

Supporting Information

A highly luminescent Mn⁴⁺ activated LaAlO₃ far-red-emitting phosphor for plant growth LEDs: Charge compensation induced Mn⁴⁺ incorporation

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Table S1. The decay lifetimes of LAO:Mn⁴⁺,Mg²⁺ with different concentrations of Mn⁴⁺.

Concentration of Mn ⁴⁺ / mol%	Monitoring wavelength / nm	Lifetime / ms
0.05	729	3.4
0.1	729	3.3
0.3	729	2.4
0.5	729	2.1
0.7	729	1.7

Table S2. The chromaticity coordinates of LAO:Mn⁴⁺,Mg²⁺ in 298–433 K.

Temperature/ K	Chromaticity coordinates (x, y)
298	0.7335, 0.2665
313	0.7334, 0.2666
333	0.7333, 0.2667
353	0.7332, 0.2668
373	0.7330, 0.2670
393	0.7328, 0.2672
413	0.7326, 0.2674
433	0.7317, 0.2683

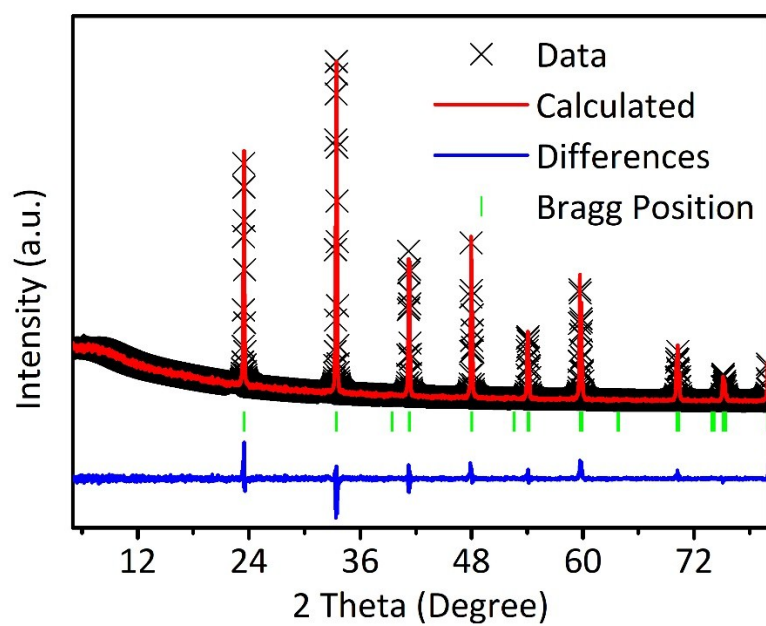


Fig. S1. Rietveld refinement patterns LAO:0.001Mn⁴⁺ phosphor.

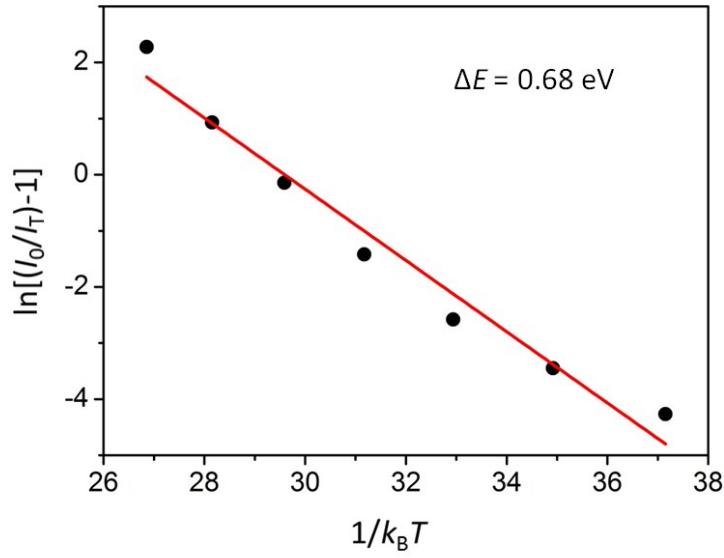


Fig. S2. The plotting line of $\ln[(I_0/I_T)-1]$ vs. $1/k_B T$ for LAO:Mn⁴⁺,Mg²⁺ phosphor.

