

Synthesis, Structural Properties and Reactivity of Ruthenocene-based Pincer Pd(II) Tetrahydroborate

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Experimental Section

All manipulations were performed under a dry argon atmosphere using standard Schlenk technique. Commercially available argon (99.9%) was additionally purified from traces of oxygen and moisture by sequential passage through Ni/Cr catalyst column and 4 Å molecular sieves. The HPLC grade solvents (Acros Organics) were used for sample preparation after additional purification by standard procedures. Dichloromethane (DCM) and toluene were dehydrated over CaH_2 and Na/benzophenone, respectively. All solvents were freshly distilled under argon prior to use. Deuterated were distilled and degassed by three freeze–pump–thaw cycles prior to use. {2,5-Bis(*tert*-butylphosphinomethyl)ruthenocen-1-yl}palladium chloride ($^{t\text{Bu}}\text{PC}^{\text{Ru}}\text{P}\text{PdCl}$ (where $^{t\text{Bu}}\text{PC}^{\text{Ru}}\text{P} = \kappa^3\text{-}\{2,5-(^{t\text{Bu}}\text{PCH}_2)_2\text{C}_5\text{H}_2\}\text{Ru}(\text{C}_5\text{H}_5)\})$ was prepared as previously described.¹

FTIR spectra (Figures S10–S11, S24, S38 and S45) were measured on Shimadzu IR Prestige 21 FTIR spectrometer. NMR spectra (Figures S4–S15, S25, S27–S28, S30–S31, S35–S39, S42–S44) were recorded on a Bruker Avance II 400 MHz and 500 MHz spectrometers. ^1H chemical shifts are reported in parts per million (ppm) downfield to tetramethylsilane (TMS) and were calibrated against the residual solvent resonance, while $^{31}\text{P}\{^1\text{H}\}$ signals were referenced to 85% H_3PO_4 and ^{11}B – to $\text{BF}_3\cdot\text{Et}_2\text{O}$. **Elemental analyses** were carried out in the Laboratory of Microanalysis of INEOS RAS. Because the automatic CH analyzer cannot be used for the Ru-containing compounds due to formation of volatile oxides, the classic manual technique was used. The sample was burned in a stream of oxygen at 950 °C followed by trapping CO_2 and water by the Ascaris (asbestos impregnated with NaOH) and Anhydron (anhydrous magnesium perchlorate), respectively, and the

analysis of the mass changes. Attempted high-resolution MS analysis of this series of complexes revealed easy (^tBuPC^{Ru}P)Pd-X → (^tBuPC^{Ru}P)Pd⁺ + X⁻ dissociation (X = BH₄⁻, OC(O)H⁻, OC(O)OH⁻), the only ions observed at the conditions required for ionization being (^tBuPC^{Ru}P)Pd⁺ (m/z 653.12) and its adduct with acetonitrile (used as a solvent) (^tBuPC^{Ru}P)Pd(NCCH₃)⁺ (m/z 694.15) for all compounds.

X-ray diffraction. Single crystals of **1** (Figure S1), **2** (Figure S22), **4** (Figure S32) and **5** (Figure S39) were obtained by slow solvent evaporation from toluene/hexane mixture. X-ray diffraction experiments were carried out with a SMART APEX2 DUO CCD diffractometer for the compound **2** and **4** with a SMART APEX2 CCD diffractometer for **1** and **5**, using graphite-monochromated Mo-K_α radiation ($\lambda = 0.71073 \text{ \AA}$, ω -scans) at 120 K. The structures were solved by direct method and refined by the full-matrix least-squares against F^2 in anisotropic approximation for non-hydrogen atoms. Hydrogen atoms of the borohydride anion in **1** were found in difference Fourier synthesis and their coordinates and isotropic displacement parameters were refined freely. Hydrogen atoms of the borohydride anion in **2** and OH group in **5** were found in difference Fourier synthesis and refined isotropically in the riding model. Positions of other hydrogen atoms were calculated, and all of them were refined in isotropic approximation in the riding model. In **2**, the anion is, apparently, disordered by two positions. The best obtained model of this disorder resulted in R1 = 0.0465, wR2 = 0.1634 and GOOF= 1.147. Crystal data and structure refinement parameters for **1**, **2**, **4** and **5** are given in Table S1. All calculations were performed using the SHELXTL software.²

Full crystallographic data have been deposited with the CCDC as 1882669 (**1**), 1882671 (**2**), 1892812 (**4**), 1882670 (**5**). These data can be obtained free of charge via <http://www.ccdc.cam.ac.uk/conts/retrieving.html>, or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: (+44) 1223-336-033; or e-mail: deposit@ccdc.cam.ac.uk.

Preparation of (^tBuPC^{Ru}P)Pd($\eta^{1,2}$ -BH₄) (**1**)

(^tBuPC^{Ru}P)PdCl (100 mg, 0.145 mmol) was added to a suspension of NaBH₄ (50 mg, 1.32 mmol) in absolute EtOH (10 mL) at room temperature. The reaction mixture was refluxed for 1 h, then another portion of NaBH₄ (50 mg, 1.32 mmol) was added and the boiling was continued for 1 h. The reaction mixture was then cooled and evaporated *in vacuo* to a minimum volume. To the residue 10 ml of distilled H₂O was added, the resulting suspension was stirred at room temperature for 1 h. The reaction mass was extracted (3 × 10 ml) with hexane-CH₂Cl₂ (3:1). The solution was evaporated, and the residue was dried *in vacuo* at room temperature. The product is a pale yellow powder.

Yield: 93 mg (96%). The crystals suitable for XRD analysis were grown by slow solvent evaporation from toluene/hexane mixture under argon flow.

Anal. Calc. (%) for C₂₈H₅₁BP₂RuPd (%): C 50.34, H 7.71. **Found**: C, 50.61; H, 7.90. **FTIR (KBr pellet, v in cm⁻¹)**: 3097, 2989, 2959, 2926, 2989, 2866, 2364 (B–H_{term}^{as}), 2292 (B–H_{term}^s), 1993 (B–H_{br}–Pd), 1840 (B–H_{br}–Pd), 1474, 1466, 1411, 1391, 1369, 1363, 1180, 1168, 1129, 1099, 1053.

NMR (298 K, C₆D₆, δ in ppm): ¹H: 0.20 (br m, 4H, BH₄⁻), 1.18, (vt, J_{HP} = 13.2 Hz 18H, CH₃), 1.40 (vt, J_{HP} = 14.2 Hz, 18H, CH₃), 2.34 (dvt, J_{HH} = 16.7 Hz, J_{HP} = 8.8 Hz, 2H, CH_AH_BP), 2.64 (dvt, J_{HH} = 16.7 Hz, J_{HP} = 5.6 Hz, 2H, CH_AH_BP), 4.44 (s, 5H, C₅H₅), 4.65 (s, 2H, C₅H₂). ¹¹B{¹H}: -35.75 (br m). ³¹P{¹H}: 89.91 (s).

NMR (298 K, toluene-d₈, δ in ppm): ¹H: 0.02 (br m, 4H, BH₄⁻), 1.20, (vt, J_{HP} = 13.2 Hz 18H, CH₃), 1.41 (vt, J_{HP} = 14.2 Hz, 18H CH₃), 2.37 (dvt, J_{HH} = 16.6 Hz, J_{HP} = 8.6 Hz, 2H, CH_AH_BP), 2.66 (dvt, J_{HH} = 16.6 Hz, J_{HP} = 5.4 Hz, 2H, CH_AH_BP), 4.39 (s, 5H, C₅H₅), 4.63 (s, 2H, C₅H₂). ¹¹B{¹H}: -35.85 (br m). ³¹P{¹H}: 89.75 (s).

Preparation of (^tBuPC^{Ru}P)Pd(η^{1,2}-BD₄) (1-d₄)

Complex **1-d₄** was prepared by the same procedure using NaBD₄ instead of NaBH₄, EtOD as a solvent and D₂O for hydrolysis of excess NaBD₄. Yield: 21 mg (86%).

FTIR (KBr pellet, v in cm⁻¹): 3096, 2988, 2960, 2899, 2868, 1775 (B–D_{term}^{as}), 1750 (B–D_{term}^s), 1680 (B–D_{br}–Pd), 1622 (B–D_{br}–Pd), 1474, 1466, 1411, 1391, 1369, 1337, 1319, 1300, 1266, 1180, 1168, 1140, 1129, 1099.

NMR (298 K, C₆D₆, δ in ppm): ²H: 0.14 (s, BD₄⁻), ¹H: 1.18, (vt, J_{HP} = 13.2 Hz, 18H, CH₃), 1.40 (vt, J_{HP} = 14.2 Hz, 18H CH₃), 2.34 (dvt, J_{HH} = 16.6 Hz, J_{HP} = 8.6 Hz, 2H, CH_AH_BP), 2.64 (dvt, J_{HH} = 16.6 Hz, J_{HP} = 5.4 Hz, 2H, CH_AH_BP), 4.44 (s, 5H, C₅H₅), 4.66 (s, 2H, C₅H₂). ¹¹B{¹H}: -36.30 (br m). ³¹P{¹H}: 89.83 (s).

Reaction of **1** and **1-d₄** with (CF₃)₂CHOH (HFIP).

Formation of {(^tBuPC^{Ru}P)Pd}₂(μ²:η^{1,2}-BH₄) [B{OCH(CF₃)₂}₄] (**2**).

1 (3.5 mg, 0.0052 mmol) was weighed into standard NMR tube and dissolved in toluene-d₈ (0.5 mL). After addition of HFIP (15 μL, ca. 25 molar equiv.) by syringe the immediate formation of bubbles was observed. The reaction was monitored by ³¹P NMR spectroscopy. After two days, the quantitative conversion of complex **1** into the product **2** was observed. The crystals suitable for XRD analysis were grown by slow solvent evaporation from toluene/hexane mixture under argon flow. Scaling-up the preparation of this complex seems to be a non-trivial task since the crystallization

relies on slow solvent evaporation and is sensitive to surface area - reaction mixture volume ratio, solvent vapor pressure etc. For this reason the element analysis was not performed for this compound.

FTIR (KBr, v in cm⁻¹): 3422, 3097, 2963, 2920, 2871, 2416 (B–H_{term^{as}}), 2389 (B–H_{term^s}), 2029 (B–H_{br^s}), 1918 (B–H_{br^{as}}), 1474, 1381, 1372, 1292, 1262, 1216, 1180, 1138, 1101, 1054, 1020, 1000.

NMR (298 K, C₆D₆, δ in ppm): ¹H: -1.0 (br s, 4H, BH₄⁻), 1.32–1.07, (m, 72H, CH₃), 1.40 (vt, J_{HP} = 14.2 Hz, 18H, CH₃), 2.34 (dvt, J_{HH} = 16.7 Hz, J_{HP} = 8.8 Hz, 2H, CH_AH_BP), 2.64 (dvt, J_{HH} = 16.7 Hz, J_{HP} = 5.6 Hz, 2H, CH_AH_BP), 4.44 (s, 5H, C₅H₅), 4.65 (s, 2H, C₅H₂). ¹¹B{¹H}: -39.0 (br m). ³¹P{¹H}: δ_{centr} 90.58 (J_{PP} 329 Hz, δ_A 93.9, δ_B 87.3).

Formation of {(^tBuPC^{Ru}P)Pd}₂(μ²:η^{1,2}-BD₄) [B{OCH(CF₃)₂}₄] (2-d₄)

Complex **2-d₄** was prepared by the same procedure using **1-d₄** instead **1**.

FTIR (KBr, v in cm⁻¹): 1817 (B–D_{term^{as}}), 1592 (B–D_{br}), 1475, 1468, 1448, 1412, 1395, 1381, 1372, 1324, 1292, 1262, 1100, 1053, 1020, 997.

NMR (298 K, C₆D₆, δ in ppm): ³¹P{¹H}: δ_{centr} 90.6 (J_{PP} 330 Hz, δ_A 94.0, δ_B 87.2).

Reaction of 1 with an excess pyridine

1 (6 mg, 0.0089 mmol) was weighed into standard NMR tube and dissolved in toluene-d₈ (0.5 mL), then pyridine (15 μL, ca. 25 molar equivalents) was added by syringe. The formation of (^tBuPC^{Ru}P)PdH (**3**) was monitored by ¹H NMR spectroscopy following the signal of C₅H₂ protons at 4.76 ppm. The rate constant was obtained from the difference between the integral area of C₅H₂ signals of **1** at 4.62 ppm and **3** at 4.76 ppm.

Preparation of (^tBuPC^{Ru}P)PdH (3)

To a solution of (^tBuPC^{Ru}P)PdCl (60 mg, 0.087 mmol) in 40 mL of absolute Et₂O at 0 °C was added LiAlH₄ (27 mg (0.71 mmol)). The reaction mixture was stirred at this temperature for 1 h, then degassed distilled H₂O (0.07 ml, 3.89 mmol) was added and stirring was continued for 30 min. The supernatant solution was then decanted and evaporated in vacuum. The residue was extracted with pentane, the solution was filtered, concentrated in vacuum to a minimum volume and cooled to -15° C. The precipitated white crystals were separated and dried in vacuum. Yield: 52 mg (73%). According to the NMR data, the product has a purity of 80–85%.

NMR (298 K, C₆D₆, δ in ppm): ¹H: -2.92 (t, J_{HP} = 14.1 Hz, 1H, PdH), 1.20, (vt, J_{HP} = 13.0 Hz 18H, CH₃), 1.31 (vt, J_{HP} = 13.8 Hz, 18H, CH₃), 2.69 (dvt, J_{HH} = 16.5 Hz, J_{HP} = 8.2 Hz, 2H, CH_AH_BP), 2.89 (dvt, J_{HH} = 16.5 Hz, J_{HP} = 5.6 Hz, 2H, CH_AH_BP), 4.44 (s, 5H, C₅H₅), 4.86 (s, 2H, C₅H₂). ³¹P{¹H}: 107.27 (s).

Reaction of 3 with CO₂. Formation of (tBuPCRuP)Pd(HCO₂) (4)

3 (7 mg, 0.01 mmol) was weighed into medium-walled NMR tube fitted with a resealable Teflon valve and dissolved in benzene-d₆ (0.5 mL). The tube was pressurized with 4 bar CO₂. The reaction was complete in less than 10 min, giving solution of **4**. The crystals suitable for XRD analysis were grown by slow solvent evaporation of benzene solution used for NMR measurements.

Anal. Calc. (%) for C₂₉H₄₈P₂RuPd (%): C 49.89, H 6.93. **Found:** C, 50.61; H, 7.90.

FTIR (KBr, ν in cm⁻¹):

3097, 2959, 2943, 2924, 2899, 2867, 2761, 2669, 1620 (C=O), 1473 (C–O), 1393, 1370, 1316, 1263, 1181, 1100.

NMR (298 K, C₆D₆, δ in ppm): ¹H: 1.23, (vt, J_{HP} = 12.72 Hz, 18H, CH₃), 1.36 (vt, J_{HP} = 13.99 Hz, 18H, CH₃), 2.27 (dvt, J_{HH} = 16.69 Hz, J_{HP} = 8.42 Hz, 2H, CH_AH_BP), 2.53 (dvt, J_{HH} = 16.64 Hz, J_{HP} = 6.2 Hz, 2H, CH_AH_BP), 4.51 (s, 5H, C₅H₅), 4.65 (s, 2H, C₅H₂), 9.17 (s, 1H, HCO). ¹³C{¹H}: 167.82 (s, COO⁻), 97.42 (vt, 2C, J_{CP} = 28.61 Hz, 2C, 2,5-C₅H₂), 72.88 (s, C₅H₅), 68.70 (vt, J_{CP} = 17.61 Hz 2C, 3,4-C₅H₂), 35.86 (vt, J_{CP} = 11 Hz, 1C CH₂P), 34.68 (vt, J_{CP} = 13.2 Hz, 1C CH₂P), 29.82 (vt, J_{CP} = 5.87 Hz, 9C, C(CH₃)₃), 29.59 (vt, J_{CP} = 8.07Hz, 9C, C(CH₃)₃), 25.26 (vt, J_{CP} = 21.3 Hz, 2C, C(CH₃)₃). ³¹P{¹H}: 83.5 (s).

Characterization of (tBuPCRuP)Pd(HCO₃) (5)

Slow evaporation of benzene solution of (tBuPCRuP)PdH (**3**) under argon flow resulted in crystals which turned out being Pd(II) hydrocarbonate (**5**). The amount of this crystalline phase was enough for the spectroscopic (FTIR, NMR) characterisation and X-ray diffraction analysis (see the main text). it appears to be the most stable crystal phase in the mixture of hydride **3**, formate **4** and carbonate **5**, due to strong hydrogen bonding.

FTIR (KBr, ν in cm⁻¹): 3094, 2958, 2898, 2866, 2626, 2278, 1845, 1606 (C=O), 1466 (C–O), 1412 (C–O), 1392, 1371, 1355, 1264, 1180, 1162, 1099.

NMR (298 K, C₆D₆, δ in ppm, J in Hz): ¹H: 1.25, (vt, J_{HP} = 12.8 Hz, 18H, CH₃), 1.45 (vt, J_{HP} = 14.2 Hz, 18H, CH₃), 2.26 (dvt, J_{HH} = 16.6 Hz, J_{HP} = 8.5 Hz, 2H, CH_AH_BP), 2.51 (dvt, J_{HH} = 16.6 Hz, J_{HP} = 5.0 Hz, 2H, CH_AH_BP), 4.48 (s, 5H, C₅H₅), 4.62 (s, 2H, C₅H₂). ¹³C{¹H}: 162.43 (s, CO₂OH), 109.89 (bt, 1C, 1-C₅H₂), 97.52 (vt, J_{CP} = 29 Hz, 2C, 2,5-C₅H₂), 72.91 (s, C₅H₅), 67.63 (vt, J_{CP} = 17.3 Hz 2C, 3,4-C₅H₂), 35.49 (vt, J_{CP}

= 10.9 Hz, 1C, CH₂P), 34.34 (vt, J_{CP} = 13.17 Hz, 1C, CH₂P), 29.55(vt, J_{CP} = 5.9 Hz, 9C, C(CH₃)₃), 29.36 (vt, J_{CP} = 7.27 Hz, 9C, C(CH₃)₃), 25.26 (vt, J_{CP} = 20.89 Hz, 2C, C(CH₃)₃). ³¹P{¹H}: 82.7 (s).

Computational details

The computations were carried out on “real” complexes without any ligand simplification. Full geometry optimizations were performed without symmetry constraints using the M06³, B3LYP⁴⁻⁶, BP86⁷ and PBE0⁸ methods implemented in Gaussian09 (Revision D.01)⁹ software package. Effective core potentials (ECPs) and their associated SDD basis set supported with *f*-function polarization were used to represent the innermost electrons of the ruthenium and palladium atoms.¹⁰ The basis sets used were 6-311++G(d,p) for the PCP ligand (with the exception of ^tBu groups at phosphorus atoms¹¹), BH₄⁻ fragment, palladium bound hydrogen and 6-31G for ^tBu groups.^{12, 13} The M06-optimized geometries quite well reproduced for all X-ray determined structures (**1**, **2**, **4**, **5**): the difference between the calculated and experimentally observed Pd–C, Pd–P and Pd…B bonds are less than 0.05 Å (Tables S5, S29, S37 and S41).

Frequency calculations were performed for all optimized complexes at the same level of theory to confirm a character of local minima on the potential energy surface and are reported without the use of scaling factors. The nature of all the stationary points on the potential energy surfaces was confirmed by an absence of any imaginary frequencies in vibrational analysis.¹⁴ Transition state (TS) structures showed only one negative eigenvalue in their diagonalized force constant matrices, and their associated eigenvectors were confirmed to correspond to the motion along the reaction coordinate under consideration using the Intrinsic Reaction Coordinate (IRC) method.¹⁵ ZPVE correction was determined from the unscaled harmonic frequencies.^{16, 17}

Inclusion of nonspecific solvent effects in the calculations was performed by using the SMD method.¹⁸ The geometries of all species were optimized in toluene (ϵ = 2.4) or tetrahydrofuran (ϵ = 7.4) by SMD method. The calculations were carried out with ultrafine integration grid and very tight SCF option in order to improve the accuracy of the optimization procedure.

The Wiberg bond indices¹⁹ (WBI) as measure of electron distribution between two atoms²⁰ and natural atomic charges were calculated using the natural-bond orbital (NBO) analysis²¹ implemented in *Gaussian09*.⁹ Topological analysis of the electron-density distribution function $\rho(r)$ was performed using the AIMALL²² program package based on the wave function obtained by the M06 calculations. The delocalization index (DI)²³⁻²⁵ is a measure of electrons that are shared or

exchanged between two atoms or basins, that obtained from the integration of the Fermi hole density, is directly related to the bond order.^{26, 27} The X–Y bond ellipticity, ε_{X-Y} , was defined as $\varepsilon = (\lambda_1/\lambda_2 - 1)$, where λ_1 and λ_2 are the negative eigenvalues of the Hessian of the electron density at the bond critical point ordered such that $\lambda_1 < \lambda_2 < 0$.²⁸⁻³⁰

DHB Strength. The OH band shifts values, $\Delta\nu_{OH}$ ($\Delta\nu_{OH} = \nu_{OH}^{bond} - \nu_{OH}^{free}$), can be used to estimate the hydrogen bond formation enthalpy (ΔH°_{exp}). Introduced by Logansen for classical hydrogen bonds of organic acids and bases³¹⁻³³ the $\Delta H^\circ_{exp}/\Delta\nu_{OH}$ correlation (Eqn. S1) proved to be applicable to dihydrogen bonds.³⁴⁻³⁶

$$\Delta H^\circ = -\frac{75 \cdot |\Delta\nu_{OH}|}{720 + |\Delta\nu_{OH}|} \quad (S1)$$

The basicity factors (E_j), characterizing the proton accepting ability of the base participating in a hydrogen bond, can be calculated by the so-called “rule of factors” equation (Eqn. S2)³¹⁻³³ and are used to compare the properties of bases independent of a partner acid and solvent.

$$E_j = \frac{\Delta H^\circ}{\Delta H_{11}^\circ \cdot P_i} \quad (S2)$$

where ΔH_{11} is the enthalpy of the “standard” complex between phenol ($P_1 = 1.0$) and Et₂O ($E_1 = 1.0$) which is equal to -16.7 kJ/mol in toluene.³¹

DFT calculations of DHB complexes

According to DFT calculation **syn-1** in toluene interacts with proton donors (MeOH, TFE, HFIP) with formation monodentate (**Ia**) complexes on BH_{term} ligand and bifurcate DHB complexes (**IIa** and **IIab**) with coordination on two BH_{term} ligands or BH_{term} and BH_{br} ligands (Figures S16–S18, Table S9). Complexes **Ia** type characterized by short B–H_{term}···O–H distances (1.696–1.789 Å with O–H···H(B) angles 165–172°). Bifurcate complexes characterized by short B–H_{term}···O–H distances for primary interaction (1.719–1.939 Å with O–H···H(B) angles 145–173°) and longer B–H···O–H distances for the secondary contacts (1.954–2.258 Å with O–H···H(B) angles 147–123°). DHB formation causes typical elongation of O–H bond of proton donors (by 0.010–0.015 Å) and B–H bonds (0.003–0.008 Å) involved in DHB formation. It should be noted that in complexes **Ia** and **IIa** DHB formation causes elongation of Pd–H(B) (by 0.009–0.019 Å) and contraction of B–H(Pd) (by 0.004–0.019 Å).

Generally, the re-polarization of hydrogen atoms involved in DHB formation was anticipated.³⁷ During the DHB formation the increase of positive charge on the OH proton

($\Delta q^{\text{NBO}}_{[(\text{O})\text{H}]}$ is $0.032 \div 0.047$) of proton donors (MeOH, TFE, HFIP) and increase of negative charge for the BH_{term} ligand ($\Delta q^{\text{NBO}}_{[(\text{B})\text{Hterm}]}$ is $-0.002 \div -0.022$) were observed (Table S10). It should be noted that for bifurcate DHB complexes of **IIab_anti** type (with coordination on BH_{term} and BH_{br} ligand) for some complexes the significant increase of negative charge was observed on BH_{br} ligand ($\Delta q^{\text{NBO}}_{[(\text{B})\text{Hbr}]}$ increased up to -0.081). The changes of Wiberg bond indices¹⁹ (ΔWBI , Table S10) also reflect the depopulation of OH ($\Delta \text{WIB}_{\text{OH}}$ is $-0.097 \div -0.147$) and BH bonds ($\Delta \text{WIB}_{\text{BHterm}}$ is $-0.011 \div -0.043$) with simultaneous appearance of the electron density for the H···H contacts ($\Delta \text{WIB}_{\text{H}\cdots\text{H}}$ are $0.009 \div 0.030$ for primary contact $\text{B}-\text{H}_{\text{term}}\cdots\text{O}-\text{H}$ and up to 0.007 for secondary $\text{B}-\text{H}_{\text{term}}\cdots\text{O}-\text{H}$ or $\text{B}-\text{H}_{\text{br}}\cdots\text{O}-\text{H}$ interactions). It should be noted that in all DHB complexes the depopulation of Pd–H(B) was observed ($\Delta \text{WIB}_{\text{PdHbr}}$ is $-0.007 \div -0.041$). This fact indicates the weakening of the interaction between palladium atom and the BH_4^- fragment as the result of DHB formation. Previously the same phenomenon was observed in the case of copper(I) tetrahydroborates.^{38, 39}

The DHB formation between **1** and ROH also confirmed by energy donation (E^2 , estimated from second-order perturbative analysis of donor-acceptor interactions, Table S10) from $\sigma(\text{B}-\text{H})$ to $\sigma^*(\text{O}-\text{H})$ ($E^2 = 10.0\text{--}31.0$ kJ/mol for the primary contact $\text{B}-\text{H}_{\text{term}}\cdots\text{O}-\text{H}$ and $E^2 = 3.3\text{--}7.9$ kJ/mol for the B–H_{term}···O–H or B–H_{br}···O–H interactions). According to QTAIM analysis the (+3;−1) critical point ($p_c = 0.02$ a.u.) was found only for the $\text{B}-\text{H}_{\text{term}}\cdots\text{O}-\text{H}$ primary contact (Figures S19–S21, Table S11). The presence of B–H···O–H secondary interactions causes deviation of O–H···H(B) angles from linearity and reflected in the values of H···H bond ellipticity ($\epsilon_{\text{H}\cdots\text{H}} = 0.17\text{--}0.26$ for **Ia** and $\epsilon_{\text{H}\cdots\text{H}} = 0.37\text{--}0.78$ for **IIab** DHB complexes). Extremely high $\epsilon_{\text{H}\cdots\text{H}}$ ellipticity values observed for **IIa** DHB complexes ($\epsilon_{\text{H}\cdots\text{H}} = 1.17\text{--}8.31$) correlate with the location the minima of molecular electrostatic potential (MEP) in **1** (Figure 3), which is located almost equidistantly from BH_{term} ligands.

Description of crystal structure of complex **2**

In complex **2** two ruthenocene palladacycle fragments (**A**) and (**B**) are rotated by ca. 90° relative to each other (Figure S22). The structural parameters of ruthenocene palladacycle fragments are different: one (**A**) is similar to initial borohydride complex **1**, the structure of the second (**B**) is significantly distorted. In both fragments the palladium atoms have a distorted square-planar geometry with “gull wing” pincer arms conformation, with angles P(1)–Pd–P(2) 156.66(4) (**A**) and P(1A)–Pd–P(2A) 154.24(4) Å (**B**), being smaller compare to 160.58(2) in **1**.

The carbon atoms of methylene bridges are situated on the Cp-ring plane (**A**) C(15) 0.014 Å, C(6) 0.011 Å, (**B**) (C(15A) 0.042 Å, C(6A) 0.000 Å), for **1** (C(11) 0.009 Å C(12) 0.038 Å), the phosphorus atoms in fragment (**A**) rise above the plane on P(1) 0.251 Å and P (2) 0.483 Å, in fragment (**B**) on P(1A) 0.268 Å and P(2A) 0.178 Å, whereas these distances are 0.367 Å, 0.379 Å in **1**. Palladium atoms are deviated from the plane of the Cp-ring in the direction opposite to the ruthenium atom by 0.150 Å (**A**) and 0.132 Å (**B**). This deviation is much smaller compared to initial borohydride complex **1** (0.322 Å). At the same time the Cp-rings appear differently inclined in complexes **2** [4.05° (**A**), and 1.42° (**B**)], and **1** (2.42°), while the distance between the Cp rings centroids of **2** (3.623 Å (**A**) and 3.630 Å (**B**)) is almost the same as in **1** (3.636 Å).

This disposition of two ruthenocene palladacycle fragments (**A**) and (**B**) in **2** rotated by ca. 90° relative to each other (Figure S22) leads to non-equivalence of phosphorus nuclei in palladacycle fragments, which is manifested as AB spin system in phosphorus NMR.

Table S1. Crystal data and structure refinement parameters for **1**, **2**, **4** and **5**.

