

**Reversible borohydride formation from aluminium hydrides and {H(9-BBN)}₂:
structural, thermodynamic and reactivity studies**

Alexa Caise, Eugene L. Kolychev, Jamie Hicks, M. Ángeles Fuentes, Jose M. Goicoechea
and Simon Aldridge

Supporting Information

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1. Variable temperature NMR experiments

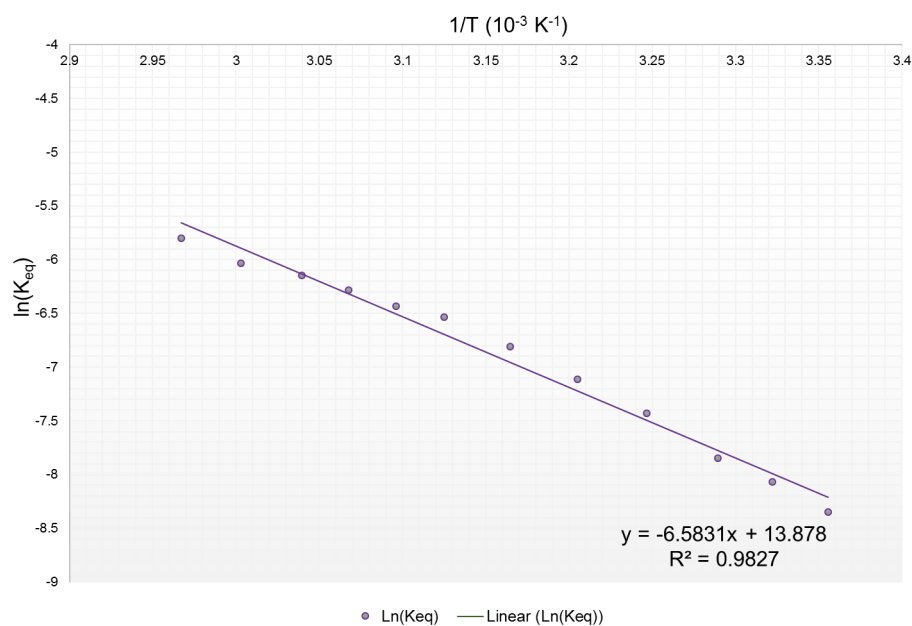


Figure s1: A Van't Hoff plot for the dissociation of the borane from $[\text{Dipp1}]\text{Al}(\text{Et})\{\text{H}_2(9\text{-BBN})\}$. Temperature range from 295 K to 345 K.

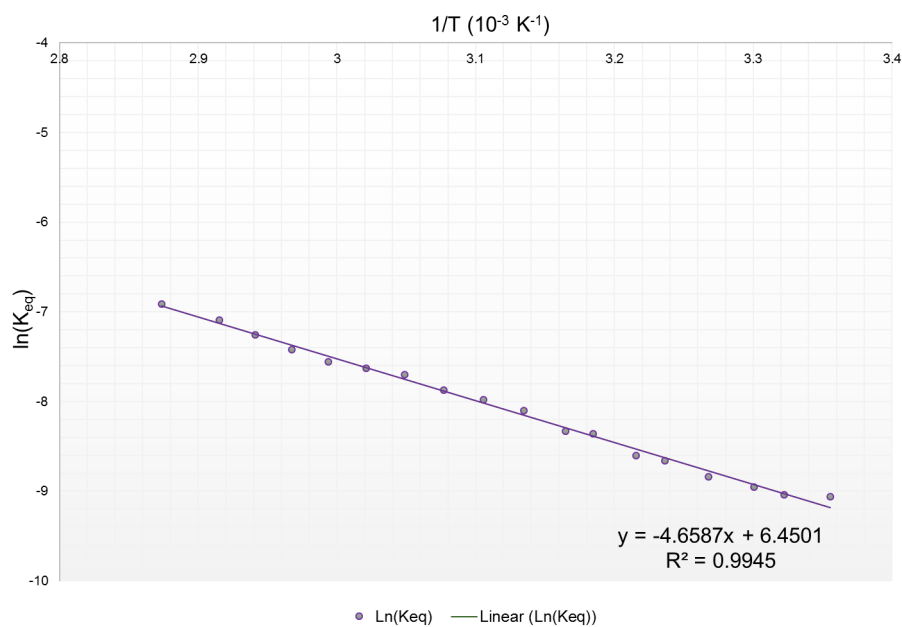


Figure s2: A Van't Hoff plot for the dissociation of the borane from $[\text{Dipp1}]\text{Al}(\text{Me})\{\text{H}_2(9\text{-BBN})\}$. Temperature range from 295 K to 358 K.

