

Supporting Information

Ligands directed assembly engineering of trapezoidal {Ti₅} building blocks stabilized by dimethylglyoxime

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1. Materials and Methods

All reagents and solvents employed are commercially available and are used as received without further purification. Dimethylglyoxime were bought from Alfa Aesar. $\text{Ti}(\text{O}^i\text{Pr})_4$ (96%), Terephthalic acid, 2-aminoterephthalic acid, 2-Nitroterephthalic acid, 2,5-Dihydroxyterephthalic acid, 2,6-Naphthalenedicarboxylic acid, 4,4'-Biphenyldisulfonic acid, 1,2,4,5-Benzenetetracarboxylic acid and tetrakis(4-carboxyphenyl)porphyrin were bought from Admas-beta. The phase purity of products were confirmed by PXRD using a Rigaku Dmax2500 diffractometer with Cu $\text{K}\alpha$ radiation ($\lambda = 1.54056 \text{ \AA}$) with a step size of $5^\circ/\text{min}$. Thermogravimetric analyses (TGA) were performed using a NETSCH STA-449C thermoanalyzer with a heating rate of $10^\circ\text{C}/\text{min}$ under a nitrogen atmosphere. Fourier transform infrared (FT-IR) spectra were recorded with a Spectrum One FT-IR Spectrometer in the $400\text{-}4000 \text{ cm}^{-1}$ range. The UV-vis diffuse reflection data were recorded at room temperature using a powder sample with BaSO_4 as a standard on a Perkin-Elmer Lambda950 UV-vis spectrophotometer and scanned at 200-800 nm in the reflectance mode with application of the Kubelka-Munk equation, $(F(R) = (1-R)^2/2R)$,¹ where R representing the reflectance. The elemental analyses were performed on an EA1110 CHNS-O CE elemental analyzer.

2. Syntheses

Synthesis of PTC-211

Dimethylglyoxime (0.035 g, 0.3 mmol), terephthalic acid (0.033 g, 0.2 mmol), and isopropyl alcohol (6 ml) were mixed at room temperature and then dropwise $\text{Ti}(\text{O}^i\text{Pr})_4$ (0.5 ml, 1.6 mmol) was added. The resultant solution was heated at 100°C for three days in a glass vial with a polyethylene screw cap. After cooling to and kept at room temperature for a week, yellow rod-like crystals of **PTC-211** were obtained (yield: 70% based on dimethylglyoxime). EA (%) calculated for $\text{C}_{82}\text{H}_{170}\text{N}_4\text{O}_{36}\text{Ti}_{10}$ (2266.90): C, 43.44; H, 7.56; N, 2.47. Found: C, 44.03; H, 7.34; N, 2.39. FT-IR (KBr pellet, cm^{-1}): 2975(m), 2926(w), 2867(w), 2357(w), 1688(m), 1554(s), 1370(s), 1220(s), 1006(s), 945(m), 858(s), 816(w), 738(s), 684(w), 615(w).

Synthesis of PTC-212

It was synthesized in the same way as that of **PTC-211** except that terephthalic acid was replaced by 2-aminoterephthalic acid (0.036 g, 0.2 mmol). Yellow rod-like crystals of **PTC-212** were obtained (yield: 72% based on dimethylglyoxime). EA (%) calculated for $\text{C}_{82}\text{H}_{171}\text{N}_5\text{O}_{36}\text{Ti}_{10}$ (2281.91): C, 43.16; H, 7.55; N, 3.07. Found: C, 43.28; H, 7.46; N, 2.98. FT-IR (KBr pellet, cm^{-1}): 3349(w), 2968(m), 2928(w), 2862(w), 2354(w), 2324(w), 1609(w), 1543(m), 1430(w), 1366(m), 1324(w), 1253(w), 1122(s), 995(s), 956(s), 858(s), 759(s), 627(s), 590(s), 558(s), 445(m).

Synthesis of PTC-213

It was synthesized in the same way as that of **PTC-211** except that terephthalic acid was replaced by 2-Nitroterephthalic acid (0.042 g, 0.2 mmol). Yellow rod-like crystals of **PTC-213** were obtained (yield: 40% based on dimethylglyoxime). EA (%) calculated for $\text{C}_{82}\text{H}_{169}\text{N}_5\text{O}_{38}\text{Ti}_{10}$ (2311.89): C, 42.60; H, 7.37; N, 3.03. Found: C, 42.76; H, 7.23; N, 2.96. FT-IR (KBr pellet, cm^{-1}): 2975(m), 2926(w), 2857(w), 2352(w), 1585(m), 1543(m), 1460(w), 1376(s), 1317(m), 1131(s), 1082(m), 995(s), 955(s), 857(m), 769(m), 604(s), 456(m).

Synthesis of PTC-214

It was synthesized in the same way as that of **PTC-211** except that terephthalic acid was replaced by 2,6-Naphthalenedicarboxylic acid (0.043 g, 0.2 mmol). Yellow rod-like crystals of **PTC-214** were obtained (yield: 47% based on dimethylglyoxime). EA (%) calculated for $\text{C}_{86}\text{H}_{172}\text{N}_4\text{O}_{36}\text{Ti}_{10}$ (2316.95): C, 44.58; H, 7.48; N, 2.42. Found: C, 44.65; H, 7.41; N, 2.36. FT-IR (KBr pellet, cm^{-1}): 2989(w), 2356(w), 2324(w), 1619(m), 1541(m), 1492(w), 1404(s), 1355(s), 1195(s), 1068(m), 960(w), 924(w), 764(m), 647(w), 569(w), 461(m).

Synthesis of PTC-215

It was synthesized in the same way as that of **PTC-211** except that terephthalic acid was replaced by 4,4'-Biphenyldisulfonic acid (0.064 g, 0.2 mmol). Yellow rod-like crystals of **PTC-215** were obtained (yield: 77% based on Dimethylglyoxime). EA (%) calculated for $C_{86}H_{174}N_4O_{38}S_2Ti_{10}$ (2415.10): C, 42.77; H, 7.26; N, 2.32. Found: C, 42.84; H, 7.23; N, 2.28. FT-IR (KBr pellet, cm^{-1}): 2968(m), 2928(w), 2870(w), 2360(m), 2332(m), 1608(m), 1460(w), 1373(m), 1324(w), 1266(w), 1138(s), 992(s), 953(s), 845(m), 757(m), 718(m), 591(s), 463(m).

