

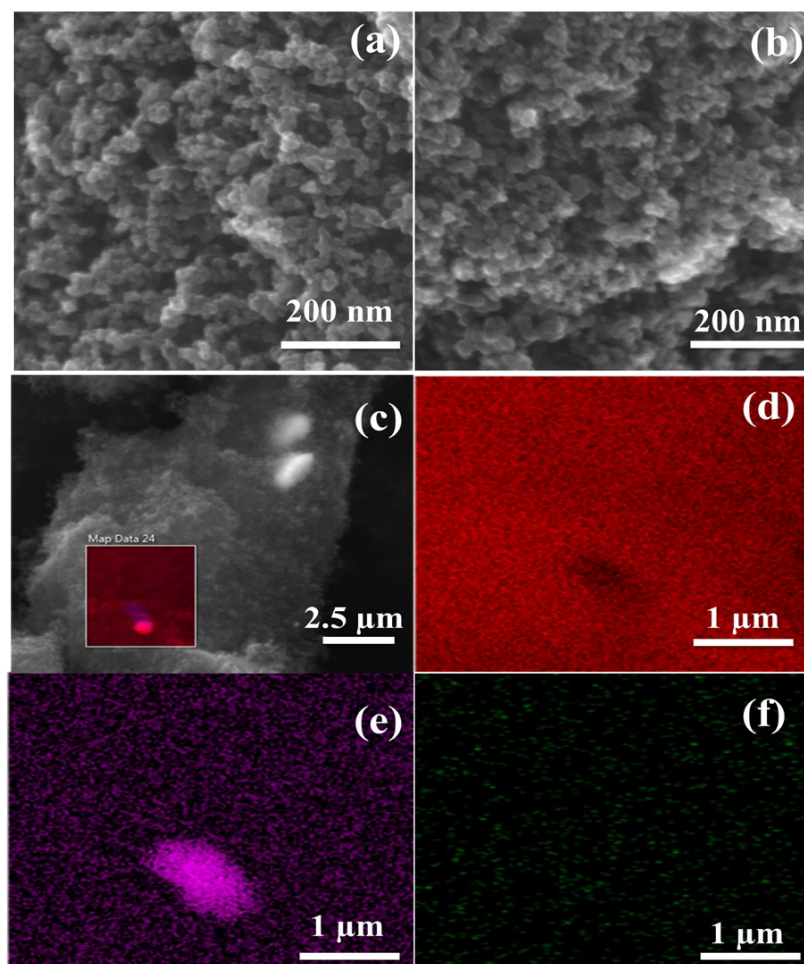
## Supporting Information

# NH<sub>3</sub>-Plasma Pre-Treated Carbon Supported Active Iron-Nitrogen Catalyst for Oxygen Reduction in Acid and Alkaline Electrolyte

