

Electronic Supporting Information

Highly Soluble Fluorine Containing Cu(I) AlkylPyrPhos TADF Complexes

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1. General Information

Column Chromatography. The phosphine ligands described in this study were purified *via* column chromatography. For the stationary phase of the column silica gel 60 (SiO_2 , 0.040×0.063 mm, 260 – 400 mesh ASTM) and sea sand (annealed and purified with hydrochloric acid), purchased from *Merck* respectively, were used. The solvents used for the column chromatography were purchased from *Fisher* in HPLC-grade and the described mixtures were prepared of the corresponding volumes of the solvents. Fractions were collected manually. Product control was performed *via* thin layer chromatography.

Thin Layer Chromatography (TLC). TLC aluminum plates with a fluorescence indicator, purchased from *Merck* (art. number 105554, silica gel 60 F_{254} , thickness 0.2 mm), were used for the analytical thin layer chromatography. The detection on the TLC plates was carried out *via* fluorescence quenching under UV light ($\lambda = 254$ nm) with a UV-lamp from *VWR*.

Nuclear Magnetic Resonance Spectroscopy (NMR). All NMR spectra of the AlkylPyrPhos ligands and corresponding copper complexes, described in this study, were recorded on a *Bruker* Avance 400 NMR instrument with 400 MHz for ^1H NMR, 101 MHz for ^{13}C NMR, 162 MHz for ^{31}P NMR and 376 MHz for ^{19}F NMR spectra. The NMR spectra were measured at room temperature in deuterated solvents commercially acquired from *Eurisotop*. The chemical shift δ is displayed in parts per million [ppm] and the ^1H NMR and ^{13}C NMR resonances of the solvents themselves were used as references: chloroform-d₁ (CDCl_3): 7.26 ppm for the ^1H NMR and 77.2 ppm for the ^{13}C NMR and dimethyl sulfoxide-d₆ (DMSO-d_6): 2.50 ppm for the ^1H NMR and 39.5 ppm for the ^{13}C NMR. Evaluation of the resonances was done according to first order spectra. For the characterization of centrosymmetrical signals, the signal's median point was chosen, for multiplets the signal range. For the description of couplings, the following abbreviations were used: d = doublet, t = triplet, dd = doublet of doublet and m = multiplet. Coupling constants ' J ' are given in Hertz [Hz], starting with the largest value. Coupling constants are given with their respective number of bindings and binding partners, as far as they could be determined, written as index of the coupling constants ($^XJ_{HH}$ and $^XJ_{CP}$). Resonances of the ^1H NMR and ^{13}C NMR spectra were assigned with *H-C-Heteronuclear Single Quantum Coherence HSQC* NMR spectra and *H-C-heteronuclear multiple bond correlation* NMR spectra HMBC and the help of *distortionless polarization by polarization transfer* NMR spectra DEPT90 and DEPT135.

Mass Spectrometry (MS). The mass spectrometrical identifications were performed either with the *electron ionization* (EI) or with the *fast atom bombardment* (FAB) mass spectrometry. An instrument by *Finnigan*, model MAT 95 (70 eV), was used. For FAB measurements 3-nitrobenzyl alcohol (3-NBA) served as internal standard. For the interpretation of the spectra the molecular peak $[\text{M}]^+$, the protonated molecular peak $[\text{M}+\text{H}]^+$ and characteristic fragment peaks are indicated with their mass to charge ratio (m/z) and their intensity in percent, relative to the base peak (100%). For the high resolution mass spectrometry the following abbreviation was used: calcd = calculated.

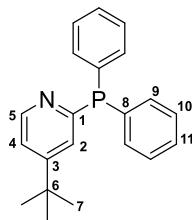
Infrared Spectroscopy (IR). Infrared spectra were recorded on a *Bruker*, model IFS 88. The powders were measured by the *attenuated total reflection* (ATR) method. The absorption is given in wave numbers $\bar{\nu}$ [cm^{-1}] and was measured in the range from 3600 cm^{-1} to 400 cm^{-1} . The characterization of absorption bands was done according to the transmission strength T with the following abbreviations: vs = very strong (0–10% T), s = strong (11–40% T), m = medium (41–70% T), w = weak (71–90% T) and vw = very weak (91–100 % T).

Elemental analysis. The ratio of carbon (C), hydrogen (H) and nitrogen (N) in mass percentage of the synthesized compounds in this work were determined with an *Elementar*, model vario MICRO cube. The weight of the samples taken was measured with the balance Sartorius M2P. Acetanilide served as standard. The values for the elemental analysis were given with the following designations: Anal. calcd = calculated and found = actual values for the mass percentages of the CHN-analysis.

Melting point (mp). The melting curves of the compounds, synthesized in this study, were measured on a *Laboratory Devices Inc.*, model Mel-Temp II and were not corrected.

2. Experimental

Ligands



(2-(Diphenylphosphino)-4-tert-butylpyridine (**3**)).

The title ligand **3** was synthesized according to the same literature procedure as ligand **2**.¹ Colorless, microcrystalline solid (69% yield). The complete numeration for the exact NMR classification of ligand **3** is given on the left.

R_f (CH/EE 4/1) = 0.57; **R_f** (CH/EE 6/1) = 0.39; **R_f** (CH/EE 10/1) = 0.34. – **¹H NMR** (400 MHz, DMSO-d₆) δ [ppm] = 8.58 (d, ³J_{HH} = 5.2 Hz, 1H, H₅), 7.42 – 7.39 (m, 6H, H₁₀, H₁₁), 7.36 – 7.30 (m, 5H, H₄, H₉), 7.08 – 7.07 (m, 1H, H₂), 1.13 (s, 9H, H₇). –

¹³C NMR (101 MHz, DMSO-d₆) δ [ppm] = 162.2 (d, ¹J_{CP} = 3.9 Hz, 1C, C_{q1}), 159.1 (d, ³J_{CP} = 3.5 Hz, 1C, C_{q3}), 150.1 (d, ³J_{CP} = 11.9 Hz, 1C, C₅), 136.2 (d, ¹J_{CP} = 10.9 Hz, 2C, C₈), 133.7 (d, ²J_{CP} = 19.7 Hz, 4C, C₉), 129.1 (s, 2C, C₁₁), 128.6 (d, ³J_{CP} = 7.1 Hz, 4C, C₁₀), 124.6 (d, ²J_{CP} = 21.0 Hz, 1C, C₂), 119.7 (s, 1C, C₄), 34.4 (s, 1C, C_{q6}), 29.9 (s, 3C, C₇). – **³¹P NMR** (162 MHz, DMSO-d₆) δ [ppm] = -4.61 (s, 1P). – **MS** (EI, 70 eV) m/z [%] = 320/319/318 (14/68/62) [M]⁺, 211 (50), 196 (100). – **HRMS** (C₂₁H₂₂NP) calcd: 319.1490; found: 319.1491. – **IR** (ATR) $\tilde{\nu}$ [cm⁻¹] = 3065 (vw), 3044 (vw), 2964 (w), 2930 (w), 2866 (w), 1578 (m), 1530 (w), 1477 (w), 1454 (w), 1434 (m), 1377 (m), 1309 (w), 1266 (w), 1152 (w), 1092 (w), 1068 (w), 1028 (w), 988 (w), 846 (m), 743 (m), 696 (s), 633 (w), 495 (m), 469 (m), 429 (w), 401 (w). – **Anal.** calcd for C₂₁H₂₂NP (319.2): C 78.97, H 6.94, N 4.39; found: C 79.06, H 6.92, N 4.39. – **mp** = 111–113 °C. A molecular structure of the ligand was obtained.

Cu(I) Complexes

[*(2-(Diphenylphosphino)pyridine)(tris(4-trifluoromethylphenyl)phosphine)Cu₂I₂*] (**Cu-1c**).

The title compound was synthesized according to the general procedure. Complex **Cu-1c** could not be precipitated due to the high solubility in every kind of solvent. Therefore the solvent was removed completely and the obtained solid was washed with very small amounts of *n*-pentane and diethyl ether. Yellow powder (96% yield).

¹H NMR (400 MHz, DMSO-d₆) δ [ppm] = 8.73 (bs, 1H, H_{Pyr}), 7.96 (bs, 1H, H_{Pyr}), 7.84 – 7.27 (m, 36H). – **¹³C NMR** (101 MHz, DMSO-d₆) δ [ppm] = 137.1 (d, *J* = 24.0 Hz), 134.5 (d, *J* = 14.0 Hz), 133.3 (d, *J* = 13.3 Hz), 131.8 (d, *J* = 26.1 Hz), 131.0 (s), 130.5 (d, *J* = 32.1 Hz), 130.1 (d, *J* = 7.4 Hz), 128.6 (d, *J* = 8.0 Hz), 127.9 (s), 125.5 (bd, *J* = 4.3 Hz), 125.1 (s), 122.4 (s), 119.7 (s). – **³¹P NMR** (162 MHz, CDCl₃) δ [ppm] = -4.71 (bs, 1P, P_{PyrPhos}), -12.65 (bs, 2P, P_{(CF₃-Ph)₃P}). – **¹⁹F NMR** (376 MHz, DMSO-d₆) δ [ppm] = -66.08 (s, 18F, F_{(CF₃-Ph)₃P}). – **MS** (FAB, 3-NBA) m/z [%] = 1637 (1) [M+Cu]⁺, 1374 (1) [Cu₃I₂P₂]⁺, 1184 (3) [Cu₂IP₂]⁺, 792 (17) [CuLP]⁺, 705 (75) [Cu₃I₂L]⁺, 589 (21) [CuL₂]⁺, 515 (80) [Cu₂IL]⁺, 467 (16) [P+H]⁺, 326 (100) [CuL]⁺, 264 (19) [L+H]⁺, 185 (14) [PPh₂]⁺. – **IR** (ATR) $\tilde{\nu}$ [cm⁻¹] = 3055 (vw), 1607 (vw), 1436 (vw), 1397 (w), 1319 (m), 1166 (w), 1122 (m), 1059 (m), 1014 (w), 829 (w), 764 (vw), 742 (w), 702 (w), 693 (w), 634 (vw), 597 (w), 518 (w), 411 (w). – **Anal.** Calcd for C₅₉H₃₈Cu₂F₁₈I₂NP₃ (1574.8): C 44.94, H 2.43, N 0.89; found: C 44.81, H 2.16, N 0.95.

[(2-(Diphenylphosphino)pyridine)(tris(4-chlorophenyl)phosphine)₂Cu₂I₂] (Cu-1d).

Complex **Cu-1d** was synthesized according to the general procedure. Pale yellow powder (66% yield).

¹H NMR (400 MHz, DMSO-d₆) δ [ppm] = 8.70 (bs, 1H, H_{Pyr}), 7.92 (t, ³J_{HH} = 8.0 Hz, 1H, H_{Pyr}), 7.66 – 7.30 (m, 36H). – **¹³C NMR** (101 MHz, DMSO-d₆) δ [ppm] = 135.3 (d, J_{CP} = 16.8 Hz), 133.4 (d, J_{CP} = 11.5 Hz), 131.4 (d, J_{CP} = 27.3 Hz), 130.1 (s), 128.8 (d, J_{CP} = 8.6 Hz), 128.6 (d, J_{CP} = 7.5 Hz). – **³¹P NMR** (162 MHz, DMSO-d₆) δ [ppm] = -5.06 (bs, 1P, P_{PyrPhos}), -13.06 (bs, 2P, P_{(Cl-Ph)3P}). – **MS** (FAB, 3-NBA) m/z [%] = 1433 (5) [M+Cu]⁺, 1371 (4) [M+H]⁺, 1243 (7) [M-I]⁺, 1069 (16) [Cu₃I₂LP]⁺, 980 (13) [Cu₂IP₂]⁺, 968 (63) [Cu₃I₂L₂]⁺, 879 (43) [Cu₂ILP]⁺, 806 (20) [Cu₃I₂P]⁺, 790 (29) [CuP₂]⁺. – **IR** (ATR) $\tilde{\nu}$ [cm⁻¹] = 1573 (vw), 1557 (vw), 1478 (w), 1454 (vw), 1433 (vw), 1384 (w), 1183 (vw), 1079 (w), 1012 (w), 817 (w), 767 (vw), 743 (m), 694 (w), 631 (vw), 554 (w), 541 (w), 519 (w), 494 (m), 438 (w), 425 (w), 384 (w). – **Anal.** calcd for C₅₃H₃₈Cl₆Cu₂I₂NP₃ (1370.7): C 46.28, H 2.78, N 1.02; found: C 46.75, H 2.80 N 1.16. A molecular structure of the complex was obtained.

[(4-Methyl-2-(diphenylphosphino)pyridine)(tris(4-chlorophenyl)phosphine)₂Cu₂I₂] (Cu-2d).

The complex was obtained according to the general procedure. Pale yellow powder (55% yield).

¹H NMR (400 MHz, DMSO-d₆) δ [ppm] = 8.66 (bs, 1H, H_{Pyr}), 7.44 – 7.31 (m, 36H), 2.28 (s, 3H, H_{Me}). – **¹³C NMR** (101 MHz, DMSO-d₆) δ [ppm] = 135.3 (d, J_{CP} = 15.6 Hz), 133.3 (m), 131.4 (d, J_{CP} = 27.6 Hz), 130.1 (s) 128.8 (d, J_{CP} = 8.8 Hz), 128.6 (d, J_{CP} = 7.8 Hz), 21.8 (s, 1C, C_{Me}). – **³¹P NMR** (162 MHz, DMSO-d₆) δ [ppm] = -4.73 (bs, 1P, P_{MePyrPhos}), -12.98 (bs, 2P, P_{(Cl-Ph)3P}). – **MS** (FAB, 3-NBA) m/z [%] = 1447 (2) [M+Cu]⁺, 1275 (1) [M-I]⁺, 1083 (2) [Cu₃I₂LP]⁺, 893 (8) [Cu₂ILP]⁺, 790 (5) [CuP₂]⁺, 719 (28) [Cu₃I₂L]⁺, 704 (10) [CuLP]⁺, 529 (56) [Cu₂IL]⁺, 426 (7) [CuP]⁺, 340 (66) [CuL]⁺, 278 (21) [L+H]⁺. – **IR** (ATR) $\tilde{\nu}$ [cm⁻¹] = 2950 (vw), 1573 (w), 1478 (m), 1436 (w), 1385 (w), 1183 (vw), 1097 (w), 1079 (m), 1013 (m), 816 (m), 743 (m), 692 (w), 631 (vw), 556 (w), 541 (m), 519 (m), 493 (m), 464 (w), 436 (w), 383 (w). – **Anal.** calcd for C₅₄H₄₀Cl₆Cu₂I₂NP₃ (1384.7): C 46.68, H 2.90, N 1.01; found: C 46.78, H 3.09, N 1.12. A molecular structure of the complex with two molecules of dichloromethane was obtained.

[(2-(Diphenylphosphino)-4-tert-butylpyridine)(triphenylphosphine)₂Cu₂I₂] (Cu-3a).

Compound **Cu-3a** was synthesized according to the general procedure. Pale yellow powder (83% yield).

¹H NMR (400 MHz, DMSO-d₆) δ [ppm] = 8.53 (bs, 1H, H_{Pyr}), 7.62 (m, 1H, H_{Pyr}), 7.45 – 7.27 (m, 41H), 1.10 (s, 9H, H_{tertBu}). – **¹³C NMR** (101 MHz, DMSO-d₆) δ [ppm] = 133.7 (d, J_{CP} = 13.8 Hz), 133.1 (d, J_{CP} = 27.9 Hz), 129.8 (bs), 128.5 (d, J_{CP} = 8.1 Hz), 34.8 (s, C_{tertBu}), 29.7 (s, C_{tertBu}). – **³¹P NMR** (162 MHz, DMSO-d₆) δ [ppm] = -5.04 (bs, 1P, P_{tertBuPyrPhos}), -9.98 (bs, 2P, P_{PPh3}). – **MS** (FAB, 3-NBA) m/z [%] = 1285 (2) [M+Cu]⁺, 1096 (2) [M-I]⁺, 1023 (4) [Cu₃I₂LP]⁺, 834 (15) [Cu₂ILP]⁺, 761 (26) [Cu₃I₂L]⁺, 644 (56) [CuLP]⁺, 571 (72) [Cu₂IL]⁺, 382 (100) [CuL]⁺, 325 (28) [CuP]⁺, 320 (22) [L+H]⁺, 263 (16) [P+H]⁺. – **IR** (ATR) $\tilde{\nu}$ [cm⁻¹] = 2961 (vw), 1585 (vw), 1479 (w), 1434 (w), 1382 (vw), 1156 (vw), 1093 (w), 1027 (vw), 998 (vw), 844 (vw), 741 (w), 692 (m), 518 (m), 504 (m), 427 (vw). – **Anal.** calcd for C₅₇H₅₂Cl₆Cu₂I₂NP₃ (1223.0): C 55.89, H 4.28, N 1.14; found: C 55.51, H 4.27, N 1.17.

[(2-(Diphenylphosphino)-4-tert-butylpyridine)(tris(4-fluorophenyl)phosphine)₂Cu₂I₂] (Cu-3b).

The complex **Cu-3b** was synthesized according to the general procedure. Pale yellow, microcrystalline solid (57% yield).

¹H NMR (400 MHz, DMSO-d₆) δ [ppm] = 8.50 (bs, 1H, H_{Pyr}), 7.62 (bs, 1H, H_{Pyr}), 7.57 – 7.00 (m, 35H), 1.11 (s, 9H, H_{tertBu}). – **¹³C-NMR** (101 MHz, DMSO-d₆) δ [ppm] = 164.4 (s), 162.0 (s), 136.0 (q, J = 15.5 Hz, J = 8.3 Hz), 133.3 (bd, J = 13.0 Hz), 130.0 (s), 129.0 (d, J = 27.9 Hz), 128.5 (d, J = 7.7 Hz), 115.9 (dd, J = 21.2 Hz, J = 9.3 Hz), 34.9 (s, C_{tertBu}), 29.7 (s, C_{tertBu}). – **³¹P NMR** (162 MHz, DMSO-d₆) δ [ppm] = -4.76 (bs, 1P, P_{tertBuPyrPhos}), -14.47 (bs, 2P, P_{(F-Ph)3P}). – **¹⁹F NMR** (376 MHz, DMSO-d₆) δ [ppm] = -114.82 (s, 6F, F_{(F-Ph)3P}). – **MS** (FAB, 3-NBA) m/z [%] = 1393 (1) [M+Cu]⁺, 1077 (3) [Cu₃I₂LP]⁺, 887 (7) [Cu₂ILP]⁺, 884 (2) [Cu₂IP₂]⁺, 761 (32) [Cu₃I₂L]⁺, 698 (25) [CuLP]⁺, 695 (6) [CuP₂]⁺, 571 (74) [Cu₂IL]⁺, 382 (100) [CuL]⁺, 320 (22) [L+H]⁺, 317 (14) [P+H]⁺. – **IR** (ATR) $\tilde{\nu}$ [cm⁻¹] = 2962 (vw), 1586 (m), 1493 (m), 1434 (w), 1393 (w), 1225 (m), 1158 (m), 1092 (w), 1013 (w), 824 (m), 742 (w), 693 (w), 634 (w), 520 (m), 467 (w), 432 (m). – Anal. calcd for **C₅₇H₄₆Cu₂F₆I₂NP₃** (1330.9): C 51.37, H 3.48, N 1.05; found: C 51.26, H 3.48, N 1.13. A molecular structure of the complex was obtained.

[(4-(Cyclopentylmethyl)-2-(diphenylphosphino)pyridine)(triphenylphosphine)₂Cu₂I₂] (Cu-4a).

Complex **Cu-4a** was obtained according to the general procedure. Pale yellow powder (99% yield).

¹H NMR (400 MHz, DMSO-d₆) δ [ppm] = 8.45 (bs, 1H, H_{Pyr}), 7.45 – 7.22 (m, 43H), 2.52 (bs, 1H, H_{CH2}), 1.88 (sept, ³J_{HH} = 7.6 Hz, 1H, H_{CH}), 1.52 – 1.41 (m, 4H, H_{Alkyl}), 1.40 – 1.36 (m, 2H, H_{Alkyl}), 1.01 – 0.93 (m, 2H, H_{Alkyl}). – **¹³C NMR** (101 MHz, DMSO-d₆) δ [ppm] = 133.7 (d, J_{CP} = 13.8 Hz), 133.1 (d, J_{CP} = 27.8 Hz), 129.8 (s), 128.4 (d, J_{CP} = 8.4 Hz), 31.6 (s), 24.3 (s). – **³¹P NMR** (162 MHz, DMSO-d₆) δ [ppm] = -5.35 (bs, 1P, P_{CyclopentPyrPhos}), -11.10 (bs, 2P, P_{PPh3}). – **MS** (FAB, 3-NBA) m/z [%] = 1311 (1) [M+Cu]⁺, 1049 (3) [Cu₃I₂LP]⁺, 860 (6) [Cu₂ILP]⁺, 787 (21) [Cu₃I₂L]⁺, 670 (49) [CuLP]⁺, 597 (62) [Cu₂IL]⁺, 587 (33) [CuP₂]⁺, 408 (100) [CuL]⁺, 346 (21) [L+H]⁺, 325 (42) [CuP]⁺, 263 (25) [P+H]⁺. – **IR** (ATR) $\tilde{\nu}$ [cm⁻¹] = 3047 (vw), 2949 (w), 2860 (vw), 1812 (vw), 1586 (w), 1542 (vw), 1479 (w), 1433 (w), 1388 (w), 1308 (vw), 1182 (w), 1156 (vw), 1093 (w), 1068 (w), 1027 (w), 997 (w), 846 (vw), 740 (m), 691 (m), 618 (vw), 515 (m), 468 (w), 425 (w). – Anal. calcd for **C₅₉H₅₄Cu₂I₂NP₃** (1249.0): C 56.65, H 4.35, N 1.12; found: C 56.18, H 4.40, N 1.16. A molecular structure of the complex with two molecules of dichloromethane was obtained.

[(4-(Cyclopentylmethyl)-2-(diphenylphosphino)pyridine)(tris(4-fluorophenyl)phosphine)₂Cu₂I₂] (Cu-4b).

The title complex was obtained via the general procedure. Pale yellow powder (91% yield).

¹H NMR (400 MHz, DMSO-d₆) δ [ppm] = 8.64 (bs, 1H, H_{Pyr}), 7.80 – 7.22 (m, 36H), 2.57 (d, ³J_{HH} = 7.5 Hz, 2H, H_{CH2}), 1.91 (sept, ³J_{HH} = 7.4 Hz, 1H, H_{CH}), 1.54 – 1.43 (m, 4H, H_{Alkyl}), 1.42 – 1.36 (m, 2H, H_{Alkyl}), 1.04 – 0.95 (m, 2H, H_{Alkyl}). – **¹³C NMR** (101 MHz, DMSO-d₆) δ [ppm] = 164.5 (s), 162.0 (s), 136.0 (q, J = 15.8 Hz, J = 8.3 Hz), 133.3 (bs), 130.1 (s), 128.9 (d, J = 27.7 Hz), 128.6 (d, J = 6.8 Hz), 115.9 (dd, J = 21.3 Hz, J = 9.5 Hz), 31.6 (s), 24.3 (s). – **³¹P NMR** (162 MHz, DMSO-d₆) δ [ppm] = -5.09 (bs, 1P, P_{CyclopentPyrPhos}), -12.90 (bs, 2P, P_{(F-Ph)3P}). – **¹⁹F NMR** (376 MHz, DMSO-d₆) δ [ppm] = -114.79 (s, 6F, F_{(F-Ph)3P}). – **MS** (FAB, 3-NBA) m/z [%] = 1419 (1) [M+Cu]⁺, 1103 (2) [Cu₃I₂LP]⁺, 913 (6) [Cu₂ILP]⁺, 884 (3) [Cu₂IP₂]⁺, 787 (36) [Cu₃I₂L]⁺, 724 (25) [CuLP]⁺, 695 (9) [CuP₂]⁺, 597 (72) [Cu₂IL]⁺, 408 (100) [CuL]⁺, 378 (24) [CuP]⁺, 346 (21) [L+H]⁺, 317 (14) [P+H]⁺. – **IR** (ATR) $\tilde{\nu}$ [cm⁻¹] = 3053 (vw), 2949 (vw), 2860 (vw), 1586 (m), 1493 (m), 1434 (w), 1392 (w), 1300 (vw), 1223 (m), 1158 (m), 1092 (w), 1013 (w), 825 (m), 742 (w), 693 (m), 634 (w), 517 (m), 496 (w), 467 (m), 443 (m), 431 (m). – Anal. calcd for **C₅₉H₄₈Cu₂F₆I₂NP₃** (1356.9): C 52.15, H 3.56, N 1.03; found: C 51.14, H 3.50, N 1.10. A molecular structure of the complex with one molecule of dichloromethane was obtained.