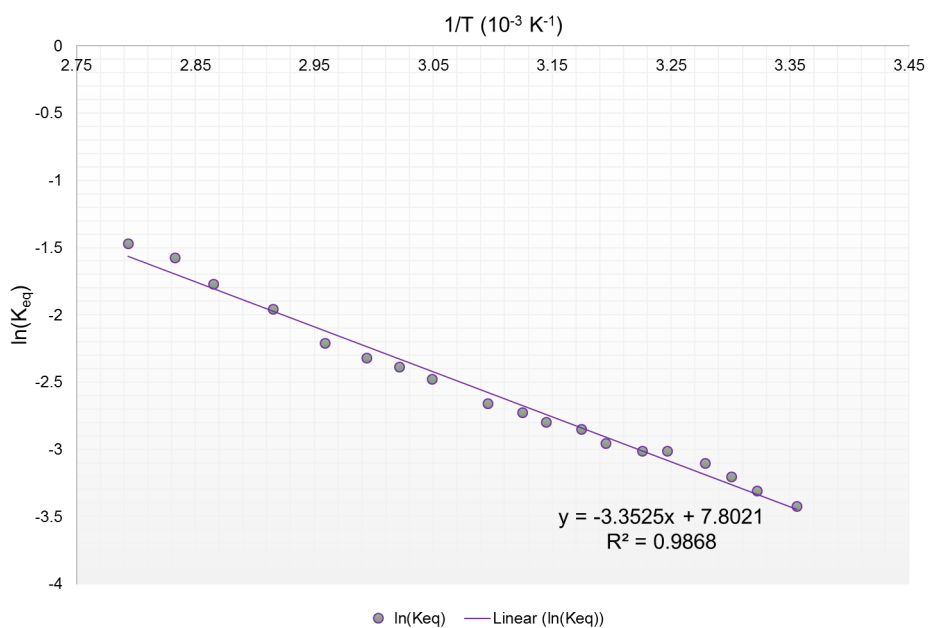


Figure s3: A Van't Hoff plot for the dissociation of the borane from $[\text{Dipp1}]\text{Al}(\text{OTf})\{\text{H}_2(9\text{-BBN})\}$. Temperature range from 298 K to 333 K.

