

Supplementary Information

Promoting photocatalytic hydrogen production by core-shell **CdS@MoO_x photocatalyst connected by S-Mo “bridge”**

Cankun Jiang, Lulu Zhang, Fan Gao, Xueyan Huang, Rui Lei, Yun Ye, Jie Yuan, Ping Liu*

Research Institute of Photocatalysis, State Key Laboratory of Photocatalysis on Energy and Environment, Fuzhou University, Fuzhou 350002, P. R. China

* Corresponding author. Tel.: +86-591-22865876; fax: +86-591-2286-5876; E-mail: liuping@fzu.edu.cn.

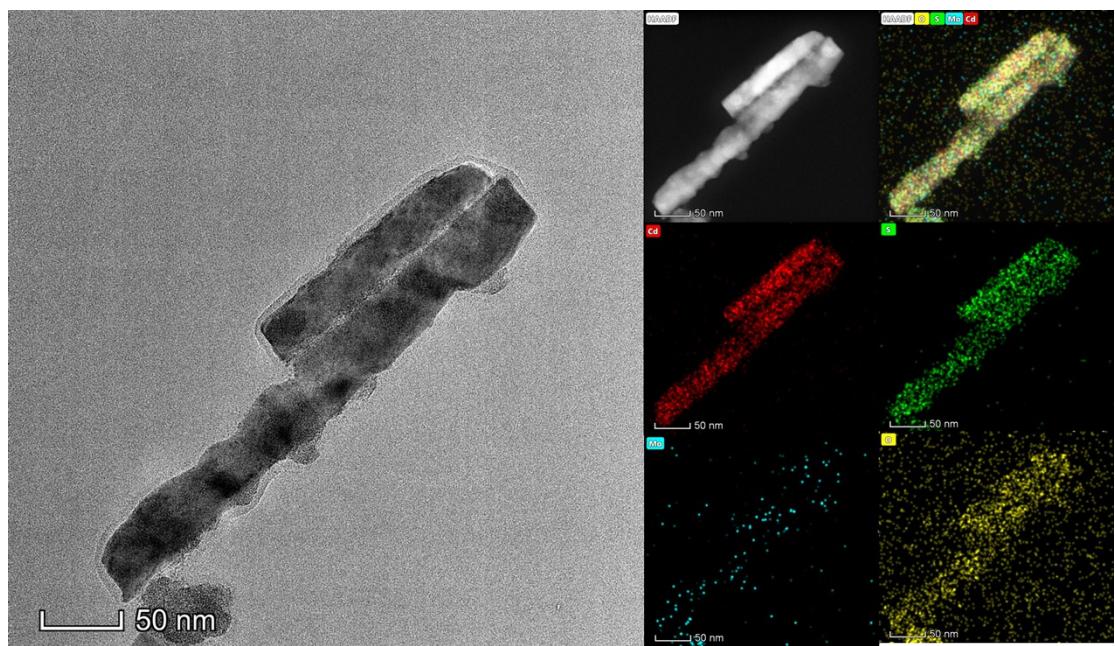


Fig.S1. TEM-EDX image of CdS@MoO_x-0.9 composite and the corresponding element mapping for Cd, S, Mo, O.