3. Spectroscopic Data

Ligands

4-Methyl-2-(diphenylphosphino)pyridine (MePyrPhos)

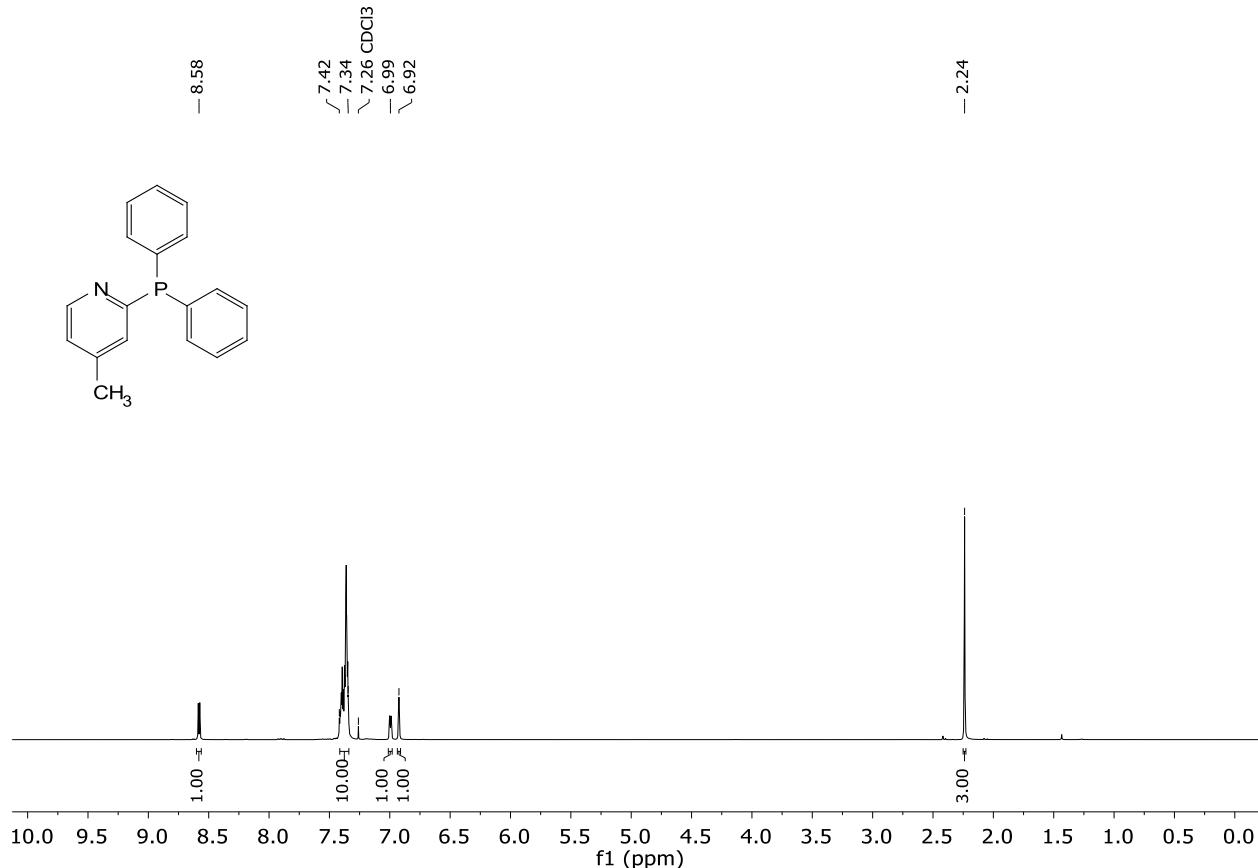


Figure S1. ¹H NMR of **MePyrPhos (2)** in CDCl₃, 400 MHz.

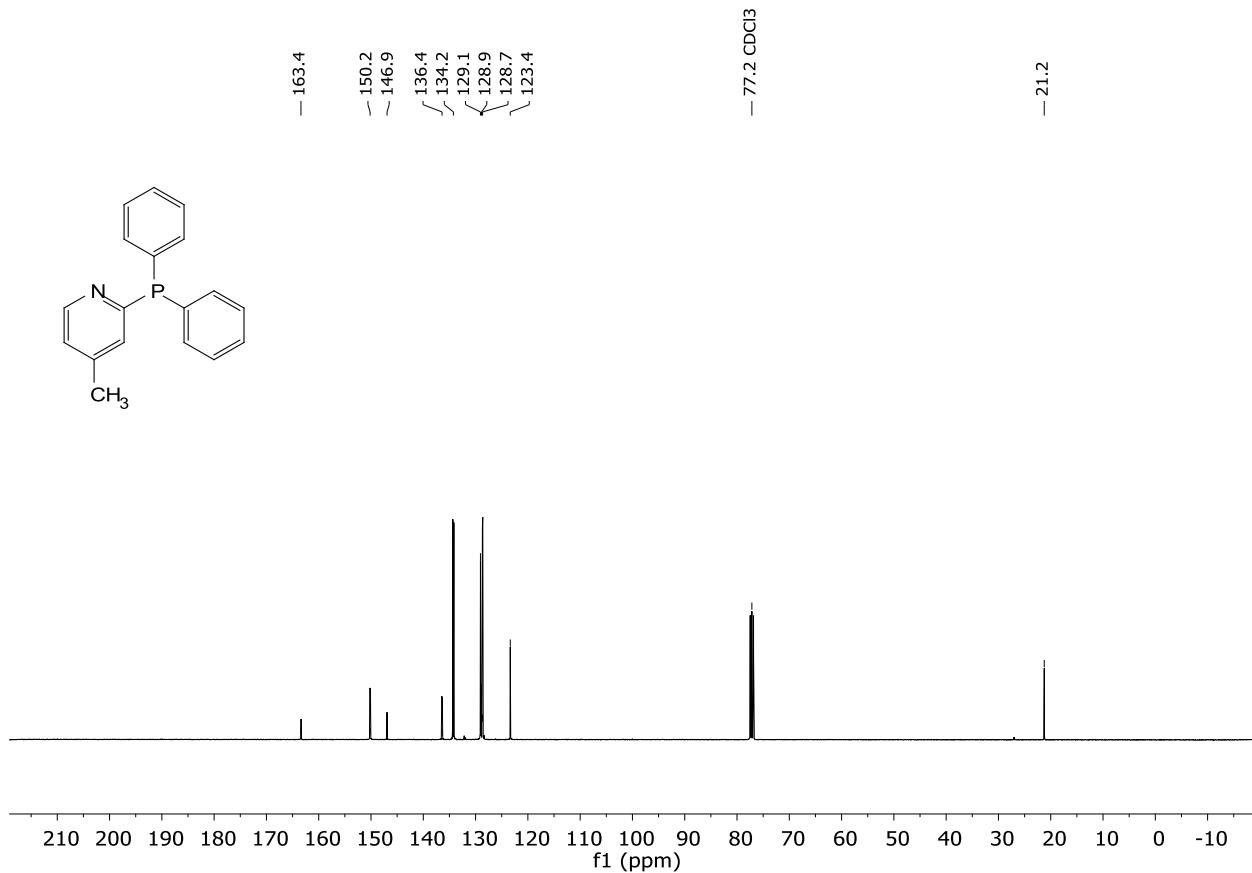


Figure S2. ^{13}C NMR of **MePyrPhos (2)** in CDCl_3 , 400 MHz.

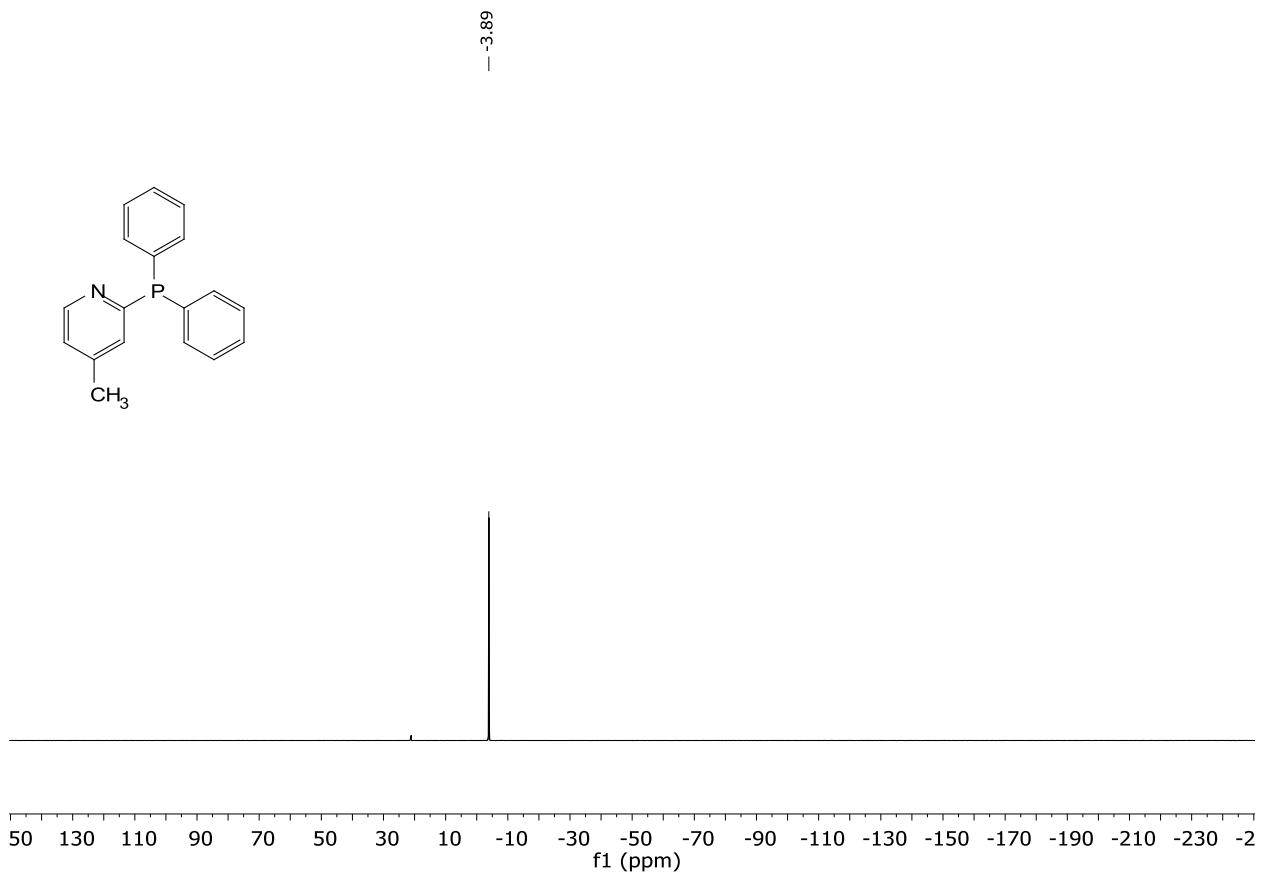


Figure S3. ^{31}P NMR of **MePyrPhos (2)** in CDCl_3 , 400 MHz.

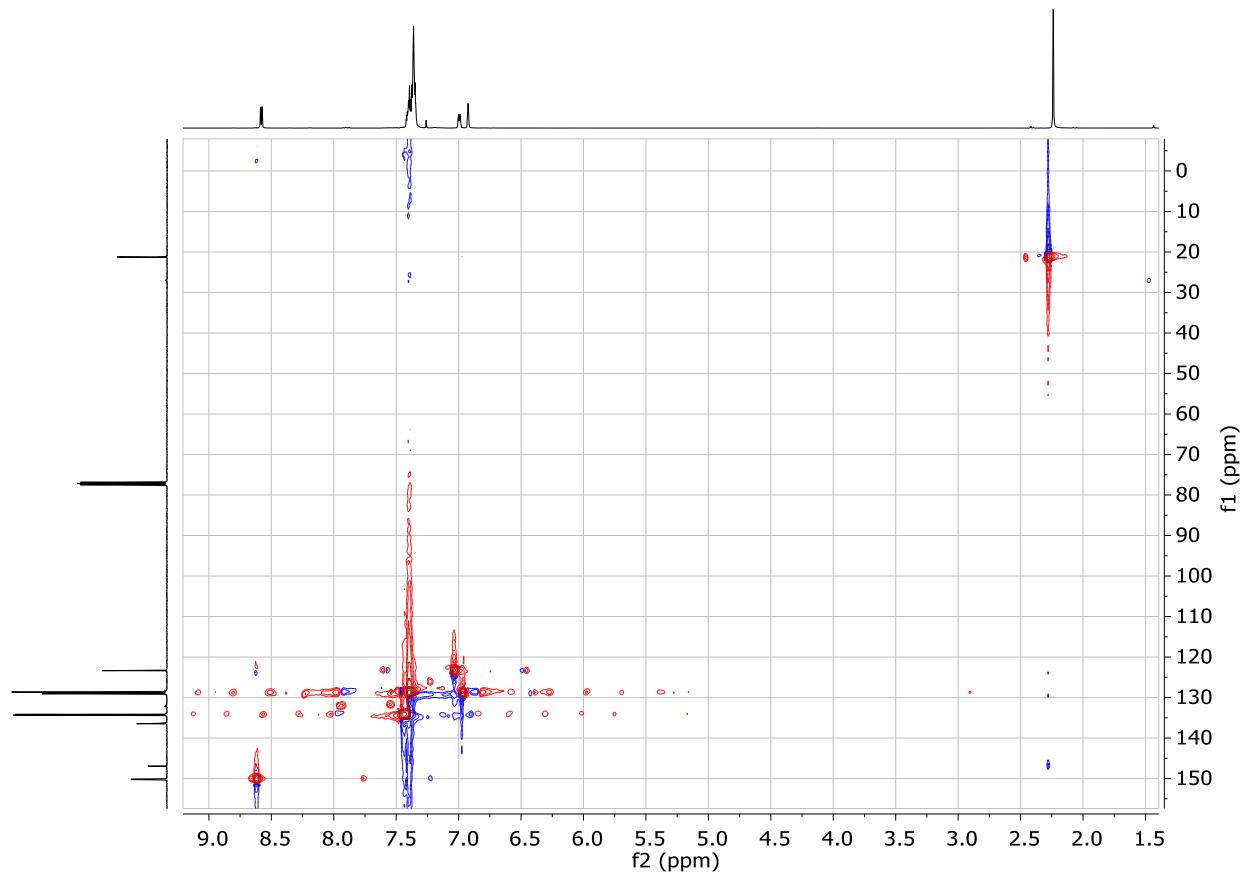


Figure S4. HSQC of **MePyrPhos (2)** in CDCl_3 , 400 MHz.

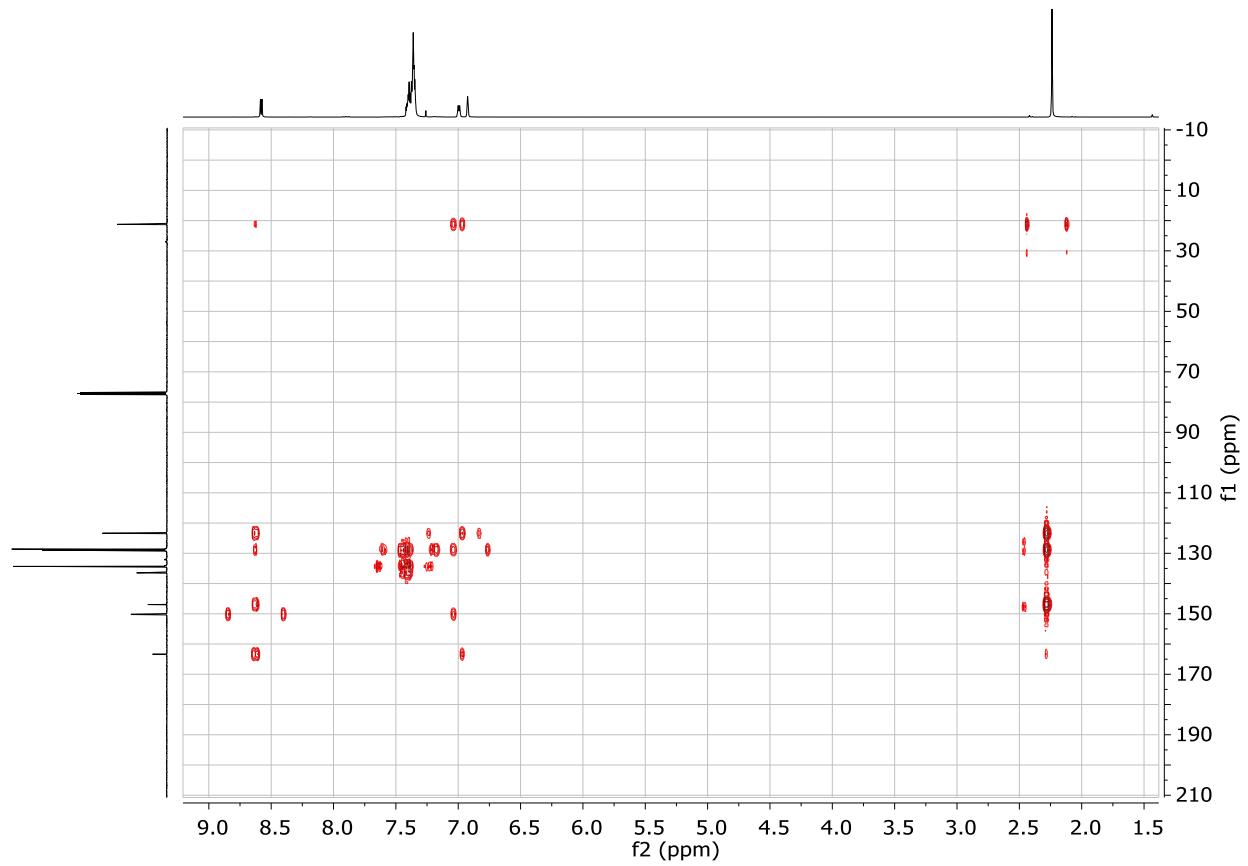


Figure S5. HMBC of **MePyrPhos (2)** in CDCl_3 , 400 MHz.

*2-(Diphenylphosphino)-4-tertbutylpyridine (**tertBuPyrPhos**)*

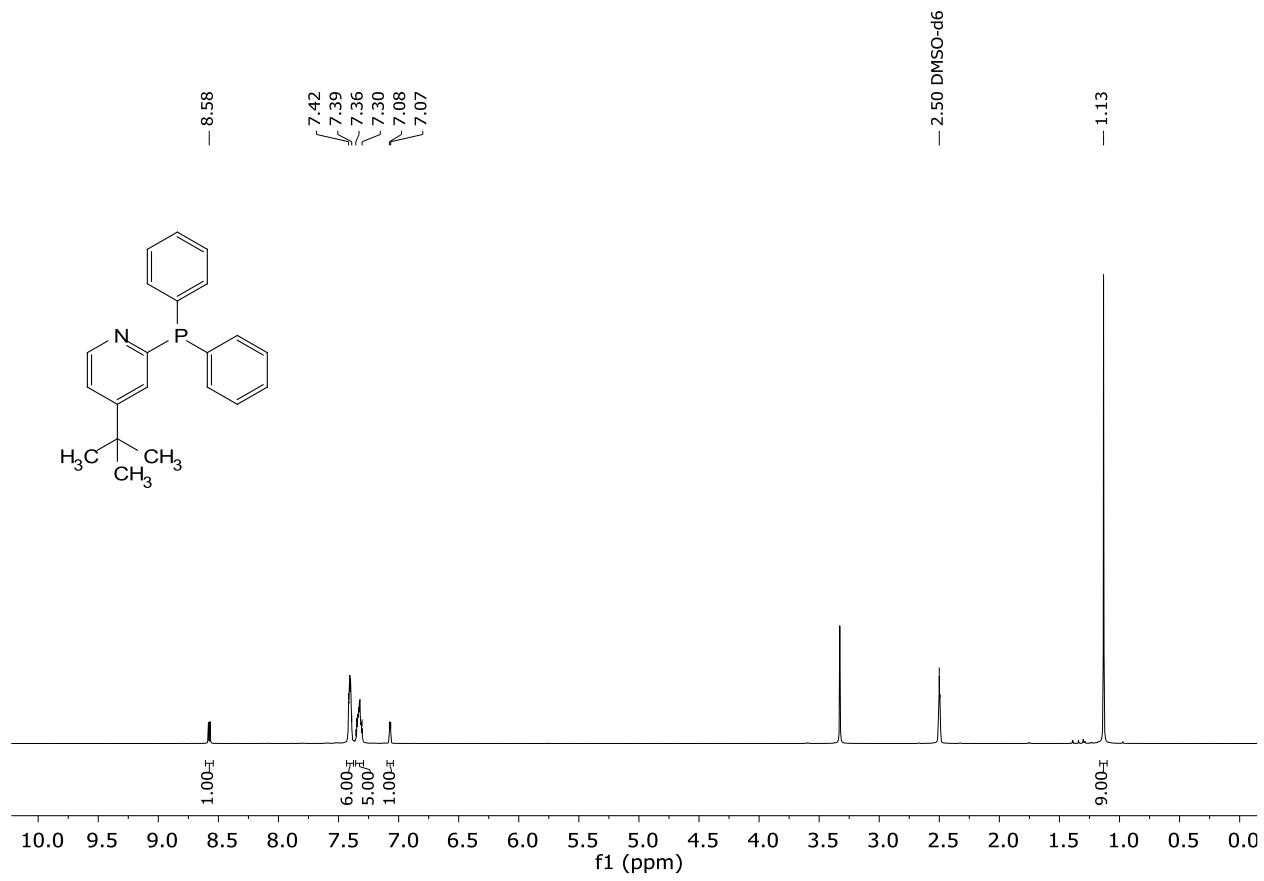


Figure S6. ^1H NMR of **tertBuPyrPhos** (**3**) in DMSO-d_6 , 400 MHz.