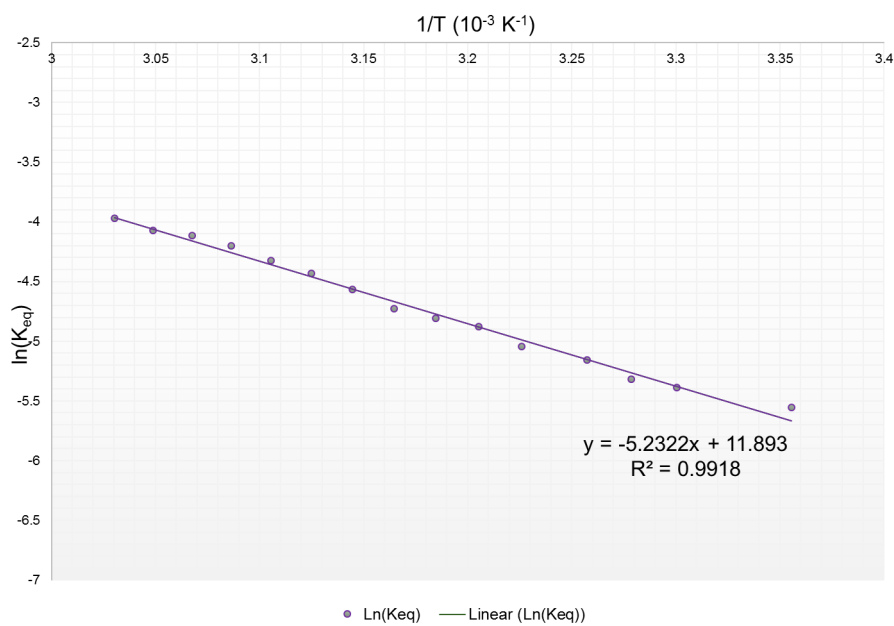
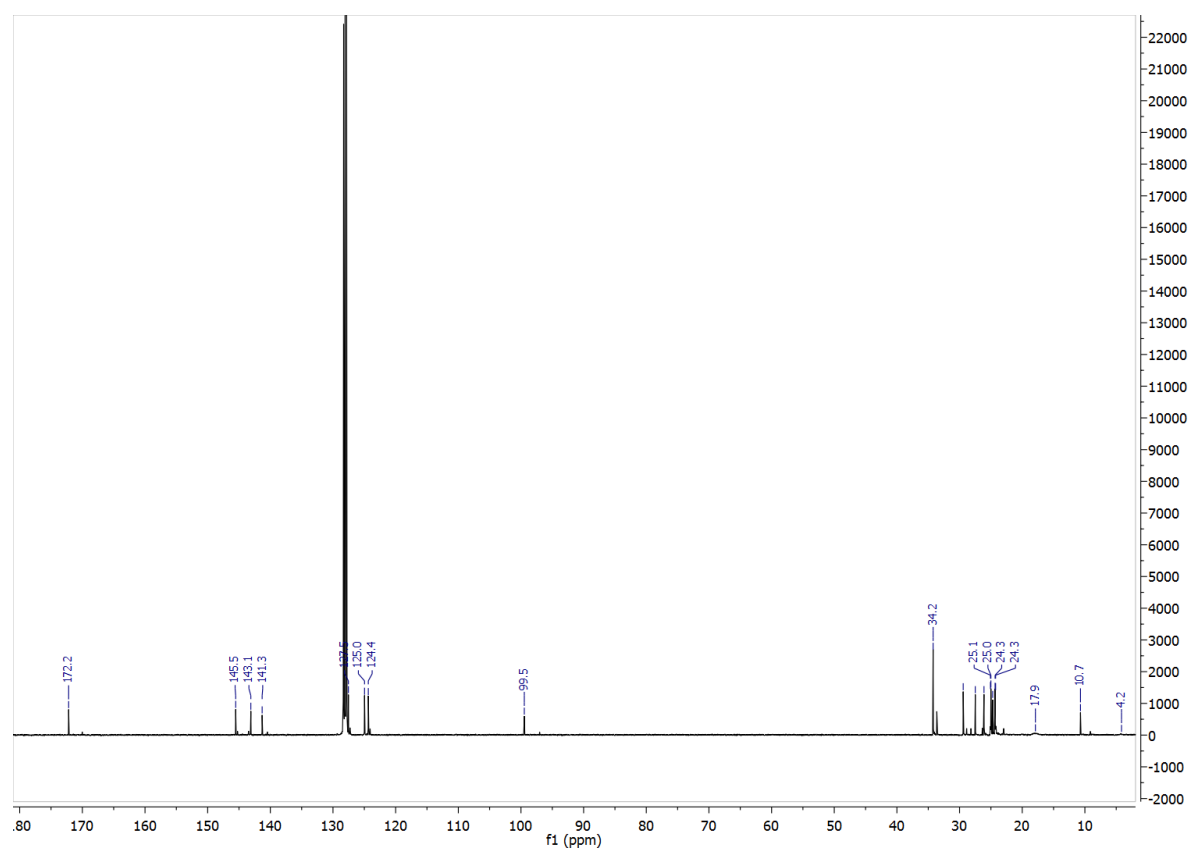