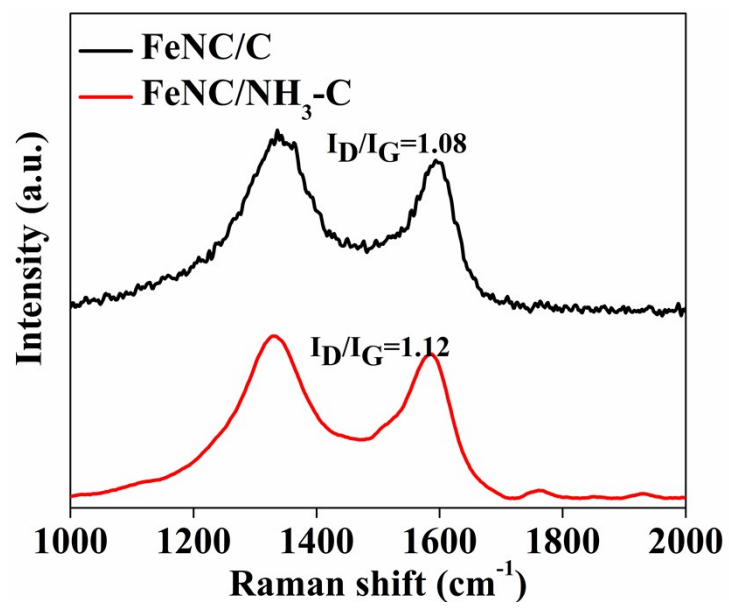
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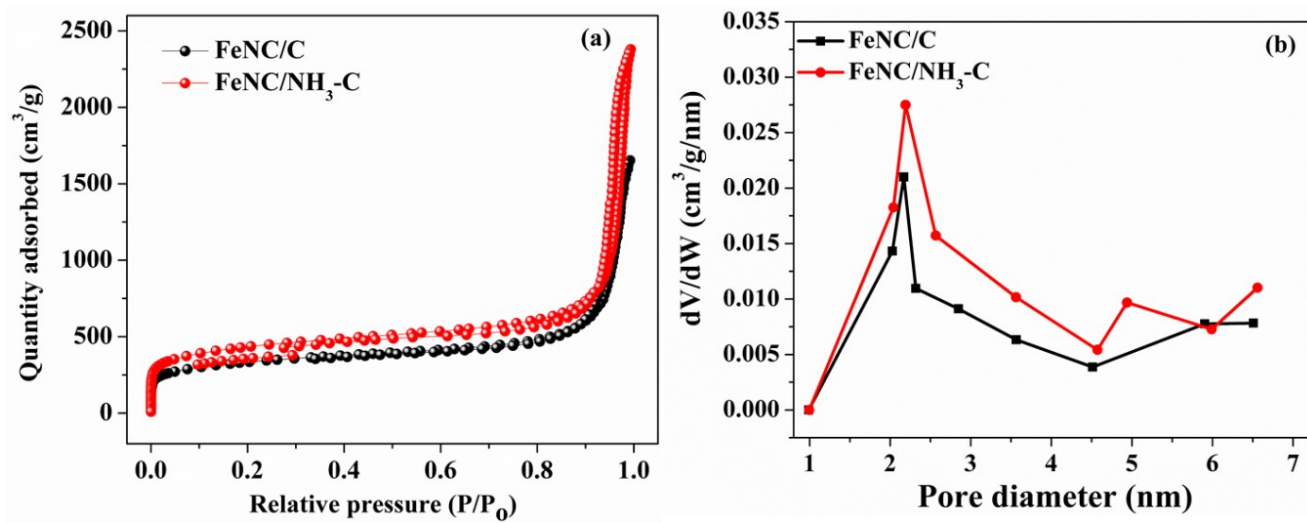
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**Fig. S1** SEM image of FeNC/C (a), FeNC/NH<sub>3</sub>-C (b), elemental mapping of FeNC/NH<sub>3</sub>-C on the selected area in square (c), carbon (d), iron (e) and nitrogen (f) respectively



**Fig. S2** Raman spectra of FeNC/C and FeNC/NH<sub>3</sub>-C.



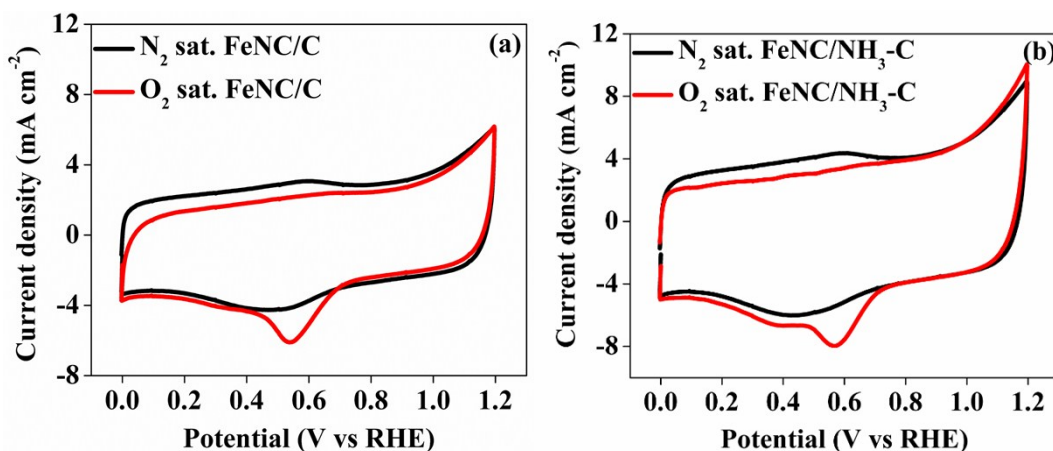
**Fig. S3** (a) Nitrogen adsorption and desorption isotherm; (b) Pore size distribution of FeNC/C and FeNC/NH<sub>3</sub>-C