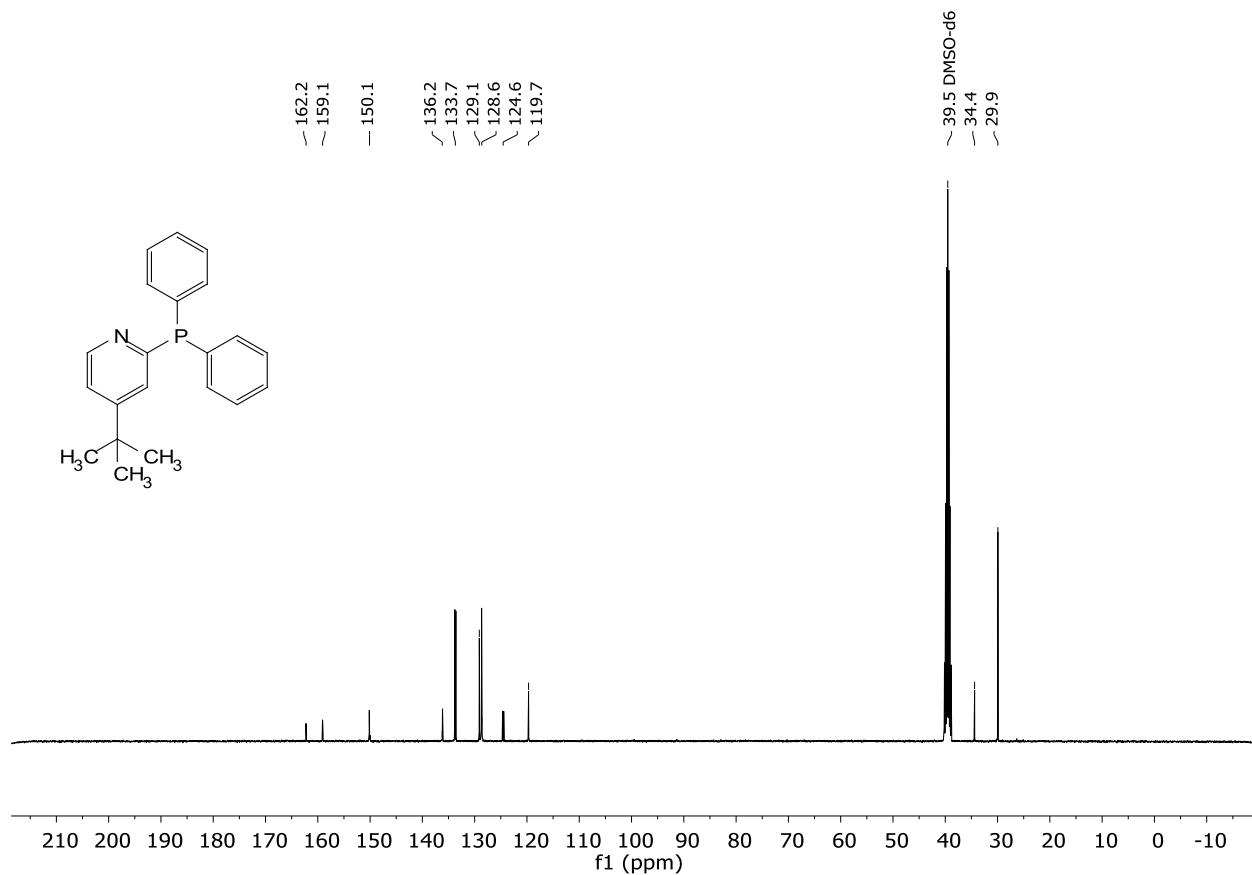


Figure S7. ^{13}C NMR of **tertBuPyrPhos (3)** in DMSO-d_6 , 400 MHz.

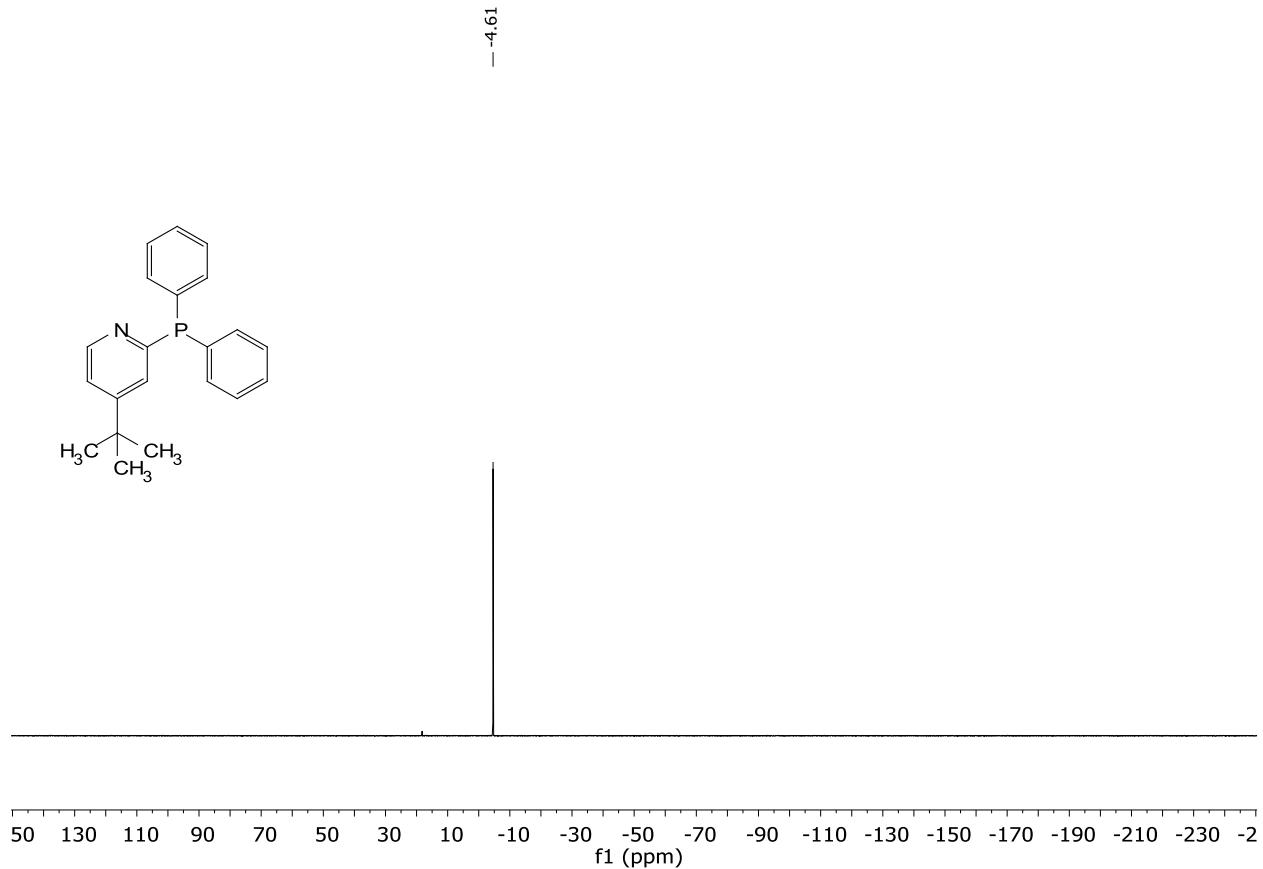


Figure S8. ^{31}P NMR of *tert*BuPyrPhos (**3**) in DMSO-d_6 , 400 MHz.

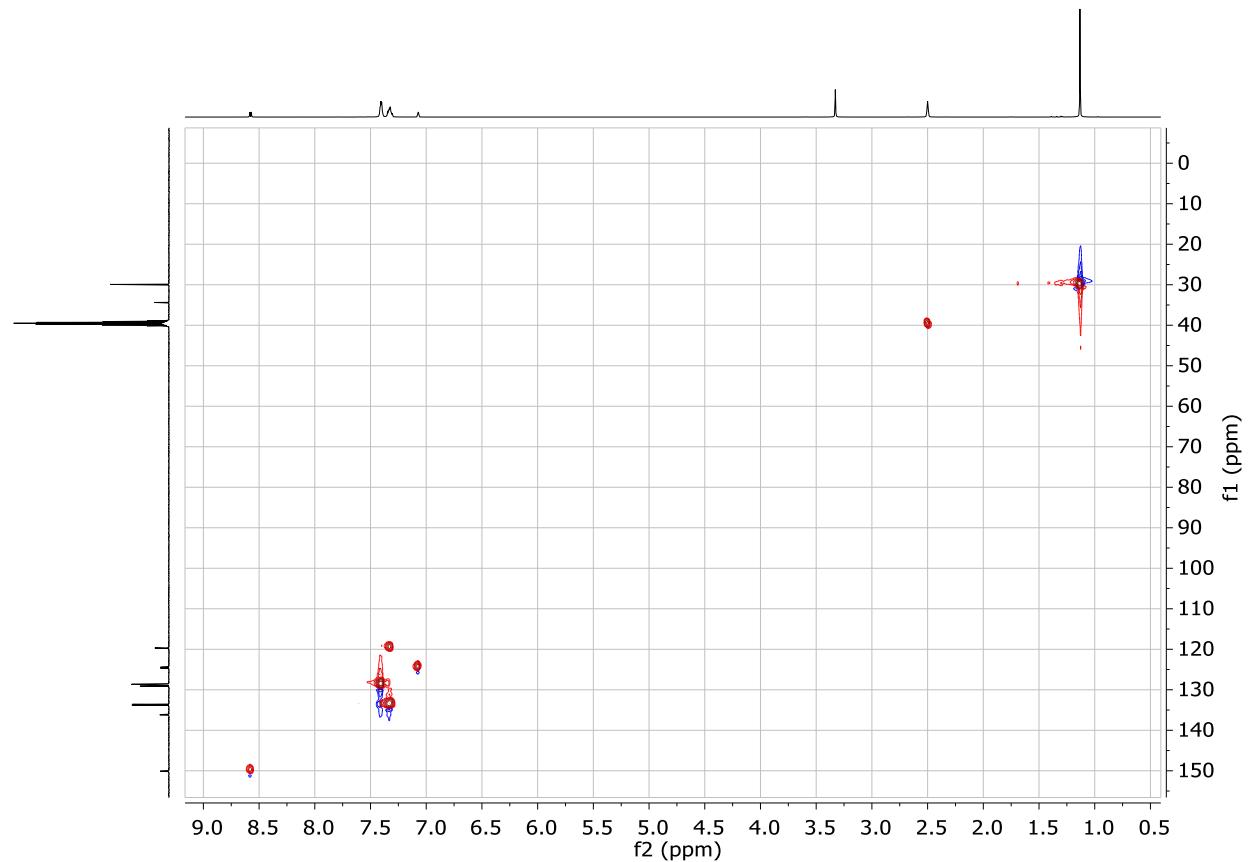


Figure S9. HSQC of *tert*BuPyrPhos (**3**) in DMSO-d_6 , 400 MHz.

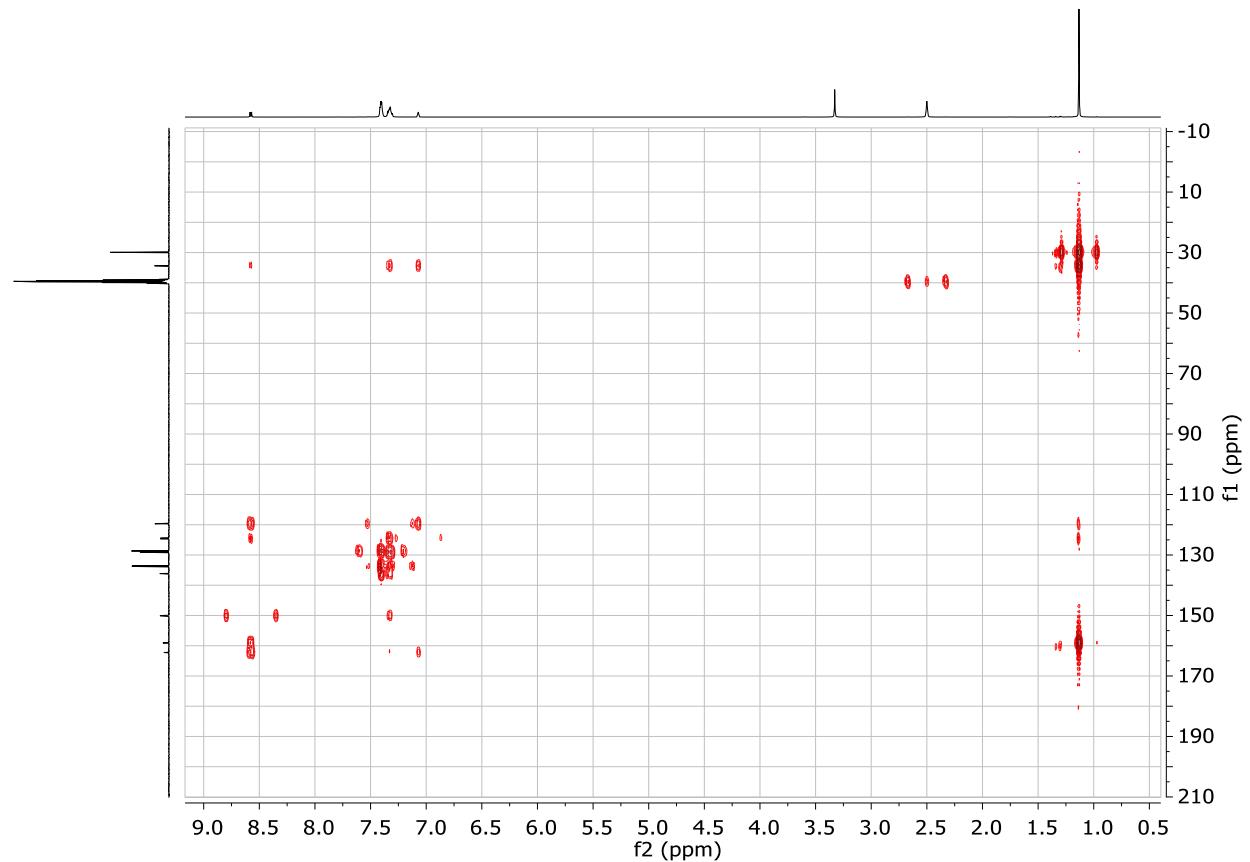


Figure S10. HMBC of **tertBuPyrPhos** (**3**) in DMSO-d_6 , 400 MHz.

*(4-(Cyclopentylmethyl)-2-(diphenylphosphino)pyridine (**CyclopentPyrPhos**)*

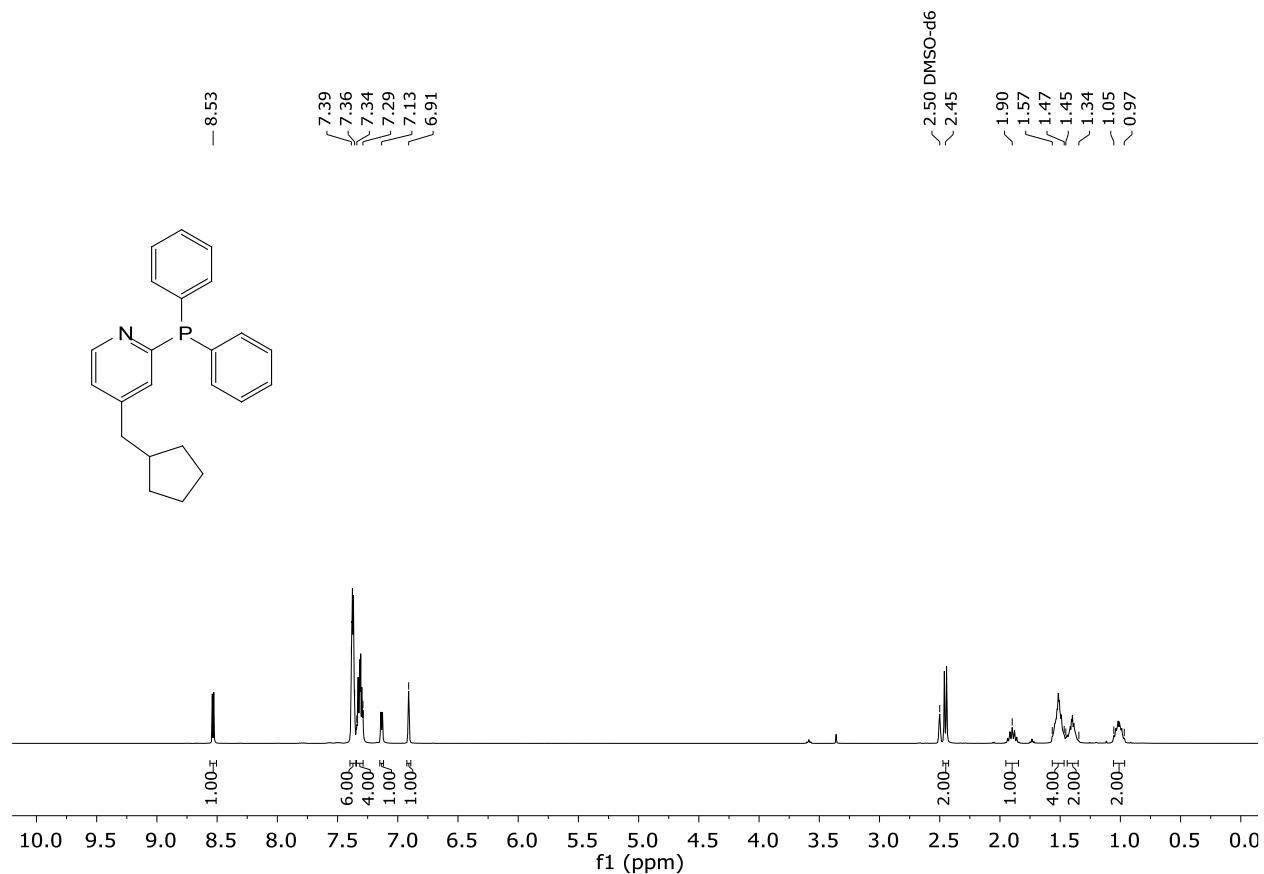


Figure S11. ¹H NMR of ligand **CyclopentPyrPhos** (**4**) in DMSO-d₆, 400 MHz.

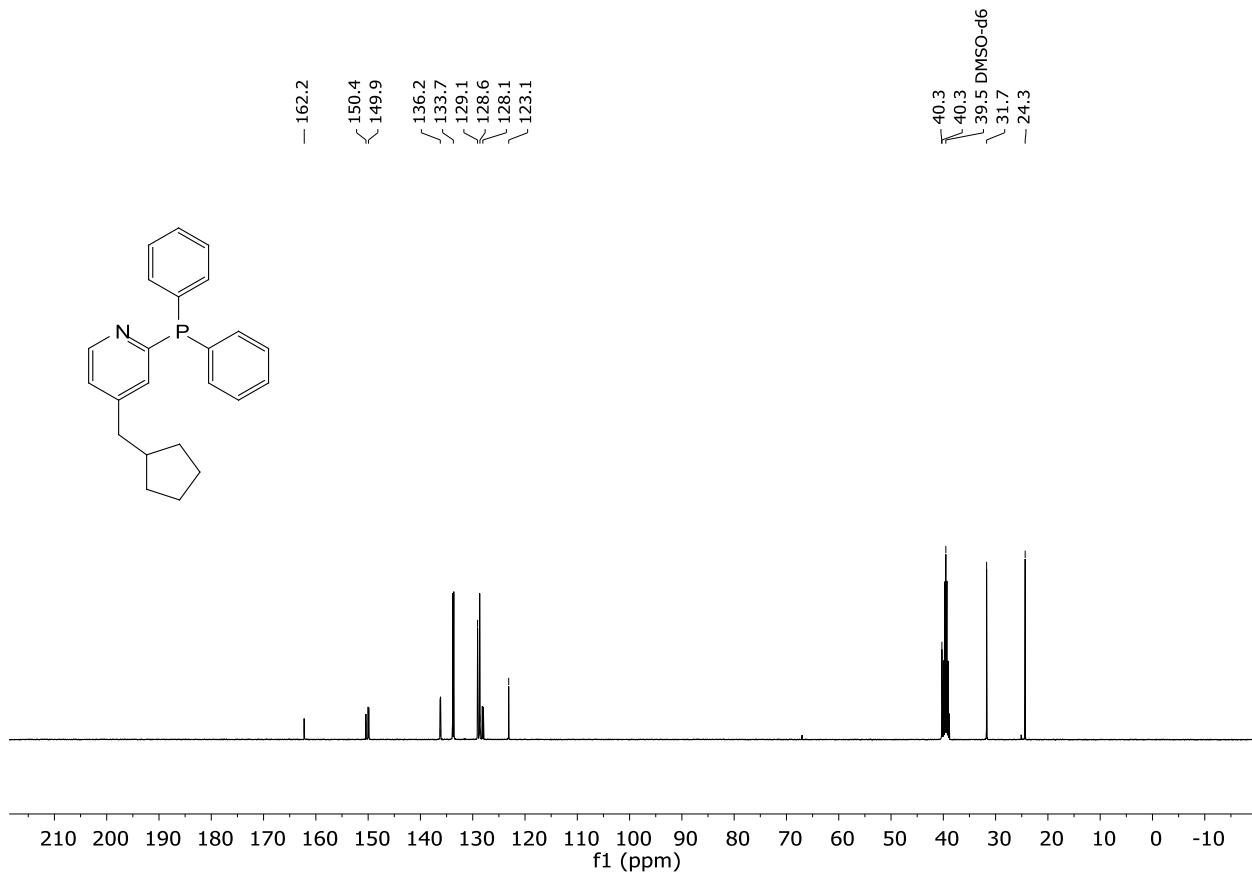


Figure S12. ^{13}C NMR of ligand **CyclopentPyrPhos (4)** in DMSO-d_6 , 400 MHz.

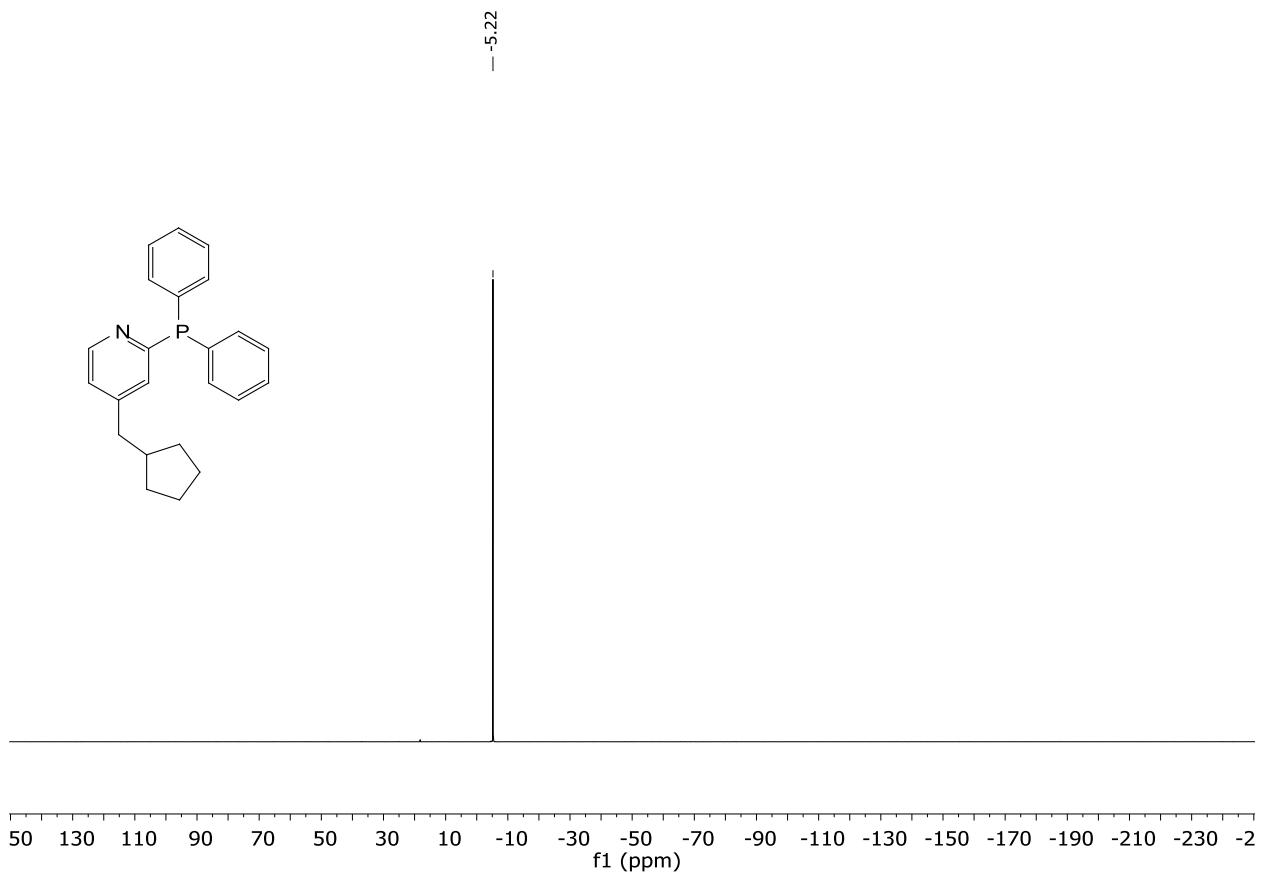


Figure S13. ^{31}P NMR of ligand **CyclopentPyrPhos (4)** in DMSO-d_6 , 400 MHz.