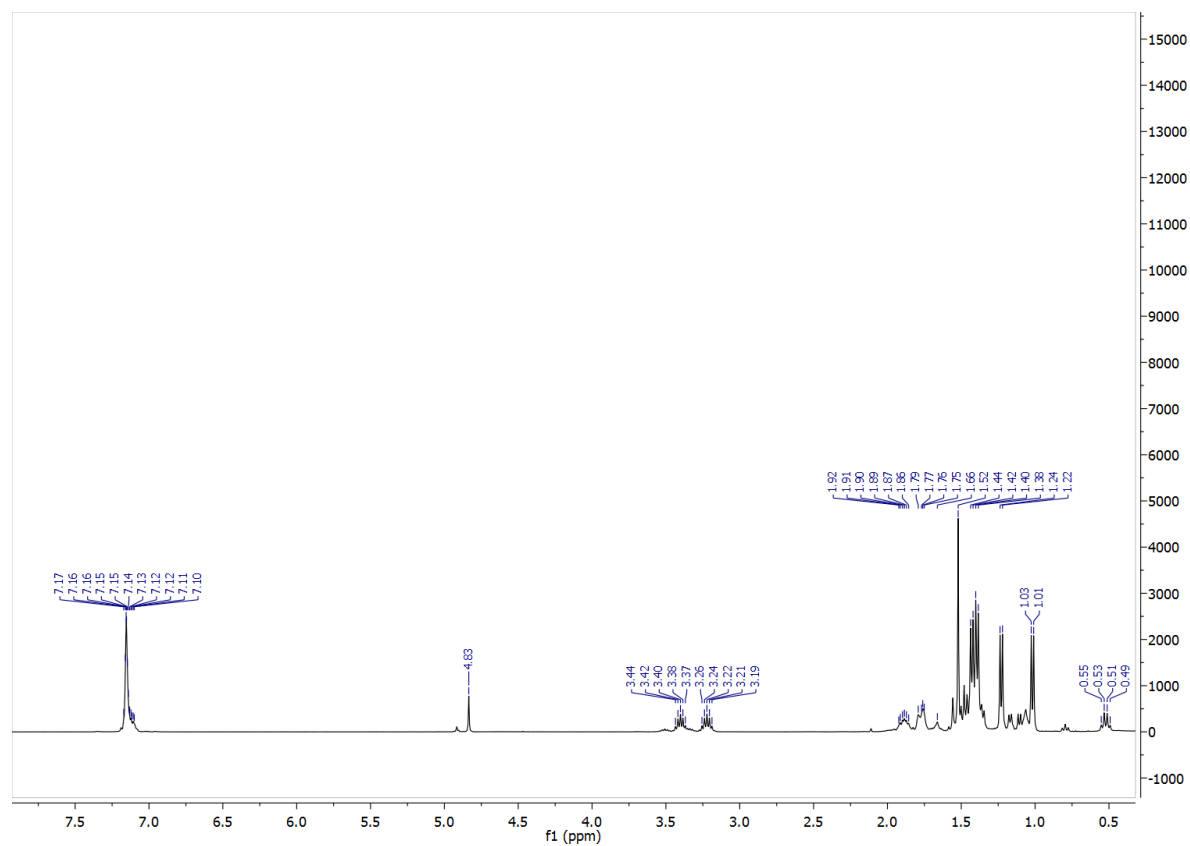
