

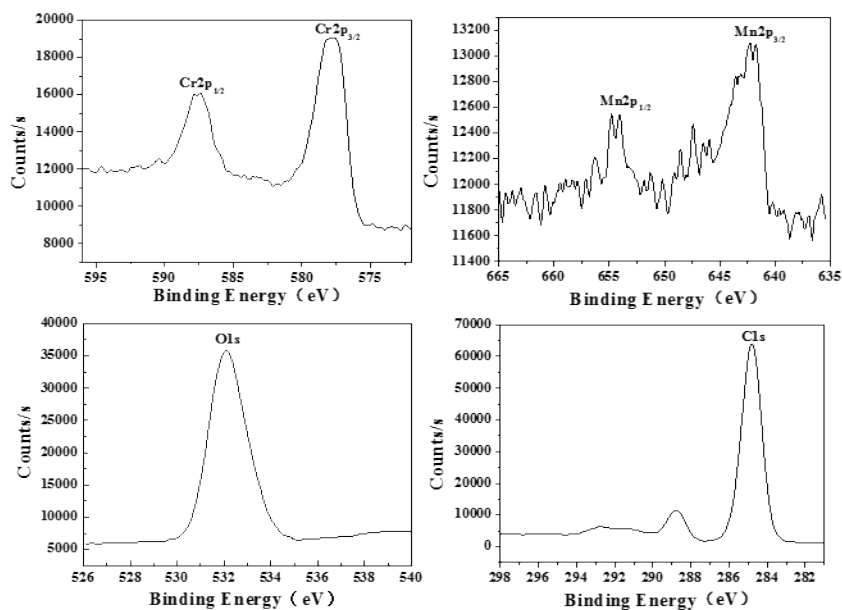


*Electronic Supplementary Information (ESI) for*

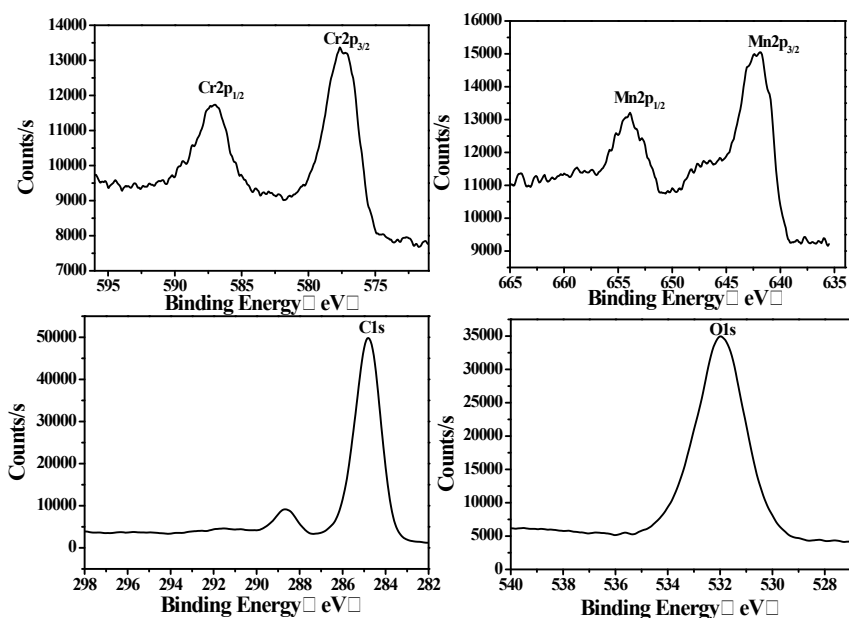
## Hyperstable Chromium(III)/Manganese(II) Bimetallic Wheel Cluster with Visible Photoactivity

