

Supplementary Information

Ultrafine AuPd Nanoparticles Supported on Amine Functionalized Monochlorotriazinyl β -Cyclodextrin as Highly Active Catalysts for Hydrogen Evolution from Formic Acid Dehydrogenation

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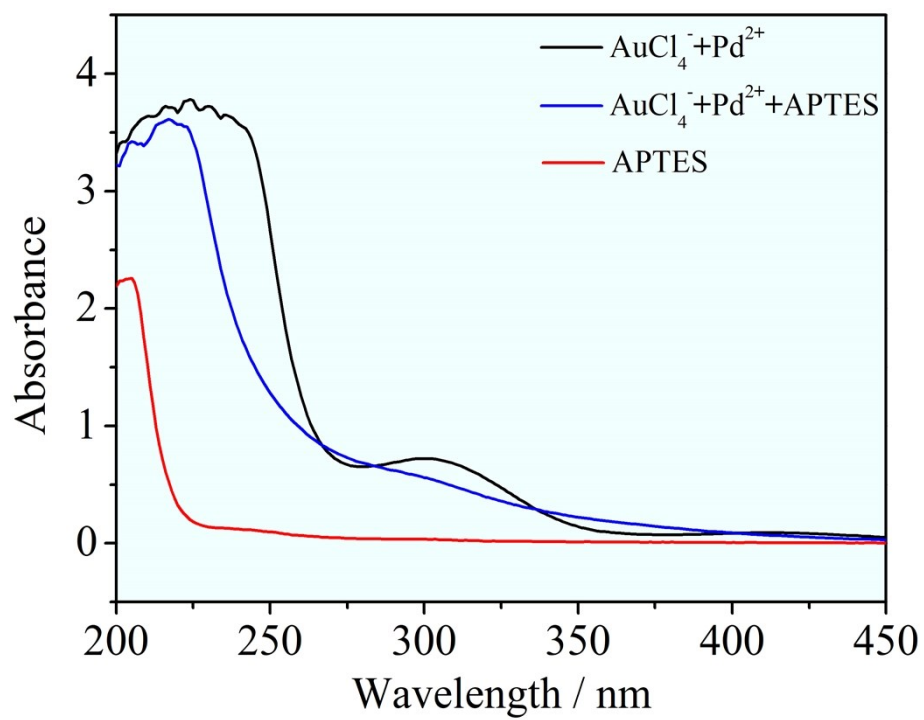


Fig. S1 UV-vis spectra of aqueous solutions containing various species.