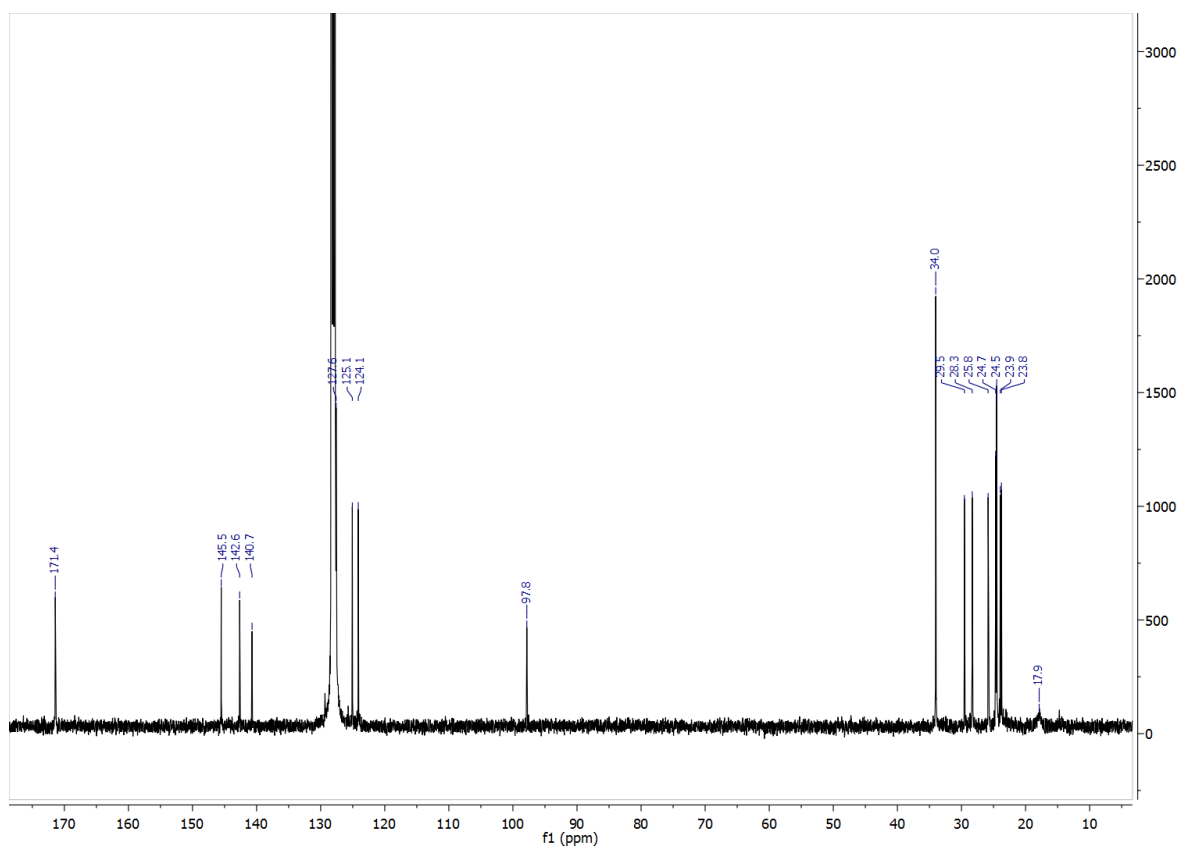
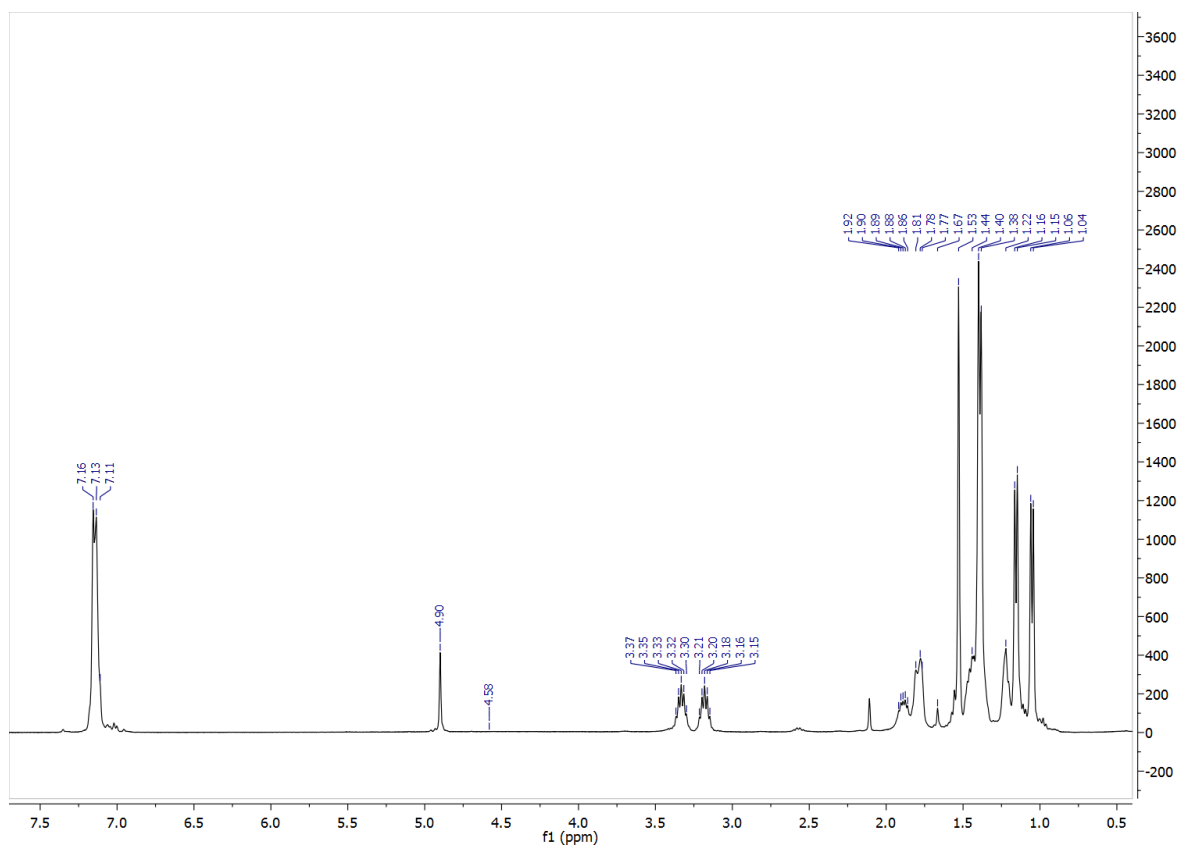