	1	2	4	5
Empirical formula	C ₂₈ H ₅₁ BP ₂ PdRu	C ₆₈ H ₁₀₂ B ₂ F ₂₄ O ₄ P ₄ Pd ₂ Ru ₂	C ₃₂ H ₅₁ O ₂ P ₂ PdRu	C ₄₁ H ₆₀ O ₃ P ₂ PdRu
Formula weight	667.90	1999.93	737.13	870.30
Crystal system	Monoclinic	Triclinic	Triclinic	Monoclinic
Space group	P2 ₁ /c	P-1	P-1	P2 ₁ /n
Z	4	2	2	4
a, Å	10.9154(4)	14.6724(11)	9.8920(5)	14.0995(9)
b, Å	18.1161(6)	15.4144(11)	12.0215(6)	19.7160(13)
c, Å	15.8325(6)	20.2741(15)	15.1339(7)	15.7422(10)
α, °	90	99.8970(10)	109.1100(10)	90
β, °	106.5607(7)	99.7330(10)	105.3890(10)	110.4874(12)
γ, °	90	99.8720(10)	96.9360(10)	90
V, Å ³	3000.92(19)	4356.1(6)	1596.37(14)	4099.3(5)
D _{calc} (g cm ⁻¹)	1.478	1.525	1.534	1.410
Linear absorption, μ (cm ⁻¹)	12.24	9.12	11.63	9.2
F(000)	1376	2016	758	1800
2θ _{max} , °	60	54	60	60
Reflections measured	38541	73348	21508	52539
Independent reflections	8753	18998	9295	11939
Observed reflections [<i>I</i> > 2σ(<i>I</i>)]	8016	16889	7335	10242
Parameters	326	1009	355	449
R1	0.0309	0.0465	0.0352	0.0317
wR2	0.0591	0.1634	0.0764	0.0763
GOF	1.195	1.147	1.002	1.056
Δρ _{max} / Δρ _{min} (e Å ⁻³)	0.550/-0.850	3.587/-1.870	0.638/-0.831	2.145/-1.008

Table S2. The distances (in Å) between two ^tBu groups in **syn**- and **anti**-position in complexes **1**, **4** and **5**.

Complex	syn-position		anti-position	
1	H18C···H26B	3.40	H14B···H22C	2.61
			H16C···H24B	3.53
4	H19C···H26A	3.39	H15A···H22C	2.38
			H16C···H24A	3.71
5	H19B···H26C	3.50	H16B···H24C	2.33
			H15C···H22B	3.16

Table S3. Geometry parameters of crystal and DFT-optimized structures of **1**.

Distances	1	M06	BP86	B3LYP
Pd–C	1.980(2)	1.999	1.996	1.997
Pd–P₁	2.3443(7)	2.390	2.388	2.405
Pd–P₂	2.3454(7)	2.397	2.390	2.407
Pd···B	2.587(3)	2.563	2.546	2.662
Pd–H₁(B)	1.87(2)	1.816	1.779	1.771
Pd···H₂(B)	2.54(3)	2.482	2.491	2.696
B–H₁	1.11(2)	1.284	1.301	1.292
B–H₂	1.10(3)	1.221	1.227	1.214
B–H₃	1.06(3)	1.213	1.219	1.211
B–H₄	1.04(4)	1.213	1.219	1.211
C–Pd–B	170.3(1)	166.37	166.69	166.92
P₁–Pd–P₂	160.58(2)	158.27	159.68	159.71
Pd–H₁–B	118(2)	110.32	110.54	119.90
Pd–H₂–B	80(2)	79.68	78.41	75.37

Figure S1. (a) General view of molecular structure of **1** (thermal ellipsoids of all non-hydrogen atoms are drawn at the 50% probability level. (b) M06-optimized geometry of *syn*-**1** configuration. Hydrogen atoms of the ligand are omitted for clarity.

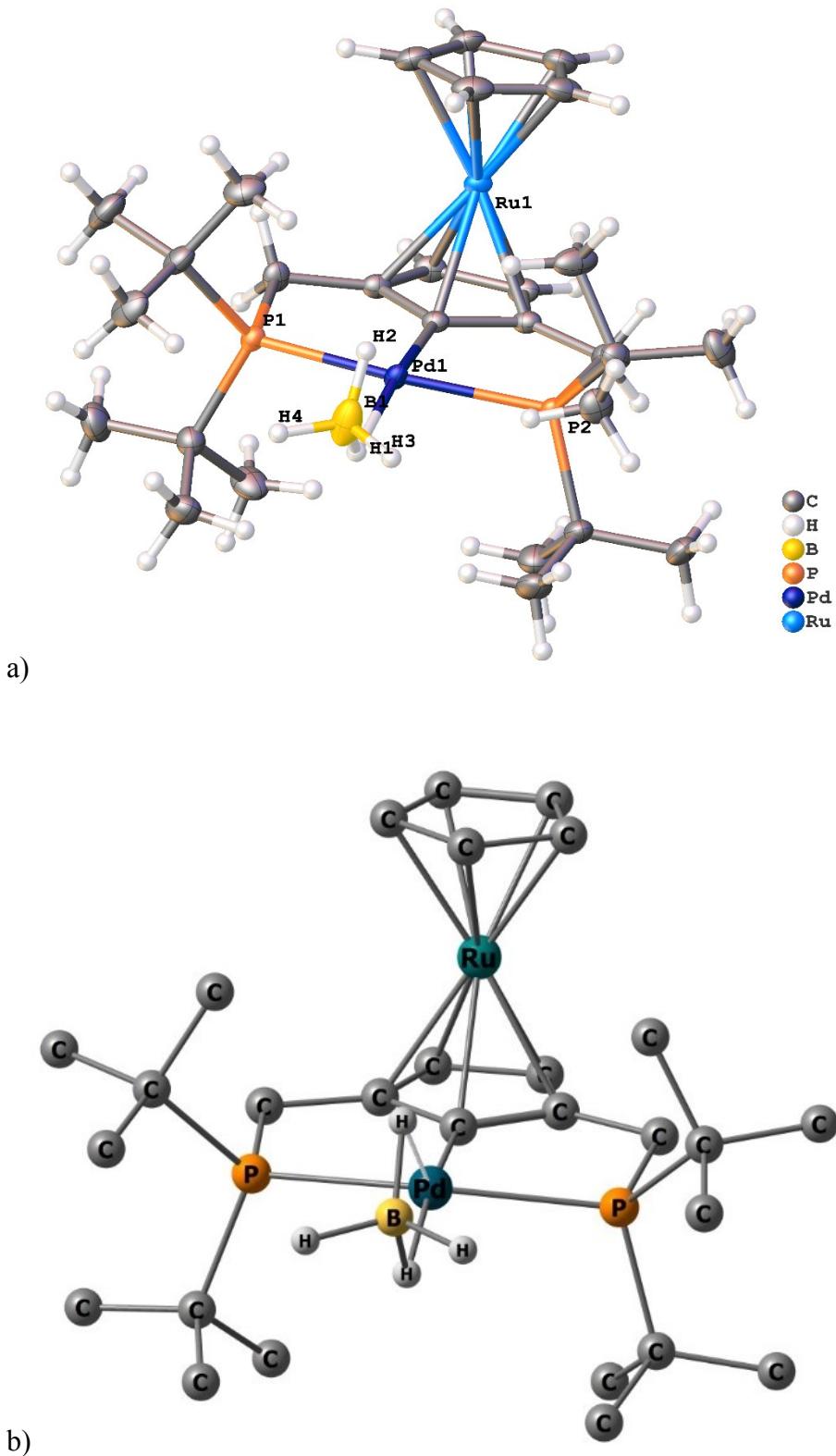


Figure S2. (a) M06-optimized geometry of **anti-1** configuration. (b) QTAIM molecular graph of M06-optimized geometry of **anti-1** configuration. Hydrogen atoms of the ligand are omitted for clarity. Red spheres denote critical points (+3;-1), yellow spheres denote ring critical points (+3;+1) and green spheres denote cage critical points (+3;+3).

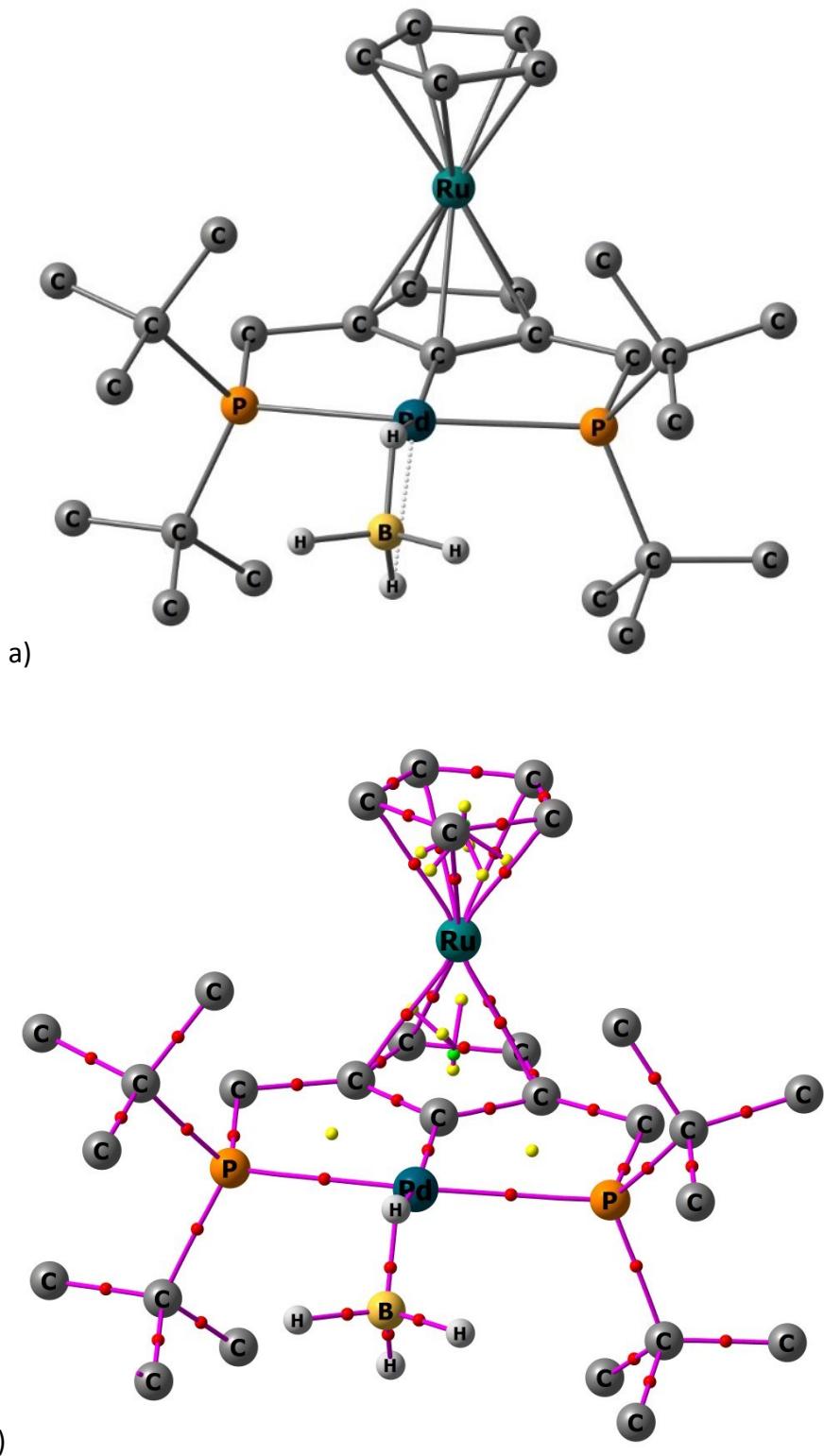


Figure S3. (a) Energy scan of shortening Pd···H₂ coordinate (step size = 0.02 Å). (b) Energy profile of the BH_{br}/BH_{term} exchange for **1**.

a)

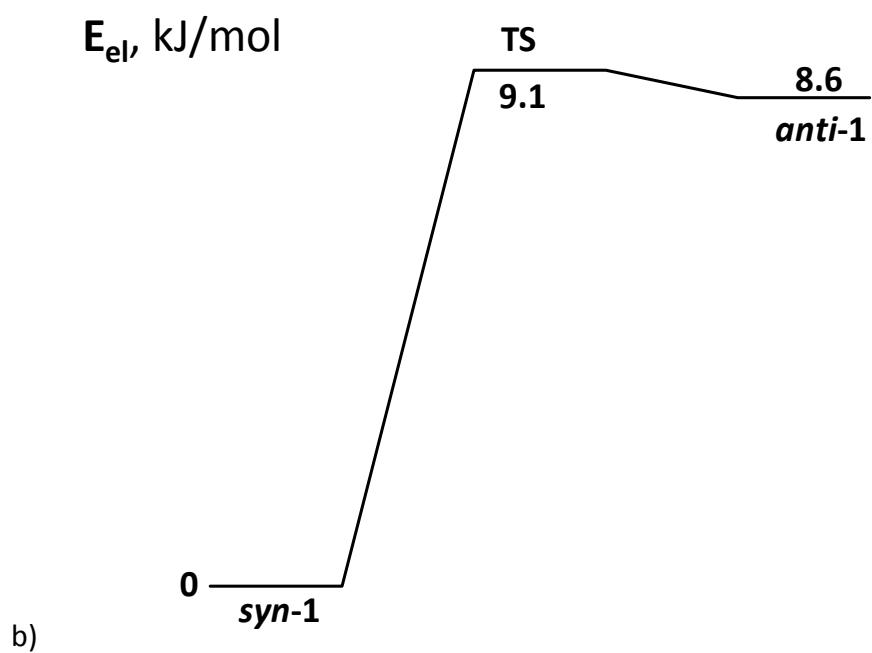
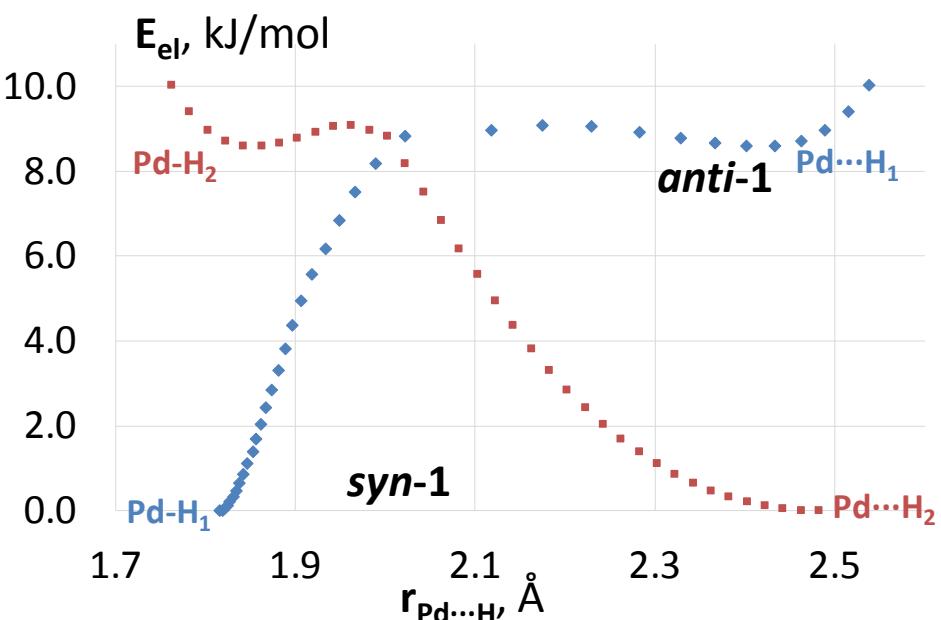


Table S4. The main donor-acceptor orbital interactions and donation energy (E^2 and their sum ΣE^2 in kJ/mol) estimated from second-order perturbative analysis of donor-acceptor interactions for isomeric *syn*-**1** and *anti*-**1** configurations and *syn,syn*-**2**.

Compound	Donor orbital	Acceptor orbital	E^2	ΣE^2
<i>syn</i> - 1	γ_{B-H1}	$d_z^2(Pd)^*$	137.7	
		γ_{Pd-C}^*	57.3	195.0
	γ_{B-H2}	$d_z^2(Pd)^*$	38.9	
		$d_{xz}(Pd)^*$	30.1	71.5
		γ_{Pd-C}^*	2.5	
<i>anti</i> - 1	γ_{B-H2}	$d_z^2(Pd)^*$	124.3	
		γ_{Pd-C}^*	52.3	184.1
		$d_{xz}(Pd)^*$	7.5	
	γ_{B-H1}	$d_z^2(Pd)^*$	70.3	
		$d_{xz}(Pd)^*$	21.3	94.6
		γ_{Pd-C}^*	2.9	
<i>syn,syn</i> - 2	γ_{B-H1}	$d_z^2(Pd)^*$	63.6	
		γ_{Pd-C}^*	43.9	107.5
	γ_{B-H2}	$d_z^2(Pd)^*$	20.9	
		$d_{xz}(Pd)^*$	16.3	41.4
		γ_{Pd-C}^*	4.2	
	γ_{B-H3}	$d_z^2(Pd)^*$	62.8	
		γ_{Pd-C}^*	49.0	111.3
		$d_{xz}(Pd)^*$	15.5	
γ_{B-H4}	γ_{Pd-C}^*	$d_{xz}(Pd)^*$	16.3	36.4
		$d_z^2(Pd)^*$	4.2	

Table S5. QTAIM delocalization indices (DI), sum of donation energy (sum ΣE^2 in kJ/mol) estimated from second-order perturbative analysis of donor-acceptor interactions and Wiberg bond indices (WIB) for **1** isomeric *syn*-**1** and *anti*-**1** configurations and *syn,syn*-**2**.

Compound	DI ₁	DI ₂	DI ₁ /DI ₂	$\Sigma_1 E^2$	$\Sigma_2 E^2$	$\Sigma_1 E^2 / \Sigma_2 E^2$	WIB ₁	WIB ₂	WIB ₁ /WIB ₂
<i>syn</i> - 1	0.45	0.14	3.1	195.0	71.5	2.7	0.36	0.11	3.2
<i>anti</i> - 1	0.43	0.16	2.6	184.1	94.6	1.9	0.31	0.11	2.8
<i>syn,syn</i> - 2	0.38	0.14	2.7	107.5	41.4	2.6	0.20	0.06	3.5
	0.39	0.13	2.9	111.3	36.4	3.1	0.21	0.05	3.8

Figure S4. ^1H NMR spectrum (400 MHz) of **1** in C_6D_6 .

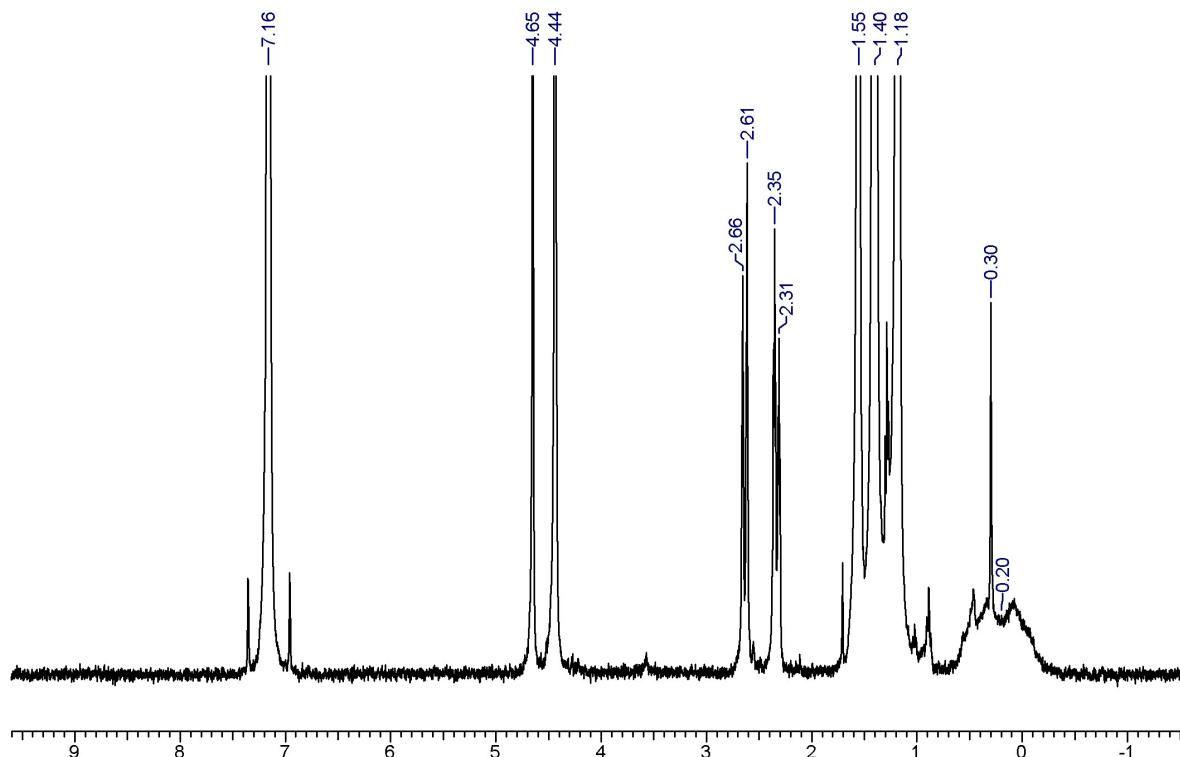


Figure S5. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum (128.4 MHz) of **1** in C_6D_6 .

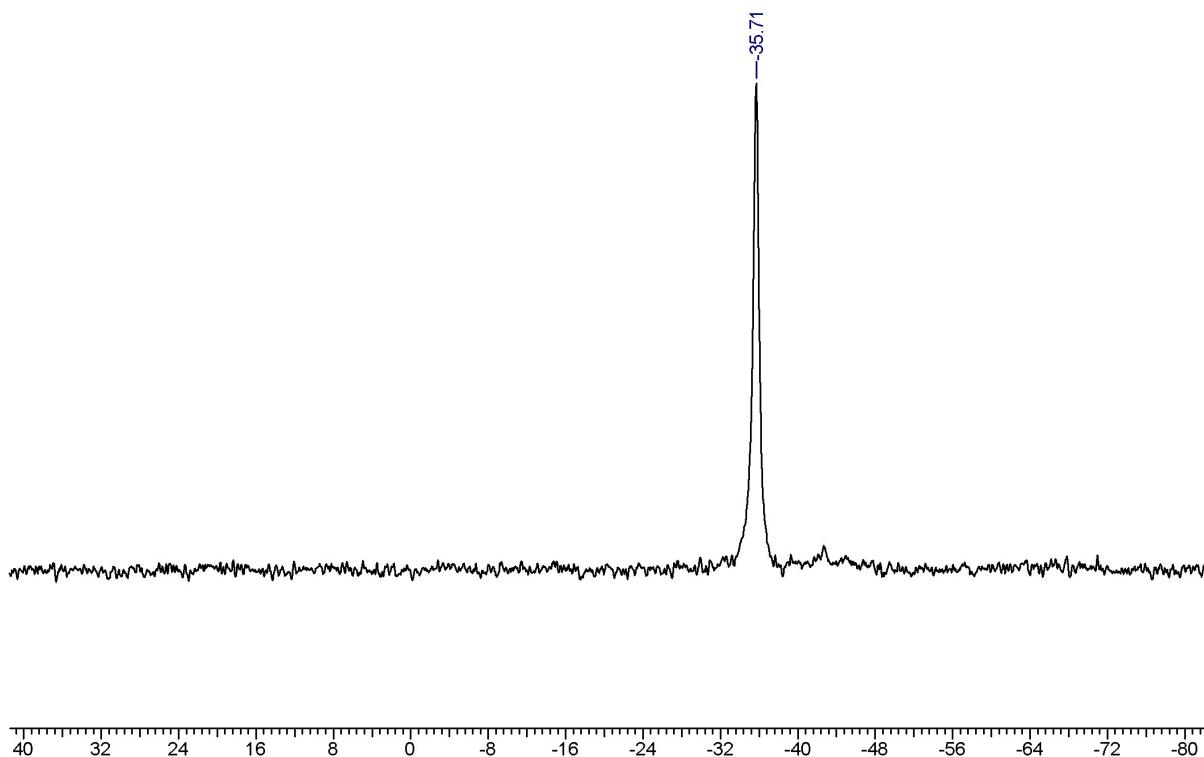


Figure S6. $^{31}\text{P}\{\text{H}\}$ NMR spectrum (162.0 MHz) of **1** in C_6D_6 .

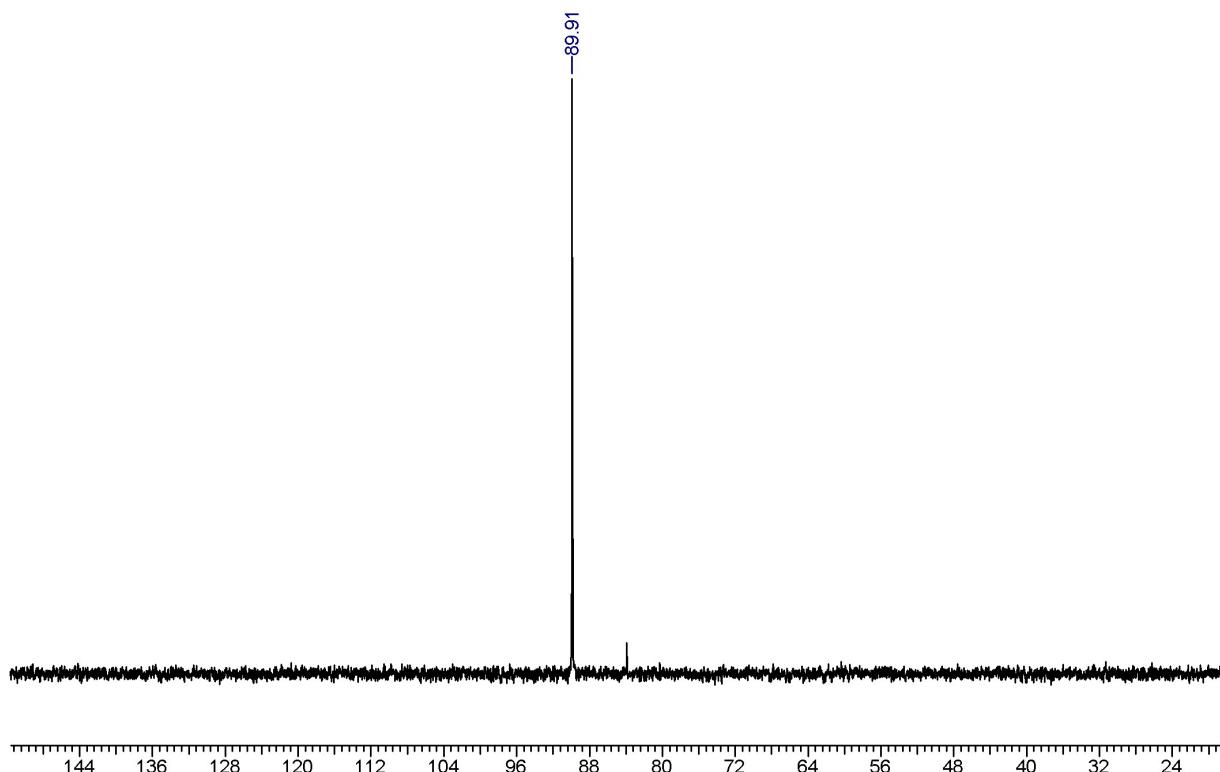


Figure S7. ^1H NMR spectrum (500 MHz) of **1** in toluene- d_8 .

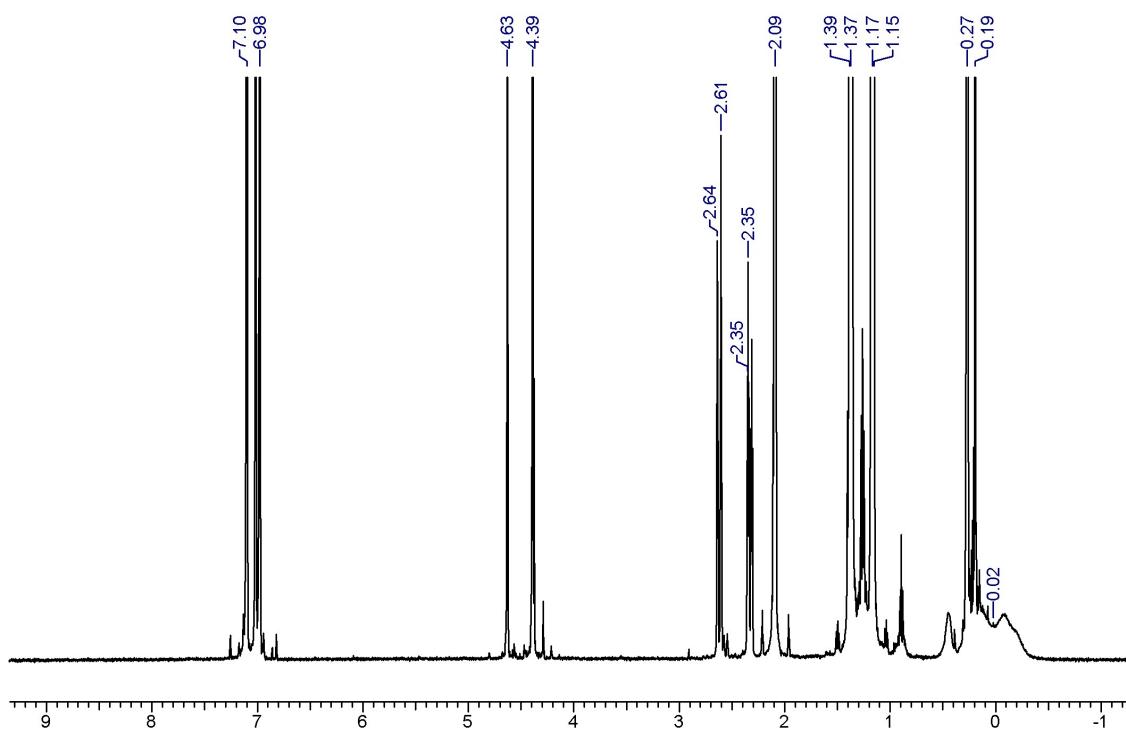


Figure S8. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum (160.5 MHz) of **1** in toluene-d₈.

