

Supplementary information

Tuning the supramolecular isomerism of MOF-74 by controlling the synthesis conditions

Andreea Gheorghe^a, Inhar Imaz^b, Jarl Ivar van der Vlugt^c, Daniel Maspoch^{b,d}, and Stefania Tanase*^a

- ^a Heterogeneous Catalysis and Sustainable Chemistry, van 't Hoff Institute for Molecular Sciences, University of Amsterdam, Science Park 904, 1098 XH, Amsterdam, The Netherlands.
- ^b Catalan Institute of Nanoscience and Nanotechnology (ICN2), CSIC and Barcelona Institute of Science and Technology, Campus UAB, Bellaterra 08193, Barcelona, Spain.
- ^c Bioinspired, Homogeneous & Supramolecular Catalysis, van 't Hoff Institute for Molecular Sciences, University of Amsterdam, Science Park 904, 1098 XH Amsterdam, The Netherlands.
- ^d ICREA, Pg. Lluís Companys 23, 08010 Barcelona, Spain.

*To whom correspondence should be addressed.

E-mail: s.grecea@uva.nl

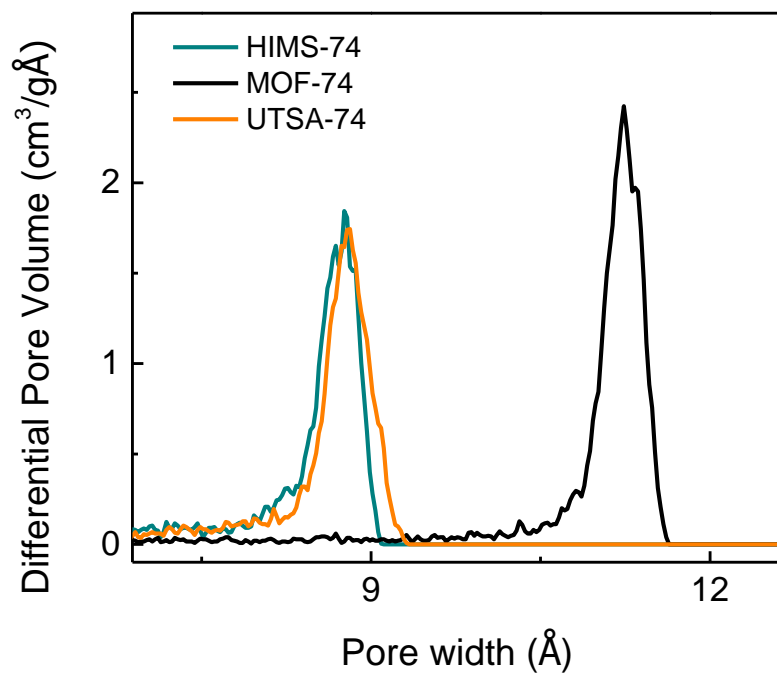


Fig. S1 Pore size distributions (PSDs) calculated from the single-crystal crystallographic data using iRASP¹ for UTSA-74, HIMS-74 and MOF-74².

