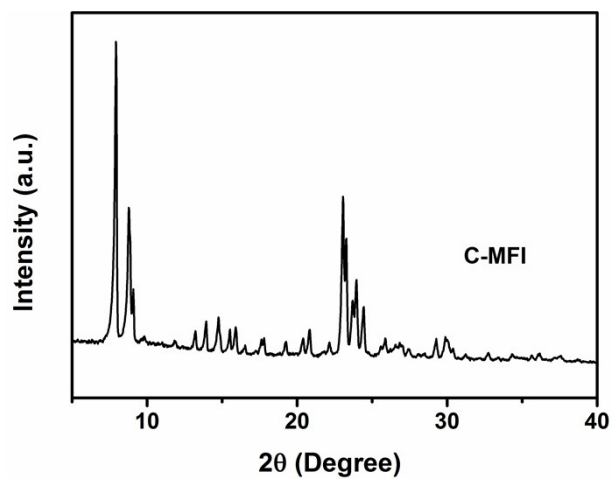


Figure S2. XRD pattern of C-MFI.

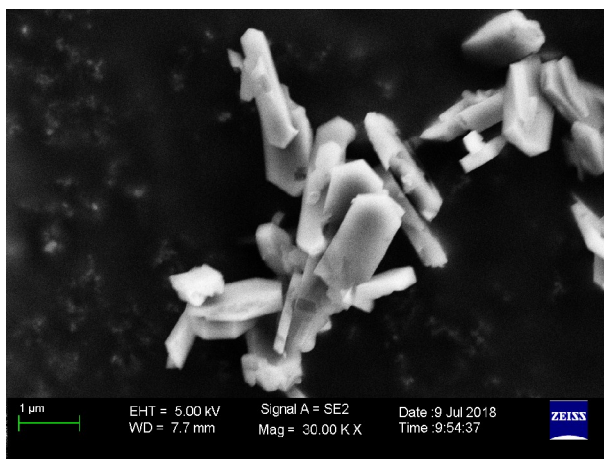


Figure S3. SEM image of C-MFI.

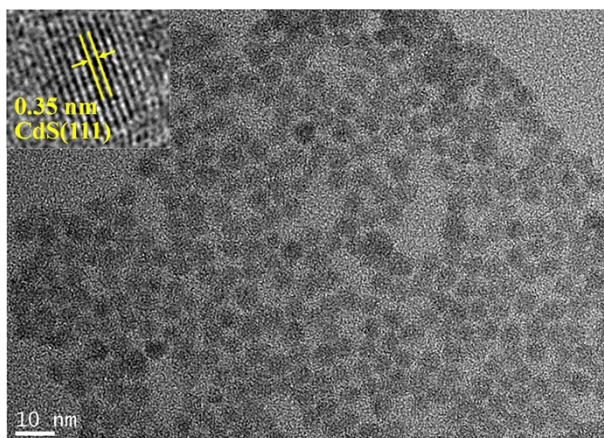


Figure S4. TEM image of the CdS nanoparticles.

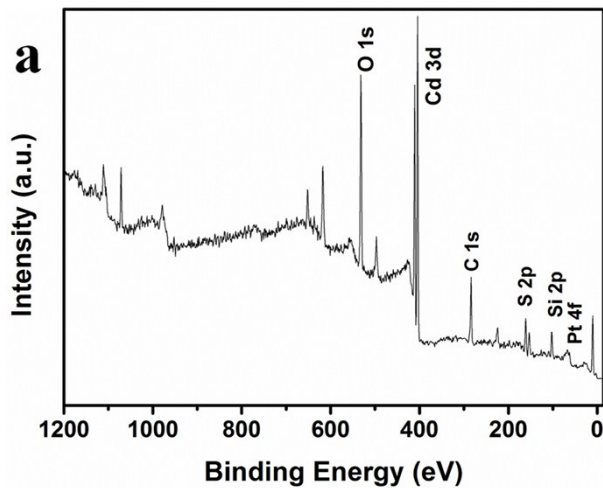


Figure S5. Full XPS spectrum of CdS/Pt/NL-MFI.

