

## Electronic Supplementary Information (ESI)

### Models of the iron-only hydrogenase enzyme: Structure, electrochemistry and catalytic activity of $\text{Fe}_2(\text{CO})_3(\mu\text{-dithiolate})(\mu,\kappa^1,\kappa^2\text{-triphos})$

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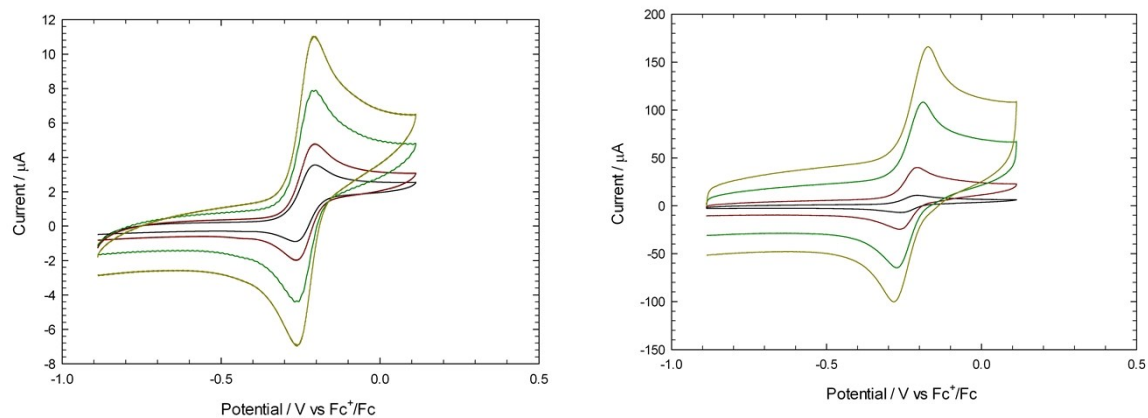
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- (i) CVs of oxidation of **1** over a range of scan rates
- (ii) CVs of oxidation of **3** over a range of scan rates
- (iii) Electrocatalytic performance of **4** on addition of strong acid
- (iv) Oxidation response for **3** on addition of strong acid
- (v) Computational Details
- (vi) Crystallographic data and structure refinement for **1**
- (vii) Crystallographic data and structure refinement for **3**

**(i) CVs of oxidation of 1 over a range of scan rates**

**1** undergoes oxidation at  $E_{1/2} = -0.35$  V in a process that was found to be reversible ( $i_p^{ox}/i_p^{red}$  ca. 1) over all scan rates studied ( $0.01 - 10$  V s<sup>-1</sup>).



*Fig ESI 1: CV of 0.5 mM complex 1 in 0.1 M TBAPF<sub>6</sub>/ CH<sub>2</sub>Cl<sub>2</sub> over range of scan rate. Left: 0.01 V s<sup>-1</sup> (black); 0.02 V s<sup>-1</sup> (brown); 0.05 V s<sup>-1</sup> (dark green); 0.1 V s<sup>-1</sup> (light green). Right: 0.1 V s<sup>-1</sup> (black); 1 V s<sup>-1</sup> (brown); 5 V s<sup>-1</sup> (dark green); 10 V s<sup>-1</sup> (light green).*

## (ii) CVs of oxidation of **3** over a range of scan rates

Oxidation of **3** takes place reversibly at the same potential as for **1** ( $E_{1/2} = -0.35$  V), however a second oxidation couple of similar peak height is observed at  $E_{1/2} = -0.11$  V, suggestive of a further reversible one electron oxidation. The scan rate dependence of the two couples shows the relative peak heights and separations of the two processes do not change significantly with scan rate. The behaviour observed may be attributed to consecutive loss of two electrons from the same orbital; thus the first couple is assigned to the  $[\text{FeFe}]^{0/+}$  process and the second to  $[\text{FeFe}]^{+/2+}$ .

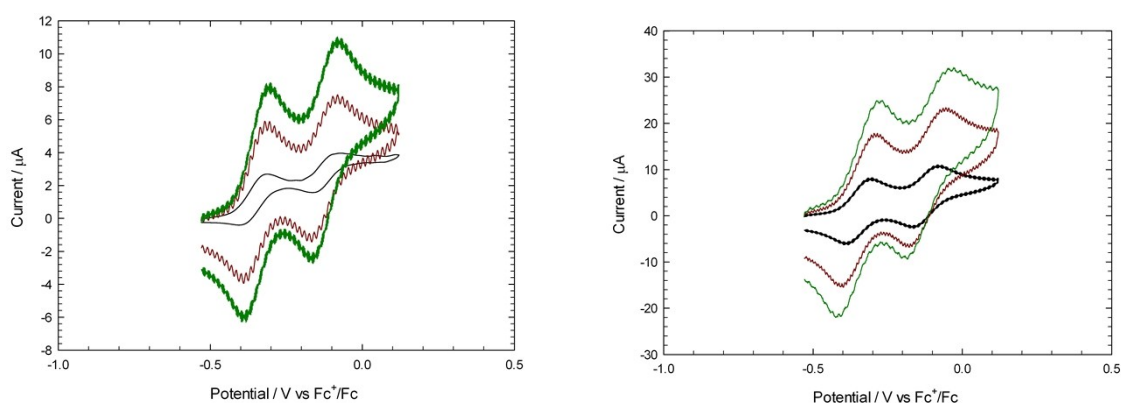


Fig ESI 2: CV of 0.5 mM complex **3** in 0.1 M TBAPF<sub>6</sub>/CH<sub>2</sub>Cl<sub>2</sub> over range of scan rate. Left: 0.01 V s<sup>-1</sup> (black); 0.02 V s<sup>-1</sup> (brown); 0.05 V s<sup>-1</sup> (dark green); 0.1 V s<sup>-1</sup> (light green). Right: 0.1 V s<sup>-1</sup> (black); 1 V s<sup>-1</sup> (brown); 5 V s<sup>-1</sup> (dark green); 10 V s<sup>-1</sup> (light green).

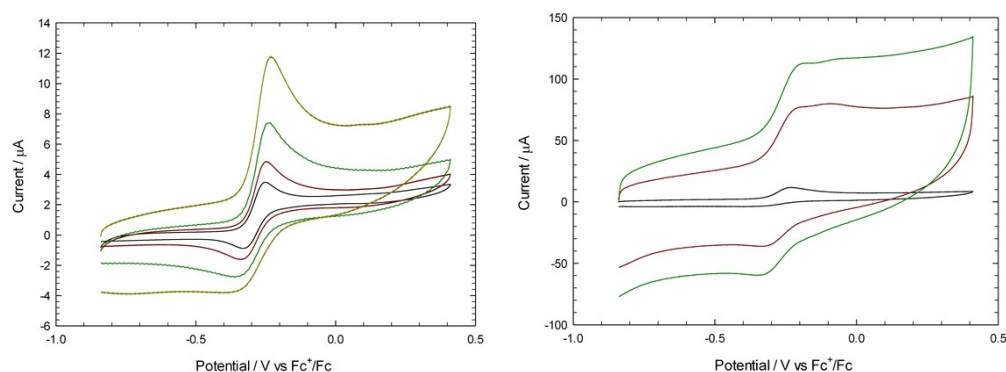


Fig ESI 3: CV of 0.5 mM complex **3** in 0.1 M TBAPF<sub>6</sub>/CH<sub>3</sub>CN over range of scan rate. Left: 0.01 V s<sup>-1</sup> (black); 0.02 V s<sup>-1</sup> (brown); 0.05 V s<sup>-1</sup> (dark green); 0.1 V s<sup>-1</sup> (light green). Right: 0.1 V s<sup>-1</sup> (black); 1 V s<sup>-1</sup> (brown); 5 V s<sup>-1</sup> (dark green); 10 V s<sup>-1</sup> (light green).

We find that in CH<sub>3</sub>CN, **3** instead undergoes a single, irreversible two-electron oxidation at slow scans, which resolves into two consecutive one electron processes as scan rate is increased. This is consistent with further stabilisation of the **3**<sup>2+</sup> species via formation of an adduct  $[\mathbf{3}(\text{CH}_3\text{CN})]^{2+}$  with the coordinating solvent, as isolated

and characterised for similar compounds by Olsen *et al.* (M.T. Olsen, T.B. Rauchfuss, S.R. Wilson, J. Am. Chem. Soc. 132 (2010) 17733-17740.)

### (iii) Electrocatalytic performance of **4** on addition of strong acid

The performance of open-bridged **4** was investigated in the presence of  $\text{HBF}_4 \cdot \text{Et}_2\text{O}$  in electrolytes composed of  $\text{TBAPF}_6$ ,  $\text{TBABF}_4$  and  $\text{TBAClO}_4$ . Broadly the same features can be identified as for **1** and hence the reaction mechanism shown in Fig 4 is proposed to also apply for **4**. However it was found that the identity of supporting anion influenced the relative currents observed at potentials **A**, **B** and **C** in a different manner than for **1**.

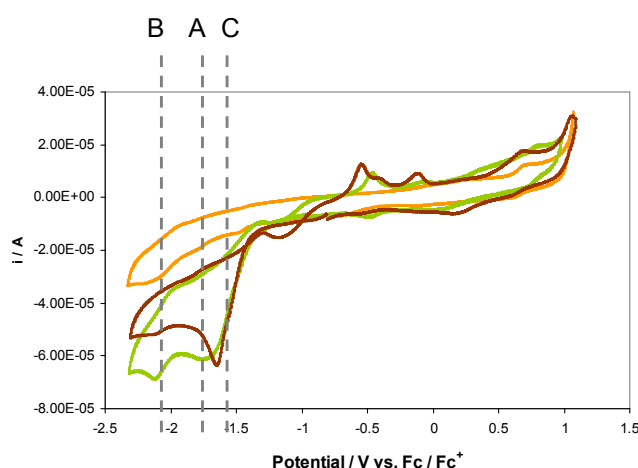
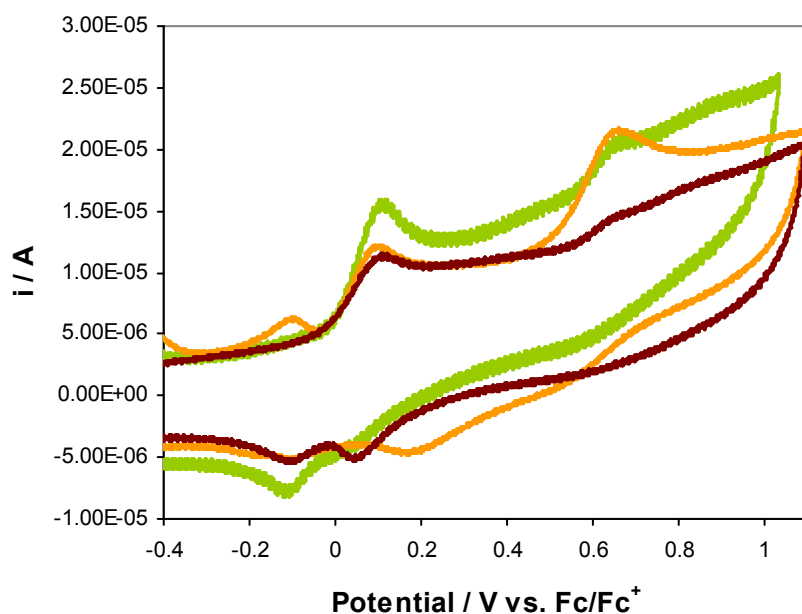


Fig EC4: CV of 0.5 mM complex **4** with 10 molar equivalent of  $\text{HBF}_4 \cdot \text{Et}_2\text{O}$  (scan rate  $0.1 \text{ V s}^{-1}$ ) in 0.1 M  $\text{TBAPF}_6 / \text{CH}_2\text{Cl}_2$  (orange); 0.1 M  $\text{TBABF}_4 / \text{CH}_2\text{Cl}_2$  (green) in 0.1 M  $\text{TBAClO}_4 / \text{CH}_2\text{Cl}_2$  (brown). A, B and C refer to catalytic processes shown in Figure 4 in the main manuscript.

In  $\text{TBAPF}_6$  the currents were all very inhibited compared to those for **1**, reaching a maximum after only 4 equivalents of acid were introduced. This indicates that for **4**,  $\text{PF}_6^-$  plays a retarding role in intramolecular proton transfer leading to  $\text{H}_2$  elimination. This may be related to the open and flexible nature of the unlinked bridging groups, which may allow closer approach of the anion to the iron centres, where it could occupy a position that blocks the transfer. In the presence of  $\text{BF}_4^-$  or  $\text{ClO}_4^-$ , currents at potential C attributed to reduction of  $4(\mu\text{H})\text{H}^+$  are enhanced in comparison to those observed for  $1(\mu\text{H})\text{H}^+$  (Fig 4). This indicates that for **4** these anions facilitate more efficient production of the intermediate  $4\text{H}$  complex after  $\text{H}_2$  elimination from  $4\text{H}(\mu\text{H})\text{H}$ , suggesting  $\text{ClO}_4^-$  and  $\text{BF}_4^-$  may promote the intramolecular proton transfer steps necessary to form  $4(\mu\text{H})\text{-H}_2$ . That this process is accelerated more for **4** than **1**, suggests that optimal positioning of the anions with respect to the pdt or  $(\text{SMe})_2$  bridging groups is important.

**(iv) Oxidation response for 3 on addition of strong acid**



*Fig EC5: CV of oxidation response of 0.5 mM complex 3 with 10 molar equivalent of  $\text{HBF}_4\text{-Et}_2\text{O}$  (scan rate  $0.1 \text{ V s}^{-1}$ ) in 0.1 M  $\text{TBAPF}_6 / \text{CH}_2\text{Cl}_2$  (brown); 0.1 M  $\text{TBABF}_4 / \text{CH}_2\text{Cl}_2$  (green) in 0.1 M  $\text{TBAClO}_4 / \text{CH}_2\text{Cl}_2$  (orange).*

In excess acid the oxidation peak for protonated **3** is still observed at 0.15 V, the same potential as noted for  $3\text{H}_\text{N}^+$  in the presence of  $\text{TBAPF}_6$  and  $\text{TBABF}_4$ . There is only a very small increase in current at 0.75 V, where  $3\text{H}_\text{Fe}^+$  would be expected to undergo oxidation, indicating that this species is not present in solution at high concentration even in excess acid. In  $\text{TBAClO}_4$  peaks are observed at +0.15 V and at +0.75 V suggesting a mixture of species protonated at nitrogen and iron are present in solution.

## (v) Computational Details

B3LYP geometries and energies for all optimized structures

### Species A

HF energy= -3841.81363212

No Imaginary frequency

Zero-point correction = 0.692797 (Hartree/Particle)

Thermal correction to Energy = 0.742326

Thermal correction to Enthalpy = 0.743270

Thermal correction to Gibbs Free Energy = 0.607836

Sum of electronic and zero-point Energies = -3841.120835

Sum of electronic and thermal Energies = -3841.071306

Sum of electronic and thermal Enthalpies = -3841.070362

Sum of electronic and thermal Free Energies = -3841.205796

### Coordinates: A

Fe	-0.03090000	6.09640000	7.02150000
Fe	-0.61120000	7.55000000	5.03320000
P	-0.47060000	5.00660000	8.94120000
P	0.39550000	4.12660000	6.03790000
P	-0.61130000	6.29820000	3.12620000
S	-2.12160000	6.05060000	5.99580000
S	-0.41340000	8.37390000	7.21810000
O	2.76240000	6.67270000	7.66860000
O	2.22510000	8.17770000	4.66390000
O	-1.83000000	9.93080000	3.83990000
C	1.66370000	6.42570000	7.39900000
C	1.10550000	7.91650000	4.82160000
C	-1.39850000	8.95640000	4.29510000
C	-3.46870000	6.91730000	6.92720000
H	-4.38570000	6.72460000	6.35570000
H	-3.57440000	6.41220000	7.89250000
C	-3.29090000	8.42000000	7.12330000
H	-3.25830000	8.92140000	6.14860000
H	-4.18440000	8.80340000	7.64370000
C	-2.06240000	8.81000000	7.94050000
H	-2.10710000	8.35190000	8.93690000
H	-2.04110000	9.89810000	8.07850000
C	-0.09480000	3.18580000	8.62080000
H	0.97090000	3.05020000	8.84830000
H	-0.66160000	2.53460000	9.29760000
C	-0.36320000	2.82570000	7.14800000
H	-1.43770000	2.86960000	6.92740000
H	-0.03490000	1.80350000	6.92360000
C	-0.37150000	3.70050000	4.38370000
H	-0.12730000	2.64970000	4.17460000
H	-1.45970000	3.77130000	4.49650000
C	0.12450000	4.57200000	3.21570000
H	-0.08500000	4.06240000	2.26730000

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C	0.53380000	5.38410000	10.44680000
C	1.06000000	6.67530000	10.61050000
H	0.89600000	7.42390000	9.83870000
C	1.79610000	7.00170000	11.75370000
H	2.19930000	8.00630000	11.86450000
C	2.02180000	6.04050000	12.74060000
H	2.60180000	6.29220000	13.62640000
C	1.50100000	4.75150000	12.58780000
H	1.67150000	3.99880000	13.35530000
C	0.75640000	4.42750000	11.45270000
H	0.34230000	3.42600000	11.35660000
C	-2.17340000	4.94540000	9.68060000
C	-3.13270000	4.01610000	9.24400000
H	-2.88190000	3.28480000	8.48020000
C	-4.42630000	4.01930000	9.77210000
H	-5.15140000	3.28720000	9.42160000
C	-4.78800000	4.95660000	10.74260000
H	-5.79580000	4.95990000	11.15270000
C	-3.84470000	5.88820000	11.18380000
H	-4.11270000	6.62000000	11.94350000
C	-2.55020000	5.88190000	10.65920000
H	-1.82360000	6.60320000	11.02600000
C	2.14170000	3.56950000	5.75890000
C	3.07750000	4.50010000	5.27050000
H	2.78410000	5.53440000	5.11710000
C	4.39250000	4.11720000	4.99430000
H	5.09800000	4.85660000	4.62120000
C	4.80110000	2.79960000	5.21110000
H	5.82770000	2.50300000	5.00550000
C	3.88470000	1.86530000	5.69900000
H	4.19330000	0.83640000	5.87430000
C	2.56700000	2.24570000	5.96510000
H	1.87540000	1.49420000	6.33690000
C	0.36210000	7.04550000	1.73600000
C	-0.26940000	7.93460000	0.84810000
H	-1.33770000	8.11770000	0.93180000
C	0.45760000	8.58530000	-0.15020000
H	-0.05220000	9.26930000	-0.82580000
C	1.83040000	8.36030000	-0.28110000
H	2.39720000	8.86580000	-1.06030000
C	2.47070000	7.48600000	0.59880000
H	3.54080000	7.30780000	0.51190000
C	1.74520000	6.83890000	1.60230000
H	2.27690000	6.18210000	2.28520000
C	-2.23380000	5.95820000	2.28500000
C	-3.42870000	6.49340000	2.78650000
H	-3.40370000	7.08940000	3.69230000
C	-4.64460000	6.26090000	2.13570000
H	-5.56070000	6.68650000	2.54060000



C	-4.68340000	5.48790000	0.97470000
H	-5.62960000	5.30540000	0.46880000
C	-3.49780000	4.95490000	0.45850000
H	-3.51600000	4.35980000	-0.45250000
C	-2.28490000	5.19190000	1.10590000
H	-1.37060000	4.79370000	0.67080000

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### Species TSAB

HF energy= -3841.79698330

One imaginary frequency (249i)

Zero-point correction = 0.692466 (Hartree/Particle)