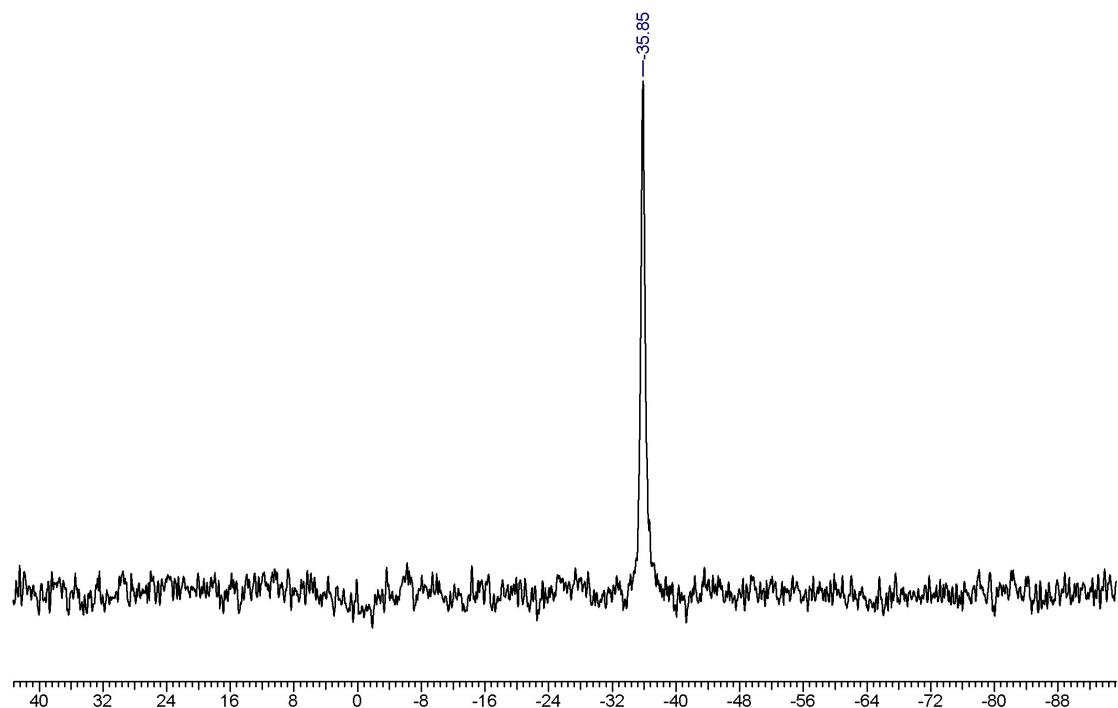


Figure S9. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum (202.5 MHz) of **1** in toluene-d₈.

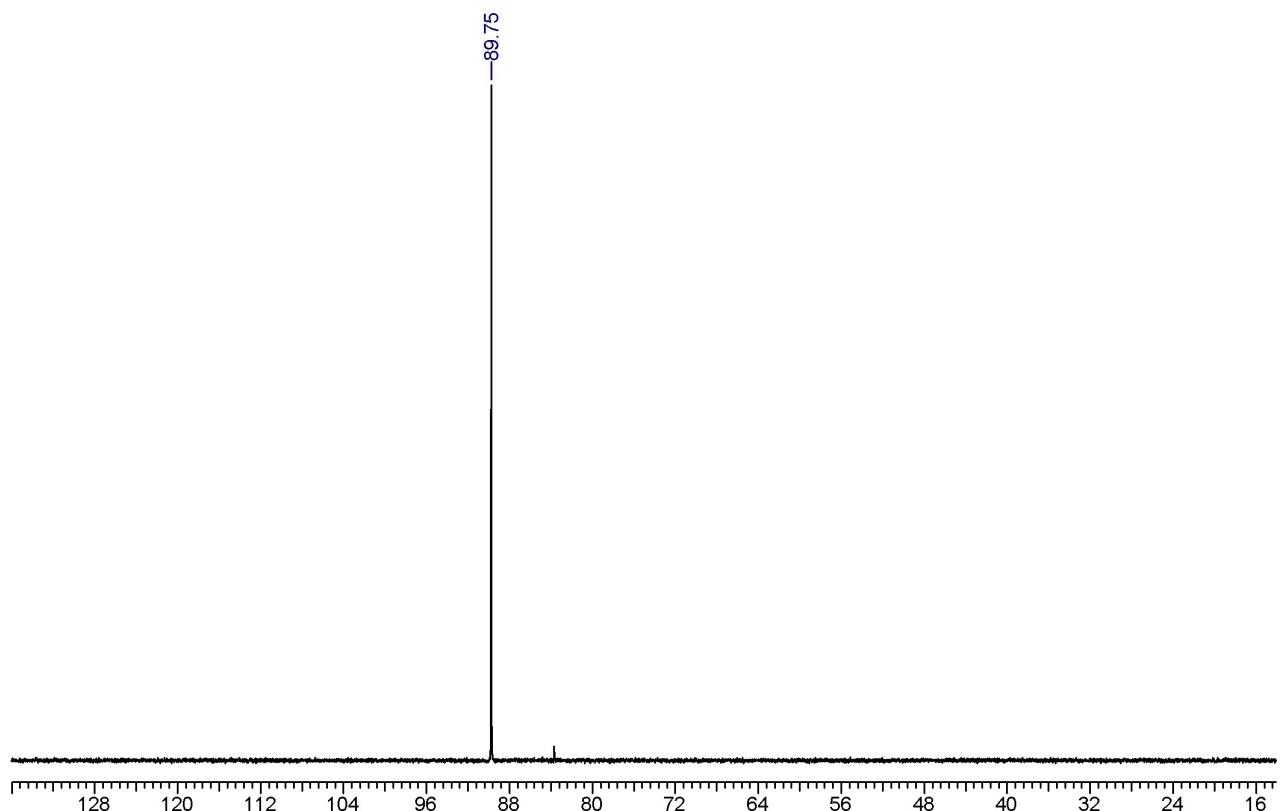


Figure S10. FTIR spectrum of **1** in KBr pellet.

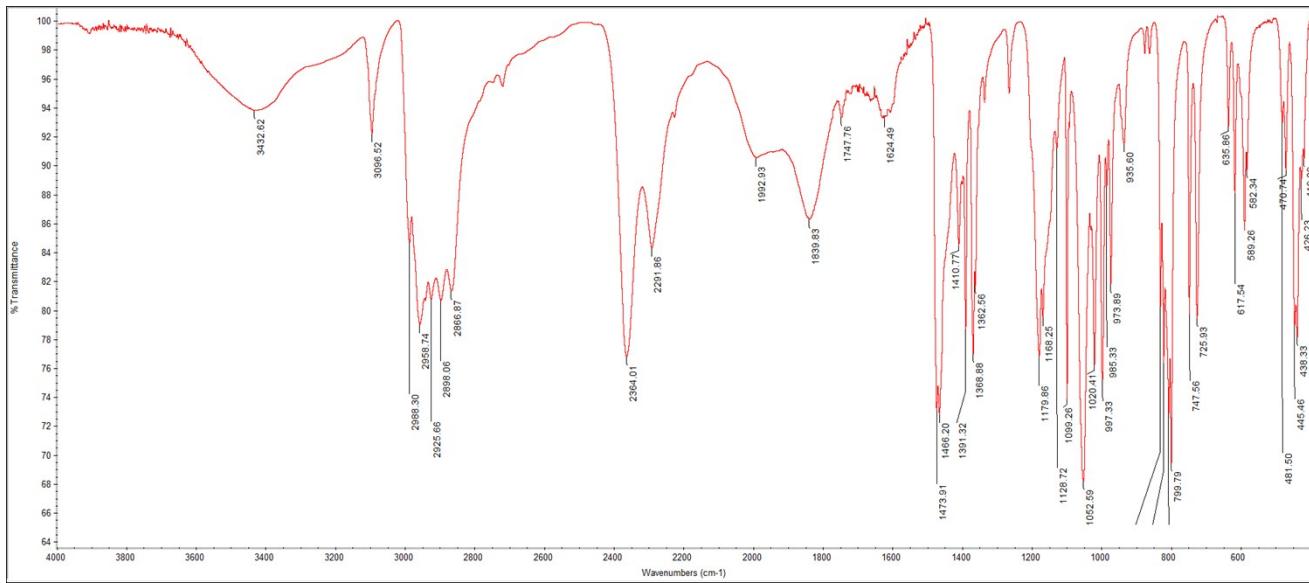


Figure S11. FTIR spectrum of **1-d₄** in KBr pellet.

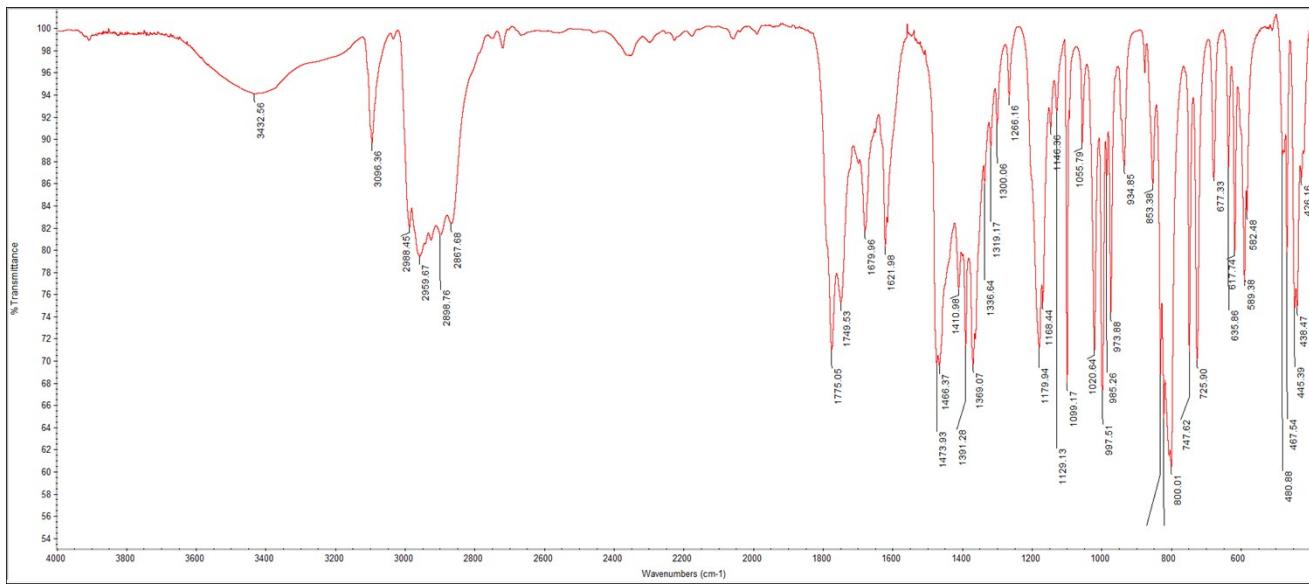


Figure S12. ^1H NMR spectrum (400 MHz) of **1-d₄** in C₆D₆.

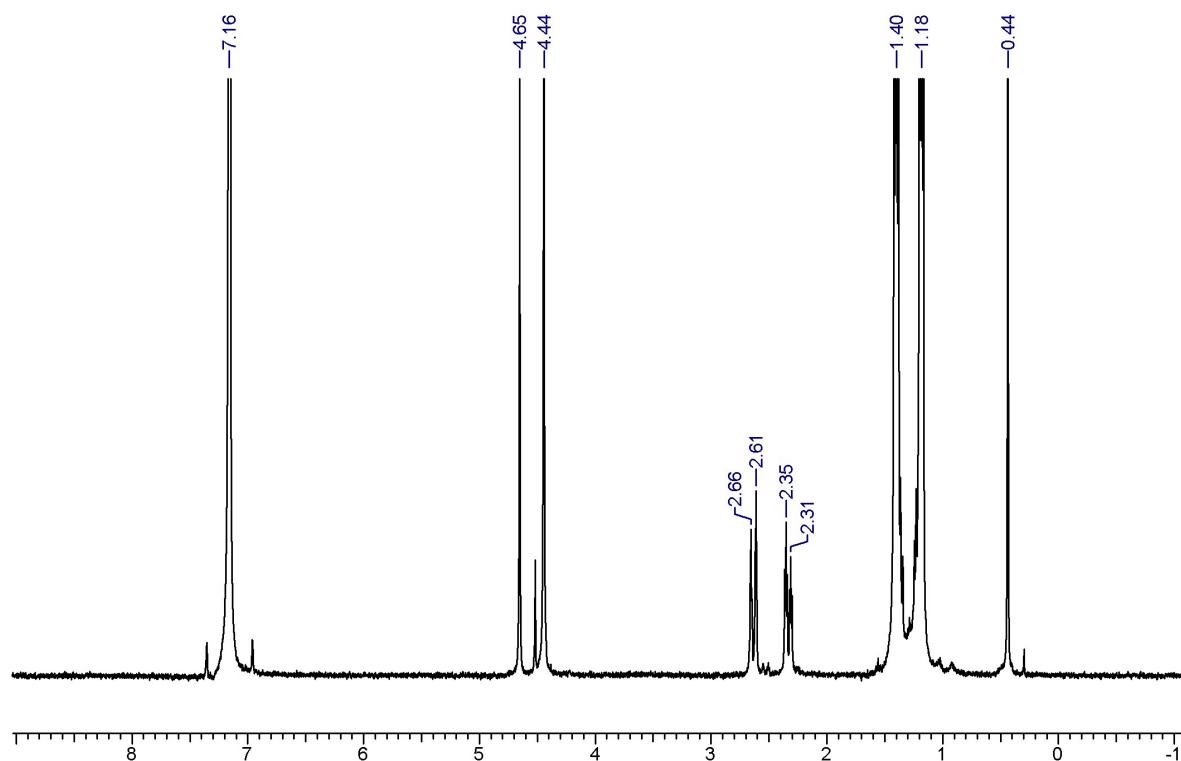


Figure S13. ^2H NMR spectrum (61.4 MHz) of **1-d₄** in C₆D₆.

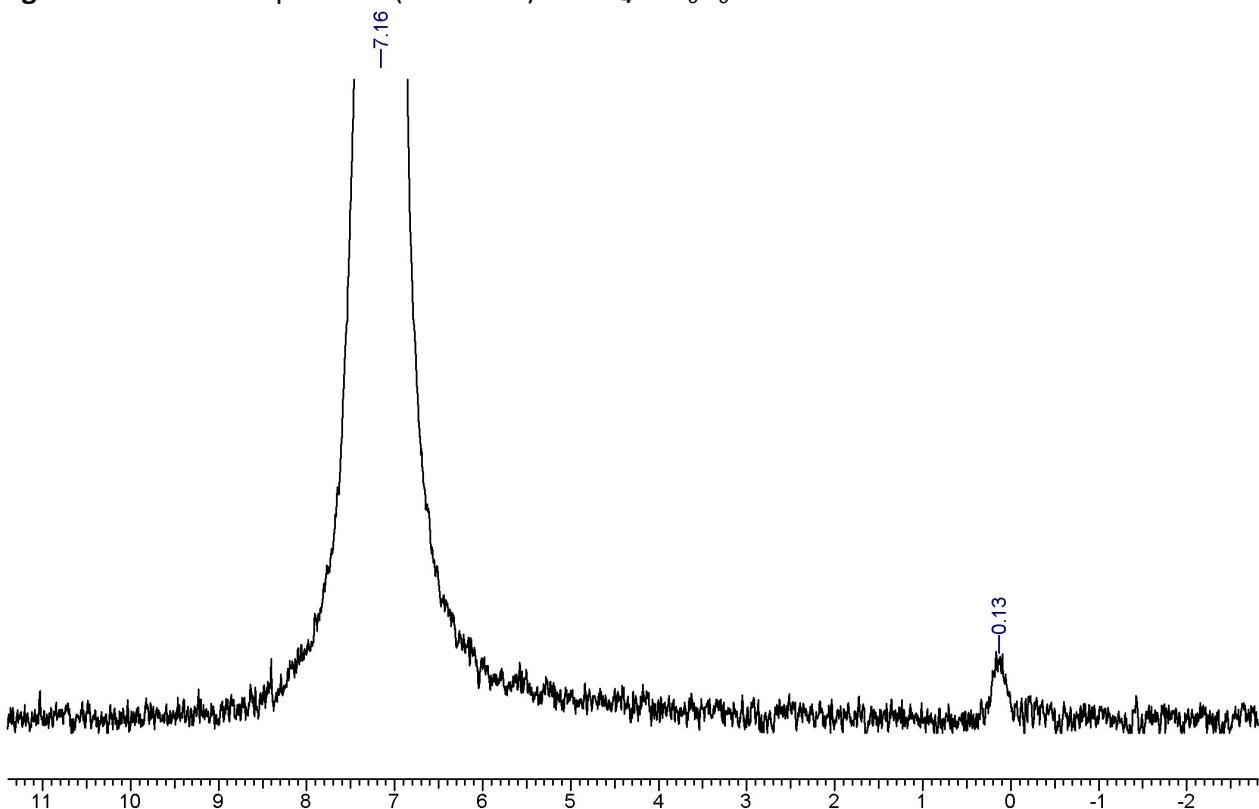


Figure S14. $^{31}\text{P}\{\text{H}\}$ NMR spectrum (162.0 MHz) of **1-d₄** in C₆D₆.

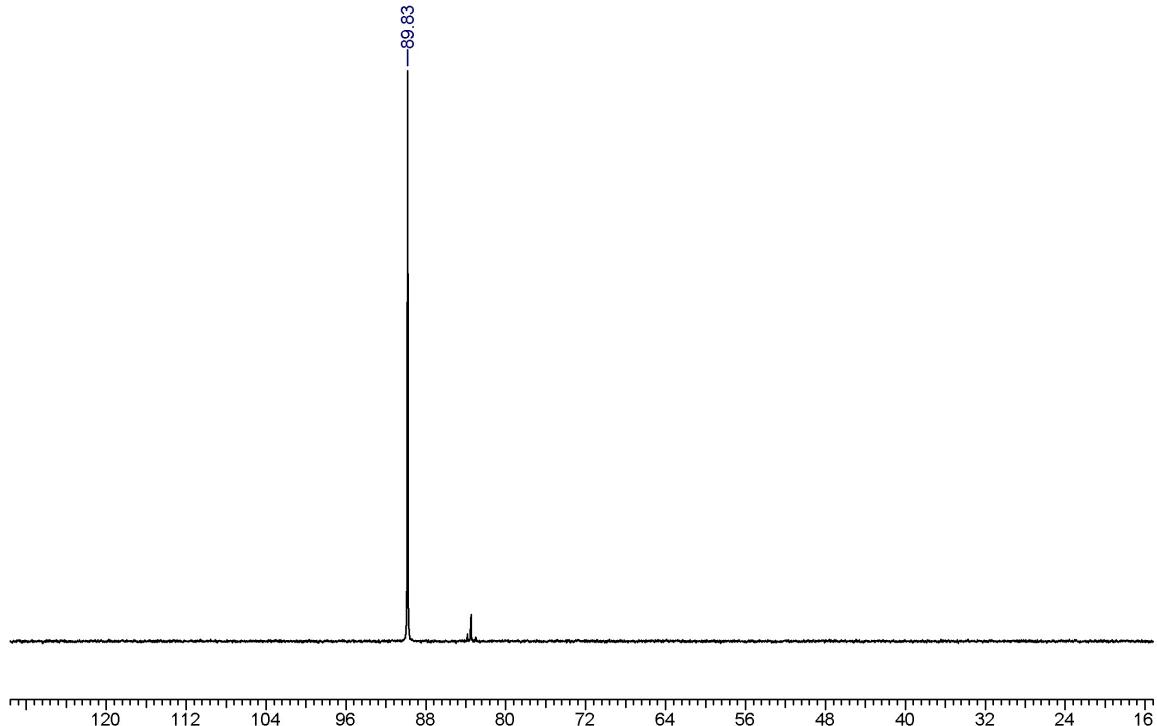


Figure S15. $^{11}\text{B}\{\text{H}\}$ NMR spectrum (128.4 MHz) of **1-d₄** in C₆D₆.

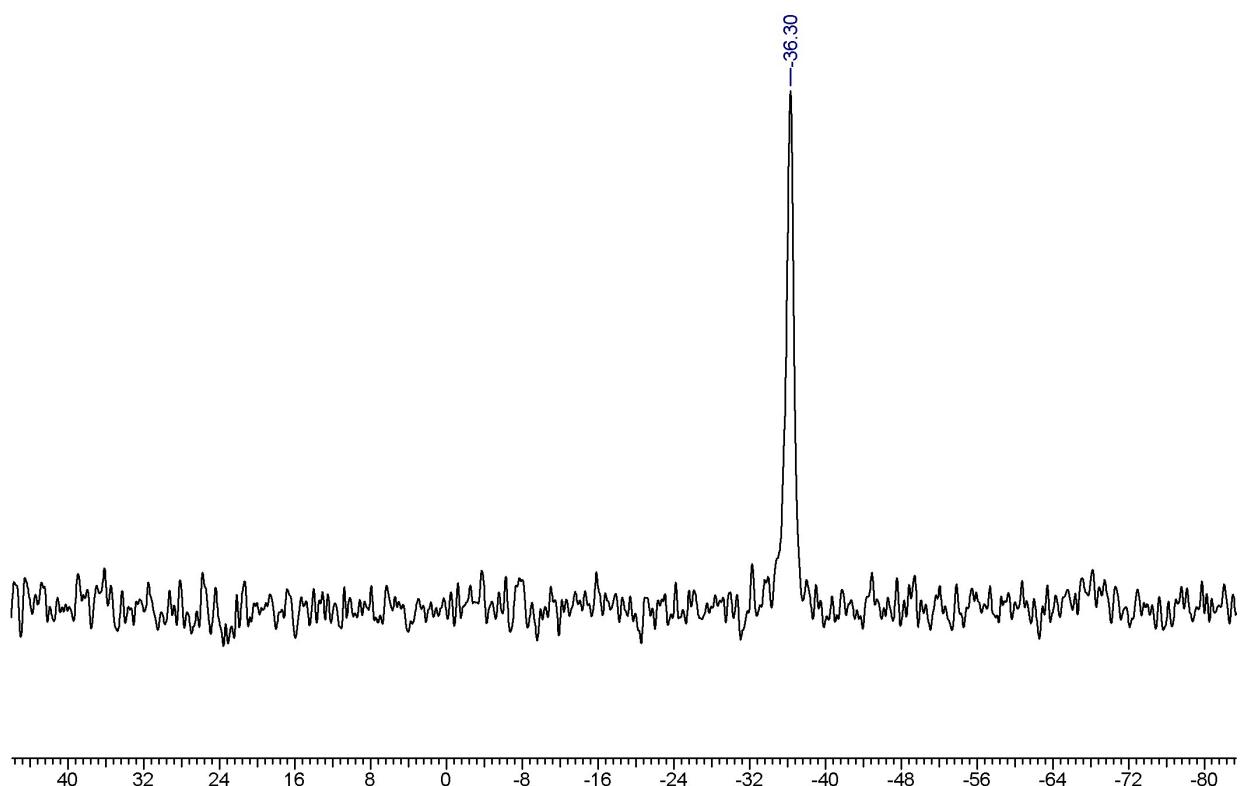


Table S6. CCDC analysis of the structures of Pd(II) tetrahydroborates.

Nº	Compound	$r_{(M\cdots B)}$	$r_{(M-H^A_{br})}$	$r_{(M\cdots H^B_{br})}$	$r_{(B-H^A_{br})}$	$r_{(B-H^B_{br})}$	$\angle_{(M-H^A-B)}$	$\angle_{(M-H^B-B)}$	CSD refcode	CSD №	Ref
1.	Pd(2,6-(<i>i</i> Pr ₂ PO) ₂ C ₆ H ₃)(η ^{1,2} -BH ₄)	2.420(2)	1.78(2)	2.21(2)	1.27(2)	1.14(2)	103(1)	86(1)	FIRCUH	1872776	⁴⁰
2.	Pd(2,6-(<i>t</i> Bu ₂ PO) ₂ C ₆ H ₃)(η ^{1,2} -BH ₄)	2.464(4)	1.91(3)	2.11(3)	1.18(3)	1.25(4)	103(2)	91(2)	FIRCOB	1872775	⁴⁰
3.	Pd[{2,5-(<i>t</i> Bu ₂ PCH ₂) ₂ C ₅ H ₂ }Ru(C ₅ H ₅)](η ^{1,2} -BH ₄)	2.587(3)	1.87(2)	2.54(3)	1.11(2)	1.10(3)	118(2)	80(2)	—	1882669	This work
4.	Pd[{2,5-(<i>t</i> Bu ₂ PCH ₂) ₂ C ₅ H ₂ }Fe(C ₅ H ₅)](η ^{1,2} -BH ₄)	2.613(9)	2.00(4)	2.56(6)	0.91(4)	1.13(5)	123(4)	80(3)	EYUJEN	248899	^{41, 42}

Table S7. ¹¹B and ¹H NMR characteristics of Pd(II) tetrahydroborates.

Nº	Compound	Solvent	¹¹ B, δ _{BH4} (ppm)	¹ H, δ _{BH4} (ppm)	Ref
1.	Pd[{2,5-(<i>t</i> Bu ₂ PCH ₂) ₂ C ₅ H ₂ }Ru(C ₅ H ₅)](η ^{1,2} -BH ₄)	C ₆ D ₆	-35.8	+0.2	This work
		toluene-d ₈	-35.9	0.0	This work
2.	Pd[{2,5-(<i>t</i> Bu ₂ PCH ₂) ₂ C ₅ H ₂ }Fe(C ₅ H ₅)](η ^{1,2} -BH ₄)	CDCl ₃	-37.3 (83.1 Hz)	-0.5 (98 Hz)	^{41, 42}
3.	Pd[{2,5-(<i>i</i> Pr ₂ PCH ₂) ₂ C ₅ H ₂ }Fe(C ₅ H ₅)](η ^{1,2} -BH ₄)	C ₆ D ₆	-37.7 (81.3 Hz)	+0.3 (97/55 Hz)	^{41, 42}
4.	Pd(2,6-(<i>i</i> Pr ₂ PO) ₂ C ₆ H ₃)(η ^{1,2} -BH ₄)	C ₆ D ₆	-39.5 (81.0 Hz)	+0.2 (77 Hz)	⁴⁰
5.	Pd(2,6-(<i>t</i> Bu ₂ PO) ₂ C ₆ H ₃)(η ^{1,2} -BH ₄)	C ₆ D ₆	-39.8 (82.4 Hz)	+0.4 (75 Hz)	⁴⁰

Table S8. Chemical shifts (δ_H in ppm) of selected 1H NMR signals of **1** in the presence of fluorinated alcohols [CH_2FCH_2OH (MFE), CF_3CH_2OH (TFE) and $(CF_3)_2CHOH$ (HFIP)] in toluene-d₈.

	BH_4^-	$-CH_2-$			$-C_5H_2-$	C_5H_5 (Cp)	$tBu-$		
free 1	0.06	2.68	2.64	2.38	2.35	4.67	4.43	1.41	1.20
MFE									
1.4 eqv	0.05	2.67	2.64	2.38	2.35	4.67	4.43	1.41	1.20
3.5 eqv	0.04	2.67	2.64	2.38	2.35	4.66	4.42	1.41	1.20
17.0 eqv	0.00	2.67	2.63	2.38	2.35	4.66	4.42	1.40	1.20
TFE									
1.65 eqv	0.03	2.67	2.64	2.38	2.35	4.66	4.42	1.40	1.20
4.65 eqv	0.01	2.66	2.63	2.38	2.34	4.66	4.42	1.39	1.19
14.0 eqv	-0.05	2.65	2.62	2.36	2.33	4.65	4.41	1.37	1.18
HFIP									
0.7 eqv	0.05	2.67	2.63	2.38	2.33	4.67	4.43	1.40	1.20
1.8 eqv	-0.01	2.65	2.61	2.36	2.32	4.66	4.42	1.38	1.18
7.0 eqv	-0.11	2.62	2.58	2.34	2.29	4.64	4.40	1.33	1.15
13.85 eqv	-0.18	2.60	2.56	2.32	2.28	4.63	4.39	1.32	1.14

Figure S16. M06-optimized geometries of DHB complexes of **1** with MeOH in toluene. Hydrogen atoms of the ligand are omitted for clarity.

