

Supporting information for:

**Starch-Regulated Copper-Terephthalic Acid as pH/Hydrogen Peroxide
Simultaneous-Responsive Fluorescence Probes for Lysosomes Imaging**

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Supporting Information Figures

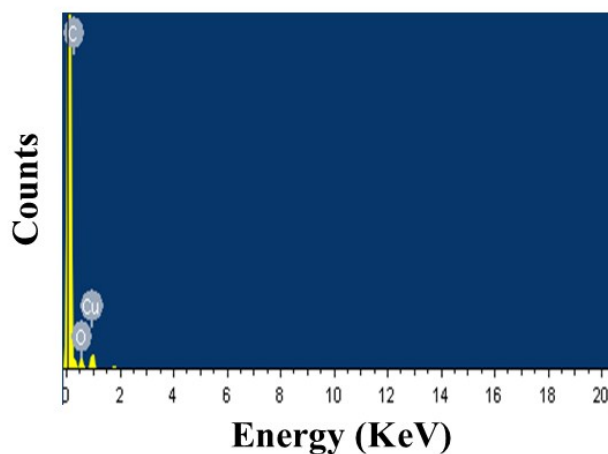


Figure S1. EDX spectrum of CuBDC-2.

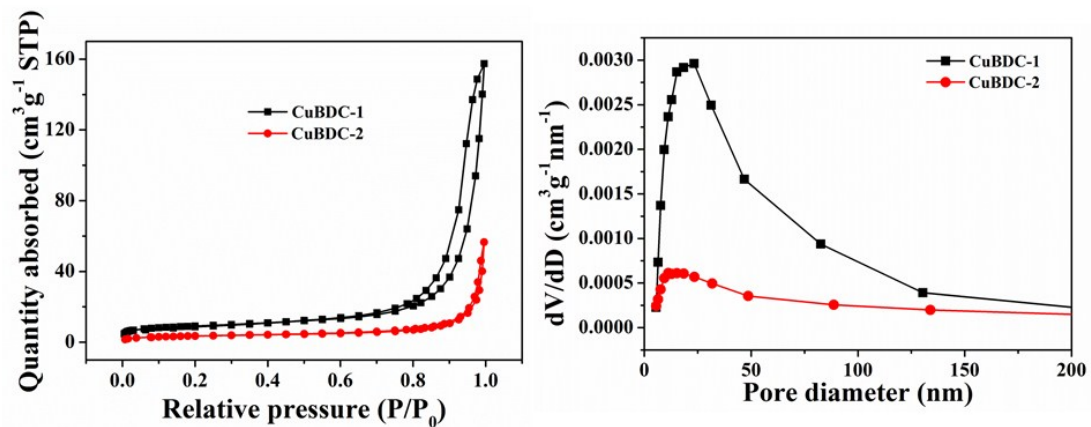


Figure S2. (a) N_2 adsorption and desorption isotherm and (b) BJH pore distribution of the CuBDC-1 and CuBDC-2.

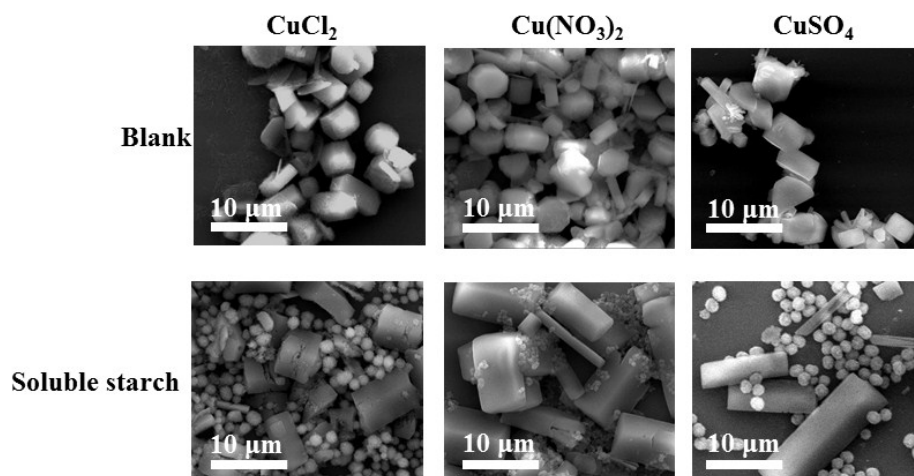


Figure S3. SEM images of CuBDC synthesized using different copper source with or without soluble starch.