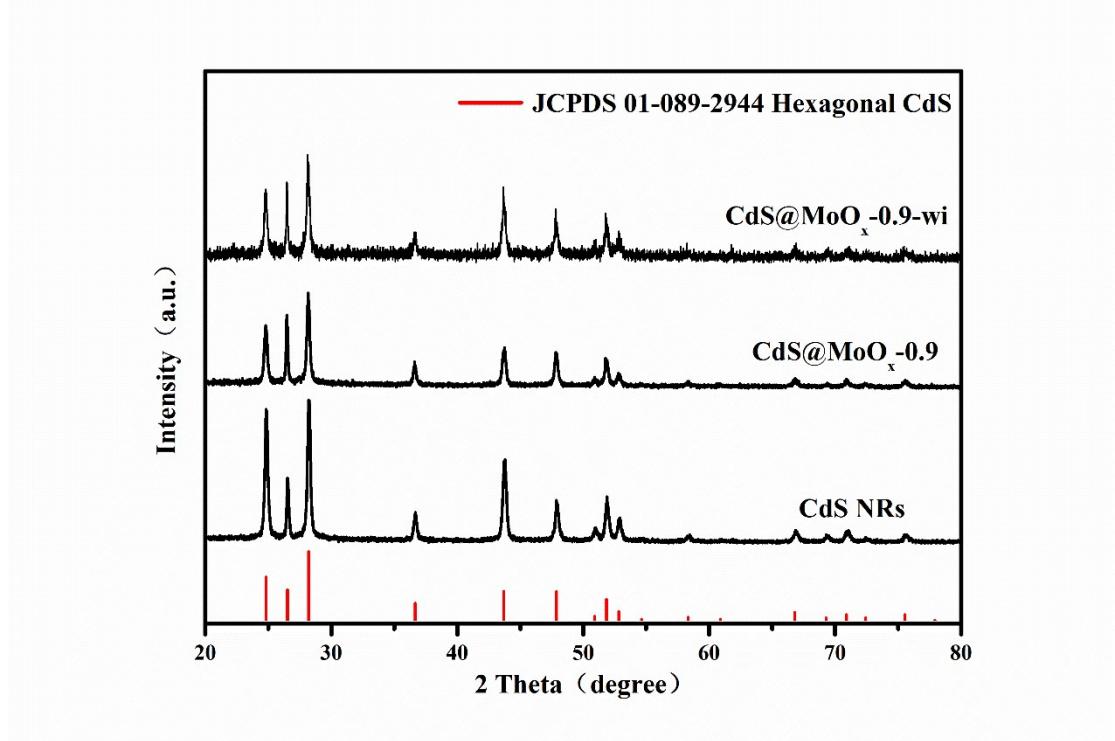


Fig.S2. Typical XRD patterns of CdS@MoO_x-0.9-wi.

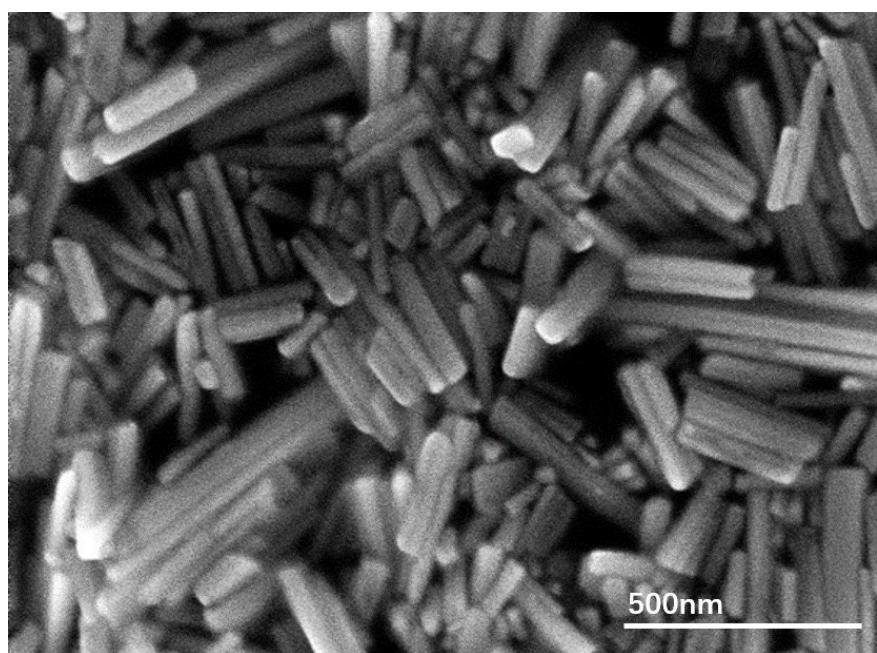


Fig.S3. FESEM image of CdS@MoO_x-0.9-wi.

