

Supporting Information

Synthesis and Reactivity of an Osmium(III) Aminoguanidine Complex

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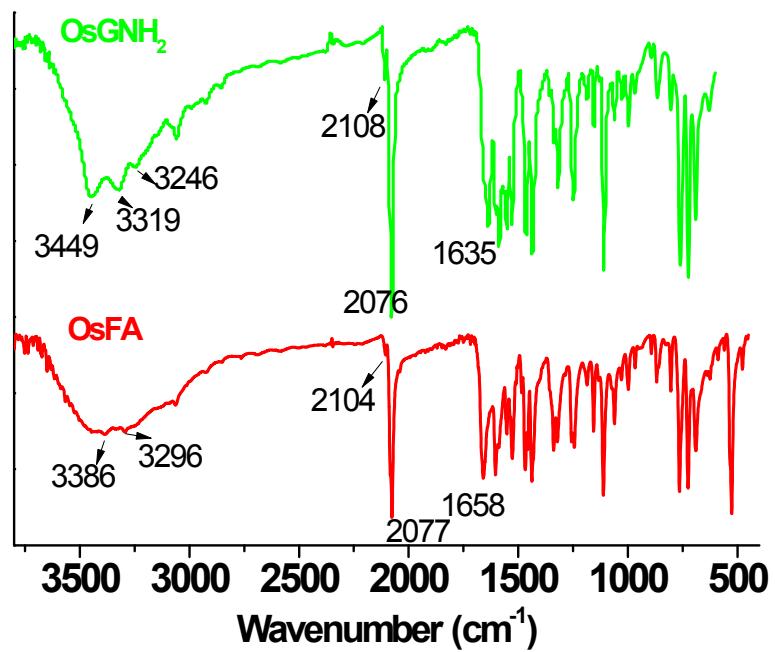


Figure S1. IR spectra of **OsGNH₂** and **OsFA**.

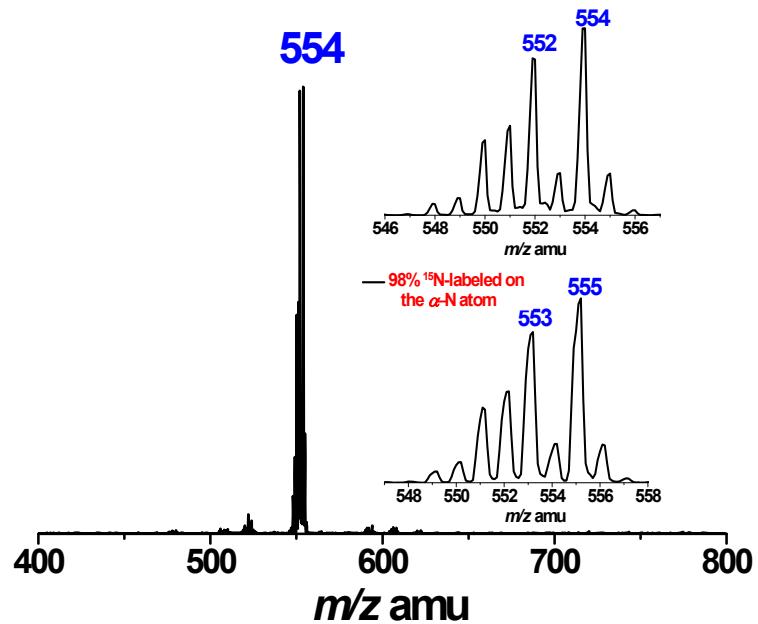


Figure S2. The ESI mass spectrum of $[\text{Os}\{\text{N}(\text{H})\text{C}(\text{NH}_2)\}(\text{NHNH}_2)(\text{L})(\text{CN})_3]^-$ (**OsGNH₂**) and experimental isotopic distribution patterns of the peaks at $m/z = 554$ and $m/z = 555$ (¹⁵N labeled on the α -N atom).

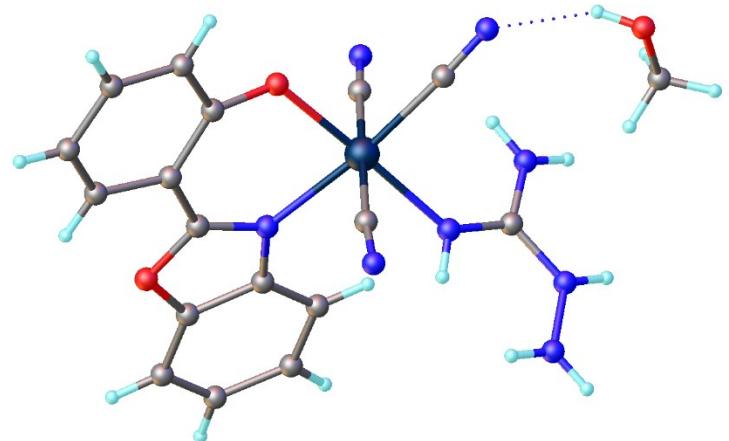


Figure S3. The H-bonding effects of a cyano group in **OsGNH₂** with MeOH.