**Table S1.** Crystal data and structure refinement summary for cluster of **1** and **2**.

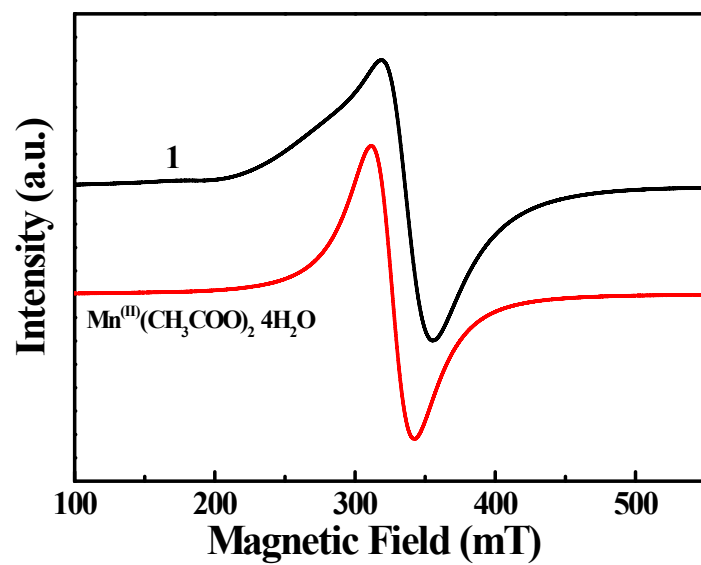
Complex Name	<b>1</b>	<b>2</b>
CCDC No.	1889241	1818720
Formulae	C <sub>112</sub> H <sub>88</sub> O <sub>40</sub> Cr <sub>7</sub> Mn	C <sub>92</sub> H <sub>72</sub> O <sub>28</sub> Cr <sub>4</sub> Mn <sub>4</sub>
Mol. wt.	2492.76	2164.05
Crystal system	Tetragonal	Triclinic
Space group	P 4/n n c	<i>P</i> -1
Temperature (K)	293	293
Wavelength (Å)	0.71073	0.71073
a, b, c /Å	a 16.8255(9) b 16.8255(9) c 23.4167(14)	a 14.682(2) b 15.378(3) c 24.143(4)
$\alpha$ , $\beta$ , $\gamma$ /°	90, 90, 90	74.940(5), 89.845(5), 65.801(5)
V/ Å <sup>3</sup>	6629.2(8)	4768.0 (14)
Z	2	2
Density / g.cm <sup>-3</sup>	1.249	1.507
Abs. Coeff. /mm <sup>-1</sup>	0.716	1.031
F(000)	2546	2203
Total no. of reflections	3365	21666
Reflections, $I > 2\sigma(I)$	1365	14925
Max. $2\theta$ /°	27.517	27.513
Ranges (h, k, l)	-21 ≤ h ≤ 21 -21 ≤ k ≤ 17 -30 ≤ l ≤ 30	-19 ≤ h ≤ 19 -19 ≤ k ≤ 19 -31 ≤ l ≤ 31
Complete to $2\theta$ (%)	0.876	98.9
Restraints	381	76
Parameters	189	1270
Goof (F <sup>2</sup> )	1.067	1.015
Rindices [ $I > 2\sigma(I)$ ]	0.0876	0.0451
Rindices (all data)	0.1996	0.0795
WR <sub>2</sub> [ $I > 2\sigma(I)$ ]	0.1867	0.1082
WR <sub>2</sub> (all data)	0.2496	0.1230