Synthesis of PTC-216

Dimethylglyoxime (0.035 g, 0.3 mmol), 2,5-Dihydroxyterephthalic acid (0.040 g, 0.2 mmol), and isopropyl alcohol (6 ml) were mixed at room temperature and then dropwise $Ti(O^iPr)_4$ (0.5 ml, 1.6 mmol) was added. The resultant solution was heated at 100°C for three days. After cooled to room temperature, red block crystals of **PTC-216** were obtained (yield: 80% based on dimethylglyoxime). EA (%) calculated for $C_{111}H_{220}O_{57}N_4Ti_{16}$ (3288.79): C, 40.53; H, 6.74; N, 1.71. Found: C, 40.37; H, 6.78; N, 1.61. FT-IR (KBr pellet, cm^{-1}): 2971(m), 2932(w), 2873(w), 2364(w), 1611(m), 1552(m), 1445(s), 1366(m), 1327(m), 1249(m), 1220(w), 1141(s), 1005(s), 907(m), 828(m), 780(w), 672(w), 613(s), 466(m).

Synthesis of PTC-217

It was synthesized in the same way as that of **PTC-211** except that terephthalic acid was replaced by 1,2,4,5-Benzenetetracarboxylic acid (0.051 g, 0.2 mmol). Yellow rod-like crystals of **PTC-217** were obtained (yield: 61% based on dimethylglyoxime). EA (%) calculated for $C_{96}H_{198}N_4O_{46}Ti_{12}$ (2718.99): C, 42.41; H, 7.34; N, 2.06. Found: C, 41.78; H, 6.93; N, 2.11. FT-IR (KBr pellet, cm^{-1}): 2968(m), 2925(w), 2857(w), 2362(w), 2331(w), 1719(m), 1581(m), 1420(m), 1360(m), 1247(m), 1121(s), 999(s), 956(s), 853(m), 820(w), 797(w), 763(m), 590(s), 503(w), 463(w).

Synthesis of PTC-218

It was synthesized in the same way as that of **PTC-211** except that terephthalic acid was replaced by meso-Tetra(4-carboxyphenyl)porphine (0.079 g, 0.1 mmol). Crimson block crystals of **PTC-218** were obtained (yield: 60% based on dimethylglyoxime). EA (%) calculated for $C_{196}H_{358}O_{72}N_{12}Ti_{20}$ (4992.30): C, 47.15; N, 3.37; H, 7.23. Found: C, 47.28; N, 3.21; H, 7.16. FT-IR (KBr pellet, cm^{-1}): 2975(m), 2916(w), 2857(w), 2358(w), 1605(m), 1543(m), 1468(w), 1390(s), 1331(m), 1135(s), 1077(m), 998(s), 959(s), 842(m), 803(w), 754(m), 725(w), 578(s), 451(m).

3. Structural information and physical characterization of PTC-211 to 218

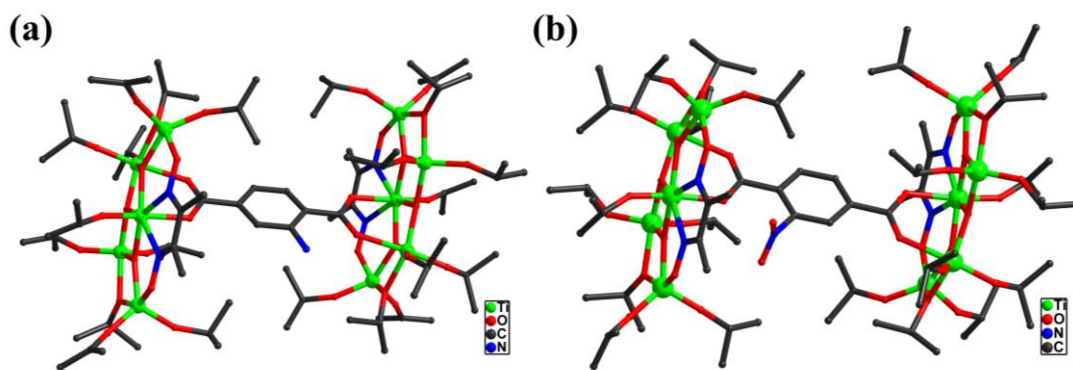


Fig. S1 The sandwich-like $\{Ti_{10}\}$ molecular cluster of PTC-212 (a), PTC-213 (b). All the H atoms are omitted for clarity. Color codes: green Ti; black C; red O; blue N.