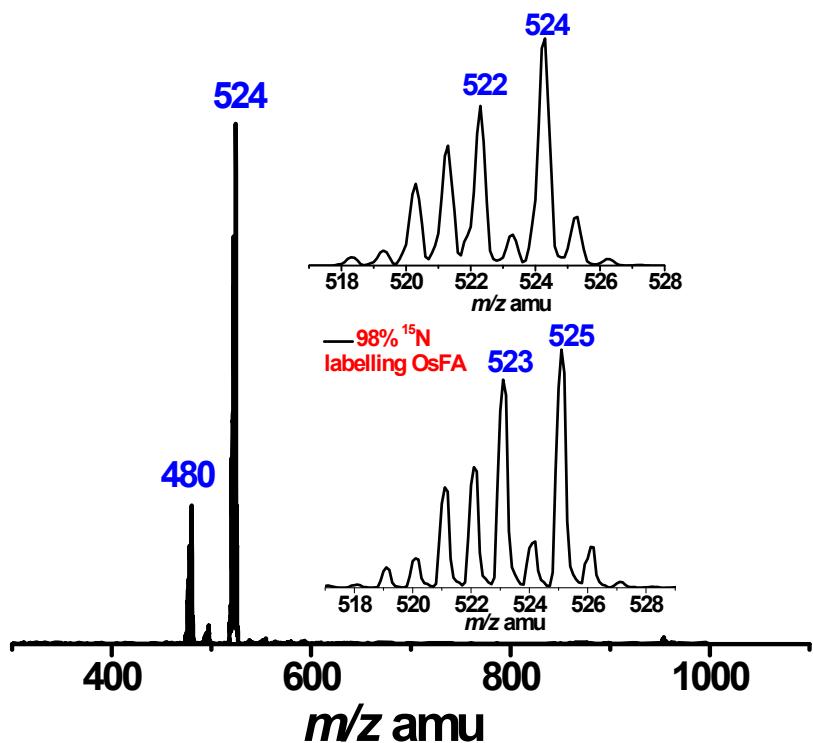


Figure S4. The ESI mass spectrum of $(\text{PPh}_4)[\text{Os}^{\text{III}}(\text{L})(\text{CN})_3(\text{NH}_2\text{-CH}=\text{NH})]$ (**OsFA**) and experimental isotopic distribution patterns of the peaks at $m/z = 524$ and $m/z = 525$ (^{15}N labelling on the $\alpha\text{-N}$ atom).

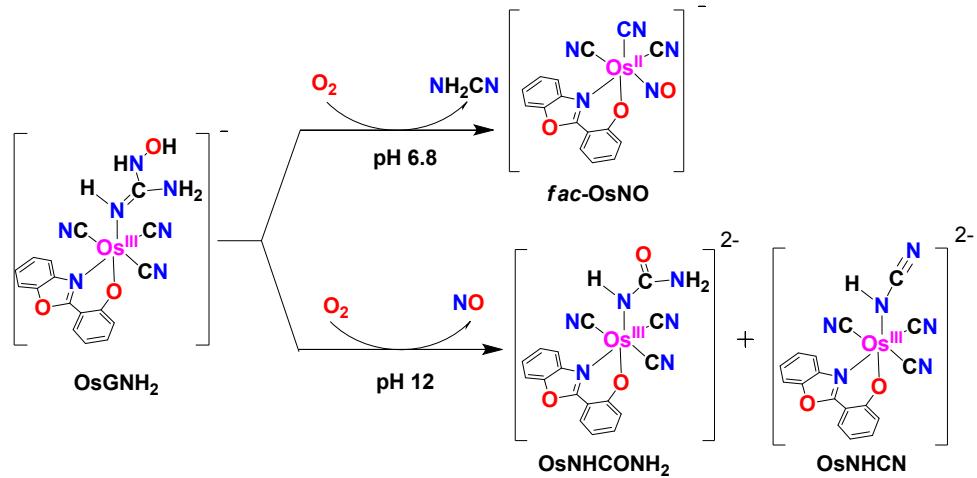


Figure S5. Aerobic oxidation of γ -OsGOH at different pH values.