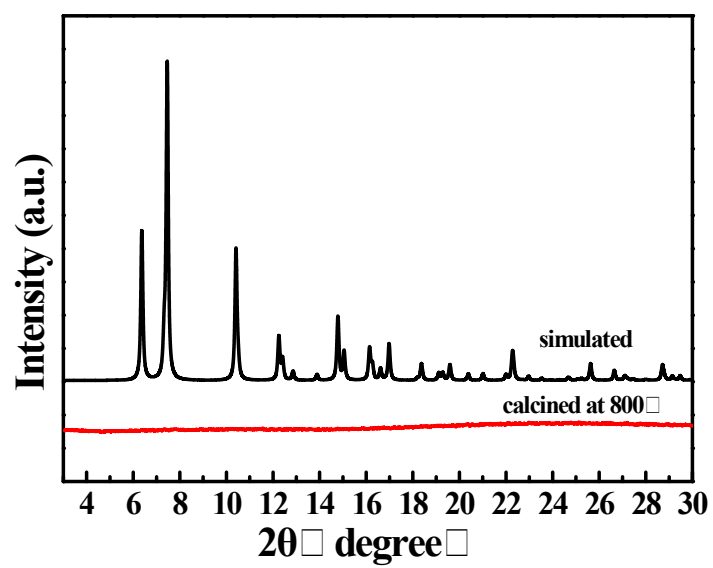
**Figure S1.** XPS high-resolution spectra of Cr 2p, Mn 2p, O 1s and C 1s core levels for cluster 1 indicates the existence oxidation state of (a)  $\text{Cr}^{3+}$  (577.8 eV for  $2p_{3/2}$  and 587.7 eV for  $2p_{1/2}$ ), (b)  $\text{Mn}^{2+}$  (642.3 eV for  $2p_{3/2}$  and 654.3 eV for  $2p_{1/2}$ ), (c) O 1s (532.1 eV), (d) C 1s (284.8 eV and 288.8 eV).



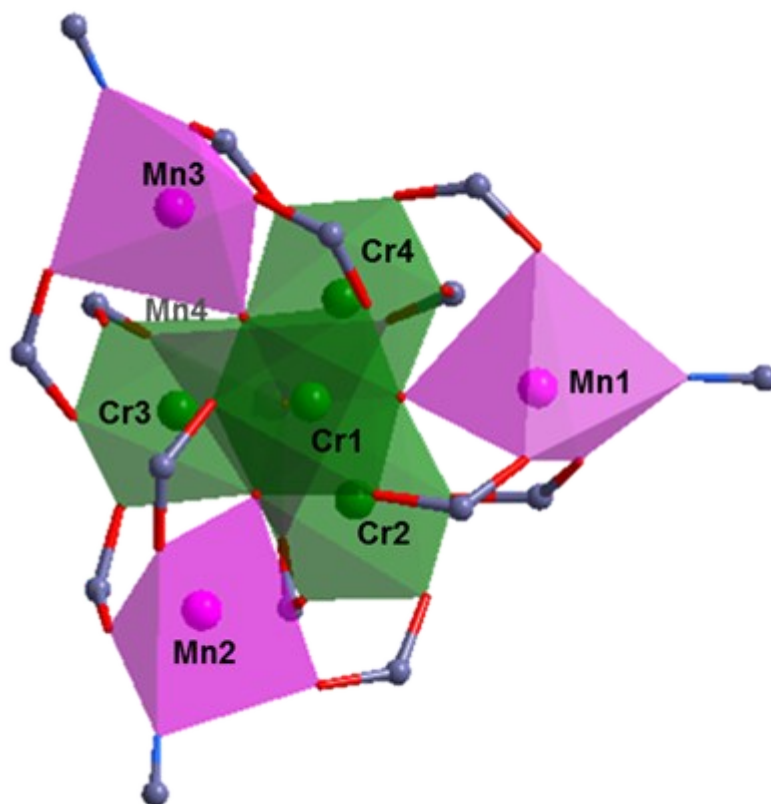
**Figure S2.** XPS high-resolution spectra of Cr 2p, Mn 2p, O 1s and C 1s core levels for cluster 2 indicates the existence oxidation state of (a)  $\text{Cr}^{3+}$  (577.4 eV for  $2p_{3/2}$  and 587.2 eV for  $2p_{1/2}$ ), (b)  $\text{Mn}^{2+}$  (642.1 eV for  $2p_{3/2}$  and 653.9 eV for  $2p_{1/2}$ ), (c) C 1s (284.8 eV and 288.7 eV), (d) O 1s (531.9 eV).