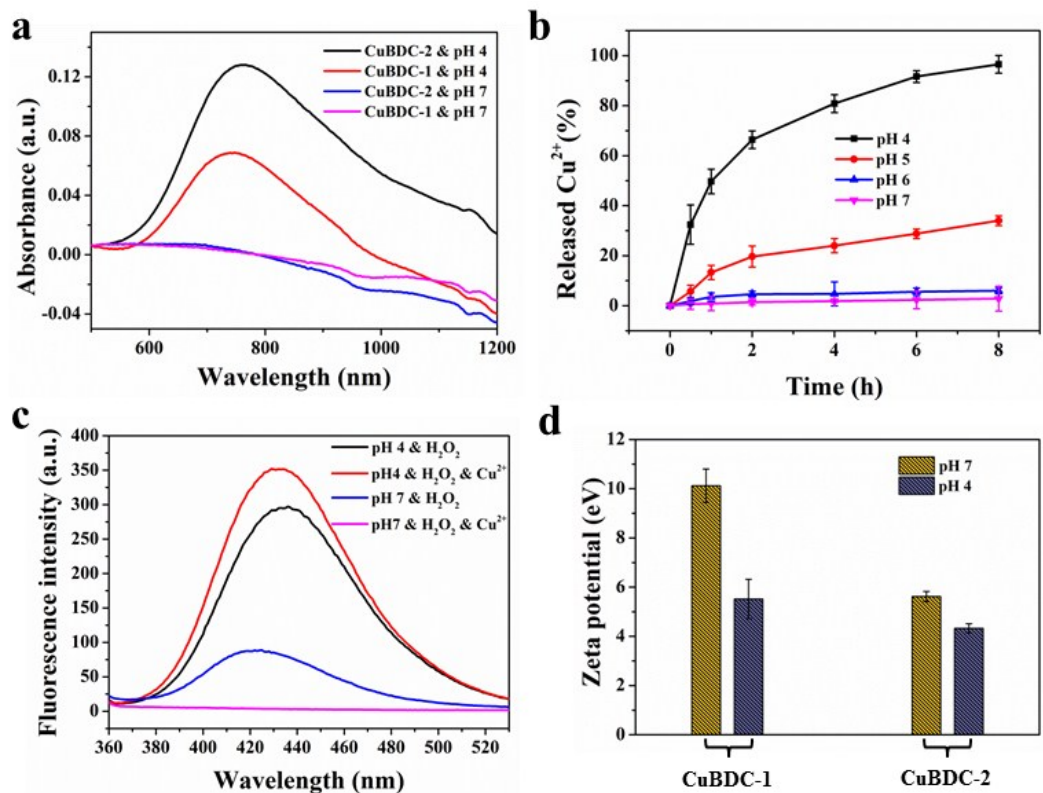


Figure S4. (a) UV-Vis spectra of supernatants withdrawn from the solution containing CuBDC-1 or CuBDC-2 at pH~4 or pH~7 after 4 h of dispersion. (b) Release of Cu²⁺ ions from CuBDC-2 in solutions at pH 4, 5, 6 and 7, respectively. (c) Fluorescence spectra of solution containing CuBDC-2 and H₂O₂ (1.5 mM) at pH 7 or pH 4 with or without extra Cu²⁺ added. (d) Zeta potential of CuBDC-1 and CuBDC-2 dispersed in solution at pH 7 or pH 4.