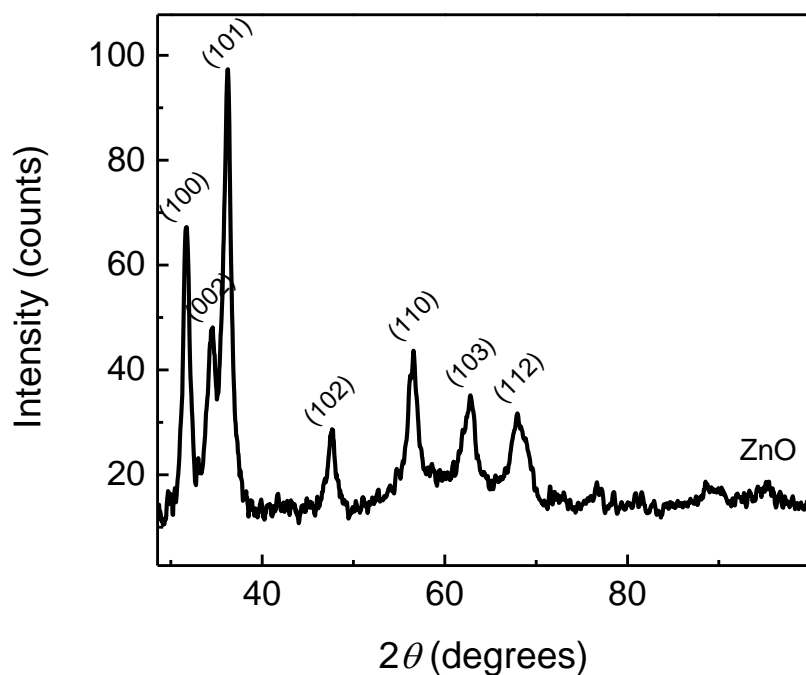


Fig. S2 PXRD pattern of the material obtained by reacting $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$, H_4dobdc , molar ratio Zn:linker 3:1, in NMP/EtOH/ H_2O (v/v/v 20/1/1), under solvothermal conditions, in the absence of *cinchona* alkaloid.

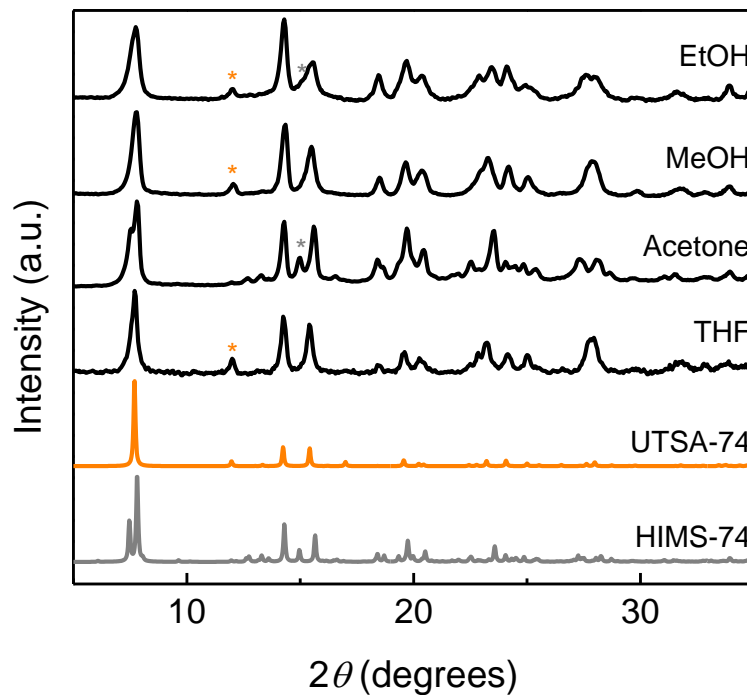


Fig. S3 PXRD pattern of the simulated UTSA-74 and HIMS-74 and HIMS-74 samples recovered after immersing in different polar solvents.

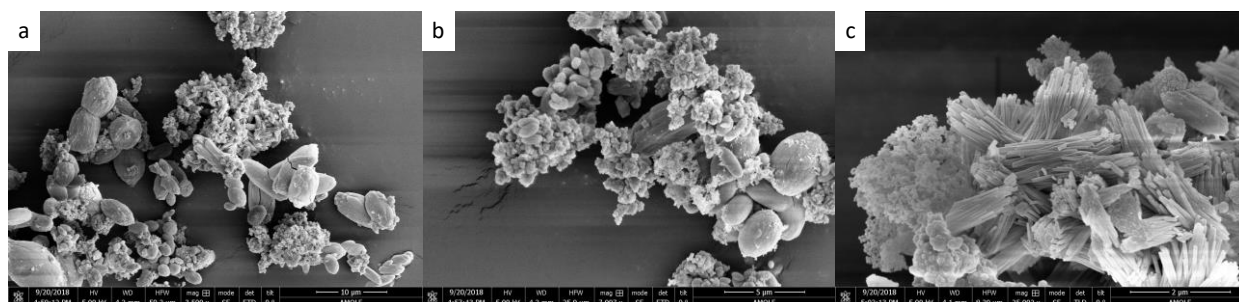


Fig. S4 SEM images of HIMS-74 particles immersed in methanol for 6 days with 3 times replenish of solvent.