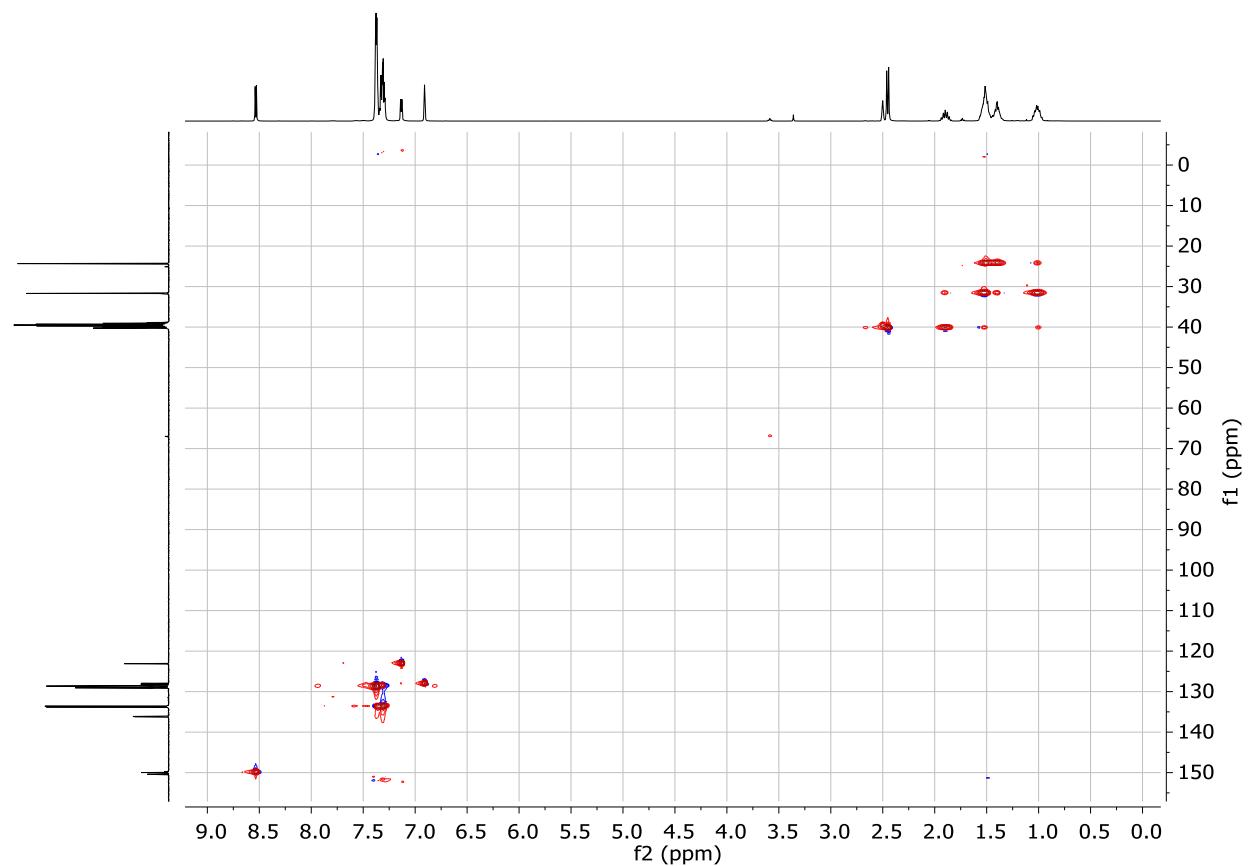


Figure S14. HSQC of ligand **CyclopentPyrPhos (4)** in DMSO-d_6 , 400 MHz.

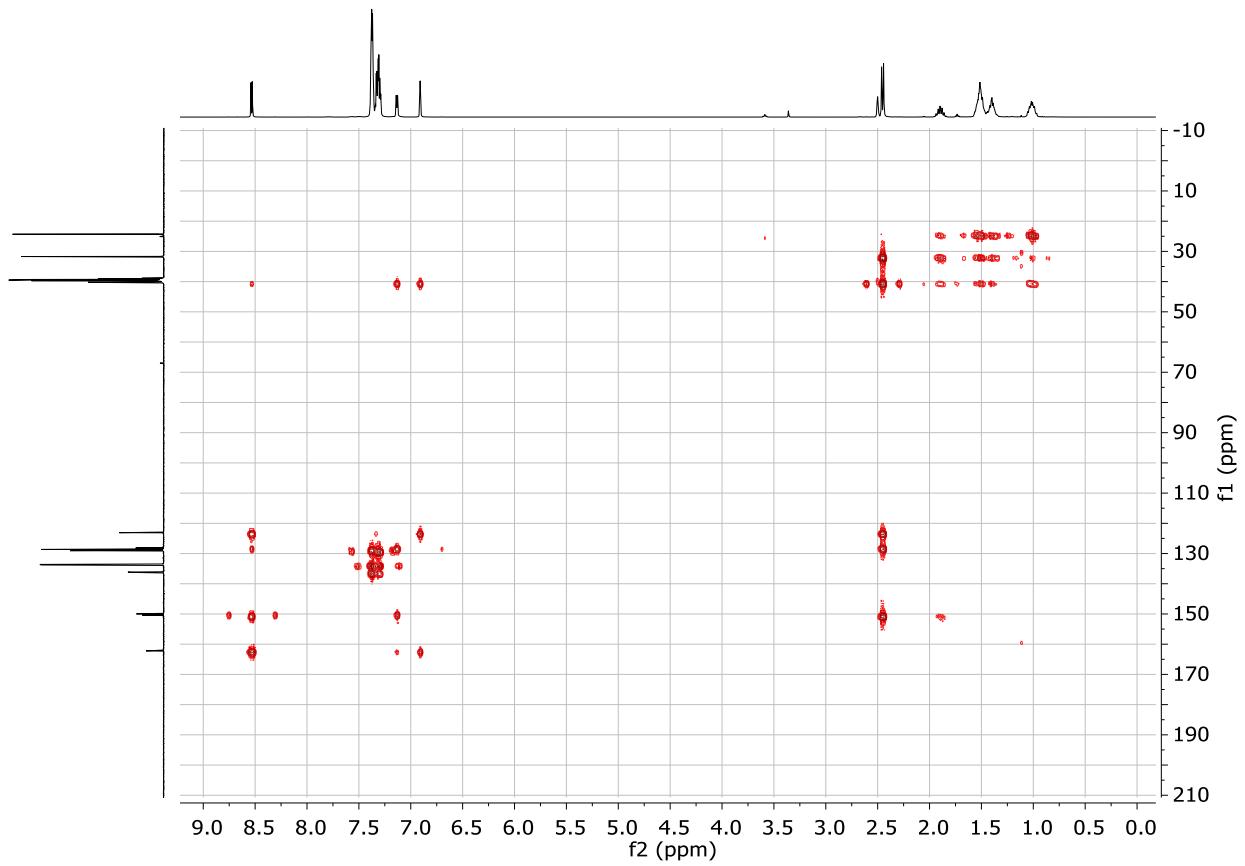


Figure S15. HMBC of ligand **CyclopentPyrPhos (4)** in DMSO-d₆, 400 MHz.

4. Crystallography

General Information. The molecular structures of the crystalline compounds **3**, **Cu-1b**, **Cu-1d**, **Cu-2b**, **Cu-2c**, **Cu-3b**, **Cu-4a** and **Cu-4b** were determined by single crystal X-ray analysis. Suitable crystals were selected and investigated on a *Stoe StadiVari* diffractometer using either Cu-K α ($\lambda = 1.54186 \text{ \AA}$) or Ga-K α ($\lambda = 1.34143 \text{ \AA}$) radiation. The crystals were kept at low temperature during data collection. Using Olex2², the structures were solved either with the ShelXS³ structure solution program using Direct Methods and refined with the ShelXL⁴ refinement package using Least Squares minimization. Non-hydrogen atoms were refined with anisotropic displacement parameters (disordered atoms and solvent molecules were refined isotropically); hydrogen atoms were modelled on idealized positions. Crystallographic and refinement data are summarized in **Table S1**.

Crystal Structure Determination of Cu-2d. The single-crystal X-ray diffraction study was carried out on a Bruker D8 Venture diffractometer with PhotonII CPAD detector at 123(2) K using Mo-K α radiation ($\lambda = 0.71073 \text{ \AA}$). Dual space methods (SHELXT)⁵ were used for structure solution and refinement was carried out using SHELXL-2014 (full-matrix least-squares on F^2)⁴. Hydrogen atoms were localized by difference electron density determination and refined using a riding model. A semi-empirical absorption correction was applied.

CCDC-1918364 (**Cu-2d**) and CCDC-1919266–1919273 (**3**, **Cu-1b**, **Cu-1d**, **Cu-2b**, **Cu-2c**, **Cu-3b**, **Cu-4a** and **Cu-4b**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via <https://www.ccdc.cam.ac.uk/structures/>.

Table S1. Crystallographic and refinement data.

Compound	3	Cu-1b	Cu-1d
Reaction code	JMB072_B	JMB049	JMB176
Emp. formula	C ₂₁ H ₂₂ NP	C ₅₃ H ₃₈ Cu ₂ F ₆ I ₂ NP ₃	C ₅₃ H ₃₈ Cl ₆ Cu ₂ I ₂ NP ₃
Moiety formula	C ₂₁ H ₂₂ NP	C ₅₃ H ₃₈ Cu ₂ F ₆ I ₂ NP ₃	C ₅₃ H ₃₈ Cl ₆ Cu ₂ I ₂ NP ₃
Molar mass	319.36	1276.63	1375.33
Temperature/K	180.15	180.15	180.15
Crystal system	monoclinic	monoclinic	monoclinic
Space group	P2 ₁ /c	P2 ₁ /c	P2 ₁
a/Å	10.9170(6)	18.9070(5)	10.61070(10)
b/Å	14.2236(10)	12.6700(2)	13.40820(10)
c/Å	11.8139(6)	20.7952(5)	18.7629(2)
α/°	90	90	90
β/°	103.375(4)	103.210(2)	93.5340(10)
γ/°	90	90	90
Volume/Å ³	1784.69(19)	4849.71(19)	2664.33(4)
Z	4	4	2
ρ _{calc} g/cm ³	1.189	1.748	1.714
μ/mm ⁻¹	0.153	12.531	12.969
F(000)	680.0	2504.0	1348.0
Radiation / Å	MoKα (λ = 0.71073)	CuKα (λ = 1.54186)	GaKα (λ = 1.34143)
2θ range /°	3.8–66	4.8–126.0	4.1–124
Refl. collected	15137	16081	29745
Independent refl.	6095 [R _{int} = 0.0279, R _σ = 0.0485]	7497 [R _{int} = 0.0293, R _σ = 0.0197]	9041 [R _{int} = 0.0410, R _σ = 0.0268]
Ind. refl. I ≥ 2σ(I)	3970	7328	8662
Data/rest./param.	6095/0/211	7497/0/604	9041/1/604
Gof	1.043	1.080	1.017
R indexes [I ≥ 2σ(I)]	R ₁ = 0.0456, wR ₂ = 0.1142	R ₁ = 0.0264, wR ₂ = 0.0710	R ₁ = 0.0748, wR ₂ = 0.2003
R indexes [all data]	R ₁ = 0.0823, wR ₂ = 0.1272	R ₁ = 0.0272, wR ₂ = 0.0714	R ₁ = 0.0769, wR ₂ = 0.2042
Diff. peak/hole /eÅ ⁻³	0.39/−0.25	0.44/−0.55	3.46/−2.89
Flack parameter			−0.013(8)
CCDC no.	1919266	1919267	1919268

Table S1. (continued).

Compound	Cu-2b	Cu-2c	Cu-2d
Reaction code	JMB057	JMB053	JMB181
Emp. formula	$C_{59}H_{52}Cu_2F_{6I_2}NP_3$	$C_{60.5}H_{40}ClCu_2F_{18I_2}NP_3$	$C_{54}H_{40}Cl_6Cu_2I_2NP_3$
Moiety formula	$C_{54}H_{40}Cu_2F_{6I_2}NP_3 \cdot C_5H_{12}$	$C_{60}H_{39}Cu_2F_{18I_2}NP_3 \cdot \frac{1}{2} CH_2Cl_2$	$C_{54}H_{40}Cl_6Cu_2I_2NP_3$
Molar mass	1362.80	1632.17	1389.36
Temperature/K	180.15	180.15	123
Crystal system	monoclinic	triclinic	monoclinic
Space group	$P2_1/n$	$P\bar{1}$	$P2_1/n$
a/Å	18.9997(6)	15.2045(10)	12.3082(4)
b/Å	12.9197(5)	15.6811(6)	24.7061(10)
c/Å	23.7991(6)	15.7442(10)	18.5816(8)
$\alpha/^\circ$	90	90.236(4)	90
$\beta/^\circ$	102.080(2)	106.170(5)	104.719(1)
$\gamma/^\circ$	90	116.415(4)	90
Volume/Å ³	5712.6(3)	3192.9(3)	5465.0(4)
Z	4	2	4
$\rho_{\text{calc}} \text{g/cm}^3$	1.585	1.698	1.689
μ/mm^{-1}	10.677	9.932	2.33
F(000)	2704.0	1596.0	2728
Radiation / Å	CuKα ($\lambda = 1.54186$)	GaKα ($\lambda = 1.34143$)	MoKα ($\lambda = 0.71073$)
2θ range /°	6.7–128.0	6.0–106.0	4.6 – 55
Refl. collected	22256	24934	286816
Independent refl.	9312 [$R_{\text{int}} = 0.1121$, $R_o = 0.0879$]	10843 [$R_{\text{int}} = 0.0150$, $R_o = 0.0186$]	12543 [$R_{\text{int}} = 0.043$, $R_o = 0.014$]
Ind. refl. $I \geq 2\sigma(I)$	7879	9479	11785
Data/rest./param.	9312/7/635	10843/4/777	12543/0/614
Gof	1.266	1.079	1.16
R indexes [$ I \geq 2\sigma(I)$]	$R_1 = 0.1216$, $wR_2 = 0.2919$	$R_1 = 0.0434$, $wR_2 = 0.1169$	$R_1 = 0.026$, $wR_2 = 0.050$
R indexes [all data]	$R_1 = 0.1290$, $wR_2 = 0.2999$	$R_1 = 0.0493$, $wR_2 = 0.1198$	$R_1 = 0.028$, $wR_2 = 0.051$
Diff. peak/hole /eÅ ⁻³	4.41/-4.41	1.67/-0.71	0.66/-0.49
CCDC no.	1919269	1919270	1918364

Table S1. (continued).

Compound	Cu-3b	Cu-4a	Cu-4b
Reaction code	JMB080_ANW012	JMB082-14	JMB83-15
Emp. formula	$C_{57}H_{46}Cu_2F_{6l_2}NP_3$	$C_{61}H_{58}Cl_4Cu_2l_2NP_3$	$C_{60}H_{50}Cl_2Cu_2F_6l_2NP_3$
Moiety formula	$C_{57}H_{46}Cu_2F_{6l_2}NP_3$	$C_{59}H_{54}Cu_2l_2NP_3 \cdot 2(CH_2Cl_2)$	$C_{59}H_{48}Cu_2F_{6l_2}NP_3 \cdot CH_2Cl_2$
Molar mass	1332.74	1420.67	1443.70
Temperature/K	150.15	160.15	180.15
Crystal system	triclinic	triclinic	monoclinic
Space group	$P\bar{1}$	$P\bar{1}$	$P2_1/c$
a/Å	12.6116(7)	11.2312(2)	19.1297(5)
b/Å	17.2115(8)	14.2065(3)	12.6256(2)
c/Å	27.1045(9)	21.3424(4)	25.3784(6)
$\alpha/^\circ$	103.391(3)	72.1220(10)	90
$\beta/^\circ$	100.214(4)	82.3480(10)	111.373(2)
$\gamma/^\circ$	101.155(4)	67.3660(10)	90
Volume/Å ³	5461.1(5)	2990.83(10)	5708.0(2)
Z	4	2	4
$\rho_{\text{calc}} \text{g/cm}^3$	1.621	1.578	1.680
μ/mm^{-1}	11.015	11.059	11.566
F(000)	2632.0	1416.0	2856.0
Radiation / Å	GaKα ($\lambda = 1.34143$)	GaKα ($\lambda = 1.34143$)	CuKα ($\lambda = 1.54186$)
2θ range /°	4.9–110.0	6.1–120.0	8.6–125.0
Refl. collected	47037	37037	38804
Independent refl.	20129 [$R_{\text{int}} = 0.0424$, $R_o = 0.0345$]	13129 [$R_{\text{int}} = 0.0253$, $R_o = 0.0194$]	9040 [$R_{\text{int}} = 0.0243$, $R_o = 0.0148$]
Ind. refl. $I \geq 2\sigma(I)$	17952	11520	8926
Data/rest./param.	20129/0/1285	13129/0/658	9040/0/682
Gof	1.090	1.033	1.199
R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0466$, $wR_2 = 0.1309$	$R_1 = 0.0316$, $wR_2 = 0.0861$	$R_1 = 0.0569$, $wR_2 = 0.1359$
R indexes [all data]	$R_1 = 0.0534$, $wR_2 = 0.1343$	$R_1 = 0.0367$, $wR_2 = 0.0885$	$R_1 = 0.0572$, $wR_2 = 0.1360$
Diff. peak/hole /eÅ ⁻³	1.68/-1.23	1.46/-0.94	1.13/-1.32
CCDC no.	1919271	1919272	1919273

Ligand

*2-(Diphenylphosphino)-4-tertbutylpyridine (**tertBuPyrPhos**)*

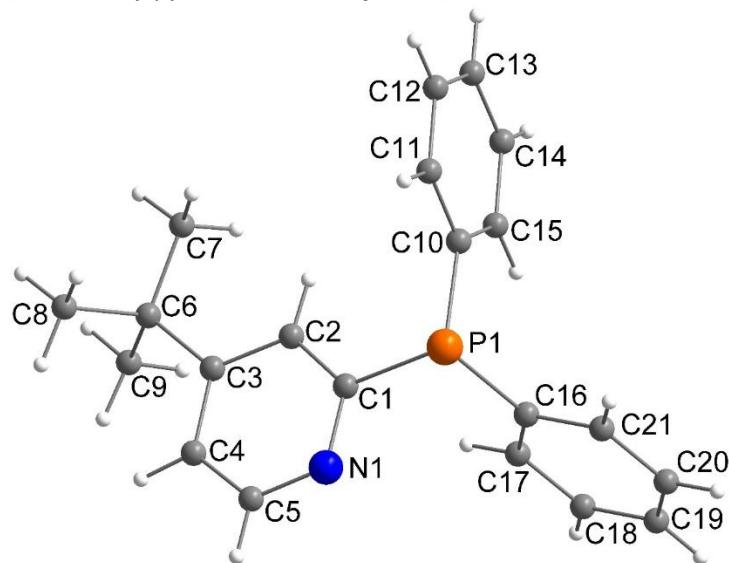


Figure S16. Molecular structure of **tertBuPyrPhos** (3).

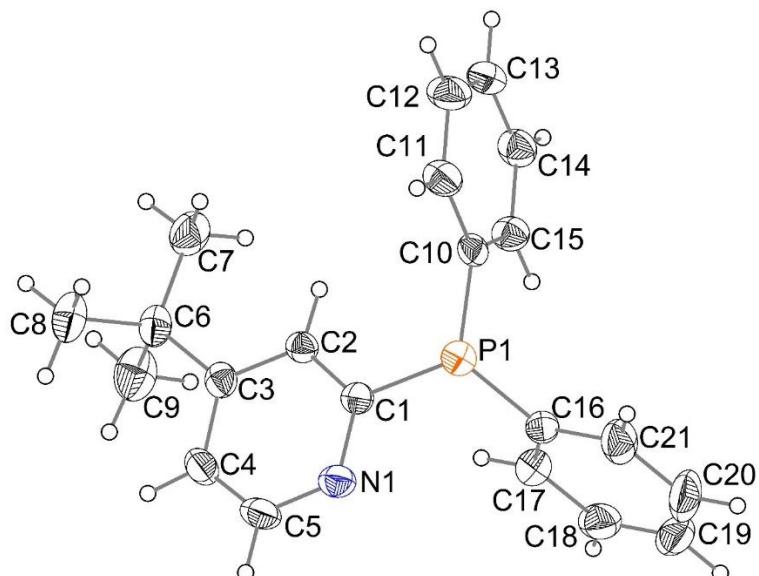


Figure S17. ORTEP plot of **tertBuPyrPhos** (3)
(displacement parameters are drawn at 50 % probability level).

Cu(I) Complexes

Additionally the molecular structure of Cu(I) complex **Cu-4a** was obtained by X-Ray analysis.

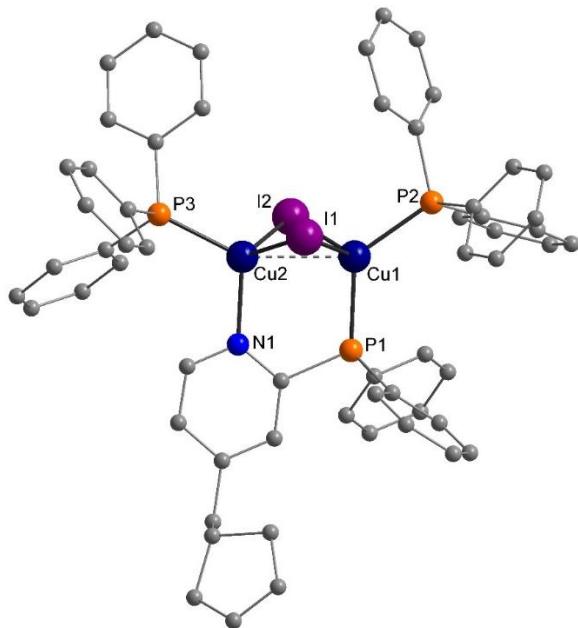


Figure S18. Molecular structure of Cu(I) complex **Cu-4a** bearing triphenylphosphine as ancillary ligands (hydrogen atoms and solvent were omitted for clarity).

ORTEP plots

In the following section the ORTEP plots of all obtained Cu(I) AlkylPyrPhos complexes **Cu-1b**, **Cu-1d**, **Cu-2b**, **Cu-2c**, **Cu-3b**, **Cu-4a** and **Cu-4b** are shown.

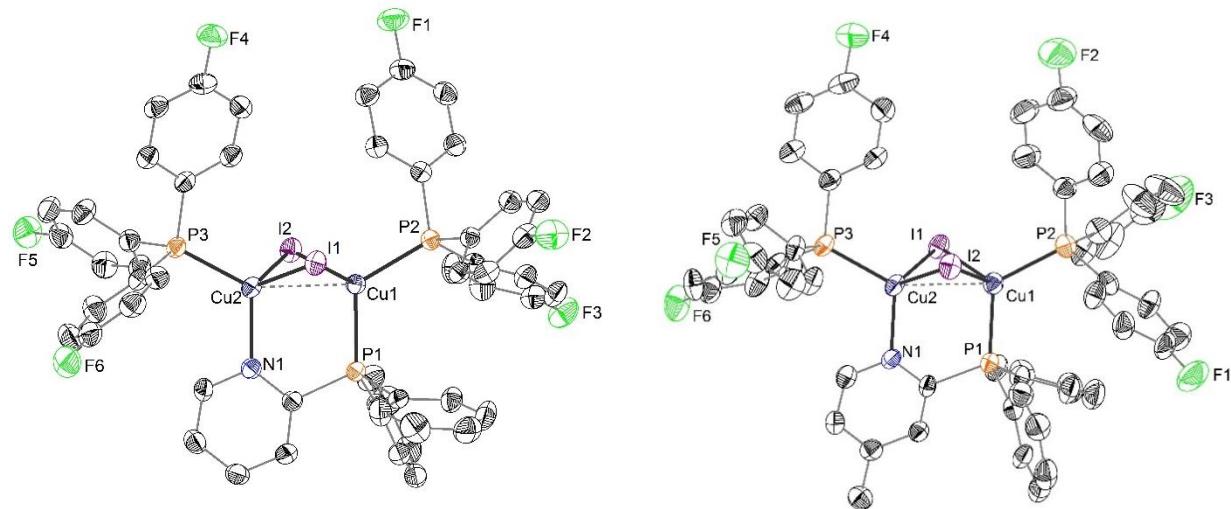


Figure S19. ORTEP plots of **Cu-1b** (left) and **Cu-2b** (right)
(displacement parameters are drawn at 50 % probability level,
hydrogen atoms and solvent (right) were omitted for clarity).