**Figure S3.** The ESR spectrum of **1** (black line) and manganese acetate tetrahydrate (red line).



**Figure S4** Simulated XRD patterns of **1** from single-crystal X-ray diffraction data (black line) and PXRD pattern of **1** after calcined at 800 °C in Ar flow using tube furnace.



**Figure S5.** Polyhedral and ball-and-stick representations of **2**.

**Table S2.** Selected bond lengths (Å) and angles (°) for **2**.

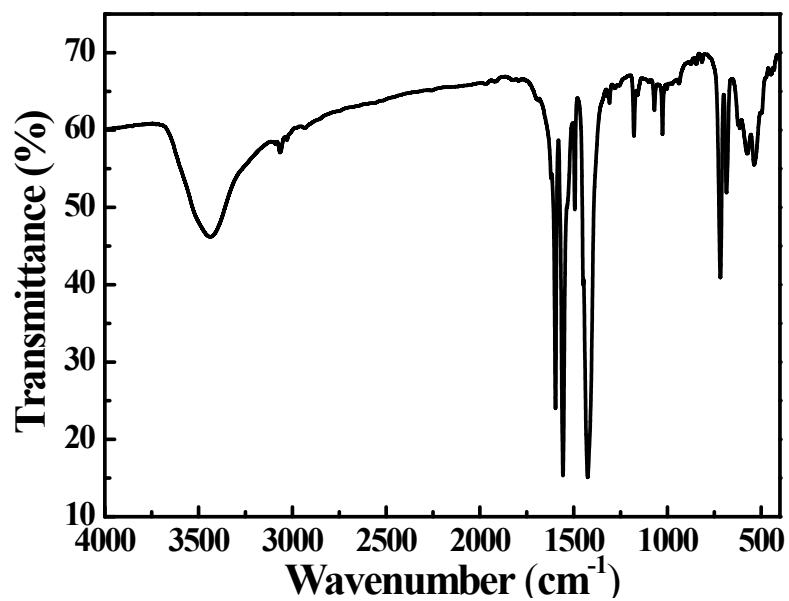
Lengths				
Cr1-O1	1.959	Mn4-O1	2.059	
Cr1-O2	1.967	Mn4-O2	2.135	
Cr1-O3	1.966	Mn4-O3	2.102	
Cr1-O4	1.988	Mn4-O4	2.100	
Cr1-O5	1.988	Mn4-O5	2.198	
Cr1-O6	2.004	Mn1-O1	2.064	
Cr2-O1	1.9568	Mn1-O2	2.073	
Cr2-O2	1.965	Mn1-O3	2.088	
Cr2-O3	1.963	Mn1-O4	2.070	
Cr2-O4	1.999	Mn1-N	2.275	
Cr2-O5	1.985	Mn2-O1	2.059	
Cr2-O6	1.993	Mn2-O2	2.080	
Cr3-O1	1.958	Mn2-O3	2.086	
Cr3-O2	1.967	Mn2-O4	2.087	
Cr3-O3	1.968	Mn2-N	2.224	
Cr3-O4	1.997	Mn3-O1	2.061	
Cr3-O5	2.010	Mn3-O2	2.095	
Cr3-O6	1.998	Mn3-O3	2.093	