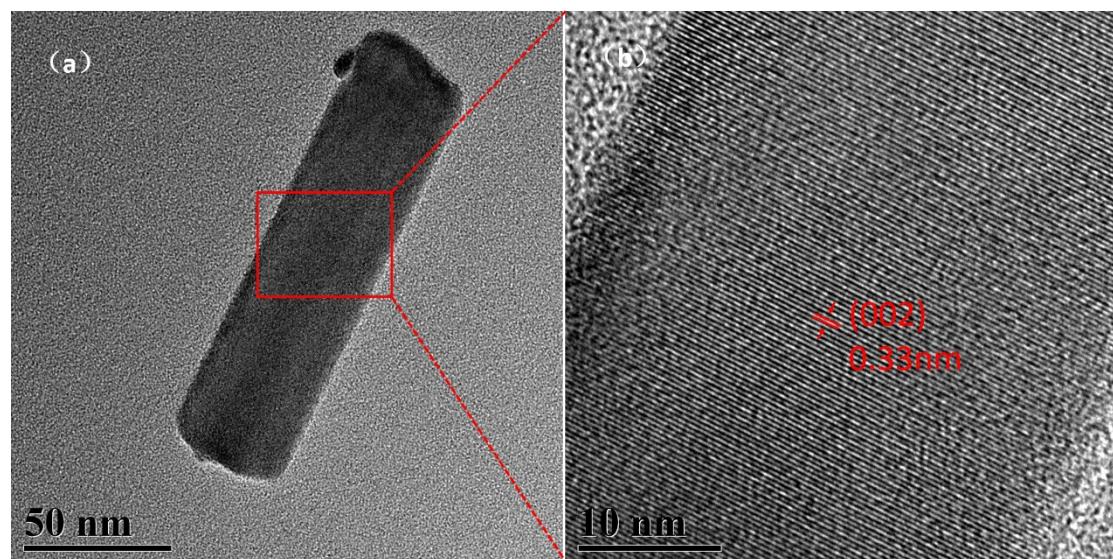


Fig.S4. (a) TEM and (b) HRTEM images of CdS@MoO_x-0.9-wi.