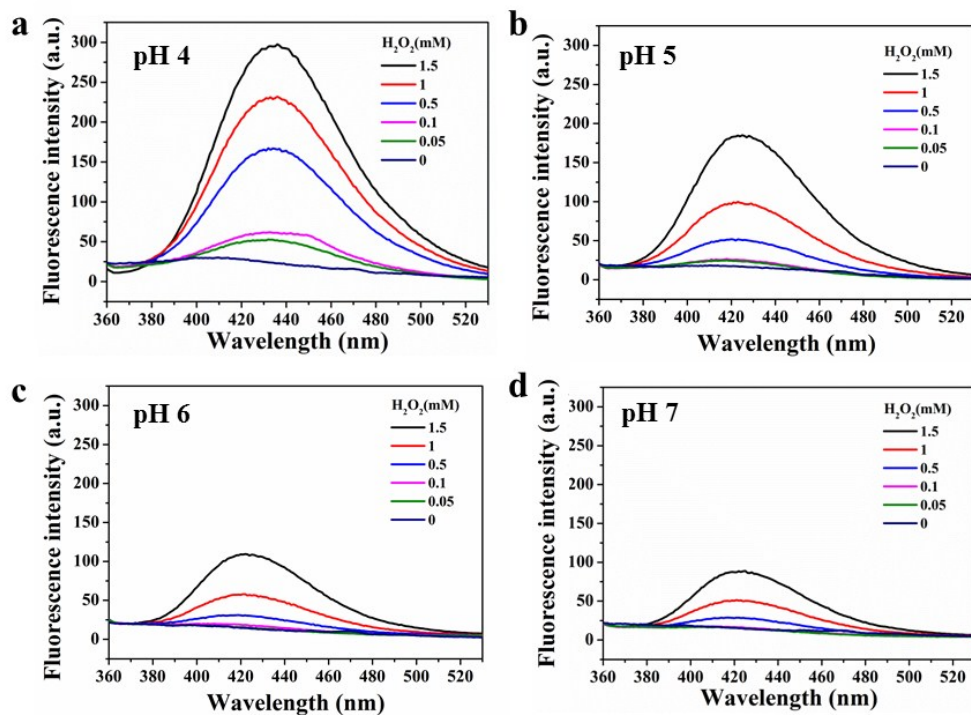


Figure S5. Fluorescence spectra of CuBDC-2 treated with different concentrations of H_2O_2 at pH~4 (a), pH~5 (b), pH~6 (c), pH~7 (d) for 4 h.

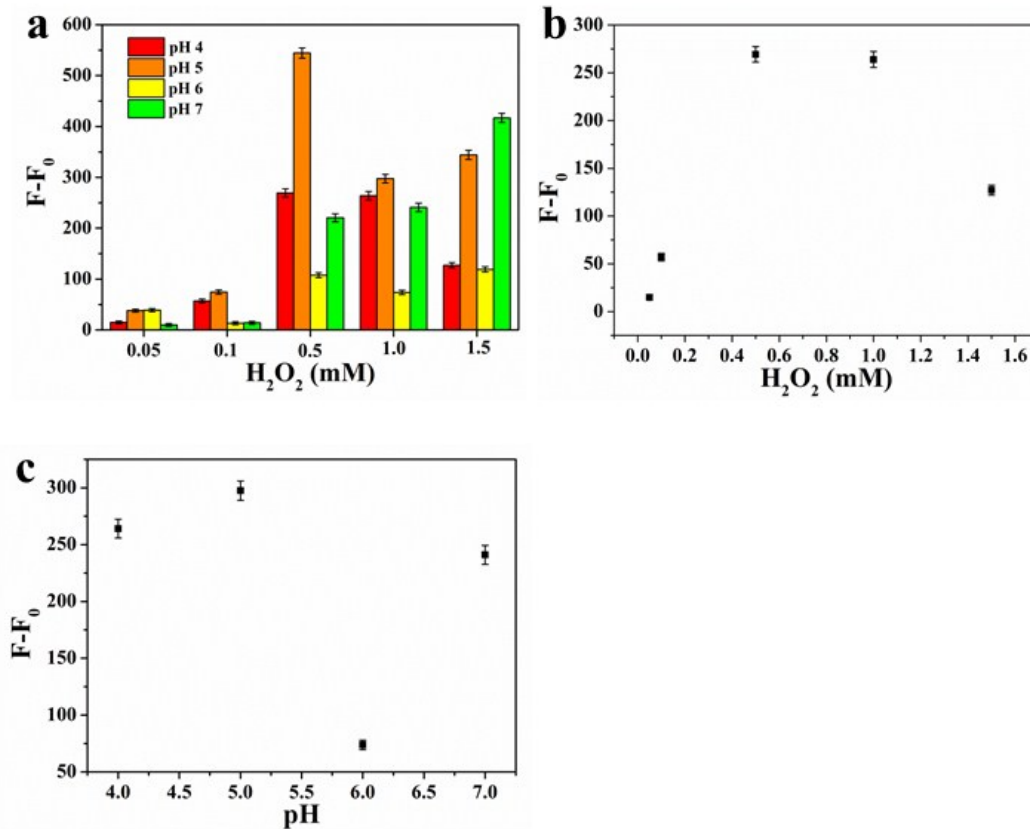


Figure S6. (a) Fluorescence intensity change of mixture of Cu^{2+} and TA treated with