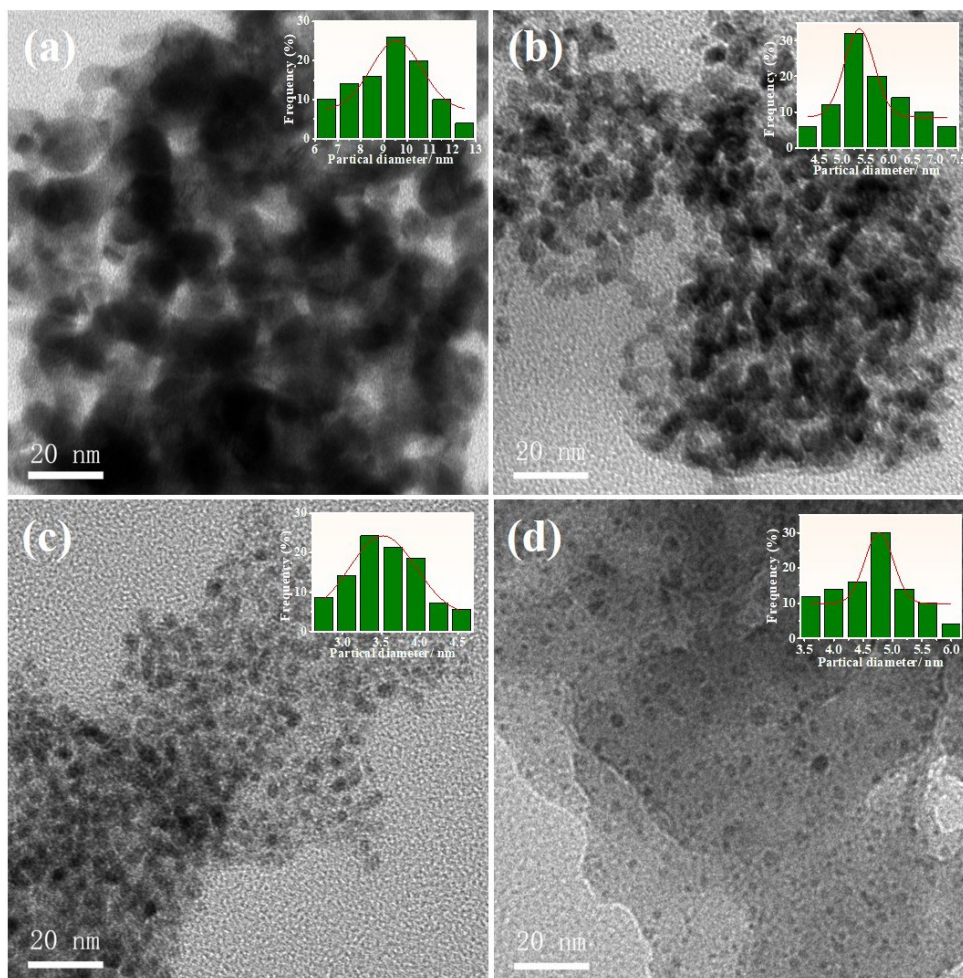


Fig. S2 TEM images and corresponding particle size distribution of (a) Au_{0.3}Pd_{0.7}, (b) Au_{0.3}Pd_{0.7}/M-β-CD, (c) Au_{0.3}Pd_{0.7}-A, (d) Au_{0.3}Pd_{0.7}/C.

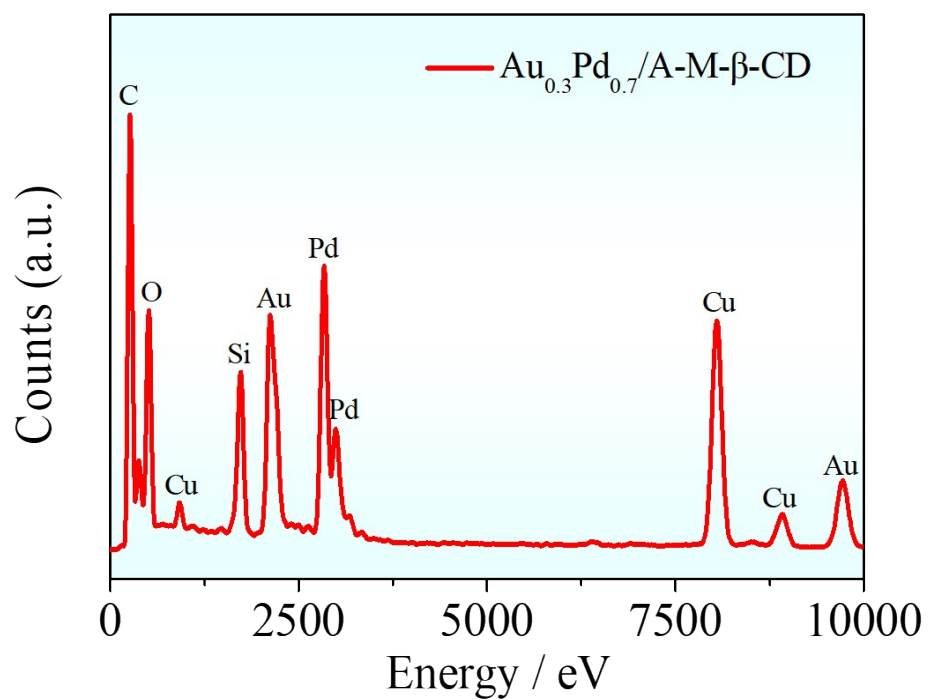


Fig. S3 EDX spectrum of Au_{0.3}Pd_{0.7}/A-M-β-CD.