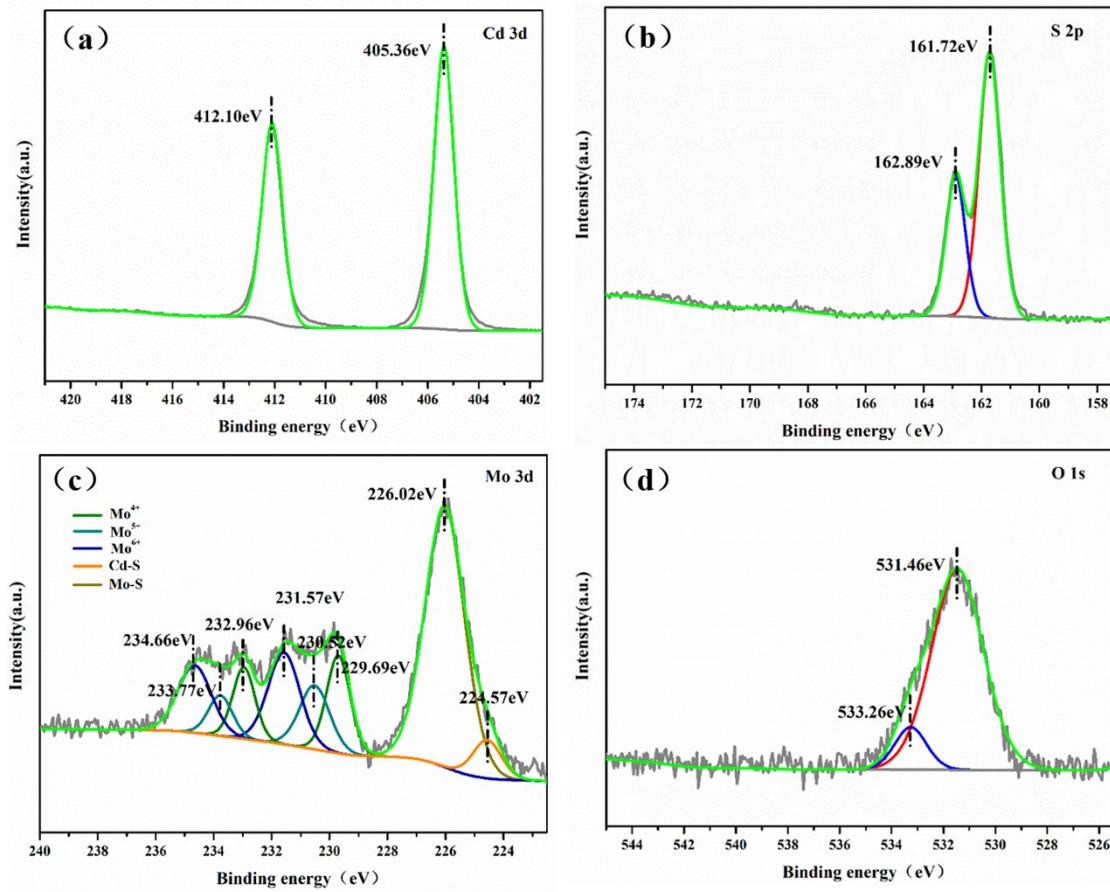


Fig.S5. The high-resolution XPS spectra of (a) Cd 3d, (b) S 2p, (c) Mo 3d and (d) O 1s of CdS@ MoO_x -0.9-wi sample.

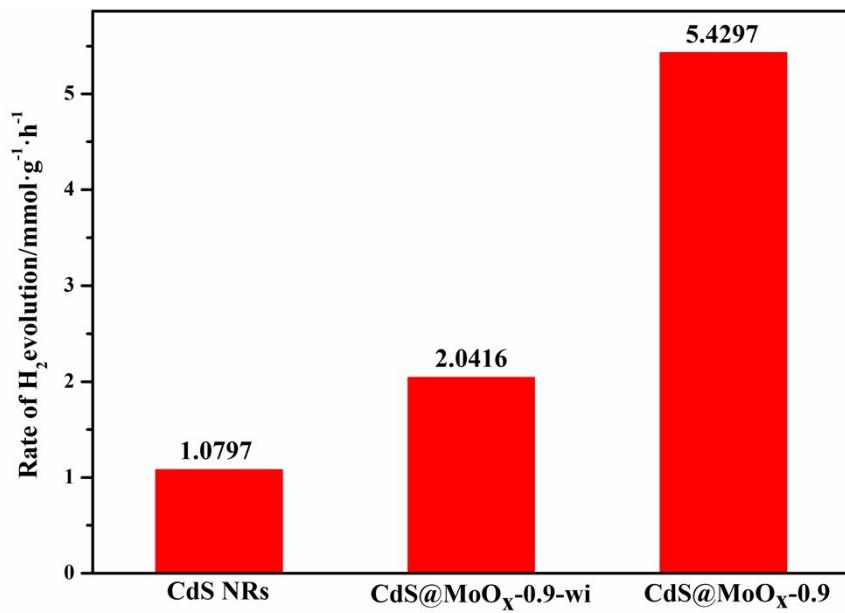


Fig.S6. Comparison of H_2 evolution rates of CdS NRs, CdS@MoO_x-0.9-wi and CdS@MoO_x-0.9.