Cr4-O1	1.970	Mn3-O4	2.097
Cr4-O2	1.970	Mn3-N	2.231
Cr4-O3	1.964		
Cr4-O4	1.987		
Cr4-O5	1.989		
Cr4-O6	2.003		
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Angles			
O1-Cr1-O2	83.21	O1-Mn2-O2	96.48
O1-Cr1-O3	83.26	O1-Mn2-O3	93.63
O1-Cr1-O4	95.38	O1-Mn2-O4	94.10
O1-Cr1-O5	175.72	O1-Mn2-O5	174.14
O1-Cr1-O6	94.26	O2-Mn2-O5	87.9
O2-Cr1-O3	82.14	O2-Mn2-O3	118.05
O2-Cr1-O4	93.96	O2-Mn2-O4	118.70
O2-Cr1-O5	94.42	O3-Mn2-O4	121.24
O2-Cr1-O6	176.95	O3-Mn2-O5	87.7
O3-Cr1-O4	175.98	O4-Mn2-O5	80.3
O3-Cr1-O5	92.89	O1-Mn4-O2	92.71
O3-Cr1-O6	95.89	O1-Mn4-O3	94.77
O4-Cr1-O5	88.34	O1-Mn4-O4	93.88
O4-Cr1-O6	87.97	O1-Mn4-N2	175.3
O5-Cr1-O6	87.99	O12-Mn4-O3	116.1
O1-Cr2-O2	82.69	O12-Mn4-O4	118.6
O1-Cr2-O3	83.14	O2-Mn4-N2	91.4
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O1-Cr2-O4	94.8	O3-Mn4-O4	123.9
O1-Cr2-O5	175.05	O3-Mn4-N2	81.3
O1-Cr2-O6	92.61	O4-Mn4-N2	86.2
O2-Cr2-O3	82.14	O1-Mn3-O2	93.78
O2-Cr2-O4	176.92	O1-Mn3-O3	93.96
O2-Cr2-O5	92.75	O1-Mn3-O4	94.65
O2-Cr2-O6	95.80	O1-Mn3-N1	174.6
O3-Cr2-O4	95.74	O2-Mn3-O3	114.6
O3-Cr2-O5	94.35	O2-Mn3-O4	118.4
O3-Cr2-O6	175.47	O2-Mn3-N1	91.3
O4-Cr2-O5	89.62	O3-Mn3-O4	125.4
O4-Cr2-O6	86.16	O3-Mn3-N1	82.2
O5-Cr2-O6	89.77	O3-Mn3-N1	84.5
O1-Cr3-O2	82.91	O1-Mn1-O2	93.75
O1-Cr3-O3	82.34	O1-Mn1-O3	94.23
O1-Cr3-O4	95.49	O1-Mn1-O4	96.38
O1-Cr3-O5	178.85	O1-Mn1-N3	177.3
O1-Cr3-O6	94.76	O2-Mn1-O3	121.6
O2-Cr3-O3	81.98	O2-Mn1-O4	115.4
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O2-Cr3-O4	95.81	O2-Mn1-N3	83.8

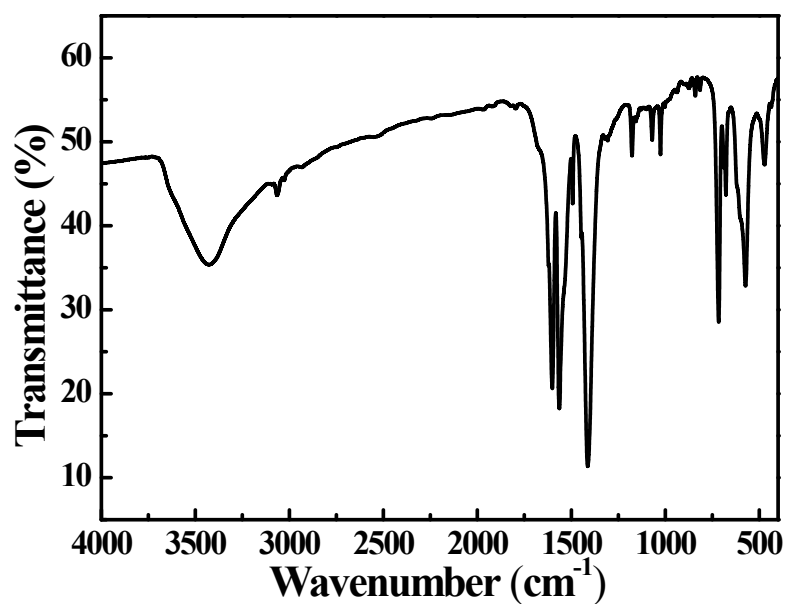
O2-Cr3-O5	96.10	O3-Mn1-O4	120.9
O2-Cr3-O6	177.20	O3-Mn1-N3	86.1
O3-Cr3-O4	177.06	O4-Mn1-N3	85.7
O3-Cr3-O5	96.95		
O3-Cr3-O6	96.19		
O4-Cr3-O5	85.18		
O4-Cr3-O6	85.95		
O5-Cr3-O6	86.22		
O1-Cr4-O2	82.85		
O1-Cr4-O3	82.67		
O1-Cr4-O4	94.90		
O1-Cr4-O5	175.86		
O1-Cr4-O6	94.62		
O2-Cr4-O3	82.68		
O2-Cr4-O4	95.21		
O2-Cr4-O5	93.71		
O2-Cr4-O6	177.17		
O3-Cr4-O4	176.95		
O3-Cr4-O5	94.63		
O3-Cr4-O6	95.77		
O4-Cr4-O5	87.68		
O4-Cr4-O6	86.25		
O5-Cr4-O6	88.77		