different concentrations of H_2O_2 and different pH values for 4 h. F indicated the fluorescence intensity of solution containing mixture of Cu^{2+} and TA and different concentrations of H_2O_2 at different pH values for 4 h, F_0 indicated the fluorescence intensity of distilled water containing mixture of Cu^{2+} and TA. (b) Relationship between the fluorescence intensity change and H_2O_2 concentration in solutions containing mixture of Cu^{2+} and TA at $\text{pH}\sim 4$ for 4 h. F indicated the fluorescence intensity of solution containing mixture of Cu^{2+} and TA and different concentrations of H_2O_2 at $\text{pH}\sim 4$ for 4 h, F_0 indicated the fluorescence intensity of distilled water containing mixture of Cu^{2+} and TA. (c) Relationship between the fluorescence intensity change and pH values in solution containing mixture of Cu^{2+} and TA and 1 mM H_2O_2 for 4 h. F indicated the fluorescence intensity of solution containing mixture of Cu^{2+} and TA and 1 mM H_2O_2 at different pH values for 4 h, F_0 indicated the fluorescence intensity of distilled water containing mixture of Cu^{2+} and TA.

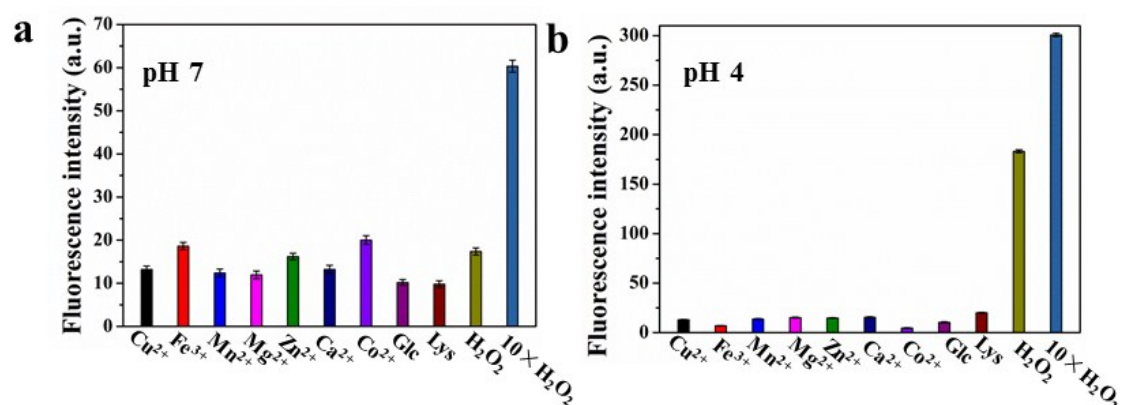


Figure S7. Fluorescence intensity of CuBDC-2 incubated with physiologically important metal ions and bio-molecules including Fe^{3+} , Cu^{2+} , Mn^{2+} , Zn^{2+} , Ca^{2+} , Mg^{2+} , Co^{2+} , glucose and L-lysine at $\text{pH}\sim 4$ or $\text{pH}\sim 7$.