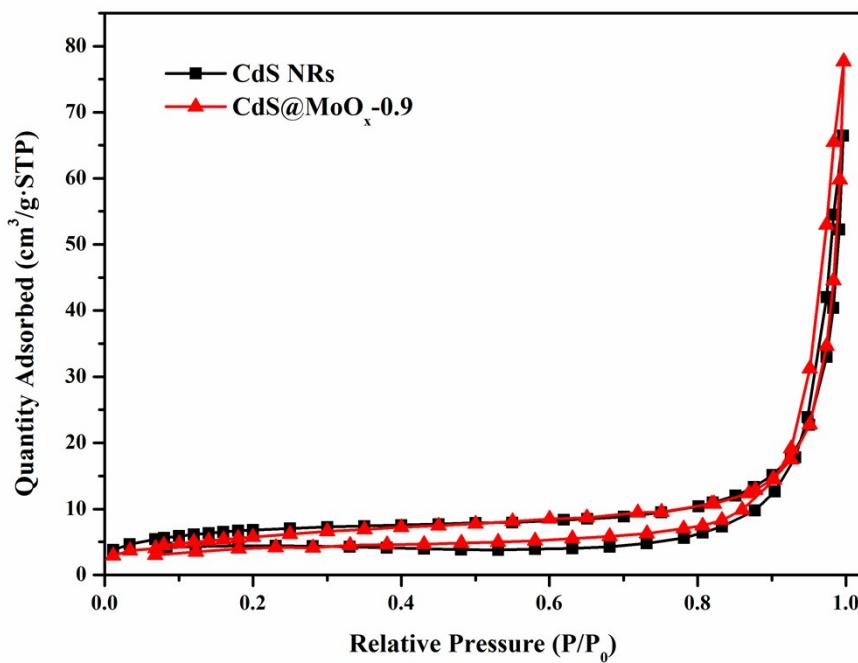


Fig.S7. Nitrogen adsorption desorption isotherms of CdS NRs and CdS@MoO_x-0.9.

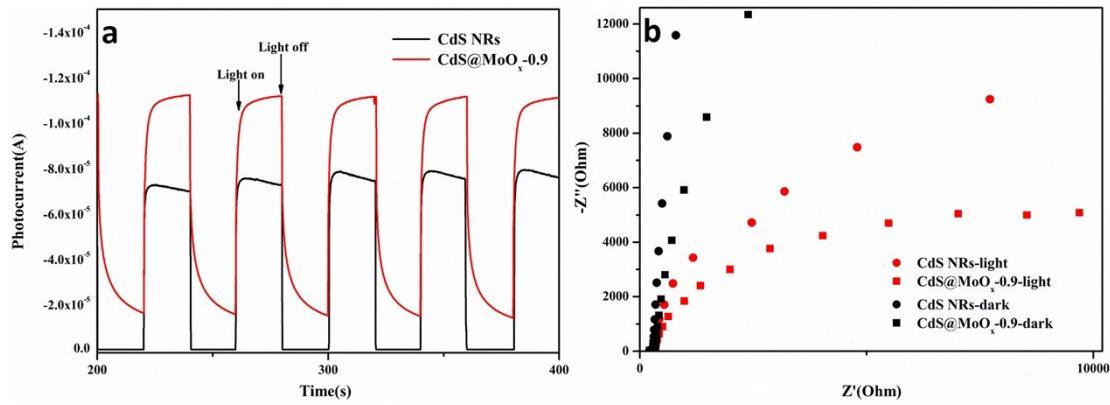


Fig.S8. (a) transient photocurrent and (b) Nyquist plots at an overpotential of 5 mV under the simulated visible-light (or dark) of CdS-NRs and CdS@MoO_x-0.9.

Photocatalyst	Sacrificial reagent	Light source	H ₂ production rate (mmol·h ⁻¹ ·g ⁻¹)	Ref.
CdS@MoO _x	10 vol.% Lactic acid	300 W Xe lamp ($\lambda > 400$ nm)	5.4297	This work
CdS/MoO _x -clusters	20 vol.% Lactic acid	300 W Xe lamp (UV-cut filter)	2.868	[1]
MoO ₂ -C/CdS	10 vol.% Lactic acid	300 W Xe lamp ($\lambda > 420$ nm)	16.08	[2]
MoO ₃ @CdS	1.5 mol·L ⁻¹ Na ₂ SO ₃ 0.2 mol·L ⁻¹ Na ₂ S	300 W Xe lamp ($\lambda > 400$ nm)	5.25	[3]
MoO ₂ /CdS	10 vol.% Lactic acid	300 W Xe lamp ($\lambda > 400$ nm)	15.11	[4]
CdS/Graphene	10 vol.% Lactic acid	350W Xe lamp ($\lambda > 400$ nm)	1.89	[5]

Table S1. Comparison of photocatalytic H₂ production performance with other CdS-based molybdenum oxide or electron transport media composite photocatalysts

References

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