Table S1. The metal loading amounts and the molar ratios of Au/Pd of Au_{0.3}Pd_{0.7}/A-M-β-CD before and after cyclic tests for dehydrogenation of FA reaction.

Sample	Au loading (mg)	Pd loading (mg)	Ratio of Au/Pd (mol/mol)
Au _{0.3} Pd _{0.7} /A-M-β-CD before cyclic tests	3.215	4.070	0.298:0.702
Au _{0.3} Pd _{0.7} /A-M-β-CD after cyclic tests	3.210	4.052	0.299:0.701

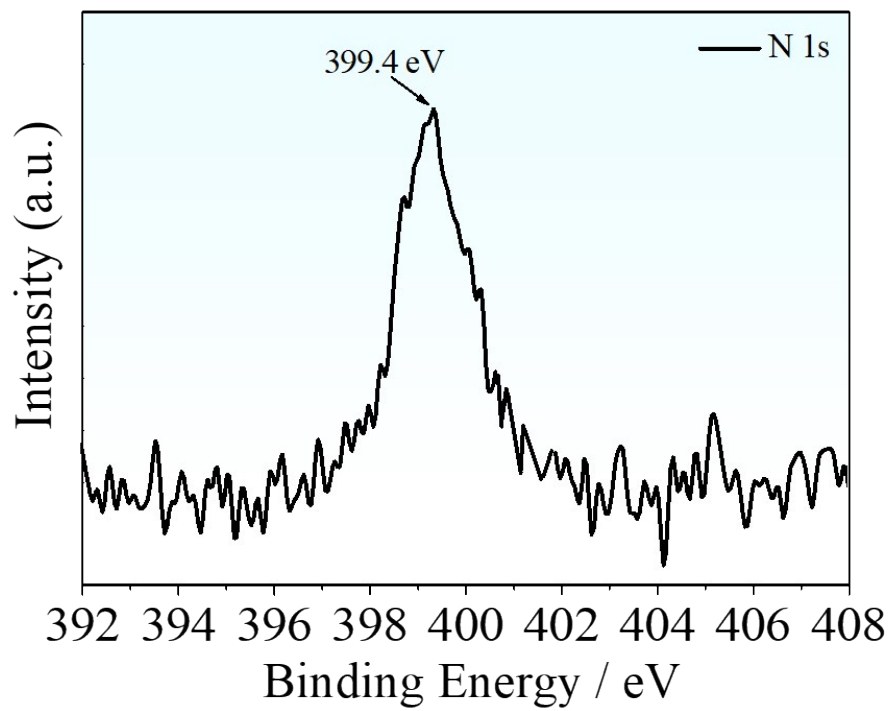


Fig. S4 High-resolution XPS spectrum of N1s for Au_{0.3}Pd_{0.7}-A.