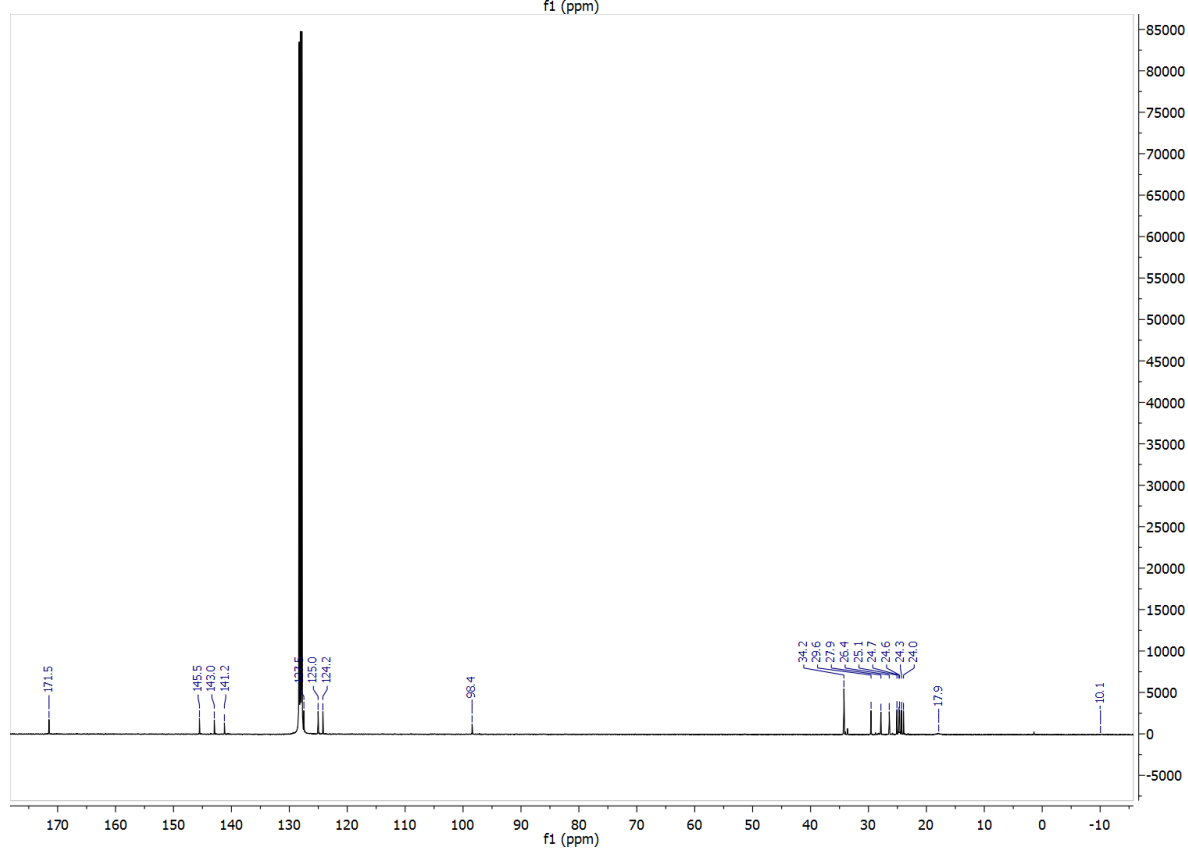
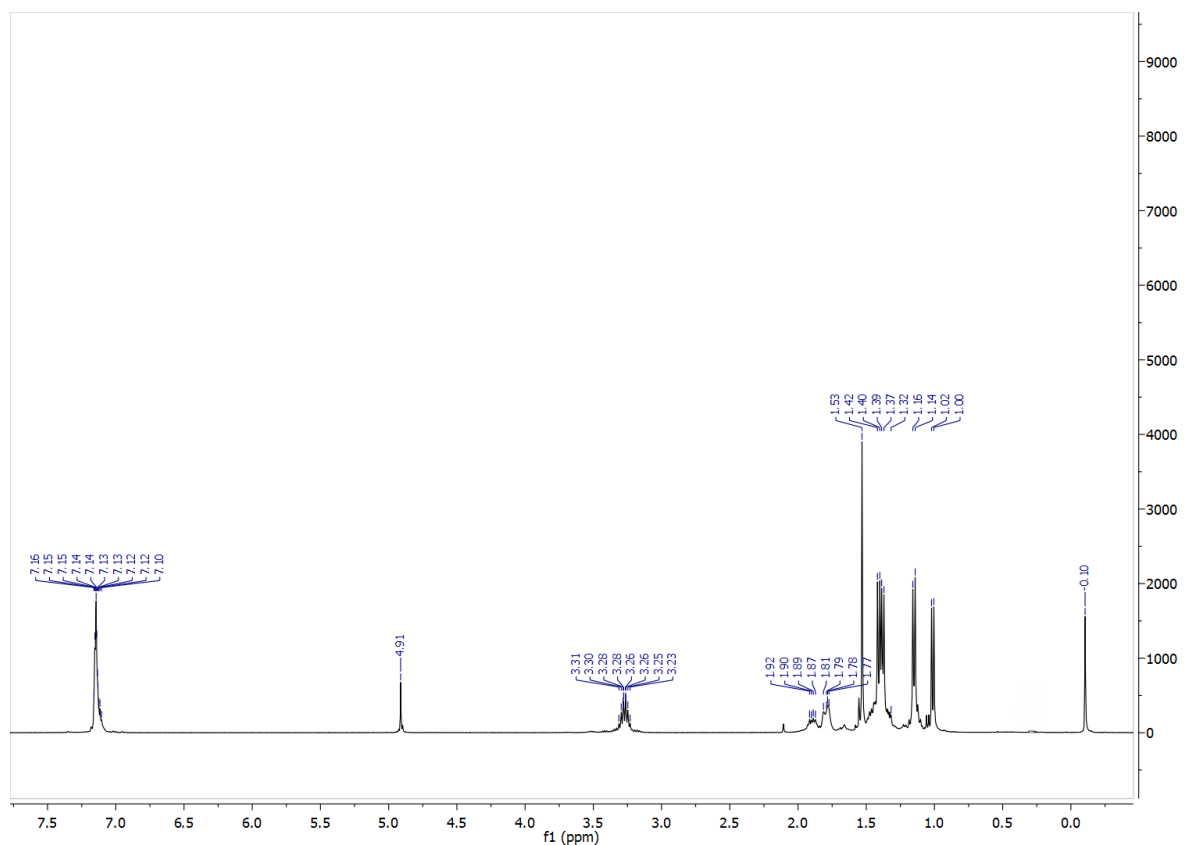
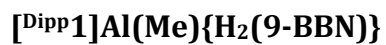
Figure s4: A Van't Hoff plot for the dissociation of the borane from $[\text{Dipp2}]\text{Al}(\text{Et})\{\text{H}_2(9\text{-BBN})\}$. Temperature range from 298 K to 333 K.

