Electronic Supplementary Information

Thorium(IV) trialkyl complexes of non-carbocyclic ligands as highly active isoprene polymerisation catalysts

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Figure S1. ¹H NMR spectrum (400 MHz) of complex **1** in C₆D₆ at 25 °C.



Figure S2. ¹³C{¹H} NMR spectrum (101 MHz) of complex 1 in C_6D_6 at 25 °C.



Figure S3. ¹H NMR spectrum (400 MHz) of complex 2 in C_6D_6 at 25 °C.



Figure S4. ¹³C{¹H} NMR spectrum (101 MHz) of complex 2 in C_6D_6 at 25 °C.



-25.33

Figure S5. ³¹P{¹H} NMR spectrum (162 MHz) of complex 2 in C_6D_6 at 25 °C.



Figure S6. ¹H NMR spectrum (400 MHz) of complex 3 in C_6D_6 at 25 °C.



Figure S7. ¹³C{¹H} NMR spectrum (101 MHz) of complex 3 in C_6D_6 at 25 °C.



Figure S8. ${}^{31}P{}^{1}H$ NMR spectrum (162 MHz) of complex 3 in C₆D₆ at 25 °C.



Figure S9. ¹H NMR spectrum (400 MHz) of complex 4 in C_6D_6 at 25 °C.



Figure S10. $^{13}C\{^{1}H\}$ NMR spectrum (101 MHz) of complex 4 in $C_{6}D_{6}$ at 25 °C.



Figure S11. The steric map of complex 1



Figure S12. The steric map of complex 2

Experimental details for the polymerization kinetics studies

The catalyst system of complex $1/[Ph_3C][B(C_6F_5)_4]/Al'Bu_3$ and complex $2/[Ph_3C][B(C_6F_5)_4]/Al'Bu_3$ were selected for investigation into the kinetics of the polymerization of isoprene.

Experimental details: Under a N₂ atmosphere at 25 °C, isoprene (0.34 g, 5 mmol) and a deuterated benzene solution (1.5 mL) of Al/Bu₃ (25 µmol) were added into a 5 mL flask containing a magnetic stir bar. A deuterated benzene solution (1 mL) of complex **1** (4.65 mg, 5 µmol) or complex **2** (5.12 mg, 5 µmol) and [Ph₃C][B(C₆F₅)₄] (9.2 mg, 10 µmol) was added quickly by an injector, the conversion of isoprene with time was monitored by ¹H NMR spectroscopy.