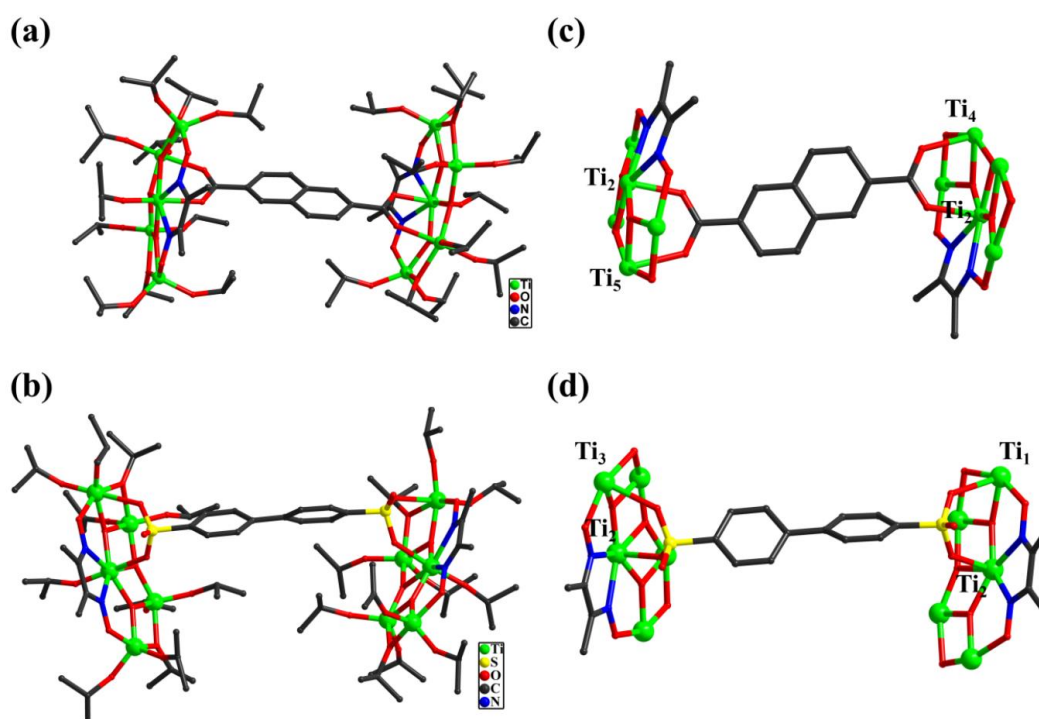


Fig. S2 The sandwich-like $\{Ti_{10}\}$ molecular cluster of PTC-214 (a, c) and PTC-215 (b, d) with different coordination environments. All the H atoms are omitted for clarity. Color codes: green Ti; black C; red O; yellow S; blue N.

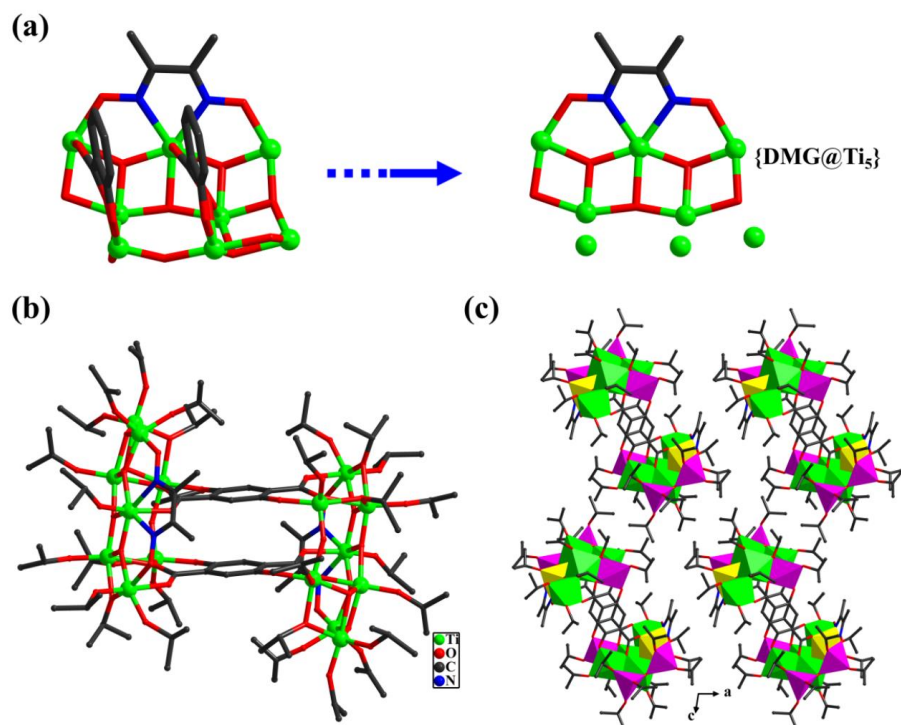


Fig. S3 The Synthetic and structural evolution of $\{\text{Ti}_8\}$ unit in **PTC-216** containing one $\{\text{DMG@Ti}_5\}$ unit (a); the sandwich-like $\{\text{Ti}_{16}\}$ molecular cluster of **PTC-216** (b); and the packing view of **PTC-216** along the b-axis (c). All the H atoms are omitted for clarity. Color codes: green Ti; black C; red O; blue N. Polyhedral color code: pink TiO_5 ; green TiO_6 ; yellow TiO_5N_2 .

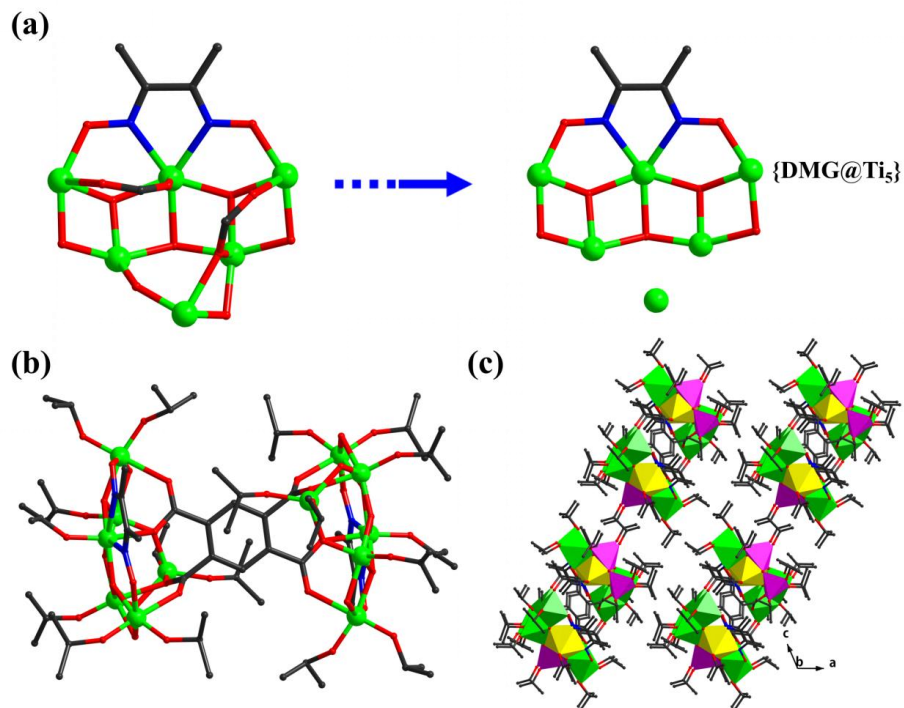


Fig. S4 The Synthetic and structural evolution of $\{\text{Ti}_6\}$ unit in **PTC-217** containing one $\{\text{DMG@Ti}_5\}$ unit (a); the sandwich-like $\{\text{Ti}_{12}\}$ molecular cluster of **PTC-217** (b); and the packing view of **PTC-217** along the b-axis (c). All the H atoms are omitted for clarity. Color codes: green Ti; black C; red O; blue N. Polyhedral color code: pink TiO_5 ; green TiO_6 ; yellow TiO_5N_2 .