**Table S1.** XPS surface atomic ratio of FeNC/C and FeNC/NH<sub>3</sub>-C

Catalyst	C (Atomic %)	N (Atomic %)	Fe (Atomic %)	O (Atomic %)	C/N ratio
FeNC/C	96.54	0.52	0.03	2.91	185.65
FeNC/NH <sub>3</sub> -C	96.45	0.53	0.04	2.98	181.98

**Table S2.** ICP-AES and CHN/O elemental analysis of FeNC/C and FeNC/NH<sub>3</sub>-C

Catalyst	Fe from ICP-AES (wt %)	CHN/O Elemental analysis			
		C (wt %)	N (wt %)	H (wt %)	O (wt %)
FeNC/C	0.14	96.51	1.21	0.08	2.06
FeNC/NH <sub>3</sub> -C	0.22	96.27	1.30	0.09	2.12

**Fig. S4** Comparison of N<sub>2</sub> and O<sub>2</sub> saturated cyclic voltammogram in 0.5 M H<sub>2</sub>SO<sub>4</sub> at a scan rate of 50 mV s<sup>-1</sup> studied with (a) FeNC/C (b) FeNC/NH<sub>3</sub>-C

**Table S3. ORR activity of FeNC/C and FeNC/NH<sub>3</sub>-C compared with 20% Pt/C in 0.5 M H<sub>2</sub>SO<sub>4</sub>**

<b>Catalyst</b>	<b>Onset potential (V vs RHE)</b>	<b>Half wave Potential (V vs RHE)</b>	<b>Limiting current density at 0.2 V vs RHE (mA cm<sup>-2</sup>)</b>
FeNC /C	0.75	0.58	-5.3
FeNC/NH <sub>3</sub> -C	0.80	0.63	-5.3
20% Pt/C	0.97	0.87	-5.5

**Table S4. ORR activity of FeNC/C and FeNC/NH<sub>3</sub>-C compared with 20% Pt/C in 0.1 M KOH**

<b>Catalyst</b>	<b>Onset potential (V vs RHE)</b>	<b>Half wave Potential (V vs RHE)</b>	<b>Limiting current density at 0.2 V vs RHE (mA cm<sup>-2</sup>)</b>
FeNC/C	0.99	0.85	-5.74
FeNC/NH <sub>3</sub> -C	1.0	0.86	-5.81
20% Pt-C	1.0	0.87	-5.16