**Table S3.** Summary of stability test of cluster **1** and **2** in different solvent for at least 24 h.

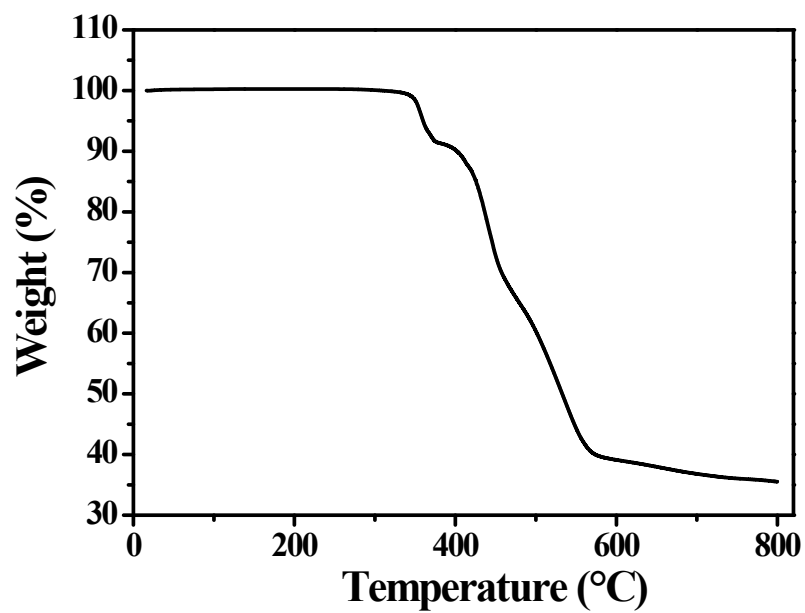
	<b>1</b>	<b>2</b>
methanol	no	yes
ethanol	no	yes
CHCl <sub>3</sub>	no	yes
CCl <sub>4</sub>	no	no
CH <sub>2</sub> Cl <sub>2</sub>	no	no
acetone	no	yes
isopropanol	no	yes
propylene oxide	no	yes
cyclohexanol	no	yes
nitrobenzene	no	no
cyclohexane	no	no
DMSO	no	yes
DMF	yes	yes