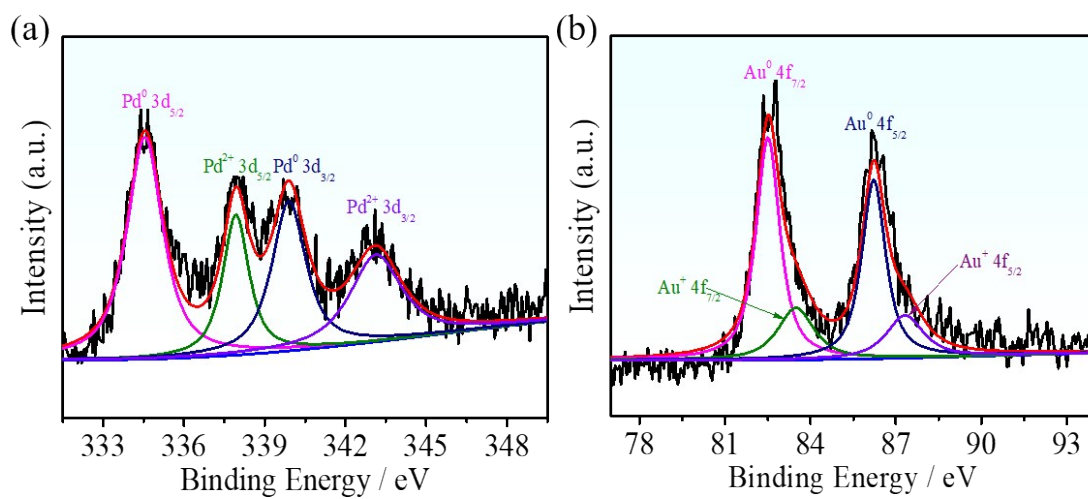


Fig. S5 High-resolution XPS spectra of (a) Pd 3d and (b) Au 4f for Au_{0.3}Pd_{0.7}/A-M-β-CD.

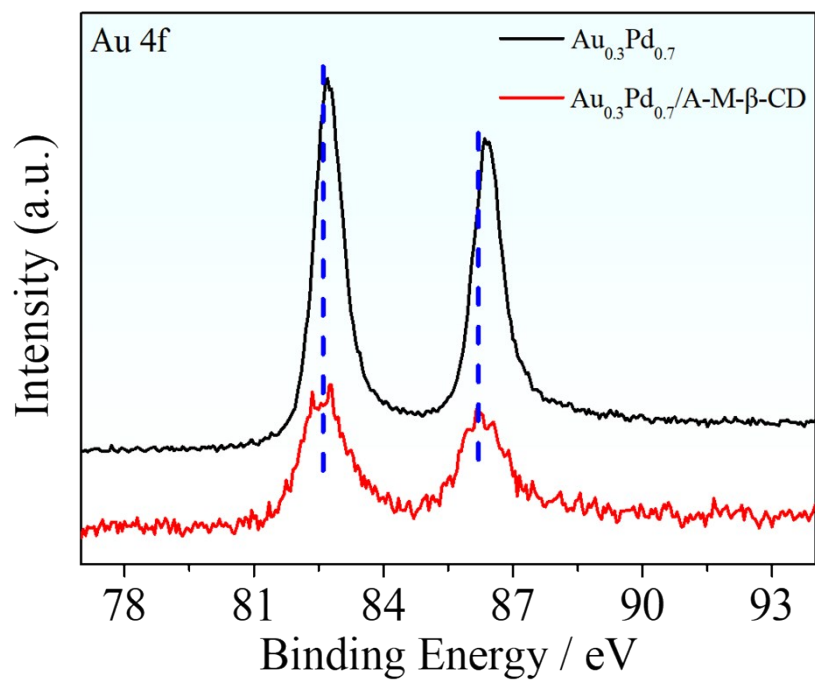


Fig. S6 High-resolution XPS spectra of Au 4f for Au_{0.3}Pd_{0.7}/A-M-β-CD and Au_{0.3}Pd_{0.7} NPs.

