

Supporting Information

Flexible and Conductive Metallic Paper-based Current Collector with Energy Storage

Capability in Supercapacitor Electrodes

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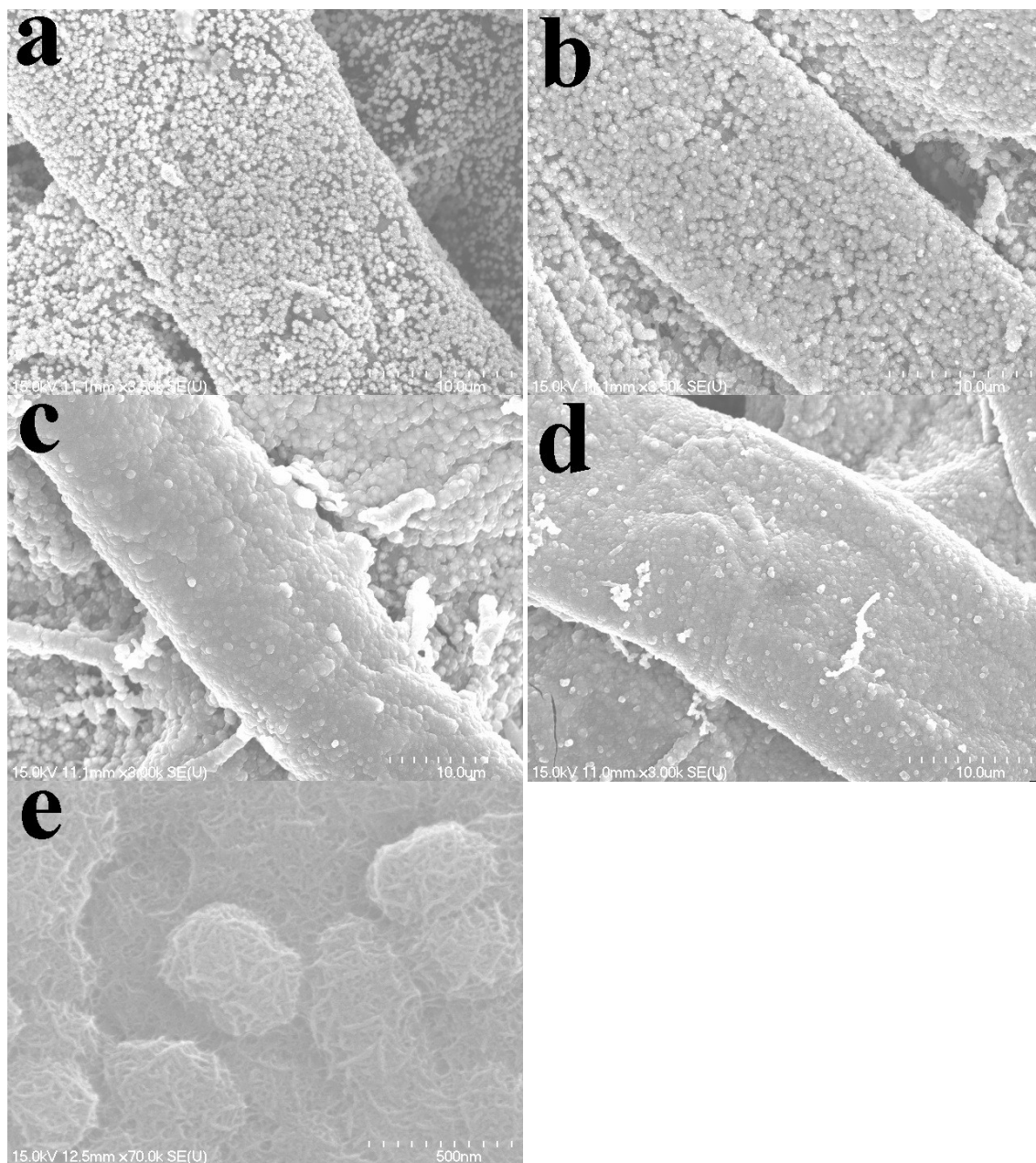


Figure S1. SEM images of Ni-paper-MnO₂ electrodes with different soaking time: (a) 5 min; (b) 10 min; (c) 20 min; (d) 30 min; (e) 30 min (high magnification).