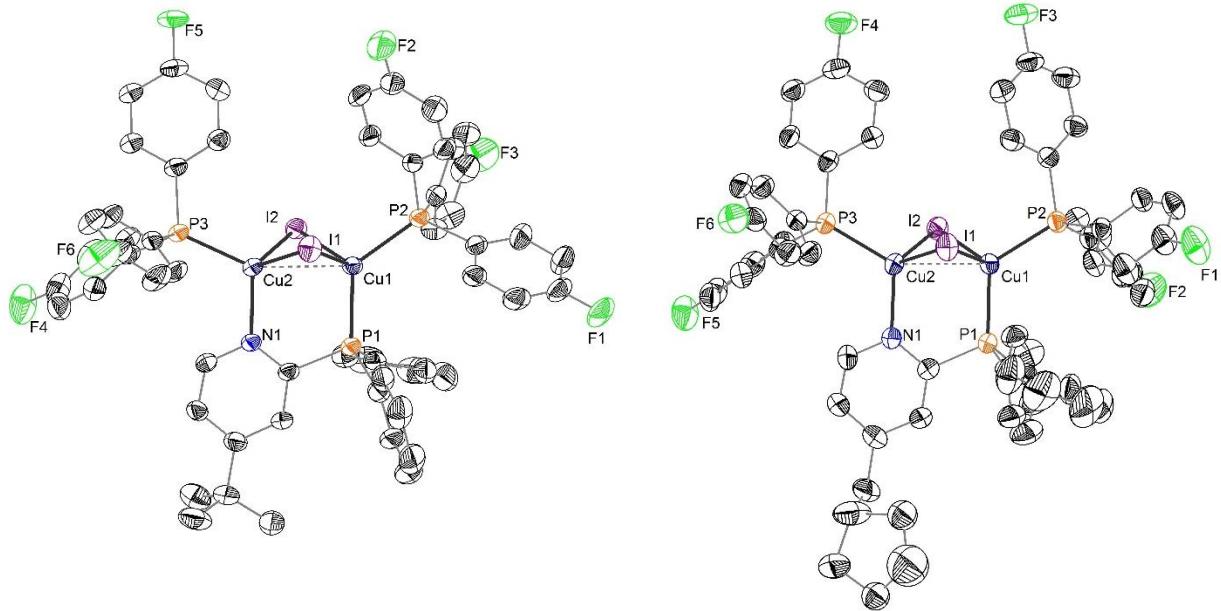


Figure S20. ORTEP plots of **Cu-3b** (left) and **Cu-4b** (right)
(displacement parameters are drawn at 50 % probability level,
hydrogen atoms and solvent and minor disordered parts (right) were omitted for clarity).

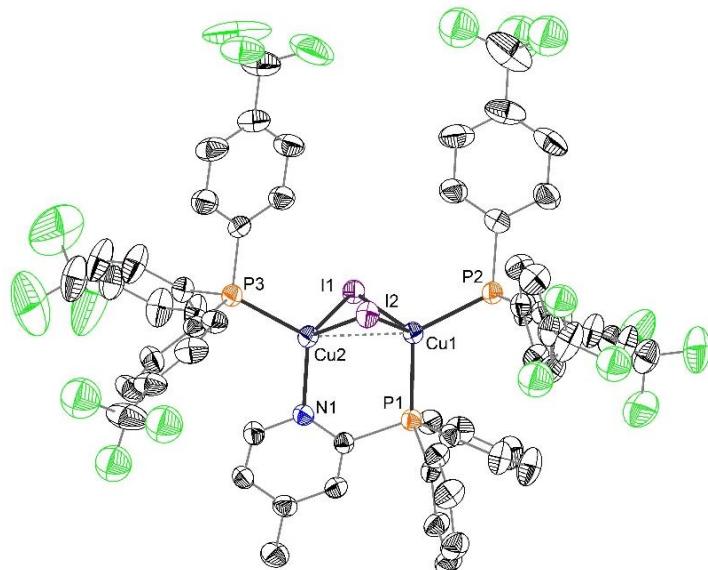


Figure S21. ORTEP plot of **Cu-2c**
(displacement parameters are drawn at 50 % probability level,
hydrogen atoms, solvent and minor disordered parts were omitted for clarity).

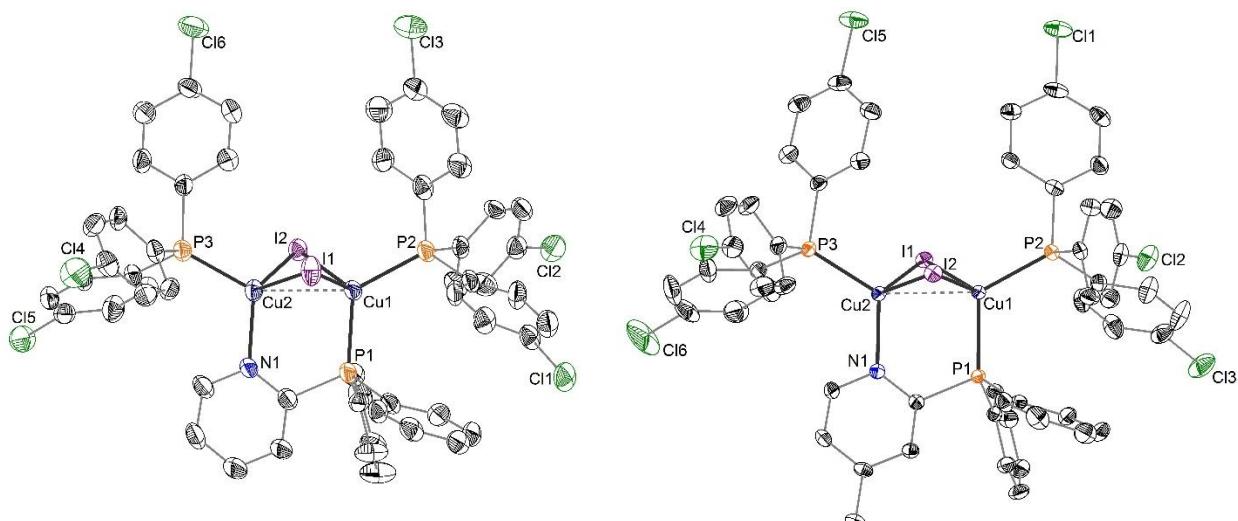


Figure S22. ORTEP plots of **Cu-1d** (left) and **Cu-2d** (right)
(displacement parameters are drawn at 50 % probability level,
hydrogen atoms were omitted for clarity).

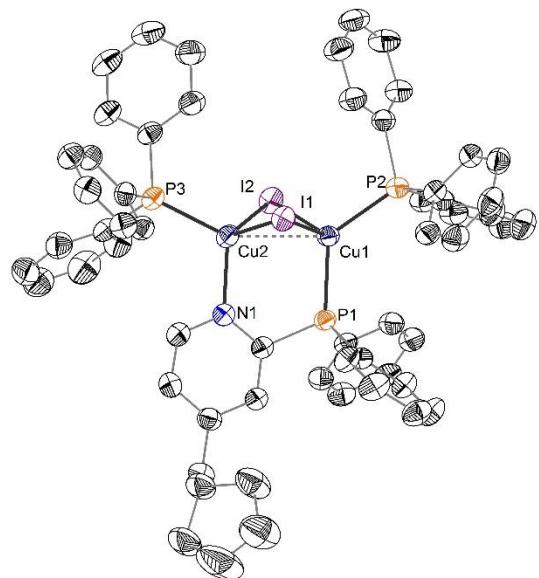


Figure S23. ORTEP plot of **Cu-4a**
(displacement parameters are drawn at 50 % probability level,
hydrogen atoms and solvent were omitted for clarity).

5. Theoretical Calculations

Computational Details. The different complexes were optimized using the `ridft` module within the standard jobex script. The energy threshold within a self-consistent (SCF) calculation was set to $10^{-10} E_h$ whereas convergence thresholds for the structure optimization were $10^{-8} E_h$ and $10^{-6} E_h/a_0$ for the total SCF energy and the Cartesian gradients, respectively. The structures used for the calculations are given in section **9. Structures in XYZ**.

For the calculation of IR absorption spectra of **Cu-1b** and **Cu-2b** the crystal structures were used as input structures. For geometry optimization and frequency calculation the DFT functional B3LYP with dispersion correction (D_3)⁶, RI-approximation (resolution of identity) and the def2-TZVP basis set were used. The calculations of energies, gradients, force constants and frequencies were performed and taken from Turbomole 7.3⁷. Obtained vibrational frequencies were convolved with a Gaussian profile with a FWHM = 8 cm⁻¹. Triplet states were calculated as ground state using UDFT.

The S₁ and T₁ structures were optimized and the respective vertical transitions were calculated by TD-DFT (B3LYP-D3(BJ)/def2-TZVP) and considered for calculation of the singlet-triplet energy gaps of **Cu-1b** and **Cu-2b**.

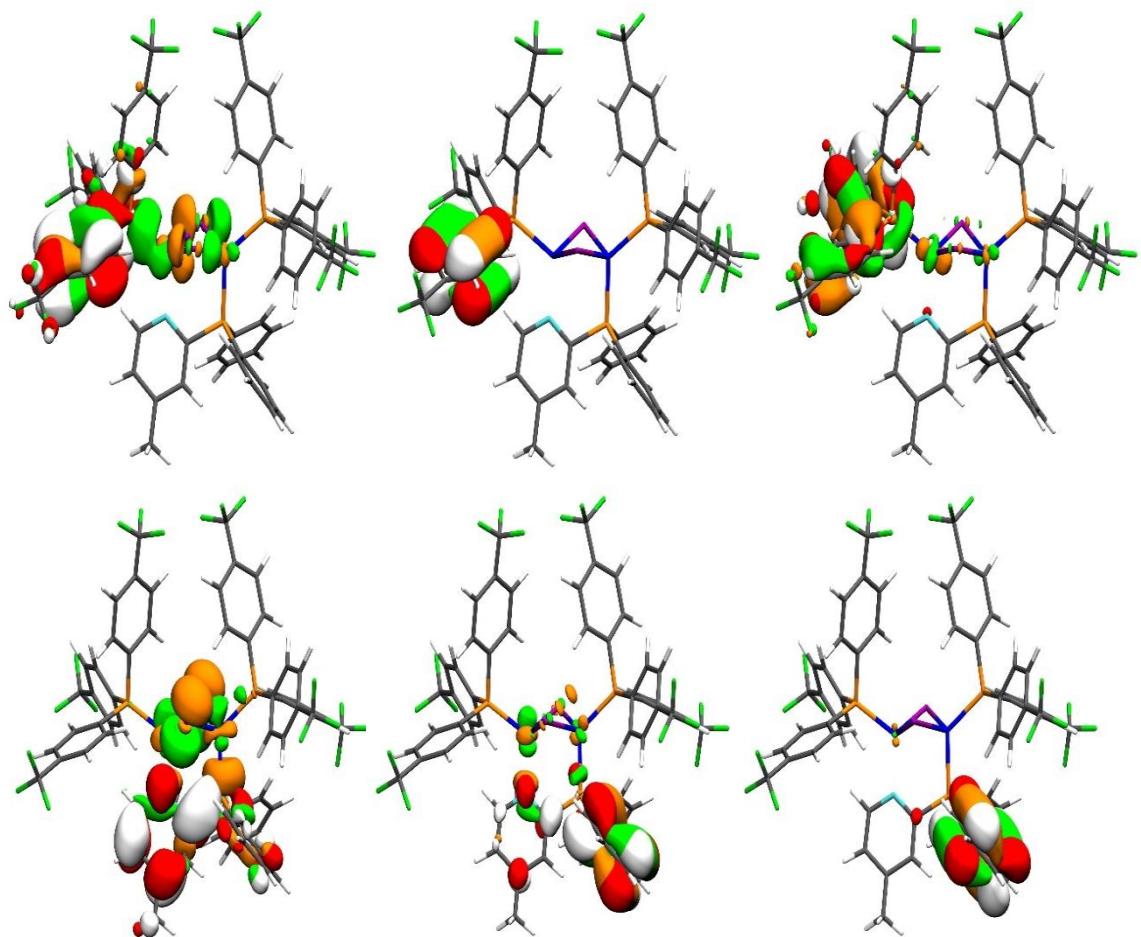


Figure S24. First three natural transition orbitals of **Cu-2c** for the first triplet excitation (top: 80%, 10%, 6%) and second triplet excitation (bottom: 79%, 12%, 5%) at the S₀ structure. The colours green/orange indicate the hole while red/white correspond to the electron. (isovalue: $\pm 0.03 \alpha_0^{-3/2}$, Cu = blue, P = orange, I = purple, N = cyan, F = green, C = grey, H = white).

6. Absorption Spectra

Absorption spectra measurements. Absorption spectra were recorded on a *Thermo Scientific*, model Evolution 201. 10 mg of each sample were dissolved in 100 mL spectroscopical grade dichloromethane respectively. 1 mL of this stock solution was diluted 1:10 to give concentrations of 0.01 mg/mL.

Experimental Data

Table S2. Overview of the absorption maxima, concentration of the samples, extinction and molar extinction coefficient at the absorption maxima for all Cu(I) complexes and ligands investigated in this study respectively.

compound	absorption maxima [nm]	c ($\times 10^{-6}$) [mol/L]	E(absorption maximum)	molar extinction coefficient at absorption maxima ($\times 10^{-4}$) [L/(mol × cm)]
Cu-1b	250	7.96	0.283	3.55
Cu-1c	265	6.61	0.255	3.86
Cu-1d	250	7.44	0.371	4.99
Cu-2b	250	8.08	0.296	3.66
Cu-2c	266	6.34	0.233	3.67
Cu-2d	250	7.49	0.368	4.91
Cu-3a	250	8.28	0.335	4.05
Cu-3b	250	7.56	0.288	3.80
Cu-4a	250	8.17	0.302	3.70
Cu-4b	250	7.45	0.258	3.46

compound	absorption maxima [nm]	c ($\times 10^{-5}$) [mol/L]	E(absorption maximum)	molar extinction coefficient at absorption maxima [L/(mol × cm)]
ligand 1	258	3.81	0.372	9753
ligand 2	257	3.62	0.335	9250
ligand 3	257	3.18	0.287	9013
ligand 4	259	2.93	0.250	8529
ligand a	263	3.84	0.360	9365
ligand b	257	3.23	0.348	10785
ligand c	270	2.19	0.244	11180
ligand d	266	2.83	0.389	13755

Comparison plots of the absorption spectra of the copper complexes **Cu-1b – Cu-4b** and their corresponding bridging and ancillary ligands, measured of the solution with the concentrations given in **Table S2** at room temperature.

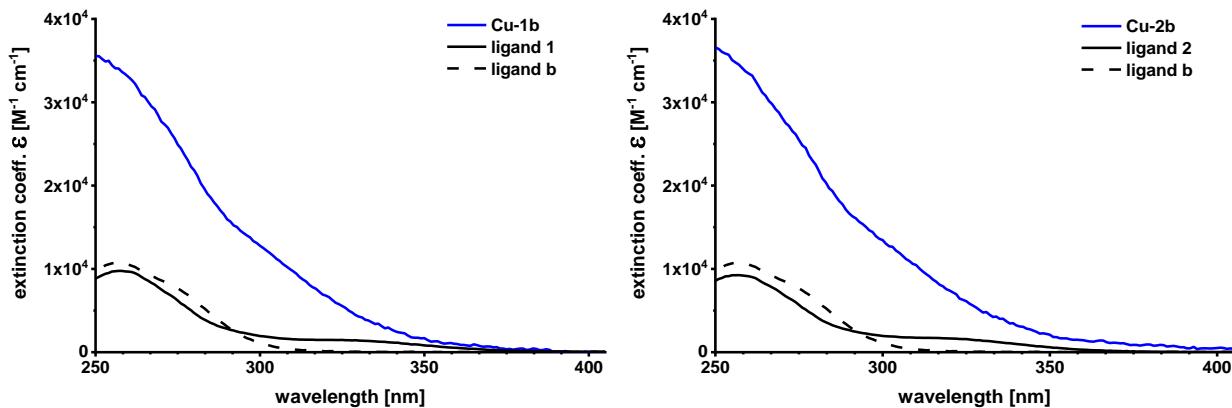


Figure S25. UV/Vis spectra of **Cu-1b**, **Cu-2b** and the corresponding ligands **1**, **2** and **b**.

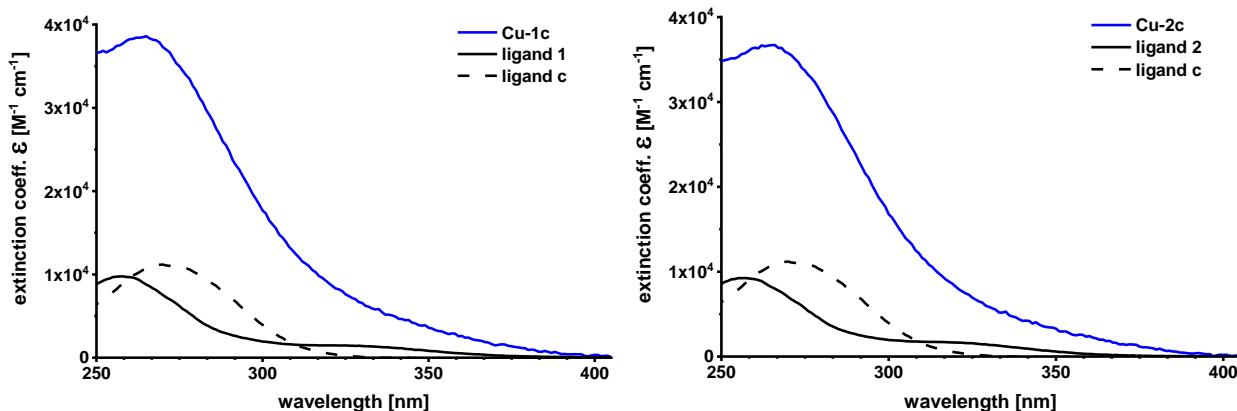


Figure S26. UV/Vis spectra of **Cu-1c**, **Cu-2c** and the corresponding ligands **1**, **2** and **c**.

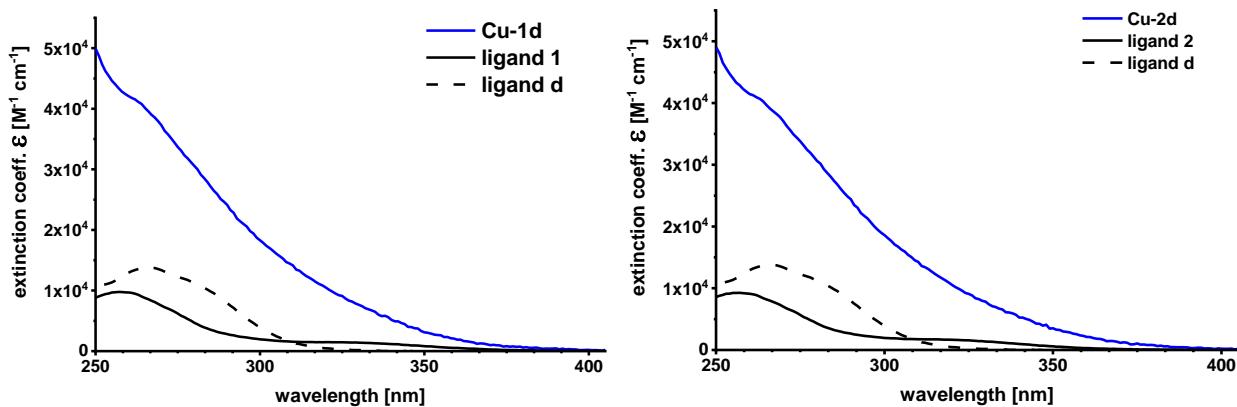


Figure S27. UV/Vis spectra of **Cu-1d**, **Cu-2d** and the corresponding ligands **1**, **2** and **d**.

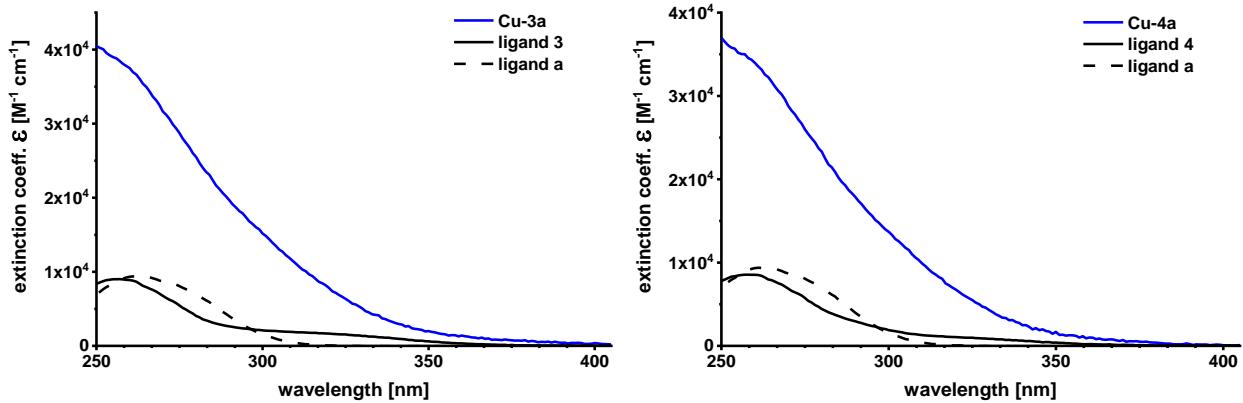


Figure S28. UV/Vis spectra of **Cu-3a**, **Cu-4a** and the corresponding ligands **3**, **4** and **a**.

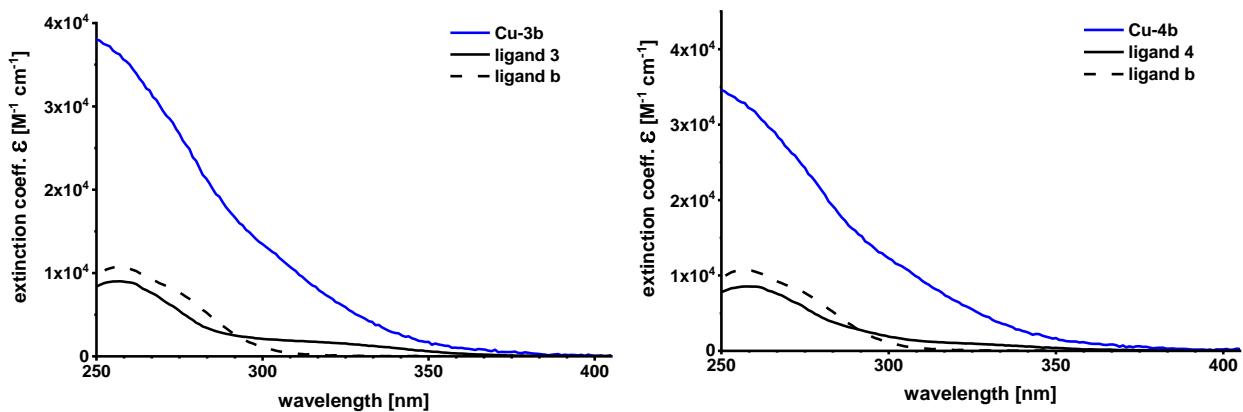


Figure S29. UV/Vis spectra of **Cu-3b**, **Cu-4b** and the corresponding ligands **3**, **4** and **b**.