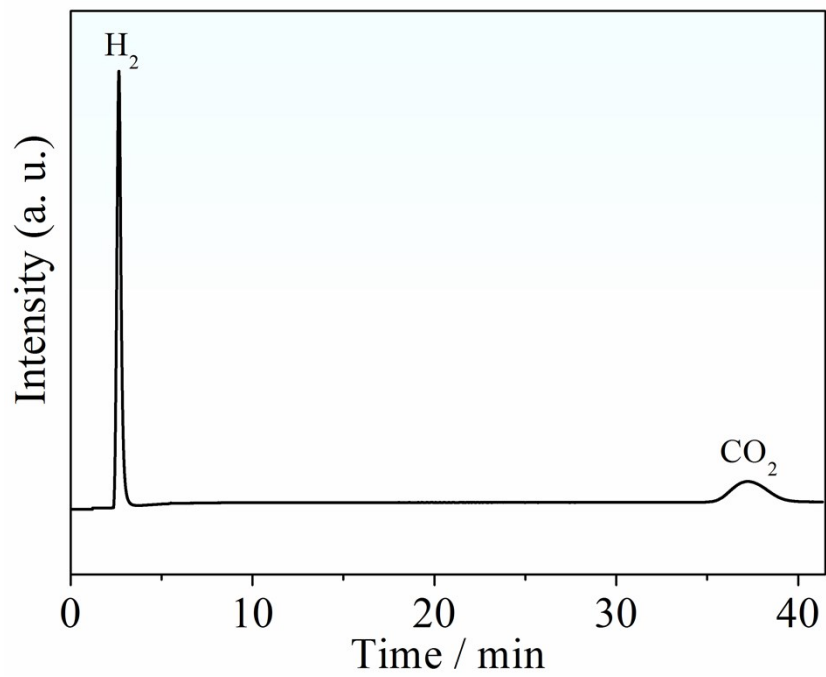


Fig. S7 GC spectrum using TCD for the evolved gas from FA aqueous solution (1.0 M, 5.0 mL) over Au_{0.3}Pd_{0.7}/A-M- β -CD at 323 K.

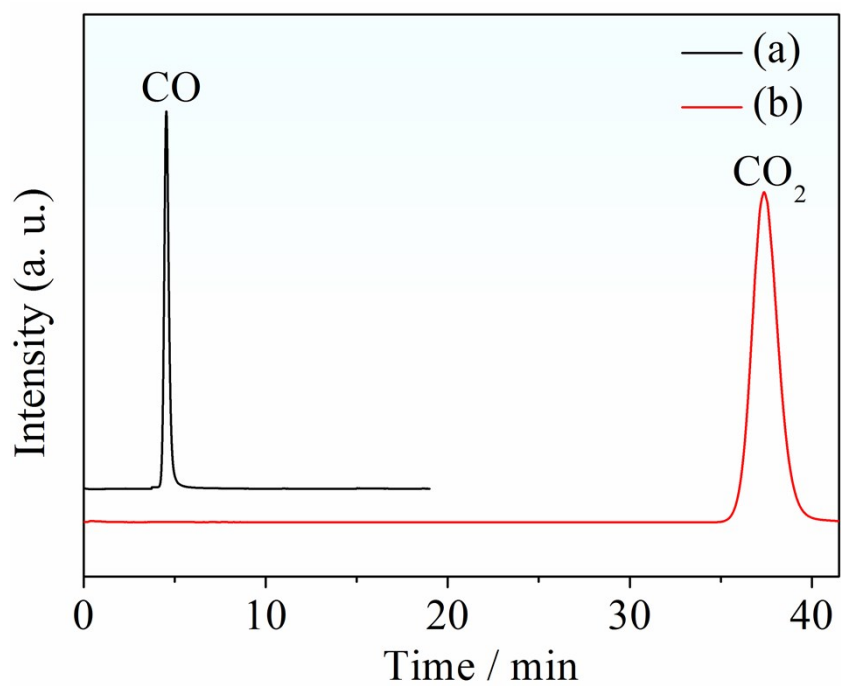


Fig. S8 GC spectrum using FID-Methanator for the (a) commercial pure CO, and (b) evolved gas from FA aqueous solution (1.0 M, 5.0 mL) over Au_{0.3}Pd_{0.7}/A-M-β-CD at 323 K.