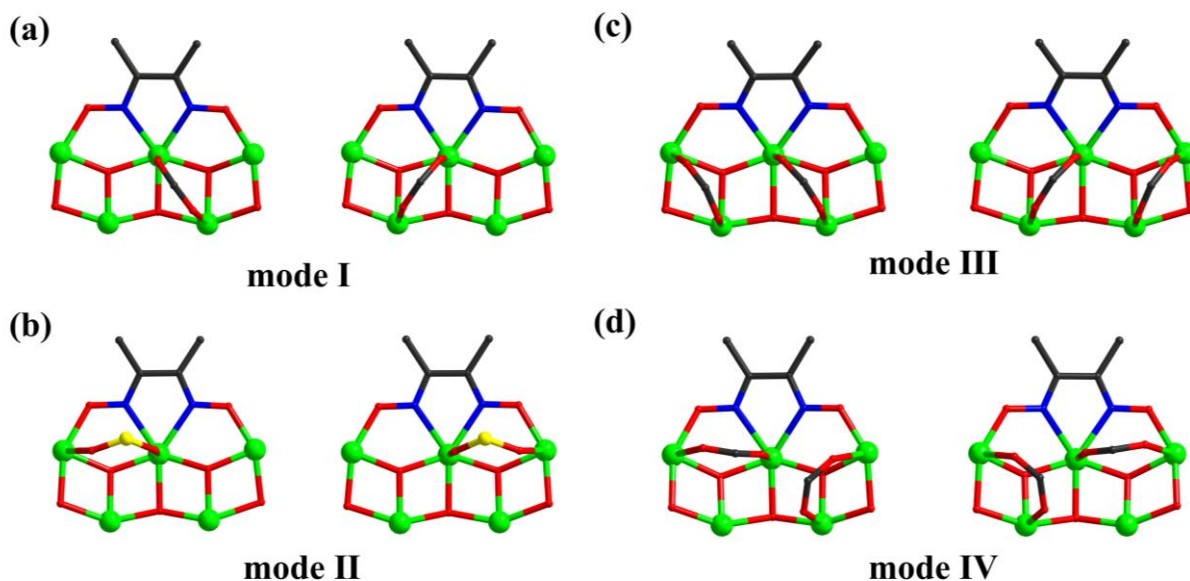


Fig. S5 Bridging coordination modes of $\{\text{DMG@Ti}_5\}$ unit. Color codes: green Ti; black C; red O; yellow S; blue N.

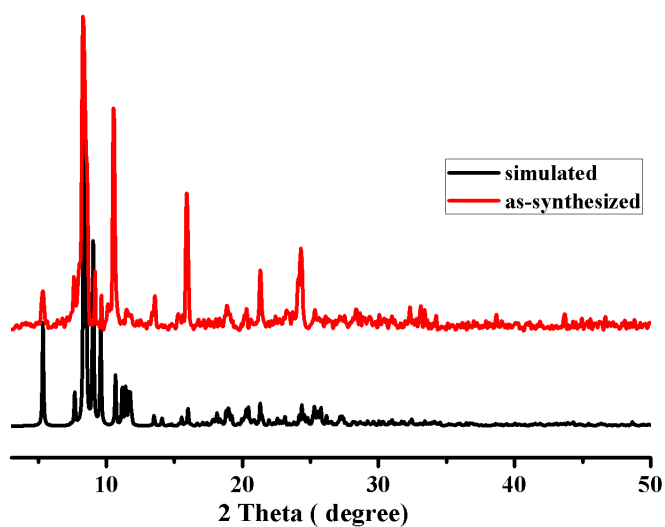


Fig. S6 The PXRD of PTC-211: simulated pattern (black), experimental (red).

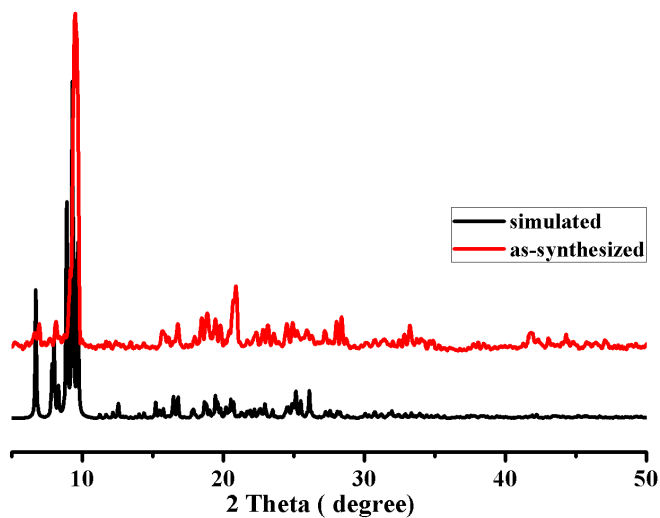


Fig. S7 The PXRD of PTC-212: simulated pattern (black), experimental (red).

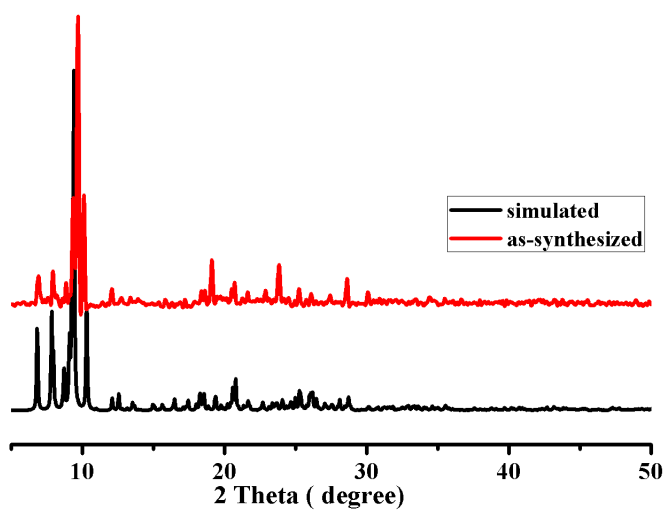


Fig. S8 The PXRD of PTC-213: simulated pattern (black), experimental (red).