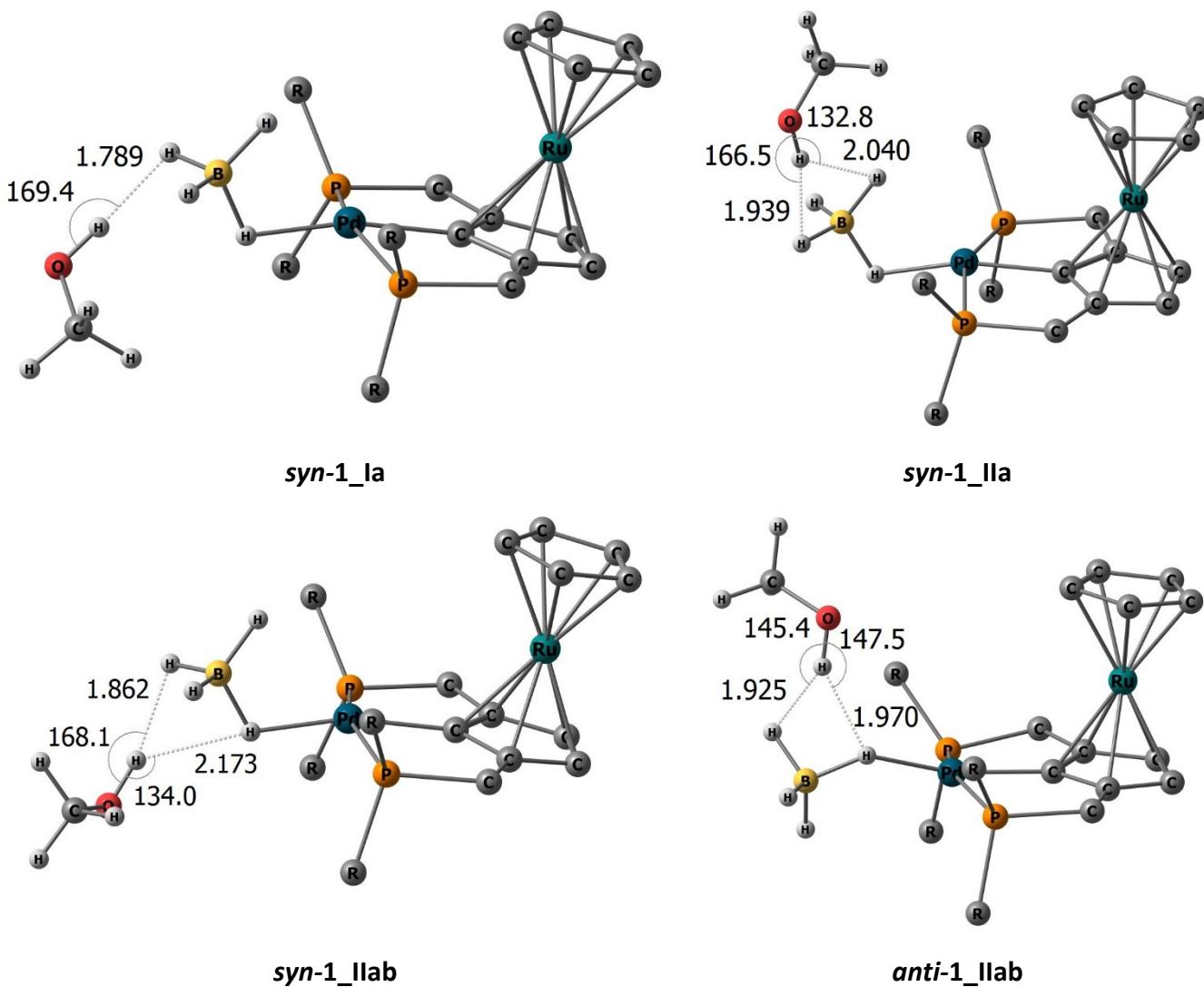


Figure S17. M06-optimized geometries of DHB complexes of **1** with TFE in toluene. Hydrogen atoms of the ligand are omitted for clarity.

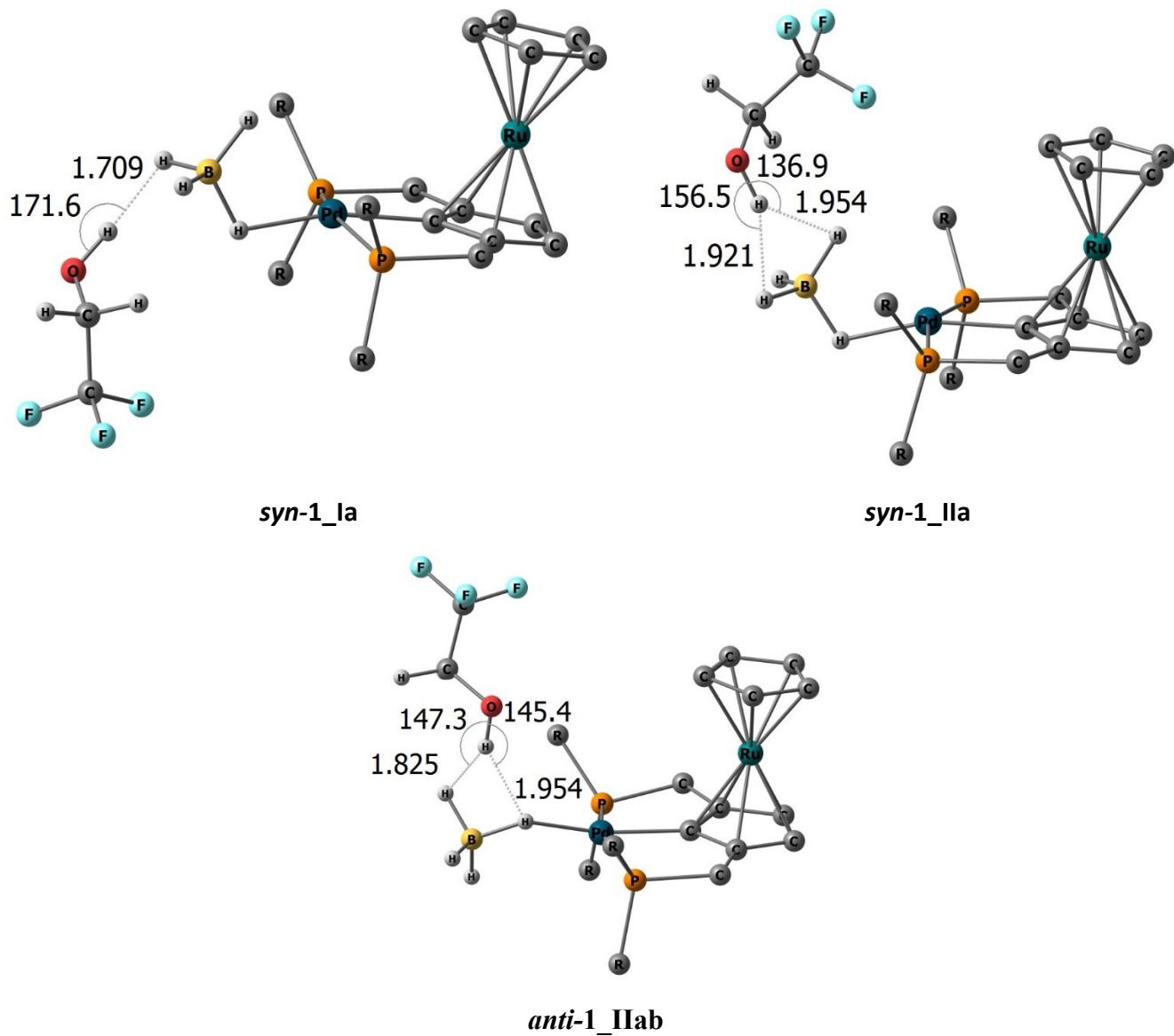


Figure S18. M06-optimized geometries of DHB complexes of **1** with HFIP in toluene. Hydrogen atoms of the ligand are omitted for clarity.

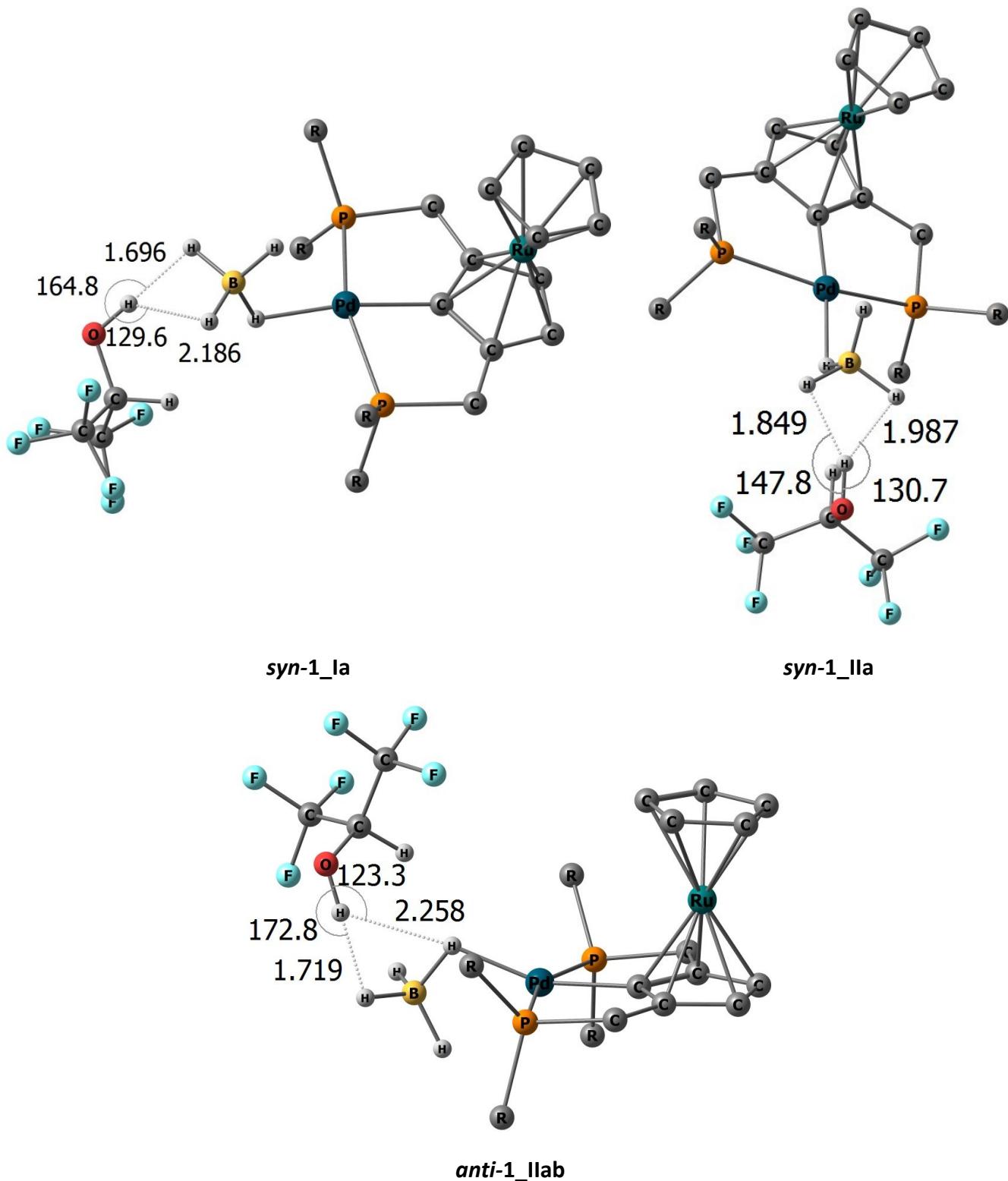


Figure S19. QTAIM molecular graph of M06-optimized geometries of DHB complexes of **1** with MeOH in toluene. Hydrogen atoms of the ligand are omitted for clarity. Red spheres denote critical points (+3;−1), yellow spheres denote ring critical points (+3;+1) and green spheres denote cage critical points (+3;+3).

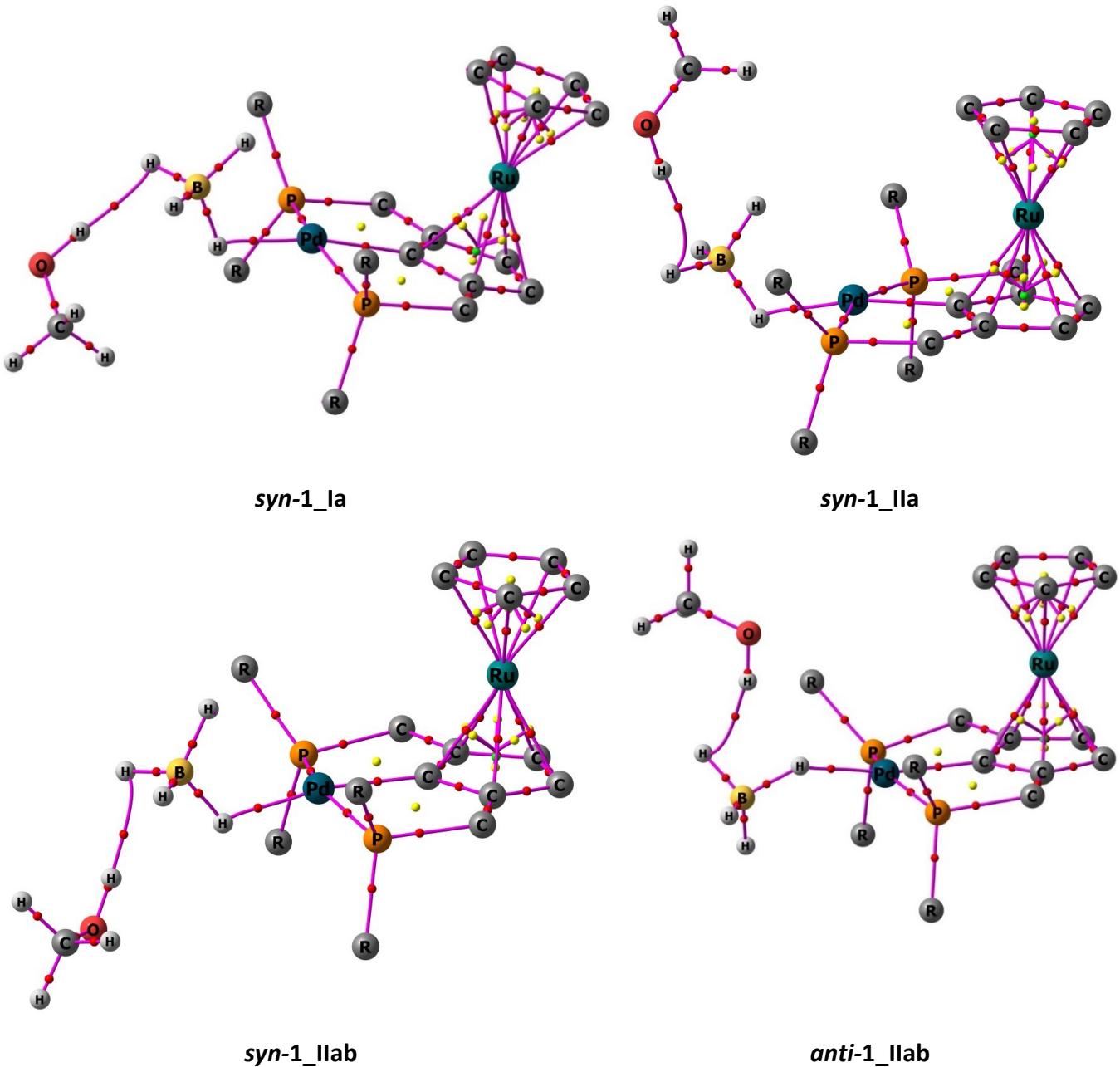


Figure S20. QTAIM molecular graph of M06-optimized geometries of DHB complexes of **1** with TFE in toluene. Hydrogen atoms of the ligand are omitted for clarity. Red spheres denote critical points (+3;-1), yellow spheres denote ring critical points (+3;+1) and green spheres denote cage critical points (+3;+3).

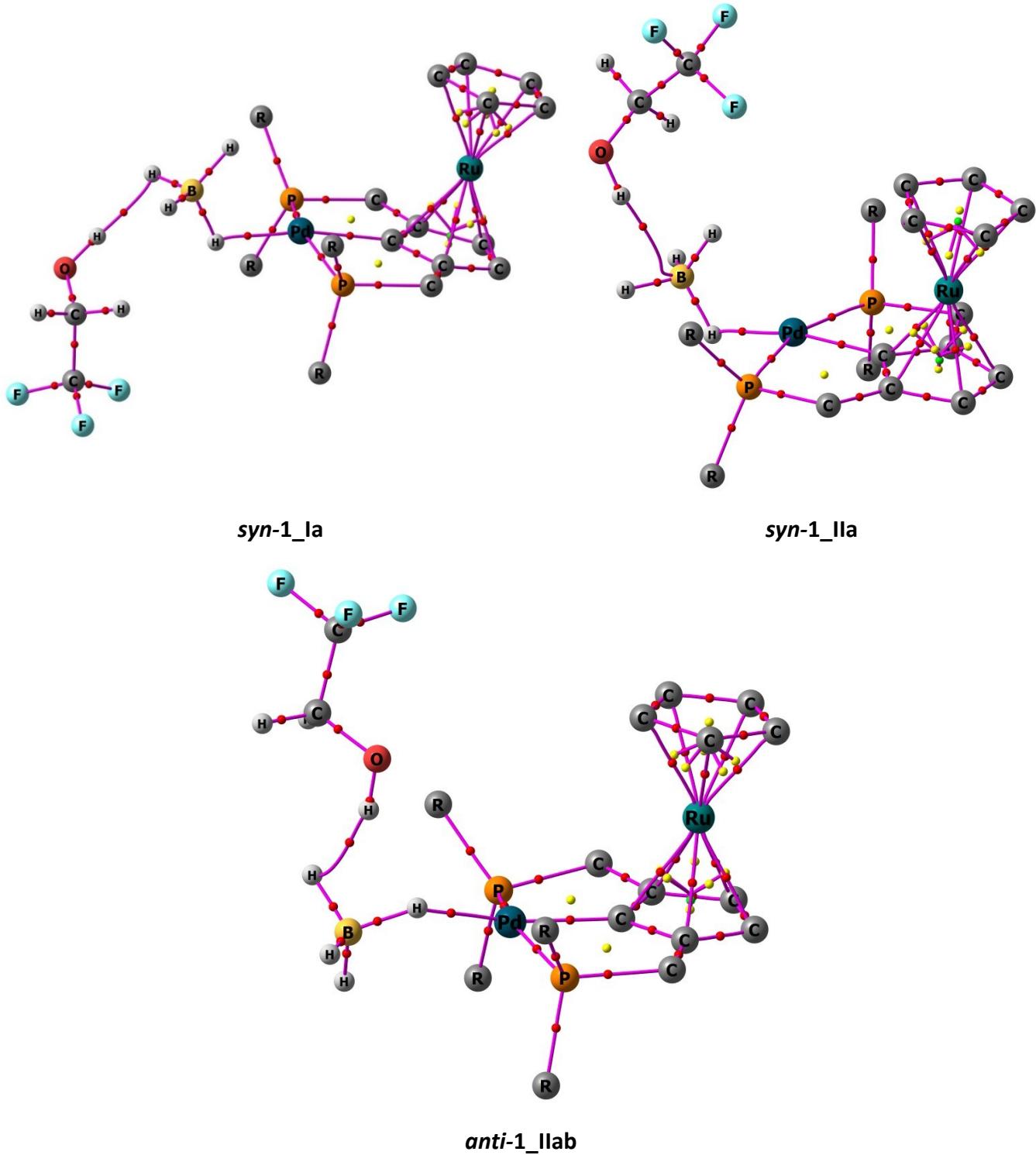
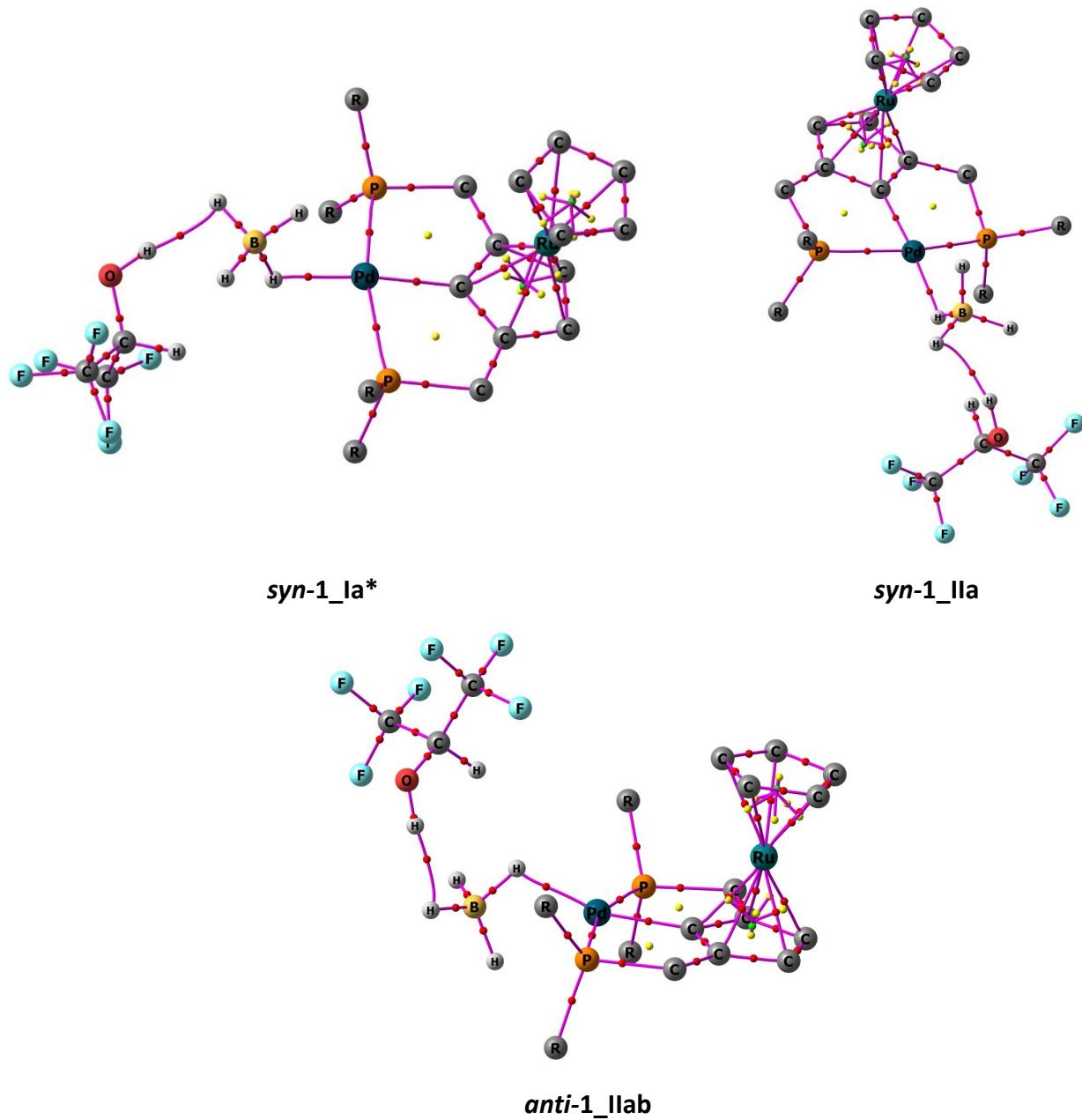


Figure S21. QTAIM molecular graph of M06-optimized geometries of DHB complexes of **1** with HFIP in toluene. Hydrogen atoms of the ligand are omitted for clarity. Red spheres denote critical points (+3;-1), yellow spheres denote ring critical points (+3;+1) and green spheres denote cage critical points (+3;+3).



MeOH					
	$r_{(H_B \cdots H_O)}$	$\angle OH \cdots H$	$r_{(O-H)}$	$\Delta r_{(O-H)}$	$\Delta r_{(B-H)}$
<i>syn-1_la</i>	1.789	169	0.970	0.010	0.003
	1.939	166	0.969	0.010	0.004
<i>syn-1_llla</i>	2.040	133			0.000
	1.862	168	0.969	0.010	0.007
<i>syn-1_llab</i>	2.173	134			-0.005
	1.925	145	0.969	0.010	0.006
<i>anti-1_llab</i>	1.970	147			0.004
TFE					
	$r_{(H_B \cdots H_O)}$	$\angle OH \cdots H$	$r_{(O-H)}$	$\Delta r_{(O-H)}$	$\Delta r_{(B-H)}$
<i>syn-1_la</i>	1.709	172	0.973	0.010	0.006
	1.921	157	0.973	0.010	0.004
<i>syn-1_llla</i>	1.954	137			0.001
	1.825	147	0.972	0.010	0.008
<i>anti-1_llab</i>	1.954	145			0.001
HFIP					
	$r_{(H_B \cdots H_O)}$	$\angle OH \cdots H$	$r_{(O-H)}$	$\Delta r_{(O-H)}$	$\Delta r_{(B-H)}$
<i>syn-1_la*</i>	1.696	165	0.978	0.015	0.007
	2.186	130			
<i>syn-1_llla</i>	1.849	148	0.975	0.013	0.007
	1.987	131			0.002
<i>anti-1_llab</i>	1.719	173	0.978	0.015	0.008
	2.258	123			

Table S9. Computed geometric parameters (distances $r_{(X-Y)}$ in Å, angles in degrees) for DHB complexes **1**·ROH in toluene.

* coordination on BH_{br} ligand generally leads to switching ***syn-1*** configuration to ***anti-1***.

Table S10. Changes of NPA charges (Δq), Wiberg bond indexes (WBI) and its changes (ΔWBI) and energetic characteristics (in kJ/mol) for DHB complexes **1**·ROH in toluene.

MeOH							
	$\Delta q[(O)H]$	$\Delta q[H(B)]$	$\Delta WBI[OH]$	$\Delta WBI[HB]$	$WBI[H \cdots H]$	$\Delta WBI[PdH(B)]$	$E^2[a]$
syn-1_Ia	0.032	-0.010	-0.097	-0.022	0.022	-0.025	16.7
syn-1_IIa	0.036	-0.003 0.000	-0.103	-0.011 0.000	0.009 0.003	-0.030	10.0 3.8
syn-1_IIab	0.034	-0.002 -0.005	-0.113	-0.021 0.035	0.013 0.001	-0.024	13.4
anti-1_IIab	0.039	-0.014 -0.076	-0.100	-0.018 -0.027	0.010 0.003	-0.007	10.0 3.3
TFE							
	$\Delta q[(O)H]$	$\Delta q[H(B)]$	$\Delta WBI[OH]$	$\Delta WBI[HB]$	$WBI[H \cdots H]$	$\Delta WBI[PdH(B)]$	$E^2[a]$
syn-1_Ia	0.041	-0.011	-0.132	-0.035	0.016	-0.035	26.4
syn-1_IIa	0.039	-0.004 0.005	-0.119	-0.015 0.002	0.011 0.007	-0.041	13.0 7.9
anti-1_IIab	0.047	-0.022 -0.081	-0.147	-0.031 -0.013	0.019 0.004	-0.017	18.4 5.0
HFIP							
	$\Delta q[(O)H]$	$\Delta q[H(B)]$	$\Delta WBI[OH]$	$\Delta WBI[HB]$	$WBI[H \cdots H]$	$\Delta WBI[PdH(B)]$	$E^2[a]$
syn-1_Ia*	0.037	-0.007 0.005	-0.147	-0.040 -0.008	0.030 0.002	-0.038	31.0 3.8
syn-1_IIa	0.035	0.001 0.003	-0.123	-0.021 -0.012	0.015 0.007	-0.039	13.0 7.9
anti-1_IIab	0.037	-0.003 -0.001	-0.147	-0.043 0.060	0.030 0.000	-0.041	29.7

* coordination on BH_{br} ligand generally leads to switching **syn-1** configuration to **anti-1**.

[a] E^2 – donation energy from $\sigma(B-H)$ of **1** to $\sigma^*(O-H)$ estimated from second-order perturbative analysis of donor–acceptor interactions within natural-bond orbital (NBO) analysis.

Table S11. Electron density at (3;−1) critical point of H···H bond (ρ_c in a.u.), Laplacian of electron density ($\nabla^2\rho_c$) and bond ellipticity (ϵ) for DHB complexes **1**·ROH in toluene.

MeOH			
	ρ_c	$\nabla^2\rho_c$	$\epsilon_{\text{H}\cdots\text{H}}$
<i>syn-1_la</i>	0.02	0.053	0.20
<i>syn-1_llaa</i>	0.02	0.046	2.03
<i>syn-1_llab</i>	0.02	0.050	0.37
<i>anti-1_llab</i>	0.02	0.048	0.78
TFE			
	ρ_c	$\nabla^2\rho_c$	$\epsilon_{\text{H}\cdots\text{H}}$
<i>syn-1_la</i>	0.02	0.060	0.17
<i>syn-1_llaa</i>	0.02	0.050	8.31
<i>anti-1_llab</i>	0.02	0.055	0.51
HFIP			
	ρ_c	$\nabla^2\rho_c$	$\epsilon_{\text{H}\cdots\text{H}}$
<i>syn-1_la*</i>	0.02	0.061	0.26
<i>syn-1_llaa</i>	0.02	0.052	1.17
<i>anti-1_llab</i>	0.02	0.060	0.23

* coordination on BH_{br} ligand generally leads to switching ***syn-1*** configuration to ***anti-1***.

Table S12. Computed formation energy for DHB complexes **1**·ROH (ΔE_{el} , ΔE_{ZPVE} , ΔH° in kJ/mol), energy of DHB formation ($\Delta H^{\text{theor}}(\Delta v)$ and $\Delta H^{\text{theor}}(\Delta A)$ in kJ/mol) and energy of H···H interaction ($E_{H\cdots H}$ in kJ/mol).