Thermal correction to Energy = 0.741659

Thermal correction to Enthalpy = 0.742603

Thermal correction to Gibbs Free Energy = 0.606570

Sum of electronic and zero-point Energies = -3841.104518

Sum of electronic and thermal Energies = -3841.055324

Sum of electronic and thermal Enthalpies = -3841.054380

Sum of electronic and thermal Free Energies = -3841.190413

### Coordinates: TSAB

Fe	0.01220000	6.00850000	7.04890000
Fe	-0.58550000	7.48930000	5.05760000
P	-0.42530000	4.91230000	8.95240000
P	0.42970000	4.05160000	6.04370000
P	-0.59630000	6.24100000	3.14650000
S	-2.05270000	5.99680000	6.02630000
S	-0.35820000	8.26910000	7.22000000
O	2.81400000	6.59120000	7.65460000
O	2.22400000	8.21390000	4.62520000
O	-1.99290000	9.83340000	4.01080000
C	1.71180000	6.34040000	7.40230000
C	1.11530000	7.92050000	4.79540000
C	-1.46930000	8.87350000	4.39510000
C	-3.45530000	6.79180000	6.92000000
H	-4.20170000	6.95010000	6.13230000
H	-3.85300000	6.00260000	7.56490000
C	-3.30790000	8.09920000	7.74680000
H	-4.01770000	8.82240000	7.32650000
H	-3.68070000	7.87270000	8.75200000
C	-1.95990000	8.85140000	7.92400000
H	-1.75190000	8.96430000	8.99490000
H	-2.05870000	9.86660000	7.52350000
C	-0.05120000	3.09300000	8.62310000
H	1.01550000	2.95840000	8.84690000
H	-0.61370000	2.43580000	9.29770000
C	-0.32350000	2.74180000	7.14850000
H	-1.39910000	2.78700000	6.93240000
H	0.00430000	1.72120000	6.91620000

C	-0.34650000	3.63710000	4.39020000
H	-0.10340000	2.58740000	4.17400000
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H	-0.07120000	4.00880000	2.27410000
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C	1.49940000	4.63200000	12.62210000
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H	0.32110000	3.32830000	11.38590000
C	-2.13280000	4.85740000	9.68300000
C	-3.07980000	3.89800000	9.28690000
H	-2.82010000	3.13980000	8.55270000
C	-4.37190000	3.90280000	9.81980000
H	-5.08750000	3.14710000	9.50130000
C	-4.74300000	4.87070000	10.75610000
H	-5.74850000	4.87410000	11.17180000
C	-3.81130000	5.83210000	11.15740000
H	-4.08640000	6.58700000	11.89150000
C	-2.51970000	5.82550000	10.62600000
H	-1.80220000	6.57190000	10.95950000
C	2.17410000	3.49850000	5.74940000
C	3.10200000	4.43330000	5.25400000
H	2.80150000	5.46650000	5.10580000
C	4.41580000	4.05570000	4.96560000
H	5.11540000	4.79800000	4.58710000
C	4.83100000	2.73930000	5.17740000
H	5.85680000	2.44670000	4.96230000
C	3.92260000	1.80110000	5.67280000
H	4.23670000	0.77330000	5.84450000
C	2.60570000	2.17620000	5.95080000
H	1.92010000	1.42220000	6.32830000
C	0.35870000	6.99380000	1.74670000
C	-0.28310000	7.89830000	0.88180000
H	-1.34820000	8.08850000	0.98940000
C	0.42930000	8.55490000	-0.12300000
H	-0.08830000	9.25050000	-0.78070000
C	1.79750000	8.32090000	-0.28350000
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C	2.44800000	7.43150000	0.57350000
H	3.51470000	7.24620000	0.46330000
C	1.73700000	6.77810000	1.58340000
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C	-2.22500000	5.89300000	2.31910000
C	-3.41960000	6.40940000	2.84030000
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C	-4.64100000	6.16930000	2.20280000
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C	-4.68550000	5.40720000	1.03480000
H	-5.63590000	5.21900000	0.53880000
C	-3.50050000	4.89280000	0.49890000
H	-3.52340000	4.30640000	-0.41780000
C	-2.28230000	5.13720000	1.13360000
H	-1.36890000	4.75330000	0.68380000

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### Species B

HF energy= -3841.81232981

No Imaginary frequency

Zero-point correction = 0.692653 (Hartree/Particle)

Thermal correction to Energy = 0.742257

Thermal correction to Enthalpy = 0.743201

Thermal correction to Gibbs Free Energy = 0.607063

Sum of electronic and zero-point Energies = -3841.119676

Sum of electronic and thermal Energies = -3841.070073

Sum of electronic and thermal Enthalpies = -3841.069128

Sum of electronic and thermal Free Energies = -3841.205266

### Coordinates: B

Fe	-0.04420000	6.04880000	7.09410000
Fe	-0.54890000	7.48440000	5.06240000
P	-0.46940000	4.95730000	9.02780000
P	0.36120000	4.07330000	6.10870000
P	-0.62500000	6.19130000	3.17720000
S	-2.12970000	6.10270000	6.05700000
S	-0.29280000	8.36800000	7.19730000
O	2.75250000	6.55680000	7.77880000
O	2.28320000	8.10870000	4.63040000
O	-1.82380000	9.86580000	3.93370000
C	1.65330000	6.33460000	7.49050000
C	1.16600000	7.84730000	4.79740000
C	-1.35680000	8.88960000	4.34780000
C	-3.42560000	7.09170000	6.94120000
H	-3.89420000	7.73170000	6.18170000
H	-4.18000000	6.37000000	7.27520000
C	-2.96540000	7.92900000	8.13150000
H	-3.85290000	8.42780000	8.55560000
H	-2.57690000	7.26790000	8.91410000
C	-1.92470000	8.99400000	7.80420000
H	-1.69930000	9.57900000	8.70480000
H	-2.30120000	9.69340000	7.04670000
C	0.06370000	3.17140000	8.72680000

H	1.15060000	3.13770000	8.88210000
H	-0.40160000	2.49010000	9.44950000
C	-0.27030000	2.75760000	7.28300000
H	-1.35740000	2.72270000	7.13630000
H	0.10670000	1.75190000	7.06380000
C	-0.49610000	3.61520000	4.50910000
H	-0.30370000	2.54770000	4.33140000
H	-1.57490000	3.73640000	4.66410000
C	-0.01440000	4.41710000	3.28640000
H	-0.31980000	3.89700000	2.37040000
H	1.08030000	4.44350000	3.27500000
C	0.50500000	5.43750000	10.51990000
C	0.85060000	6.78820000	10.68980000
H	0.56100000	7.51410000	9.93300000
C	1.57920000	7.20020000	11.80880000
H	1.84250000	8.24980000	11.92220000
C	1.98080000	6.26730000	12.76720000
H	2.55610000	6.58700000	13.63390000
C	1.64690000	4.91920000	12.60670000
H	1.95890000	4.18660000	13.34870000
C	0.91010000	4.50770000	11.49410000
H	0.64740000	3.45700000	11.39270000
C	-2.17530000	4.70040000	9.73110000
C	-3.19240000	4.18230000	8.90680000
H	-2.99620000	3.99170000	7.85520000
C	-4.47200000	3.94260000	9.41100000
H	-5.23960000	3.53580000	8.75510000
C	-4.77020000	4.23540000	10.74480000
H	-5.76990000	4.05630000	11.13560000
C	-3.77680000	4.76750000	11.56800000
H	-3.99710000	5.00730000	12.60660000
C	-2.49030000	4.99380000	11.06850000
H	-1.73090000	5.40040000	11.72990000
C	2.10160000	3.55590000	5.72120000
C	3.01030000	4.52100000	5.25010000
H	2.70320000	5.55850000	5.16710000
C	4.31450000	4.16640000	4.89500000
H	4.99880000	4.93410000	4.53970000
C	4.73890000	2.84110000	5.00930000
H	5.75710000	2.56600000	4.74150000
C	3.84760000	1.87060000	5.47260000
H	4.16670000	0.83420000	5.56550000
C	2.54130000	2.22400000	5.81960000
H	1.87020000	1.44300000	6.16730000
C	0.40120000	6.84760000	1.77830000
C	-0.16830000	7.75180000	0.86410000
H	-1.22440000	8.00020000	0.93300000
C	0.60480000	8.33420000	-0.14190000
H	0.14230000	9.03070000	-0.83840000
C	1.96280000	8.02530000	-0.25380000

H	2.56520000	8.47730000	-1.03910000
C	2.54250000	7.13690000	0.65360000
H	3.60100000	6.89450000	0.58220000
C	1.77120000	6.55800000	1.66450000
H	2.25790000	5.88870000	2.36910000
C	-2.26470000	5.94690000	2.33400000
C	-3.43140000	6.53550000	2.84090000
H	-3.37660000	7.12440000	3.74970000
C	-4.65920000	6.36340000	2.19340000
H	-5.55290000	6.82930000	2.60380000
C	-4.73800000	5.59880000	1.02890000
H	-5.69340000	5.46350000	0.52540000
C	-3.58040000	5.01290000	0.50660000
H	-3.62930000	4.42380000	-0.40730000
C	-2.35610000	5.18860000	1.15180000
H	-1.46270000	4.74880000	0.71290000

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### Species A\_alt

HF energy= -3841.79355711

No imaginary frequency

Zero-point correction = 0.692796 (Hartree/Particle)

Thermal correction to Energy = 0.742328

Thermal correction to Enthalpy = 0.743272

Thermal correction to Gibbs Free Energy = 0.608161

Sum of electronic and zero-point Energies = -3841.100761

Sum of electronic and thermal Energies = -3841.051229

Sum of electronic and thermal Enthalpies = -3841.050285

Sum of electronic and thermal Free Energies = -3841.185396

### Coordinates: A\_alt

Fe	1.29340000	0.45760000	0.21000000
Fe	-1.08710000	-0.01900000	0.88080000
P	3.33840000	1.36900000	0.19730000
P	0.82930000	1.56940000	-1.66570000
P	-1.47460000	-1.29350000	-0.96050000
S	0.75000000	-1.24560000	1.68480000
S	0.12260000	1.80650000	1.67990000
O	2.31470000	-1.78120000	-1.35690000
O	-3.01920000	-1.07030000	2.81590000
O	-2.88480000	2.03720000	-0.18340000
C	1.87820000	-0.87940000	-0.76460000
C	-2.26070000	-0.64820000	2.04860000
C	-2.17010000	1.21310000	0.20080000
C	3.54270000	2.12370000	-1.53770000
H	3.41510000	3.20790000	-1.43520000
H	4.55950000	1.95380000	-1.90970000
C	2.49740000	1.57650000	-2.53470000
H	2.73600000	0.54330000	-2.81590000

H	2.49720000	2.17910000	-3.45140000
C	-0.39490000	0.71650000	-2.81390000
H	-1.32370000	1.28890000	-2.70510000
H	-0.06460000	0.86090000	-3.85290000
C	-0.65510000	-0.78470000	-2.58570000
H	0.28070000	-1.34670000	-2.64220000
H	-1.28270000	-1.14960000	-3.40920000
C	1.07910000	-0.90470000	3.47590000
H	0.78890000	-1.82040000	4.00640000
H	2.16860000	-0.80240000	3.57330000
C	0.71060000	1.62650000	3.42560000
H	1.79360000	1.80430000	3.41960000
H	0.24300000	2.45090000	3.97720000
C	3.77730000	2.78010000	1.31790000
C	4.65390000	2.60540000	2.40220000
H	5.14640000	1.64870000	2.55410000
C	4.91220000	3.65510000	3.28860000
H	5.59710000	3.49870000	4.11990000
C	4.30150000	4.89730000	3.10640000
H	4.50620000	5.71530000	3.79410000
C	3.42180000	5.08080000	2.03580000
H	2.93480000	6.04260000	1.88640000
C	3.15570000	4.03110000	1.15540000
H	2.44690000	4.19090000	0.34670000
C	4.82350000	0.27510000	0.36930000
C	4.67680000	-1.00070000	0.93500000
H	3.69070000	-1.33880000	1.24500000
C	5.78350000	-1.84150000	1.09220000
H	5.65010000	-2.82970000	1.52790000
C	7.04930000	-1.41860000	0.68410000
H	7.90960000	-2.07490000	0.80110000
C	7.20970000	-0.14660000	0.12440000
H	8.19500000	0.19200000	-0.19060000
C	6.10670000	0.69430000	-0.02730000
H	6.25330000	1.69020000	-0.44130000
C	0.30910000	3.33360000	-1.94020000
C	-0.09140000	4.16220000	-0.88060000
H	-0.10170000	3.77340000	0.13320000
C	-0.48830000	5.48290000	-1.11980000
H	-0.79860000	6.10680000	-0.28390000
C	-0.49590000	5.99420000	-2.41780000
H	-0.80680000	7.02080000	-2.60190000
C	-0.10560000	5.17750000	-3.48370000
H	-0.11150000	5.56460000	-4.50100000
C	0.29210000	3.86140000	-3.24620000
H	0.58730000	3.24410000	-4.09270000
C	-3.27960000	-1.29610000	-1.42390000
C	-4.17550000	-2.00740000	-0.60580000
H	-3.79940000	-2.57610000	0.24140000
C	-5.54500000	-2.00310000	-0.86910000

H	-6.21750000	-2.56630000	-0.22500000
C	-6.05110000	-1.27810000	-1.95240000
H	-7.12000000	-1.27090000	-2.15620000
C	-5.17500000	-0.56220000	-2.76830000
H	-5.55620000	0.00940000	-3.61250000
C	-3.80080000	-0.57160000	-2.50700000
H	-3.14670000	0.00090000	-3.15910000
C	-1.13670000	-3.11810000	-0.93860000
C	-0.95880000	-3.86010000	-2.11960000
H	-0.97550000	-3.36830000	-3.08850000
C	-0.76710000	-5.24180000	-2.07180000
H	-0.62710000	-5.79770000	-2.99700000
C	-0.76010000	-5.90810000	-0.84320000
H	-0.61140000	-6.98560000	-0.80680000
C	-0.94520000	-5.18370000	0.33530000
H	-0.93880000	-5.69210000	1.29750000
C	-1.12780000	-3.79870000	0.28890000
H	-1.24840000	-3.24120000	1.21200000
C	0.37770000	0.29960000	4.09720000
H	0.67640000	0.35880000	5.15690000
H	-0.70810000	0.14670000	4.08600000

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### Species C

HF energy= -3841.81352471

No Imaginary frequency

Zero-point correction = 0.692461 (Hartree/Particle)

Thermal correction to Energy = 0.742150

Thermal correction to Enthalpy = 0.743094

Thermal correction to Gibbs Free Energy = 0.607012

Sum of electronic and zero-point Energies = -3841.121064

Sum of electronic and thermal Energies = -3841.071375

Sum of electronic and thermal Enthalpies = -3841.070431

Sum of electronic and thermal Free Energies = -3841.206512

### Coordinates: C

Fe	1.02150000	0.38860000	0.75410000
Fe	-1.25560000	-0.77920000	1.00170000
P	2.59140000	-0.90400000	-0.39070000
P	0.72220000	1.47340000	-1.22520000
P	-2.79780000	0.07000000	-0.42730000
S	0.62240000	-1.26850000	2.28290000
S	-0.82800000	1.42070000	1.64880000
O	2.83640000	2.17970000	2.15640000
O	-0.41980000	-2.58570000	-1.13550000
O	-3.08930000	-2.42860000	2.57160000
C	2.16960000	1.44760000	1.55130000
C	-0.69280000	-1.85780000	-0.26590000
C	-2.39710000	-1.71580000	1.97070000

C	2.24210000	-0.65060000	-2.20150000
H	3.03630000	-1.04040000	-2.84950000
H	1.33830000	-1.23600000	-2.40920000
C	1.98440000	0.84150000	-2.45830000
H	1.64870000	1.00670000	-3.49060000
H	2.90240000	1.42490000	-2.31310000
C	-0.84560000	1.17940000	-2.21510000
H	-0.77970000	1.80970000	-3.11300000
H	-0.81040000	0.13580000	-2.55420000
C	-2.16790000	1.47830000	-1.48030000
H	-2.94760000	1.76030000	-2.19920000
H	-2.05870000	2.33360000	-0.80290000
C	0.47480000	-0.61330000	4.00950000
H	0.39300000	-1.49860000	4.65220000
H	1.43140000	-0.12530000	4.24040000
C	-0.66080000	1.62650000	3.48310000
H	0.26680000	2.18390000	3.66610000
H	-1.49570000	2.27260000	3.78220000
C	4.36260000	-0.36830000	-0.19680000
C	4.87600000	-0.20980000	1.10220000
H	4.22860000	-0.36430000	1.96150000
C	6.20790000	0.15200000	1.30910000
H	6.58220000	0.26630000	2.32450000
C	7.05150000	0.37890000	0.21880000
H	8.08750000	0.67040000	0.37860000
C	6.55420000	0.23260000	-1.07720000
H	7.20160000	0.40670000	-1.93470000
C	5.22380000	-0.14410000	-1.28290000
H	4.87080000	-0.26330000	-2.30370000
C	2.78940000	-2.73860000	-0.22430000
C	3.15140000	-3.27500000	1.02370000
H	3.27850000	-2.62130000	1.88090000
C	3.33860000	-4.64860000	1.18260000
H	3.61580000	-5.04290000	2.15820000
C	3.16210000	-5.51250000	0.09890000
H	3.30420000	-6.58410000	0.22380000
C	2.79780000	-4.99220000	-1.14390000
H	2.65290000	-5.65530000	-1.99460000
C	2.61270000	-3.61670000	-1.30510000
H	2.32030000	-3.24220000	-2.28100000
C	0.89700000	3.31030000	-1.38250000
C	0.86780000	4.12630000	-0.24130000
H	0.73240000	3.67460000	0.73630000
C	1.00230000	5.51360000	-0.35420000
H	0.97620000	6.13010000	0.54210000
C	1.17120000	6.10430000	-1.60740000
H	1.27900000	7.18370000	-1.69420000
C	1.20240000	5.30270000	-2.75280000
H	1.33610000	5.75490000	-3.73380000
C	1.06290000	3.91880000	-2.64050000



H	1.09070000	3.31280000	-3.54440000
C	-4.34070000	0.83230000	0.26640000
C	-4.45700000	1.07890000	1.64180000
H	-3.63980000	0.80420000	2.30140000
C	-5.61100000	1.67160000	2.16250000
H	-5.68690000	1.85270000	3.23300000
C	-6.66240000	2.02570000	1.31500000
H	-7.56170000	2.48490000	1.72100000
C	-6.55870000	1.78080000	-0.05770000
H	-7.37720000	2.04630000	-0.72430000
C	-5.40800000	1.18620000	-0.57730000
H	-5.35050000	0.98040000	-1.64420000
C	-3.50550000	-1.12150000	-1.65870000
C	-3.55870000	-0.88500000	-3.04150000
H	-3.16820000	0.03810000	-3.46160000
C	-4.11210000	-1.83290000	-3.90880000
H	-4.14000000	-1.63120000	-4.97810000
C	-4.62560000	-3.02920000	-3.40680000
H	-5.05590000	-3.76630000	-4.08180000
C	-4.57990000	-3.27610000	-2.03150000
H	-4.97410000	-4.20690000	-1.62880000
C	-4.02030000	-2.33500000	-1.16750000
H	-3.98010000	-2.54710000	-0.10210000
C	-0.68850000	0.33390000	4.29260000
H	-0.66500000	0.59480000	5.36370000
H	-1.63680000	-0.18880000	4.11620000

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### Species TSCD

HF energy= -3841.79675387

One imaginary frequency (*247i*)

Zero-point correction = 0.692139 (Hartree/Particle)