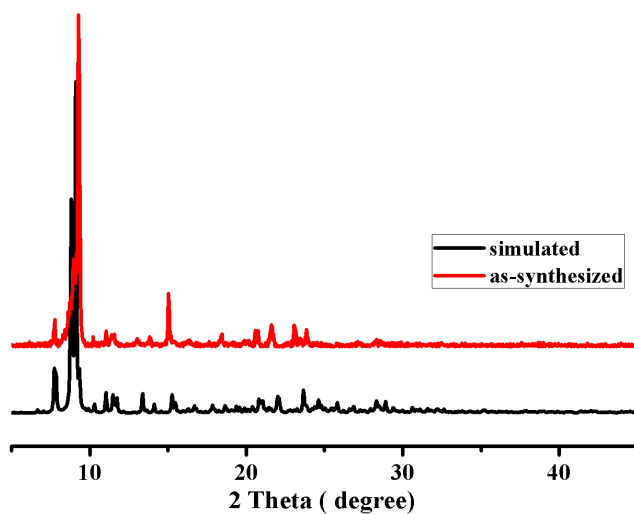


Fig. S9 The PXRD of PTC-214: simulated pattern (black), experimental (red).

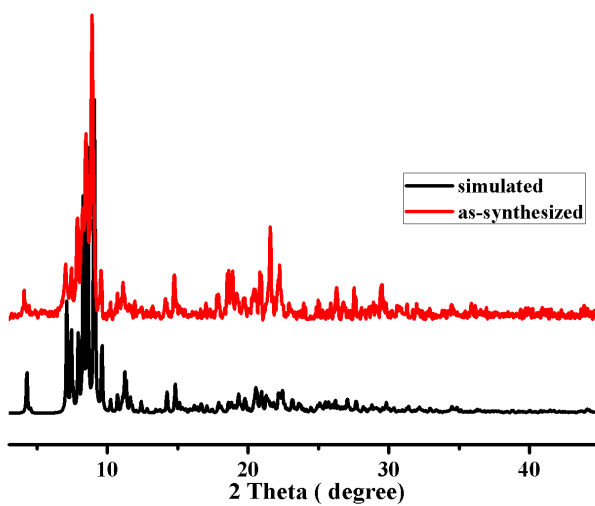


Fig. S10 The PXRD of PTC-215: simulated pattern (black), experimental (red).

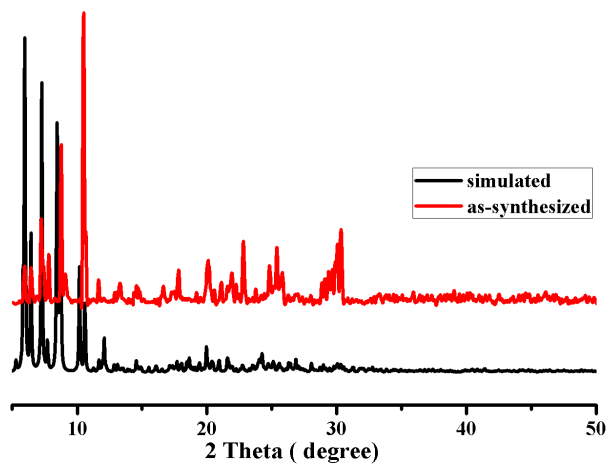


Fig. S11 The PXRD of PTC-216: simulated pattern (black), experimental (red).

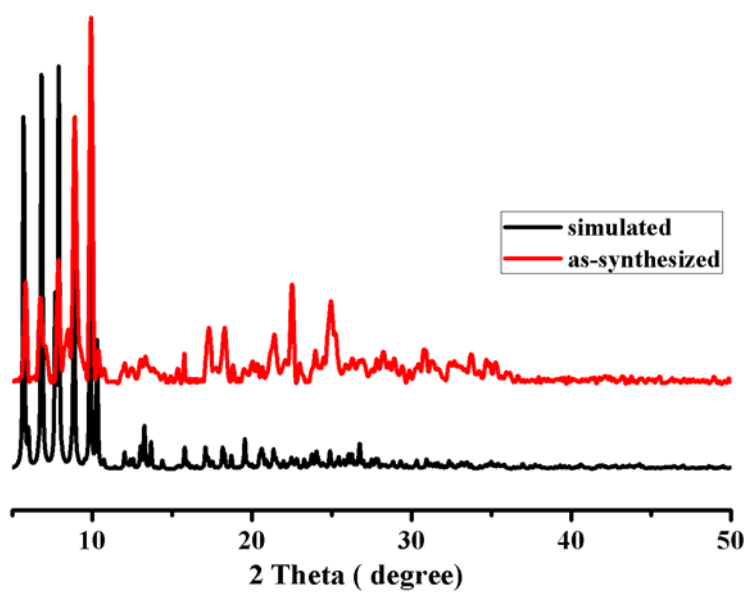


Fig. S12 The PXRD of PTC-217: simulated pattern (black), experimental (red).

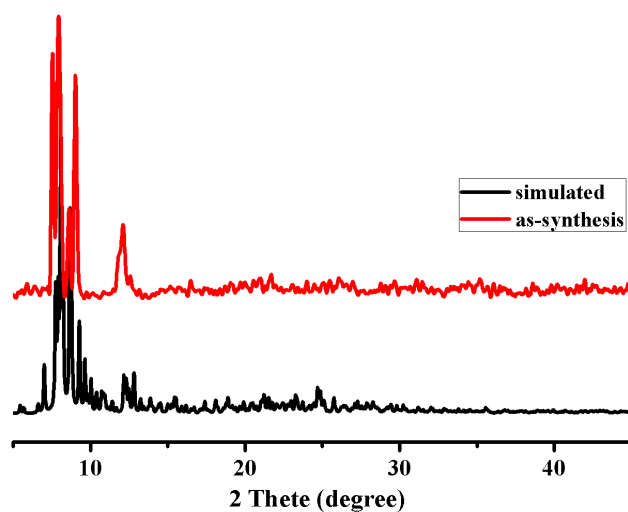


Fig. S13 The PXRD of PTC-218: simulated pattern (black), experimental (red).