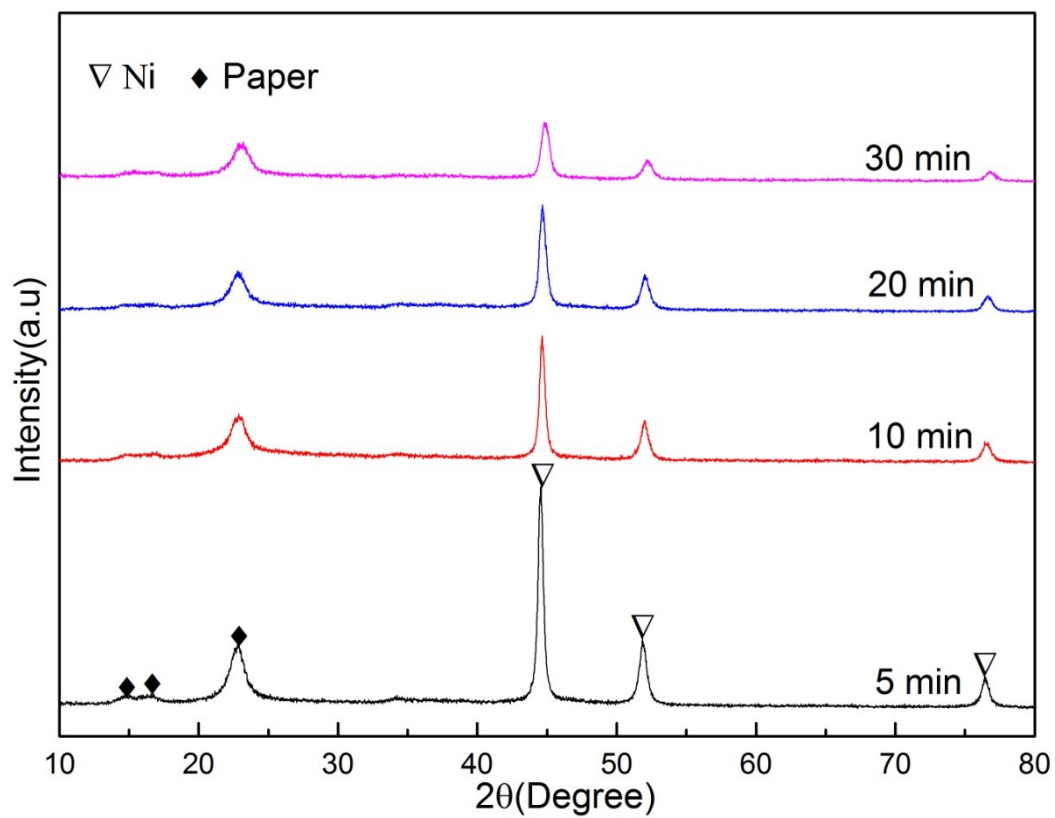


Figure S2. XRD patterns of Ni-paper-MnO₂ electrodes with different soaking time.