The value of ΔE is calculated based on a modified Arrhenius equation (S1):

$$I_T = \frac{I_0}{1 + A \exp(-\Delta E / k_B T)} \quad (S1)$$

in which I_0 is the initial PL intensity at room temperature; I_T is the PL intensity at a given temperature T ; ΔE is the activation energy; k_B and A refer to Boltzmann constant (8.617E-5 eV/K) and an absolute constant for a definite host. According to the plots of $\ln[(I_0/I_T)-1]$ vs. $1/k_B T$ for LAO:0.001Mn⁴⁺,0.001Mg²⁺ phosphor, the value of ΔE can be roughly derived from the slope to be 0.68 eV.

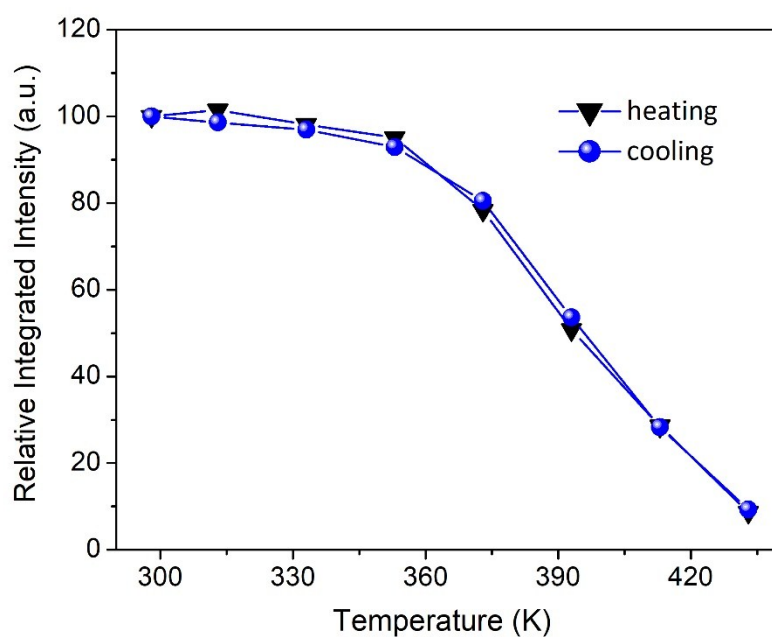


Fig. S3 The PL intensity of the LAO:Mn⁴⁺,Mg²⁺ phosphor in a heating-and-cooling cycle.

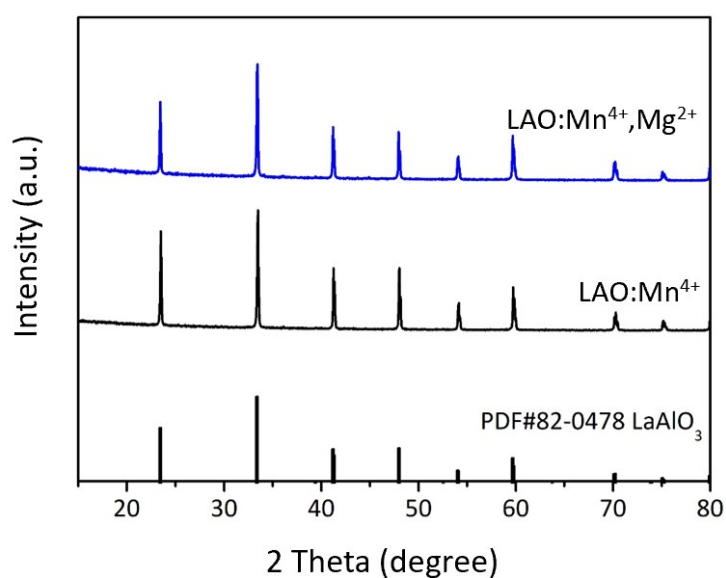


Fig. S4. XRD patterns of LAO:Mn⁴⁺,Mg²⁺ and LAO:Mn⁴⁺ phosphors.