Thermal correction to Energy = 0.741506

Thermal correction to Enthalpy = 0.742451

Thermal correction to Gibbs Free Energy = 0.605874

Sum of electronic and zero-point Energies = -3841.104615

Sum of electronic and thermal Energies = -3841.055248

Sum of electronic and thermal Enthalpies = -3841.054303

Sum of electronic and thermal Free Energies = -3841.190880

### Coordinates: TSCD

Fe	1.07880000	0.40800000	0.70680000
Fe	-1.19920000	-0.79400000	0.96480000
P	2.63040000	-0.89310000	-0.44830000
P	0.74840000	1.47390000	-1.27250000
P	-2.75550000	0.06460000	-0.44600000
S	0.67270000	-1.26460000	2.20600000
S	-0.76750000	1.37920000	1.61870000
O	2.90440000	2.23640000	2.04780000

O	-0.42460000	-2.66970000	-1.14170000
O	-2.99310000	-2.35650000	2.66510000
C	2.23020000	1.48730000	1.47200000
C	-0.69010000	-1.91980000	-0.29080000
C	-2.32040000	-1.68370000	1.99990000
C	2.26080000	-0.64860000	-2.25650000
H	3.04760000	-1.04130000	-2.91180000
H	1.35440000	-1.23460000	-2.45240000
C	2.00040000	0.84250000	-2.51590000
H	1.65580000	1.00420000	-3.54590000
H	2.91980000	1.42600000	-2.37990000
C	-0.82580000	1.17830000	-2.25210000
H	-0.76510000	1.81120000	-3.14860000
H	-0.79140000	0.13540000	-2.59380000
C	-2.14300000	1.47400000	-1.50810000
H	-2.92980000	1.75220000	-2.22080000
H	-2.03070000	2.33110000	-0.83320000
C	0.35580000	-0.80980000	3.96660000
H	-0.39700000	-1.53940000	4.28350000
H	1.28240000	-1.07550000	4.48930000
C	-0.49880000	1.71790000	3.41320000
H	0.24510000	2.52200000	3.42340000
H	-1.44150000	2.16500000	3.75020000
C	4.40440000	-0.36000000	-0.27520000
C	4.92580000	-0.18510000	1.01850000
H	4.28200000	-0.32410000	1.88340000
C	6.26020000	0.17440000	1.21250000
H	6.64110000	0.30180000	2.22390000
C	7.09800000	0.38300000	0.11410000
H	8.13610000	0.67290000	0.26360000
C	6.59250000	0.22050000	-1.17700000
H	7.23550000	0.38060000	-2.04040000
C	5.25960000	-0.15430000	-1.36960000
H	4.89990000	-0.28700000	-2.38640000
C	2.82950000	-2.72730000	-0.27710000
C	3.21020000	-3.25780000	0.96790000
H	3.35270000	-2.59930000	1.81910000
C	3.39900000	-4.63050000	1.13100000
H	3.69110000	-5.02020000	2.10410000
C	3.20560000	-5.49980000	0.05440000
H	3.34920000	-6.57090000	0.18240000
C	2.82300000	-4.98550000	-1.18540000
H	2.66510000	-5.65280000	-2.03050000
C	2.63620000	-3.61070000	-1.35070000
H	2.32980000	-3.24110000	-2.32420000
C	0.91320000	3.31190000	-1.43390000
C	0.85110000	4.13060000	-0.29600000
H	0.70180000	3.67920000	0.67980000
C	0.97200000	5.51900000	-0.40950000
H	0.92060000	6.13740000	0.48440000

C	1.16070000	6.10840000	-1.66050000
H	1.25870000	7.18870000	-1.74780000
C	1.22450000	5.30430000	-2.80270000
H	1.37360000	5.75560000	-3.78190000
C	1.09780000	3.91920000	-2.68970000
H	1.14980000	3.31160000	-3.59140000
C	-4.28700000	0.83080000	0.27050000
C	-4.37690000	1.09130000	1.64550000
H	-3.54670000	0.82430000	2.29220000
C	-5.52160000	1.68770000	2.18230000
H	-5.57700000	1.87950000	3.25220000
C	-6.58990000	2.03170000	1.35200000
H	-7.48200000	2.49350000	1.77080000
C	-6.51240000	1.77330000	-0.01990000
H	-7.34400000	2.03110000	-0.67310000
C	-5.37100000	1.17530000	-0.55570000
H	-5.33330000	0.95940000	-1.62150000
C	-3.48530000	-1.12570000	-1.66480000
C	-3.54690000	-0.89860000	-3.04860000
H	-3.15130000	0.01800000	-3.47820000
C	-4.11600000	-1.84770000	-3.90470000
H	-4.15020000	-1.65350000	-4.97520000
C	-4.63710000	-3.03510000	-3.38980000
H	-5.07960000	-3.77300000	-4.05600000
C	-4.58330000	-3.27250000	-2.01300000
H	-4.98350000	-4.19640000	-1.60060000
C	-4.00800000	-2.33050000	-1.16050000
H	-3.96090000	-2.53390000	-0.09340000
C	-0.07750000	0.61170000	4.41660000
H	0.74240000	1.03150000	5.01260000
H	-0.90740000	0.47370000	5.12100000

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### Species D

HF energy= -3841.81241277

No Imaginary frequency

Zero-point correction = 0.692432 (Hartree/Particle)

Thermal correction to Energy = 0.742135

Thermal correction to Enthalpy = 0.743079

Thermal correction to Gibbs Free Energy = 0.606374

Sum of electronic and zero-point Energies = -3841.119980

Sum of electronic and thermal Energies = -3841.070278

Sum of electronic and thermal Enthalpies = -3841.069334

Sum of electronic and thermal Free Energies = -3841.206038

### Coordinates: D

Fe	1.09250000	0.43930000	0.72240000
Fe	-1.17800000	-0.74760000	0.99560000
P	2.61810000	-0.88650000	-0.43870000

P	0.74520000	1.49550000	-1.25950000
P	-2.73570000	0.08510000	-0.43070000
S	0.70230000	-1.24130000	2.24620000
S	-0.76420000	1.43590000	1.65040000
O	3.00280000	2.31150000	1.87910000
O	-0.45710000	-2.73330000	-1.03060000
O	-2.98830000	-2.22130000	2.75690000
C	2.28420000	1.53750000	1.39860000
C	-0.70480000	-1.94120000	-0.21500000
C	-2.30920000	-1.58300000	2.06510000
C	2.24190000	-0.63720000	-2.24430000
H	3.02170000	-1.03650000	-2.90390000
H	1.32890000	-1.21420000	-2.43730000
C	1.99460000	0.85640000	-2.50090000
H	1.65170000	1.02160000	-3.53080000
H	2.91910000	1.43160000	-2.36410000
C	-0.82580000	1.21310000	-2.24730000
H	-0.76110000	1.86180000	-3.13220000
H	-0.79090000	0.17630000	-2.60660000
C	-2.14340000	1.49580000	-1.50170000
H	-2.93570000	1.76000000	-2.21360000
H	-2.04010000	2.35800000	-0.83180000
C	0.45700000	-0.65310000	3.98490000
H	-0.48490000	-1.09130000	4.33820000
H	1.26760000	-1.10970000	4.56700000
C	-0.66100000	1.61100000	3.49410000
H	-0.56640000	2.68600000	3.69220000
H	-1.63530000	1.28920000	3.88510000
C	4.40040000	-0.38000000	-0.27330000
C	4.92920000	-0.21760000	1.01900000
H	4.28560000	-0.34750000	1.88560000
C	6.27020000	0.11870000	1.20890000
H	6.65710000	0.23670000	2.21910000
C	7.10700000	0.31660000	0.10770000
H	8.15040000	0.58840000	0.25410000
C	6.59390000	0.16710000	-1.18190000
H	7.23630000	0.31930000	-2.04720000
C	5.25430000	-0.18500000	-1.37050000
H	4.88850000	-0.30840000	-2.38630000
C	2.79640000	-2.72350000	-0.27150000
C	3.17470000	-3.26280000	0.97040000
H	3.32270000	-2.60990000	1.82470000
C	3.35290000	-4.63780000	1.12660000
H	3.64260000	-5.03430000	2.09770000
C	3.15230000	-5.50040000	0.04610000
H	3.28760000	-6.57320000	0.16890000
C	2.77360000	-4.97720000	-1.19110000
H	2.61050000	-5.63890000	-2.03950000
C	2.59710000	-3.60020000	-1.34940000
H	2.29370000	-3.22430000	-2.32150000

C	0.91830000	3.33310000	-1.41940000
C	0.79230000	4.15570000	-0.28940000
H	0.59220000	3.70650000	0.67860000
C	0.91260000	5.54420000	-0.40250000
H	0.81110000	6.16580000	0.48500000
C	1.16490000	6.12980000	-1.64400000
H	1.26330000	7.21010000	-1.73020000
C	1.29180000	5.32170000	-2.77800000
H	1.49010000	5.76990000	-3.74990000
C	1.16500000	3.93640000	-2.66650000
H	1.26500000	3.32620000	-3.56220000
C	-4.28160000	0.84020000	0.26930000
C	-4.37770000	1.13270000	1.63730000
H	-3.54460000	0.89630000	2.29120000
C	-5.53200000	1.72370000	2.15920000
H	-5.59080000	1.94070000	3.22410000
C	-6.60490000	2.03060000	1.32050000
H	-7.50460000	2.48820000	1.72760000
C	-6.52160000	1.74090000	-0.04470000
H	-7.35630000	1.96980000	-0.70470000
C	-5.37020000	1.14840000	-0.56530000
H	-5.32840000	0.90880000	-1.62580000
C	-3.45050000	-1.12230000	-1.64220000
C	-3.51000000	-0.90830000	-3.02820000
H	-3.12220000	0.00820000	-3.46480000
C	-4.06850000	-1.86980000	-3.87750000
H	-4.10140000	-1.68530000	-4.94970000
C	-4.58120000	-3.05690000	-3.35350000
H	-5.01570000	-3.80430000	-4.01430000
C	-4.53000000	-3.28110000	-1.97440000
H	-4.92420000	-4.20450000	-1.55490000
C	-3.96530000	-2.32680000	-1.12880000
H	-3.92030000	-2.51980000	-0.05950000
C	0.46760000	0.85860000	4.19560000
H	1.43540000	1.26700000	3.88210000
H	0.38290000	1.05710000	5.27700000

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**Species C\_alt**

HF energy= -3841.79222866

No Imaginary frequency

Zero-point correction = 0.692219 (Hartree/Particle)

Thermal correction to Energy = 0.741957

Thermal correction to Enthalpy = 0.742902

Thermal correction to Gibbs Free Energy = 0.606399

Sum of electronic and zero-point Energies = -3841.100009

Sum of electronic and thermal Energies = -3841.050271

Sum of electronic and thermal Enthalpies = -3841.049327

Sum of electronic and thermal Free Energies = -3841.185830

Coordinates: C\_alt

Fe	1.30090000	0.44520000	0.52330000
Fe	-1.14180000	-0.43920000	0.89500000
P	2.93890000	-0.41140000	-0.95030000
P	0.64700000	1.65870000	-1.23010000
P	-1.43560000	-1.49060000	-1.07870000
S	0.81580000	-1.53230000	1.53330000
S	-0.21040000	1.35910000	1.99840000
O	3.27310000	2.03880000	1.95130000
O	-2.98870000	-1.89440000	2.63610000
O	-3.11090000	1.57100000	0.05990000
C	2.55300000	1.37130000	1.32840000
C	-2.25160000	-1.31480000	1.95400000
C	-2.32480000	0.77410000	0.34840000
C	2.91810000	0.64730000	-2.51490000
H	3.93570000	0.82120000	-2.88480000
H	2.40450000	0.04790000	-3.27630000
C	2.16280000	1.96360000	-2.29340000
H	1.89110000	2.43010000	-3.25050000
H	2.77790000	2.68480000	-1.74010000
C	-0.49730000	0.83520000	-2.48720000
H	-1.50810000	1.20480000	-2.28220000
H	-0.22890000	1.19680000	-3.49090000
C	-0.49600000	-0.71090000	-2.49840000
H	0.51890000	-1.11720000	-2.43430000
H	-0.90800000	-1.06020000	-3.45460000
C	1.04300000	-1.47030000	3.37430000
H	0.87500000	-2.49290000	3.73570000
H	2.10310000	-1.23400000	3.53230000
C	0.28730000	0.96120000	3.73890000
H	1.31830000	1.31550000	3.86240000
H	-0.35480000	1.58490000	4.37270000
C	4.65650000	-0.18440000	-0.29000000
C	5.21780000	-1.18900000	0.51660000
H	4.66800000	-2.11060000	0.69530000
C	6.48260000	-1.02290000	1.08260000
H	6.90290000	-1.81340000	1.70140000
C	7.20830000	0.14960000	0.85240000
H	8.19500000	0.27820000	1.29250000
C	6.65820000	1.15550000	0.05610000
H	7.21230000	2.07440000	-0.12550000
C	5.38990000	0.99140000	-0.50760000
H	4.97410000	1.79700000	-1.10850000
C	3.13070000	-2.10490000	-1.69280000
C	2.49310000	-3.21970000	-1.13010000
H	1.85440000	-3.09330000	-0.26160000
C	2.68080000	-4.49550000	-1.67500000
H	2.16940000	-5.34480000	-1.22860000
C	3.50270000	-4.67220000	-2.78770000

H	3.64490000	-5.66430000	-3.21220000
C	4.14700000	-3.56780000	-3.35560000
H	4.79600000	-3.69700000	-4.21990000
C	3.96760000	-2.29740000	-2.80960000
H	4.49560000	-1.45520000	-3.25120000
C	-0.02420000	3.37440000	-1.11180000
C	0.43810000	4.18620000	-0.06140000
H	1.11490000	3.77050000	0.67970000
C	0.02390000	5.51540000	0.04690000
H	0.39020000	6.12660000	0.86940000
C	-0.86670000	6.05200000	-0.88570000
H	-1.19680000	7.08520000	-0.79630000
C	-1.33540000	5.25450000	-1.93230000
H	-2.03040000	5.66360000	-2.66310000
C	-0.91510000	3.92730000	-2.04690000
H	-1.28950000	3.33120000	-2.87500000
C	-3.20150000	-1.43670000	-1.66850000
C	-4.16990000	-2.08180000	-0.87770000
H	-3.87080000	-2.59360000	0.03390000
C	-5.51470000	-2.07940000	-1.24680000
H	-6.24460000	-2.58770000	-0.62000000
C	-5.92280000	-1.42330000	-2.41220000
H	-6.97270000	-1.41490000	-2.69770000
C	-4.97380000	-0.77810000	-3.20510000
H	-5.27920000	-0.26290000	-4.11400000
C	-3.62360000	-0.78740000	-2.83850000
H	-2.91100000	-0.27520000	-3.47870000
C	-1.11120000	-3.30330000	-1.30030000
C	-1.09320000	-3.89840000	-2.57380000
H	-1.21040000	-3.28810000	-3.46630000
C	-0.94640000	-5.27890000	-2.71260000
H	-0.93450000	-5.72310000	-3.70600000
C	-0.82670000	-6.08920000	-1.57870000
H	-0.72280000	-7.16730000	-1.68710000
C	-0.84780000	-5.50950000	-0.30850000
H	-0.75790000	-6.13230000	0.57950000
C	-0.98670000	-4.12460000	-0.16980000
H	-0.98900000	-3.67640000	0.81910000
C	0.15780000	-0.50090000	4.15490000
H	0.42830000	-0.58090000	5.22090000
H	-0.89080000	-0.80980000	4.07920000

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**Species A<sub>ox</sub>**

HF energy= -3841.64381641

No Imaginary frequency

Zero-point correction = 0.692144 (Hartree/Particle)

Thermal correction to Energy = 0.742872

Thermal correction to Enthalpy = 0.743817

Thermal correction to Gibbs Free Energy = 0.602927  
 Sum of electronic and zero-point Energies = -3840.951672  
 Sum of electronic and thermal Energies = -3840.900944  
 Sum of electronic and thermal Enthalpies = -3840.900000  
 Sum of electronic and thermal Free Energies = -3841.040889

Coordinates:  $A_{ox}$

Fe	0.02460000	6.24970000	7.10820000
Fe	-0.49790000	7.62550000	4.88250000
P	-0.53170000	4.95390000	9.11700000
P	0.50300000	4.20880000	5.98100000
P	-0.59100000	6.30070000	2.98930000
S	-2.01980000	6.24750000	6.03730000
S	-0.09690000	8.57560000	6.98110000
O	2.82120000	6.64320000	8.00350000
O	2.29490000	8.33950000	4.26680000
O	-1.85300000	9.94180000	3.63150000
C	1.74140000	6.46650000	7.66040000
C	1.19910000	8.06350000	4.48070000
C	-1.34700000	9.02120000	4.09700000
C	-3.25270000	7.22560000	7.01810000
H	-4.21210000	7.05810000	6.51460000
H	-3.31540000	6.75760000	8.00660000
C	-2.97820000	8.72160000	7.13670000
H	-3.01250000	9.19250000	6.14770000
H	-3.79920000	9.16690000	7.72030000
C	-1.67000000	9.07860000	7.83370000
H	-1.63570000	8.65250000	8.84450000
H	-1.58210000	10.16630000	7.93540000
C	-0.10000000	3.21040000	8.58170000
H	0.96300000	3.07600000	8.81940000
H	-0.66140000	2.48250000	9.18020000
C	-0.34520000	2.96480000	7.08100000
H	-1.41020000	3.06810000	6.83890000
H	-0.06160000	1.94000000	6.81150000
C	-0.25450000	3.78350000	4.32860000
H	-0.00610000	2.73000000	4.14200000
H	-1.34450000	3.84660000	4.43260000
C	0.24550000	4.63720000	3.15060000
H	0.09260000	4.08760000	2.21450000
H	1.32570000	4.80230000	3.23510000
C	0.43310000	5.22350000	10.65310000
C	0.97750000	6.49230000	10.91130000
H	0.85090000	7.29790000	10.19090000
C	1.68740000	6.73040000	12.09160000
H	2.10530000	7.71660000	12.27980000
C	1.86540000	5.70210000	13.01890000
H	2.42370000	5.88510000	13.93420000
C	1.32480000	4.43580000	12.77160000
H	1.45860000	3.63430000	13.49460000



C	0.60660000	4.19730000	11.59980000
H	0.17700000	3.21220000	11.43060000
C	-2.27310000	4.86980000	9.71160000
C	-3.20940000	3.98690000	9.14770000
H	-2.91540000	3.28680000	8.37030000
C	-4.53720000	3.98890000	9.58220000
H	-5.24700000	3.29190000	9.14210000
C	-4.95040000	4.87670000	10.57850000
H	-5.98410000	4.87610000	10.91620000
C	-4.02720000	5.76050000	11.14410000
H	-4.33750000	6.44770000	11.92820000
C	-2.69830000	5.75870000	10.71550000
H	-1.98730000	6.43820000	11.18010000
C	2.24800000	3.66230000	5.79710000
C	3.20320000	4.60110000	5.36540000
H	2.91450000	5.63600000	5.19810000
C	4.53060000	4.22080000	5.15550000
H	5.25570000	4.95980000	4.82290000
C	4.92470000	2.90060000	5.38550000
H	5.96010000	2.60550000	5.23180000
C	3.98600000	1.96010000	5.81800000
H	4.28770000	0.93120000	6.00030000
C	2.65570000	2.33470000	6.01770000
H	1.94550000	1.58030000	6.34650000
C	0.25840000	7.05130000	1.53340000
C	-0.46590000	7.91580000	0.69300000
H	-1.52860000	8.07530000	0.85870000
C	0.16480000	8.56550000	-0.36920000
H	-0.41090000	9.22930000	-1.01010000
C	1.52600000	8.36080000	-0.61020000
H	2.01610000	8.86520000	-1.43960000
C	2.25420000	7.50460000	0.21830000
H	3.31380000	7.33810000	0.03720000
C	1.62790000	6.85750000	1.28630000
H	2.22380000	6.20680000	1.92090000
C	-2.24870000	5.86060000	2.30490000
C	-3.42430000	6.41440000	2.83340000
H	-3.36930000	7.08390000	3.68510000
C	-4.66920000	6.10960000	2.27450000
H	-5.57030000	6.55170000	2.69370000
C	-4.75450000	5.24410000	1.18320000
H	-5.72280000	5.00600000	0.74870000
C	-3.58910000	4.69110000	0.64250000
H	-3.64660000	4.02520000	-0.21560000
C	-2.34630000	5.00090000	1.19420000
H	-1.45110000	4.58590000	0.73610000