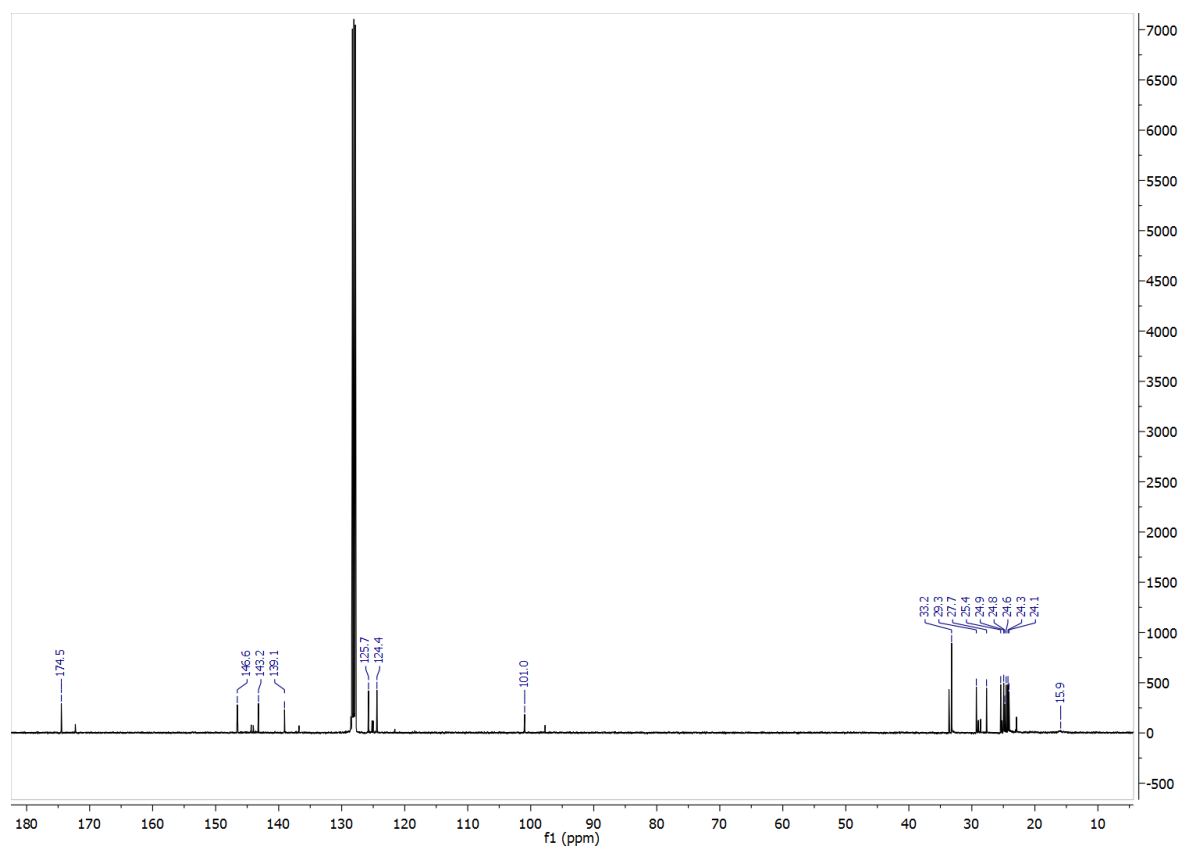
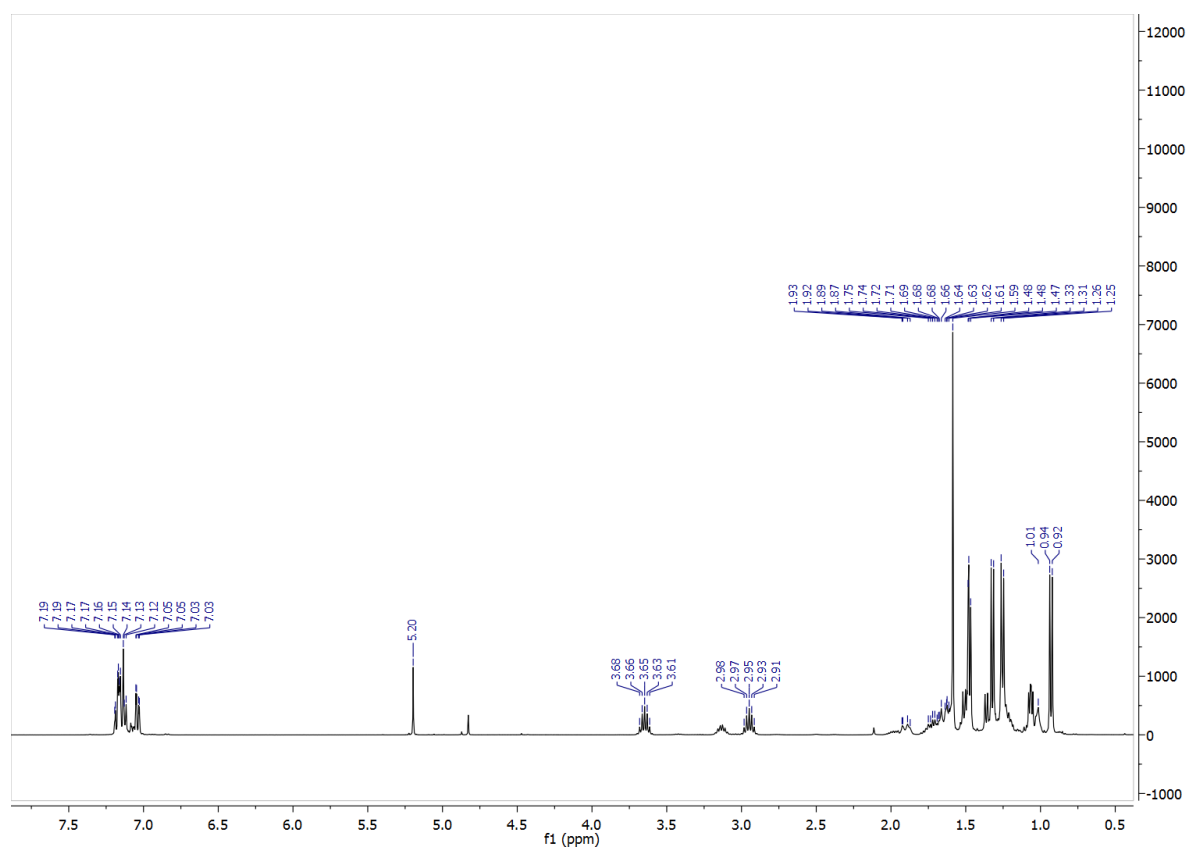
2. ^1H and ^{13}C NMR spectra of novel compounds