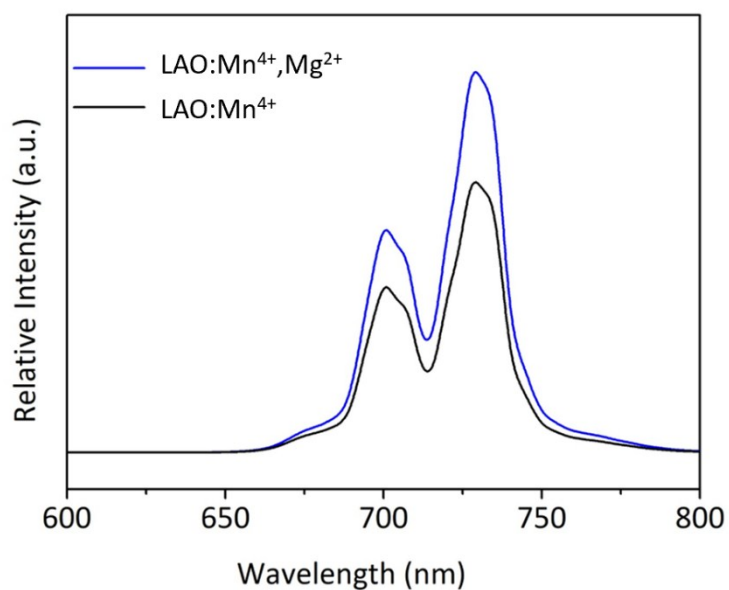


Fig. S5. PL spectra of LAO:Mn⁴⁺,Mg²⁺ and LAO:Mn⁴⁺ phosphors ($\lambda_{\text{ex}} = 340$ nm).

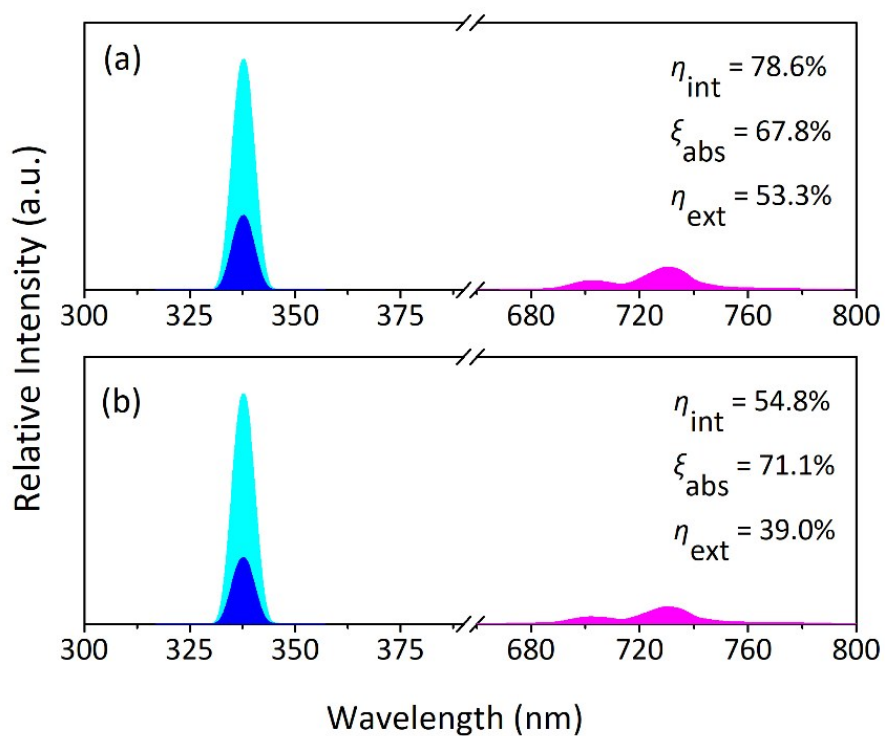


Fig. S6. Quantum yields of LAO:Mn⁴⁺,Mg²⁺ and LAO:Mn⁴⁺ phosphors ($\lambda_{\text{ex}} = 340$ nm).