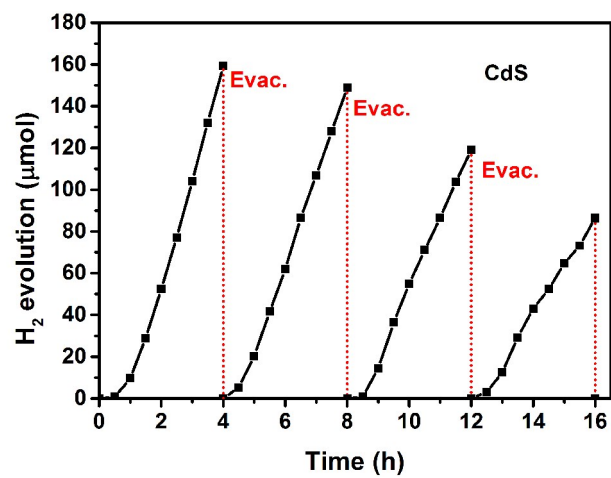


Figure S6. Photocatalytic stability of the pure CdS nanoparticles.

Table S1. ICP and NH₃-TPD results of C-MFI and NL-MFI.

Sample	Si/Al	Weak acid		Strong acid		Total acid amount (mmol·g ⁻¹)
		Temperature (°C)	Amount (mmol·g ⁻¹)	Temperature (°C)	Amount (mmol·g ⁻¹)	
C-MFI	50	166	0.16	367	0.13	0.29
NL-MFI	49	169	0.18	364	0.12	0.30

Table S2. Comparison of the photocatalytic H₂ evolution efficiency of CdS/Pt/NL-MFI with those of the CdS-based and zeolite-based photocatalysts reported in the literature.

Photocatalyst	photocatalyst amount (mg); Reaction solution (mL)	Sacrificial reagent	Light source	H ₂ evolution rate (μmol/h)	QE (%); wavelength (nm)	Ref.
CdS/Pt/NL-MFI	60; 100	20 vol% Lactic acid	300 W Xe, λ>420 nm	2152.7	39.4% 450 nm	This work
CdS/Pt/Ga ₂ O ₃ -U	100; 100	10 vol% Lactic acid	300 W Xe, λ>420 nm	995.8	43.6% 460 nm	S1
CdS/Pt/In ₂ O ₃ -U				1032.2	45.3% 460 nm	
N-MoS ₂ /CdS	50; 100	10 vol% Lactic acid	300 W Xe, λ>420 nm	~456		S2
Pd-CdS/ZSM-5	600; 600	10 vol% ethanol	7 W UVC, λ~254 nm	~668.8		S3
MoS ₂ /CdS-TiO ₂	20; 80	10 vol% Lactic acid	300 W Xe, λ>420 nm	280	19.3% 420 nm	S4
MoS ₂ /CdS-P25				94.2		
Pt/CdS-TiO ₂				70		
Pt -CdS/ZnO	10; 25	0.25 M Na ₂ S, 0.35 M Na ₂ SO ₄	225 W Xe arc, λ: 420~720 nm	221.2		S5
CdS/WS ₂ /graphene	8;8	0.35 M Na ₂ S, 0.25 M Na ₂ SO ₃	500 W Xe arc, λ>420 nm	~14.7	21.2% 420 nm	S6
Pt/Zn _{0.5} Cd _{0.5} S	30; 120	0.1 M Na ₂ S, 0.1 M Na ₂ SO ₃	500 W Xe arc, λ>400 nm	114.3	7.15%	S7
		0.15 M ascorbic acid		164.9	8.56%	
Pd@CdS/PdS	10; 50	0.1 M Na ₂ S, 0.1 M Na ₂ SO ₃	300 W Xe, λ>400 nm	892		S8
		0.5 M Na ₂ S, 0.5 M Na ₂ SO ₃		1448		
CdS/Pt/graphene	20; 80	10 vol% Lactic acid	350 W Xe arc, λ>420 nm	~22.4	22.5% 420 nm	S9