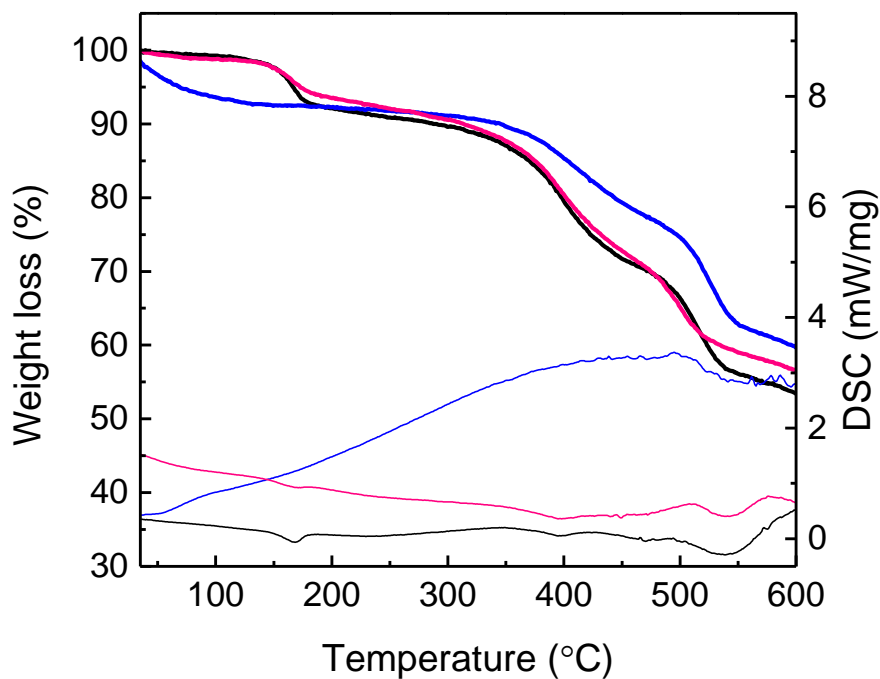


Fig. S5 TGA-DSC curves of as synthesized HIMS-74 (black) and the HIMS-74 samples recovered after exposure to methanol (blue) or acetone (pink).

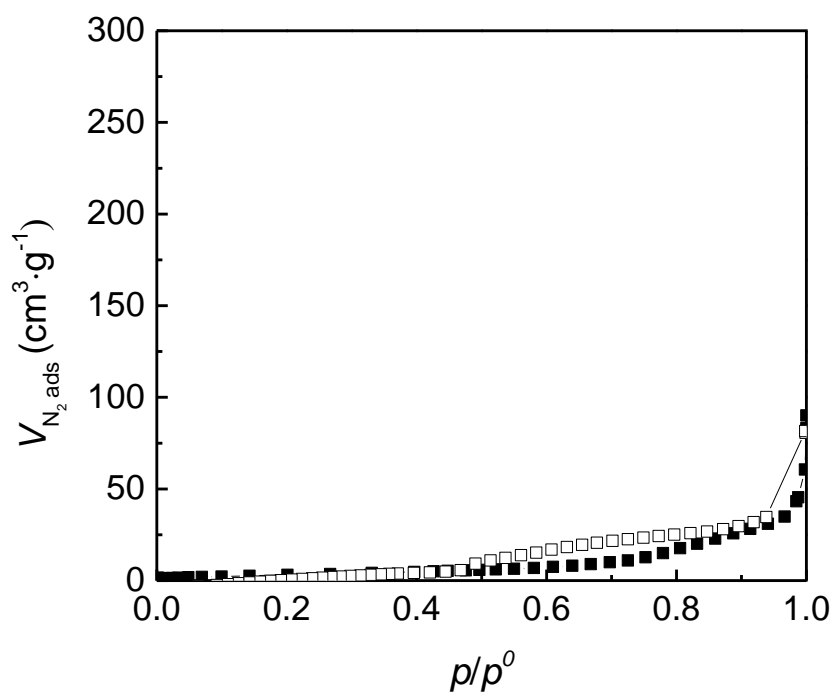


Fig. S6 N₂ adsorption isotherm for pretreated HIMS-74 recovered after immersion in MeOH at 77K.