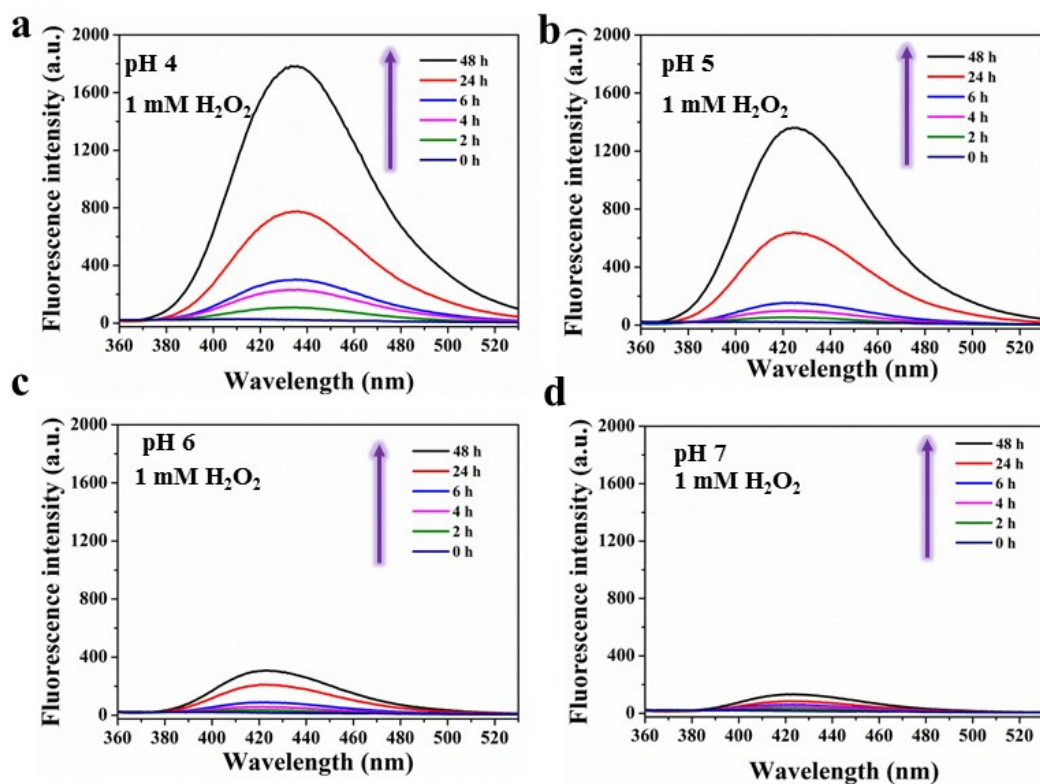


Figure S8. Fluorescence spectra of CuBDC-2 treated with 1mM H₂O₂ at pH~4 (a), pH~5 (b), pH~6 (c) or at pH~7 (d) over 48 h.

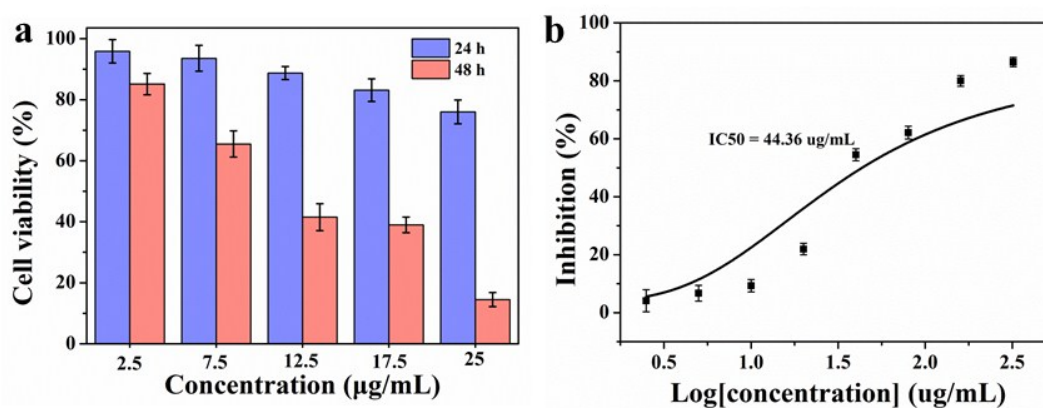


Figure S9. (a) Cell viability of HeLa cells treated with CuBDC-2 at 2.5, 7.5, 12.5, 17.5, 25 µg/mL for 24 h or 48 h. (b) Growth inhibition of HeLa cells by different concentrations of CuBDC-2. IC₅₀ was determined by sigmoidal curve fitting.