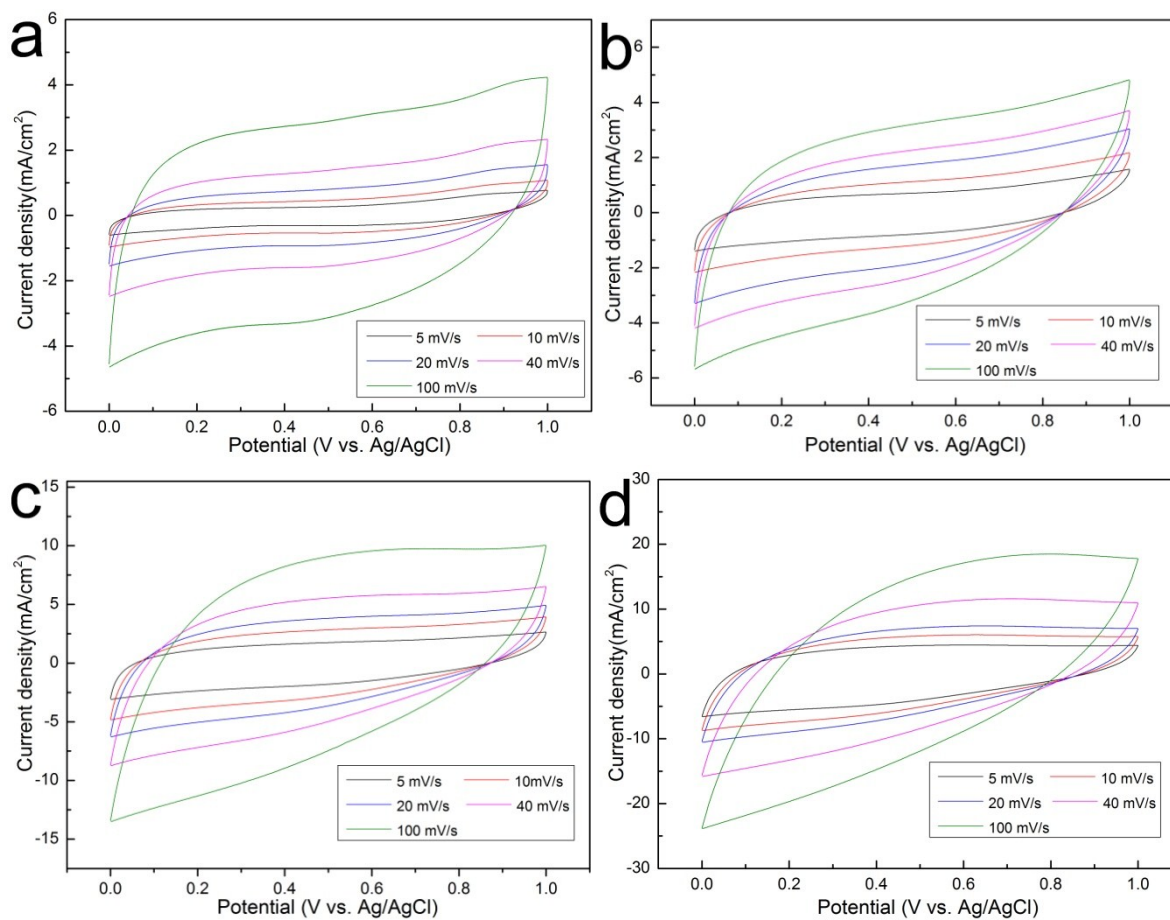


Figure S3. CV curves of Ni-paper-MnO₂ electrodes: (a) Ni-paper-5;(b) Ni-paper-10;(c) Ni-paper-20;(d) Ni-paper-30.