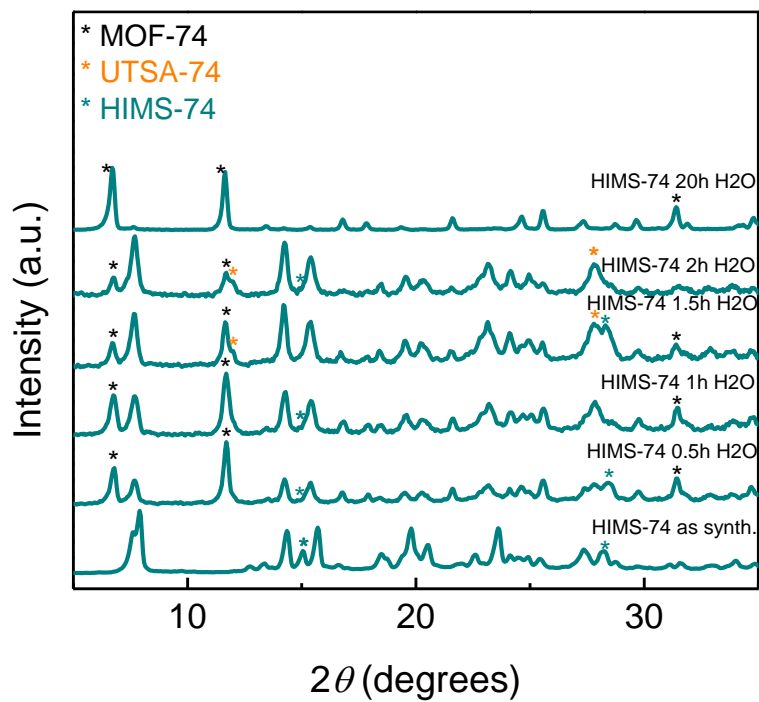


Fig. S7 PXRD patterns of the as synthesized HIMS-74 before and after immersion in H₂O at different stages showing its conversion to MOF-74. PXRD also indicates the presence of UTSA-74 phase.