MeOH						
	ΔE_{el}	ΔE_{ZPVE}	ΔH°	$\Delta H^{\text{theor}}(\Delta v)^{[a]}$	$\Delta H^{\text{theor}}(\Delta A)^{[b]}$	$E_{H\cdots H}^{[c]}$
<i>syn-1_Ia</i>	-31.0	-26.4	-25.5	-14.3	-16.7	-13.8
<i>syn-1_Ila</i>	-32.2	-28.0	-26.8	-13.8	-13.0	-10.9
<i>syn-1_IIab</i>	-31.4	-27.2	-25.9	-12.1	-14.6	-12.1
<i>anti-1_IIab</i>	-19.2	-17.6	-15.5	-12.6	-10.9	-12.6
TFE						
	ΔE_{el}	ΔE_{ZPVE}	ΔH°	$\Delta H^{\text{theor}}(\Delta v)^{[a]}$	$\Delta H^{\text{theor}}(\Delta A)^{[b]}$	$E_{H\cdots H}^{[c]}$
<i>syn-1_Ia</i>	-42.3	-38.1	-36.4	-16.7	-26.8	-17.2
<i>syn-1_Ila</i>	-44.8	-41.0	-39.2	-14.6	-22.2	-13.4
<i>anti-1_IIab</i>	-25.9	-25.0	-22.2	-14.6	-19.7	-15.5
HFIP						
	ΔE_{el}	ΔE_{ZPVE}	ΔH°	$\Delta H^{\text{theor}}(\Delta v)^{[a]}$	$\Delta H^{\text{theor}}(\Delta A)^{[b]}$	$E_{H\cdots H}^{[c]}$
<i>syn-1_Ia*</i>	-59.8	-54.0	-52.7	-21.3	-29.3	-18.8
<i>syn-1_Ila</i>	-61.5	-54.8	-54.0	-21.8	-23.4	-14.2
<i>anti-1_IIab</i>	-52.7	-47.7	-46.4	-21.3	-21.3	-18.0

* coordination on BH_{br} ligand generally leads to switching ***syn-1*** configuration to ***anti-1***. [a] DHB formation enthalpy in gas phase calculated from Δv_{OH} by formula $-\Delta H^\circ_{HB}(\Delta v) = 75 \cdot |\Delta v_{XH}| / (720 + |\Delta v_{XH}|)$, see refs. ³¹⁻³⁶; [b] DHB formation enthalpy in gas phase calculated from ΔA_{OH} by formula $-\Delta H^\circ_{HB}(\Delta A) = 12.1 \cdot (\nabla A_{\text{bonded}} - \nabla A_{\text{free}})$, see refs. ³¹⁻³⁶; [c] $E_{H\cdots H} = 0.5 \cdot V(r)$ in kJ/mol, see refs. ^{43, 44}

Figure S22. General view of molecular structure of **2** (thermal ellipsoids of all non-hydrogen atoms are drawn at the 50% probability level).

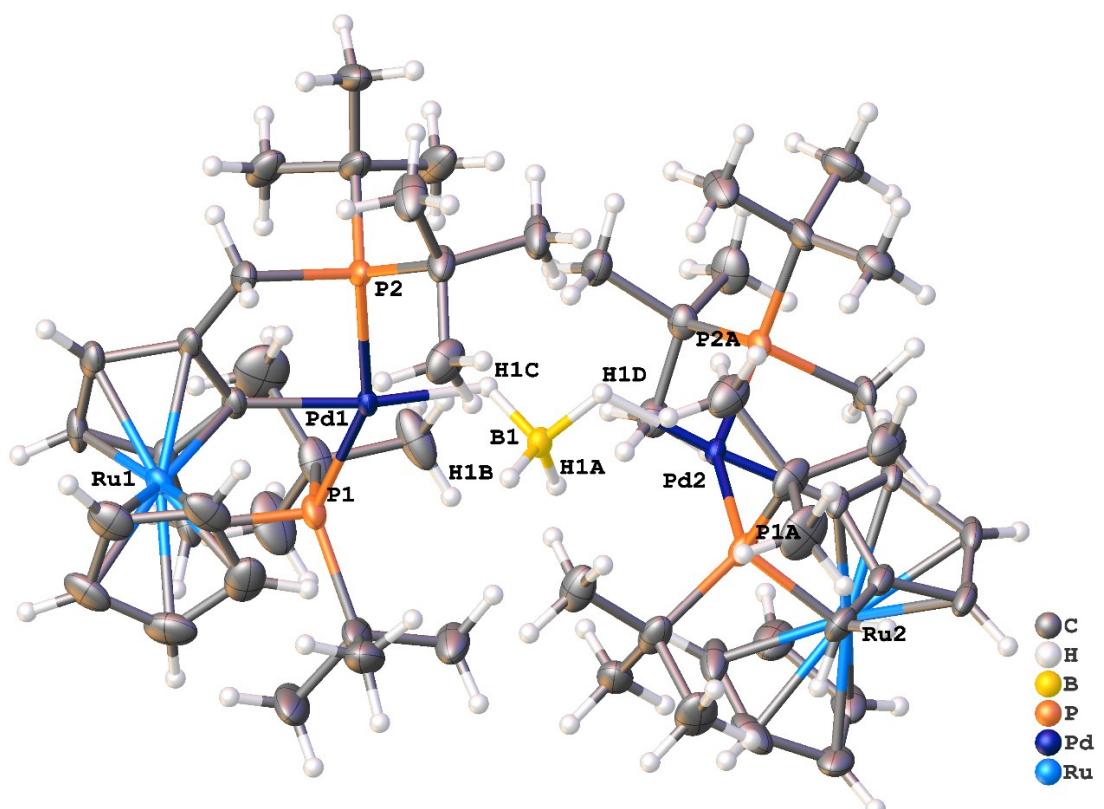
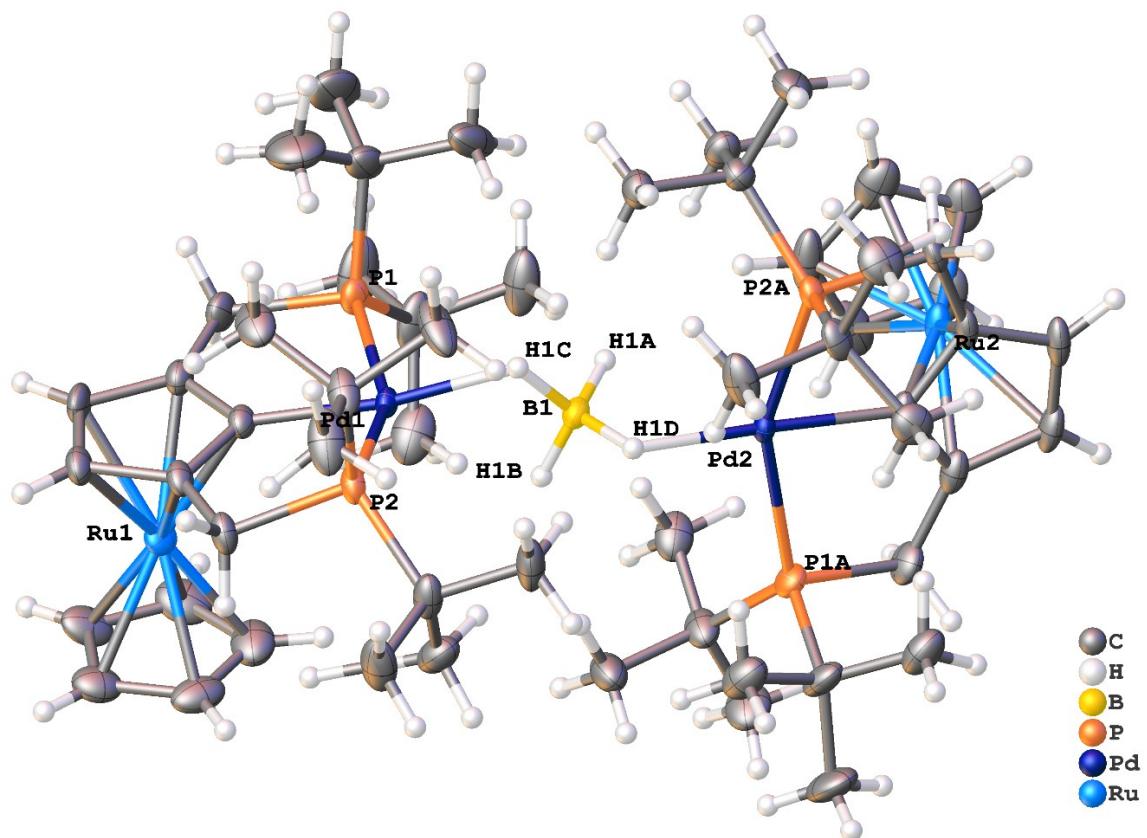


Table S13. Geometry parameters of crystal and M06-optimized structure of **2**.

Distances	2	M06
Pd ₁ –C	1.980(3)	1.996
Pd ₂ –C	1.981(4)	1.994
Pd ₁ –P ₁	2.349(1)	2.390
Pd ₁ –P ₂	2.3913(9)	2.436
Pd ₂ –P ₃	2.380(1)	2.410
Pd ₂ –P ₄	2.382(1)	2.421
Pd ₁ ···B	2.577(5)	2.585
Pd ₂ ···B	2.555(5)	2.577
Pd ₁ –H ₁ (B)	1.866	1.901
Pd ₁ ···H ₂ (B)	2.434	2.452
Pd ₂ –H ₃ (B)	1.926	1.895
Pd ₂ ···H ₄ (B)	2.490	2.481
B–H ₁	1.190	1.267
B–H ₂	1.073	1.208
B–H ₃	1.256	1.272
B–H ₄	1.070	1.207
C–Pd ₁ –B	164.4(1)	162.41
C–Pd ₂ –B	156.3(1)	158.14
P ₁ –Pd ₁ –P ₂	156.66(4)	156.50
P ₃ –Pd ₂ –P ₄	154.24(4)	154.80
Pd ₁ –H ₁ –B	113.2	107.13
Pd ₁ –H ₂ –B	85.3	81.93
Pd ₂ –H ₃ –B	104.9	107.85
Pd ₂ –H ₄ –B	81.2	81.09

Figure S23. QTAIM molecular graph of M06-optimized geometry of **2**. Hydrogen atoms of the ligand are omitted for clarity. Red spheres denote critical points (+3;-1), yellow spheres denote ring critical points (+3;+1) and green spheres denote cage critical points (+3;+3).

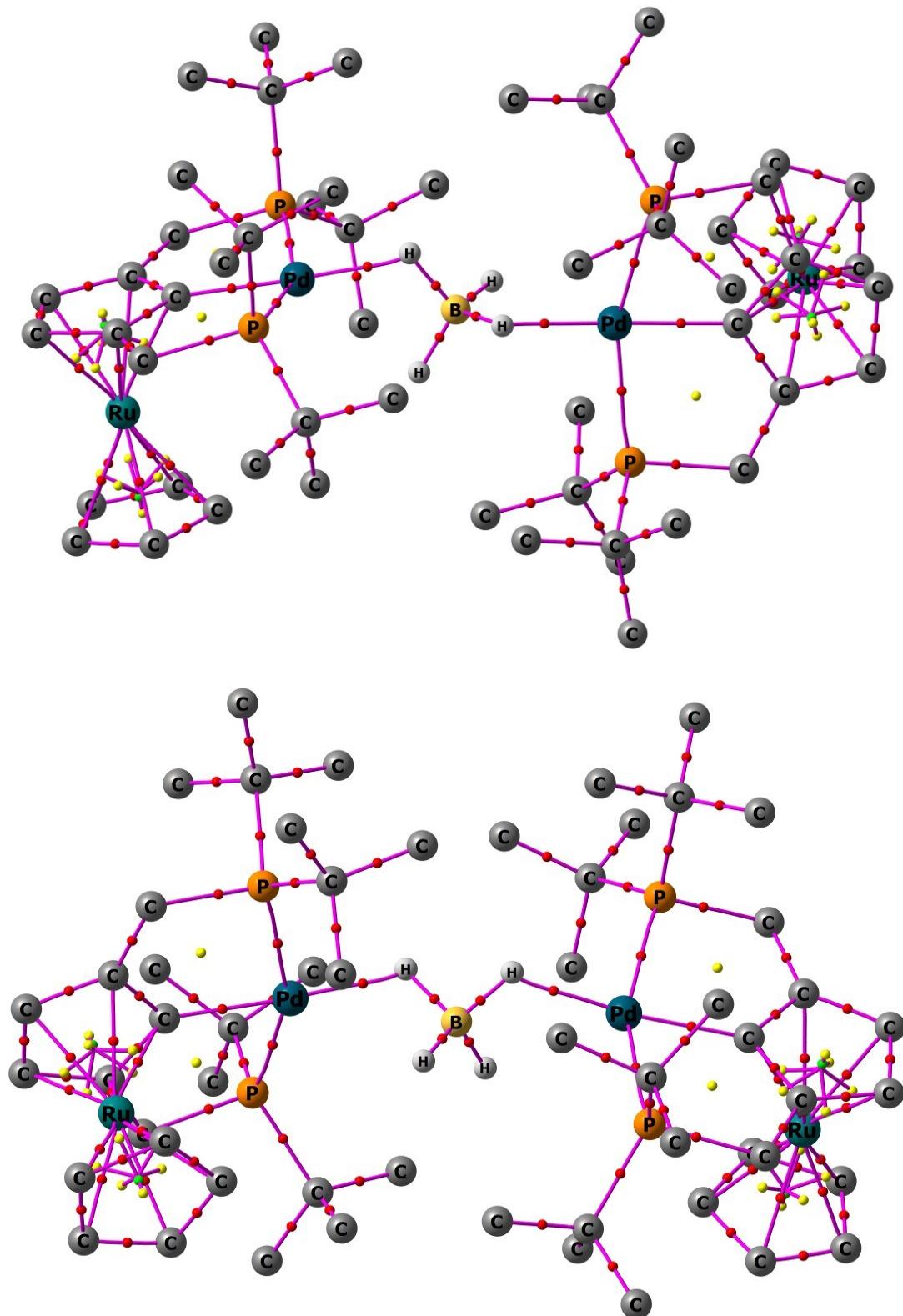


Figure S24. FTIR spectra of **2** and **2-d₄** in KBr pellet.

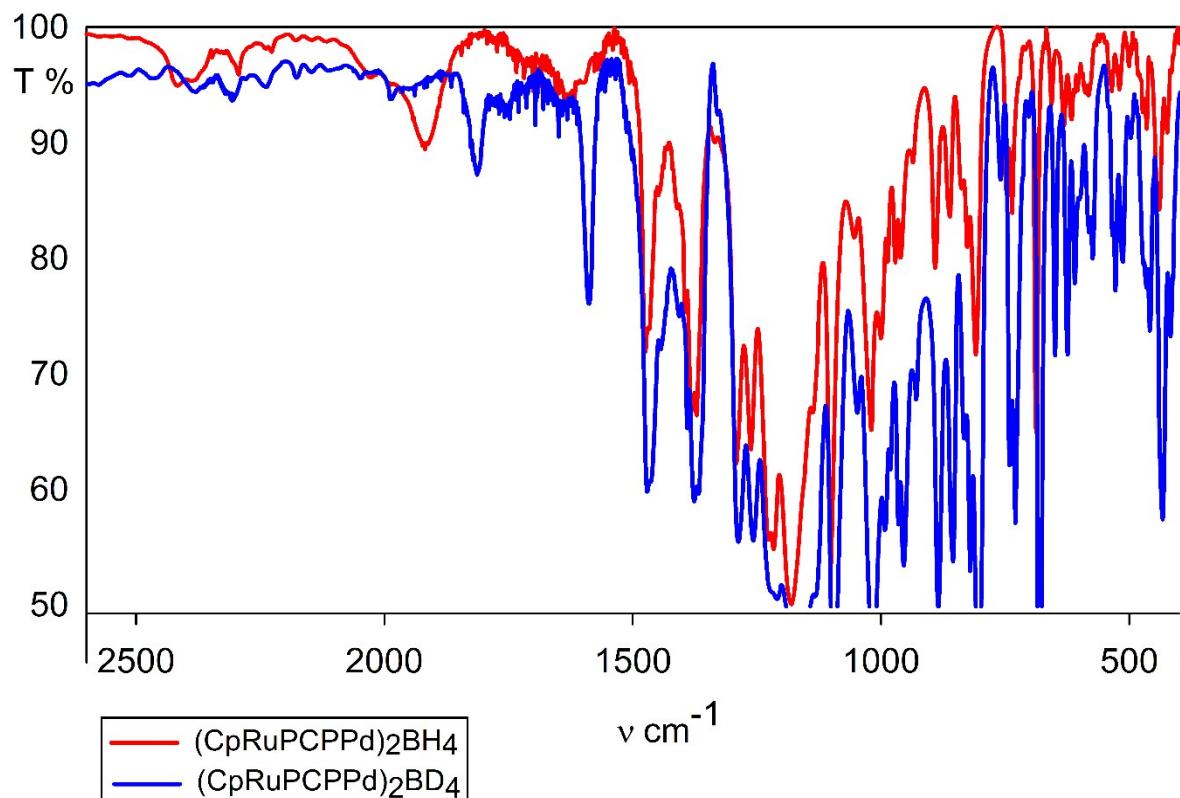


Figure S25. The $^{31}\text{P}\{^1\text{H}\}$ monitoring of the reaction of **1-d₄** (spectrum 1) with 20-fold excess of HFIP (spectrum 2). Spectra (3–5) are measured at 0.5 h, 18 h and 40 h after mixing.

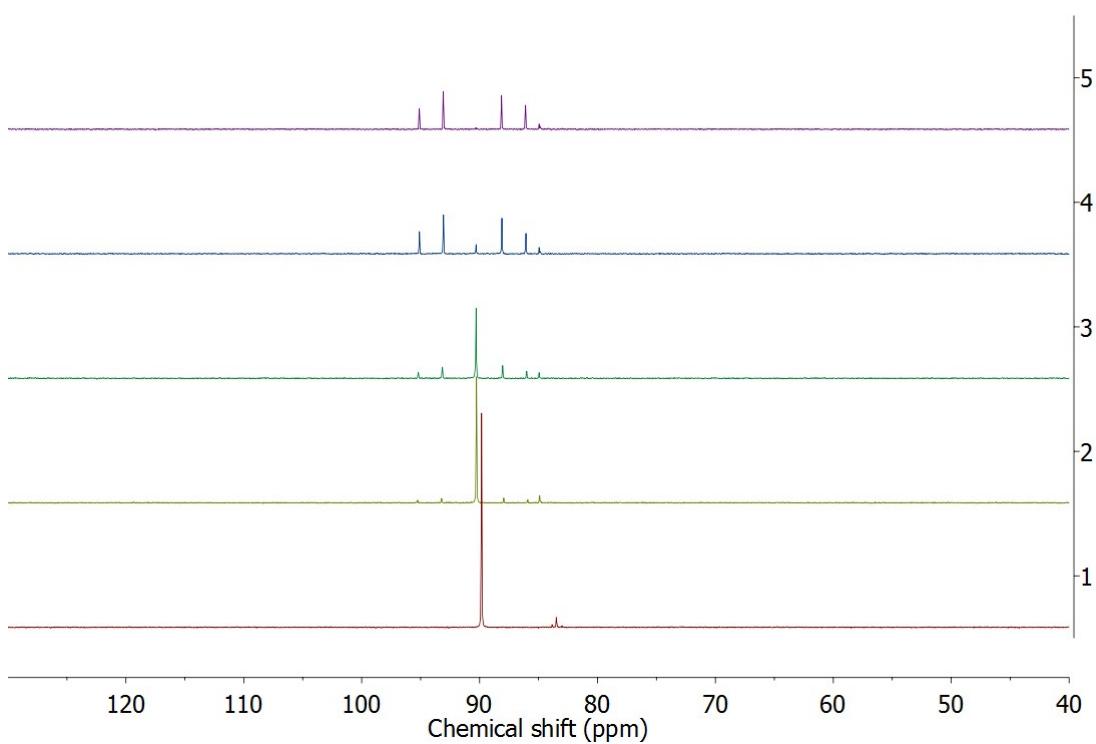


Figure S26. M06-optimized geometry of **3**. Hydrogen atoms of the ligand are omitted for clarity.

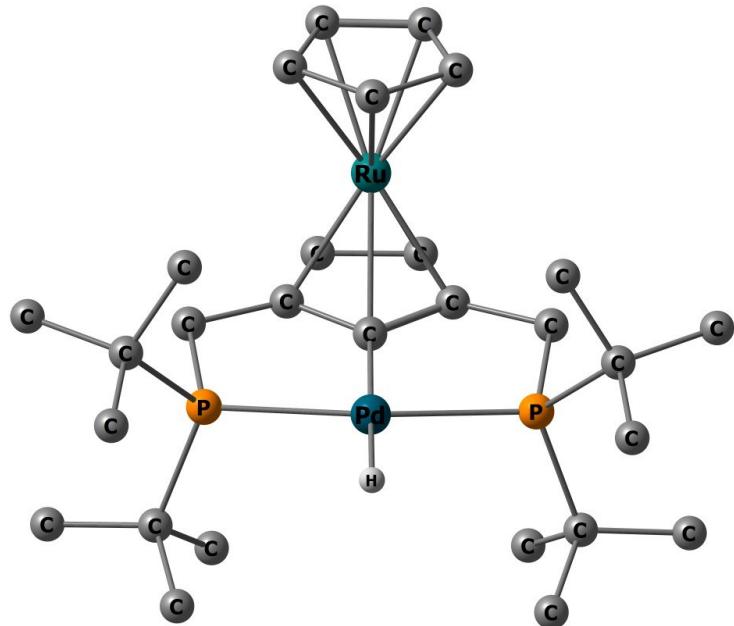


Figure S27. ^1H NMR monitoring of BH_3 abstraction from **1** by pyridine excess (15 mol. equiv.) in toluene- d_8 .

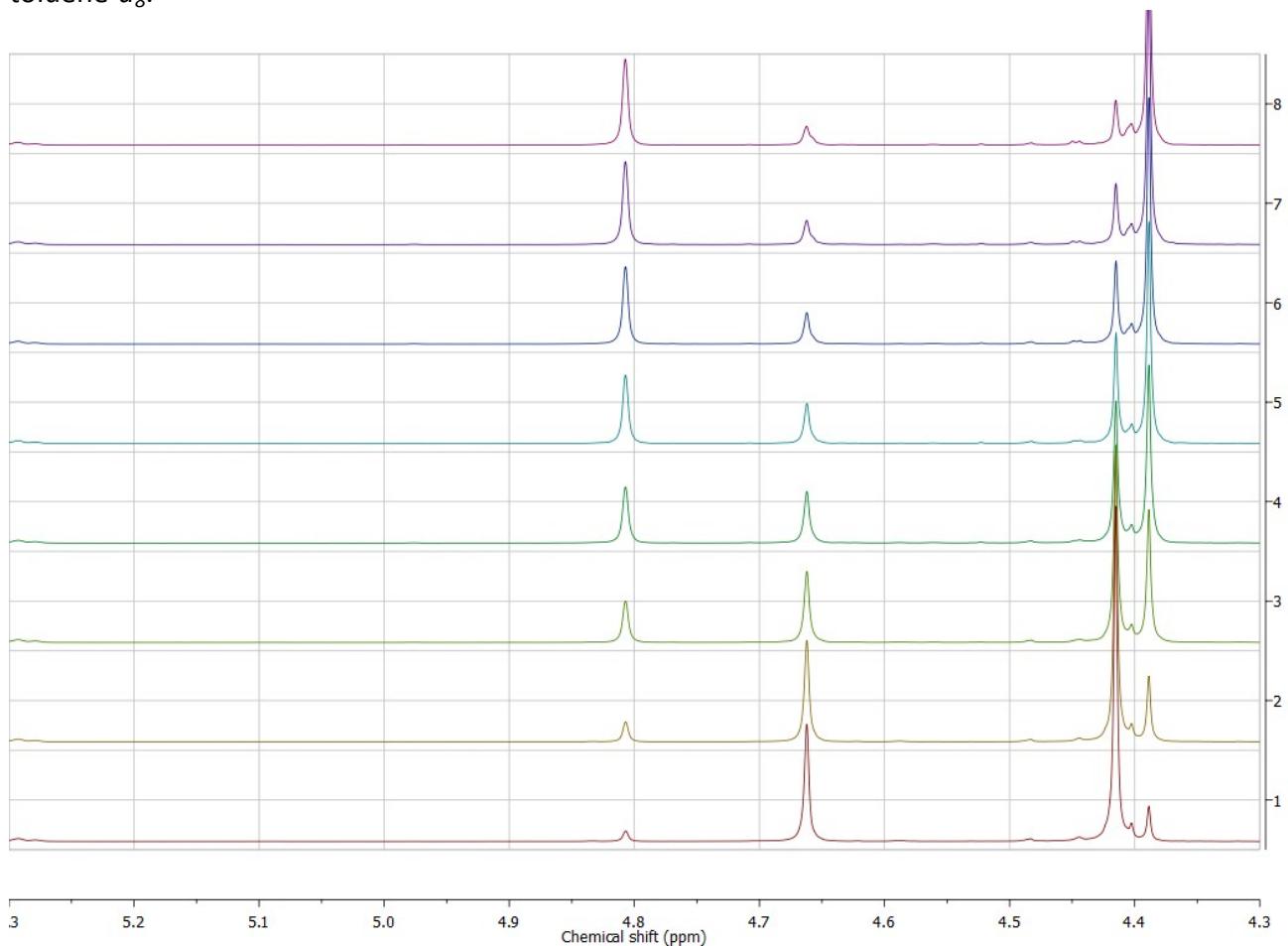


Figure S28. The linear dependence of $\ln(1/K)$ vs time (s) of BH_3 abstraction from **1** by pyridine excess (15 mol. equiv.) in toluene- d_8 .

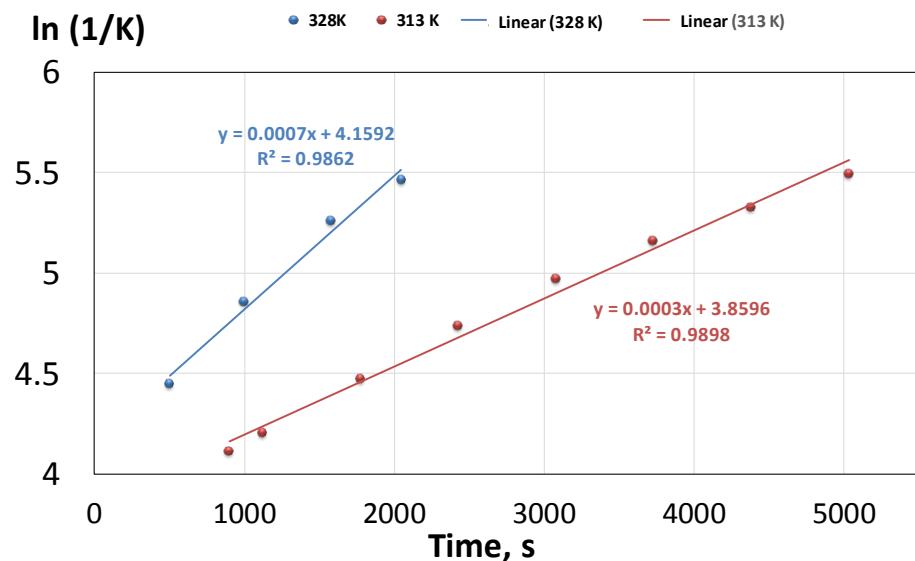


Figure S29. Computed energy profiles of BH_3 abstraction from **1**: (a) by THF in THF and Py in toluene at 298 K and (b) by Py in toluene at 298–328 K .