**Species C<sub>ox</sub>**

HF energy= -3841.63588462

No Imaginary frequency

Zero-point correction = 0.692551 (Hartree/Particle)  
 Thermal correction to Energy = 0.742942  
 Thermal correction to Enthalpy = 0.743886  
 Thermal correction to Gibbs Free Energy = 0.604530  
 Sum of electronic and zero-point Energies = -3840.943334  
 Sum of electronic and thermal Energies = -3840.892943  
 Sum of electronic and thermal Enthalpies = -3840.891999  
 Sum of electronic and thermal Free Energies = -3841.031354

Coordinates: C<sub>ox</sub>

Fe	1.02610000	0.38390000	0.64630000
Fe	-1.19020000	-0.89360000	1.00720000
P	2.74180000	-0.66960000	-0.58640000
P	0.60310000	1.60730000	-1.26870000
P	-2.87050000	-0.13000000	-0.38150000
S	0.80320000	-1.31770000	2.18650000
S	-0.86530000	1.32250000	1.66720000
O	2.72150000	2.28190000	2.06470000
O	-0.23830000	-2.38630000	-1.35050000
O	-2.57600000	-3.18570000	2.30630000
C	2.09670000	1.52870000	1.45970000
C	-0.52140000	-1.77230000	-0.41350000
C	-2.06890000	-2.30420000	1.77830000
C	2.33480000	-0.29690000	-2.36470000
H	3.16240000	-0.54970000	-3.03720000
H	1.50090000	-0.95820000	-2.62720000
C	1.91500000	1.17200000	-2.52450000
H	1.56400000	1.36420000	-3.54640000
H	2.76310000	1.84310000	-2.34020000
C	-0.96830000	1.14030000	-2.18150000
H	-0.99620000	1.77200000	-3.07960000
H	-0.84860000	0.10770000	-2.53180000
C	-2.29370000	1.32660000	-1.40260000
H	-3.10030000	1.57710000	-2.10250000
H	-2.22980000	2.17260000	-0.70860000
C	0.66340000	-0.72640000	3.93550000
H	0.64610000	-1.63830000	4.54300000
H	1.59820000	-0.19790000	4.16270000
C	-0.59910000	1.46030000	3.49860000
H	0.31360000	2.04760000	3.65550000
H	-1.44270000	2.06750000	3.84590000
C	4.42910000	0.02820000	-0.31170000
C	4.92590000	0.09440000	1.00230000
H	4.30860000	-0.22890000	1.83720000
C	6.21290000	0.57060000	1.25480000
H	6.58110000	0.61030000	2.27740000
C	7.02080000	1.00090000	0.19920000
H	8.02160000	1.37820000	0.39590000
C	6.53770000	0.94370000	-1.10950000
H	7.16120000	1.27290000	-1.93770000

C	5.25350000	0.45570000	-1.36540000
H	4.91280000	0.40770000	-2.39590000
C	3.03320000	-2.48960000	-0.54430000
C	3.43330000	-3.08750000	0.66420000
H	3.54630000	-2.48780000	1.56210000
C	3.69020000	-4.45720000	0.72770000
H	4.00040000	-4.90200000	1.67050000
C	3.54720000	-5.25260000	-0.41220000
H	3.74610000	-6.32060000	-0.36170000
C	3.14950000	-4.66950000	-1.61660000
H	3.03860000	-5.27960000	-2.51020000
C	2.89430000	-3.29760000	-1.68500000
H	2.58900000	-2.87440000	-2.63680000
C	0.56350000	3.44510000	-1.30820000
C	0.52530000	4.20140000	-0.12650000
H	0.49650000	3.70360000	0.83680000
C	0.51010000	5.59820000	-0.17980000
H	0.48080000	6.17050000	0.74460000
C	0.53290000	6.25470000	-1.41110000
H	0.52290000	7.34160000	-1.45100000
C	0.56970000	5.51160000	-2.59540000
H	0.58970000	6.01710000	-3.55820000
C	0.58370000	4.11810000	-2.54530000
H	0.61730000	3.55990000	-3.47890000
C	-4.39510000	0.51560000	0.42020000
C	-4.46920000	0.66190000	1.81380000
H	-3.61790000	0.39160000	2.43230000
C	-5.63130000	1.15710000	2.41160000
H	-5.67960000	1.26150000	3.49320000
C	-6.72690000	1.51310000	1.62270000
H	-7.63140000	1.89800000	2.08820000
C	-6.66370000	1.36590000	0.23330000
H	-7.51810000	1.63390000	-0.38410000
C	-5.50790000	0.86500000	-0.36570000
H	-5.48120000	0.72980000	-1.44500000
C	-3.51970000	-1.34150000	-1.60940000
C	-3.50550000	-1.12340000	-2.99640000
H	-3.10170000	-0.20510000	-3.41330000
C	-4.01780000	-2.08630000	-3.87190000
H	-3.99920000	-1.90100000	-4.94360000
C	-4.55330000	-3.27500000	-3.37470000
H	-4.95190000	-4.02210000	-4.05710000
C	-4.57840000	-3.50010000	-1.99490000
H	-4.99930000	-4.42080000	-1.59750000
C	-4.06350000	-2.54360000	-1.12050000
H	-4.09700000	-2.73270000	-0.05020000
C	-0.54260000	0.14590000	4.26820000
H	-0.50230000	0.38250000	5.34280000
H	-1.46940000	-0.42110000	4.10760000

(vi) Table 1. Crystal data and structure refinement for str0432 (1).

Identification code	str0432	
Chemical formula	$C_{41}H_{41}Cl_2Fe_2O_3P_3S_2$	
Formula weight	921.37	
Temperature	150(2) K	
Radiation, wavelength	MoK $\alpha$ , 0.71073 Å	
Crystal system, space group	triclinic, P1bar	
Unit cell parameters	a = 9.4439(8) Å	$\alpha =$
99.7830(10)°	b = 9.6258(9) Å	$\beta = 95.331(2)^\circ$
	c = 22.437(2) Å	$\gamma = 94.278(2)^\circ$
Cell volume	1992.8(3) Å <sup>3</sup>	
Z	2	
Calculated density	1.536 g/cm <sup>3</sup>	
Absorption coefficient $\mu$	1.127 mm <sup>-1</sup>	
F(000)	948	
Crystal colour and size	orange, 0.16 × 0.15 × 0.04 mm <sup>3</sup>	
Data collection method	Bruker SMART APEX diffractometer	
	$\omega$ rotation with narrow frames	
$\theta$ range for data collection	1.85 to 28.29°	
Index ranges	h -12 to 12, k -12 to 12, l -29 to 28	
Completeness to $\theta = 26.00^\circ$	98.4 %	
Reflections collected	17384	
Independent reflections	9087 ( $R_{int} = 0.0228$ )	
Reflections with $F^2 > 2\sigma$	7520	
Absorption correction	semi-empirical from equivalents	
Min. and max. transmission	0.8403 and 0.9563	
Structure solution	direct methods	
Refinement method	Full-matrix least-squares on $F^2$	
Weighting parameters a, b	0.0503, 0.9529	
Data / restraints / parameters	9087 / 0 / 478	
Final R indices [ $F^2 > 2\sigma$ ]	R1 = 0.0405, wR2 = 0.0950	
R indices (all data)	R1 = 0.0510, wR2 = 0.0998	
Goodness-of-fit on $F^2$	1.046	
Largest and mean shift/su	0.001 and 0.000	
Largest diff. peak and hole	1.659 and -1.013 e Å <sup>-3</sup>	

Table 2. Atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ ) for str0432.  $U_{\text{eq}}$  is defined as one third of the trace of the orthogonalized  $U^{\text{ij}}$  tensor.

	x	y	z	$U_{\text{eq}}$
Fe(1)	0.13516(3)	0.76806(3)	0.319378(14)	0.01580(9)
Fe(2)	0.08546(4)	0.89030(3)	0.229439(14)	0.01622(9)
P(1)	0.08711(7)	0.68444(6)	0.40143(3)	0.01742(13)
P(2)	0.15237(6)	0.54995(6)	0.27423(3)	0.01711(13)
P(3)	0.03901(6)	0.73007(6)	0.14534(3)	0.01703(13)
S(1)	-0.08223(6)	0.75661(6)	0.26652(3)	0.01738(12)
S(2)	0.12665(7)	1.00513(6)	0.32849(3)	0.02051(13)
O(1)	0.43955(19)	0.8259(2)	0.35641(9)	0.0306(4)
O(2)	0.3926(2)	0.9232(2)	0.22207(9)	0.0348(5)
O(3)	-0.0007(2)	1.1229(2)	0.17081(9)	0.0398(5)
C(1)	0.3189(3)	0.8001(2)	0.34176(11)	0.0204(5)
C(2)	0.2709(3)	0.9086(3)	0.22498(11)	0.0223(5)
C(3)	0.0273(3)	1.0285(3)	0.19386(11)	0.0230(5)
C(4)	-0.2142(3)	0.8682(3)	0.29942(12)	0.0251(5)
C(5)	-0.1709(3)	1.0259(3)	0.31130(12)	0.0272(6)
C(6)	-0.0390(3)	1.0665(3)	0.35590(11)	0.0266(6)
C(7)	0.1040(3)	0.4903(2)	0.38567(11)	0.0217(5)
C(8)	0.0600(3)	0.4335(2)	0.31817(11)	0.0204(5)
C(9)	0.0676(3)	0.4818(3)	0.19608(11)	0.0206(5)
C(10)	0.1172(3)	0.5612(3)	0.14687(11)	0.0201(5)
C(11)	0.2066(2)	0.7478(3)	0.47113(10)	0.0188(5)
C(12)	0.2650(3)	0.8881(3)	0.48240(11)	0.0242(5)
C(13)	0.3562(3)	0.9422(3)	0.53420(11)	0.0254(5)
C(14)	0.3927(3)	0.8562(3)	0.57547(11)	0.0222(5)
C(15)	0.3343(3)	0.7169(3)	0.56554(11)	0.0248(5)
C(16)	0.2411(3)	0.6628(3)	0.51406(11)	0.0223(5)
C(17)	-0.0896(3)	0.6876(2)	0.42977(11)	0.0195(5)
C(18)	-0.2103(3)	0.6263(3)	0.39049(11)	0.0217(5)
C(19)	-0.3446(3)	0.6236(3)	0.41050(12)	0.0246(5)
C(20)	-0.3616(3)	0.6814(3)	0.46998(12)	0.0265(6)
C(21)	-0.2434(3)	0.7427(3)	0.50926(12)	0.0272(6)
C(22)	-0.1086(3)	0.7471(3)	0.48931(11)	0.0234(5)
C(23)	0.3291(3)	0.4837(3)	0.26948(10)	0.0204(5)
C(24)	0.4356(3)	0.5627(3)	0.24687(11)	0.0241(5)
C(25)	0.5676(3)	0.5126(3)	0.23905(12)	0.0298(6)
C(26)	0.5973(3)	0.3846(3)	0.25468(13)	0.0331(6)
C(27)	0.4947(3)	0.3053(3)	0.27820(13)	0.0334(6)
C(28)	0.3610(3)	0.3544(3)	0.28529(11)	0.0257(5)
C(29)	0.0984(3)	0.7788(3)	0.07578(10)	0.0200(5)
C(30)	0.0038(4)	0.7926(6)	0.02752(16)	0.0719(15)
C(31)	0.0516(4)	0.8337(7)	-0.02416(19)	0.0904(19)
C(32)	0.1923(3)	0.8623(4)	-0.02822(13)	0.0389(7)
C(33)	0.2881(3)	0.8487(4)	0.01921(13)	0.0395(7)
C(34)	0.2406(3)	0.8082(4)	0.07094(13)	0.0397(8)
C(35)	-0.1510(3)	0.6762(3)	0.12117(10)	0.0196(5)
C(36)	-0.2490(3)	0.7756(3)	0.13237(12)	0.0254(5)
C(37)	-0.3931(3)	0.7406(3)	0.11272(13)	0.0312(6)
C(38)	-0.4406(3)	0.6062(3)	0.08309(13)	0.0336(6)

C(39)	-0.3451(3)	0.5065(3)	0.07225(13)	0.0352(7)
C(40)	-0.2004(3)	0.5408(3)	0.09038(12)	0.0282(6)
C(41)	0.3661(4)	0.1912(5)	0.12401(17)	0.0640(11)
Cl(1)	0.53156(9)	0.12141(10)	0.12418(4)	0.0513(2)
Cl(2)	0.32746(11)	0.27003(12)	0.06063(6)	0.0759(3)

Table 3. Bond lengths [Å] and angles [°] for str0432.

Fe(1)–C(1)	1.750(3)	Fe(1)–P(2)	2.1953(7)
Fe(1)–P(1)	2.2016(7)	Fe(1)–S(1)	2.2589(7)
Fe(1)–S(2)	2.2644(7)	Fe(1)–Fe(2)	2.5279(5)
Fe(2)–C(3)	1.760(3)	Fe(2)–C(2)	1.761(3)
Fe(2)–P(3)	2.2109(7)	Fe(2)–S(1)	2.2718(7)
Fe(2)–S(2)	2.2918(7)	P(1)–C(11)	1.827(2)
P(1)–C(17)	1.841(3)	P(1)–C(7)	1.863(2)
P(2)–C(23)	1.836(2)	P(2)–C(8)	1.838(2)
P(2)–C(9)	1.848(2)	P(3)–C(29)	1.828(2)
P(3)–C(35)	1.835(2)	P(3)–C(10)	1.839(2)
S(1)–C(4)	1.831(2)	S(2)–C(6)	1.835(3)
O(1)–C(1)	1.153(3)	O(2)–C(2)	1.156(3)
O(3)–C(3)	1.155(3)	C(4)–C(5)	1.515(4)
C(5)–C(6)	1.506(4)	C(7)–C(8)	1.528(3)
C(9)–C(10)	1.534(3)	C(11)–C(12)	1.392(3)
C(11)–C(16)	1.397(3)	C(12)–C(13)	1.380(3)
C(13)–C(14)	1.380(4)	C(14)–C(15)	1.385(4)
C(15)–C(16)	1.385(3)	C(17)–C(22)	1.394(3)
C(17)–C(18)	1.400(3)	C(18)–C(19)	1.384(4)
C(19)–C(20)	1.385(4)	C(20)–C(21)	1.383(4)
C(21)–C(22)	1.388(4)	C(23)–C(28)	1.397(3)
C(23)–C(24)	1.400(4)	C(24)–C(25)	1.385(4)
C(25)–C(26)	1.378(4)	C(26)–C(27)	1.386(4)
C(27)–C(28)	1.394(4)	C(29)–C(30)	1.369(4)
C(29)–C(34)	1.370(4)	C(30)–C(31)	1.390(4)
C(31)–C(32)	1.351(5)	C(32)–C(33)	1.361(4)
C(33)–C(34)	1.387(4)	C(35)–C(36)	1.388(4)
C(35)–C(40)	1.392(4)	C(36)–C(37)	1.390(4)
C(37)–C(38)	1.371(4)	C(38)–C(39)	1.373(4)
C(39)–C(40)	1.388(4)	C(41)–Cl(2)	1.744(4)
C(41)–Cl(1)	1.747(4)		
C(1)–Fe(1)–P(2)	93.26(8)	C(1)–Fe(1)–P(1)	94.47(8)
P(2)–Fe(1)–P(1)	88.37(3)	C(1)–Fe(1)–S(1)	162.54(8)
P(2)–Fe(1)–S(1)	88.13(2)	P(1)–Fe(1)–S(1)	102.97(3)
C(1)–Fe(1)–S(2)	87.99(8)	P(2)–Fe(1)–S(2)	157.61(3)
P(1)–Fe(1)–S(2)	113.84(3)	S(1)–Fe(1)–S(2)	84.24(2)
C(1)–Fe(1)–Fe(2)	106.46(8)	P(2)–Fe(1)–Fe(2)	101.66(2)
P(1)–Fe(1)–Fe(2)	156.05(2)	S(1)–Fe(1)–Fe(2)	56.328(18)
S(2)–Fe(1)–Fe(2)	56.820(18)	C(3)–Fe(2)–C(2)	101.97(12)
C(3)–Fe(2)–P(3)	93.47(8)	C(2)–Fe(2)–P(3)	94.58(8)
C(3)–Fe(2)–S(1)	117.39(9)	C(2)–Fe(2)–S(1)	140.62(8)
P(3)–Fe(2)–S(1)	84.90(2)	C(3)–Fe(2)–S(2)	100.30(8)
C(2)–Fe(2)–S(2)	88.60(8)	P(3)–Fe(2)–S(2)	164.92(3)
S(1)–Fe(2)–S(2)	83.33(2)	C(3)–Fe(2)–Fe(1)	154.24(8)
C(2)–Fe(2)–Fe(1)	88.00(8)	P(3)–Fe(2)–Fe(1)	109.52(2)
S(1)–Fe(2)–Fe(1)	55.846(18)	S(2)–Fe(2)–Fe(1)	55.787(19)
C(11)–P(1)–C(17)	101.93(11)	C(11)–P(1)–C(7)	103.86(11)
C(17)–P(1)–C(7)	100.77(11)	C(11)–P(1)–Fe(1)	117.57(8)
C(17)–P(1)–Fe(1)	122.77(8)	C(7)–P(1)–Fe(1)	107.32(8)
C(23)–P(2)–C(8)	104.10(11)	C(23)–P(2)–C(9)	100.61(11)
C(8)–P(2)–C(9)	101.86(11)	C(23)–P(2)–Fe(1)	119.72(8)
C(8)–P(2)–Fe(1)	106.59(8)	C(9)–P(2)–Fe(1)	121.47(8)