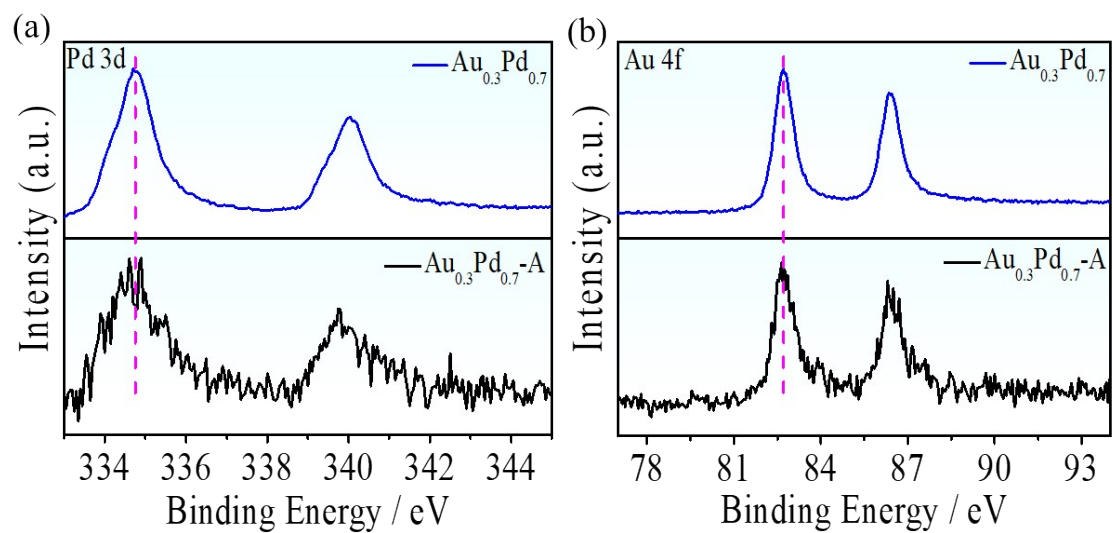


Fig. S9 High resolution XPS spectra of (a) Pd 3d and (b) Au 4f for Au_{0.3}Pd_{0.7}-A and Au_{0.3}Pd_{0.7} NPs.

Table S2. Comparisons of catalytic activities for the dehydrogenation of FA catalyzed by previously reported heterogeneous catalysts with the as-synthesized in this work.

Catalyst	Temp. (K)	$n_{\text{catalyst}}/n_{\text{FA}}$	Additive	TOF (h^{-1})	E_a (kJ/mol)	Ref.
<i>Without additive</i>						
Au@Pd/N-mrGO	298	0.0200	None	89.1 ^a	---	12
Au ₄₂ Pd ₅₈	323	0.0100	None	382 ^a	22 ± 1	46
Pd _{IMP} /CNF	303	0.0188	None	563.2 ^b	27.50	47
Pd/CN _{0.25}	298	0.0075	None	752 ^b	48.80	48
Ni _{0.4} Pd _{0.6} /NH ₂ -N-rGO	298	0.0200	None	954.3 ^a	---	41
Pd-MnO _x /SiO ₂ -NH ₂	323	0.1068	None	1300 ^a	61.9	49
Cr _{0.4} Pd _{0.6} /MIL-101-NH ₂	323	0.0200	None	2009 ^a	43.50	27
Pd/A-SEP-NH _{2(0.9)}	333	0.0315	None	5587 ^b	44.5	50
Au _{0.3} Pd _{0.7} /A-M-β-CD	323	0.0200	None	7352 ^b	39.50	This work
<i>With Additive</i>						
Pd-CNTs-in	303	0.0214	HCOONa	1135.8 ^a	36.60	51
Pd-B/C	303	0.0143	HCOONa	1184 ^b	---	7
Co ₅ Pd ₅ /CTF-600	323	0.0070	HCOOK	2129 ^a	35.94	52
(Co ₆)Ag _{0.1} Pd _{0.4} /RGO	323	0.0200	HCOONa	2739 ^b	43.10	53
Pd/S-1-in-K	323	0.0100	HCOONa	3027 ^b	39.2	54
Pd/PDA-rGO	323	0.0150	HCOONa	3810 ^b	54.30	55
(Co ₃) _E Au _{0.6} Pd _{0.4} /rGO	323	0.0200	HCOONa	4840 ^a	---	16
Au ₂ Pd ₃ @(P)N-C	303	0.0170	HCOONa	5400 ^a	---	56

a. Initial TOF values calculated based on total metal.

b. Initial TOF values calculated based on total Pd atoms.