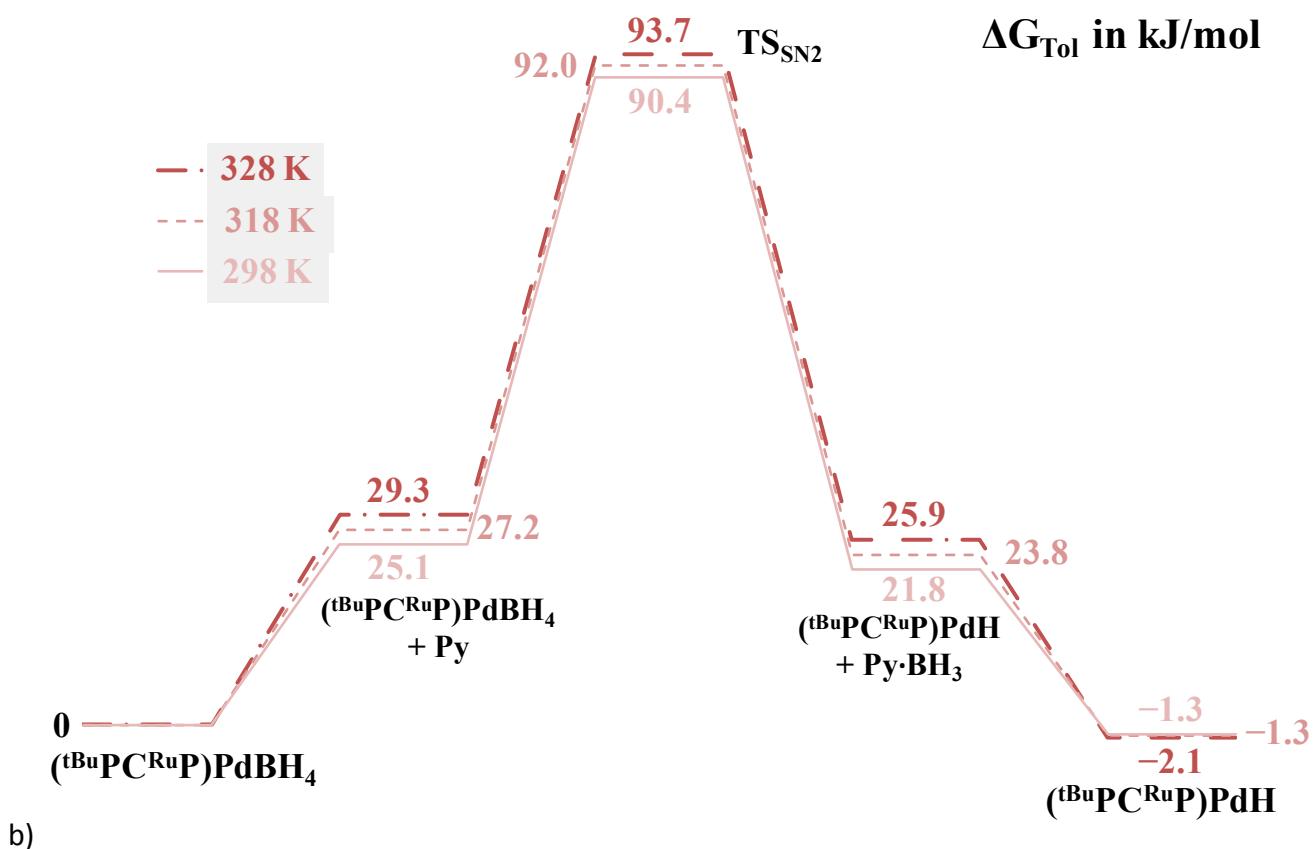
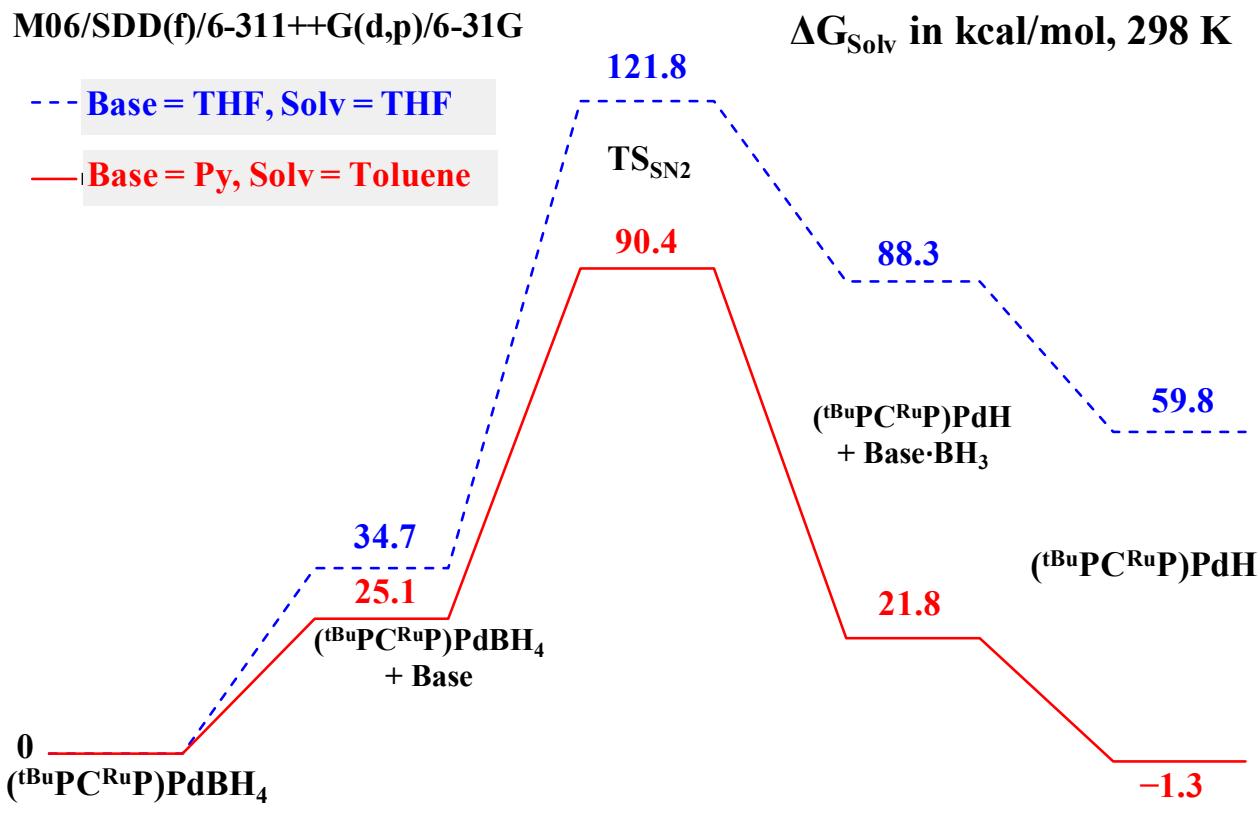


Figure S30. ^1H NMR spectrum (500 MHz) of **3** in toluene-d₈.

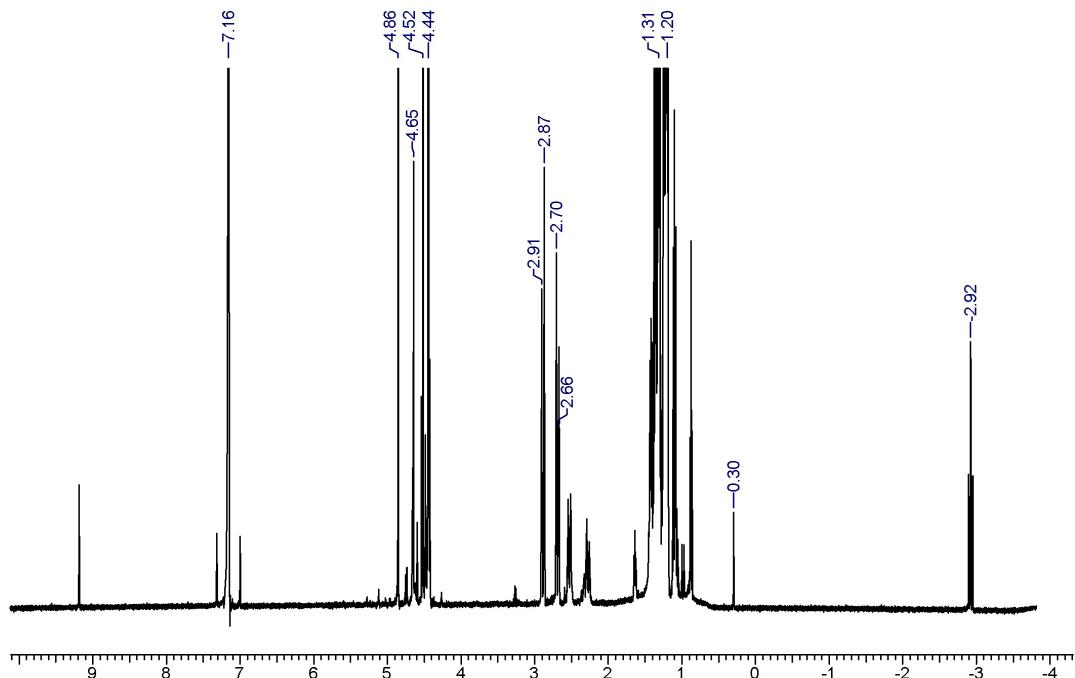


Figure S31. ^{31}P NMR spectrum (202.5 MHz) of **3** in toluene-d₈.

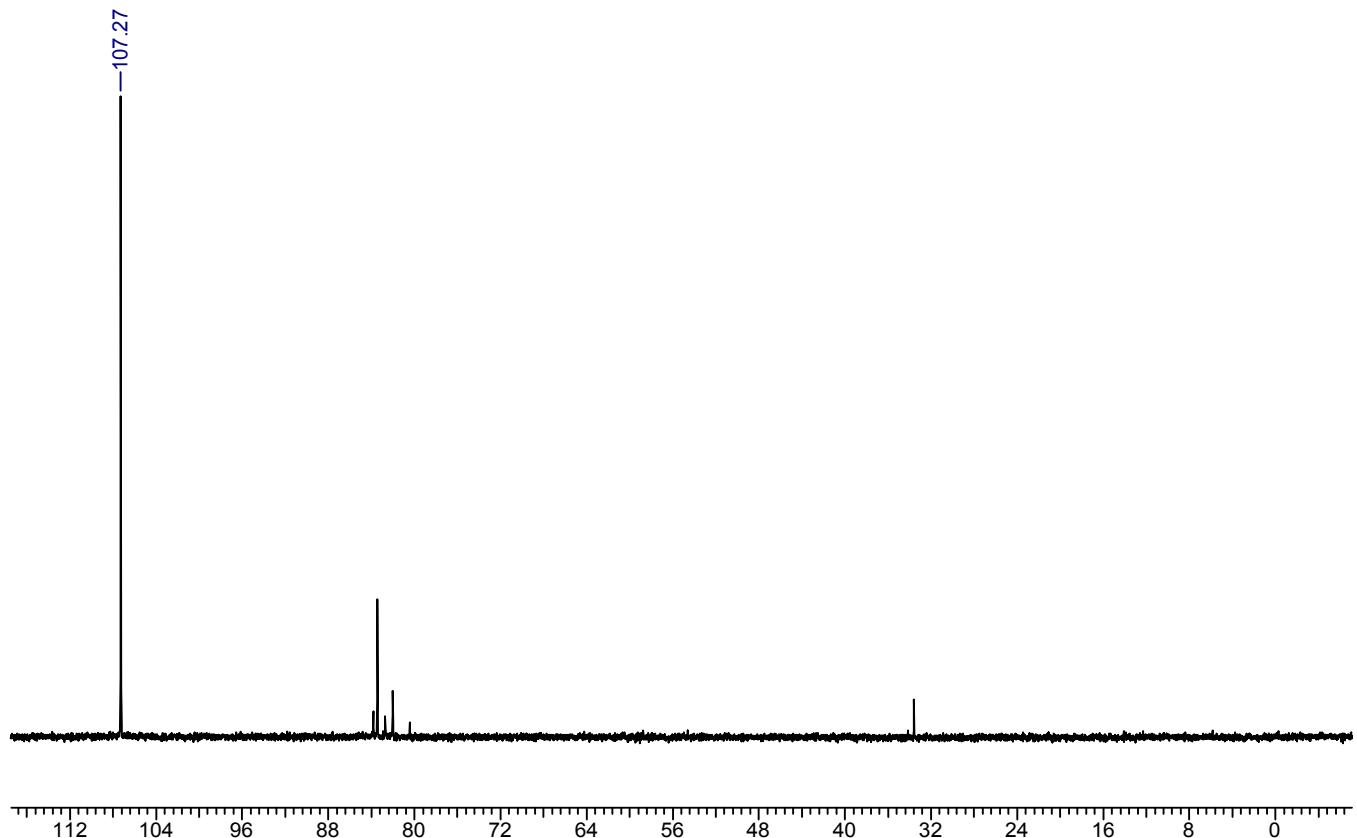


Figure S32. General view of molecular structure of **4** (thermal ellipsoids of all non-hydrogen atoms are drawn at the 50% probability level).

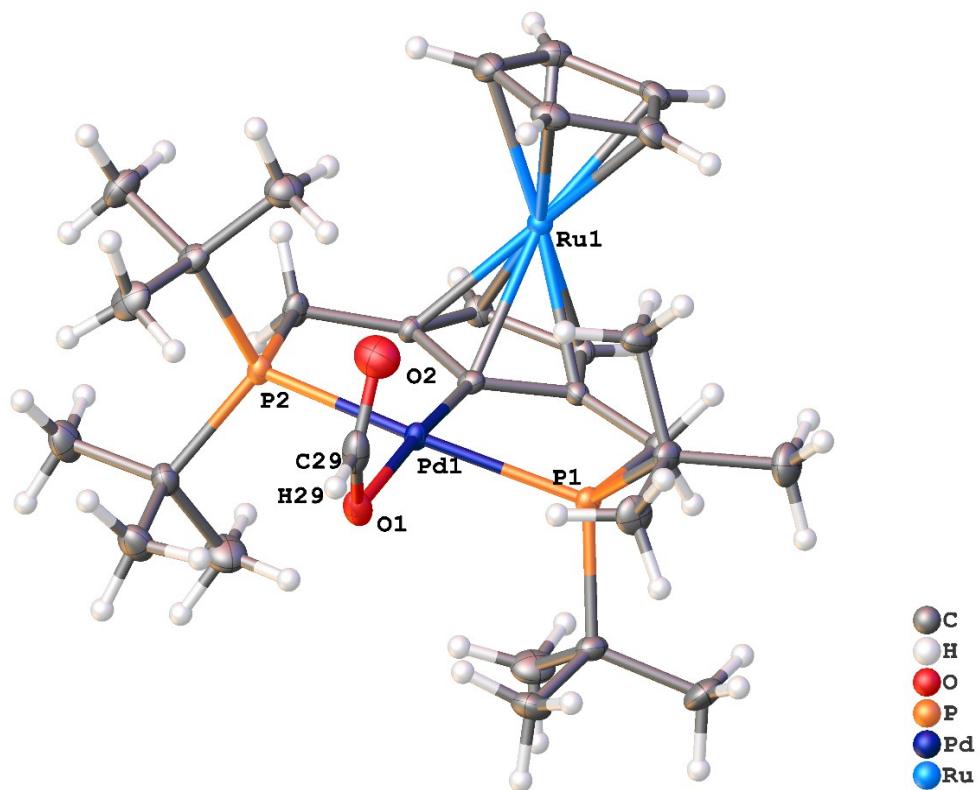


Figure S33. QTAIM molecular graph of M06-optimized geometry of **4**. Hydrogen atoms of the ligand are omitted for clarity. Red spheres denote critical points (+3;-1), yellow spheres denote ring critical points (+3;+1) and green spheres denote cage critical points (+3;+3).

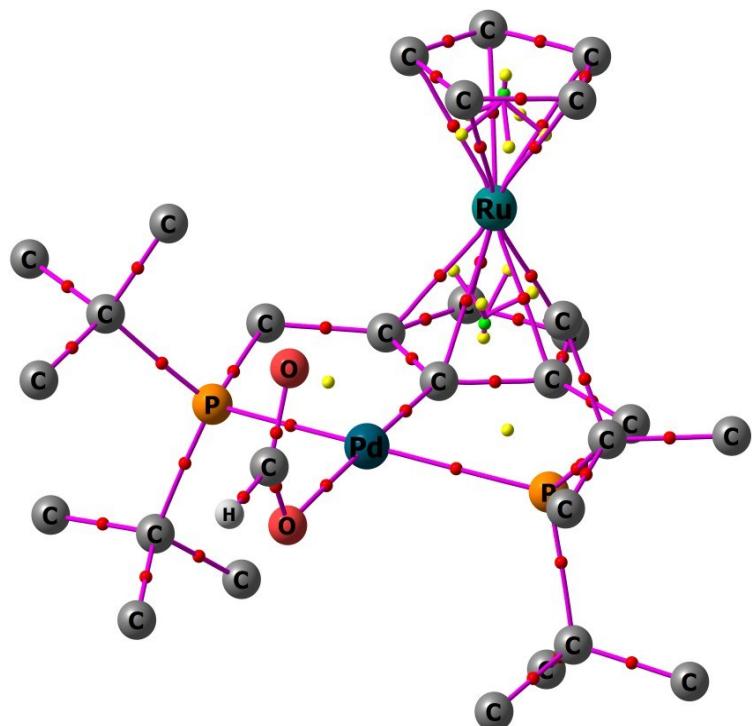


Table S14. Geometry parameters of crystal and M06-optimized structure of **4**.

Distances	4	M06
Pd–C	1.966(3)	1.979
Pd–P ₁	2.3401(8)	2.372
Pd–P ₂	2.3463(9)	2.376
Pd–O ₁	2.167(2)	2.166
Pd···O ₂	3.184(2)	3.192
Pd···C(H)	2.962(3)	2.978
C–Pd–C(H)	162.0(1)	160.54
P ₁ –Pd–P ₂	160.90(3)	159.36
Pd–O ₁ –C(H)	117.7(2)	117.81
Pd–O ₂ –C(H)	68.4(2)	68.88

Figure S34. Crystal packing of **4** (thermal ellipsoids of all non-hydrogen atoms are drawn at the 50% probability level).

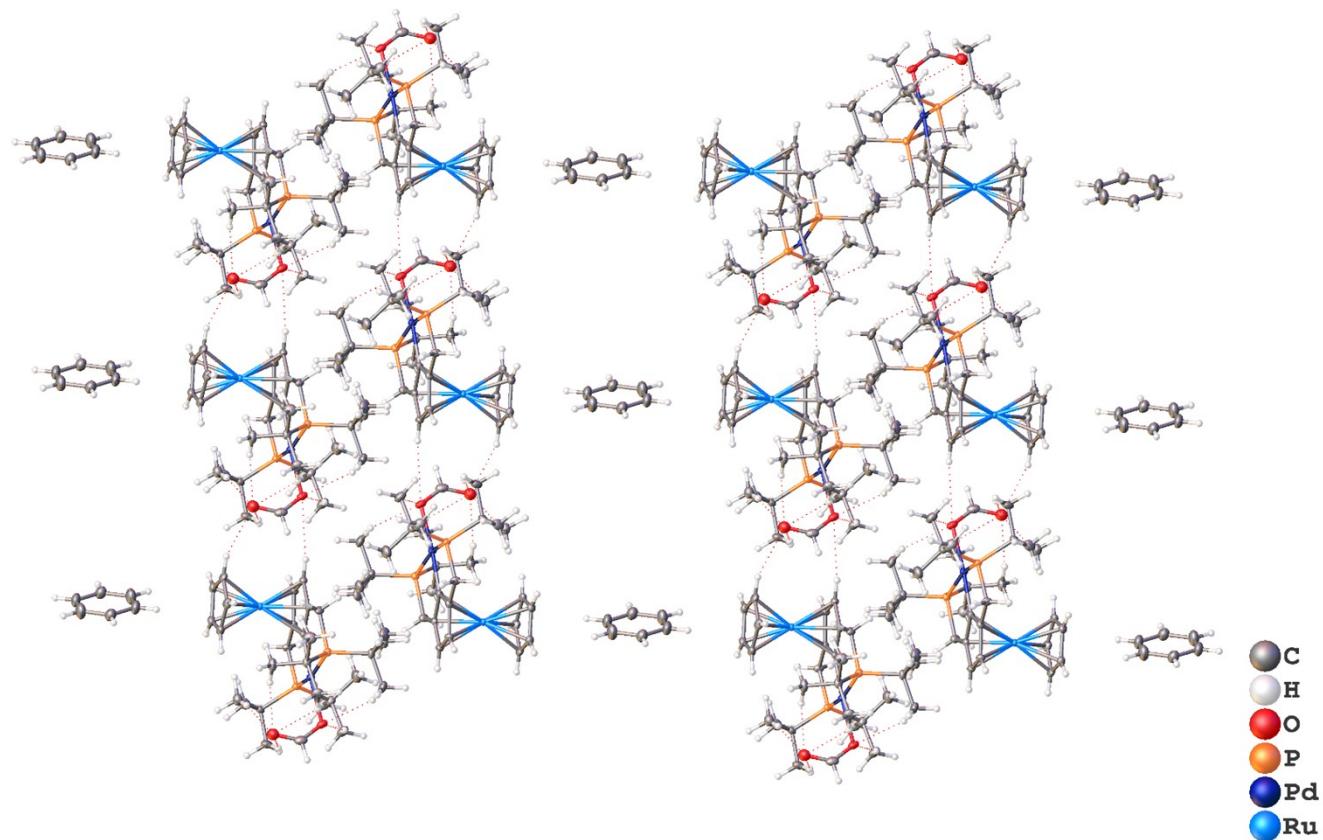


Figure S35. ^1H NMR spectrum (400 MHz) of **4** in C_6D_6 .

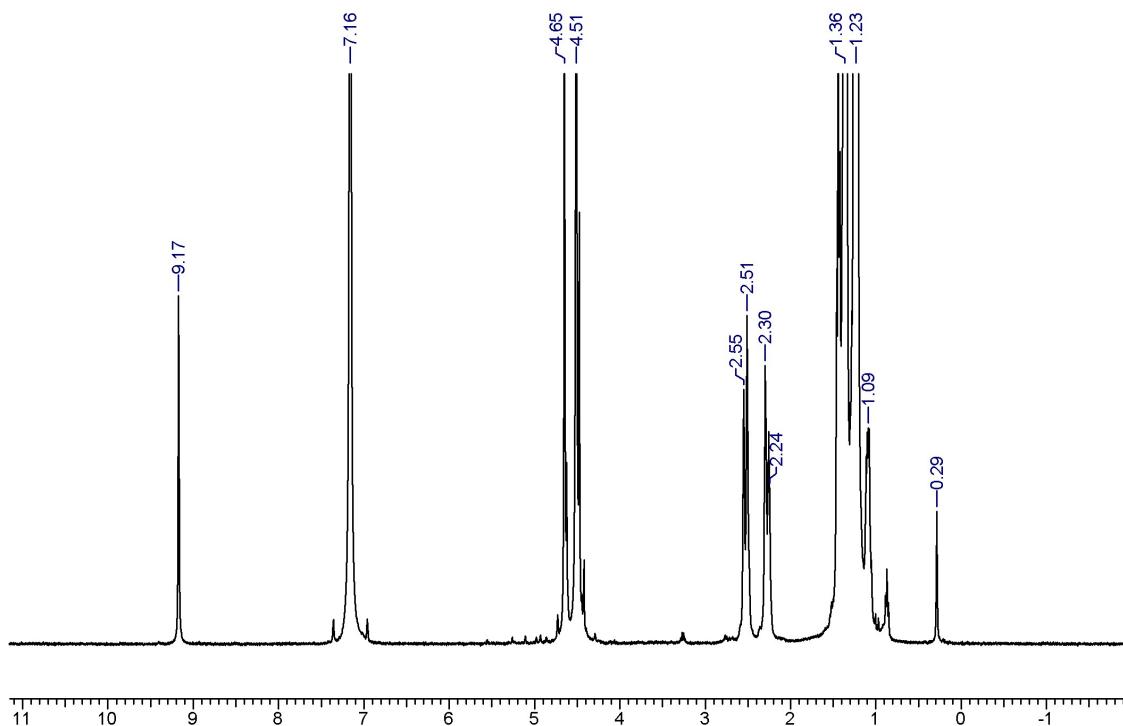


Figure S36. ^{13}C NMR spectrum (100.6 MHz) of **4** in C_6D_6 .

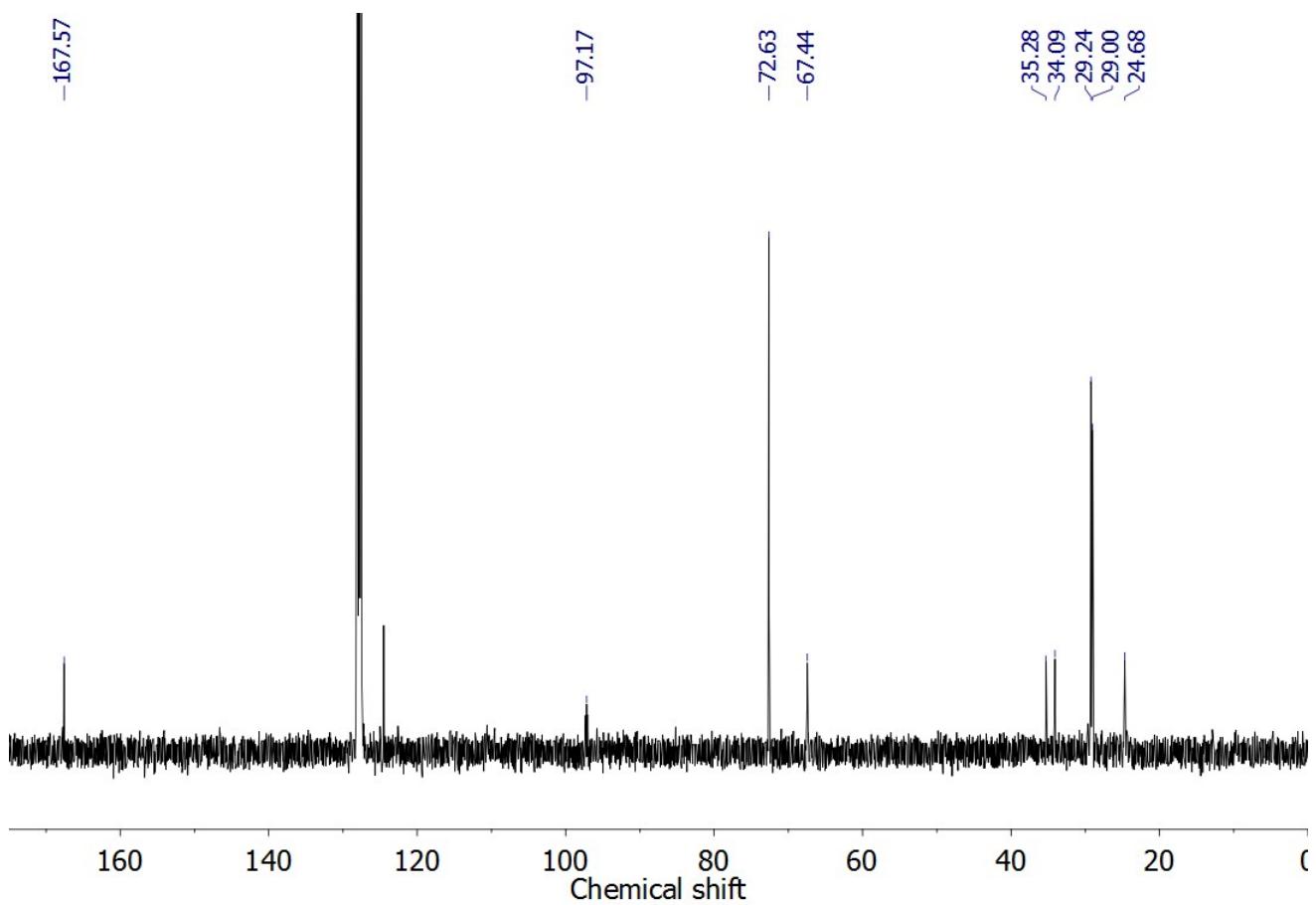


Figure S37. $^{31}\text{P}\{\text{H}\}$ NMR spectrum (162.0 MHz) of **4** in C_6D_6 .

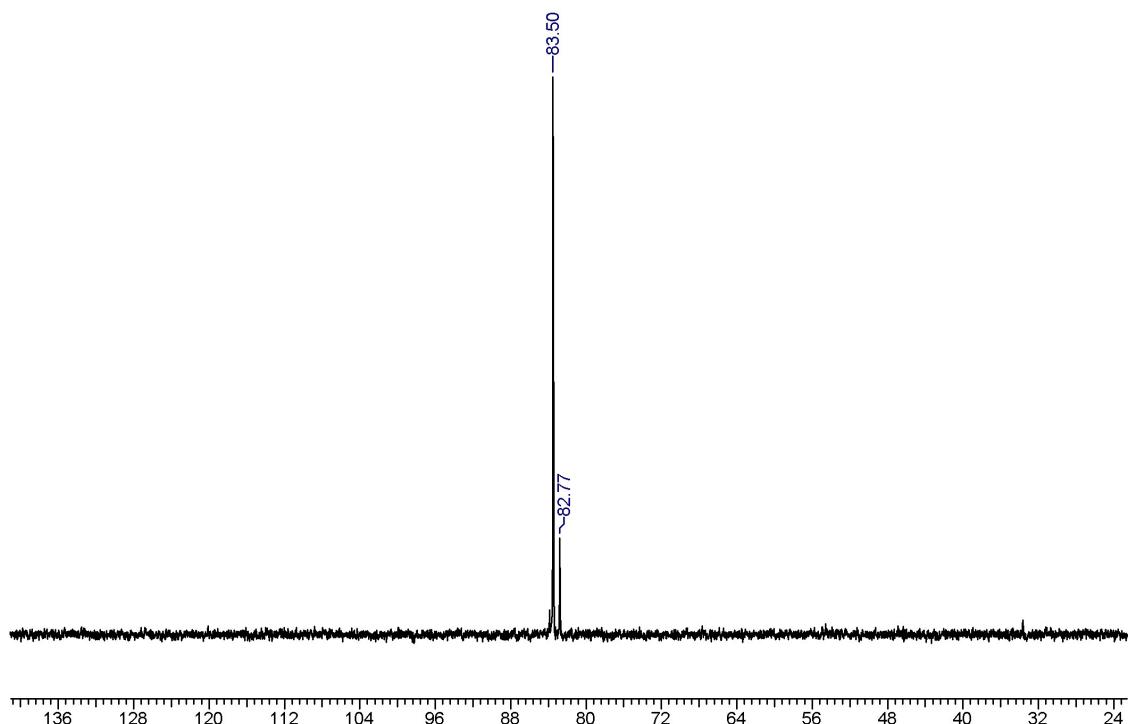


Figure S38. FTIR spectrum of **4** in KBr pellet.

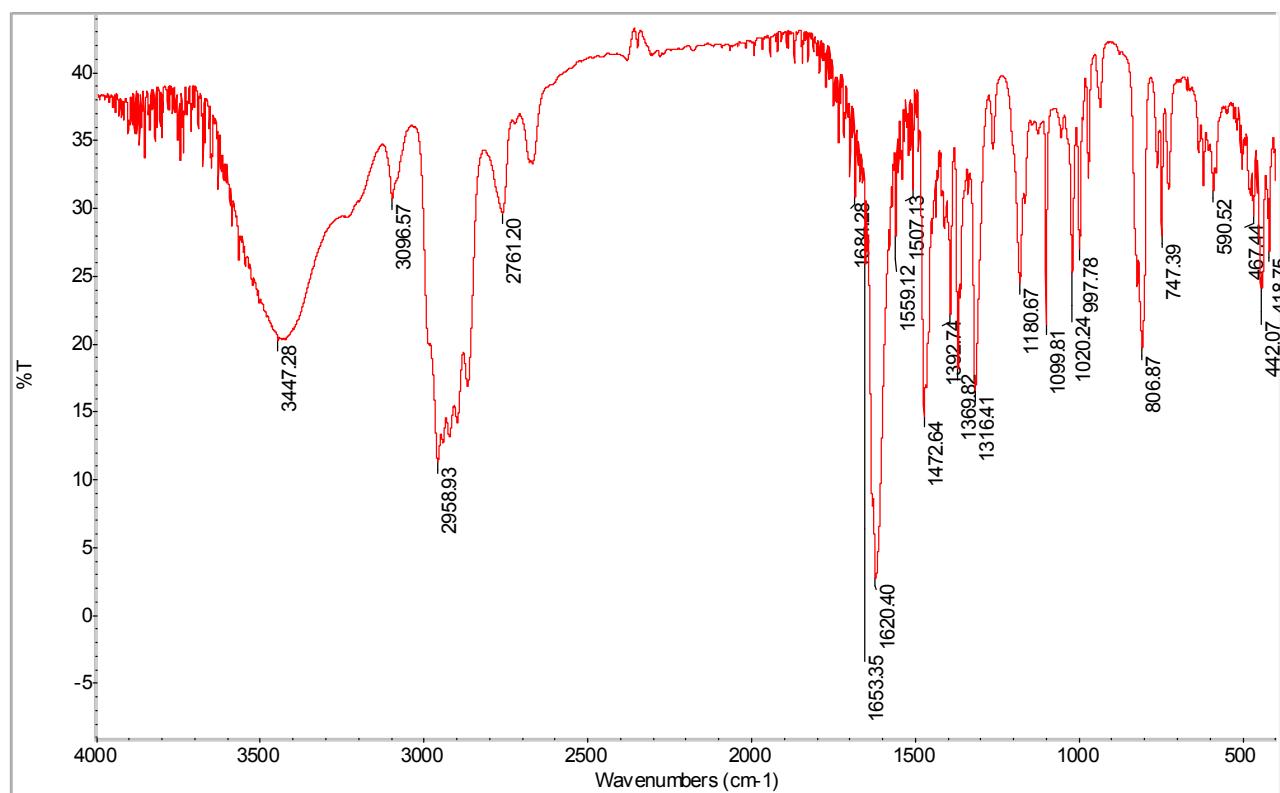


Figure S39. General view of molecular structure of **5** (thermal ellipsoids of all non-hydrogen atoms are drawn at the 50% probability level).