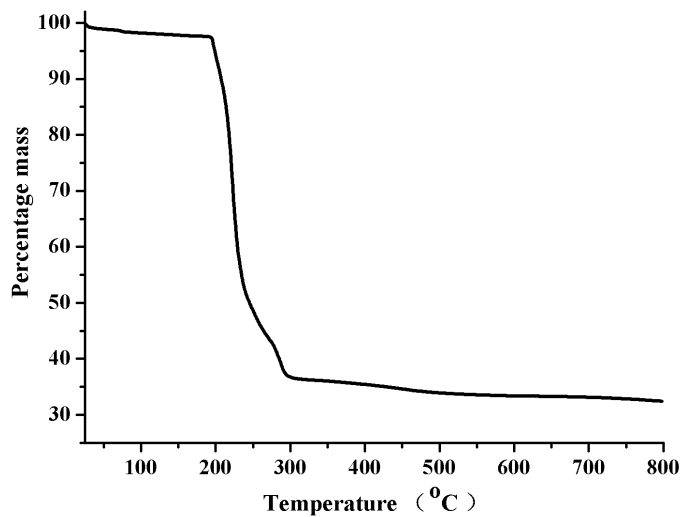


Fig. S14 TGA curve of PTC-211.

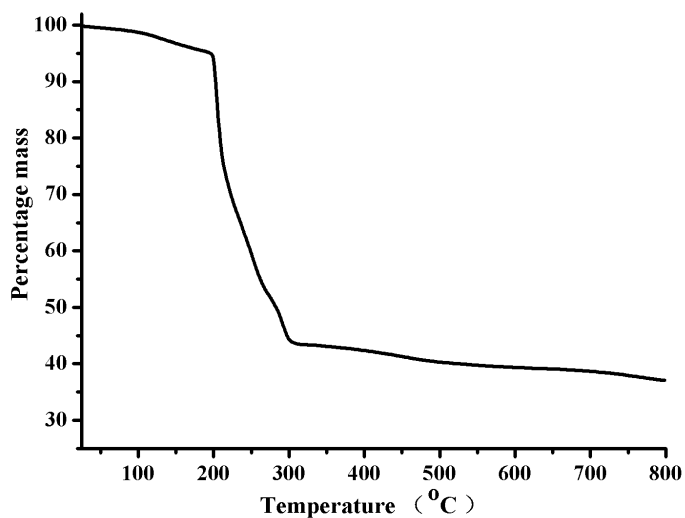


Fig. S15 TGA curve of PTC-212.

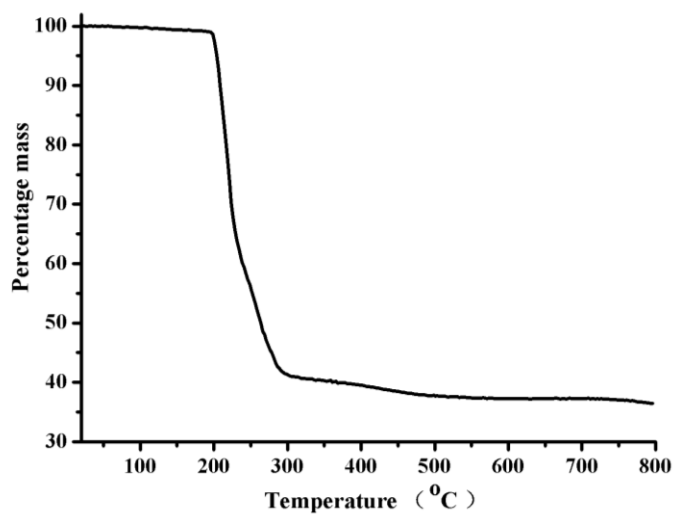


Fig. S16 TGA curve of PTC-213.

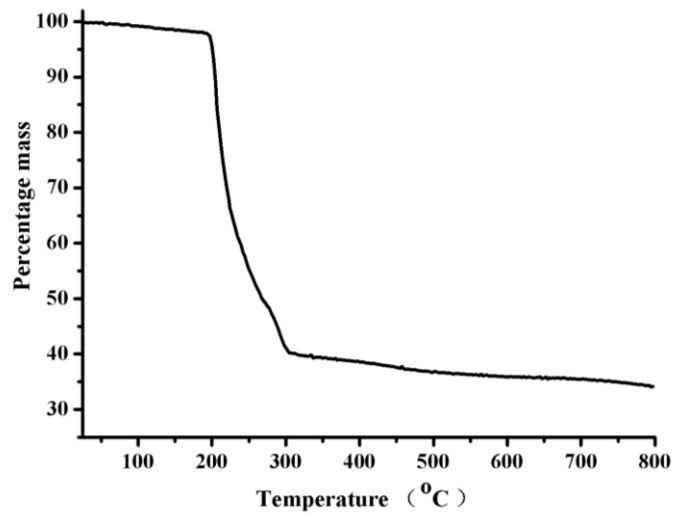


Fig. S17 TGA curve of PTC-214.

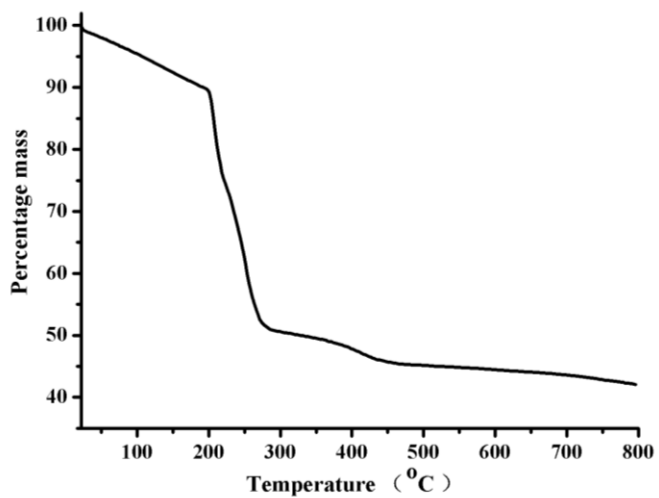


Fig. S18 TGA curve of PTC-215.