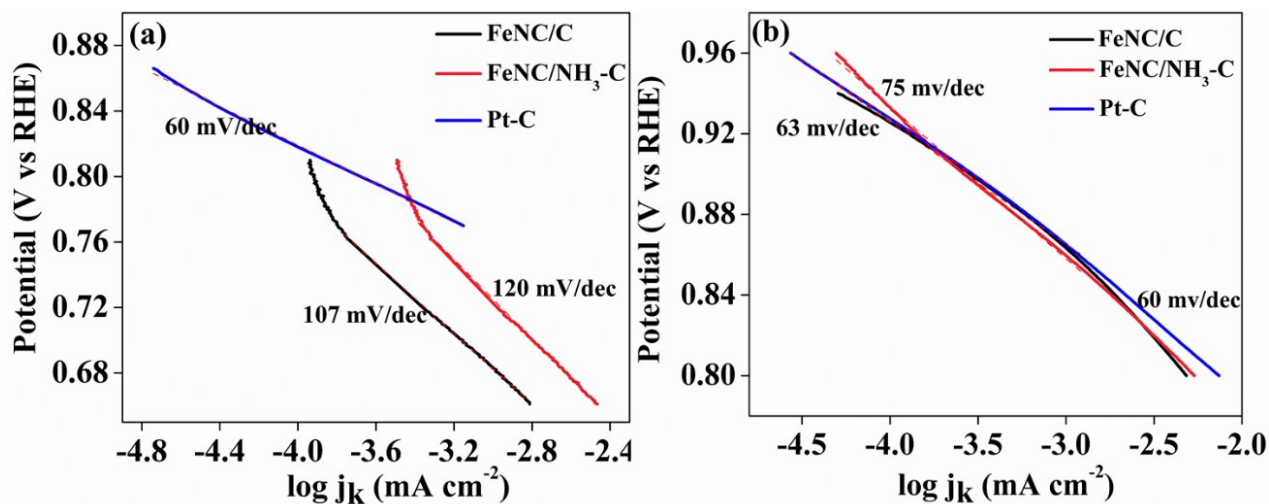


Fig. S5 Tafel plot of FeNC/C, FeNC/NH<sub>3</sub>-C and Pt-C in (a) 0.5 M H<sub>2</sub>SO<sub>4</sub> and (b) 0.1 M KOH.

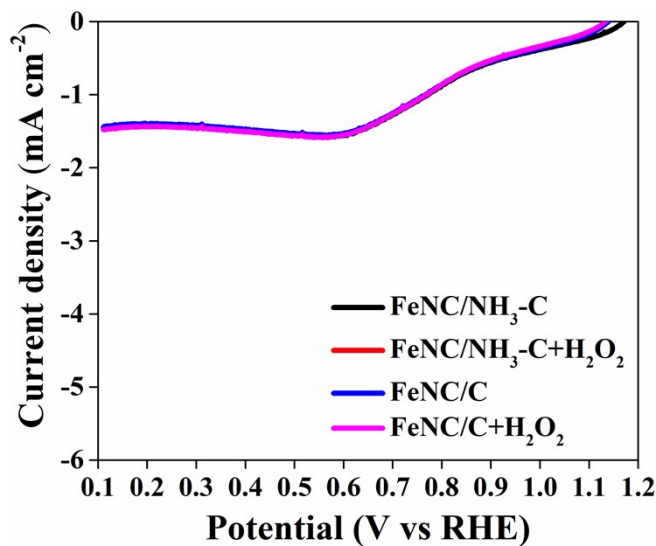


Fig. S6 LSV of FeNC/C and FeNC/NH<sub>3</sub>-C in an electrolyte solution after adding 0.01 M peroxide in 0.5 M H<sub>2</sub>SO<sub>4</sub> under N<sub>2</sub> with 5 mV s<sup>-1</sup> scan rate

**Table S5.** A Comparative study of  $E_{\text{onset}}$ ,  $E_{1/2}$  towards ORR in acid and alkaline condition for several reported M-N-C catalysts

Catalyst	Onset potential	Half-wave potential vs RHE	Electrolyte	Reference
N-Fe/G (60) 900 s	0.834 V	0.716 V	0.1 M HClO <sub>4</sub>	1
FeNC-900	0.85 V	0.72 V	0.1 M HClO <sub>4</sub>	2
Fe-N-CNF	0.79 V	0.60 V	0.5 M H <sub>2</sub> SO <sub>4</sub>	3
Fe-N-C HNSs-750	0.78 V	0.54 V	0.5 M H <sub>2</sub> SO <sub>4</sub>	4
Fe-N/C 120	0.88 V	/	0.1 M HClO <sub>4</sub>	5
Fe-CNT-PA	0.80 V	/	0.5 M H <sub>2</sub> SO <sub>4</sub>	6
<b>FeNC/NH<sub>3</sub>-C</b>	<b>0.80 V</b>	<b>0.63 V</b>	<b>0.5 M H<sub>2</sub>SO<sub>4</sub></b>	<b>This work</b>
Fe-N-CC	0.94	0.83	0.1 M KOH	7
N-Fe-C@CNTs	1.01	0.88	0.1 M KOH	8
C-Fe(OH) <sub>3</sub> @ ZIF-1000	0.99	0.88	0.1 M KOH	9
Fe/Fe <sub>2.5</sub> C/Fe <sub>3</sub> N/N-CNT-30	0.93	0.79	0.1 M KOH	10
(MC-O2350-NH31050)	0.978	0.871	0.1 M NaOH	11
N-C <sub>9</sub> S <sub>8</sub> /G	0.941	0.76	0.1 M KOH	12
<b>FeNC/NH<sub>3</sub>-C</b>	<b>1 V</b>	<b>0.86</b>	<b>0.1 M KOH</b>	<b>This work</b>

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