Further absorption spectra were recorded using a *Perkin-Elmer* Lambda 900 double beam UV/VIS/NIR spectrophotometer with concentrations in the region of $2 \cdot 10^{-5}$ M.

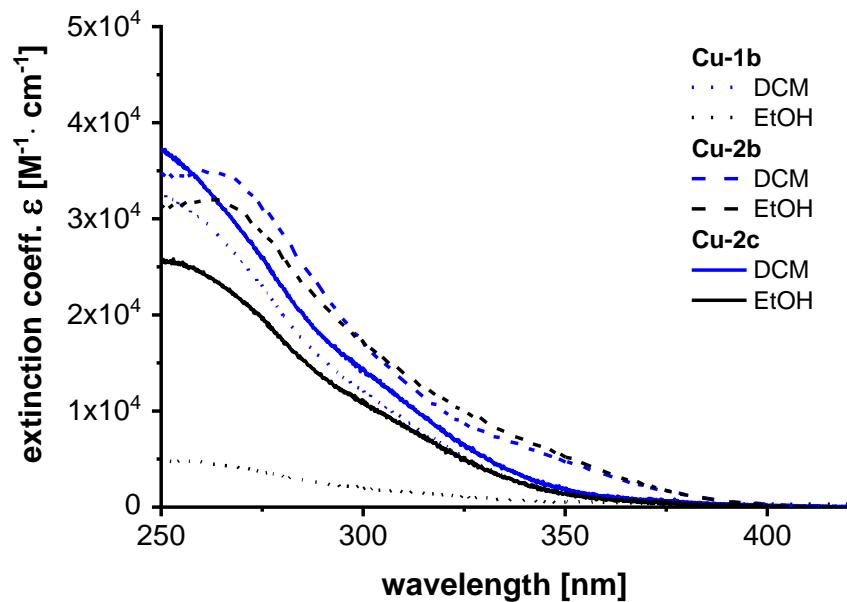


Figure S30. UV/Vis spectra of the complexes **Cu-1b**, **Cu-2b** and **Cu-2c**, measured in dichloromethane (blue) and ethanol (black).

Absorption measurements of the Cu(I) complexes in neat films

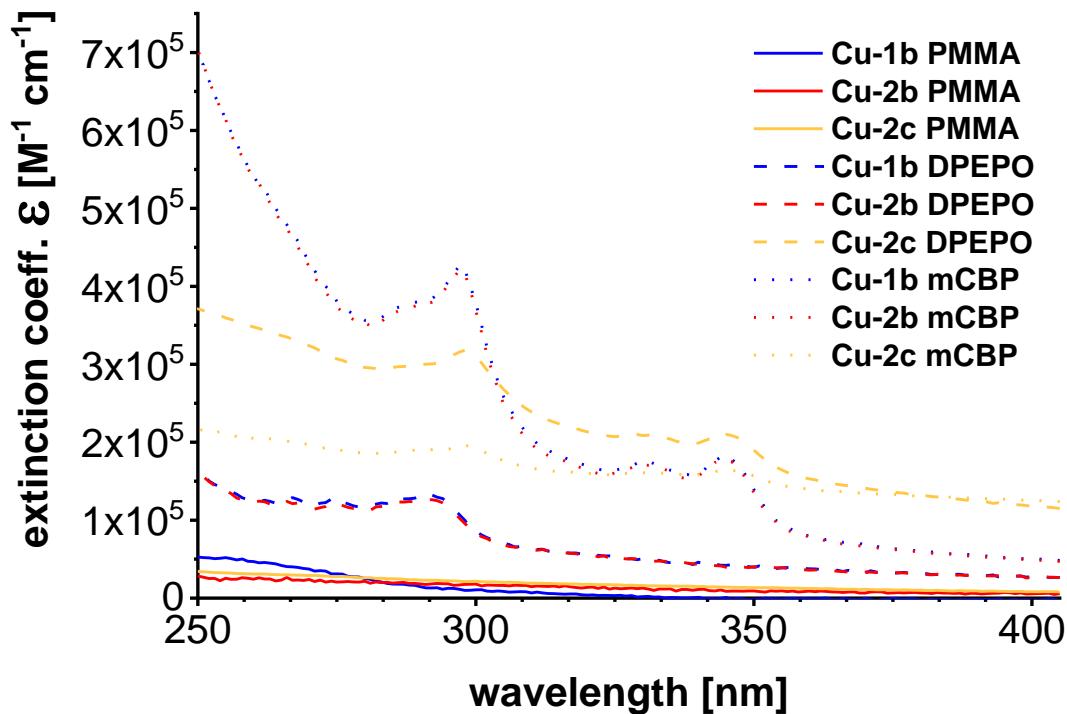


Figure S31. UV/Vis spectra of the complexes **Cu-1b**, **Cu-2b** and **Cu-2c**, measured of the PMMA, DPEPO and mCBP films respectively at room temperature.

Computational Data

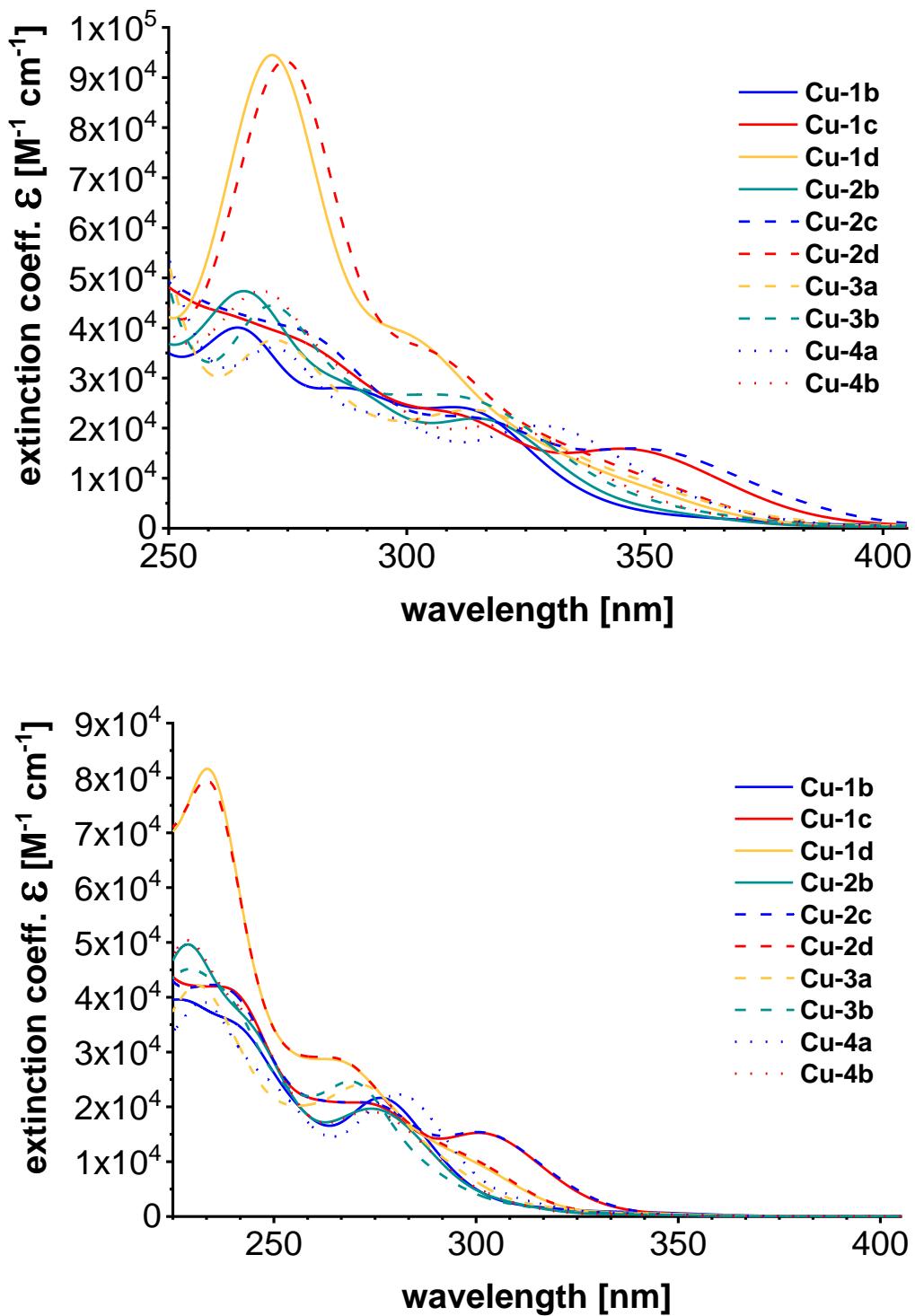


Figure S32. UV/Vis spectra computed using evGW/BSE^{8,9} with PBE0¹⁰ (top) and CAM-B3LYP¹¹ (bottom) with the def2-TZVP basis (def2-SV(P) for hydrogen).

7. Photophysics

Powder measurements

Photophysical measurements. Photoluminescence spectroscopy and TCSPC (*Time-correlated single-photon counting*). Steady-state emission spectra were measured by a *Horiba Scientific*, model FluoroMax-4 equipped with a 150 W Xenon-Arc lamp, excitation- and emissions monochromators and a Hamamatsu R928 photomultiplier and a time-correlated single-photon counting option. Emissions and excitation spectra were corrected using standard correction fits. Excited state lifetimes were determined employing the same system using the TCSPC method with FM-2013 equipment and a Horiba Yvon TCSPC hub. Excitation sources: NanoLED 370 (wavelength: 371 nm, pulse duration: 1.1 ns), NanoLED 290 (wavelength: 294 nm, pulse duration: <1 ns), SpectraLED 310 (wavelength: 314 nm), SpectraLED 355 (wavelength: 355 nm). Data analysis (exponential fit) was done using the software suite DataStation and DAS6 analysis software. The fit is specified using the chi-squared-test. For the photoluminescence quantum yield (PLQY) measurements an *Absolute PL Quantum Yield Measurement C9920-03G* system (*Hamamatsu Photonics*) was used. Quantum yields and CIE coordinates were determined using the software U6039-05 version 3.6.0. Emission maxima are given in nm, quantum yields Φ in % and CIE coordinates as x, y values. The PLQYs were measured with an integrating sphere set up of the powder of the Cu(I) complexes with an excitation wavelength of 350 nm. The yield was calculated using the following equation, wherein n_{photon} denotes the photon count and Int . the intensity.

$$\Phi_{PL} = \frac{n_{\text{photon}, \text{emitted}}}{n_{\text{photon}, \text{absorbed}}} = \frac{\int \frac{\lambda}{hc} [\text{Int}_{\text{emitted}}^{\text{sample}}(\lambda) - \text{Int}_{\text{absorbed}}^{\text{sample}}(\lambda)] d\lambda}{\int \frac{\lambda}{hc} [\text{Int}_{\text{emitted}}^{\text{reference}}(\lambda) - \text{Int}_{\text{absorbed}}^{\text{reference}}(\lambda)] d\lambda}$$

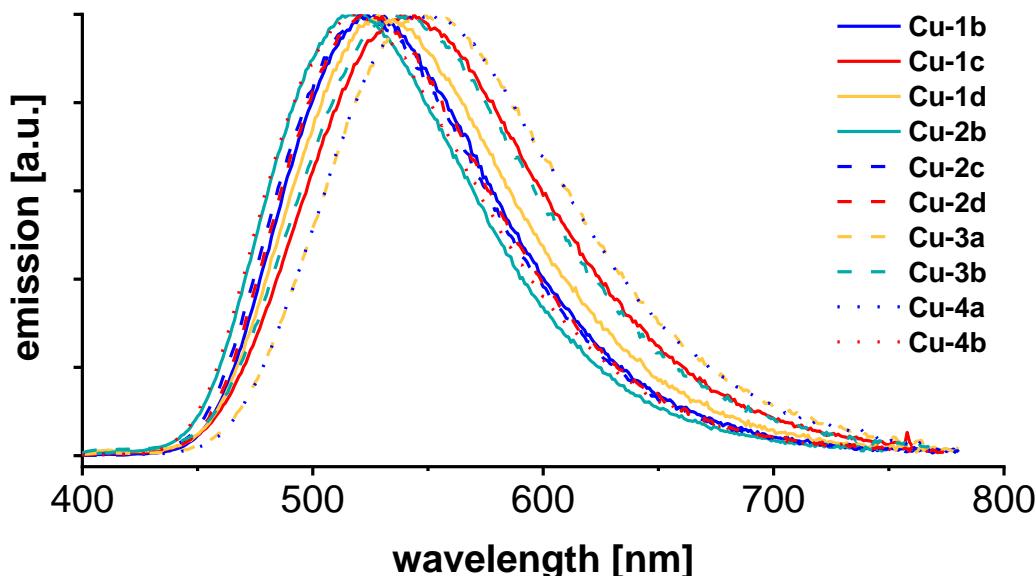


Figure S33. Overview of all emission spectra of the Cu(I) AlkylPyrPhos complexes, measured of the powder samples at room temperature with 350 nm excitation wavelength.

Measurements of the Cu(I) complexes in neat films

The photophysical measurements were also performed of neat films of 10 wt% of the copper complex emitters **Cu-1b**, **Cu-2b** and **Cu-2c** in the corresponding host materials. Hereby, poly(methylmethacrylate) (PMMA), purchased from *Sigma-Aldrich* (art. number 182230-25G, used without further purification), bis(2-(diphenylphosphino)phenyl)ether oxide (DPEPO), purchased also from *Sigma-Aldrich* (98% purity, sublimed before usage) and 3,3-di(9H-carbazol-9-yl)biphenyl (mCBP) purchased from *Jilin OLED materials* in >98% purity (sublimed before usage) served as host materials (**Figure S34**).

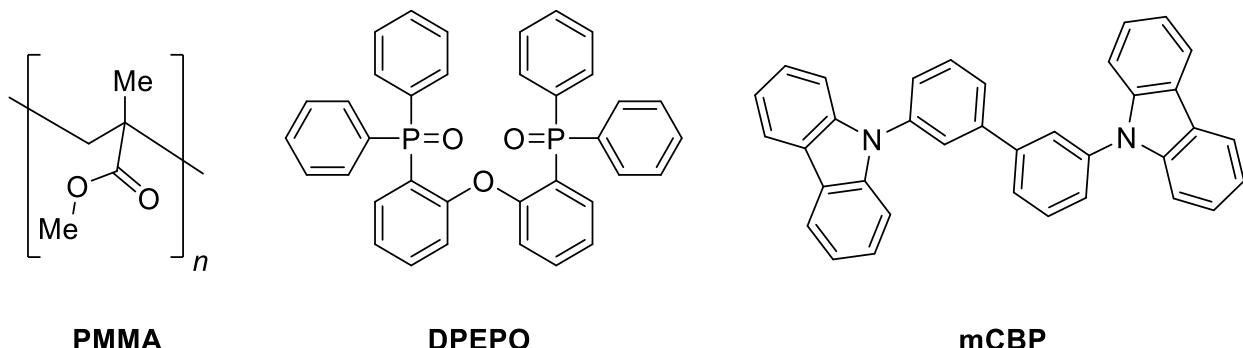


Figure S34. Host materials PMMA (left), DPEPO (center) and mCBP (right).

Sample Preparation. Stock solutions of the copper complexes and the host materials were prepared respectively. Dichloromethane was used as solvent.

Stock solution 1: 10 mg of the copper complex emitter was dissolved in 1 mL of DCM.

Stock solution 2: 10 mg of the host material was dissolved in 1 mL of DCM.

The stock solutions 1 and 2 were mixed by volume with an Eppendorf pipette in a ratio of 10/90 (stock solution 1/stock solution 2). In this case, the ratio by volume also corresponded to the ratio by weight, as both stock solutions had a concentration of 10 mg/mL.

Film Preparation. For the film preparation, 50 µL of the mixed solution (cf. section **sample preparation**) were applied on a commercially available fused quartz substrate with an Eppendorf pipette. Spin coating was performed with a Wafer Spinner Spin 150 (*APT GmbH*) with the following spin coating program:

Step 1: Speed 400 rpm, time 5 s, acceleration 1000

Step 2: Speed 1000 rpm, time 20 s, acceleration 1000

Step 3: Speed 4000 rpm, time 10 s, acceleration 1000

Subsequently, the spin-coated film was heated on a heating plate for 30 – 60 seconds at 70 °C. During the spin coating and due to the subsequent heating, the solvent was removed and a film with 10% by weight of copper complex emitter and 90% by weight PMMA (DPEPO, mCBP) was achieved. This film (nm scale thickness) was used to measure the PL, PLQY, lifetime and CIE coordinates were determined. For this the same equipment as described for the measurements in powder was used (see **photophysical measurements**). All emission spectra were plotted using the software ORIGIN®.

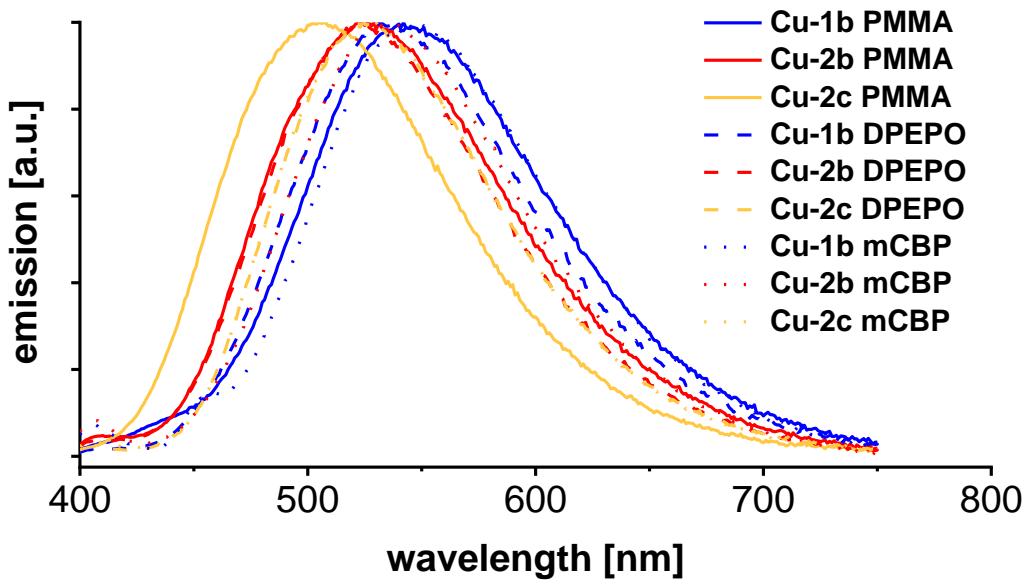


Figure S35. Overview of the emission spectra of the complexes **Cu-1b**, **Cu-2b** and **Cu-2c**, measured of the neat films (PMMA, DPEPO and mCBP as hosts) at room temperature with 350 nm excitation wavelength.

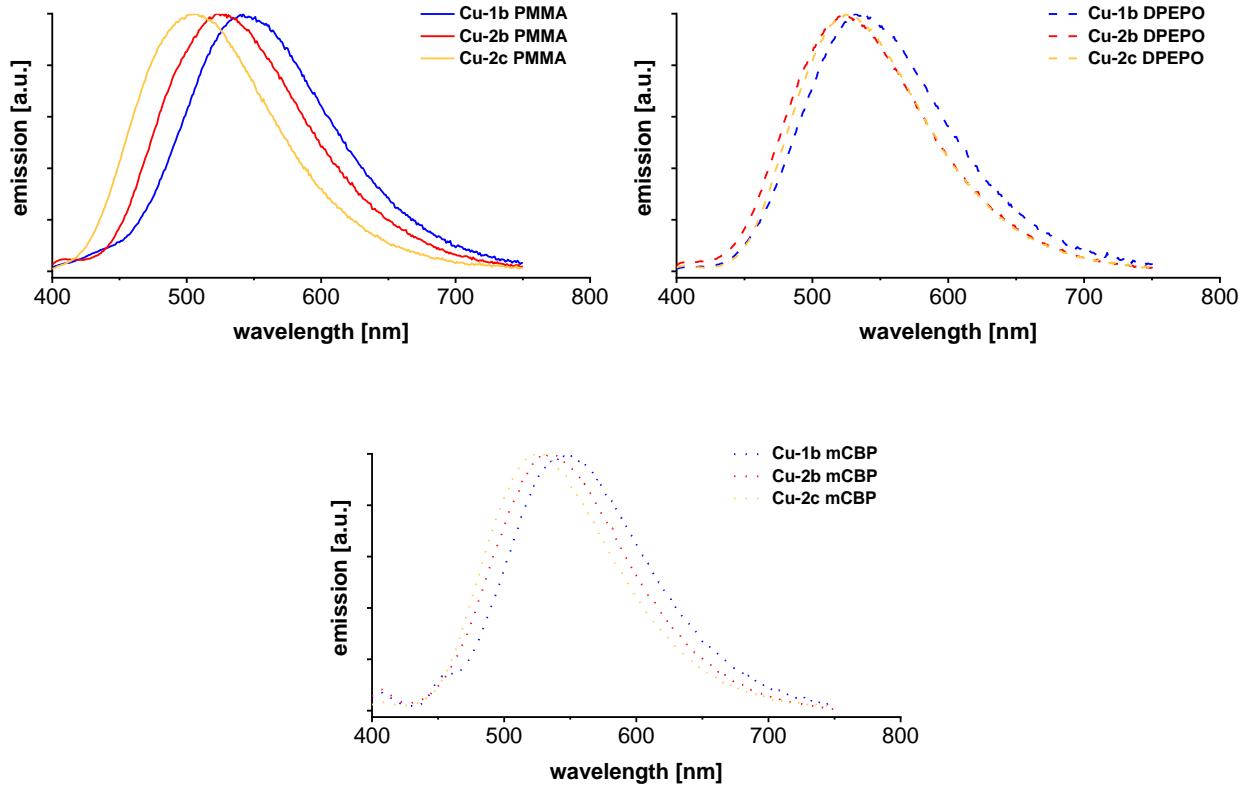


Figure S36. Comparison of the emission spectra of **Cu-1b**, **Cu-2b** and **Cu-2c** in PMMA (top, left), DPEPO (top, right) and mCBP (bottom) films, measured at rt with 350 nm excitation wavelength.

The photophysical properties of the complexes **Cu-1b**, **Cu-2b** and **Cu-2c**, measured of the neat films (PMMA, DPEPO and mCBP as hosts) at room temperature with 350 nm excitation wavelength, are given in **Table S3**.

Table S3. Photophysics and CIE-coordinates of 10wt% **Cu-1b**, **Cu-2b** and **Cu-2c** in PMMA, DPEPO and mCBP films, respectively, ordered according to the Cu(I) complex emitter (top) and host (bottom).

Complex	Host	λ_{PL} [nm]	Φ_{PL} [%]	τ [μs]	CIE X	CIE Y
Cu-1b	PMMA	542	59	9.5	0.39	0.53
	DPEPO	532	64	7.2	0.36	0.51
	mCBP	544	54	7.3	0.38	0.52
Cu-2b	PMMA	529	78	7.8	0.35	0.52
	DPEPO	526	65	6.9	0.32	0.52
	mCBP	529	60	6.7	0.36	0.53
Cu-2c	PMMA	505	77	10.0	0.27	0.44
	DPEPO	526	69	8.3	0.32	0.52
	mCBP	526	60	8.4	0.32	0.52

Complex	Host	λ_{PL} [nm]	Φ_{PL} [%]	τ [μs]	CIE X	CIE Y
Cu-1b	PMMA	542	59	9.5	0.39	0.53
Cu-2b		529	78	7.8	0.35	0.52
Cu-2c		505	77	10.0	0.27	0.44
Cu-1b	DPEPO	532	64	7.2	0.36	0.51
Cu-2b		526	65	6.9	0.32	0.52
Cu-2c		526	69	8.3	0.32	0.52
Cu-1b	mCBP	544	54	7.3	0.38	0.52
Cu-2b		529	60	6.7	0.36	0.53
Cu-2c		526	60	8.4	0.32	0.52

Measurements of the Cu(I) complexes in the KBr matrix

Pellet preparation. For the preparation of KBr pellets the copper complexes (0.9 mg) were mixed with dry KBr, (200 mg, stored in a compartment dryer at 80 °C, purchased from Merck) and ground to a homogenous mixture. This mixture was filled in an evacuable pellet die with a diameter of 13 mm and sintered at a pressure of 0.75 GPa at room temperature.