C(29)–P(3)–C(35)	101.09(11)	C(29)–P(3)–C(10)	101.80(11)
C(35)–P(3)–C(10)	103.61(11)	C(29)–P(3)–Fe(2)	116.95(8)
C(35)–P(3)–Fe(2)	115.42(8)	C(10)–P(3)–Fe(2)	115.78(8)
C(4)–S(1)–Fe(1)	118.18(9)	C(4)–S(1)–Fe(2)	110.07(9)
Fe(1)–S(1)–Fe(2)	67.83(2)	C(6)–S(2)–Fe(1)	113.31(9)
C(6)–S(2)–Fe(2)	110.68(9)	Fe(1)–S(2)–Fe(2)	67.39(2)
O(1)–C(1)–Fe(1)	177.7(2)	O(2)–C(2)–Fe(2)	178.8(2)
O(3)–C(3)–Fe(2)	175.1(2)	C(5)–C(4)–S(1)	115.39(18)
C(6)–C(5)–C(4)	112.5(2)	C(5)–C(6)–S(2)	115.23(17)
C(8)–C(7)–P(1)	109.18(16)	C(7)–C(8)–P(2)	108.04(16)
C(10)–C(9)–P(2)	115.67(17)	C(9)–C(10)–P(3)	114.16(16)
C(12)–C(11)–C(16)	118.3(2)	C(12)–C(11)–P(1)	118.42(18)
C(16)–C(11)–P(1)	123.24(18)	C(13)–C(12)–C(11)	121.2(2)
C(12)–C(13)–C(14)	120.0(2)	C(13)–C(14)–C(15)	119.7(2)
C(14)–C(15)–C(16)	120.4(2)	C(15)–C(16)–C(11)	120.3(2)
C(22)–C(17)–C(18)	118.3(2)	C(22)–C(17)–P(1)	122.67(19)
C(18)–C(17)–P(1)	119.04(18)	C(19)–C(18)–C(17)	120.8(2)
C(18)–C(19)–C(20)	120.3(2)	C(21)–C(20)–C(19)	119.5(2)
C(20)–C(21)–C(22)	120.4(2)	C(21)–C(22)–C(17)	120.6(2)
C(28)–C(23)–C(24)	118.0(2)	C(28)–C(23)–P(2)	122.9(2)
C(24)–C(23)–P(2)	119.05(18)	C(25)–C(24)–C(23)	121.0(2)
C(26)–C(25)–C(24)	120.3(3)	C(25)–C(26)–C(27)	120.0(3)
C(26)–C(27)–C(28)	119.9(3)	C(27)–C(28)–C(23)	120.8(3)
C(30)–C(29)–C(34)	117.2(2)	C(30)–C(29)–P(3)	122.0(2)
C(34)–C(29)–P(3)	120.77(19)	C(29)–C(30)–C(31)	120.8(3)
C(32)–C(31)–C(30)	121.2(3)	C(31)–C(32)–C(33)	118.9(3)
C(32)–C(33)–C(34)	120.0(3)	C(29)–C(34)–C(33)	121.8(3)
C(36)–C(35)–C(40)	118.6(2)	C(36)–C(35)–P(3)	118.75(19)
C(40)–C(35)–P(3)	122.6(2)	C(35)–C(36)–C(37)	120.5(2)
C(38)–C(37)–C(36)	120.2(3)	C(37)–C(38)–C(39)	119.9(3)
C(38)–C(39)–C(40)	120.6(3)	C(39)–C(40)–C(35)	120.2(3)
Cl(2)–C(41)–Cl(1)	112.1(2)		



Table 4. Anisotropic displacement parameters ( $\text{\AA}^2$ ) for str0432. The anisotropic displacement factor exponent takes the form:  $-2\pi^2[h^2a^*2U^{11} + \dots + 2hka^*b^*U^{12}]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Fe(1)	0.01918(18)	0.01477(16)	0.01276(16)	0.00378(12)	-0.00172(12)	
-0.00167(12)						
Fe(2)	0.01958(18)	0.01635(17)	0.01244(16)	0.00408(13)	-0.00026(12)	
-0.00116(13)						
P(1)	0.0217(3)	0.0163(3)	0.0136(3)	0.0043(2)	-0.0016(2)	-0.0021(2)
P(2)	0.0202(3)	0.0156(3)	0.0150(3)	0.0038(2)	-0.0012(2)	-0.0003(2)
P(3)	0.0177(3)	0.0197(3)	0.0135(3)	0.0036(2)	-0.0003(2)	0.0007(2)
S(1)	0.0189(3)	0.0181(3)	0.0150(3)	0.0044(2)	-0.0001(2)	-0.0003(2)
S(2)	0.0294(3)	0.0159(3)	0.0150(3)	0.0027(2)	-0.0005(2)	-0.0020(2)
O(1)	0.0238(10)	0.0319(10)	0.0325(10)	0.0022(8)	-0.0053(8)	-0.0024(8)
O(2)	0.0251(11)	0.0463(12)	0.0343(11)	0.0128(9)	0.0051(8)	-0.0040(9)
O(3)	0.0633(15)	0.0262(10)	0.0316(11)	0.0150(9)	-0.0034(10)	0.0026(10)
C(1)	0.0270(13)	0.0163(11)	0.0175(11)	0.0039(9)	0.0004(10)	0.0003(9)
C(2)	0.0290(14)	0.0213(12)	0.0152(11)	0.0026(9)	0.0000(10)	-0.0026(10)
C(3)	0.0302(14)	0.0206(12)	0.0170(11)	0.0028(10)	-0.0002(10)	-0.0008(10)
C(4)	0.0216(13)	0.0293(14)	0.0269(13)	0.0076(11)	0.0068(10)	0.0072(10)
C(5)	0.0343(15)	0.0268(14)	0.0243(13)	0.0072(11)	0.0104(11)	0.0114(11)
C(6)	0.0429(16)	0.0202(12)	0.0179(12)	0.0014(10)	0.0103(11)	0.0058(11)
C(7)	0.0271(13)	0.0185(12)	0.0203(12)	0.0074(10)	0.0013(10)	-0.0001(10)
C(8)	0.0250(13)	0.0165(11)	0.0190(11)	0.0032(9)	0.0016(9)	-0.0017(9)
C(9)	0.0241(13)	0.0173(12)	0.0182(11)	0.0009(9)	-0.0035(9)	-0.0001(9)
C(10)	0.0212(12)	0.0232(12)	0.0157(11)	0.0029(9)	-0.0005(9)	0.0038(10)
C(11)	0.0194(12)	0.0215(12)	0.0147(11)	0.0039(9)	-0.0007(9)	-0.0010(9)
C(12)	0.0329(14)	0.0211(12)	0.0181(12)	0.0057(10)	-0.0020(10)	-0.0012(10)
C(13)	0.0321(14)	0.0205(13)	0.0211(12)	0.0011(10)	-0.0003(10)	-0.0042(10)
C(14)	0.0216(12)	0.0282(13)	0.0147(11)	0.0016(10)	-0.0015(9)	-0.0023(10)
C(15)	0.0297(14)	0.0268(13)	0.0186(12)	0.0086(10)	-0.0032(10)	0.0034(11)
C(16)	0.0275(13)	0.0200(12)	0.0195(12)	0.0065(10)	0.0001(10)	-0.0016(10)
C(17)	0.0238(12)	0.0172(11)	0.0186(11)	0.0080(9)	0.0006(9)	0.0007(9)
C(18)	0.0260(13)	0.0223(12)	0.0171(11)	0.0069(10)	0.0001(10)	-0.0016(10)
C(19)	0.0234(13)	0.0227(13)	0.0277(13)	0.0081(11)	-0.0014(10)	-0.0005(10)
C(20)	0.0253(13)	0.0253(13)	0.0322(14)	0.0100(11)	0.0076(11)	0.0073(10)
C(21)	0.0350(15)	0.0243(13)	0.0235(13)	0.0043(10)	0.0056(11)	0.0075(11)
C(22)	0.0300(14)	0.0197(12)	0.0193(12)	0.0023(10)	-0.0012(10)	0.0024(10)
C(23)	0.0231(12)	0.0214(12)	0.0154(11)	0.0015(9)	-0.0025(9)	0.0028(10)
C(24)	0.0252(13)	0.0242(13)	0.0211(12)	0.0029(10)	-0.0036(10)	0.0013(10)
C(25)	0.0232(13)	0.0358(15)	0.0271(14)	-0.0004(12)	-0.0004(11)	-0.0018(11)
C(26)	0.0283(15)	0.0365(16)	0.0306(15)	-0.0036(12)	-0.0052(11)	0.0094(12)
C(27)	0.0378(16)	0.0302(15)	0.0315(15)	0.0048(12)	-0.0059(12)	0.0123(12)
C(28)	0.0302(14)	0.0241(13)	0.0229(13)	0.0056(10)	0.0000(10)	0.0033(11)
C(29)	0.0230(12)	0.0218(12)	0.0157(11)	0.0039(9)	0.0034(9)	0.0028(10)
C(30)	0.0240(17)	0.162(5)	0.045(2)	0.068(3)	0.0006(14)	0.000(2)
C(31)	0.038(2)	0.202(6)	0.055(2)	0.089(3)	0.0029(18)	0.013(3)
C(32)	0.0383(17)	0.057(2)	0.0284(15)	0.0212(14)	0.0129(13)	0.0085(15)
C(33)	0.0270(15)	0.065(2)	0.0263(14)	0.0104(14)	0.0076(12)	-0.0067(14)
C(34)	0.0226(14)	0.076(2)	0.0210(14)	0.0140(14)	-0.0016(11)	-0.0015(14)
C(35)	0.0200(12)	0.0253(13)	0.0133(11)	0.0052(9)	0.0002(9)	-0.0017(9)
C(36)	0.0250(13)	0.0239(13)	0.0252(13)	0.0031(10)	-0.0037(10)	0.0003(10)
C(37)	0.0231(14)	0.0373(16)	0.0330(15)	0.0058(12)	-0.0012(11)	0.0074(11)
C(38)	0.0222(14)	0.0424(17)	0.0337(15)	0.0068(13)	-0.0052(11)	-0.0036(12)
C(39)	0.0316(15)	0.0313(15)	0.0367(16)	-0.0005(13)	-0.0082(12)	-0.0069(12)
C(40)	0.0272(14)	0.0266(14)	0.0279(14)	0.0004(11)	-0.0026(11)	0.0004(11)
C(41)	0.045(2)	0.107(4)	0.041(2)	0.009(2)	0.0106(16)	0.016(2)
Cl(1)	0.0481(5)	0.0584(5)	0.0507(5)	0.0115(4)	0.0149(4)	0.0103(4)
Cl(2)	0.0570(6)	0.0723(7)	0.0973(9)	0.0368(6)	-0.0205(6)	-0.0151(5)

Table 5. Hydrogen coordinates and isotropic displacement parameters ( $\text{\AA}^2$ ) for str0432.

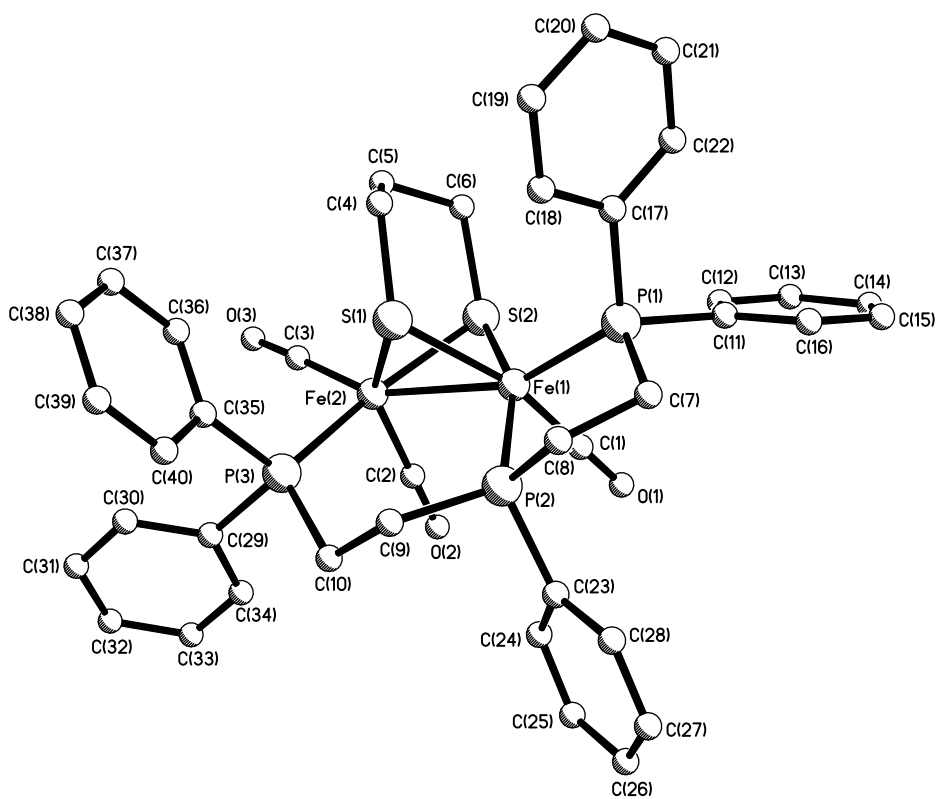
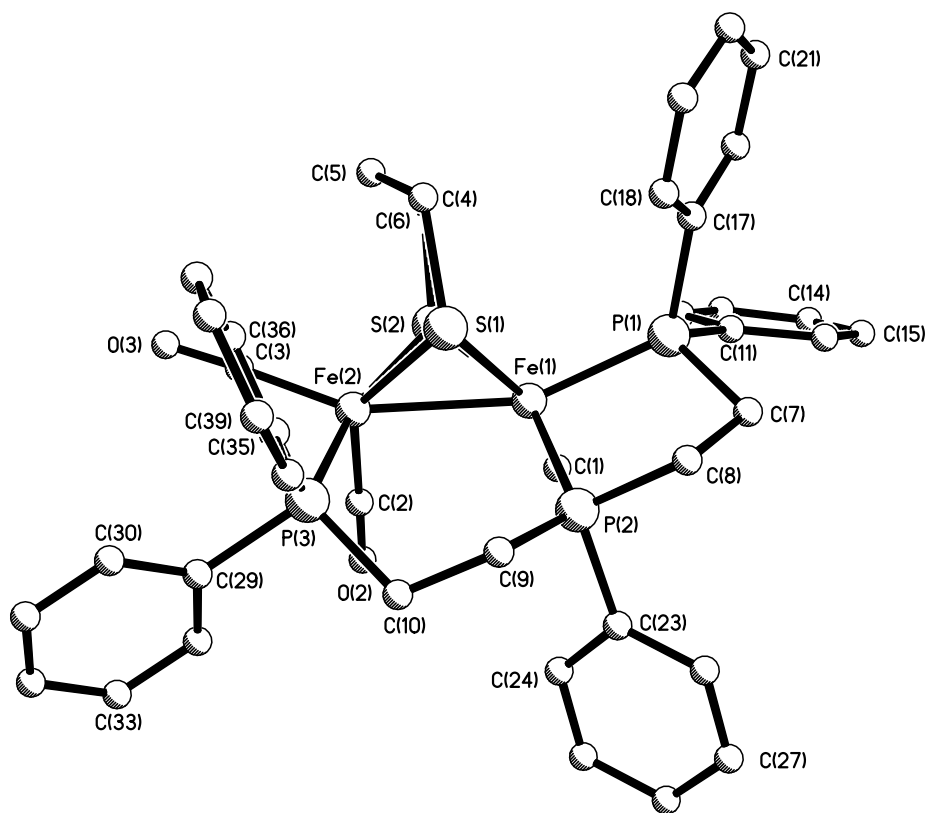
	x	y	z	U
H(4A)	-0.3039	0.8493	0.2716	0.030
H(4B)	-0.2349	0.8397	0.3383	0.030
H(5A)	-0.1530	1.0561	0.2724	0.033
H(5B)	-0.2505	1.0768	0.3273	0.033
H(6A)	-0.0547	1.0280	0.3932	0.032
H(6B)	-0.0263	1.1710	0.3673	0.032
H(7A)	0.2039	0.4717	0.3966	0.026
H(7B)	0.0419	0.4419	0.4105	0.026
H(8A)	-0.0447	0.4323	0.3090	0.024
H(8B)	0.0868	0.3356	0.3074	0.024
H(9A)	0.0862	0.3813	0.1850	0.025
H(9B)	-0.0369	0.4848	0.1960	0.025
H(10A)	0.0925	0.4994	0.1067	0.024
H(10B)	0.2225	0.5799	0.1535	0.024
H(12A)	0.2416	0.9476	0.4539	0.029
H(13A)	0.3937	1.0386	0.5415	0.030
H(14A)	0.4577	0.8925	0.6105	0.027
H(15A)	0.3584	0.6580	0.5942	0.030
H(16A)	0.2005	0.5675	0.5079	0.027
H(18A)	-0.1998	0.5861	0.3496	0.026
H(19A)	-0.4255	0.5819	0.3833	0.030
H(20A)	-0.4538	0.6790	0.4837	0.032
H(21A)	-0.2545	0.7820	0.5502	0.033
H(22A)	-0.0284	0.7911	0.5165	0.028
H(24A)	0.4171	0.6520	0.2367	0.029
H(25A)	0.6380	0.5667	0.2228	0.036
H(26A)	0.6881	0.3507	0.2493	0.040
H(27A)	0.5154	0.2176	0.2895	0.040
H(28A)	0.2907	0.2992	0.3011	0.031
H(30A)	-0.0957	0.7739	0.0294	0.086
H(31A)	-0.0159	0.8419	-0.0572	0.108
H(32A)	0.2239	0.8913	-0.0636	0.047
H(33A)	0.3875	0.8670	0.0169	0.047
H(34A)	0.3086	0.8007	0.1039	0.048
H(36A)	-0.2172	0.8682	0.1536	0.030
H(37A)	-0.4590	0.8099	0.1198	0.037
H(38A)	-0.5393	0.5822	0.0701	0.040
H(39A)	-0.3784	0.4132	0.0522	0.042
H(40A)	-0.1349	0.4717	0.0818	0.034
H(41A)	0.2905	0.1144	0.1244	0.077
H(41B)	0.3658	0.2624	0.1614	0.077

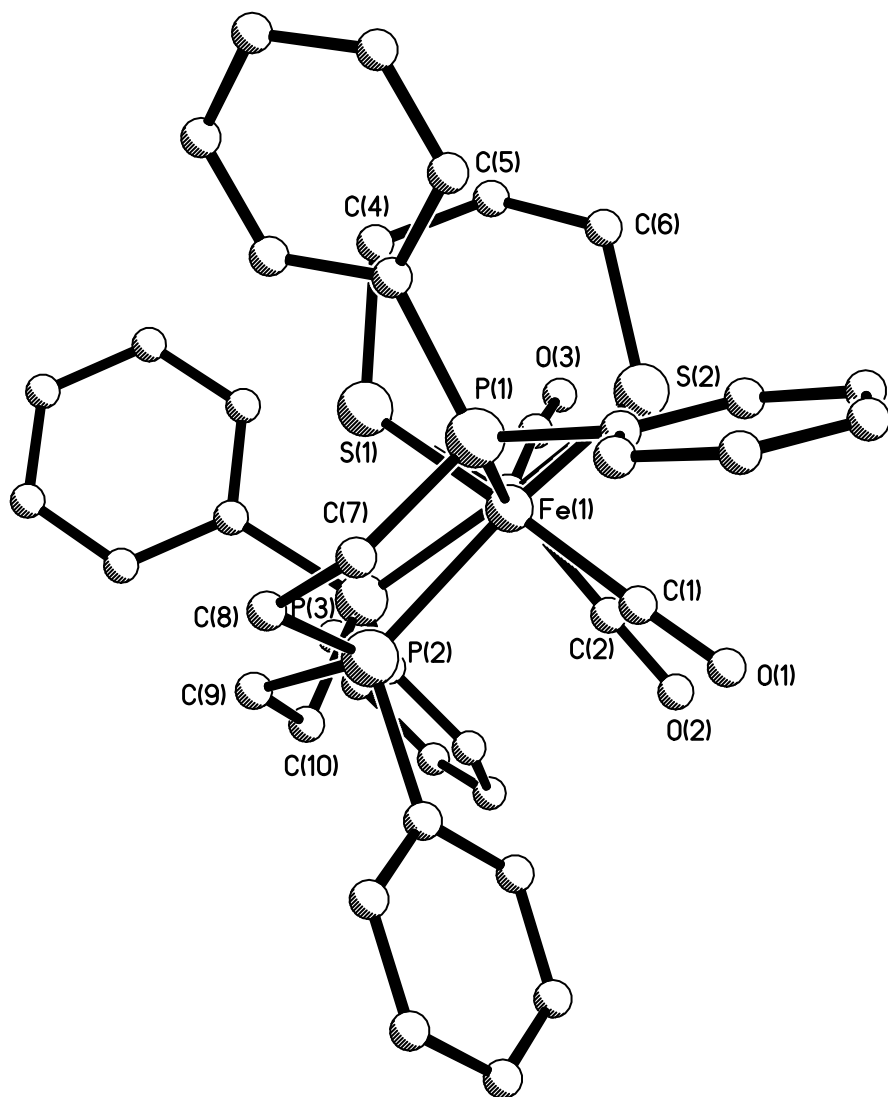
Table 6. Torsion angles [°] for str0432.