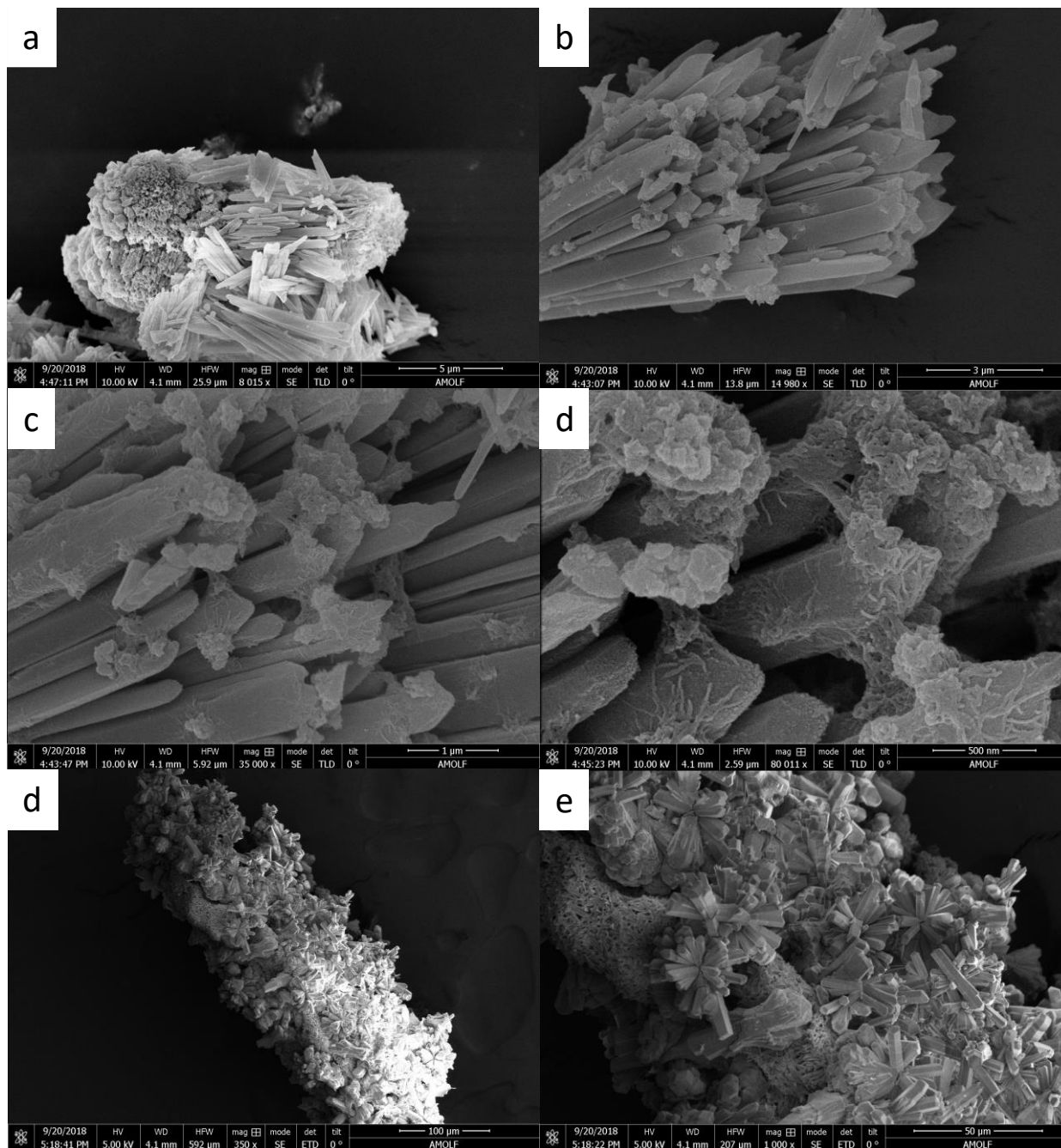


Fig. S8 Solvent-mediated isomerization of HIMS-74 to MOF-74 by immersion in H_2O and imaging of the crystals after 1.5h (a-d) and 20h (f-h) exposure time.