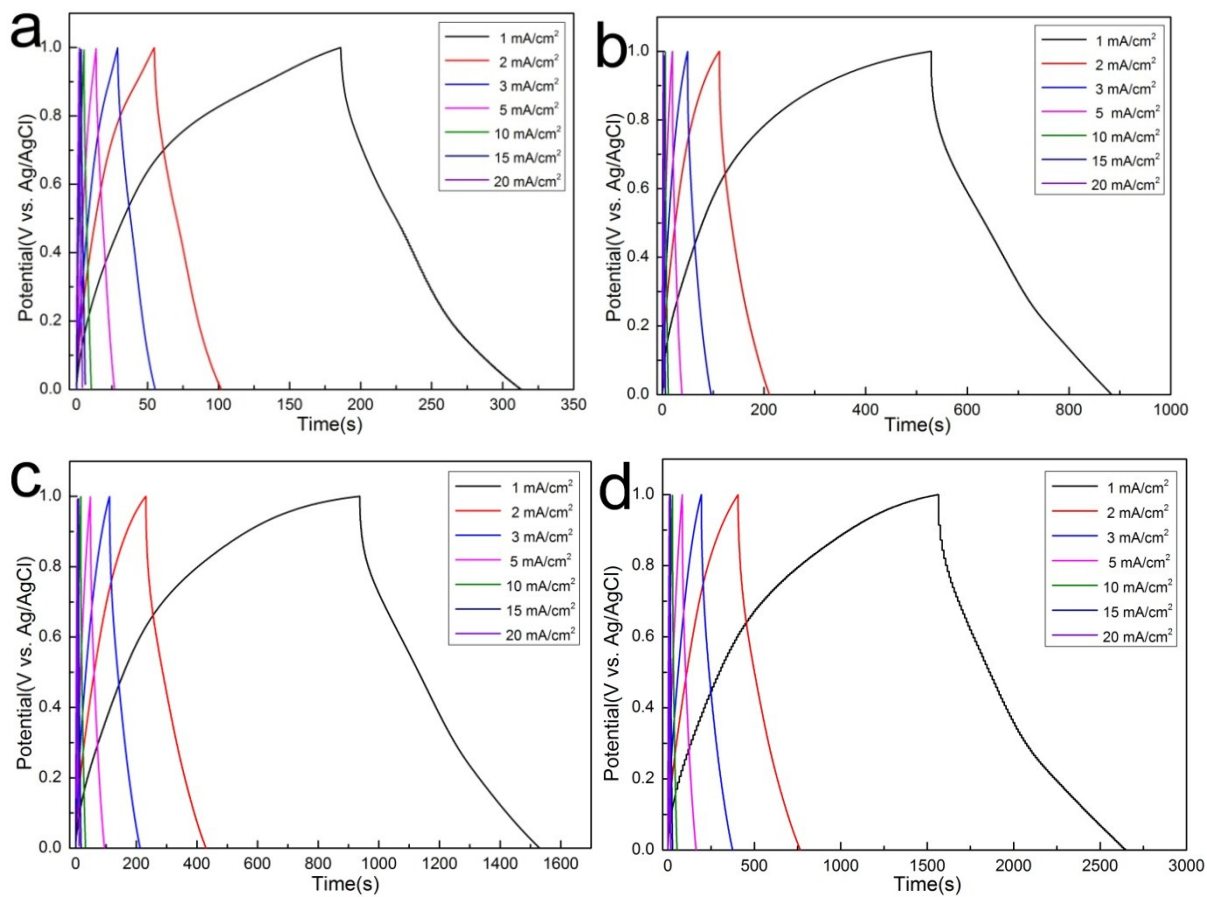


Figure S4. GCD curves of Ni-paper-MnO₂ electrodes: (a) Ni-paper-5;(b) Ni-paper-10;(c) Ni-paper-20;(d) Ni-paper-30.

Table.S1 Comparison of as prepared supercapacitor with some reported supercapacitors.

Materials	Current collector	Electrolyte	Capacitance	Stability (cycles)	Ref.
MnO ₂ nanostructure	Graphite foam	1M Na ₂ SO ₄	201 F/g (1A/g)	96.7% (1000)	S1
MnO ₂	Graphite fiber	1M Na ₂ SO ₄	245 F/g (1A/g)	80% (1000)	S2
MnO ₂ nanowires	PVDF membrane	0.5M Na ₂ SO ₄	118 F/g (200 mV/s)	95.3% (1000)	S3
MnO ₂ nanorods	Carbon nanofibers	1M Na ₂ SO ₄	291 F/g (1A/g)	90.9% (5000)	S4
MnO ₂ nanoflowers	Graphite paper	6%NH ₄ HC O ₃	368.3 F/g (0.2A/g)	98.4% (3000)	S5
MnO ₂ nanosheets	Carbon sphere	1M Na ₂ SO ₄	231 F/g (0.5A/g)	96% (1000)	S6
MnO ₂ nanoparticle	3D graphene	1M Na ₂ SO ₄	324 F/g (0.4A/g)	91.1% (5000)	S7
MnO ₂ /reduced graphene oxide	Ni fibers	1M Na ₂ SO ₄	119.4 mF/cm ² (0.5mA/cm ²)		S8
MnO ₂ nanorods	Porous carbon	6M KOH	196.2 F/g (1A/g)	78.5% (5000)	S9
MnO ₂ nanosheets	Carbon fibers	1M Na ₂ SO ₄	115.3 F/g (0.5A/g)	85.6% (2000)	S10
MnO ₂	Ni Paper	1M Na ₂ SO ₄	1095 mF/cm ² (1 mA/cm ²) 352 F/g (0.33 F/g)	73% (2000)	This work