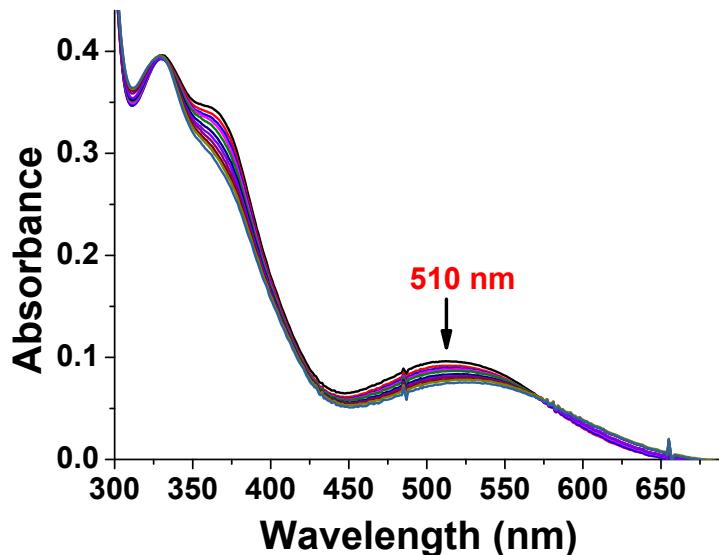


Figure S6. Spectral changes of the reaction of OsGNH_2 (5×10^{-5} M) and H_2O_2 (2.5×10^{-3} M) in CH_3OH at 25°C .

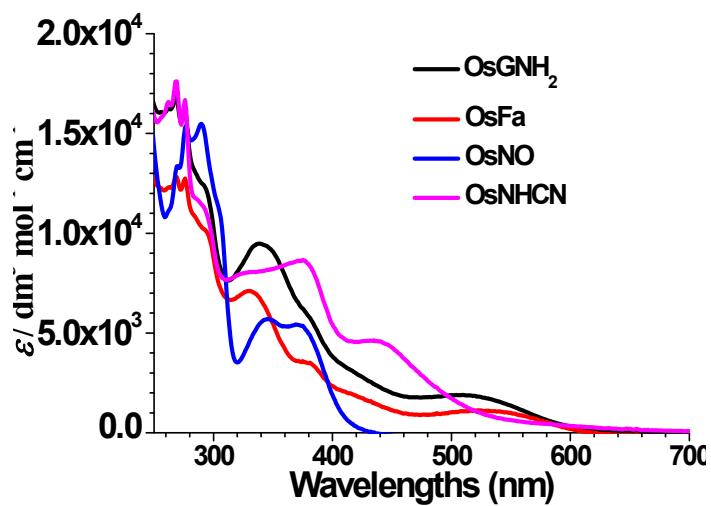


Figure S7. UV/vis absorption spectra of **OsGNH₂**, **OsFa**, *mer*-**OsNO** and **OsNHCN** in MeCN.