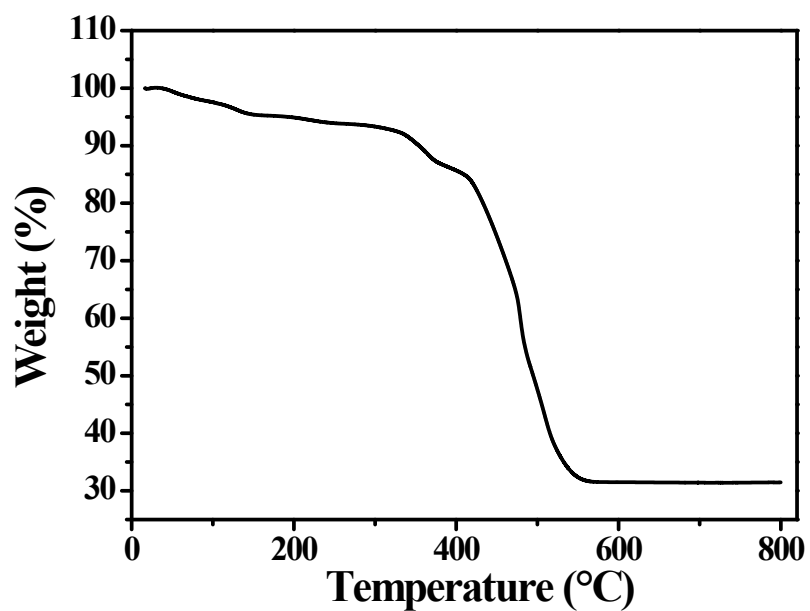
**Figure S6.** Infrared spectra of **1**: characteristic bands of the coordinated carboxylate in  $1400 - 1600 \text{ cm}^{-1}$  as well as  $M - O$  ( $M = \text{Cr, Mn}$ ) vibrations in  $400 - 800 \text{ cm}^{-1}$ , but no free carboxyl group vibrations in  $1650 - 1750 \text{ cm}^{-1}$ . The broad vibration around  $3400 \text{ cm}^{-1}$  arises from water molecules.



**Figure S7.** Infrared spectra of **2**: characteristic bands of the coordinated carboxylate in  $1400 - 1600 \text{ cm}^{-1}$  as well as  $M - O$  ( $M = \text{Cr, Mn}$ ) vibrations in  $400 - 800 \text{ cm}^{-1}$ , but no free carboxyl group vibrations in  $1650 - 1750 \text{ cm}^{-1}$ . The broad vibration around  $3400 \text{ cm}^{-1}$  arises from water molecules.



**Figure S8.** TG curve of cluster 1. The cluster can keep intact above 300 °C, following by decomposition of 1.



**Figure S9.** TG curve of cluster 2. The cluster can keep intact above 200 °C, following by decomposition of 2.



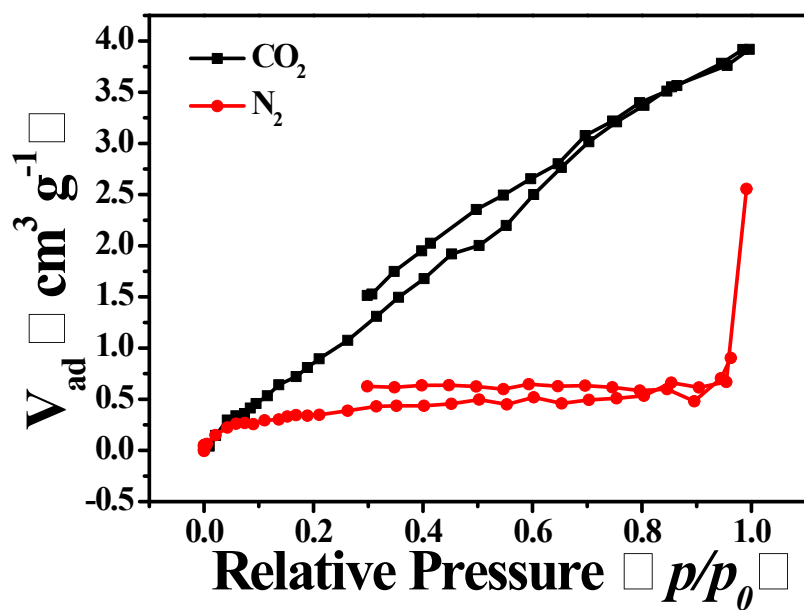


Figure S10.  $\text{CO}_2$  sorption isotherms at 298 K and  $\text{N}_2$  sorption isotherm at 77 K over evacuated **1**.