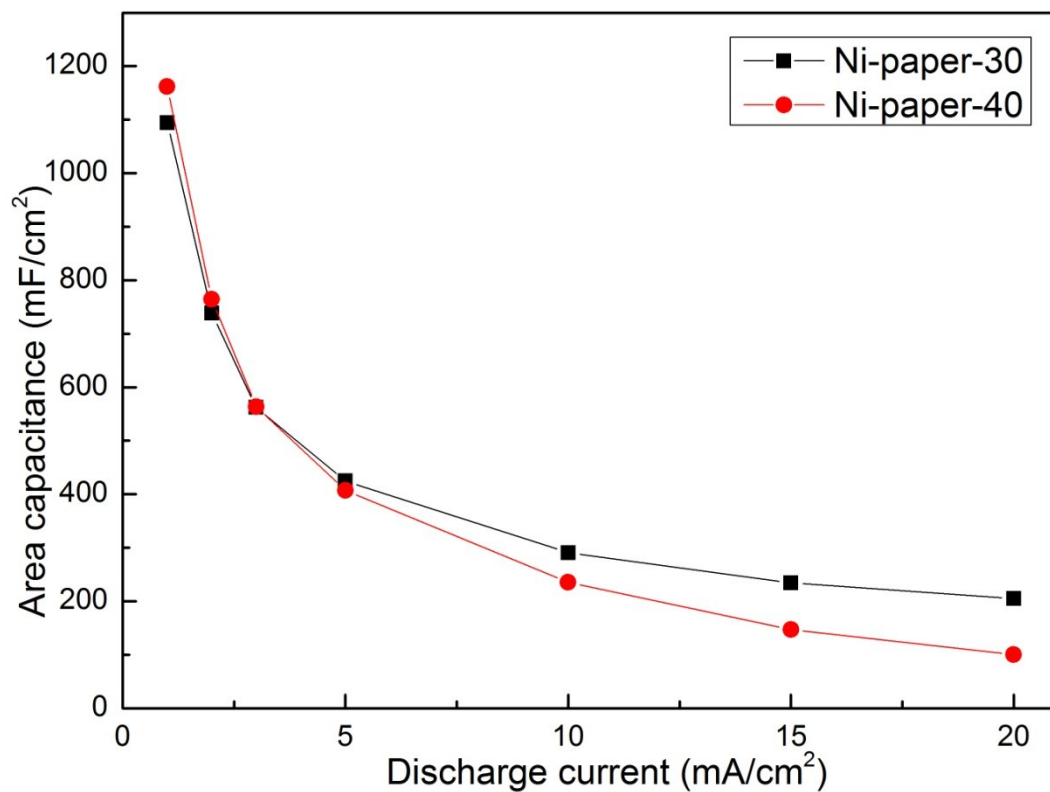


Figure S5. Areal specific capacitance measured at different current densities in the potential range of 0–1.0 V for Ni-paper-30 and Ni-paper-40.