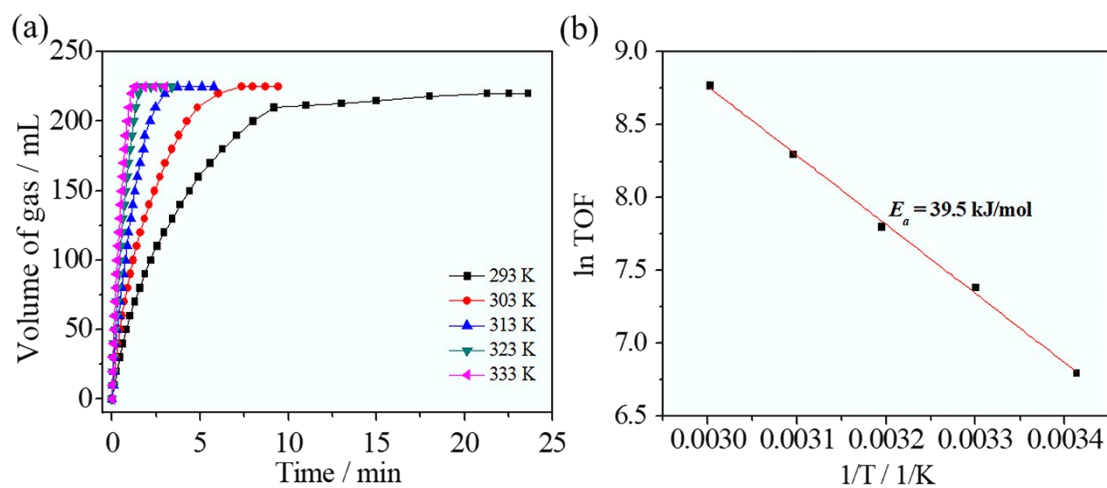


Fig. S10 (a) Plot of volume of gas over time for the dehydrogenation of FA catalyzed by Au_{0.3}Pd_{0.7}/A-M-β-CD at different temperatures; (b) Arrhenius plot of ln TOF vs. 1/T for Au_{0.3}Pd_{0.7}/A-M-β-CD.

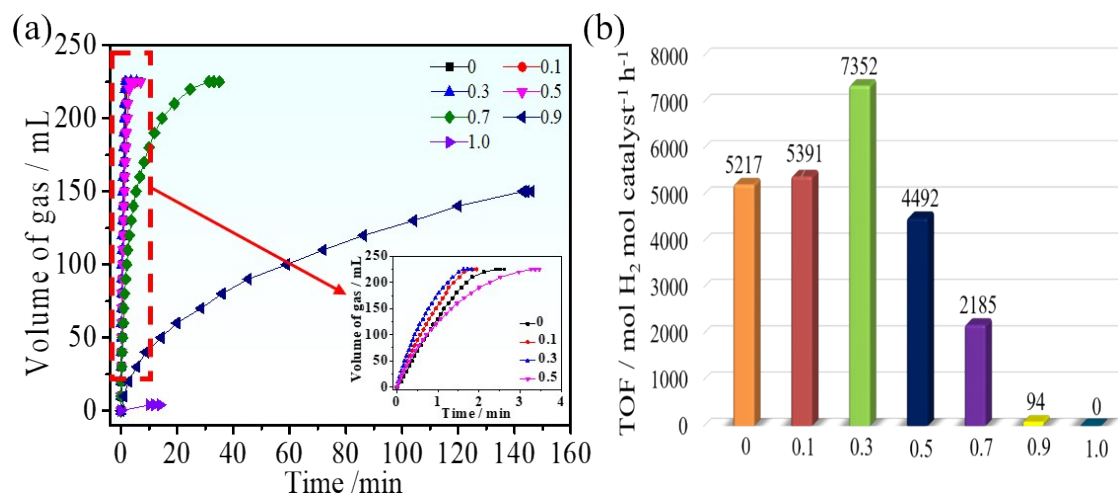


Fig. S11 (a) Time-course plots for the dehydrogenation of FA (1.0 M, 5.0 mL) catalyzed by Au_xPd_{1-x}/A-M-β-CD (x = 0, 0.1, 0.3, 0.5, 0.7, 0.9 and 1.0) at 323 K and the inset shows the corresponding larger plots; (b) their related initial TOF values.