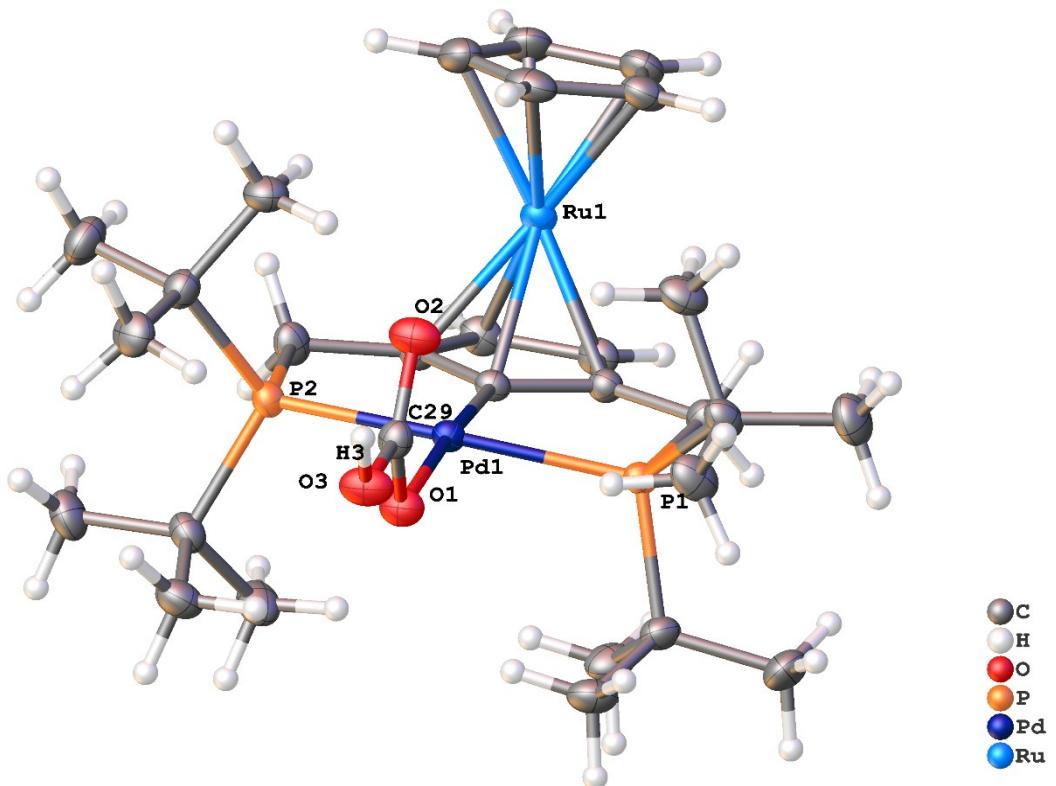


Figure S40. QTAIM molecular graph of M06-optimized geometry of **5**. Hydrogen atoms of the ligand are omitted for clarity. Red spheres denote critical points (+3;-1), yellow spheres denote ring critical points (+3;+1) and green spheres denote cage critical points (+3;+3).

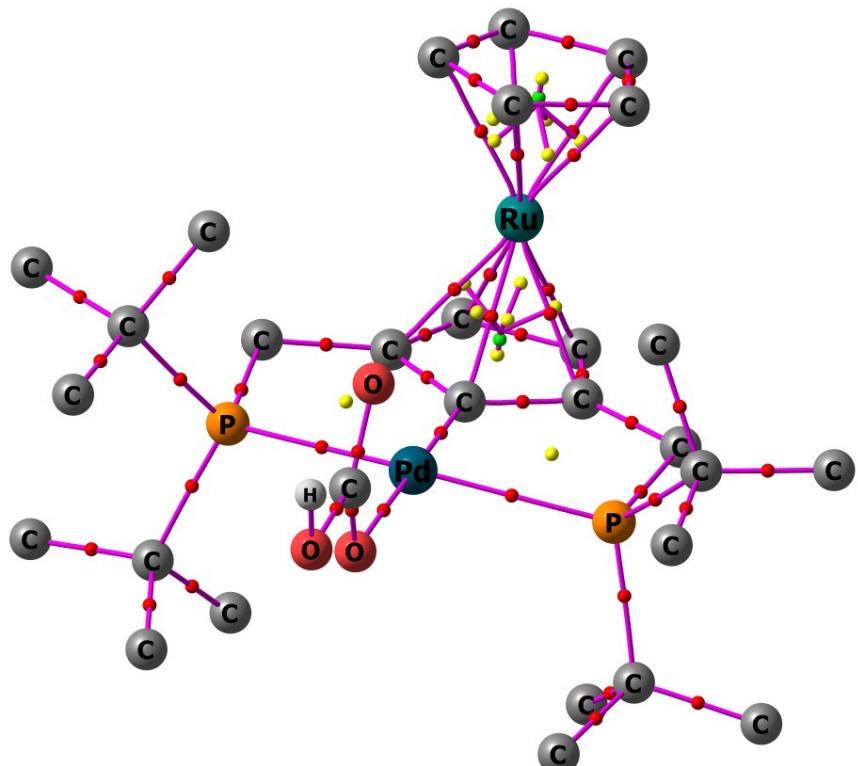


Table S15. Geometry parameters of crystal and M06-optimized structures of **5**.

Distances	5	M06
Pd–C	1.955(2)	1.975
Pd–P ₁	2.3207(6)	2.372
Pd–P ₂	2.3321(7)	2.374
Pd–O ₁	2.128(2)	2.166
Pd···O ₂	3.004(1)	3.204
Pd···C(OH)	2.905(2)	2.973
C–Pd–C(OH)	162.49(7)	159.75
P ₁ –Pd–P ₂	161.03(2)	159.42
Pd–O ₁ –C(OH)	115.7(1)	117.92
Pd–O ₂ –C(OH)	71.7(1)	68.07

Figure S41. Crystal packing of **5** (thermal ellipsoids of all non-hydrogen atoms are drawn at the 50% probability level).

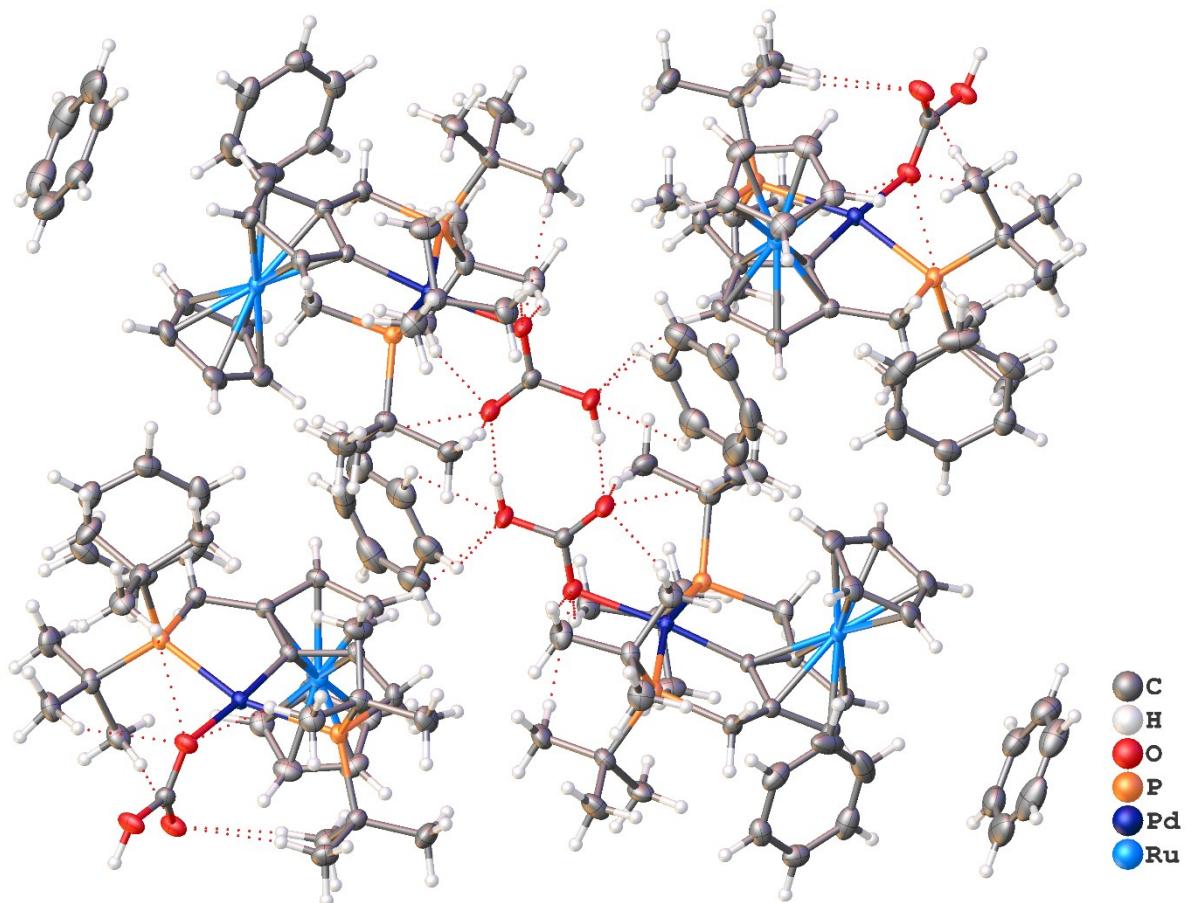


Figure S42. ^1H NMR spectrum (500 MHz) of **5** in C_6D_6 .

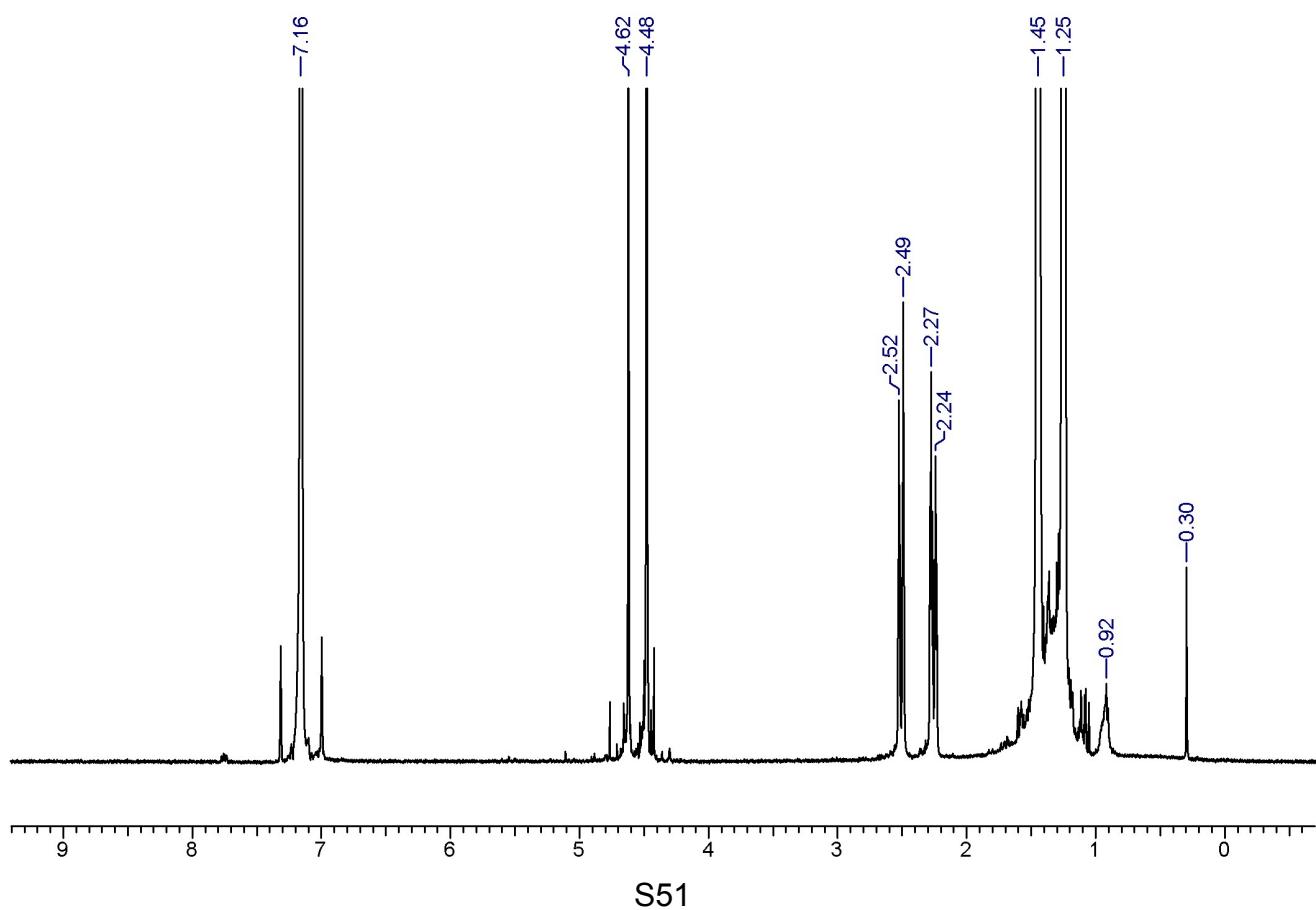


Figure S43. ^{13}C NMR spectrum (125.8 MHz) of **5** in C_6D_6 .

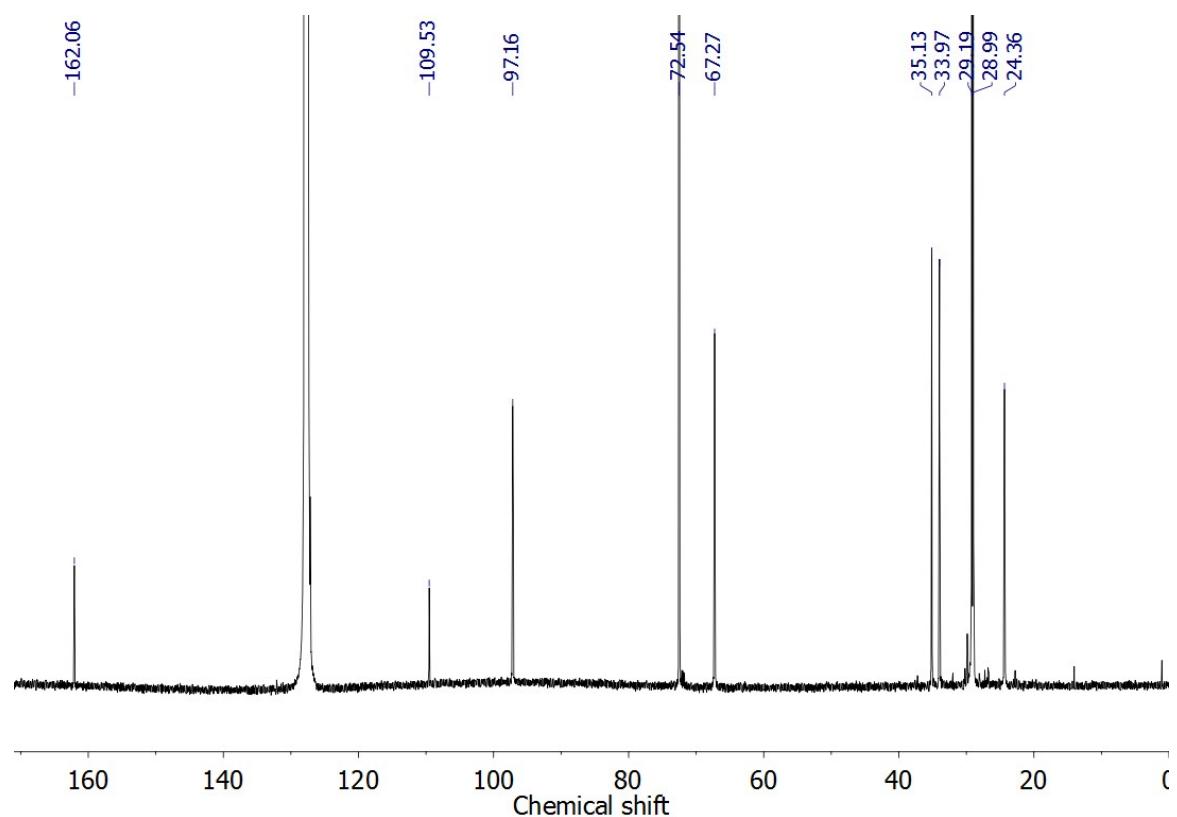


Figure S44. ^{31}P NMR spectrum (202.5 MHz) of **5** in C_6D_6 .

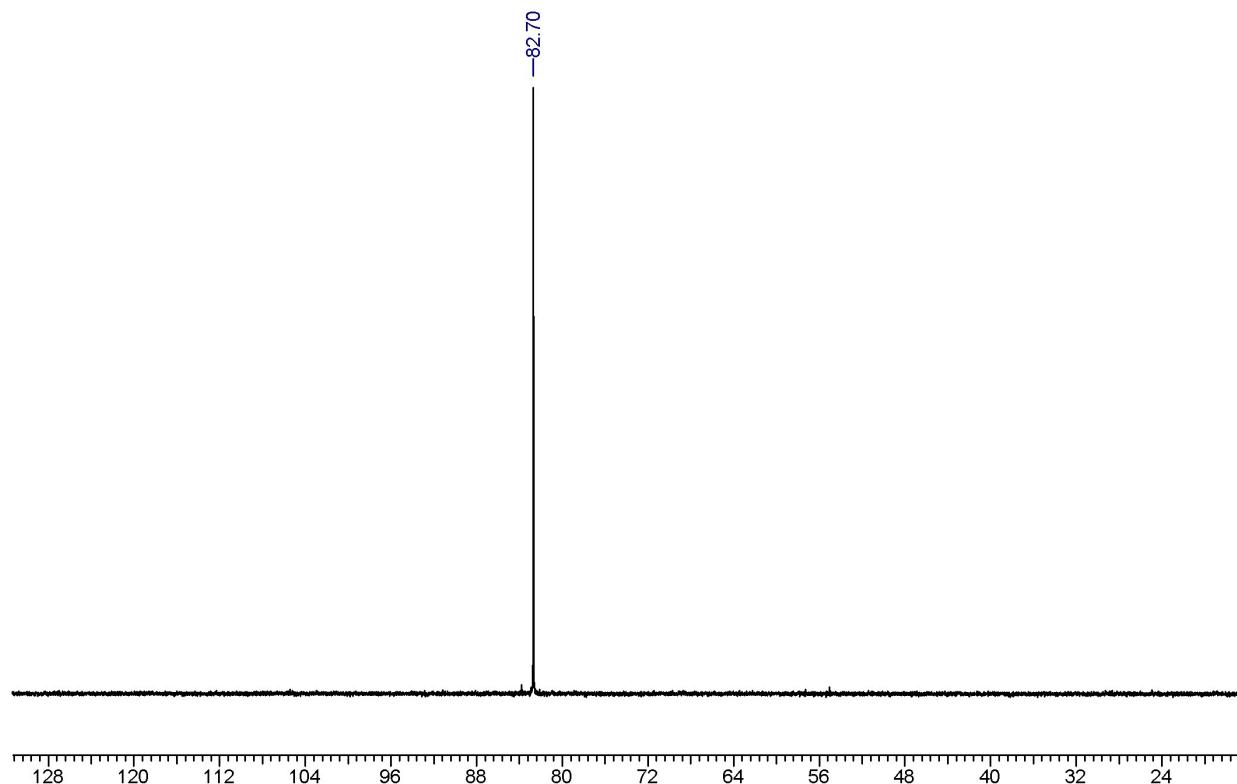
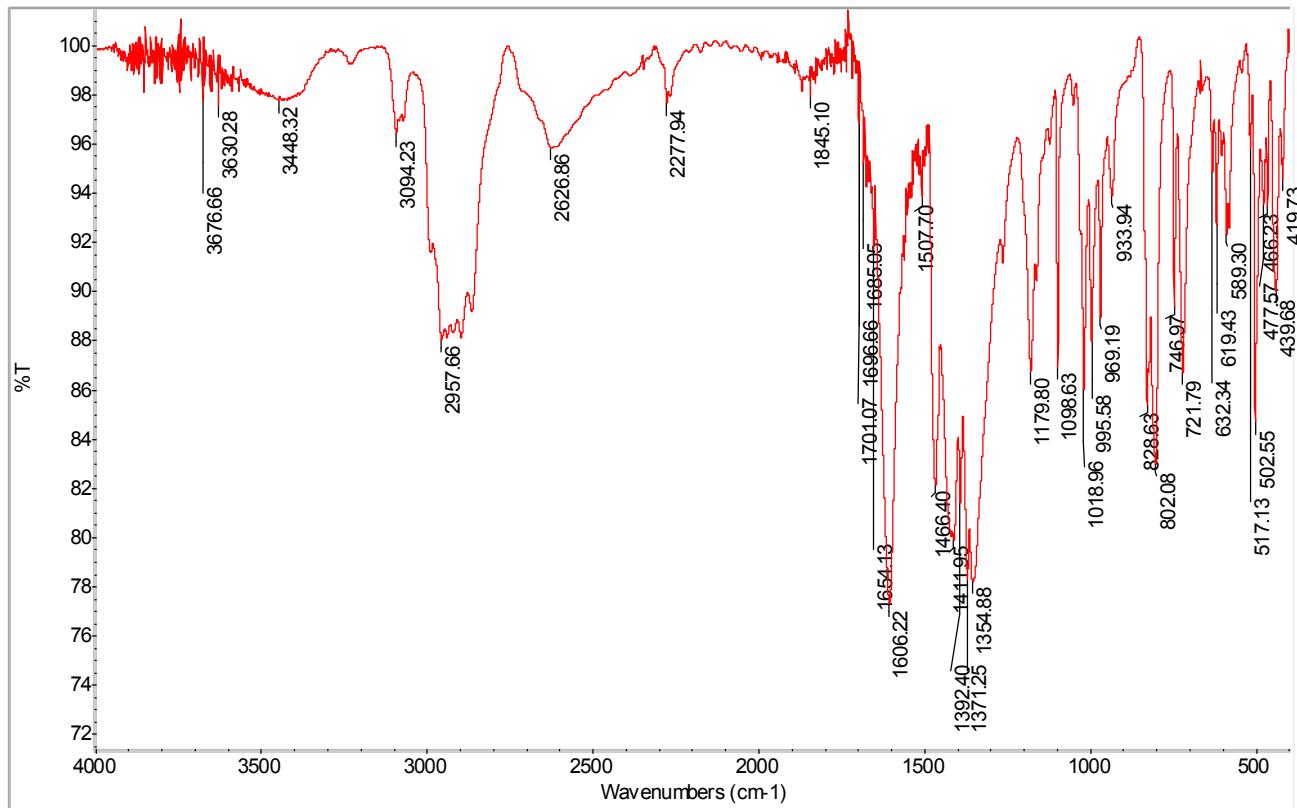


Figure S45. FTIR spectrum of **5** in KBr pellet.



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Table S16. DFT-optimized geometries (Cartesian coordinates) and electronic energies in toluene or in tetrahydrofuran.

