Supporting information for:

## Actinyl-cation interactions: Experimental and theoretical assessment of Np(VI) and U(VI) tetrachloro systems.

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## I. Crystal Structure Information

	LiNp	NaNp	KNp
Empirical formula	C <sub>16</sub> H <sub>32</sub> Cl <sub>4</sub> Li <sub>2</sub> NpO <sub>10</sub>	C <sub>24</sub> H <sub>52</sub> Cl <sub>4</sub> Na <sub>2</sub> NpO <sub>16</sub>	C <sub>24</sub> H <sub>49</sub> Cl <sub>4</sub> K <sub>2</sub> NpO <sub>14.50</sub>
Formula weight	777.00	1021.43	1026.63
Space group	Fmmm	$P2_1/n$	$P2_{1}/c$
a	13.115(8)	14.117(3)	10.0239(17)
b	13.371(7)	9.517(2)	9.9861(17)
С	14.838(10)	14.336(3)	19.349(3)
α	90	90	90
β	90	101.388(8)	95.187(7)
γ	90	90	90
V	2602(3)	1888.2(7)	1928.9(6)
Ζ	4	2	2
$\rho$ (g/cm <sup>3</sup> )	1.902	1.797	1.768
$\mu$ (mm <sup>-1</sup> )	4.447	3.122	3.244
F(000)	1372	1014	1016
$\theta$ range for data	2.253 to 26.442	2.260 to 28.346	2.938 to 28.351
collection (°)			
Limiting indices	-16< <i>h</i> <16	-18 <h<18< td=""><td>-13<h<13< td=""></h<13<></td></h<18<>	-13 <h<13< td=""></h<13<>
-	-16< <i>k</i> <16	-12< <i>k</i> <12	-13 <k<12< td=""></k<12<>
	-18< <i>l</i> <18	-19< <i>l</i> <19	-25< <i>l</i> <25
Reflections	11439/766	33982 / 4677	37361 / 4764
collected / unique			
R <sub>int</sub>	0.0494	0.0653	0.0545
Data / restraints /	766 / 0 / 78	4677 / 0 / 318	4764 / 3 / 311
parameters			
GOF on F <sup>2</sup>	1.07	1.10	0.99
Final R indices	$R_1 = 0.0419$	$R_1 = 0.0204$	$R_1 = 0.0175$
[I>2o(I)]	$wR_2 = 0.1088$	$wR_2 = 0.0439$	$wR_2 = 0.0392$
R indices (all data)	$R_1 = 0.0419$	$R_1 = 0.0284$	$R_1 = 0.0244$
	$wR_2 = 0.1088$	$wR_2 = 0.0468$	$wR_2 = 0.0409$
Largest diff. peak and hole	2.26 and -1.67	0.832 and -0.999	0.665 and -0.785

 Table S1.
 Select Crystallographic parameters for, LiNp, NaNp, and KNp.

Np(1)-O(1)	1.776 (12)	Li(1)-O(1)	1.92(3)	
Np(1)-O(1)	1.776 (12)	Li(1)-O(2A)	2.03(2)	
Np(1)-Cl(1)	2.626(3)	Li(1)- O(2A)	2.03(2)	
Np(1)-Cl(1)	2.626(3)	Li(1)- O(2B)	2.08(1)	
Np(1)-Cl(1)	2.626(3)	Li(1)- O(2B)	2.08(1)	
Np(1)-Cl(1)	2.626(3)			
O(1)-Np(1)-O(1)	180.0			
O(1)-Np(1)-C1(1)	91.3(6)			
O(1)-Np(1)-C1(1)	88.7(6)			

Table S2. Select bond lengths (Å) and angles (°) for LiNp.

 Table S3. Select bond lengths (Å) and angles (°) for NaNp.

Np(1)-O(1)	1.757(2)	C(1)-O(7)	1.417(3)	
$Np(1)-O(1)^{a}$	1.757(2)	C(1)-C(2)	1.500(3)	
Np(1)-Cl(1)	2.6637(7)	O(2)-C(2)	1.423(3)	
$Np(1)-Cl(1)^a$	2.6637(7)	O(2)-C(3)	1.427(3)	
Np(1)-Cl(2)	2.6675(8)	O(3)-C(4)	1.431(3)	
$Np(1)-Cl(2)^{a}$	2.6676(8)	O(3)-C(5)	1.436(3)	
$O(1)^{a}-Np(1)-O(1)$	180.0	C(3)-C(4)	1.497(4)	
O(1)-Np(1)-Cl(1)	90.85(6)	O(4)-C(7)	1.423(3)	
O(1)-Np(1)-Cl(2)	89.58(6)	O(4)-C(6)	1.431(3)	
		O(5)-C(8)	1.420(3)	
$Na(1)-O(8)^{b}$	2.355(2)	O(5)-C(9)	1.429(3)	
Na(1)-O(3)	2.486(2)	C(5)-C(6)	1.499(3)	
$Na(1)-O(7)^{b}$	2.550(2)	O(6)-C(10)	1.428(3)	
$Na(1)-O(2)^{b}$	2.563(2)	O(6)-C(11)	1.431(3)	
$Na(1)-O(5)^{b}$	2.586(2)	C(7)-C(8)	1.503(4)	
$Na(1)-O(4)^{b}$	2.646(2)	O(7)-C(12)	1.426(3)	
$Na(1)-O(6)^{b}$	2.880(2)	C(9)-C(10)	1.498(4)	
$Na(1)-O(3)^{b}$	2.972(2)	C(11)-C(12)	1.507(4)	

a:-x+1,-y+1,-z; b:-x+1,-y+1,-z+1.