Photophysical measurements. Temperature dependent emission spectra of KBr pellets were recorded using a Fluoromax-2 (*Jobin-Yvon*) fluorescence spectrometer. Luminescence lifetimes were determined by TCSPC using a DeltaFlex (*Horiba Scientific*) spectrometer with an analogous setup as applied for powder samples. Excitation sources: NanoLED 390 (wavelength: 389 nm, pulse duration: 1.3 ns), NanoLED 350 (wavelength: 345 nm, pulse duration: <1 ns). Long-wave pass filters (cutoff at 500/510 nm) were set between sample and emission monochromator to suppress the influence of scattered excitation light. Decay curves were analysed by multiexponential fits with the software ORIGIN®.

Table S4. Temperature dependent emission maxima and lifetimes in the KBr matrix determined by TCSPC of **Cu-1b** and **Cu-2b**. A is the contribution of the two lifetime components.

Complex	T [K]	λ_{EX} [nm]	λ_{PL} [nm]	τ_1 [μs]	τ_2 [μs]	A ₁ [%]	A ₂ [%]
Cu-1b	290	345	529	/	5.0	/	100
		389	529	/	5.1	/	100
	20	345	533	0.4	35	1	99
		389	533	0.4	38	1	99
Cu-2b	290	345	515	/	3.9	/	100
		389	515	/	4.0	/	100
	20	345	519	0.8	33	1	99
		389	519	0.4	31	1	99

Table S5. Measured temperature-dependent emission maxima and deduced singlet-triplet energy gaps $\Delta E_{\text{ST}}(\text{exp})$.

	$\lambda_{\text{PL}}(290 \text{ K})$ [nm]	$\lambda_{\text{PL}}(20 \text{ K})$ [nm]	$\Delta E_{\text{ST}}(\text{exp})$ [cm ⁻¹]	$\Delta E_{\text{ST}}(\text{exp})$ [eV]
Cu-1b	527	533	214	0.027
Cu-2b	514	518	150	0.019

Measurements of the Cu(I) complexes in solution

Static emission spectra in solution were recorded using a Fluorolog 3-22 τ (*Horiba Jobin-Yvon*) fluorescence spectrometer. Spectroscopic grade solvents were purchased from *Merck*. The solutions were prepared using the common Schlenk technique with concentrations in the range of $2 \cdot 10^{-5}$ M and were measured in 1 cm x 1 cm quartz cuvettes. Lifetimes were determined by TCSPC using the same setup as used for KBr pellets. Excitation sources: NanoLED 350 (wavelength: 345 nm, pulse duration: <1 ns), NanoLED 320 (wavelength: 313 nm, pulse duration: <1 ns). The instrumental response function (IRF) of 125 ps (FWHM) was collected by use of LUDOX®. Decay curves were analysed by monoexponential fits with the software ORIGIN®.

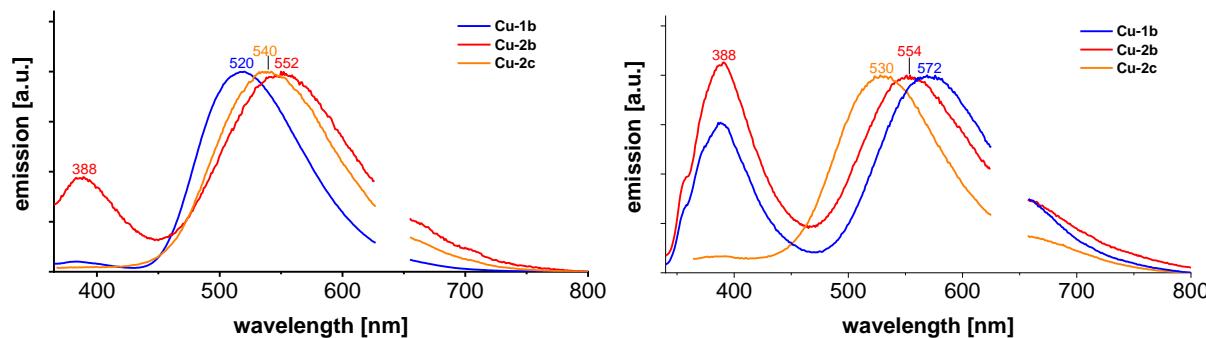


Figure S37. Emission spectra of **Cu-1b**, **Cu-2b** and **Cu-2c** in ethanol (left) and chlorobenzene (right) measured at room temperature with 320 nm excitation wavelength (second-order Rayleigh scattering cut out for clarity).

Table S6. Emission maxima and lifetimes of **Cu-1b**, **Cu-2b** and **Cu-2c** measured by TCSPC at room temperature.

Complex	Solvent	λ_{EX} [nm]	λ_{PL} [nm]	τ [μs]
Cu-1b	chlorobenzene	313	570	0.3
	DCM	345	578	0.3
	EtOH	345	518	4.9
Cu-2b	chlorobenzene	313	552	0.4
	DCM	345	558	0.2
	EtOH	313	550	0.2
Cu-2c	chlorobenzene	345	532	0.4
	DCM	345	538	0.3
	EtOH	345	532	0.1

Step-scan FTIR spectroscopy on the Cu(I) complexes

Step-scan FTIR measurements. The experimental setup has been described in detail before, so that only a brief description of the general setup is given here.¹²⁻¹⁴ All the time-resolved FTIR experiments were performed with the FTIR spectrometer Bruker Vertex 80v, operated in the step-scan mode. A liquid-nitrogen-cooled mercury cadmium telluride (MCT) detector (*Kolmar Tech.*, Model KV100-1-B-7/190) with a rise time of 25 ns, connected to a fast preamplifier and a 14-bit transient recorder board (Spectrum Germany, M3I4142, 400 MS/s), was used for signal detection and processing. The laser setup includes a Q-switched Nd:YAG laser (Innolas SpitLight Evo I) generating pulses with a band-width of 6–9 ns at a repetition rate of 100 Hz. The third harmonic (355 nm) of the Nd:YAG laser was used directly for sample excitation. The UV pump beam was attenuated to about 2.0 mJ per shot at a diameter of 9 mm. The beam was directed onto the sample and adjusted to have a maximal overlap with the IR beam of the spectrometer. The sample chamber was equipped with anti-reflection-coated germanium filters to prevent the entrance of laser radiation into the detector and interferometer compartments. The KBr pellets were prepared as described above (see **pellet preparation**) and cryogenically cooled (20 K or 290 K) with a closed cycle helium cryostat (ARS Model DE-202A). The cryo cooler was equipped with a homebuilt pellet holder and CaF₂ windows. The cryostat was hanging from a rack into the sample compartment without touching the spectrometer to prevent vibrations of the cooler to be transferred to the instrument. The temporal resolution of the 14-bit transient recorder board was set to 20 ns. The step-scan measurement was started 1.8 μ s before the Q-Switch of the Nd:YAG laser was triggered, which means that the experiment was initiated 2.2 μ s before the laser pulse reached the sample. Hence, this time was set as zero point in all spectra. The time delay between the start of the experiment and the laser pulse was controlled with a Stanford Research Systems DG535 delay generator. The spectral region was limited by undersampling to 988 – 1975 cm⁻¹ with a spectral resolution of 4 cm⁻¹ resulting in 555 interferogram points. An IR broad band filter (850 – 1750 cm⁻¹) and the CaF₂ windows (no IR transmission < 1000 cm⁻¹) of the cryostat prevented problems when performing a Fourier transformation (*i.e.* no IR intensity outside the measured region should be observed).

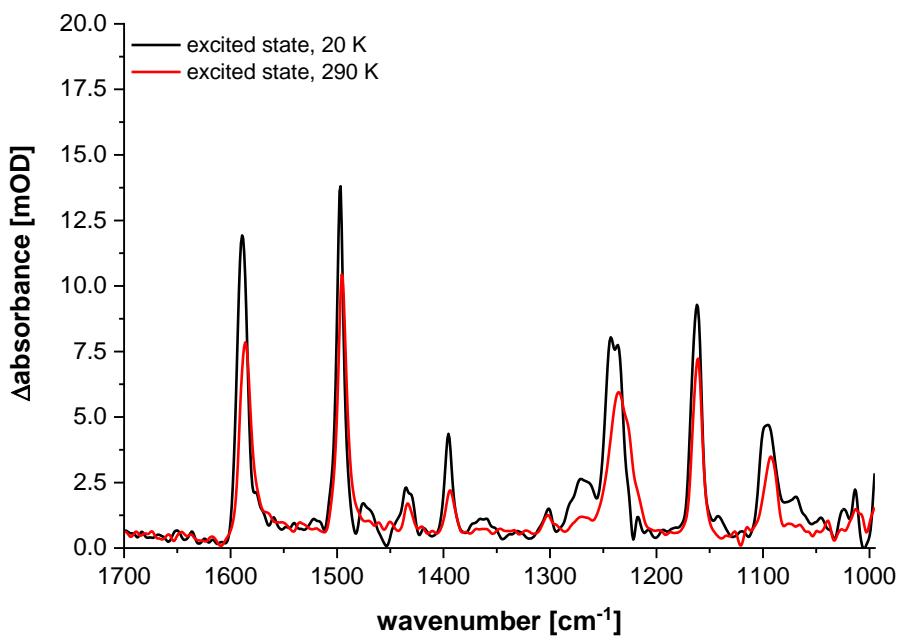


Figure S38. Excited state spectra at 20 K (black) and 290 K (red) of **Cu-2b**.

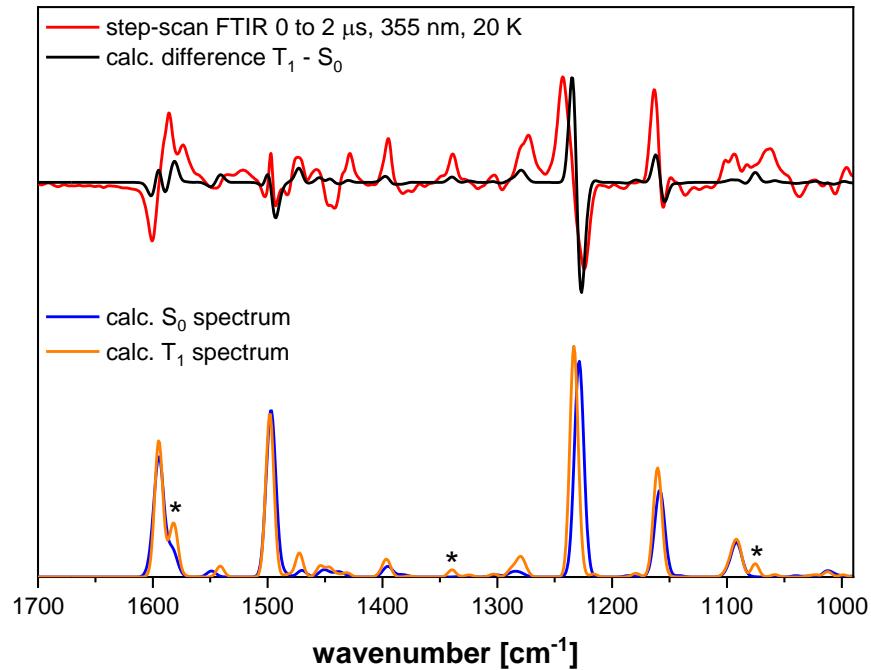


Figure S39. Step-scan difference spectrum 0 to 2 μs after excitation ($\lambda_{\text{EX}} = 355 \text{ nm}$) (red), calculated S_0 spectrum (blue), calculated T_1 spectrum (orange) (B3LYP/def2-TZVP, $\text{FWHM} = 8 \text{ cm}^{-1}$, Gaussian profile, scaled by 0.975) and calculated difference spectrum $T_1 - S_0$ (black) of **Cu-2b**. The T_1 specific bands are marked with asterisks.

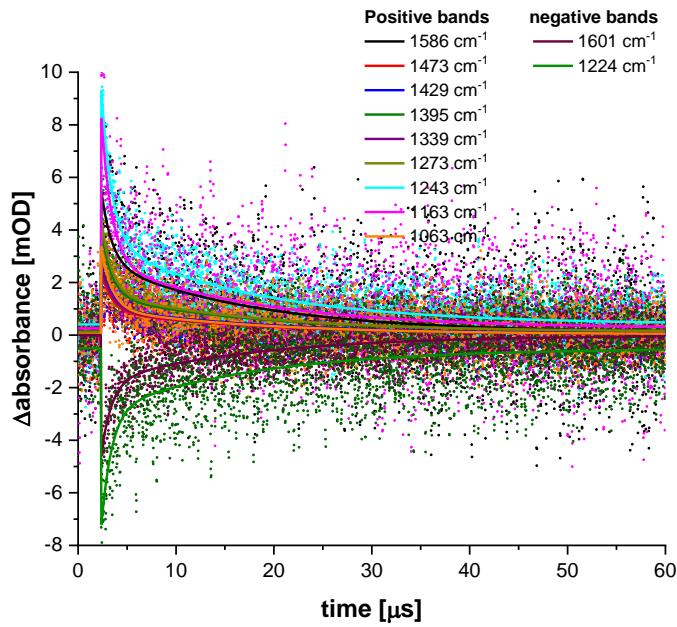


Figure S40. Time traces of the IR signal of nine positive and three negative bands (dots) in the step-scan spectrum of **Cu-2b** at 20 K with biexponential fits (convolution with Gaussian pulse), resulting in decay times of $\tau_1 = 1053 \pm 49$ ns and $\tau_2 = 15447 \pm 494$ ns.

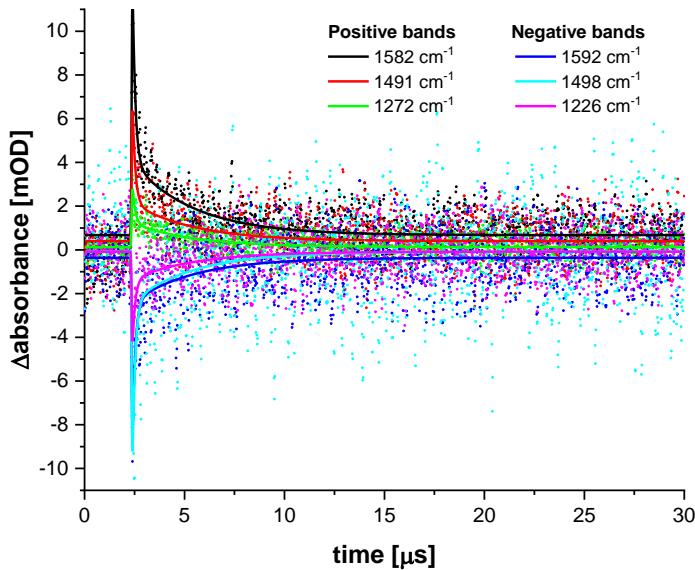
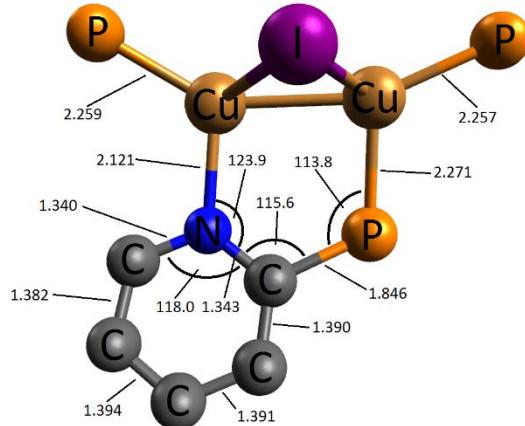


Figure S41. Time traces of the IR signal of nine positive and three negative bands (dots) in the step-scan spectrum of **Cu-2b** at 290 K with biexponential fits (convolution with Gaussian pulse), resulting in decay times of $\tau_1 = 139 \pm 16$ ns and $\tau_2 = 3083 \pm 209$ ns.

optimized S₀ geometry



optimized T₁ geometry

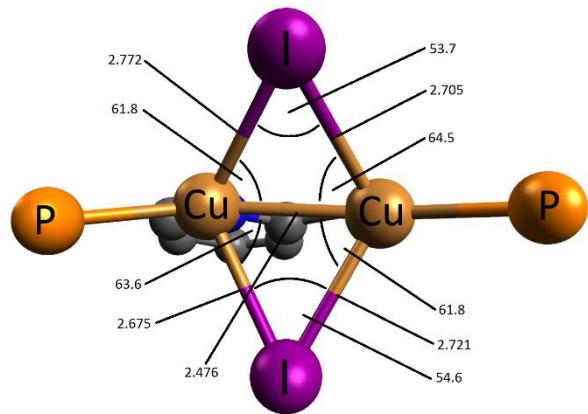
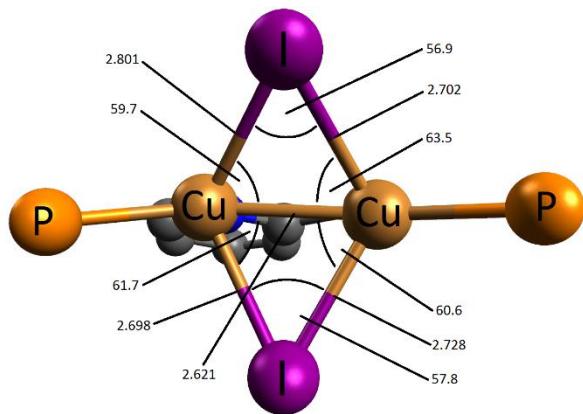
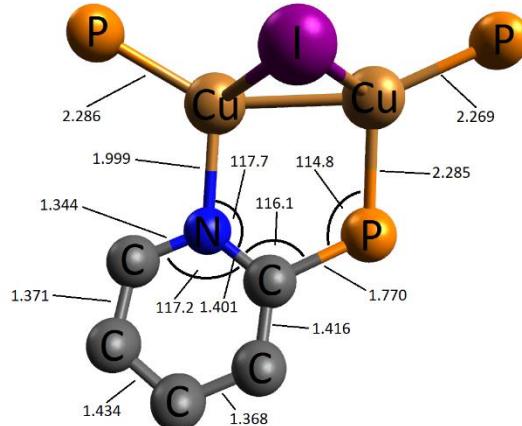


Figure S42. Bond lengths (Å) and angles (°) in the central Cu₂I₂ unit of **Cu-2b** in the S₀ and T₁ electronic state, calculated with B3LYP-D3(BJ)/def2-TZVP.

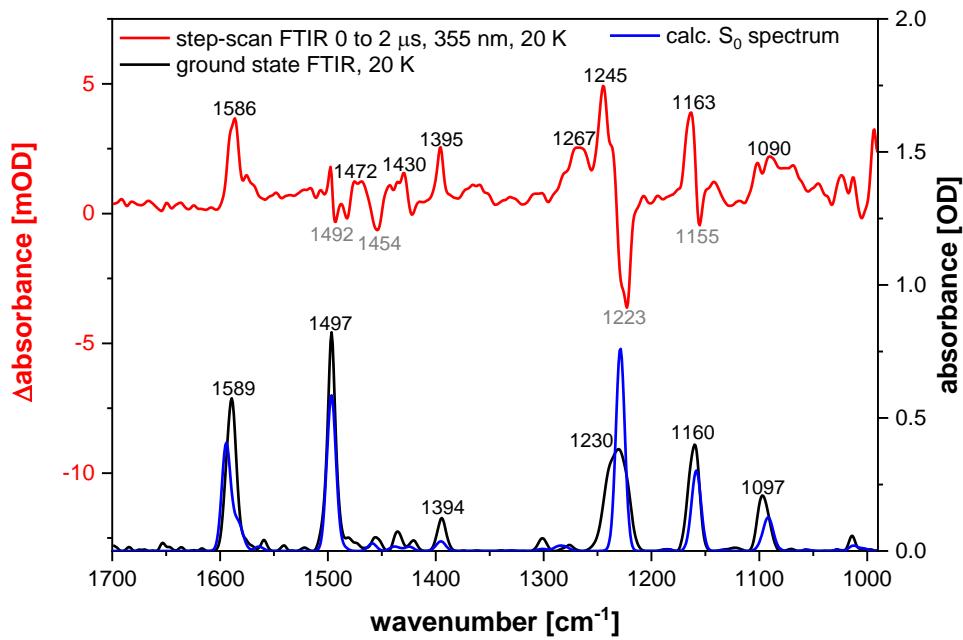


Figure S43. FTIR ground state spectrum at 20 K (black), calculated S_0 spectrum (B3LYP-D3(BJ)/def2-TZVP, FWHM = 8 cm^{-1} , Gaussian profile, scaled by 0.975) (blue) and step-scan difference spectrum 0 to 2 μs after excitation ($\lambda_{\text{EX}} = 355 \text{ nm}$) (red) of **Cu-1b**.

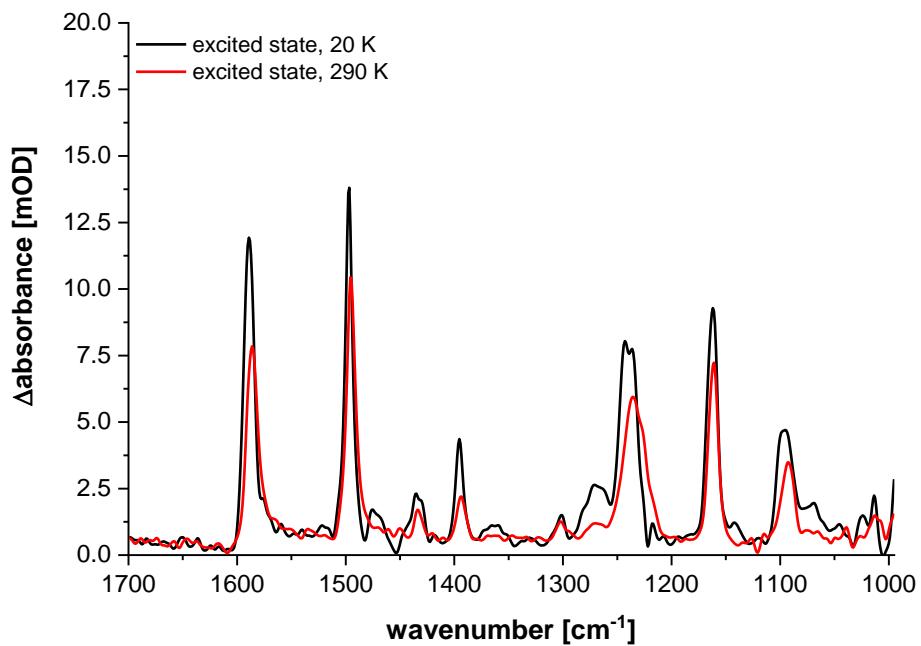


Figure S44. Excited state spectra at 20 K (black) and 290 K (red) of **Cu-1b**.