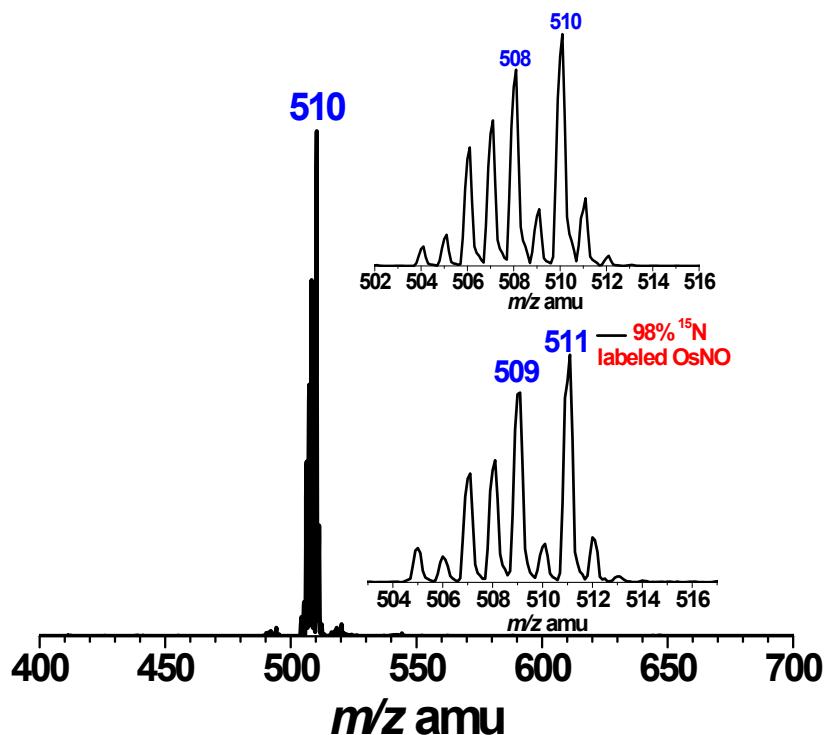


Figure S8. The ESI mass spectrum of $(\text{PPh}_4)[\text{Os}^{\text{II}}(\text{NO})(\text{L})(\text{CN})_3]^-$ (**mer-OsNO**) and experimental isotopic distribution patterns of the peaks at $m/z = 510$ and $m/z = 511$ (^{15}N labelling on the α -N atom).

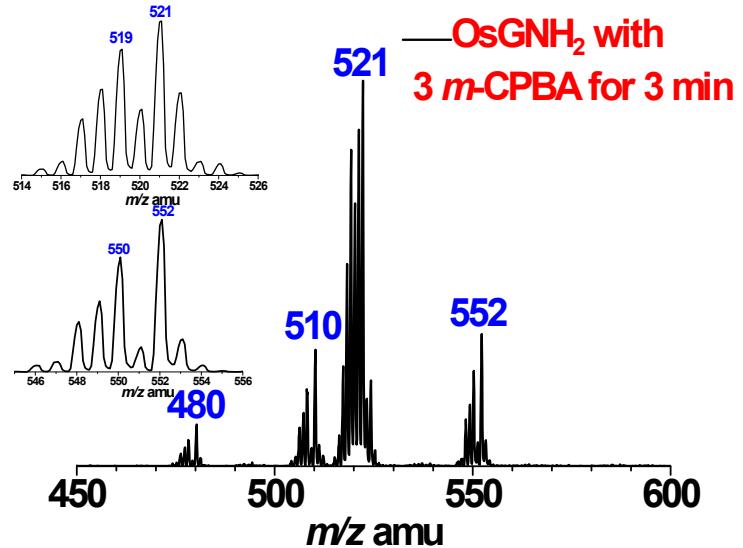


Figure S9. The ESI mass spectrum of *mer*-OsGNH₂ with 3 equiv. of *m*-CPBA for 3 min and insert shows experimental isotopic distribution patterns of *m/z* = 552 and *m/z* = 521.

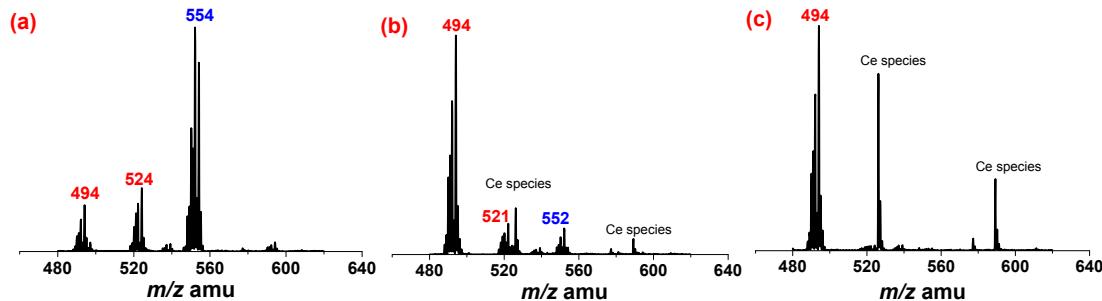


Figure S10. ESI/MS of OsGNH₂ (0.1 mM) with 10 equiv. of Ce(IV). (a) 1 min; (b) 3 min; (c) 10 min.