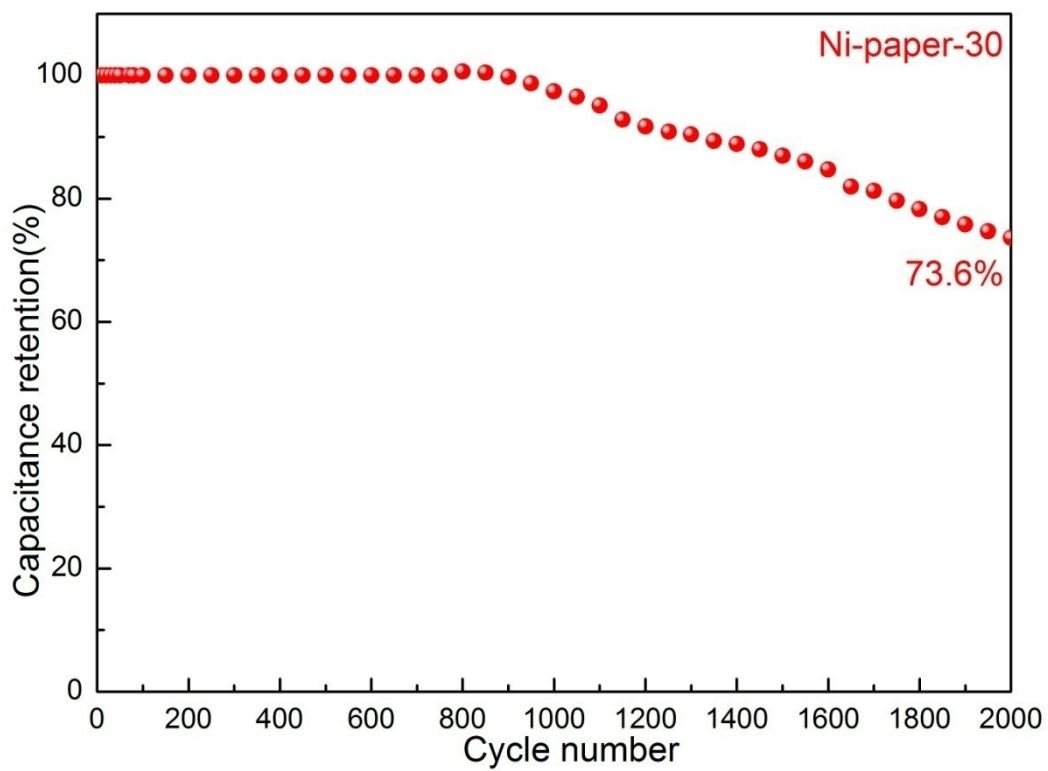


Figure S6. Cycling stability of the Ni-paper-30 at a current density of 10 mA/cm².

