## **Supporting Information**

## Unusual Formation of NiCo<sub>2</sub>O<sub>4</sub>@MnO<sub>2</sub>/Nickel Foam/MnO<sub>2</sub> Sandwich as Advanced Electrodes for Hybrid Supercapacitors

Chunli Guo,<sup>1</sup> Jie Li,<sup>1</sup> haibo Li,<sup>2</sup> huaiping Zhang,<sup>1</sup> Lifeng Hou,<sup>1</sup> Yinghui Wei,<sup>1</sup> Jing Liu,<sup>3</sup> Yanting Chu,<sup>1</sup> and Shenglin Xiong<sup>4\*</sup>

<sup>1</sup>College of Materials Science and Engineering, Taiyuan University of Technology, Taiyuan, Shanxi, 030024, P.R. China

<sup>2</sup>School of Materials Science and Engineering, Liaocheng University, Liaocheng, Shandong 252059, PR China

<sup>3</sup>College of Materials Science and Engineering, Qingdao University of Science and Technology, Qingdao 266042, China

<sup>4</sup>Key Laboratory of Colloid and Interface Chemistry, Ministry of Education, School of Chemistry and Chemical Engineering, and State Key Laboratory of Crystal Materials, Shandong University, Jinan, 250100, PR China

\*Correspondence and requests for materials should be addressed to S.L.X. (email: chexsl@sdu.edu.cn).



**Fig. S1** FESEM images of the as-synthesized cobalt-nickel hydroxide precursor nanowire arrays grown on NF.



Fig. S2 (a, b) The different distributions of cobalt-nickel hydroxide precursors on NF.



Fig. S3 EDX spectrum of the  $NiCo_2O_4@MnO_2/NF/MnO_2$  sandwiches.



Fig. S4 XRD patterns of the  $NiCo_2O_4$ @MnO<sub>2</sub>/NF/MnO<sub>2</sub> sandwiches.



Fig. S5 The corresponding EDX mapping image showing the distribution of Mn, Co, and Ni for  $NiCo_2O_4@MnO_2$  core-shell structures.



Fig. S6 XPS spectrum of survey for the as-fabricated  $NiCo_2O_4@MnO_2/NF/MnO_2$  sandwiches.



Fig. S7 FESEM images of the side A (A) and side B (B) of  $NiCo_2O_4@MnO_2/NF/MnO_2$  sandwiches after 30,000 cycles.



Fig. S8 the GCD curves of HSC at 0.1 A  $g^{-1}$  (A), 1.5 A  $g^{-1}$  (B) and 2.5 A  $g^{-1}$  (C).

**Table S1.** Electrochemical performance of the  $NiCo_2O_4@MnO_2/NF/MnO_2$  sandwich electrode in this study, compared with some other  $NiCo_2O_4$ -based electrodes reported in previous literature.

Electrode materials	Specific capacity	Electrolyte	Cycling stability	Refs.
NiCo <sub>2</sub> O <sub>4</sub> nanosheets	404.6 C g <sup>-1</sup> at 1 A g <sup>-1</sup>	6 M KOH	93.2% after 6000 cycles	[48]
3D network-like mesoporous $NiCo_2O_4$	465.5 C g⁻¹ at 3 A g⁻¹	2 M KOH	125.2% after 1000 cycles	[49]
NiCo <sub>2</sub> O <sub>4</sub> -decorated porous carbon nanosheets	238.7 C g <sup>-1</sup> at 2 A g <sup>-1</sup>	6 M KOH	98.0% after 3000 cycles	[50]
$NiCo_2O_4@MnO_2$ core-shell nanosheets	1.08 C cm <sup>-2</sup> at 3 mA cm <sup>-2</sup>	1 M NaOH	92.6% after 2000 cycles	[27]
NiCo <sub>2</sub> O <sub>4</sub> /MnO <sub>2</sub> heterostructured nanosheet	0.44 C cm <sup>-2</sup> at 2 mA cm <sup>-2</sup>	2 M KOH	110.0% after 6000 cycles	[51]
$NiCo_2O_4@MnO_2$ core-shell nanowire	1.01 C cm <sup>-2</sup> at 2 mA cm <sup>-2</sup>	1 M NaOH	113.6% after 8000 cycles	[23]
NiCo <sub>2</sub> O <sub>4</sub> @MnO <sub>2</sub> /NF/MnO <sub>2</sub>	1.70 C cm <sup>-2</sup> at 2 mA cm <sup>-2</sup>	6 M KOH	90.0% after 30,000 cycles	Our work

Current density	Discharge time	Specific capacity	Energy density	Power density
( <b>A g</b> ⁻¹)	(s)	(C g⁻¹)	(Wh kg⁻¹)	(W kg⁻¹)
0.1	2407.0	240.6	53.5	80.0
0.5	322.5	161.3	35.8	400.0
1.0	149.2	149.3	33.2	801.1
1.5	89.0	133.4	29.7	1201.3
2.0	64.6	129.3	28.7	1599.4
2.5	50.0	125.0	27.8	2001.6
3.0	40.4	121.3	27.0	2405.9
5.0	22.0	110.1	24.5	4009.1
10.0	9.3	93.0	20.7	8012.9

**Table S2.** the discharge time, specific capacity, energy densities and power densities of the HSC at various current densities.