C(1)–Fe(1)–Fe(2)–C(3)	–100.5(2)	P(2)–Fe(1)–Fe(2)–C(3)	162.5(2)
P(1)–Fe(1)–Fe(2)–C(3)	49.4(2)	S(1)–Fe(1)–Fe(2)–C(3)	82.7(2)
S(2)–Fe(1)–Fe(2)–C(3)	–24.2(2)	C(1)–Fe(1)–Fe(2)–C(2)	13.37(11)
P(2)–Fe(1)–Fe(2)–C(2)	–83.61(8)	P(1)–Fe(1)–Fe(2)–C(2)	163.26(10)
S(1)–Fe(1)–Fe(2)–C(2)	–163.39(8)	S(2)–Fe(1)–Fe(2)–C(2)	89.66(8)
C(1)–Fe(1)–Fe(2)–P(3)	107.52(8)	P(2)–Fe(1)–Fe(2)–P(3)	10.55(3)
P(1)–Fe(1)–Fe(2)–P(3)	–102.58(6)	S(1)–Fe(1)–Fe(2)–P(3)	–69.23(3)
S(2)–Fe(1)–Fe(2)–P(3)	–176.18(3)	C(1)–Fe(1)–Fe(2)–S(1)	176.76(8)
P(2)–Fe(1)–Fe(2)–S(1)	79.78(3)	P(1)–Fe(1)–Fe(2)–S(1)	–33.35(6)
S(2)–Fe(1)–Fe(2)–S(1)	–106.94(3)	C(1)–Fe(1)–Fe(2)–S(2)	–76.30(8)
P(2)–Fe(1)–Fe(2)–S(2)	–173.27(3)	P(1)–Fe(1)–Fe(2)–S(2)	73.59(6)
S(1)–Fe(1)–Fe(2)–S(2)	106.94(3)	C(1)–Fe(1)–P(1)–C(11)	27.65(12)
P(2)–Fe(1)–P(1)–C(11)	120.79(9)	S(1)–Fe(1)–P(1)–C(11)	–151.50(9)
S(2)–Fe(1)–P(1)–C(11)	–62.12(10)	Fe(2)–Fe(1)–P(1)–C(11)	–123.50(10)
C(1)–Fe(1)–P(1)–C(17)	155.49(12)	P(2)–Fe(1)–P(1)–C(17)	–111.37(10)
S(1)–Fe(1)–P(1)–C(17)	–23.66(10)	S(2)–Fe(1)–P(1)–C(17)	65.71(10)
Fe(2)–Fe(1)–P(1)–C(17)	4.34(12)	C(1)–Fe(1)–P(1)–C(7)	–88.82(12)
P(2)–Fe(1)–P(1)–C(7)	4.32(9)	S(1)–Fe(1)–P(1)–C(7)	92.03(9)
S(2)–Fe(1)–P(1)–C(7)	–178.59(9)	Fe(2)–Fe(1)–P(1)–C(7)	120.03(10)
C(1)–Fe(1)–P(2)–C(23)	–4.19(12)	P(1)–Fe(1)–P(2)–C(23)	–98.57(9)
S(1)–Fe(1)–P(2)–C(23)	158.39(9)	S(2)–Fe(1)–P(2)–C(23)	88.44(11)
Fe(2)–Fe(1)–P(2)–C(23)	103.36(9)	C(1)–Fe(1)–P(2)–C(8)	113.37(12)
P(1)–Fe(1)–P(2)–C(8)	18.99(9)	S(1)–Fe(1)–P(2)–C(8)	–84.05(9)
S(2)–Fe(1)–P(2)–C(8)	–154.00(10)	Fe(2)–Fe(1)–P(2)–C(8)	–139.08(8)
C(1)–Fe(1)–P(2)–C(9)	–130.92(12)	P(1)–Fe(1)–P(2)–C(9)	134.70(10)
S(1)–Fe(1)–P(2)–C(9)	31.66(10)	S(2)–Fe(1)–P(2)–C(9)	–38.29(13)
Fe(2)–Fe(1)–P(2)–C(9)	–23.37(10)	C(3)–Fe(2)–P(3)–C(29)	45.92(12)
C(2)–Fe(2)–P(3)–C(29)	–56.39(12)	S(1)–Fe(2)–P(3)–C(29)	163.13(9)
S(2)–Fe(2)–P(3)–C(29)	–158.12(12)	Fe(1)–Fe(2)–P(3)–C(29)	–145.89(9)
C(3)–Fe(2)–P(3)–C(35)	–72.79(12)	C(2)–Fe(2)–P(3)–C(35)	–175.11(12)
S(1)–Fe(2)–P(3)–C(35)	44.42(9)	S(2)–Fe(2)–P(3)–C(35)	83.16(14)
Fe(1)–Fe(2)–P(3)–C(35)	95.39(9)	C(3)–Fe(2)–P(3)–C(10)	165.94(12)
C(2)–Fe(2)–P(3)–C(10)	63.63(12)	S(1)–Fe(2)–P(3)–C(10)	–76.85(9)
S(2)–Fe(2)–P(3)–C(10)	–38.10(15)	Fe(1)–Fe(2)–P(3)–C(10)	–25.87(9)
C(1)–Fe(1)–S(1)–C(4)	–112.1(3)	P(2)–Fe(1)–S(1)–C(4)	152.99(10)
P(1)–Fe(1)–S(1)–C(4)	65.10(10)	S(2)–Fe(1)–S(1)–C(4)	–48.08(10)
Fe(2)–Fe(1)–S(1)–C(4)	–101.66(10)	C(1)–Fe(1)–S(1)–Fe(2)	–10.4(3)
P(2)–Fe(1)–S(1)–Fe(2)	–105.34(2)	P(1)–Fe(1)–S(1)–Fe(2)	166.76(2)
S(2)–Fe(1)–S(1)–Fe(2)	53.58(2)	C(3)–Fe(2)–S(1)–C(4)	–37.76(13)
C(2)–Fe(2)–S(1)–C(4)	139.96(15)	P(3)–Fe(2)–S(1)–C(4)	–129.03(9)
S(2)–Fe(2)–S(1)–C(4)	60.41(9)	Fe(1)–Fe(2)–S(1)–C(4)	113.20(9)
C(3)–Fe(2)–S(1)–Fe(1)	–150.96(9)	C(2)–Fe(2)–S(1)–Fe(1)	26.76(13)
P(3)–Fe(2)–S(1)–Fe(1)	117.77(2)	S(2)–Fe(2)–S(1)–Fe(1)	–52.79(2)
C(1)–Fe(1)–S(2)–C(6)	–145.08(12)	P(2)–Fe(1)–S(2)–C(6)	121.25(11)
P(1)–Fe(1)–S(2)–C(6)	–51.09(10)	S(1)–Fe(1)–S(2)–C(6)	50.57(9)
Fe(2)–Fe(1)–S(2)–C(6)	103.72(9)	C(1)–Fe(1)–S(2)–Fe(2)	111.20(8)
P(2)–Fe(1)–S(2)–Fe(2)	17.53(8)	P(1)–Fe(1)–S(2)–Fe(2)	–154.80(3)
S(1)–Fe(1)–S(2)–Fe(2)	–53.14(2)	C(3)–Fe(2)–S(2)–C(6)	62.04(13)
C(2)–Fe(2)–S(2)–C(6)	163.94(12)	P(3)–Fe(2)–S(2)–C(6)	–93.55(14)
S(1)–Fe(2)–S(2)–C(6)	–54.67(9)	Fe(1)–Fe(2)–S(2)–C(6)	–107.52(9)

C(3)–Fe(2)–S(2)–Fe(1)	169.56(9)	C(2)–Fe(2)–S(2)–Fe(1)	–88.54(8)
P(3)–Fe(2)–S(2)–Fe(1)	13.97(11)	S(1)–Fe(2)–S(2)–Fe(1)	52.85(2)
P(2)–Fe(1)–C(1)–O(1)	151(6)	P(1)–Fe(1)–C(1)–O(1)	–120(6)
S(1)–Fe(1)–C(1)–O(1)	57(6)	S(2)–Fe(1)–C(1)–O(1)	–7(6)
Fe(2)–Fe(1)–C(1)–O(1)	48(6)	C(3)–Fe(2)–C(2)–O(2)	29(10)
P(3)–Fe(2)–C(2)–O(2)	123(10)	S(1)–Fe(2)–C(2)–O(2)	–149(10)
S(2)–Fe(2)–C(2)–O(2)	–71(10)	Fe(1)–Fe(2)–C(2)–O(2)	–127(10)
C(2)–Fe(2)–C(3)–O(3)	–13(3)	P(3)–Fe(2)–C(3)–O(3)	–108(3)
S(1)–Fe(2)–C(3)–O(3)	166(3)	S(2)–Fe(2)–C(3)–O(3)	78(3)
Fe(1)–Fe(2)–C(3)–O(3)	98(3)	Fe(1)–S(1)–C(4)–C(5)	59.7(2)
Fe(2)–S(1)–C(4)–C(5)	–15.2(2)	S(1)–C(4)–C(5)–C(6)	–61.2(3)
C(4)–C(5)–C(6)–S(2)	68.2(3)	Fe(1)–S(2)–C(6)–C(5)	–70.5(2)
Fe(2)–S(2)–C(6)–C(5)	2.9(2)	C(11)–P(1)–C(7)–C(8)	–157.57(17)
C(17)–P(1)–C(7)–C(8)	97.14(18)	Fe(1)–P(1)–C(7)–C(8)	–32.39(18)
P(1)–C(7)–C(8)–P(2)	48.2(2)	C(23)–P(2)–C(8)–C(7)	83.58(18)
C(9)–P(2)–C(8)–C(7)	–172.14(17)	Fe(1)–P(2)–C(8)–C(7)	–43.88(18)
C(23)–P(2)–C(9)–C(10)	–75.29(19)	C(8)–P(2)–C(9)–C(10)	177.70(18)
Fe(1)–P(2)–C(9)–C(10)	59.6(2)	P(2)–C(9)–C(10)–P(3)	–77.6(2)
C(29)–P(3)–C(10)–C(9)	–171.62(17)	C(35)–P(3)–C(10)–C(9)	–66.98(19)
Fe(2)–P(3)–C(10)–C(9)	60.42(19)	C(17)–P(1)–C(11)–C(12)	–101.5(2)
C(7)–P(1)–C(11)–C(12)	154.1(2)	Fe(1)–P(1)–C(11)–C(12)	35.8(2)
C(17)–P(1)–C(11)–C(16)	77.8(2)	C(7)–P(1)–C(11)–C(16)	–26.6(2)
Fe(1)–P(1)–C(11)–C(16)	–144.97(19)	C(16)–C(11)–C(12)–C(13)	0.8(4)
P(1)–C(11)–C(12)–C(13)	–179.9(2)	C(11)–C(12)–C(13)–C(14)	1.0(4)
C(12)–C(13)–C(14)–C(15)	–1.9(4)	C(13)–C(14)–C(15)–C(16)	0.9(4)
C(14)–C(15)–C(16)–C(11)	0.9(4)	C(12)–C(11)–C(16)–C(15)	–1.8(4)
P(1)–C(11)–C(16)–C(15)	179.0(2)	C(11)–P(1)–C(17)–C(22)	8.2(2)
C(7)–P(1)–C(17)–C(22)	115.0(2)	Fe(1)–P(1)–C(17)–C(22)	–126.13(18)
C(11)–P(1)–C(17)–C(18)	–170.94(19)	C(7)–P(1)–C(17)–C(18)	–64.1(2)
Fe(1)–P(1)–C(17)–C(18)	54.7(2)	C(22)–C(17)–C(18)–C(19)	–0.7(4)
P(1)–C(17)–C(18)–C(19)	178.49(18)	C(17)–C(18)–C(19)–C(20)	–0.2(4)
C(18)–C(19)–C(20)–C(21)	0.4(4)	C(19)–C(20)–C(21)–C(22)	0.3(4)
C(20)–C(21)–C(22)–C(17)	–1.2(4)	C(18)–C(17)–C(22)–C(21)	1.4(4)
P(1)–C(17)–C(22)–C(21)	–177.76(19)	C(8)–P(2)–C(23)–C(28)	12.4(2)
C(9)–P(2)–C(23)–C(28)	–92.8(2)	Fe(1)–P(2)–C(23)–C(28)	131.25(19)
C(8)–P(2)–C(23)–C(24)	–170.14(19)	C(9)–P(2)–C(23)–C(24)	84.6(2)
Fe(1)–P(2)–C(23)–C(24)	–51.3(2)	C(28)–C(23)–C(24)–C(25)	1.5(4)
P(2)–C(23)–C(24)–C(25)	–176.07(19)	C(23)–C(24)–C(25)–C(26)	–1.3(4)
C(24)–C(25)–C(26)–C(27)	0.2(4)	C(25)–C(26)–C(27)–C(28)	0.8(4)
C(26)–C(27)–C(28)–C(23)	–0.6(4)	C(24)–C(23)–C(28)–C(27)	–0.5(4)
P(2)–C(23)–C(28)–C(27)	176.9(2)	C(35)–P(3)–C(29)–C(30)	11.3(3)
C(10)–P(3)–C(29)–C(30)	117.9(3)	Fe(2)–P(3)–C(29)–C(30)	–114.9(3)
C(35)–P(3)–C(29)–C(34)	–171.2(2)	C(10)–P(3)–C(29)–C(34)	–64.6(3)
Fe(2)–P(3)–C(29)–C(34)	62.6(3)	C(34)–C(29)–C(30)–C(31)	0.5(7)
P(3)–C(29)–C(30)–C(31)	178.1(4)	C(29)–C(30)–C(31)–C(32)	–0.5(9)
C(30)–C(31)–C(32)–C(33)	0.7(8)	C(31)–C(32)–C(33)–C(34)	–0.9(6)
C(30)–C(29)–C(34)–C(33)	–0.8(5)	P(3)–C(29)–C(34)–C(33)	–178.4(3)
C(32)–C(33)–C(34)–C(29)	1.0(5)	C(29)–P(3)–C(35)–C(36)	–93.9(2)
C(10)–P(3)–C(35)–C(36)	160.90(19)	Fe(2)–P(3)–C(35)–C(36)	33.3(2)
C(29)–P(3)–C(35)–C(40)	84.1(2)	C(10)–P(3)–C(35)–C(40)	–21.1(2)
Fe(2)–P(3)–C(35)–C(40)	–148.74(19)	C(40)–C(35)–C(36)–C(37)	–0.6(4)
P(3)–C(35)–C(36)–C(37)	177.5(2)	C(35)–C(36)–C(37)–C(38)	1.4(4)

C(36)–C(37)–C(38)–C(39)	–0.7(4)	C(37)–C(38)–C(39)–C(40)	–0.8(5)
C(38)–C(39)–C(40)–C(35)	1.6(4)	C(36)–C(35)–C(40)–C(39)	–0.9(4)
P(3)–C(35)–C(40)–C(39)	–178.9(2)		





(vii) Table 1. Crystal data and structure refinement for str0735 (**3**).

Identification code	str0735	
Chemical formula	$C_{46}H_{44.50}Fe_2NO_{3.25}P_3S_2$	
Formula weight	932.06	
Temperature	150(2) K	
Radiation, wavelength	MoK $\alpha$ , 0.71073 Å	
Crystal system, space group	monoclinic, C2/c	
Unit cell parameters	a = 47.771(9) Å	$\alpha = 90^\circ$
	b = 11.747(2) Å	$\beta = 97.984(4)^\circ$
	c = 15.183(3) Å	$\gamma = 90^\circ$
Cell volume	8438(3) Å <sup>3</sup>	
Z	8	
Calculated density	1.467 g/cm <sup>3</sup>	
Absorption coefficient $\mu$	0.944 mm <sup>-1</sup>	
F(000)	3860	
Crystal colour and size	red, 0.14 × 0.13 × 0.09 mm <sup>3</sup>	
Data collection method	Bruker SMART APEX diffractometer	
	$\omega$ rotation with narrow frames	
$\theta$ range for data collection	2.45 to 28.34°	
Index ranges	h -61 to 63, k -15 to 15, l -20 to 20	
Completeness to $\theta = 26.00^\circ$	99.4 %	
Reflections collected	35672	
Independent reflections	10136 ( $R_{int} = 0.0550$ )	
Reflections with $F^2 > 2\sigma$	7618	
Absorption correction	semi-empirical from equivalents	
Min. and max. transmission	0.8792 and 0.9199	
Structure solution	direct methods	
Refinement method	Full-matrix least-squares on $F^2$	
Weighting parameters a, b	0.0609, 2.3320	
Data / restraints / parameters	10136 / 0 / 518	
Final R indices [ $F^2 > 2\sigma$ ]	R1 = 0.0500, wR2 = 0.1107	
R indices (all data)	R1 = 0.0736, wR2 = 0.1203	
Goodness-of-fit on $F^2$	1.049	
Largest and mean shift/su	0.001 and 0.000	
Largest diff. peak and hole	0.707 and -0.497 e Å <sup>-3</sup>	