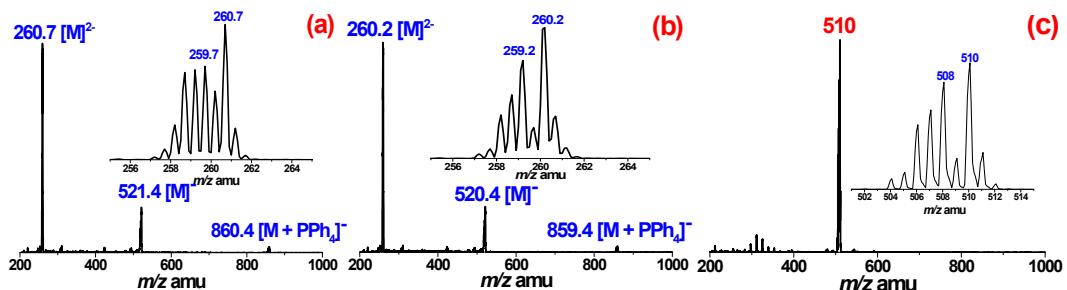


Figure S11. ESI/MS of OsNHCN (0.5 mM) with 10 equiv. of *m*-CPBA. (a) before mixing; (b) 5 min after mixing; (c) after 30 min.

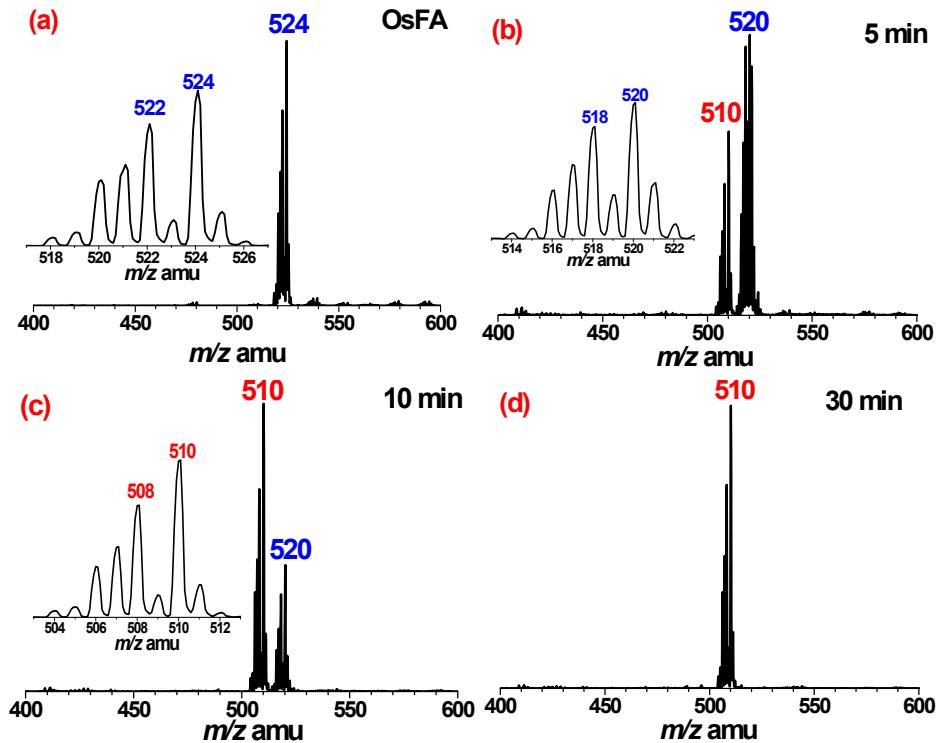


Figure S12. ESI/MS trace of **OsFA** (1.5 mM) with 10 equiv. of *m*-CPBA. (a) before mixing; (b) 5 min after mixing; (c) 10 min after mixing; (d) 30 min after mixing.

Table S1. Summary of crystal data, data collection and structure refinement for **OsGNH₂** and **OsFaA**.

Compound	OsGNH₂	OsFA
Formula	PC ₄₁ H ₃₄ N ₈ O ₂ OsP·CH ₃ OH	C ₄₁ H ₃₂ N ₆ O ₂ OsP
<i>M</i> _r	923.97	861.90
<i>T</i> / K	153	153
Crystal syst	orthorhombic	orthorhombic
Space group	P2 ₁ 2 ₁ 2 ₁	P2 ₁ 2 ₁ 2 ₁
<i>a</i> / Å	11.73820 (10)	11.76370 (10)
<i>b</i> / Å	16.3662 (2)	14.7379 (3)
<i>c</i> / Å	20.8351(2)	20.9329 (3)
α , (°)	/	/
β , (°)	/	/
γ , (°)	/	/
<i>V</i> / Å ³	4002.62 (7)	3629.18 (10)
<i>Z</i>	4	4
ρ_{calcd} , Mg m ⁻³	1.533	1.577
Unique refl.	7164	5736
Final R indices,	0.027	0.044
I > 2σ(I), wR ₂ (all data)	0.071	0.115
GOF	1.03	1.04
No. of par.	499	461