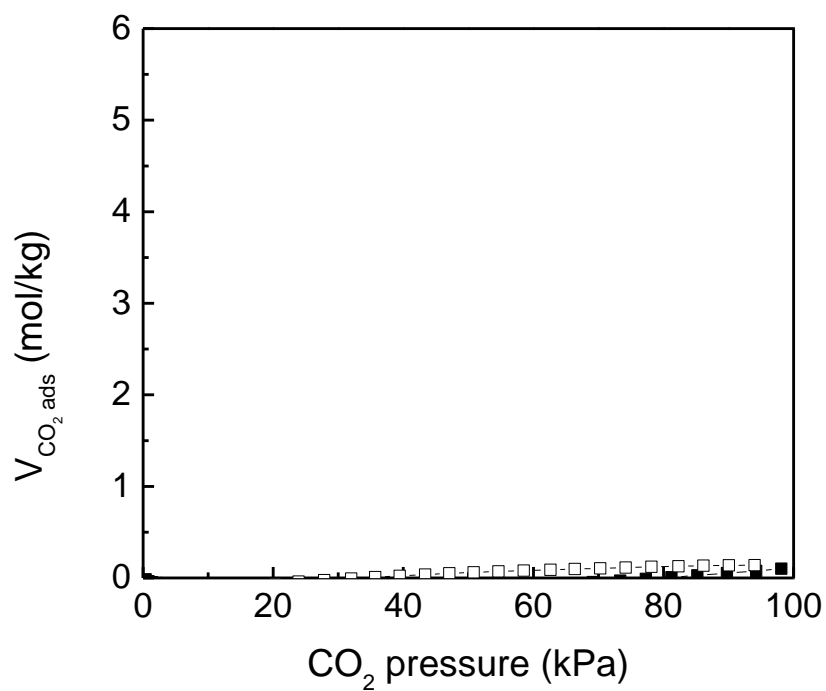


Fig. S9 CO₂ adsorption isotherm for activated HIMS-74 at 273K.

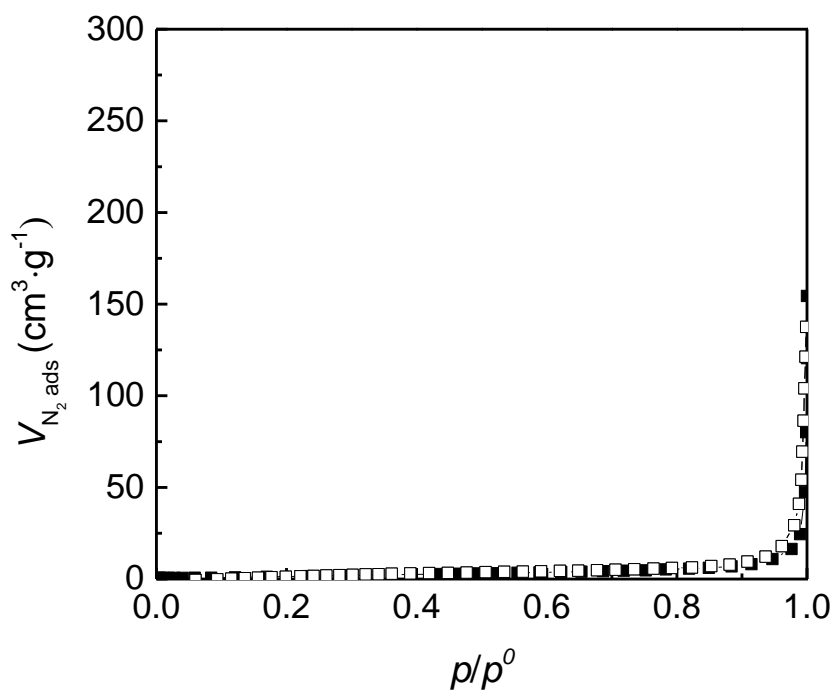


Fig. S10 N₂ adsorption isotherm for activated HIMS-74 as-synthesized at 77K.

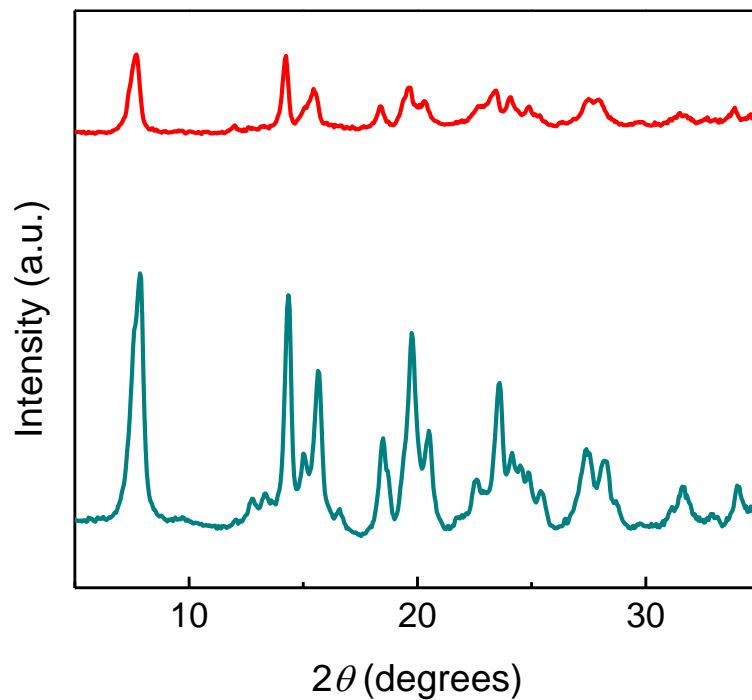


Fig. S11 PXRD patterns of the as synthesized HIMS-74 (cyan) and activated by vacuum heating with 1 °/min from room temperature to 150 °C and a hold of 2h.