Table 2. Atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ ) for str0735.  $U_{\text{eq}}$  is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

	x	y	z	$U_{\text{eq}}$
Fe(1)	0.092868(8)	0.18890(3)	0.49035(2)	0.01557(10)
Fe(2)	0.139123(8)	0.28551(3)	0.46162(3)	0.01703(10)
P(1)	0.056025(15)	0.23383(6)	0.38739(5)	0.01758(15)
P(2)	0.094368(15)	0.02960(6)	0.41550(5)	0.01703(15)
P(3)	0.165583(15)	0.16558(6)	0.39719(5)	0.01996(16)
S(1)	0.104820(14)	0.36803(6)	0.52766(4)	0.01828(15)
S(2)	0.135642(14)	0.14610(6)	0.56169(4)	0.01764(15)
N(1)	0.14372(5)	0.3269(2)	0.67774(15)	0.0230(5)
O(1)	0.05954(5)	0.1263(2)	0.62946(14)	0.0362(5)
O(2)	0.11626(5)	0.37997(19)	0.28922(14)	0.0332(5)
O(3)	0.18251(5)	0.45444(19)	0.51976(16)	0.0387(6)
C(1)	0.07194(6)	0.1497(2)	0.57127(19)	0.0222(6)
C(2)	0.12483(6)	0.3423(2)	0.35745(19)	0.0220(6)
C(3)	0.16625(6)	0.3848(3)	0.49867(19)	0.0243(6)
C(4)	0.05143(6)	0.1217(2)	0.30127(18)	0.0221(6)
C(5)	0.06068(6)	0.0065(2)	0.34296(19)	0.0227(6)
C(6)	0.11957(6)	0.0222(2)	0.33465(18)	0.0222(6)
C(7)	0.15066(6)	0.0219(2)	0.37784(19)	0.0222(6)
C(8)	0.11705(7)	0.3812(3)	0.64706(18)	0.0262(7)
C(9)	0.14102(6)	0.2045(2)	0.67418(18)	0.0221(6)
C(10)	0.15443(6)	0.3673(3)	0.76788(19)	0.0269(7)
C(11)	0.18349(6)	0.3212(3)	0.8008(2)	0.0268(7)
C(12)	0.18846(7)	0.2665(3)	0.8821(2)	0.0360(8)
C(13)	0.21497(9)	0.2255(4)	0.9137(2)	0.0499(10)
C(14)	0.23697(8)	0.2400(4)	0.8652(3)	0.0521(11)
C(15)	0.23245(8)	0.2932(3)	0.7841(3)	0.0457(9)
C(16)	0.20568(7)	0.3328(3)	0.7515(2)	0.0365(8)
C(17)	0.02234(6)	0.2303(2)	0.43347(18)	0.0204(6)
C(18)	0.01701(6)	0.3164(3)	0.4921(2)	0.0275(7)
C(19)	-0.00734(7)	0.3153(3)	0.5317(2)	0.0324(7)
C(20)	-0.02690(7)	0.2284(3)	0.5129(2)	0.0325(7)
C(21)	-0.02162(7)	0.1414(3)	0.4569(2)	0.0326(7)
C(22)	0.00289(6)	0.1420(3)	0.4167(2)	0.0264(7)
C(23)	0.05115(6)	0.3649(2)	0.32049(18)	0.0206(6)
C(24)	0.05652(6)	0.4709(2)	0.35941(19)	0.0258(6)
C(25)	0.05268(6)	0.5699(3)	0.3103(2)	0.0287(7)
C(26)	0.04329(7)	0.5662(3)	0.2208(2)	0.0357(8)
C(27)	0.03789(9)	0.4615(3)	0.1805(2)	0.0511(11)
C(28)	0.04169(8)	0.3620(3)	0.2296(2)	0.0405(9)
C(29)	0.09996(6)	-0.1073(2)	0.47225(18)	0.0192(6)
C(30)	0.09848(6)	-0.1150(2)	0.56323(19)	0.0245(6)
C(31)	0.10041(8)	-0.2196(3)	0.6055(2)	0.0354(8)
C(32)	0.10444(8)	-0.3172(3)	0.5585(2)	0.0389(8)
C(33)	0.10672(7)	-0.3111(2)	0.4697(2)	0.0307(7)
C(34)	0.10409(6)	-0.2071(2)	0.4260(2)	0.0268(7)
C(35)	0.20089(6)	0.1296(2)	0.4542(2)	0.0248(6)
C(36)	0.20770(7)	0.1417(4)	0.5451(2)	0.0444(9)
C(37)	0.23354(8)	0.1043(4)	0.5885(3)	0.0549(11)
C(38)	0.25308(7)	0.0584(3)	0.5413(3)	0.0460(9)



C(39)	0.24717(7)	0.0507(4)	0.4500(3)	0.0482(10)
C(40)	0.22114(7)	0.0836(3)	0.4077(2)	0.0417(9)
C(41)	0.17341(6)	0.2084(2)	0.28781(19)	0.0227(6)
C(42)	0.16830(7)	0.1419(3)	0.2117(2)	0.0352(8)
C(43)	0.17407(8)	0.1832(3)	0.1304(2)	0.0417(9)
C(44)	0.18527(7)	0.2899(3)	0.1244(2)	0.0370(8)
C(45)	0.19067(8)	0.3559(3)	0.1988(2)	0.0388(8)
C(46)	0.18475(7)	0.3165(3)	0.2799(2)	0.0327(7)
O(4)	-0.0025(12)	-0.126(2)	0.270(3)	0.224(14)

Table 3. Bond lengths [Å] and angles [°] for str0735.

Fe(1)–C(1)	1.750(3)	Fe(1)–P(2)	2.1957(8)
Fe(1)–S(1)	2.2328(8)	Fe(1)–S(2)	2.2331(8)
Fe(1)–P(1)	2.2485(9)	Fe(1)–Fe(2)	2.5746(7)
Fe(2)–C(2)	1.764(3)	Fe(2)–C(3)	1.775(3)
Fe(2)–P(3)	2.2098(8)	Fe(2)–S(1)	2.2554(8)
Fe(2)–S(2)	2.2558(8)	P(1)–C(23)	1.841(3)
P(1)–C(17)	1.842(3)	P(1)–C(4)	1.847(3)
P(2)–C(29)	1.827(3)	P(2)–C(6)	1.837(3)
P(2)–C(5)	1.839(3)	P(3)–C(41)	1.823(3)
P(3)–C(35)	1.835(3)	P(3)–C(7)	1.840(3)
S(1)–C(8)	1.832(3)	S(2)–C(9)	1.825(3)
N(1)–C(8)	1.443(4)	N(1)–C(9)	1.445(4)
N(1)–C(10)	1.471(4)	O(1)–C(1)	1.163(3)
O(2)–C(2)	1.148(3)	O(3)–C(3)	1.143(3)
C(4)–C(5)	1.533(4)	C(6)–C(7)	1.538(4)
C(10)–C(11)	1.509(4)	C(11)–C(12)	1.383(4)
C(11)–C(16)	1.387(4)	C(12)–C(13)	1.377(5)
C(13)–C(14)	1.375(6)	C(14)–C(15)	1.372(6)
C(15)–C(16)	1.386(5)	C(17)–C(22)	1.393(4)
C(17)–C(18)	1.394(4)	C(18)–C(19)	1.382(4)
C(19)–C(20)	1.387(4)	C(20)–C(21)	1.376(5)
C(21)–C(22)	1.393(4)	C(23)–C(24)	1.387(4)
C(23)–C(28)	1.392(4)	C(24)–C(25)	1.379(4)
C(25)–C(26)	1.371(4)	C(26)–C(27)	1.382(5)
C(27)–C(28)	1.384(4)	C(29)–C(34)	1.394(4)
C(29)–C(30)	1.396(4)	C(30)–C(31)	1.383(4)
C(31)–C(32)	1.379(5)	C(32)–C(33)	1.369(5)
C(33)–C(34)	1.388(4)	C(35)–C(36)	1.380(4)
C(35)–C(40)	1.384(4)	C(36)–C(37)	1.388(5)
C(37)–C(38)	1.365(5)	C(38)–C(39)	1.379(5)
C(39)–C(40)	1.374(5)	C(41)–C(42)	1.388(4)
C(41)–C(46)	1.392(4)	C(42)–C(43)	1.389(4)
C(43)–C(44)	1.371(5)	C(44)–C(45)	1.365(5)
C(45)–C(46)	1.381(4)	O(4)–O(4A)	0.67(9)
C(1)–Fe(1)–P(2)	101.67(10)	C(1)–Fe(1)–S(1)	102.74(10)
P(2)–Fe(1)–S(1)	155.37(3)	C(1)–Fe(1)–S(2)	100.00(10)
P(2)–Fe(1)–S(2)	88.03(3)	S(1)–Fe(1)–S(2)	84.53(3)
C(1)–Fe(1)–P(1)	94.68(10)	P(2)–Fe(1)–P(1)	85.35(3)
S(1)–Fe(1)–P(1)	95.91(3)	S(2)–Fe(1)–P(1)	164.85(3)
C(1)–Fe(1)–Fe(2)	145.32(10)	P(2)–Fe(1)–Fe(2)	101.39(3)
S(1)–Fe(1)–Fe(2)	55.41(2)	S(2)–Fe(1)–Fe(2)	55.42(2)
P(1)–Fe(1)–Fe(2)	112.71(3)	C(2)–Fe(2)–C(3)	102.13(13)
C(2)–Fe(2)–P(3)	91.04(9)	C(3)–Fe(2)–P(3)	96.99(10)
C(2)–Fe(2)–S(1)	91.22(9)	C(3)–Fe(2)–S(1)	96.91(9)
P(3)–Fe(2)–S(1)	165.11(3)	C(2)–Fe(2)–S(2)	145.37(9)
C(3)–Fe(2)–S(2)	112.47(10)	P(3)–Fe(2)–S(2)	86.30(3)
S(1)–Fe(2)–S(2)	83.49(3)	C(2)–Fe(2)–Fe(1)	94.83(9)
C(3)–Fe(2)–Fe(1)	147.31(9)	P(3)–Fe(2)–Fe(1)	110.56(3)
S(1)–Fe(2)–Fe(1)	54.58(2)	S(2)–Fe(2)–Fe(1)	54.59(2)
C(23)–P(1)–C(17)	100.25(12)	C(23)–P(1)–C(4)	102.25(13)
C(17)–P(1)–C(4)	103.21(13)	C(23)–P(1)–Fe(1)	127.35(9)
C(17)–P(1)–Fe(1)	111.80(9)	C(4)–P(1)–Fe(1)	109.23(9)

C(29)–P(2)–C(6)	101.95(13)	C(29)–P(2)–C(5)	102.19(13)
C(6)–P(2)–C(5)	101.13(13)	C(29)–P(2)–Fe(1)	121.30(9)
C(6)–P(2)–Fe(1)	117.24(9)	C(5)–P(2)–Fe(1)	110.29(9)
C(41)–P(3)–C(35)	101.64(13)	C(41)–P(3)–C(7)	103.40(13)
C(35)–P(3)–C(7)	100.06(13)	C(41)–P(3)–Fe(2)	115.45(10)
C(35)–P(3)–Fe(2)	118.87(10)	C(7)–P(3)–Fe(2)	115.01(9)
C(8)–S(1)–Fe(1)	111.46(10)	C(8)–S(1)–Fe(2)	108.95(10)
Fe(1)–S(1)–Fe(2)	70.01(2)	C(9)–S(2)–Fe(1)	111.69(10)
C(9)–S(2)–Fe(2)	110.09(9)	Fe(1)–S(2)–Fe(2)	69.99(2)
C(8)–N(1)–C(9)	111.0(2)	C(8)–N(1)–C(10)	109.6(2)
C(9)–N(1)–C(10)	111.9(2)	O(1)–C(1)–Fe(1)	175.3(3)
O(2)–C(2)–Fe(2)	178.1(3)	O(3)–C(3)–Fe(2)	175.2(3)
C(5)–C(4)–P(1)	109.76(18)	C(4)–C(5)–P(2)	107.17(19)
C(7)–C(6)–P(2)	113.51(19)	C(6)–C(7)–P(3)	113.30(19)
N(1)–C(8)–S(1)	115.32(19)	N(1)–C(9)–S(2)	114.16(19)
N(1)–C(10)–C(11)	112.2(2)	C(12)–C(11)–C(16)	118.6(3)
C(12)–C(11)–C(10)	119.9(3)	C(16)–C(11)–C(10)	121.5(3)
C(13)–C(12)–C(11)	120.6(3)	C(14)–C(13)–C(12)	120.3(4)
C(15)–C(14)–C(13)	120.0(4)	C(14)–C(15)–C(16)	119.7(4)
C(15)–C(16)–C(11)	120.8(3)	C(22)–C(17)–C(18)	118.7(3)
C(22)–C(17)–P(1)	122.9(2)	C(18)–C(17)–P(1)	118.2(2)
C(19)–C(18)–C(17)	120.8(3)	C(18)–C(19)–C(20)	120.0(3)
C(21)–C(20)–C(19)	119.9(3)	C(20)–C(21)–C(22)	120.3(3)
C(17)–C(22)–C(21)	120.2(3)	C(24)–C(23)–C(28)	117.4(3)
C(24)–C(23)–P(1)	120.9(2)	C(28)–C(23)–P(1)	121.7(2)
C(25)–C(24)–C(23)	121.6(3)	C(26)–C(25)–C(24)	120.6(3)
C(25)–C(26)–C(27)	118.8(3)	C(26)–C(27)–C(28)	120.7(3)
C(27)–C(28)–C(23)	120.8(3)	C(34)–C(29)–C(30)	118.3(3)
C(34)–C(29)–P(2)	121.6(2)	C(30)–C(29)–P(2)	120.0(2)
C(31)–C(30)–C(29)	120.5(3)	C(32)–C(31)–C(30)	120.2(3)
C(33)–C(32)–C(31)	120.2(3)	C(32)–C(33)–C(34)	120.2(3)
C(33)–C(34)–C(29)	120.6(3)	C(36)–C(35)–C(40)	117.7(3)
C(36)–C(35)–P(3)	121.4(2)	C(40)–C(35)–P(3)	120.7(2)
C(35)–C(36)–C(37)	120.9(3)	C(38)–C(37)–C(36)	120.2(4)
C(37)–C(38)–C(39)	119.6(3)	C(40)–C(39)–C(38)	119.9(3)
C(39)–C(40)–C(35)	121.5(3)	C(42)–C(41)–C(46)	117.8(3)
C(42)–C(41)–P(3)	124.6(2)	C(46)–C(41)–P(3)	117.5(2)
C(41)–C(42)–C(43)	120.6(3)	C(44)–C(43)–C(42)	120.4(3)
C(45)–C(44)–C(43)	119.7(3)	C(44)–C(45)–C(46)	120.5(3)
C(45)–C(46)–C(41)	120.9(3)		

Symmetry operations for equivalent atoms

A  $-x, y, -z+1/2$

Table 4. Anisotropic displacement parameters ( $\text{\AA}^2$ ) for str0735. The anisotropic displacement factor exponent takes the form:  $-2\pi^2[h^2a^{*2}U^{11} + \dots + 2hka^*b^*U^{12}]$

	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	U <sup>12</sup>
Fe(1)	0.0178(2)	0.01426(19)	0.01475(19)	-0.00009(14)	0.00249(15)	
-0.00046(14)						
Fe(2)	0.0197(2)	0.0156(2)	0.0160(2)	0.00018(15)	0.00310(15)	
-0.00198(15)						
P(1)	0.0194(4)	0.0175(4)	0.0156(3)	0.0001(3)	0.0017(3)	-0.0006(3)
P(2)	0.0203(4)	0.0138(3)	0.0172(3)	-0.0008(3)	0.0034(3)	-0.0013(3)
P(3)	0.0212(4)	0.0198(4)	0.0198(4)	-0.0014(3)	0.0064(3)	-0.0017(3)
S(1)	0.0226(4)	0.0149(3)	0.0168(3)	-0.0003(2)	0.0007(3)	0.0011(3)
S(2)	0.0199(3)	0.0154(3)	0.0173(3)	0.0001(2)	0.0013(3)	-0.0003(2)
N(1)	0.0318(14)	0.0190(12)	0.0167(12)	-0.0012(9)	-0.0021(10)	0.0013(10)
O(1)	0.0387(13)	0.0433(14)	0.0298(12)	0.0099(10)	0.0160(10)	0.0019(11)
O(2)	0.0347(13)	0.0436(14)	0.0218(11)	0.0085(10)	0.0056(10)	0.0041(10)
O(3)	0.0333(13)	0.0355(13)	0.0485(15)	-0.0139(11)	0.0095(11)	-0.0151(11)
C(1)	0.0226(15)	0.0214(15)	0.0222(15)	-0.0003(11)	0.0024(12)	0.0033(11)
C(2)	0.0214(15)	0.0217(15)	0.0240(15)	0.0010(12)	0.0070(12)	-0.0017(11)
C(3)	0.0255(16)	0.0244(16)	0.0238(15)	-0.0029(12)	0.0059(12)	0.0007(12)
C(4)	0.0259(15)	0.0236(15)	0.0157(14)	-0.0017(11)	-0.0011(11)	0.0002(12)
C(5)	0.0251(15)	0.0185(14)	0.0227(15)	-0.0034(11)	-0.0025(12)	-0.0009(11)
C(6)	0.0281(16)	0.0210(14)	0.0187(14)	-0.0039(11)	0.0073(12)	-0.0039(12)
C(7)	0.0258(15)	0.0181(14)	0.0242(15)	-0.0007(11)	0.0091(12)	0.0011(11)
C(8)	0.0364(17)	0.0250(16)	0.0158(14)	-0.0059(11)	-0.0012(12)	0.0060(13)
C(9)	0.0309(16)	0.0206(14)	0.0139(13)	0.0018(11)	0.0003(11)	-0.0013(12)
C(10)	0.0291(17)	0.0302(17)	0.0210(15)	-0.0041(12)	0.0018(12)	0.0022(13)
C(11)	0.0259(16)	0.0266(16)	0.0266(16)	-0.0070(13)	-0.0009(13)	-0.0028(12)
C(12)	0.0361(19)	0.048(2)	0.0227(17)	-0.0055(15)	-0.0003(14)	0.0032(16)
C(13)	0.053(2)	0.066(3)	0.0280(19)	0.0001(18)	-0.0055(17)	0.016(2)
C(14)	0.032(2)	0.077(3)	0.044(2)	-0.009(2)	-0.0075(18)	0.0101(19)
C(15)	0.030(2)	0.056(2)	0.052(2)	-0.0065(19)	0.0103(17)	-0.0017(17)
C(16)	0.0371(19)	0.0379(19)	0.0342(19)	0.0007(15)	0.0042(15)	-0.0018(15)
C(17)	0.0180(14)	0.0243(15)	0.0178(14)	0.0049(11)	-0.0012(11)	-0.0001(11)
C(18)	0.0225(16)	0.0323(17)	0.0286(16)	-0.0037(13)	0.0063(13)	-0.0033(13)
C(19)	0.0293(17)	0.0385(19)	0.0303(17)	-0.0038(14)	0.0076(14)	0.0003(14)
C(20)	0.0218(16)	0.042(2)	0.0349(18)	0.0031(15)	0.0097(13)	-0.0022(14)
C(21)	0.0253(17)	0.0314(18)	0.0415(19)	0.0032(14)	0.0064(14)	-0.0075(13)
C(22)	0.0241(16)	0.0272(16)	0.0273(16)	-0.0030(12)	0.0013(13)	-0.0029(12)
C(23)	0.0186(14)	0.0223(14)	0.0211(14)	0.0033(11)	0.0040(11)	0.0008(11)
C(24)	0.0272(16)	0.0271(16)	0.0216(15)	0.0016(12)	-0.0018(12)	0.0013(12)
C(25)	0.0304(17)	0.0216(15)	0.0327(17)	0.0008(13)	-0.0003(13)	0.0005(13)
C(26)	0.046(2)	0.0279(17)	0.0306(18)	0.0122(14)	-0.0047(15)	-0.0016(15)
C(27)	0.090(3)	0.031(2)	0.0250(18)	0.0092(15)	-0.0165(19)	-0.0075(19)
C(28)	0.065(2)	0.0251(17)	0.0265(17)	0.0016(14)	-0.0106(16)	-0.0068(16)
C(29)	0.0185(14)	0.0150(13)	0.0242(15)	0.0019(11)	0.0030(11)	-0.0028(10)
C(30)	0.0315(17)	0.0185(14)	0.0233(15)	-0.0010(11)	0.0027(12)	-0.0020(12)
C(31)	0.052(2)	0.0275(17)	0.0257(17)	0.0030(13)	0.0007(15)	-0.0032(15)
C(32)	0.058(2)	0.0180(16)	0.039(2)	0.0078(14)	0.0000(17)	-0.0014(15)
C(33)	0.0400(19)	0.0149(15)	0.0375(18)	-0.0029(13)	0.0070(15)	0.0011(13)
C(34)	0.0310(17)	0.0220(15)	0.0282(16)	-0.0041(12)	0.0068(13)	-0.0023(12)
C(35)	0.0215(15)	0.0243(15)	0.0290(16)	-0.0005(12)	0.0052(12)	-0.0011(12)
C(36)	0.0260(18)	0.072(3)	0.035(2)	-0.0116(18)	0.0024(15)	0.0120(17)
C(37)	0.035(2)	0.091(3)	0.036(2)	-0.010(2)	-0.0044(17)	0.017(2)
C(38)	0.0212(17)	0.058(2)	0.057(2)	-0.0069(19)	-0.0017(16)	0.0053(16)
C(39)	0.0220(18)	0.072(3)	0.052(2)	-0.010(2)	0.0097(16)	0.0048(17)
C(40)	0.0309(19)	0.056(2)	0.039(2)	-0.0044(17)	0.0090(15)	0.0052(17)
C(41)	0.0222(15)	0.0258(15)	0.0218(15)	0.0018(12)	0.0089(12)	-0.0011(12)
C(42)	0.042(2)	0.0372(19)	0.0289(17)	-0.0061(14)	0.0130(15)	-0.0122(15)
C(43)	0.047(2)	0.056(2)	0.0238(17)	-0.0091(16)	0.0108(15)	-0.0124(18)
C(44)	0.0371(19)	0.050(2)	0.0267(17)	0.0095(15)	0.0125(15)	0.0009(16)
C(45)	0.055(2)	0.0270(17)	0.039(2)	0.0039(15)	0.0231(17)	-0.0044(16)
C(46)	0.046(2)	0.0271(17)	0.0280(17)	-0.0037(13)	0.0154(15)	-0.0052(14)