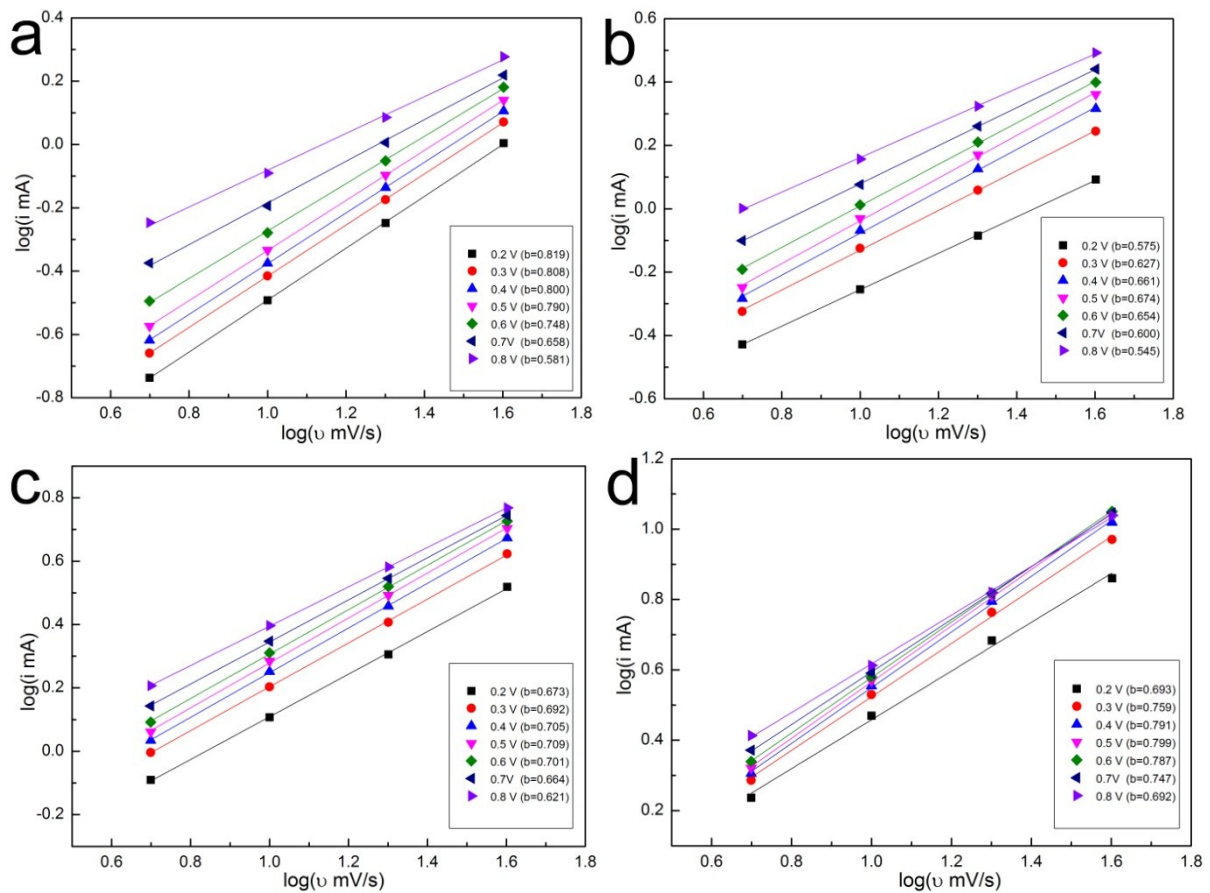


Figure S7. $\log(i)$ vs $\log(v)$ plot of the sample:(a) Ni-paper-5;(b) Ni-paper-10;(c) Ni-paper-20;(d) Ni-paper-30.

The contribution of the capacitive charge storage and the diffusion-controlled insertion processes could be quantitatively separated by the following formula:^{S11,S12}

$$i(V) = k_1 v + k_2 v^{1/2} \quad (1)$$

$$i(V)/v^{1/2} = k_1 v^{1/2} + k_2 \quad (2)$$

where $i(V)$ is the current at a given voltage, k_1 and k_2 are constants, and v is the scan rate. $k_1 v$ and $k_2 v^{1/2}$ represents the capacitive behaviour and diffusion-controlled insertion processes, respectively. k_1 and k_2 could be obtained by formula (2).

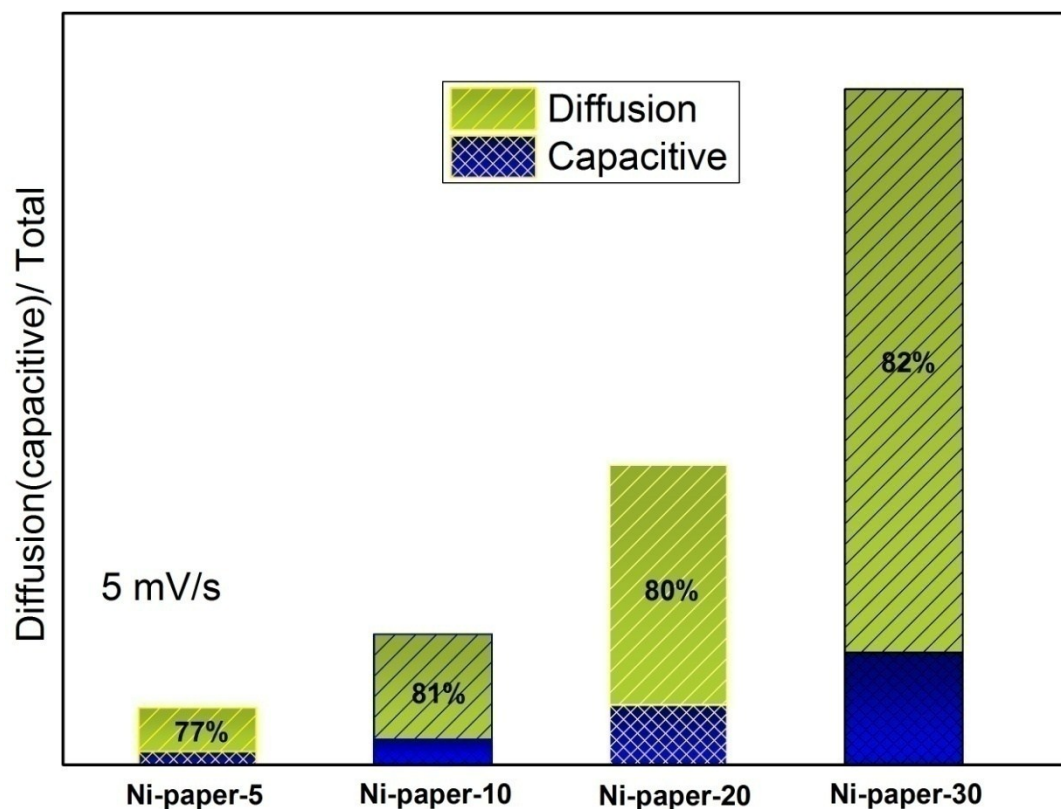


Figure S8. Diffusive and capacitive capacitance contribution of the samples at a scan rate 5 mV/s.

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