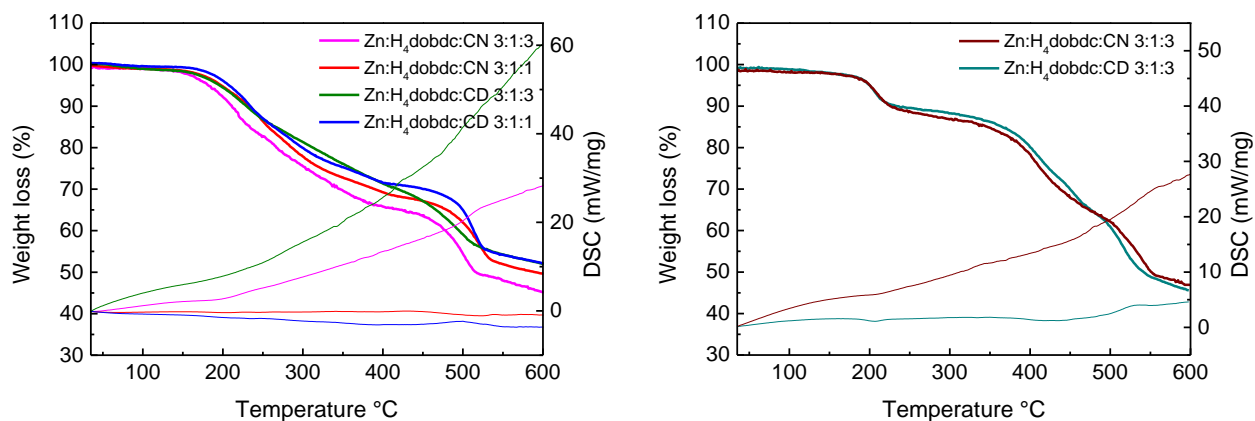


Fig. S12 TGA patterns of materials obtained in DMF (left) and materials obtained in NMP (right).