84	B3LYP _{syn} -Pd(¹⁸ PC ⁸⁰ P)(n ^{1,2} -BH ₄) (syn-1) toluene	E= -2028.24312206 Ha									
46	0.113127000	-0.048335000	-0.175060000	1	1.953253000	-0.214513000	2.299514000	6	2.227346000	-1.887587000	-2.653460000
44	-0.081111000	2.820150000	-0.676694000	1	2.504162000	1.289661000	1.537773000	1	2.480234000	-2.681548000	-3.368475000
15	-2.268094000	-0.961084000	-0.516606000	1	4.942816000	0.266692000	0.571988000	1	2.462827000	-0.928753000	-3.127031000
15	2.462666000	-0.733059000	-0.578866000	1	4.752487000	1.284342000	0.215789000	6	4.523837000	-1.970881000	-1.663177000
6	0.018265000	0.636829000	-1.241873000	1	6.041073000	-1.861974000	1.547471000	1	5.149950000	-0.921981000	-0.814165000
6	1.138838000	1.343151000	-1.792777000	1	1.5451513000	0.353719000	1.488753000	1	1.148105000	-1.924267000	-2.474585000
6	0.637703000	2.346244000	-2.690184000	1	3.127756000	-2.469751000	1.762164000	1	3.078952000	-4.260353000	-1.639261000
1	1.228016000	3.021033000	-3.294506000	5	0.220632000	-2.915452000	1.719593000	1	1.667589000	-3.712581000	-0.710668000
6	-0.797626000	2.275767000	-2.674011000	1	1.020898700	-2.704861000	0.445277000	1	3.252759000	-2.70380000	0.061400000
1	-1.464688000	2.888810000	-3.264000000	1	1.058588000	-1.900115000	2.382117000	6	3.560766000	-0.669371000	1.383366000
6	-1.178179000	1.229372000	-1.766471000	1	1.259967000	-3.513004000	1.887611000	6	2.772895000	0.158044000	2.417992000
6	-0.067581000	3.241340000	1.506278000	1	1.746842000	-3.623374000	1.890257000	1	3.422278000	0.363920000	2.379597000
1	-0.016816000	2.480583000	2.270374000	1	1.890867000	-0.387587000	2.772355000	6	2.256654000	-1.952121000	-2.672778000
6	1.047041000	3.899971000	0.898503000	1	2.441695000	1.119090000	2.007398000	1	2.502703000	-2.754862000	-3.388467000
1	2.088000000	3.734880000	1.132821000	6	4.860792000	0.080959000	1.038759000	1	2.479701000	-0.985825000	3.158818000
6	0.541858000	4.848547000	-0.044265000	1	5.462249000	0.156523000	1.954743000	1	1.169161000	-2.001823000	-2.488093000
1	1.133356000	5.517969000	-0.653476000	1	4.680593000	1.102456000	0.688600000	6	4.576680000	-1.995838000	1.666191000
6	-0.886196000	4.777301000	-0.019933000	15	2.281945000	-1.107569000	-0.037255000	1	4.850922000	-2.669303000	-2.502231000
1	-1.561669000	5.380902000	-0.607842000	15	2.381343000	-0.873067000	-0.085676000	6	1.502648900	-2.030829000	-0.028360000
6	-1.261812000	3.785177000	0.937670000	6	0.027877000	0.464528000	-0.765876000	1	4.444581000	-1.857395000	2.948328000
1	-2.272792000	3.516477000	1.204285000	6	1.089816000	1.185479000	-1.297839000	1	4.586027000	-2.611433000	1.356608000
6	-2.530984000	0.659719000	-1.440699000	6	0.593084000	2.962580000	-2.183602000	1	3.026421000	-2.626268000	2.218719000
1	-3.124022000	0.479606000	-2.341098000	1	1.185758000	2.887316000	-2.769211000	5	0.158872000	-2.930052000	2.139536000
1	-3.114360000	1.334883000	-0.817256000	6	0.839221000	1.061683000	-2.180584000	1	1.065022000	-2.876023000	0.849395000
6	2.547645000	0.912809000	-1.495393000	1	1.505987000	2.739790000	-2.764259000	1	0.095767000	-1.842208000	2.692223000
1	3.072061000	1.636281000	-0.876823000	6	-1.219497000	1.058910000	-1.293289000	1	1.195871000	-3.504331000	2.398229000
6	-2.849696000	-2.304848000	-1.753134000	6	-0.156889000	2.961203000	1.956880000	1	-0.811435000	-3.615862000	2.386211000
1	-2.088712000	-2.059348000	-0.306260400	6	0.967580000	3.624214000	1.378283000	1	3.278586000	-3.778209000	1.777190000
1	-2.327856000	-2.874592000	-3.778209000	1	2.006081000	3.443571000	1.619073000	1	4.482308000	-4.590513000	1.447247000
1	-1.004773000	-0.252212000	-0.9235281000	6	0.482308000	4.590513000	0.447247000	46	0.061299000	-1.199567000	0.310575000
1	-2.374056000	-1.119543000	-3.566423000	1	1.087214000	5.262612000	-0.145890000	44	-0.131157000	-2.661451000	-0.198750000
6	-4.369590000	-2.262782000	-0.2033254000	6	-0.943694000	4.526106000	0.448794000	15	-2.300834000	-1.116588000	-0.039306000
1	-4.600179000	-2.964219000	2.846626000	1	-1.608046000	4.113780000	-0.142098000	15	2.396090000	-0.892897000	-0.085893000
1	-4.709990000	-1.270147000	-2.350673000	6	-1.337877000	5.320997000	1.381522000	6	-0.028026000	0.487844000	-0.751261000
1	-4.954543000	-2.569649000	-1.162259000	1	-2.358450000	3.250269000	1.625545000	6	1.103176000	1.204979000	-1.298239000
6	-2.456303000	-3.709176000	-1.231365000	6	-2.562797000	0.489608000	-0.963947000	6	0.602386000	2.218052000	-2.210199000
1	-2.740389000	-4.456589000	-1.984607000	1	-3.161218000	0.298590000	-1.860178000	1	1.199566000	2.902795000	-2.803670000
1	-2.959610000	3.966746000	-0.297728000	1	-3.144554000	1.673163000	-0.339492000	6	-0.843341000	2.146326000	-2.192549000
1	-1.378976000	-3.789050000	-1.062538000	6	2.487960000	0.768106000	-0.968145000	1	1.512902000	-2.152621000	-2.787644000
6	-3.425531000	-0.823620000	1.006341000	1	2.977123000	1.489811000	-0.313713000	6	-1.230913000	1.089138000	-1.284216000
6	-6.268209000	0.101147000	2.005630000	1	3.116684000	0.674383000	-1.858584000	6	-0.122415000	3.067710000	1.975174000
1	-3.317366000	0.254433000	2.887293000	6	-2.844688000	-2.447928000	-1.251913000	1	-0.070936000	2.302562000	2.747522000
1	-2.465042000	1.083688000	1.571668000	6	-2.081379000	-2.200690000	-2.568169000	6	0.999901000	3.730861000	1.362093000
1	-1.738555000	-0.342604000	2.335495000	1	-2.297918000	-3.026590000	-3.258712000	1	2.049664000	3.567931000	1.609801000
6	-4.808096000	-0.199117000	0.690184000	1	-0.999255000	-2.7171012000	-2.405220000	6	0.490798000	4.682285000	0.408591000
1	-5.378570000	-3.138987000	1.626962000	1	-2.379452000	-1.270082000	-3.062201000	1	1.087107000	3.535137000	-0.208143000
1	-5.393816000	-0.796769500	-0.011825000	6	-4.353966000	-2.419719000	-1.537793000	6	-0.947331000	4.607378000	-0.430482000
1	-4.733108000	0.189304000	0.295188000	1	-4.572780000	-3.127311000	-2.349020000	1	-1.628666000	5.211647000	-0.166705000
6	-3.369033000	-2.205547000	1.667494000	1	-4.701351000	-1.431998000	-1.862807000	6	-1.325255000	3.613033000	1.397437000
1	-4.141670000	-2.058116000	2.633016000	1	-4.941675000	-2.287185000	-0.668387000	1	-2.344850000	3.339081000	1.666489000
1	-2.694408000	-2.721337000	1.859465000	6	-2.442295000	-3.837092000	-0.725177000	6	-0.258218800	0.968150000	-1.280922000
1	-4.280870000	-2.853383000	1.061569000	1	-2.718839000	-4.590355000	-1.475418000	1	-3.172127000	0.313867000	-1.880040000
6	3.133879000	-1.843092000	1.204400000	1	-2.944000000	-4.096901000	-0.209802000	1	-3.185656000	1.771816000	-0.340860000
6	2.324724000	-1.816967000	-3.151787000	1	-1.364773000	-3.904024000	-0.554042000	6	2.508844000	0.759820000	-0.995805000
1	2.606792000	-2.610618000	-3.857469000	6	-3.436377000	-0.968088000	1.460596000	1	3.043497000	1.480401000	-0.362455000
1	2.526627000	-0.857635000	-3.640794000	6	-2.695349000	-0.048113000	2.450414000	1	3.111758000	0.639411000	-1.911150000
1	1.247368000	-1.890828000	-2.977031000	1	-3.341189000	0.134659000	3.319080000	6	-2.886064000	-2.466100000	-1.280922000
6	4.637476000	-1.829730000	-2.155939000	1	-2.447497000	0.920538000	2.001730000	6	-2.113877000	-2.215812000	-2.607218000
1	4.909658000	-2.503910000	-2.979489000	1	-1.767007000	-0.509353000	2.800372000	6	-2.348038000	-3.036654000	-3.312885000
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1	0.060435000	11.139758000	8.428242000	6	-4.557755000	11.237637000	14.110716000	6	0.027742000	14.460406000	15.159141000	6	-3.386937000	6.444083000	10.625761000	1	-0.137748000	8.786674000	13.699691000
1	-1.604089000	8.656111000	8.829103000	6	-3.438963000	10.8747107000	15.104499000	6	-0.500153000	11.528740000	15.404923000	6	-4.715563000	6.310934000	8.994841000	1	1.201913000	9.951183000	13.656719000
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1	-3.951598000	10.622959000	16.078568000	6	-0.562605000	14.341733000	13.389542000	8	-1.617760000	6.361421000	11.481699000	1	-1.484119000	7.924794000	7.963395000	
1	-2.763437000	11.654874000	15.258393000	1	-0.650908000	15.420828000	11.347775000	6	-1.541420000	6.039726000	10.600436000	1	-2.340775000	8.678403000	9.326490000	
1	-2.916272000	9.952532000	14.787586000	6	-1.605756000	13.424227000	11.714684000	1	-2.390342000	6.718280000	10.675329000	1	-1.295266000	7.261032000	9.599602000	
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6	-6.070383000	10.516270000	8.404595000	1	-1.473744000	13.991070000	15.554357000	44	0.087194000	13.239114000	13.241643000	1	-0.914592000	9.913496000	8.414372000	
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1	-1.680464000	9.901454000	13.389190000	6	-5.571373000	10.522091000	10.551820000	1	-1.951314000	11.450699000	14.658654000	6	-4.502375000	6.350765000	10.377243000	
1	-2.916056000	8.587134000	11.240664000	1	-1.607944000	10.477732000	9.624198000	6	-1.706341000	13.672646000	14.666569000	6	-4.502375000	6.350765000	10.377243000	
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1	-3.035702000	8.342343000	10.180189000	6	-4.831284000	10.273378000	13.852699000	1	-0.562128000	15.571364000	14.974669000	6	-3.686388000	5.237066000	13.522820000	
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1	-1.110150000	6.681571000	10.521656000	1	-4.277089000	11.914707000	15.936355000	1	-1.488553000	13.956121000	15.627920000	6	-1.418970000	6.745204000	9.373057000	
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9	-3.621970000	5.031413000	16.293533000	1	13.258485000	-1.110599000	2.181657000	6	7.020896000	3.880602000	5.175853000	1	1.652316000	7.776271000	8.906795000	6	2.259735000	-1.755972000	-3.127818000
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163																			
M06	<i>syn-syn-[Pd(^{B6}PC^{R6}P),(μ,η^{12},η^1-BH₄)⁺ (2)</i>			1	15.444210000	-1.777358000	3.615515000	1	7.721838000	3.293966000	4.627026000	6	-1.035393000	8.246279000	8.877805000	1	2.496756000	-0.801467000	-3.614576000
toluene E=	-4026.96229632 Ha			6	14.777791000	-3.892803000	3.296487000	1	6.585885000	4.596213000	4.464942000	1	-1.124165000	8.017818000	7.805082000	1	1.177050000	-1.793644000	-2.948940000
44	12.907870000	-3.183229000	4.266142000	6	13.655249000	-4.425915000	2.585423000	1	7.911500000	4.292371000	-0.507616000	1	-0.831748000	7.307767000	9.405358000	1	4.792294000	-2.547024000	-2.975369000
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15	9.144556000	-1.235797000	4.300603000	5	10.597979000	2.233670000	4.339981000	1	7.141099000	3.518934000	1.969146000	6	0.935734000	8.944098000	13.378788000	1	4.832843000	-0.845845000	-2.478417000
15	12.824089000	0.187190000	6.900262000	1	9.532042000	2.403030000	3.799194000	6	5.861554000	5.365846000	1.927201000	1	1.495746000	8.268185000	14.042188000	6	2.749776000	-3.392599000	-1.321414000
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6	12.751453000	-3.870311000	6.349894000	44	7.809410000	6.383227000	1.600659000	6	6.535456000	6.029288000	-0.186825000	1	3.127747000	7.756034000	12.318513000	6	3.648392000	-0.484489000	0.865384000
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6	11.592364000	-4.313475000	5.623228000	15	11.781183000	4.861713000	1.793760300	1	5.056680000	5.655878000	5.558613000	1	2.879887000	8.131390000	10.606674000	1	3.549787000	0.562089000	2.754311000
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1	8.794579000	-3.595090000	4.905577000	6	8.655442000	7.276414000	3.432040000	1	4.149500000	11.215781000	12.215781000	1	-0.532287000	7.105581000	11.888301000	1	5.561134000	0.338669000	1.394580000
6	8.770407000	-0.973824000	2.462042000	6	8.414252000	3.171516000	2.470665000	15	-3.722332000	11.860377000	12.15781000	1	-2.643237000	8.965813000	11.890878000	1	4.766188000	1.263507000	0.114980000
6	10.147642000	-0.829952000	1.791537000	1	7.760276000	9.166728000	2.601857000	15	0.176348000	9.766000000	10.900447000	1	5.536131000	-0.295208000	-0.258801000	1	1.009740000	4.242270000	1.270533000
1	10.009740000	-0.755922000	2.702542000	6	9.221517000	8.050874000	1.311147000	6	-0.977958000	12.259491000	11.595791000	1	4.002179000	6.432740000	12.481005000	1	2.567640000	1.329870000	1.486400000
1	10.787465000	-1.699185000	1.993414000	1	9.274936000	8.664939000	0.424259000	6	0.373686000	12.477976000	11.192953000	1	1.774247000	6.692108000	10.743145000	6	4.945842000	0.247296000	0.486853000
1	10.672545000	-0.069245000	2.139991000	6	10.975681000	6.142159000	0.709737000	6	0.594467000	13.892295000	11.120278000	1	4.611370000	-1.030582000	-0.138289000	15	-2.527519000	-0.933720000	-0.511795000
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1	7.080320000	-1.553756000	7.141704000	1	14.032380000	6.228652000	0.230530000	1	2.035180000	11.161752000	11.661897000	1	-1.248363000	3.603089000	0.971449000	1	1.531047000	2.855004000	-3.223712000
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1	4.928791000	-0.295276000	-2.22268000	1	-5.038020000	9.534406000	14.246904000	6	-0.833790000	1.298967000	-1.355522000	1	-1.401686000	-1.806058000	3.983555000	1	2.905825000	-4.587356000	-0.492187000
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1	2.167020000	-3.474350000	-1.111072000	6	-0.381064000	10.151222000	8.304262000	6	-0.144251000	2.879116000	1.620688000	1	0.035661000	-2.586580000	6.526148000	6	3.189961000	-0.520660000	1.866561000
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1	3.657521000	-3.795641000	-2.027312000	6	-0.054515000	9.261757000	9.084915000	6	-1.218726000	1.168235000	-1.761175000	1	0.101874000	-2.412475000	5.445592000	1	3.108634000	-3.827089000	1.096304000
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1	-3.085239000	0.213676000	-2.503286000		11	6	-4.426240000	9.657903000	13.781593000	1	-1.551523000	13.039169000	15.756337000	95	M06_Pd(^b PC ^a P)(H) (3)+Py-BH ₃ -toluene
1	-3.113881000	1.081425000	-0.981125000		M06_Py-toluene	1	-5.092516000	9.167618000	14.507398000	6	-3.115624000	13.252027000	12.377043000	E= -228.161169630 Ha	E= -227.45035050 Ha
6	2.524601000	0.887492000	-1.395429000			1	-3.792504000	10.370591000	14.326586000	1	-3.684025000	14.010855000	11.825781000		
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6	-1.880441000	-2.246743000	-3.189105000			1	-6.889938000	11.862372000	12.690693000	1	1.590603000	11.101388000	9.838009000		
1	-2.067614000	-3.067873000	-3.896721000			6	-20.203453000	3.780293000	-0.381147000	1	-5.753632000	12.043837000	14.045733000		
1	-0.800443000	-2.210064000	-2.991158000			6	-20.203453000	2.587676000	-2.458776000	6	-6.082099000	9.287543000	11.947638000		
1	-2.174664000	-1.311226000	-3.681340000			1	-20.203453000	3.780293000	-4.251909000	1	-6.626698000	8.671599000	12.678830000		
6	-4.175761000	-2.539329000	-2.232773000			1	-20.203453000	5.931099000	-2.967915000	1	-5.434241000	8.616724000	11.369477000		
1	-4.344229000	-3.203070000	-3.072813000			1	-20.203453000	5.837644000	-0.488848000	1	-6.824476000	9.735428000	11.274639000		
1	-4.553824000	-1.553175000	-2.534875000			1	-20.203453000	1.227942000	-0.488848000	6	0.694719000	10.330866000	8.735422000		
1	-4.774365000	-2.899534000	-1.387483000			1	-20.203453000	1.629487000	-2.967915000	6	0.308082000	11.811213000	8.564727000		
6	-2.253975000	-3.900331000	-1.375721000		15	1	0.345920000	12.060195000	7.494100000	1	-5.926070000	13.731331000	11.439184000		
1	-2.480914000	-4.654814000	-2.143917000		M06_Py-BH ₃ -toluene	1	0.993400000	12.490779000	9.086496000	1	-6.963693000	12.313510000	11.741173000	E= -248.161169630 Ha	E= -274.810099136 Ha
1	-2.785352000	-4.184752000	-0.460624000			1	-0.710182000	12.011481000	8.921838000	6	-5.643736000	10.482423000	10.199919000		
1	-1.176434000	-3.933717000	-1.168168000			6	2.159553000	10.155473000	8.151800000	1	-6.248312000	10.654060000	9.296421000		
6	-3.364620000	-1.096856000	0.807766000			1	-2.286740000	10.566507000	7.303048000	1	-6.282887000	9.989458000	10.940380000		
6	-2.706914000	-0.104181000	1.782711000			6	-2.457989000	9.101260000	8.279493000	1	-4.829962000	9.789590000	9.947135000		
1	-3.345125000	0.000283000	2.673529000			6	-2.852252000	10.694012000	8.976870000	6	-4.778236000	11.194506000	13.749527000		
1	-2.585395000	0.892967000	1.337665000			7	-20.197193000	3.800068000	-0.402971000	6	-3.659479000	10.752569000	14.771024000		
1	-1.713460000	-0.450336000	2.101796000			6	-20.08158000	2.993585000	-2.449990000	1	-4.094442000	10.553684000	15.702018000		
6	-4.778989000	-0.599673000	0.476786000			1	-2.174366000	9.691297000	8.022618000	1	-2.886027000	11.524366000	14.826194000		
1	-5.348890000	-0.525568000	1.415624000			1	-0.042850000	8.417783000	7.918399000	1	-3.171454000	9.834056000	14.355122000		
1	-5.322347000	-1.281733000	-0.185384000			1	-20.408315000	5.841067000	-0.488588000	6	-5.518867000	12.391970000	14.361373000		
1	-4.777124000	0.398185000	0.020035000			6	0.653678000	7.921153000	12.423083000	1	-5.961815000	12.074086000	15.314483000		
6	-3.455038000	-2.466914000	1.494004000			1	-1.983727000	1.755688000	-0.457228000	1	-6.329486000	12.759379000	13.722507000		
1	-3.936675000	-2.340068000	2.475932000			1	-1.984097000	1.637572000	-2.950637000	1	-4.848827000	13.232420000	14.584099000		
1	-2.462641000	-2.908937000	1.656832700			5	-20.202478000	3.777636000	1.213189000	1	-0.762227000	8.733514000	13.165009000		
1	-4.064226000	-1.374646000	0.917611000			1	-21.192203000	1.535783200	1.3535793000	6	-5.753816000	10.022996000	13.577394000		
6	-3.178747000	-1.965326000	-1.856512000			1	-19.188705000	3.204112000	1.543838000	6	-6.085026000	9.691456000	14.573075000		
6	-2.643039000	-1.689383000	-3.191115000			1	-20.232755000	4.930420000	1.572157000	1	-6.649307000	10.311030000	13.012378000		
1	2.793462000	-2.439629000	-3.924539000		95	1	3.267123000	8.751510000	10.275985000	6	-0.140338000	9.103626000	8.803779000		
1	2.699121000	-0.699562000	-3.601169000		M06_synth-Pd(^b PC ^a P)(n ¹ -BH ₃)(Py-1)+Py-toluene	6	0.972757000	7.149662000	10.069930000	6	-0.568541000	10.426490000	8.142185000	E= -227.44955500 Ha	E= -227.44955500 Ha
1	1.372777100	-1.770783400	-3.092147000			1	1.358405000	12.213665000	10.500828000	1	-0.737942000	10.242972000	7.070895000		
6	4.687742000	-1.781067000	-2.056364000			1	-1.965728000	7.298025000	9.106779000	1	-0.191958000	11.213078000	8.227523000		
1	5.009366000	-2.403177000	-2.905643000			44	-0.692742000	12.007607000	14.071312000	1	-0.102183000	9.895171000	1.15052172000		
1	5.260991000	-2.101152000	-1.178114000			15	-0.431539000	12.2053000	11.573002000	1	-2.590873000	8.866750000	9.886702000		
1	4.954397000	-0.741150000	-2.289034000			5	-0.284798100	7.967723000	10.481558000	1	-1.077488000	8.619877000	11.982645000		
6	2.884035000	-3.420837000	-1.460791000			6	-1.249972000	11.974692000	11.937570000	1	-4.025351000	7.581931000	10.257458000		
1	3.294689000	-4.086918000	-2.234561000			6	-0.297766000	12.797762000	12.515422000	6	-2.624860000	7.627508000	8.510420000		
1	1.802899000	-3.601145000	-1.390111000			6	0.201536000	13.402145000	12.659012000	1	-2.140861000	9.951368000	9.931840000		
1	3.337020000	-3.699777000	-0.502688000			7	-5.523618000	6.146124000	14.322881000	1	-1.345620000	7.951552000	7.431465000		
6	3.522373500	-0.628406000	0.959516000			6	-1.135151000	13.825665000	12.951763000	6	-1.441384000	11.587289000	11.614100000		
6	2.654622000	-1.374624000	1.974567000			15	-1.434837000	6.095210000	12.505611000	1	-4.539929000	8.605210000	12.795674000		
1	3.219537000	-0.267866000	2.911091000			6	-1.249972000	12.797762000	12.515422000	6	-3.134646000	6.125623000	14.130282000		
1	1.729837000	-0.411721000	2.199894000			6	-1.028219000	10.202711000	15.277785000	1	-2.272231000	10.471547000	1.150500000		
1	2.367573000	1.134981000	1.610636000			1	-1.454367000	9.291206000	14.875747000	6	-4.150485000	6.218885000	16.28623000		
6	4.829636000	-1.438960000	0.733587000			6	0.357159000	10.537682000	15.318973000	1	-4.097514000	6.235655000	17.369728000		
1	5.356258000	-2.229589000	1.696584000			1	-1.747950000	9.203590000	14.968919000	6	-5.381094000	6.193202000	15.647040000		
1	4.655350000	-0.165106000	0.371646000			6	-0.486130000	11.822098000	15.920325000	1	-6.300064000	6.215686000	16.231353000		
1	5.501072000	-0.361172000	0.031032000			1	-1.413002000	12.349855000	16.101143000	6	-3.001896000	6.182662000	15.507961000		
6	3.841285000	-0.210450000	1.538148000			6	-0.819175000	12.284754000	16.252045000	1	-2.019199000	6.200135000	15.970808000		
1	4.298080000	-1.884879000	2.531346000		95	1	-1.057799000	13.224519000	16.731697000	1	-2.272276000	13.227762000	11.972975000		
1	4.556780000	-2.564714000	0.915162000		M06_synth-Pd(^b PC ^a P)(n ¹ -BH ₃)+Py_TS _{SH₂} -toluene	6	-1.727200000	10.755853000	12.327762000	1	-1.125584000	11.899305000	15.845645000	E= -227.41761969 Ha	E= -227.41761969 Ha
1	2.935327000	-2.623348000	1.657472000			1	-2.825321000	12.276846000	12.523542000	1	-0.190958000	7.086815000	12.977311000		
5	-0.264453000	0.645089000	4.706520000			6	-6.321621000	12.723570000	12.1032						

1	2.408349000	7.875769000	10.694085000	6	-0.536305000	-0.255825000	6.434091000	6	4.742729000	19.197541000	1.870676000
6	-0.050655000	6.797716000	11.576174000	6	0.592414000	-1.182212000	5.947164000	6	6.250364000	19.483986000	1.903861000
1	0.268591000	6.054846000	12.323511000	1	0.462717000	-0.628959000	5.572789000	1	6.519046000	20.040635000	0.993401000
1	0.254079000	6.425088000	10.590458000	1	0.208229000	-1.794971000	5.118708000	1	6.853129000	18.566607000	1.911201000
1	-1.146687000	6.859467000	11.604293000	1	0.933313000	-1.861709000	6.738733000	1	6.545942000	20.096438000	2.762146000
1	-3.210303000	8.994926000	11.646875000	6	-0.910081000	0.705006000	5.299821000	6	4.410885000	18.384157000	0.605761000
5	-2.902663000	4.663904000	9.572855000	1	-1.803550000	1.294441000	5.537666000	1	4.790961000	18.924478000	-0.274086000
1	-3.494166000	5.556182000	9.013692000	1	-1.135547000	0.117340000	4.397115000	1	3.326354000	18.252133000	0.486132000
1	-1.707043000	4.693068000	9.358748000	1	-0.092036000	1.394066000	5.048656000	1	4.881373000	17.391199000	0.622177000
1	-3.366826000	3.568052000	9.347223000	6	-1.743973000	-1.134733000	6.797435000	6	3.967815000	20.522131000	1.814300000
7	-3.069305000	4.909577000	11.158927000	1	-1.491435000	-1.856656000	7.585171000	1	4.183516000	21.016239000	0.855616000
6	-3.594597000	6.039214000	11.644771000	1	-2.060465000	-1.691967000	5.902969000	1	4.266258000	21.210484000	2.614804000
1	-3.917718800	6.773502000	10.913727000	1	-2.601788000	-0.556877000	7.149321000	1	2.881914000	20.366834000	1.869690000
6	-3.689358000	6.275528000	13.005445000	6	-0.917128000	2.139146000	8.428879000	6	-0.521829000	15.614328000	4.957129000
1	-4.097616000	7.220707000	13.345437000	6	-0.408095000	2.625332000	9.819806000	6	0.5422838000	15.413020000	6.050397000
6	-2.684443500	4.141267000	13.377576000	1	-0.613717000	1.790262000	10.574495000	1	0.091096000	15.647772000	7.025202000
1	-2.308941000	3.361557000	14.030044000	1	-0.925517000	3.485117000	10.121608000	1	1.400634000	16.082744000	5.906827000
6	-2.619932000	3.975062000	12.012094000	1	0.672447000	2.765174000	9.819639000	1	0.915625000	14.382382000	6.098216000
1	-2.201389000	3.085524000	11.552539000	6	-0.737646000	3.307783000	7.449618000	6	-1.671690000	14.628765000	5.194136000
6	-3.230333000	5.313891000	13.886850000	1	0.310689000	1.861883000	7.353705000	1	-2.027891000	14.741638000	6.229273000
1	-3.289996000	5.247304000	14.958663000	1	-1.298101000	4.172709000	7.835161000	1	-1.360897000	13.582821000	5.065028000
83				1	-1.123763000	3.058686000	6.449322000	1	-2.524133000	14.825134000	4.532910000
M06	<i>syn-Pd(^{II}PC^{II}P)(η⁻OC(O)H)</i>	(4)	toluene	6	-2.406944000	1.786266000	8.543057000	6	-1.038586000	17.058005000	5.066982000
E=	-2189.21569342	Ha		1	-2.860901000	1.582796000	7.564802000	1	-1.471629000	17.203545000	6.067978000
44	3.776708000	0.822137000	10.1063707000	1	-2.938951000	2.642115000	9.893793000	1	-1.815421000	17.287982000	4.329871000
46	0.472661000	-0.976490000	9.720315000	1	-2.572549000	0.920535000	9.197612000	1	-0.229428000	17.786747000	4.930207000
15	1.552949000	-2.688904300	10.963920000	1	-2.097350000	8.004075000	11.219527000	6	-0.860159000	15.139919000	1.873045000
15	0.082153000	0.605216000	7.996047000	1	-3.152592000	-1.154779000	11.415261000	6	0.007193000	15.207694000	0.601918000
6	2.343507000	-0.612178000	9.188023000	8	-1.580063000	-1.444716000	10.228406000	1	-0.632242000	15.019076000	-0.273448000
6	3.498989000	-1.308083000	9.663145000	8	-1.590360000	0.005145000	11.949060000	1	0.804369000	14.451134000	0.606814000
6	4.636973000	-0.808279000	8.951437000	8	-0.267063000	10.190333000	1.397314000	1	0.469167000	16.197995000	0.483082000
84				6	-2.406944000	1.786266000	8.543057000	6	-1.598277000	13.794114000	1.907703000
M06	<i>syn-Pd(^{II}PC^{II}P)(η⁻OC(O)OH)</i>	(5)	toluene	1	-2.220770000	13.715907000	1.003694000				
E=	-2264.462344160	Ha		46	2.040527000	17.033219000	3.029469000	1	-2.261790000	13.696439000	2.773438000
1	4.8103377000	0.778913000	7.374269000	44	1.6162146000	14.115643000	1.933595000	1	-0.91762000	12.938304000	1.902219000
6	2.769271000	0.351554000	8.218813000	15	4.123767000	18.139651000	3.299082000	6	-1.880272000	16.286783000	1.826440000
6	2.953781000	1.930181000	11.820539000	15	0.344678000	15.395604000	3.295518000	1	-2.426211000	16.232791000	0.873028000
6	4.118280000	1.223114000	12.240338000	8	0.785573000	18.763601000	2.679727000	1	-1.396746000	17.271683000	1.876376000
1	4.159518000	0.459091000	13.006070000	8	0.925817000	18.558737000	0.441561000	1	-2.619219000	16.217581000	2.634551000
6	5.228676000	1.713839000	11.495255000	8	-0.267063000	10.190333000	1.397314000	6	0.543650000	19.085457000	1.485606000
1	6.254674000	1.384318000	11.590236000	1	-0.385259000	20.344888000	0.453150000				
6	4.752743000	2.726050000	10.613313000	6	3.177814000	15.463500000	3.407579000				
1	5.353527000	3.298984000	9.919682000	6	4.596699000	15.483469000	3.593131000				
6	3.349068000	2.858366000	10.813908000	6	5.011774000	14.157472000	3.937754000				
1	2.699573000	3.556147000	10.300910000	1	6.020563000	13.841936000	4.175308000				
6	3.378044000	-3.284851000	10.696339000	6	3.852632000	13.314224000	3.933344000				
1	3.832989000	-2.085068000	11.644577000	1	3.841433000	12.256682000	4.167136000				
1	3.874412000	-3.315306000	10.395421000	6	2.720544000	14.119063000	3.587038000				
6	1.776681000	1.242804000	7.538074000	6	3.773337000	14.657025000	-0.162285000				
1	1.896869000	1.258601000	6.448337000	1	3.161304000	15.500521000	-0.456099000				
1	1.869713000	2.278896000	7.875882000	6	5.179762000	14.679176000	0.069769000				
6	1.2380972000	-4.375078000	10.1759246000	1	5.829994000	15.538151000	-0.035009000				
6	1.921684000	-5.539382000	10.901318000	6	5.590696000	13.366815000	0.439751000				
1	3.000041000	-5.377415000	11.036344000	1	6.601203000	13.056359000	0.672114000				
1	1.798079000	-6.453568000	10.301274000	6	4.438825000	12.529042000	0.436823000				
1	1.471885000	-5.729916000	11.883062000	1	4.420194000	11.472237000	0.667286000				
6	1.774440000	-4.294131000	8.374374000	6	3.179620000	13.325466000	0.065341000				
1	1.296783000	-3.478222000	8.176356000	1	2.300673000	12.971609000	-0.041550000				
1	1.5402474000	-5.237741000	8.220743000	6	5.371459000	16.759047000	3.466749000				
1	2.861562000	-4.152697000	8.688586000	1	6.018798000	16.751231000	2.584896000				
6	-0.276050000	-4.628484000	10.093742000	1	6.022455000	16.944554000	4.329530000				
1	-0.733555000	-4.769594000	11.078990000	6	1.268027000	13.778654000	3.451277000				
1	-0.449015000	-5.546029000	9.511511000	1	0.885266000	13.203200000	4.302895000				
1	-0.800031000	-3.799807000	9.599913000	1	1.079007000	13.175257000	2.558688000				
6	1.330468000	-2.706494000	12.834168000	6	4.189819000	19.036555000	4.957283000				
6	2.414677000	-3.486981000	13.590772000	6	5.494437000	19.804628000	5.197671000				
1	2.427181000	-4.552396000	13.336896000	1	5.502264000	20.173750000	6.234385000				
1	2.210377000	-3.407118000	14.669162000	1	5.583987000	20.677270000	4.539739000				
1	3.419976000	-3.079793000	13.422376000	1	6.384577000	19.173864000	5.065421000				
6	1.388332000	-1.226044000	13.254494000	6	4.029999000	17.966770000	6.052816000				
1	2.329782000	-0.748938000	12.948027000	1	3.954175000	18.473865000	7.025749000				
1	1.318133000	-1.165111000	14.350827000	1	4.881619000	17.276731000	6.104179000				
1	0.554369000	-0.655311000	12.822237000	1	3.116283000	17.375518000	5.908849000				
6	-0.052034000	-3.263629000	13.202887000	6	2.991345000	19.993597000					