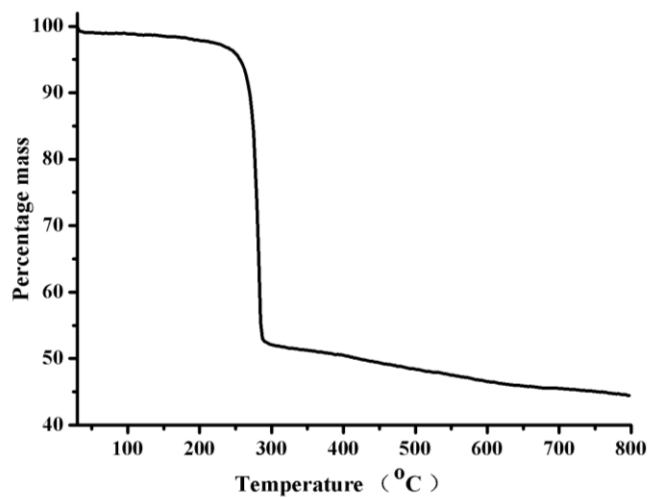


Fig. S19 TGA curve of PTC-216.

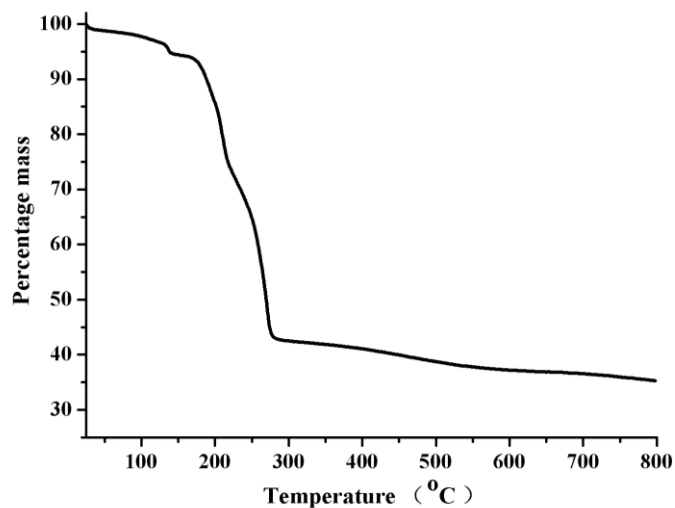


Fig. S20 TGA curve of PTC-217.

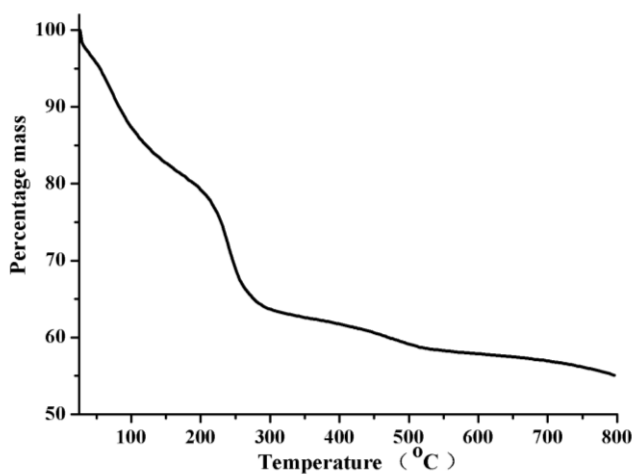


Fig. S21 TGA curve of PTC-218.

Table S1. Summary of bandgaps of the reported complexes.

No.	Complex	Bandgap	Stabilizing ligand	Bridging ligand
1	PTC-211	2.31	DMG	BDC
2	PTC-212	2.26	DMG	2-NH ₂ -BDC
3	PTC-213	2.21	DMG	2-NO ₂ -BDC
4	PTC-214	2.45	DMG	2,6-NDC
5	PTC-215	2.27	DMG	BPDC
6	PTC-216	1.78	DMG	DHPC
7	PTC-217	2.34	DMG	BETC
8	PTC-218	1.56	DMG	TCPP

4. General Methods for X-ray Crystallography

Crystallographic data of **PTC-214** was collected on a Mercury single crystal diffractometer equipped with graphite-monochromatic Mo K α radiation ($\lambda = 0.71073 \text{ \AA}$). Crystallographic data of **PTC-215**, **PTC-216** and **PTC-217** were collected on a oxford XCalibur E CCD diffractometer equipped with graphite-monochromatic Mo K α radiation ($\lambda = 0.71073 \text{ \AA}$). While crystallographic data of **PTC-211**, **PTC-212**, **PTC-213** and **PTC-218** were collected on Supernova single crystal diffractometer equipped with graphite-monochromatic Cu K α radiation ($\lambda = 1.54178 \text{ \AA}$) at room temperature. The structures were solved with direct methods using SHELXS-97² and refined with the full-matrix least-squares technique based on F^2 using the SHELXL-97³. Non-hydrogen atoms were refined anisotropically, and all hydrogen atoms bond C were generated geometrically. The X-ray crystallographic coordinates for structures reported in this article have been deposited at the Cambridge Crystallographic Data Centre (CCDC) under deposition numbers CCDC 1914954-1914961 for **PTC-211** to **PTC-218**. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via http://www.ccdc.cam.ac.uk/data_request/cif.

5. Supplementary Table S2. A summary of crystallography data for **PTC-211 to PTC-218**. Detailed data are given in **Tables S3** and **S4**.