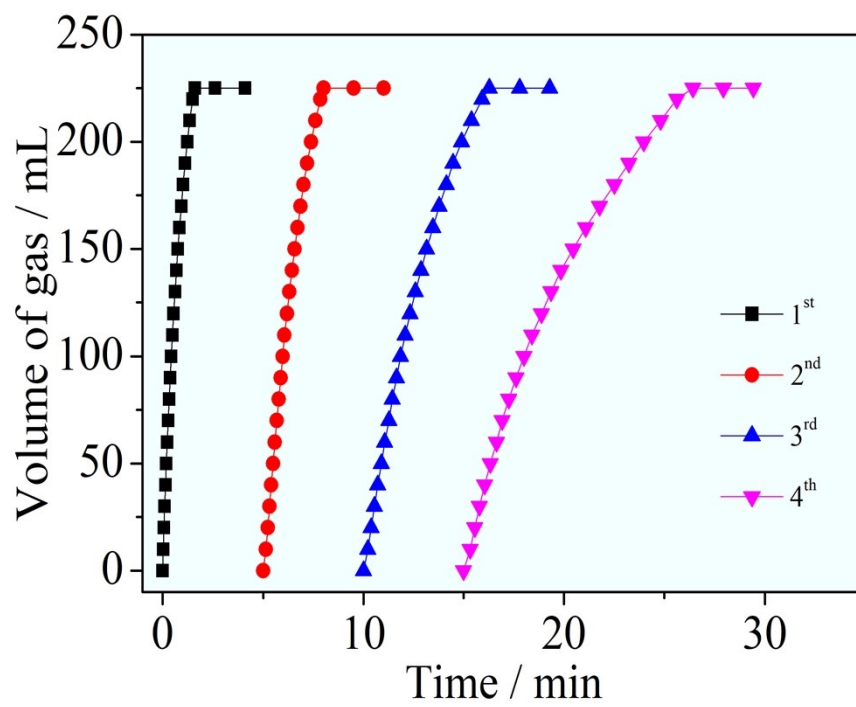


Fig. S12 Durability test of Au_{0.3}Pd_{0.7}/A-M-β-CD towards the dehydrogenation of FA.

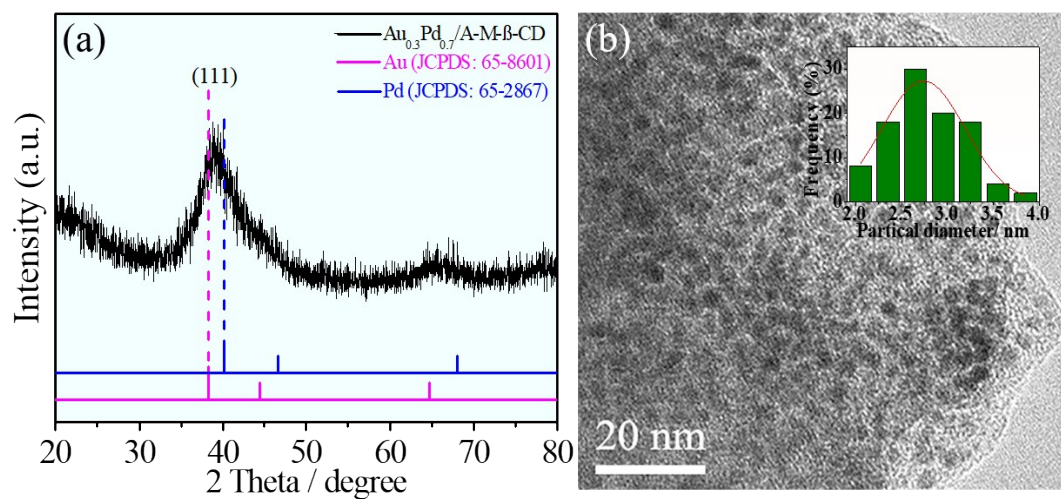


Fig. S13 (a) XRD spectrum; (b) TEM image and the corresponding particle size distribution (inset) of Au_{0.3}Pd_{0.7}/A-M-β-CD after the 4th run.