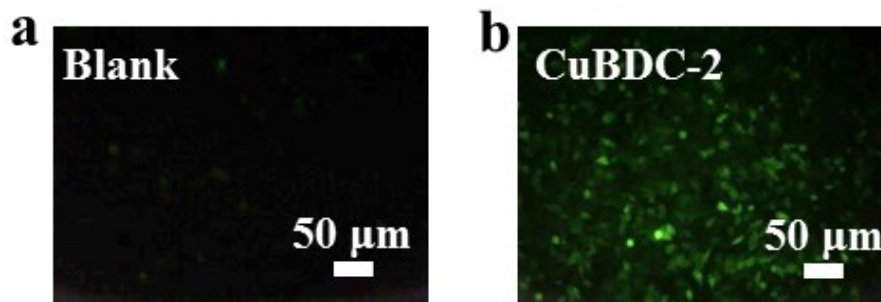


Figure S10. Fluorescence images of HeLa cells after co-incubation with CuBDC-2 for 4 h and then stained with ROS fluorescence probe DCFH-DA.

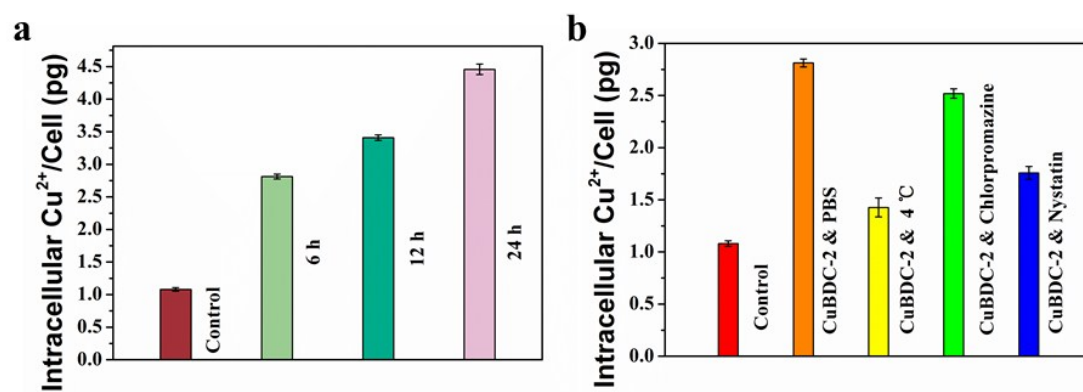


Figure S11. (a) Cell internalization of Cu²⁺ after incubated with CuBDC-2 for 6 h, 12 h or 24 h. (b) Cell internalization of Cu²⁺ after pretreated with different endocytosis inhibitors and then incubated with CuBDC-2 for 6 h.

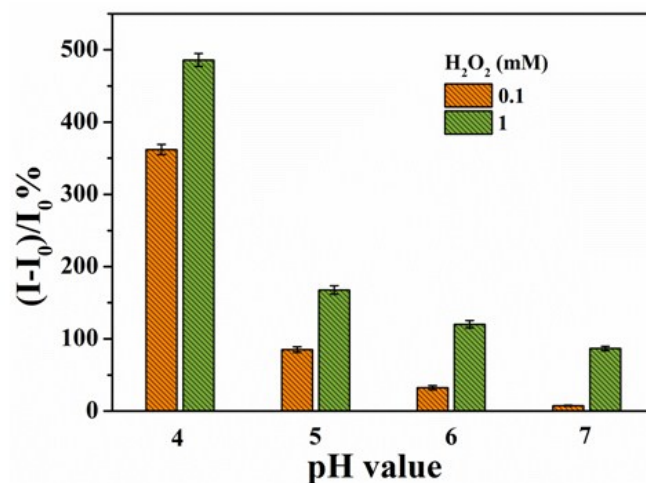


Figure S12. Fluorescence intensity of HeLa cells incubated with CuBDC-2 for 24 h and then treated with buffer solution at different pH values (pH~4, pH~5, pH~6 or pH~7), containing H_2O_2 (0.1 mM or 1mM) and nigericin. I indicated the fluorescence intensity of cells incubated with CuBDC-2 and treated with specific pH value and H_2O_2 concentration, I_0 indicated the fluorescence intensity of cells only incubated with CuBDC-2.