No.	Complex	Formula	M _w (g/mol)	Sp. Gr.	a(Å)	b(Å)	c(Å)	α (°)	β(°)	γ (°)	V(Å ³)	R/%
1.	PTC-211	Ti ₁₀ O ₆ (DMG) ₂ (BDC)(O ⁱ Pr) ₂₂	2266.90	<i>P</i> 2 ₁ / <i>n</i>	10.9245(2)	23.0505(6)	23.9209(6)	90	92.241(2)	90	6019.0(2)	6.87
2.	PTC-212	Ti ₁₀ O ₆ (DMG) ₂ (2-NH ₂ -BDC)(O ⁱ Pr) ₂₂	2281.91	<i>P</i> -1	12.2992(10)	14.3904(15)	19.0996(17)	85.923(8)	85.061(7)	66.484(9)	3085.7(5)	8.05
3.	PTC-213	Ti ₁₀ O ₆ (DMG) ₂ (2-NO ₂ -BDC)(O ⁱ Pr) ₂₂	2311.89	<i>P</i> -1	11.8011(4)	13.8482(7)	18.9934(7)	82.343(4)	85.367(3)	70.160(4)	2891.6(2)	8.55
4.	PTC-214	Ti ₁₀ O ₆ (DMG) ₂ (2,6-NDC)(O ⁱ Pr) ₂₂	2316.95	<i>P</i> 2 ₁ / <i>c</i>	19.4857(4)	26.4883(7)	22.5872(5)	90	94.059(2)	90	11629.0(5)	7.71
5.	PTC-215	Ti ₁₀ O ₆ (DMG) ₂ (BPDC)(O ⁱ Pr) ₂₂	2415.10	<i>P</i> -1	13.1334(4)	23.0232(10)	23.9960(11)	63.989(4)	78.288(3)	82.373(3)	6377.2(5)	8.43
6.	PTC-216	Ti ₁₆ O ₁₂ (DMG) ₂ (DHPC) ₂ (O ⁱ Pr) ₂₈	3288.79	<i>P</i> -1	15.4097(7)	15.8879(6)	17.9294(7)	109.617(3)	96.107(3)	95.889(3)	4066.3(3)	6.78
7.	PTC-217	Ti ₁₂ O ₈ (DMG) ₂ (BETC)(O ⁱ Pr) ₂₄	2718.99	<i>P</i> -1	14.7714(7)	15.1120(7)	16.8580(7)	93.516(4)	111.982(4)	100.404(4)	3397.7(3)	7.19
8.	PTC-218	Ti ₂₀ O ₁₂ (DMG) ₄ (TCPP)(O ⁱ Pr) ₄₄	4992.30	<i>P</i> 2 ₁ / <i>c</i>	22.1405(13)	26.7150(10)	25.288(3)	90	120.487(5)	90	12889.5(19)	12.53

Table S3. Crystallographic data and structure refinement summary for **PTC-211** to **PTC-214**.

	PTC-211	PTC-212	PTC-213	PTC-214
Empirical formula	C ₈₂ H ₁₇₀ N ₄ O ₃₆ Ti ₁₀	C ₈₂ H ₁₇₁ N ₅ O ₃₆ Ti ₁₀	C ₈₂ H ₁₆₉ N ₅ O ₃₈ Ti ₁₀	C ₈₆ H ₁₇₂ N ₄ O ₃₆ Ti ₁₀
M _r	2266.90	2281.91	2311.89	2316.95
T/K	293(2)	293(2)	99.98(13)	293(2)
Crystal system	Monoclinic	Triclinic	Triclinic	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> 2 ₁ / <i>c</i>
a/Å	10.9245(2)	12.2992(10)	11.8011(4)	19.4857(4)
b/Å	23.0505(6)	14.3904(15)	13.8482(7)	26.4883(7)
c/Å	23.9209(6)	19.0996(17)	18.9934(7)	22.5872(5)
α (°)	90	85.923(8)	82.343(4)	90
β (°)	92.241(2)	85.061(7)	85.367(3)	94.059(2)
γ (°)	90	66.484(9)	70.160(4)	90
V/Å ³	6019.0(2)	3085.7(5)	2891.6(2)	11629.0(5)
Z	2	1	1	4
Dc/mg m ⁻³	1.251	1.228	1.324	1.325
μ/mm ⁻¹	5.919	5.778	6.187	0.720
indep reflns [<i>I</i> > 2σ(<i>I</i>)]	11931	12141	11438	18245
F(000)	2396	1205	1213	4904
GOF	1.032	1.065	0.989	1.094
R ₁ ^a , wR ₂ ^b [<i>I</i> > 2σ(<i>I</i>)]	0.0691, 0.2097	0.1334, 0.3760	0.0991, 0.2935	0.0771, 0.1818
R ₁ ^a , wR ₂ ^b (all data)	0.0955, 0.2353	0.1869, 0.4470	0.1234, 0.3370	0.1024, 0.1911

$$^a R_1 = \sum(|F_o| - |F_c|) / \sum |F_o|, \quad ^b wR_2 = [\sum w(|F_o|^2 - |F_c|^2)^2 / \sum w(F_o^2)]^{1/2}.$$

Table S4. Crystallographic data and structure refinement summary for **PTC-215** to **PTC-218**.

	PTC-215	PTC-216	PTC-217	PTC-218
Empirical formula	C ₈₆ H ₁₇₄ N ₄ O ₃₈ S ₂ Ti ₁₀	C ₁₁₁ H ₂₂₀ O ₅₇ N ₄ Ti ₁₆	C ₉₆ H ₁₉₈ N ₄ O ₄₆ Ti ₁₂	C ₁₉₆ H ₃₅₈ N ₁₂ O ₇₂ Ti ₂₀
M _r	2415.10	3288.79	2718.99	4992.30
T/K	293(2)	293(2)	293(2)	293(2)
Crystal system	Triclinic	Triclinic	Triclinic	Monoclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> /Å	13.1334(4)	15.4097(7)	14.7714(7)	22.1405(13)
<i>b</i> /Å	23.0232(10)	15.8879(6)	15.1120(7)	26.7150(10)
<i>c</i> /Å	23.9960(11)	17.9294(7)	16.8580(7)	25.288(3)
α (°)	63.989(4)	109.617(3)	93.516(4)	90
β (°)	78.288(3)	96.107(3)	111.982(4)	120.487(5)
γ (°)	82.373(3)	95.889(3)	100.404(4)	90
<i>V</i> /Å ³	6377.2(5)	4066.3(3)	3397.7(3)	12889.5(19)
<i>Z</i>	2	1	1	2
Dc/mg m ⁻³	1.252	1.313	1.269	1.286
μ/mm ⁻¹	0.691	0.812	0.735	5.583
indep reflns [<i>I</i> > 2σ(<i>I</i>)]	22977	14264	11911	24068
F(000)	2534	1674	1364	5268
GOF	1.174	1.074	1.090	1.087
R ₁ ^a , wR ₂ ^b [<i>I</i> > 2σ(<i>I</i>)]	0.1219, 0.3269	0.1228, 0.3639	0.1115, 0.3081	0.1490, 0.3694
R ₁ ^a , wR ₂ ^b (all data)	0.2051, 0.4041	0.1695, 0.4190	0.1663, 0.3594	0.2870, 0.4859

$$^a R_1 = \sum(|F_o| - |F_c|) / \sum |F_o|, \quad ^b wR_2 = [\sum w(|F_o|^2 - |F_c|^2)^2 / \sum w(F_o^2)]^{1/2}.$$

6. References

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