Table 5. Hydrogen coordinates and isotropic displacement parameters ( $\text{\AA}^2$ ) for str0735.

	x	y	z	U
H(4A)	0.0313	0.1179	0.2744	0.027
H(4B)	0.0629	0.1402	0.2537	0.027
H(5A)	0.0632	-0.0495	0.2960	0.027
H(5B)	0.0462	-0.0228	0.3780	0.027
H(6A)	0.1158	-0.0478	0.2986	0.027
H(6B)	0.1164	0.0881	0.2939	0.027
H(7A)	0.1620	-0.0205	0.3390	0.027
H(7B)	0.1521	-0.0189	0.4354	0.027
H(8A)	0.1187	0.4631	0.6622	0.031
H(8B)	0.1025	0.3484	0.6800	0.031
H(9A)	0.1249	0.1819	0.7048	0.027
H(9B)	0.1583	0.1703	0.7072	0.027
H(10A)	0.1412	0.3436	0.8092	0.032
H(10B)	0.1551	0.4515	0.7678	0.032
H(12A)	0.1735	0.2572	0.9166	0.043
H(13A)	0.2181	0.1869	0.9692	0.060
H(14A)	0.2553	0.2132	0.8880	0.062
H(15A)	0.2476	0.3029	0.7503	0.055
H(16A)	0.2025	0.3682	0.6947	0.044
H(18A)	0.0303	0.3764	0.5050	0.033
H(19A)	-0.0107	0.3742	0.5719	0.039
H(20A)	-0.0439	0.2290	0.5387	0.039
H(21A)	-0.0347	0.0807	0.4455	0.039
H(22A)	0.0063	0.0819	0.3777	0.032
H(24A)	0.0630	0.4755	0.4214	0.031
H(25A)	0.0566	0.6413	0.3388	0.034
H(26A)	0.0405	0.6344	0.1871	0.043
H(27A)	0.0315	0.4577	0.1185	0.061
H(28A)	0.0378	0.2908	0.2008	0.049
H(30A)	0.0961	-0.0479	0.5964	0.029
H(31A)	0.0990	-0.2241	0.6672	0.042
H(32A)	0.1056	-0.3889	0.5877	0.047
H(33A)	0.1101	-0.3782	0.4379	0.037
H(34A)	0.1051	-0.2038	0.3640	0.032
H(36A)	0.1945	0.1760	0.5783	0.053
H(37A)	0.2376	0.1108	0.6514	0.066
H(38A)	0.2707	0.0318	0.5712	0.055
H(39A)	0.2611	0.0227	0.4165	0.058
H(40A)	0.2169	0.0746	0.3451	0.050
H(42A)	0.1608	0.0675	0.2153	0.042
H(43A)	0.1702	0.1372	0.0787	0.050
H(44A)	0.1892	0.3177	0.0687	0.044
H(45A)	0.1986	0.4297	0.1948	0.047
H(46A)	0.1885	0.3637	0.3309	0.039

Table 6. Torsion angles [°] for str0735.

C(1)–Fe(1)–Fe(2)–C(2)	–146.35(19)	P(2)–Fe(1)–Fe(2)–C(2)	82.83(10)
S(1)–Fe(1)–Fe(2)–C(2)	–88.07(10)	S(2)–Fe(1)–Fe(2)–C(2)	162.38(10)
P(1)–Fe(1)–Fe(2)–C(2)	–6.86(10)	C(1)–Fe(1)–Fe(2)–C(3)	–24.9(2)
P(2)–Fe(1)–Fe(2)–C(3)	–155.68(18)	S(1)–Fe(1)–Fe(2)–C(3)	33.42(18)
S(2)–Fe(1)–Fe(2)–C(3)	–76.14(18)	P(1)–Fe(1)–Fe(2)–C(3)	114.62(18)
C(1)–Fe(1)–Fe(2)–P(3)	120.72(16)	P(2)–Fe(1)–Fe(2)–P(3)	–10.10(3)
S(1)–Fe(1)–Fe(2)–P(3)	179.00(3)	S(2)–Fe(1)–Fe(2)–P(3)	69.44(3)
P(1)–Fe(1)–Fe(2)–P(3)	–99.80(4)	C(1)–Fe(1)–Fe(2)–S(1)	–58.28(16)
P(2)–Fe(1)–Fe(2)–S(1)	170.90(3)	S(2)–Fe(1)–Fe(2)–S(1)	–109.56(3)
P(1)–Fe(1)–Fe(2)–S(1)	81.20(3)	C(1)–Fe(1)–Fe(2)–S(2)	51.28(16)
P(2)–Fe(1)–Fe(2)–S(2)	–79.54(3)	S(1)–Fe(1)–Fe(2)–S(2)	109.56(3)
P(1)–Fe(1)–Fe(2)–S(2)	–169.24(3)	C(1)–Fe(1)–P(1)–C(23)	130.71(15)
P(2)–Fe(1)–P(1)–C(23)	–127.93(12)	S(1)–Fe(1)–P(1)–C(23)	27.36(12)
S(2)–Fe(1)–P(1)–C(23)	–63.54(18)	Fe(2)–Fe(1)–P(1)–C(23)	–27.52(12)
C(1)–Fe(1)–P(1)–C(17)	7.60(14)	P(2)–Fe(1)–P(1)–C(17)	108.95(10)
S(1)–Fe(1)–P(1)–C(17)	–95.76(10)	S(2)–Fe(1)–P(1)–C(17)	173.34(14)
Fe(2)–Fe(1)–P(1)–C(17)	–150.63(10)	C(1)–Fe(1)–P(1)–C(4)	–106.00(14)
P(2)–Fe(1)–P(1)–C(4)	–4.64(10)	S(1)–Fe(1)–P(1)–C(4)	150.65(10)
S(2)–Fe(1)–P(1)–C(4)	59.75(16)	Fe(2)–Fe(1)–P(1)–C(4)	95.77(10)
C(1)–Fe(1)–P(2)–C(29)	–43.46(14)	S(1)–Fe(1)–P(2)–C(29)	128.66(12)
S(2)–Fe(1)–P(2)–C(29)	56.35(11)	P(1)–Fe(1)–P(2)–C(29)	–137.29(11)
Fe(2)–Fe(1)–P(2)–C(29)	110.45(10)	C(1)–Fe(1)–P(2)–C(6)	–169.30(14)
S(1)–Fe(1)–P(2)–C(6)	2.82(14)	S(2)–Fe(1)–P(2)–C(6)	–69.49(11)
P(1)–Fe(1)–P(2)–C(6)	96.87(11)	Fe(2)–Fe(1)–P(2)–C(6)	–15.38(11)
C(1)–Fe(1)–P(2)–C(5)	75.76(14)	S(1)–Fe(1)–P(2)–C(5)	–112.12(12)
S(2)–Fe(1)–P(2)–C(5)	175.56(10)	P(1)–Fe(1)–P(2)–C(5)	–18.07(10)
Fe(2)–Fe(1)–P(2)–C(5)	–130.33(10)	C(2)–Fe(2)–P(3)–C(41)	23.81(14)
C(3)–Fe(2)–P(3)–C(41)	–78.55(14)	S(1)–Fe(2)–P(3)–C(41)	122.54(15)
S(2)–Fe(2)–P(3)–C(41)	169.25(11)	Fe(1)–Fe(2)–P(3)–C(41)	119.36(11)
C(2)–Fe(2)–P(3)–C(35)	144.97(14)	C(3)–Fe(2)–P(3)–C(35)	42.61(15)
S(1)–Fe(2)–P(3)–C(35)	–116.30(16)	S(2)–Fe(2)–P(3)–C(35)	–69.60(11)
Fe(1)–Fe(2)–P(3)–C(35)	–119.48(11)	C(2)–Fe(2)–P(3)–C(7)	–96.53(14)
C(3)–Fe(2)–P(3)–C(7)	161.11(14)	S(1)–Fe(2)–P(3)–C(7)	2.20(18)
S(2)–Fe(2)–P(3)–C(7)	48.90(11)	Fe(1)–Fe(2)–P(3)–C(7)	–0.98(11)
C(1)–Fe(1)–S(1)–C(8)	47.05(14)	P(2)–Fe(1)–S(1)–C(8)	–125.04(13)
S(2)–Fe(1)–S(1)–C(8)	–52.00(11)	P(1)–Fe(1)–S(1)–C(8)	143.22(11)
Fe(2)–Fe(1)–S(1)–C(8)	–103.20(11)	C(1)–Fe(1)–S(1)–Fe(2)	150.25(10)
P(2)–Fe(1)–S(1)–Fe(2)	–21.84(8)	S(2)–Fe(1)–S(1)–Fe(2)	51.20(3)
P(1)–Fe(1)–S(1)–Fe(2)	–113.58(3)	C(2)–Fe(2)–S(1)–C(8)	–158.28(14)
C(3)–Fe(2)–S(1)–C(8)	–55.90(15)	P(3)–Fe(2)–S(1)–C(8)	103.02(16)
S(2)–Fe(2)–S(1)–C(8)	56.04(11)	Fe(1)–Fe(2)–S(1)–C(8)	106.66(11)
C(2)–Fe(2)–S(1)–Fe(1)	95.06(9)	C(3)–Fe(2)–S(1)–Fe(1)	–162.56(10)
P(3)–Fe(2)–S(1)–Fe(1)	–3.65(13)	S(2)–Fe(2)–S(1)–Fe(1)	–50.62(2)
C(1)–Fe(1)–S(2)–C(9)	–48.81(14)	P(2)–Fe(1)–S(2)–C(9)	–150.31(10)
S(1)–Fe(1)–S(2)–C(9)	53.20(10)	P(1)–Fe(1)–S(2)–C(9)	145.63(14)
Fe(2)–Fe(1)–S(2)–C(9)	104.40(10)	C(1)–Fe(1)–S(2)–Fe(2)	–153.20(10)
P(2)–Fe(1)–S(2)–Fe(2)	105.29(3)	S(1)–Fe(1)–S(2)–Fe(2)	–51.19(3)
P(1)–Fe(1)–S(2)–Fe(2)	41.23(12)	C(2)–Fe(2)–S(2)–C(9)	–138.66(19)
C(3)–Fe(2)–S(2)–C(9)	38.83(15)	P(3)–Fe(2)–S(2)–C(9)	134.87(11)
S(1)–Fe(2)–S(2)–C(9)	–55.98(11)	Fe(1)–Fe(2)–S(2)–C(9)	–106.60(11)

C(2)–Fe(2)–S(2)–Fe(1)	–32.06(17)	C(3)–Fe(2)–S(2)–Fe(1)	145.43(10)
P(3)–Fe(2)–S(2)–Fe(1)	–118.53(3)	S(1)–Fe(2)–S(2)–Fe(1)	50.62(3)
P(2)–Fe(1)–C(1)–O(1)	125(3)	S(1)–Fe(1)–C(1)–O(1)	–52(3)
S(2)–Fe(1)–C(1)–O(1)	35(3)	P(1)–Fe(1)–C(1)–O(1)	–149(3)
Fe(2)–Fe(1)–C(1)–O(1)	–6(3)	C(3)–Fe(2)–C(2)–O(2)	40(8)
P(3)–Fe(2)–C(2)–O(2)	–57(8)	S(1)–Fe(2)–C(2)–O(2)	138(8)
S(2)–Fe(2)–C(2)–O(2)	–142(8)	Fe(1)–Fe(2)–C(2)–O(2)	–168(8)
C(2)–Fe(2)–C(3)–O(3)	36(3)	P(3)–Fe(2)–C(3)–O(3)	128(3)
S(1)–Fe(2)–C(3)–O(3)	–57(3)	S(2)–Fe(2)–C(3)–O(3)	–143(3)
Fe(1)–Fe(2)–C(3)–O(3)	–84(3)	C(23)–P(1)–C(4)–C(5)	168.5(2)
C(17)–P(1)–C(4)–C(5)	–87.7(2)	Fe(1)–P(1)–C(4)–C(5)	31.4(2)
P(1)–C(4)–C(5)–P(2)	–44.9(2)	C(29)–P(2)–C(5)–C(4)	171.95(19)
C(6)–P(2)–C(5)–C(4)	–83.1(2)	Fe(1)–P(2)–C(5)–C(4)	41.7(2)
C(29)–P(2)–C(6)–C(7)	–66.7(2)	C(5)–P(2)–C(6)–C(7)	–171.8(2)
Fe(1)–P(2)–C(6)–C(7)	68.2(2)	P(2)–C(6)–C(7)–P(3)	–87.8(2)
C(41)–P(3)–C(7)–C(6)	–78.9(2)	C(35)–P(3)–C(7)–C(6)	176.5(2)
Fe(2)–P(3)–C(7)–C(6)	47.9(2)	C(9)–N(1)–C(8)–S(1)	–70.0(3)
C(10)–N(1)–C(8)–S(1)	165.9(2)	Fe(1)–S(1)–C(8)–N(1)	68.5(2)
Fe(2)–S(1)–C(8)–N(1)	–6.8(2)	C(8)–N(1)–C(9)–S(2)	70.4(3)
C(10)–N(1)–C(9)–S(2)	–166.8(2)	Fe(1)–S(2)–C(9)–N(1)	–70.3(2)
Fe(2)–S(2)–C(9)–N(1)	5.5(2)	C(8)–N(1)–C(10)–C(11)	–175.1(2)
C(9)–N(1)–C(10)–C(11)	61.3(3)	N(1)–C(10)–C(11)–C(12)	–127.3(3)
N(1)–C(10)–C(11)–C(16)	53.0(4)	C(16)–C(11)–C(12)–C(13)	0.6(5)
C(10)–C(11)–C(12)–C(13)	–179.1(3)	C(11)–C(12)–C(13)–C(14)	1.0(6)
C(12)–C(13)–C(14)–C(15)	–1.6(6)	C(13)–C(14)–C(15)–C(16)	0.4(6)
C(14)–C(15)–C(16)–C(11)	1.3(6)	C(12)–C(11)–C(16)–C(15)	–1.8(5)
C(10)–C(11)–C(16)–C(15)	177.9(3)	C(23)–P(1)–C(17)–C(22)	118.6(2)
C(4)–P(1)–C(17)–C(22)	13.3(3)	Fe(1)–P(1)–C(17)–C(22)	–104.0(2)
C(23)–P(1)–C(17)–C(18)	–65.6(2)	C(4)–P(1)–C(17)–C(18)	–170.9(2)
Fe(1)–P(1)–C(17)–C(18)	71.9(2)	C(22)–C(17)–C(18)–C(19)	–1.1(4)
P(1)–C(17)–C(18)–C(19)	–177.2(2)	C(17)–C(18)–C(19)–C(20)	–0.4(5)
C(18)–C(19)–C(20)–C(21)	2.0(5)	C(19)–C(20)–C(21)–C(22)	–2.0(5)
C(18)–C(17)–C(22)–C(21)	1.1(4)	P(1)–C(17)–C(22)–C(21)	177.0(2)
C(20)–C(21)–C(22)–C(17)	0.4(5)	C(17)–P(1)–C(23)–C(24)	82.3(3)
C(4)–P(1)–C(23)–C(24)	–171.6(2)	Fe(1)–P(1)–C(23)–C(24)	–45.5(3)
C(17)–P(1)–C(23)–C(28)	–97.0(3)	C(4)–P(1)–C(23)–C(28)	9.1(3)
Fe(1)–P(1)–C(23)–C(28)	135.2(2)	C(28)–C(23)–C(24)–C(25)	0.0(5)
P(1)–C(23)–C(24)–C(25)	–179.3(2)	C(23)–C(24)–C(25)–C(26)	0.1(5)
C(24)–C(25)–C(26)–C(27)	–0.4(5)	C(25)–C(26)–C(27)–C(28)	0.4(6)
C(26)–C(27)–C(28)–C(23)	–0.3(6)	C(24)–C(23)–C(28)–C(27)	0.0(5)
P(1)–C(23)–C(28)–C(27)	179.4(3)	C(6)–P(2)–C(29)–C(34)	–40.2(3)
C(5)–P(2)–C(29)–C(34)	64.2(3)	Fe(1)–P(2)–C(29)–C(34)	–172.7(2)
C(6)–P(2)–C(29)–C(30)	143.7(2)	C(5)–P(2)–C(29)–C(30)	–112.0(2)
Fe(1)–P(2)–C(29)–C(30)	11.1(3)	C(34)–C(29)–C(30)–C(31)	–1.5(4)
P(2)–C(29)–C(30)–C(31)	174.8(2)	C(29)–C(30)–C(31)–C(32)	1.4(5)
C(30)–C(31)–C(32)–C(33)	0.4(6)	C(31)–C(32)–C(33)–C(34)	–2.1(5)
C(32)–C(33)–C(34)–C(29)	1.9(5)	C(30)–C(29)–C(34)–C(33)	–0.2(4)
P(2)–C(29)–C(34)–C(33)	–176.4(2)	C(41)–P(3)–C(35)–C(36)	150.5(3)
C(7)–P(3)–C(35)–C(36)	–103.4(3)	Fe(2)–P(3)–C(35)–C(36)	22.6(3)
C(41)–P(3)–C(35)–C(40)	–33.0(3)	C(7)–P(3)–C(35)–C(40)	73.1(3)
Fe(2)–P(3)–C(35)–C(40)	–160.9(2)	C(40)–C(35)–C(36)–C(37)	–2.7(6)
P(3)–C(35)–C(36)–C(37)	173.9(3)	C(35)–C(36)–C(37)–C(38)	2.2(7)

C(36)–C(37)–C(38)–C(39)	0.9(7)	C(37)–C(38)–C(39)–C(40)	–3.5(6)
C(38)–C(39)–C(40)–C(35)	3.0(6)	C(36)–C(35)–C(40)–C(39)	0.1(6)
P(3)–C(35)–C(40)–C(39)	–176.6(3)	C(35)–P(3)–C(41)–C(42)	103.8(3)
C(7)–P(3)–C(41)–C(42)	0.4(3)	Fe(2)–P(3)–C(41)–C(42)	–126.1(3)
C(35)–P(3)–C(41)–C(46)	–77.7(3)	C(7)–P(3)–C(41)–C(46)	178.9(2)
Fe(2)–P(3)–C(41)–C(46)	52.4(3)	C(46)–C(41)–C(42)–C(43)	–0.7(5)
P(3)–C(41)–C(42)–C(43)	177.8(3)	C(41)–C(42)–C(43)–C(44)	0.8(6)
C(42)–C(43)–C(44)–C(45)	–0.2(6)	C(43)–C(44)–C(45)–C(46)	–0.5(6)
C(44)–C(45)–C(46)–C(41)	0.6(6)	C(42)–C(41)–C(46)–C(45)	0.0(5)
P(3)–C(41)–C(46)–C(45)	–178.6(3)		

Symmetry operations for equivalent atoms