$[\text{Dipp}^1]\text{Al}(\text{Et})\{\text{H}_2(9\text{-BBN})\}$

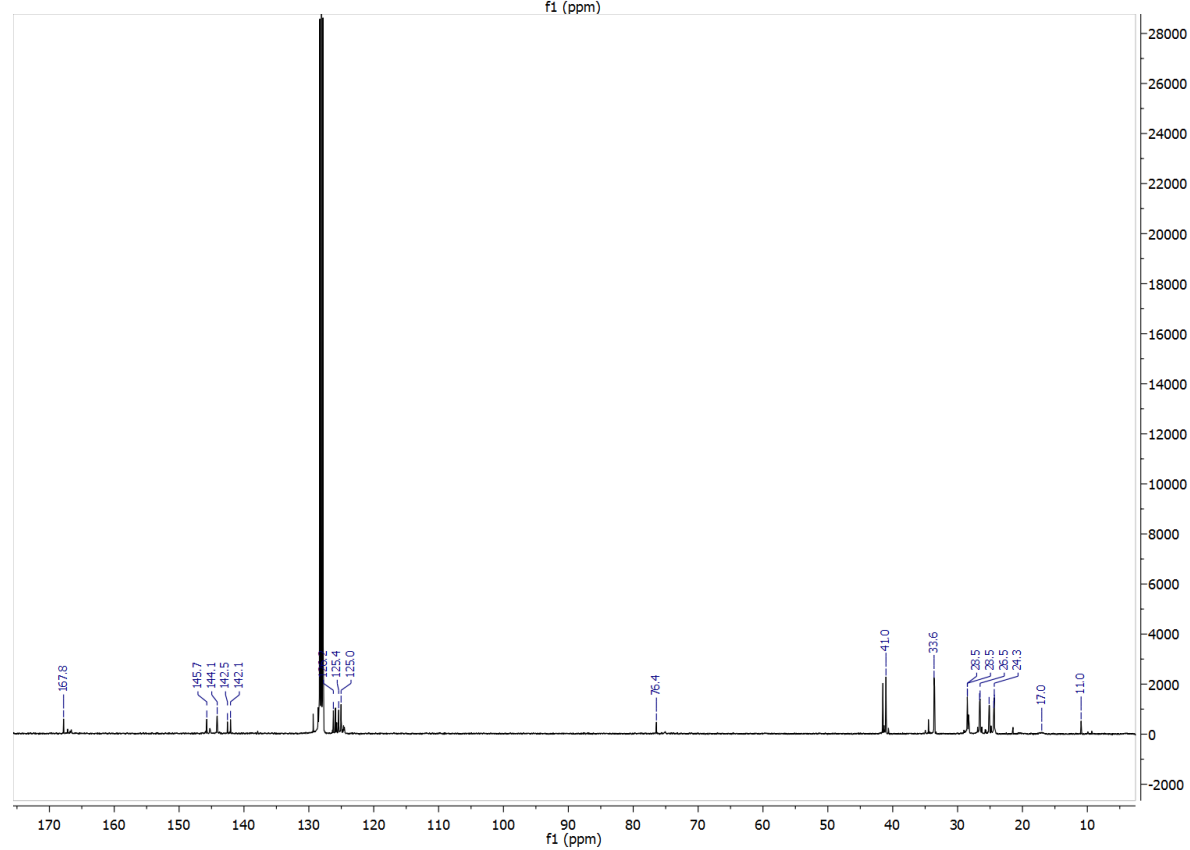
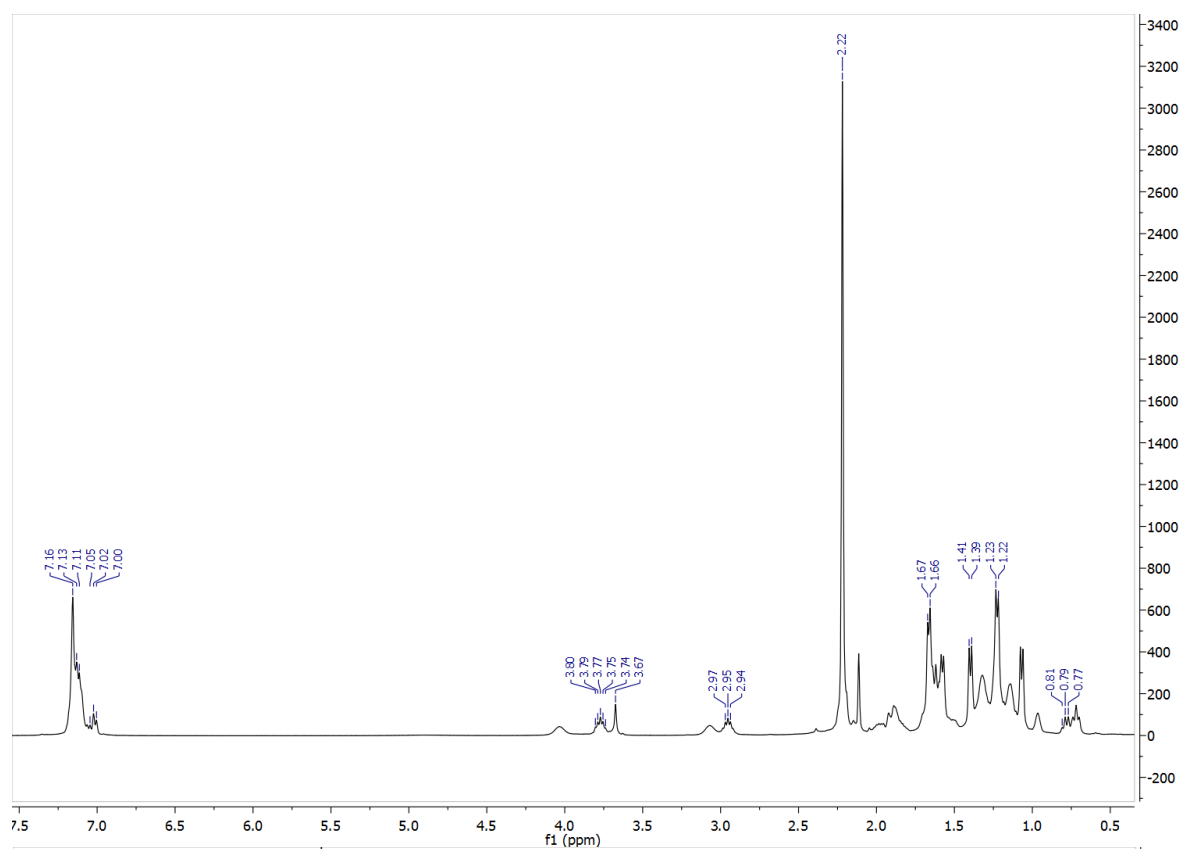








[Dipp₂]Al(Et){H₂(9-BBN)}



[Dipp₂]Al(H){H₂(9-BBN)}