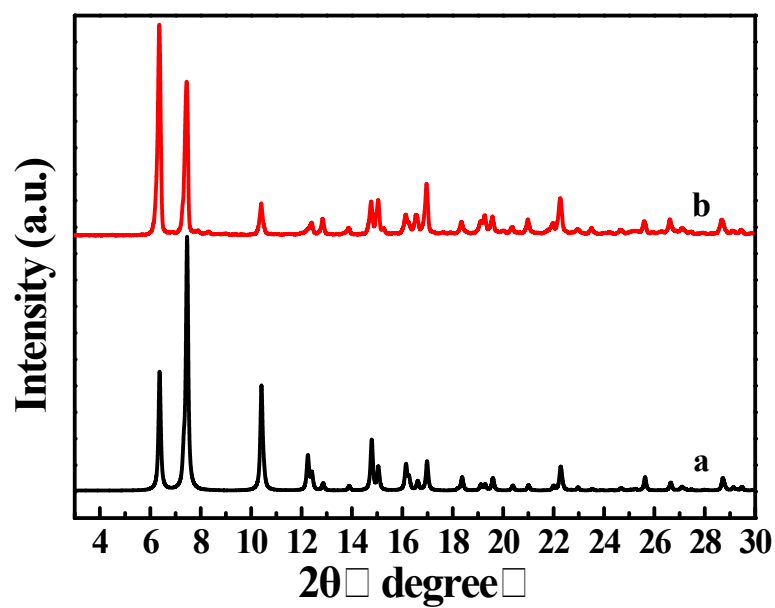
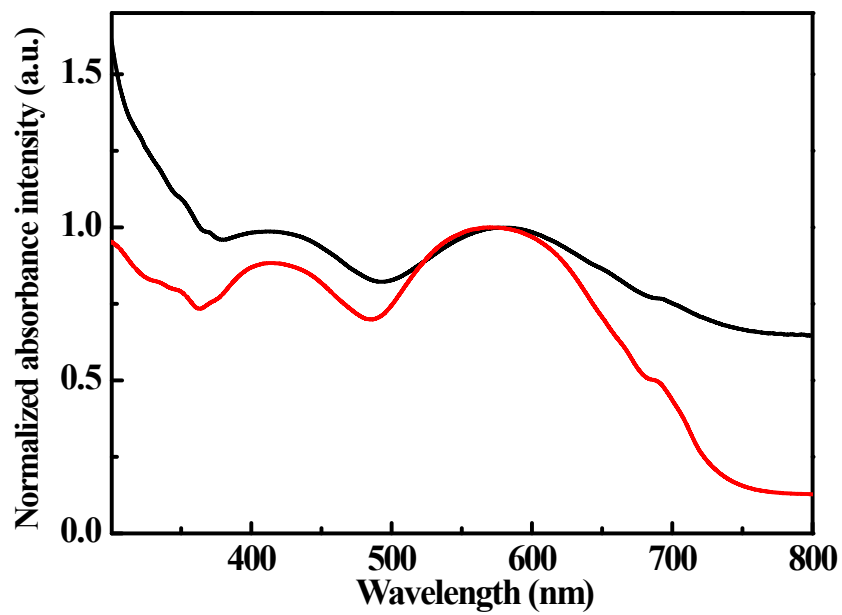


Figure S11. (a) Simulated XRD patterns of **1** nanocluster and (b) PXRD of  $\text{Cr}_8(\text{OH})_8(\text{OOCPh})_{16}$ .



**Figure S12.** Normalized UV-Vis absorption spectrum of  $\text{Cr}_8(\text{OH})_8(\text{OOCPh})_{16}$  (black line) and crystalline **1** sample ( $\lambda_{\text{max}} = 571$  nm, red line).