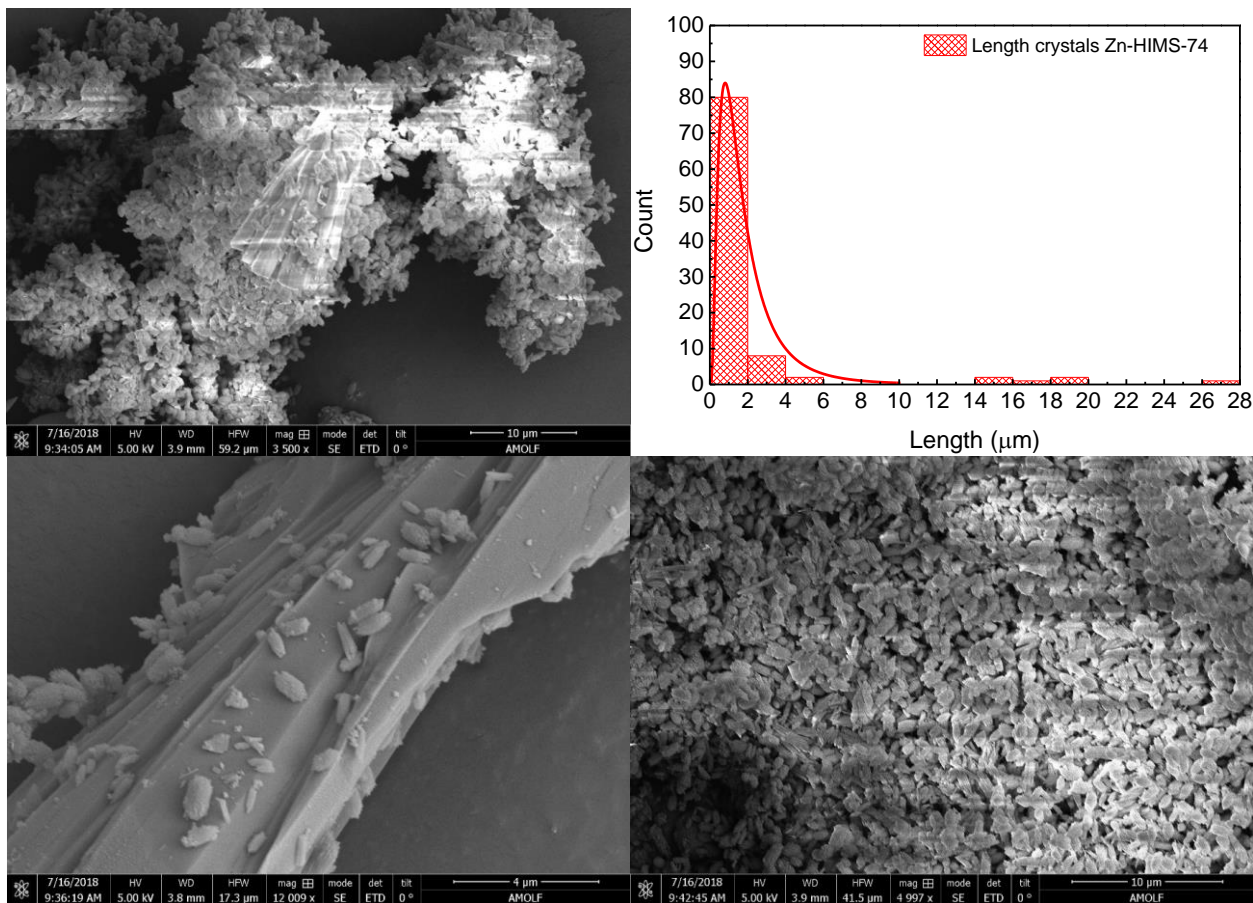


Fig. S13 SEM images of as synthesized HIMS-74 crystals and the corresponding histogram.



Fig. S14 SEM images of HIMS-74 particles synthesized at small scale (a) and large scale (b & c).

Table S1. Crystal Data and refinement information for complexes.

Complexes	HIMS-74	UTSA-74*
Formula	C ₄₄ H ₄₂ N ₄ O ₂₂ Zn ₆	C ₈ H ₂ O ₇ Zn ₂
Formula weight	1371.16	340.88
Crystal system	Monoclinic	Trigonal
Space group	<i>P</i> 2 ₁ / <i>a</i>	R-3c
a (Å)	14.030 (4)	22.970 (5)
b (Å)	23.660 (4)	22.970 (5)
c (Å)	15.820 (4)	15.910 (5)
α (°)	90	90
β (°)	113.39 (5)	90
γ (°)	90	120
Volume (Å ³)	4820(3)	7270(4)
Z	4	18
D _x (g/cm ³)	1.890	1.401
μ (mm ⁻¹)	3.212	3.164
Reflections Collected	7914	1943
Reflections Unique	7018	1817
R ₁ ^a [<i>I</i> > 2σ(<i>I</i>)]	0.0362	0.0560
wR ₂ ^b	0.0982	0.1746

$$^a R_1 = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|}$$

$$^b wR_2 = \left\{ \frac{\sum [w(|F_o|^2 - |F_c|^2)^2]}{\sum [w(|F_o|^4)]} \right\}^{1/2}$$

*The formula and derived parameters are calculated without solvent.

References

- 1 D. Dubbeldam, S. Calero and T. J. H. Vlugt, *Mol. Simul.*, 2018, **44**, 653–676.
- 2 N. L. Rosi, J. Kim, M. Eddaoudi, B. Chen, M. O’Keeffe and O. M. Yaghi, *J. Am. Chem. Soc.*, 2005, **127**, 1504–1518.