Np(1)-O(1)	1.7662(13)	O(2)-C(1)	1.424(2)	
$Np(1)-O(1)^{a}$	1.7662(13)	O(2)-C(12)	1.426(2)	
Np(1)-Cl(1)	2.6513(5)	O(3)-C(2)	1.424(2)	
$Np(1)-Cl(1)^a$	2.6513(5)	O(3)-C(3)	1.426(2)	
Np(1)-Cl(2)	2.6614(6)	O(4)-C(5)	1.423(2)	
$Np(1)-Cl(2)^{a}$	2.6614(6)	O(4)-C(4)	1.425(2)	
$O(1)^{a}-Np(1)-O(1)$	180.0	O(5)-C(7)	1.425(2)	
O(1)-Np(1)-Cl(1)	88.34(5)	O(5)-C(6)	1.426(2)	
O(1)-Np(1)-Cl(2)	89.52(5)	O(6)-C(8)	1.420(2)	
		O(6)-C(9)	1.423(2)	
K(1)-O(6)	2.8553(14)	O(7)-C(11)	1.423(2)	
K(1)-O(2)	2.8626(15)	O(7)-C(10)	1.425(2)	
K(1)-O(4)	2.8706(14)	C(1)-C(2)	1.504(3)	
K(1)-O(3)	2.8881(14)	C(3)-C(4)	1.504(3)	
K(1)-O(5)	2.9110(15)	C(5)-C(6)	1.500(3)	
K(1)-O(7)	3.0387(14)	C(7)-C(8)	1.499(3)	
K(1)-Cl(2)	3.2550(8)	C(9)-C(10)	1.499(3)	
K(1)-Cl(1)	3.3179(8)	C(11)-C(12)	1.493(3)	

 Table S4.
 Select bond lengths (Å) and angles (°) for KNp.

a: -x,-y+1,-z+1.



**Figure S1**. Crystalline packing for **LiNp**. The Np, Cl, Li, O, C atoms are depicted as maroon, blue, teal, red, and black spheres, respectively. H atoms have been removed from the image for clarity.



**Figure S2**. Crystalline packing for **NaNp1**. The Np, Cl, Na, O, C atoms are depicted as maroon, blue, yellow, red, and black ellipsoids, respectively. H atoms have been removed from the image for clarity.



**Figure S3.** Extended topology for **KNp1**. The Np, K, O, Cl, C atoms are depicted as maroon, purple, red, blue, and black ellipsoids, respectively. The H atoms have been removed for clarity.

## II. Raman Spectroscopy





Figure S4. Solid state spectra for the LiNp, NaNp, and KNp compounds.



Figure S5. Peak fitting parameters for the Np(VI) stock solution in 1 M HCl



**Figure S6.** Peak fitting parameters for the Np(VI) stock solution with  $Li^+$  concentration of 13.3 M.



**Figure S7.** DFT-optimized structures of **DFT-NaNp**, **DFT-LiNp**, **DFT-KNp** for the four structure types (I-IV), where Cl, O, Np, Na, Li, and K are represented by blue, red, maroon, yellow, turquoise, and purple, respectively.

Table S5. Selected average bond lengths from the optimized DFT-NaNp, DFT-LiNp, and DFT-KNp structures.

	Np=O (Å)	Np-Cl (Å)	
Li-I	1.756	2.696	
Li-II	1.756	2.700	
Li-III	1.763	2.686	
Li-IV	1.798	2.632	
Na-I	1.756	2.711	
Na-II	1.771	2.684	
Na-III	1.773	2.656	
Na-IV	1.754	2.695	
K-I	1.758	2.695	
K-II	1.762	2.684	
K-III	1.768	2.671	
K-IV	1.767	2.672	



**Figure S8.** DFT-optimized structures of **DFT-NaU**, **DFT-LiU**, **DFT-KU** for the four structure types (I-IV), where Cl, O, U, Na, Li, and K are represented by blue, red, orange, yellow, turquoise, and purple, respectively.

Table S6. Selected average bond lengths from the optimized DFT-NaU, DFT-LiU, and DFT	ſ-
KU structures.	

	U=O (Å)	U-Cl (Å)
Li-I	1.776	2.724
Li-II	1.777	2.715
Li-III	1.785	2.702
Li-IV	1.819	2.646
Na-I	1.776	2.728
Na-II	1.779	2.706
Na-III	1.795	2.687
Na-IV	1.774	2.712
K-I	1.779	2.711
K-II	1.783	2.700
K-III	1.789	2.688
K-IV	1.787	2.688