CdS/Au/MoS ₂	10; 50	10 vol% Lactic acid	Xe arc, $\lambda > 420$ nm	~70.1	27.9%	S10
Pd-Ti-MCM-48	2; 2	20 vol% methanol	300 W Xe, $\lambda > 280$ nm	560		S11
Pt-TiO ₂ /ZSM-5	100; 200	10 vol% methanol	300 W Xe	1000	12.6% 365 nm	S12
Pt-TiO ₂ /SBA-15				880	5.4% 365 nm	
TiO ₂ /Na-Y	100; 1000	5 vol% ethanol	100 W high pressure mercury lamp	25.1		S13
Pd-CdPdS-NaY	50; 25	0.4 M Na ₂ S, 0.3 M Na ₂ SO ₃	288 W day light fluorescent lamp	~116		S14

References

- S1. Y. X. Pan, H. Zhuang, J. Hong, Z. Fang, H. Liu, B. Liu, Y. Huang and R. Xu, *ChemSusChem*, 2014, **7**, 2537-2544.
- S2. Q. Liu, X. Li, Q. He, A. Khalil, D. Liu, T. Xiang, X. Wu and L. Song, *Small*, 2015, **11**, 5556-5564.
- S3. H. Enzweiler, P. H. Yassue-Cordeiro, M. Schwaab, E. Barbosa-Coutinho, M. H. N. Olsen Scaliante and N. R. C. Fernandes, *Int. J. Hydrogen Energy*, 2018, **43**, 6515-6525.
- S4. N. Qin, J. Xiong, R. Liang, Y. Liu, S. Zhang, Y. Li, Z. Li and L. Wu, *Appl. Catal. B: Environ.*, 2017, **202**, 374-380.
- S5. D. Ma, J. W. Shi, Y. Zou, Z. Fan, X. Ji and C. Niu, *ACS Appl. Mater. Inter.*, 2017, **9**, 25377-25386.
- S6. Q. Xiang, F. Cheng and D. Lang, *ChemSusChem*, 2016, **9**, 996-1002.
- S7. B.-J. Ng, L. K. Putri, X. Y. Kong, K. P. Y. Shak, P. Pasbakhsh, S.-P. Chai and A. R. Mohamed, *Appl. Catal. B: Environ.*, 2018, **224**, 360-367.
- S8. R. Marschall and L. Wang, *Catal. Today*, 2014, **225**, 111-135.
- S9. Q. Li, B. Guo, J. Yu, J. Ran, B. Zhang, H. Yan and J. R. Gong, *J. Am. Chem. Soc.*, 2011, **133**, 10878-10884.
- S10. R. K. Chava, J. Y. Do and M. Kang, *ACS Sustain. Chem. Eng.*, 2018, **6**, 6445-6457.
- S11. R. Peng, J. Baltrusaitis, C.-M. Wu and R. T. Koodali, *Int. J. Hydrogen Energy*, 2015, **40**, 905-918.
- S12. C. Jiang, K. Y. Lee, C. M. A. Parlett, M. K. Bayazit, C. C. Lau, Q. Ruan, S. J. A. Moniz, A. F. Lee and J. Tang, *Appl. Catal. A: Gen.*, 2016, **521**, 133-139.
- S13. A. Taheri Najafabadi and F. Taghipour, *Energy Convers. Manage.*, 2014, **82**, 106-113.
- S14. R. Sasikala, A. P. Gaikwad, V. Sudarsan, R. Rao, J. Jagannath, B. Viswanadh and S. R. Bharadwaj, *PCCP*, 2015, **17**, 6896-6904.