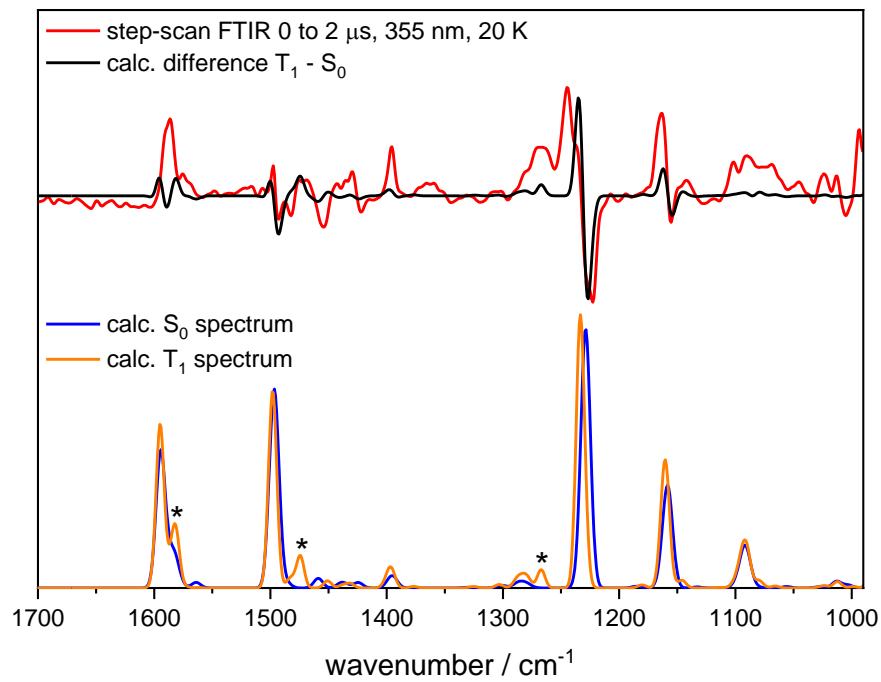
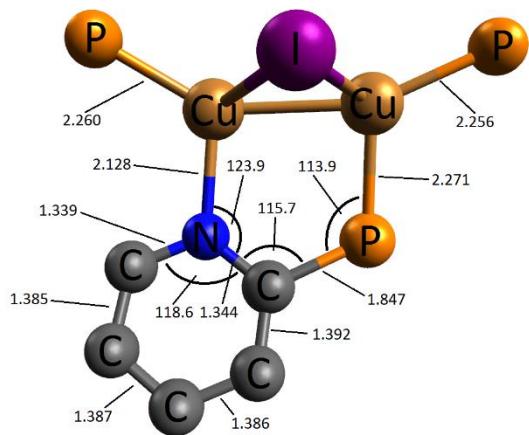


Figure S45. Step-scan difference spectrum 0 to 2 μs after excitation ($\lambda_{\text{EX}} = 355 \text{ nm}$) (red), calculated S_0 spectrum (blue), calculated T_1 spectrum (orange) (B3LYP-D3(BJ)/def2-TZVP, FWHM = 8 cm^{-1} , Gaussian profile, scaled by 0.975) and calculated difference spectrum $T_1 - S_0$ (black) of **Cu-1b**. The T_1 specific bands are marked with asterisks.

optimized S_0 geometry



optimized T₁ geometry

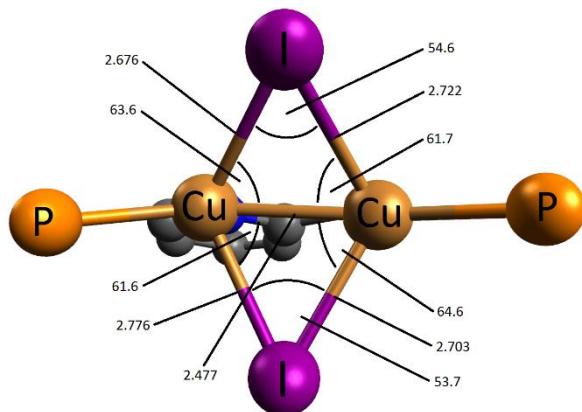
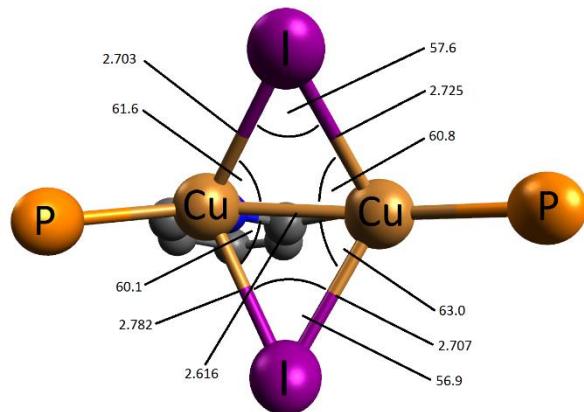
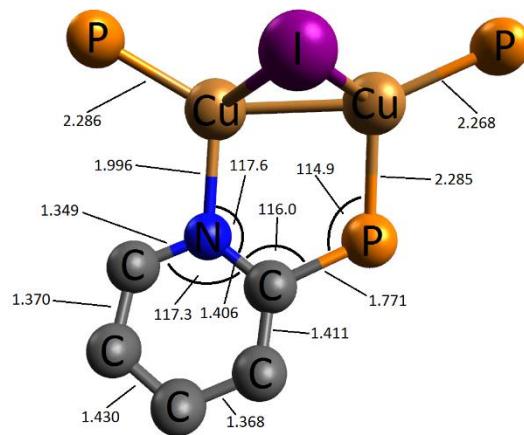


Figure S46. Bond lengths (\AA) and angles ($^{\circ}$) in the central Cu_2I_2 unit of **Cu-1b** in the S_0 and T_1 electronic state, calculated with B3LYP-D3(BJ)/def2-TZVP.

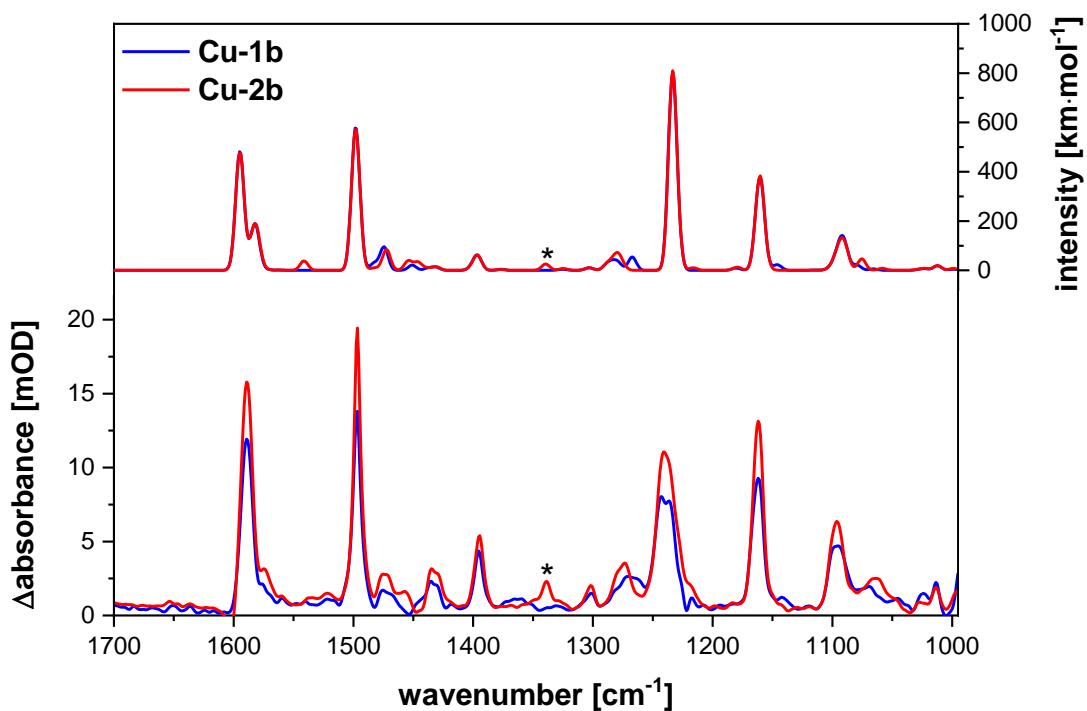
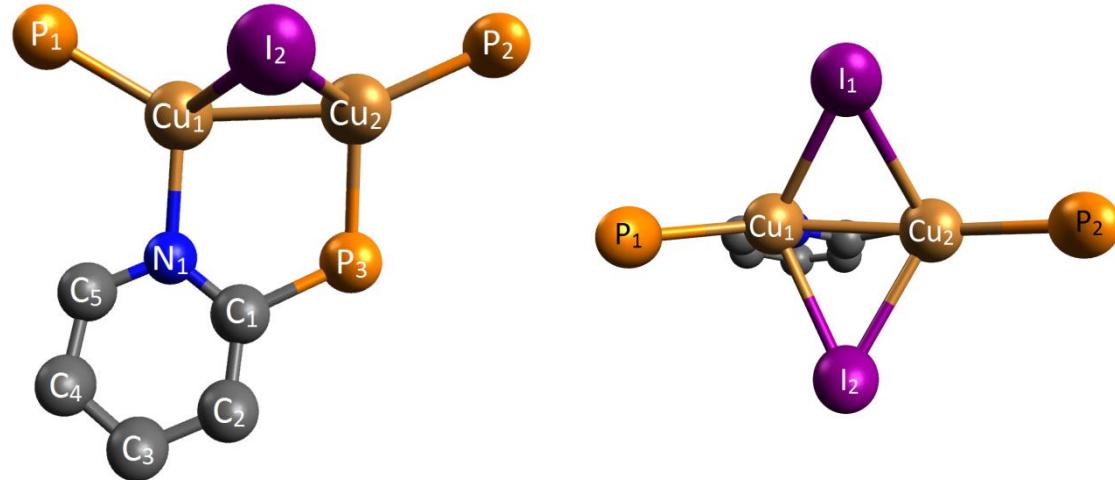


Figure S47. Excited state spectra at 20 K (bottom) and calculated T₁ spectra (top) (B3LYP-D3(BJ)/def2-TZVP, FWHM = 8 cm⁻¹, Gaussian profile, scaled by 0.975) of **Cu-1b** (blue) and **Cu-2b** (red). The band specific for **Cu-2b** is marked with an asterisk.

Table S7. Lifetimes of **Cu-1b** and **Cu-2b** determined by transient FTIR spectroscopy at 20 K and 290 K ($\lambda_{\text{EX}} = 355$ nm).

Complex	T [K]	τ_1 [μs]	τ_2 [μs]	A ₁ [%]	A ₂ [%]
Cu-1b	20	1.1	16	14	86
	290	0.1	2.4	19	81
Cu-2b	20	1.1	15	11	89
	290	0.1	3.1	13	87

Table S8. Change of geometrical parameters of **Cu-2b** and **Cu-1b** concerning the copper – iodide core with transition from the S_0 structure to the T_1 structure, calculated with B3LYP-D3(BJ)/def2-TZVP. The numbers of the atoms are shown in the figures above.



parameter	Differences	
	Cu-2b	Cu-1b
d(N1-C1)	+0.06 Å	+0.06 Å
d(Cu1-Cu2)	-0.15 Å	-0.14 Å
d(Cu1-N1)	-0.12 Å	-0.13 Å
$\angle(\text{Cu1}-\text{I1}-\text{Cu2})$	-3.2°	-3.0°
$\angle(\text{C19}-\text{I2}-\text{Cu2})$	-3.2°	-3.2°

Table S9. Characterization of vibrational modes of a) **Cu-2b** and b) **Cu-1b**. The calculated frequencies are indicated in brackets.

a)

$\tilde{\nu} / \text{cm}^{-1}$ ground state	$\tilde{\nu} / \text{cm}^{-1}$ excited state	Character of the vibration
1589 (1596)	1589 (1595)	Aromatic C–C stretching
	1575 (1580)	Aromatic C–C stretching
1497 (1498)	1497 (1498)	C–H scissoring in the fluorinated phenyl moieties
1394 (1396)	1394 (1396)	C–H scissoring, combined with aromatic C–C stretching in the fluorinated phenyl moieties
	1338 (1340)	C–H scissoring in the pyridine ring, combined with aliphatic C–H of the methyl group
1276 (1278)	1274 (1284)	C–C stretching in the phenyl rings
1231 (1229)	1241 (1234)	C–F stretching vibration
1160 (1156)	1162 (1162)	C–H scissoring
1097 (1092)	1096 (1091)	C–P stretching vibration
	1065 (1075)	C–H scissoring in the bridging ligand, combined with aliphatic C–H of the methyl group

b)

$\tilde{\nu} / \text{cm}^{-1}$ ground state	$\tilde{\nu} / \text{cm}^{-1}$ excited state	Character of the vibration
1589 (1593)	1589 (1594)	C–C stretching in the phenyl rings
	1578 (1581)	C–C stretching in the phenyl moieties
1497 (1497)	1497 (1498)	C–C stretching in the pyridine ring, combined with C–H scissoring
	1475 (1474)	C–H scissoring, combined with C–C stretching in the pyridine moiety
1394 (1396)	1395 (1397)	C–H scissoring in the fluorinated phenyl rings
1276 (1279)	1280 (1284)	Asymmetric C–C stretching in the phenyl moieties
	1271 (1268)	C–H scissoring, combined with C–C stretching in the pyridine moiety
1230 (1229)	1243 (1233)	C–F stretching vibration
1160 (1159)	1162 (1159)	C–H scissoring
1097 (1092)	1095 (1095)	C–P stretching vibration

8. References

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9. Structures in XYZ

The following structures were used for calculating electronic excitations for either the UV/Vis Spectra or vertical singlet-triplet gap.

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#Cu-1b singlet ground state
105

Cu    -1.1896460   -0.1358123   -0.0909001
Cu     1.4350184   -0.6022229   -0.0380301
I      0.2413979    0.0689196   -2.3795012
I      0.2570360    0.4598836    2.1180214
P     -1.4734687   -2.4026332    0.1445288
P     -2.8919959    1.4080169   -0.1447875
P      3.5884645    0.1807497   -0.0717902
F     -1.0947189    7.1258365   -0.0875361
F     -6.4807942    1.0864121   -4.9384822
F     -6.4693627    0.4726187    4.5754754
F      4.0745615    6.0073546   -1.3976299
F      6.4428511   -0.2870627    5.1809878
F      7.0031641   -2.7718233   -4.0309759
N      1.2627750   -2.7168160   -0.1179244
C      0.0814966   -3.3793210   -0.2142976
C      0.0225085   -4.7417664   -0.5672997
H     -0.9560619   -5.2395301   -0.6542665
C      1.2095016   -5.4494673   -0.8037901
H     1.1805861   -6.5164396   -1.0801449
C      2.4277655   -4.7677085   -0.6778883
H     3.3928014   -5.2694306   -0.8512261
C      2.4029503   -3.4101587   -0.3352118
H     3.3412510   -2.8383907   -0.2463996
C     -1.9554458   -2.9893891    1.8329263
C     -1.3858431   -4.1045797    2.4818715
H     -0.5930796   -4.6909519    1.9909501
C     -1.8163627   -4.4744155    3.7669338
H     -1.3554987   -5.3431868    4.2659589
C     -2.8253387   -3.7432496    4.4140263
H     -3.1611261   -4.0361454    5.4226276
C     -3.3957138   -2.6293063    3.7758796
H     -4.1786882   -2.0380236    4.2789901
C     -2.9544939   -2.2462476    2.5000486
H     -3.3842889   -1.3500658    2.0224930
C     -2.6755728   -3.2660522   -0.9730917
C     -3.8147121   -3.9524757   -0.5027729
H     -3.9935880   -4.0466669    0.5805664
C     -4.7250319   -4.5207000   -1.4109908
H     -5.6114294   -5.0545572   -1.0294100
C     -4.5066422   -4.4167542   -2.7939747
H     -5.2202174   -4.8679336   -3.5032553
C     -3.3728268   -3.7333614   -3.2695206
H     -3.1920076   -3.6459565   -4.3540576
C     -2.4663396   -3.1542308   -2.3687502
H     -1.5896160   -2.6008126   -2.7504929
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C	-2.3658903	3.1772465	-0.0599118
C	-1.1859344	3.5452735	-0.7438152
H	-0.5917365	2.7757297	-1.2677082
C	-0.7579609	4.8798771	-0.7648330
H	0.1566625	5.1790077	-1.3001179
C	-1.5075638	5.8455617	-0.0801255
C	-2.6740378	5.5082244	0.6185278
H	-3.2329908	6.2937838	1.1508068
C	-3.1007154	4.1712439	0.6220918
H	-4.0168772	3.9032698	1.1732125
C	-4.0000584	1.3648271	-1.6248293
C	-4.7646065	2.4755737	-2.0439495
H	-4.7009133	3.4300397	-1.4962505
C	-5.6038861	2.3865204	-3.1634148
H	-6.2016204	3.2456012	-3.5063369
C	-5.6762560	1.1740974	-3.8642900
C	-4.9262392	0.0573546	-3.4761416
H	-4.9965294	-0.8776338	-4.0536171
C	-4.0835551	0.1637710	-2.3600133
H	-3.4733991	-0.7043890	-2.0610021
C	-4.0671443	1.2340872	1.2726228
C	-5.3857469	0.7582589	1.1071733
H	-5.7836935	0.5759403	0.0960027
C	-6.2041157	0.5025759	2.2195499
H	-7.2325649	0.1263851	2.1026544
C	-5.6961554	0.7327365	3.5037216
C	-4.3957274	1.2180648	3.7002930
H	-4.0272856	1.3872148	4.7242362
C	-3.5835648	1.4593256	2.5841135
H	-2.5487418	1.8086255	2.7376739
C	3.7770138	1.9602674	-0.5354548
C	4.7462179	2.4277150	-1.4476223
H	5.4312782	1.7188845	-1.9401960
C	4.8489370	3.7958083	-1.7463713
H	5.5962767	4.1751554	-2.4608060
C	3.9787983	4.6947089	-1.1176453
C	3.0060313	4.2578564	-0.2075282
H	2.3356825	4.9920602	0.2661017
C	2.9013179	2.8888535	0.0722235
H	2.1229155	2.5346060	0.7713394
C	4.4948292	0.0802273	1.5370606
C	4.0537446	-0.8577151	2.4955025
H	3.1633769	-1.4738970	2.2836661
C	4.7120200	-0.9936605	3.7266213
H	4.3726558	-1.7149193	4.4862062
C	5.8107162	-0.1692793	4.0001411
C	6.2621459	0.7841965	3.0764837
H	7.1212116	1.4230375	3.3349654
C	5.5984236	0.9051564	1.8472028
H	5.9428998	1.6607413	1.1217122
C	4.7151646	-0.6921063	-1.2561018
C	4.2528639	-0.8743306	-2.5822778
H	3.2652547	-0.4772748	-2.8773587
C	5.0250022	-1.5660462	-3.5247519

H	4.6757325	-1.7091924	-4.5593769
C	6.2621646	-2.0976985	-3.1327810
C	6.7400159	-1.9475004	-1.8253686
H	7.7155482	-2.3800602	-1.5525994
C	5.9642464	-1.2401252	-0.8918666
H	6.3434845	-1.1169140	0.1352926

#Cu-1c singlet ground state
123

I	0.0192572	0.1078662	2.1651757
I	0.2390519	-0.3089439	-2.3365621
Cu	-1.2681004	-0.6083938	-0.1085904
Cu	1.3607625	-0.8599582	0.0686122
P	-1.4013540	-2.8862264	0.1533075
P	-3.0520904	0.8219991	-0.2891237
P	3.3990375	0.1753100	0.0980102
F	-6.8539121	-1.6703014	4.6389428
F	-6.9060427	0.3968000	5.3302823
F	-8.3755919	-0.3074672	3.8737031
F	-5.8806398	-0.6181812	-6.3141026
F	-6.4728385	1.4810989	-6.1659632
F	-7.6426997	-0.0580693	-5.1591880
F	6.8735260	-3.5441590	-4.3867456
F	8.5443759	-3.0485186	-3.0674007
F	7.9212823	-1.6490650	-4.6218702
F	-2.1125447	7.5340967	0.6968866
F	-0.0614571	6.8048428	0.5298752
F	-1.1931656	7.2905894	-1.2690537
F	3.8193253	6.4883111	-2.5199616
F	2.9341683	6.9295253	-0.5724642
F	1.6697075	6.2601487	-2.2174795
F	5.5151895	-0.2589427	6.5609946
F	7.4172121	-0.3409629	5.4977809
F	6.5377766	1.5894192	5.9984725
N	1.3710361	-2.9711633	0.0470815
C	0.2542412	-3.7378173	-0.0443990
C	0.3252261	-5.1233982	-0.2870564
H	-0.6041127	-5.7089357	-0.3687239
C	1.5759359	-5.7416839	-0.4215348
C	2.7267655	-4.9488504	-0.3093734
C	2.5740422	-3.5767879	-0.0761161
C	-2.4239561	-3.8734157	-1.0397026
C	-2.1567043	-3.6972751	-2.4183877
H	-1.3505310	-3.0129173	-2.7389586
C	-2.9163828	-4.3806421	-3.3800130
H	-2.6918458	-4.2398649	-4.4506021
C	-3.9593642	-5.2362789	-2.9822755
H	-4.5587147	-5.7687456	-3.7392048
C	-4.2336643	-5.4086160	-1.6161987
H	-5.0492208	-6.0781620	-1.2957010
C	-3.4697273	-4.7350064	-0.6470429
H	-3.6918493	-4.8837024	0.4219534
C	-1.9747889	-3.4615655	1.8135943

C	-3.2079416	-2.9479251	2.2759603
H	-3.7811525	-2.2437563	1.6493457
C	-3.7090848	-3.3166492	3.5333586
H	-4.6736973	-2.9080464	3.8762396
C	-2.9712227	-4.1833691	4.3577870
H	-3.3578770	-4.4632940	5.3515812
C	-1.7355169	-4.6806967	3.9158605
H	-1.1455154	-5.3516666	4.5621150
C	-1.2395487	-4.3268369	2.6497594
H	-0.2669697	-4.7267086	2.3224230
C	-4.3308858	0.5802741	1.0254385
C	-5.6535633	0.1811683	0.7387730
H	-5.9914535	0.0909108	-0.3056611
C	-6.5518111	-0.1096449	1.7771573
H	-7.5814443	-0.4214096	1.5423992
C	-6.1423075	0.0041958	3.1167297
C	-4.8293325	0.4179900	3.4131468
H	-4.5038171	0.5127078	4.4611041
C	-3.9288252	0.6940138	2.3777380
H	-2.8918877	0.9798413	2.6230803
C	-7.0760040	-0.3834754	4.2385604
C	-4.0325562	0.7319096	-1.8557837
C	-4.7603634	1.8291005	-2.3631513
H	-4.7438033	2.7964904	-1.8352019
C	-5.5017394	1.7017512	-3.5458569
H	-6.0596513	2.5644212	-3.9425951
C	-5.5274330	0.4738233	-4.2329914
C	-4.7970466	-0.6209336	-3.7391372
H	-4.7973417	-1.5778497	-4.2841212
C	-4.0483007	-0.4884198	-2.5613067
H	-3.4584698	-1.3441276	-2.1931387
C	-6.3774806	0.3210622	-5.4750123
C	-2.6135641	2.6164809	-0.1806449
C	-1.5615354	3.0686954	-1.0093936
H	-1.0377447	2.3594808	-1.6738876
C	-1.1632750	4.4097727	-0.9812900
H	-0.3440106	4.7534817	-1.6321411
C	-1.7932681	5.3140518	-0.1037593
C	-2.8392165	4.8727564	0.7224518
H	-3.3367206	5.5806942	1.4028189
C	-3.2526896	3.5313139	0.6791447
H	-4.0779887	3.2024191	1.3302874
C	-1.2990566	6.7413575	-0.0382589
C	4.6847940	-0.5883637	-0.9971271
C	4.3014938	-0.9172883	-2.3203823
H	3.2783609	-0.6880686	-2.6677953
C	5.2020434	-1.5478579	-3.1864308
H	4.8904578	-1.7987012	-4.2131013
C	6.4972953	-1.8791749	-2.7416776
C	6.8809569	-1.5728369	-1.4260317
H	7.8900146	-1.8373080	-1.0740887
C	5.9821457	-0.9289106	-0.5602394
H	6.3017813	-0.6921710	0.4669597
C	7.4647940	-2.5341671	-3.7011911

C	3.3420266	1.9433559	-0.4431002
C	4.0860414	2.4543187	-1.5243943
H	4.7765028	1.8032045	-2.0833870
C	3.9520969	3.7992682	-1.9054576
H	4.5327548	4.1911987	-2.7544844
C	3.0820618	4.6474068	-1.2019846
C	2.3449488	4.1467362	-0.1101970
H	1.6601310	4.8097943	0.4416913
C	2.4645025	2.8023830	0.2583892
H	1.8561940	2.4105561	1.0920838
C	2.8866866	6.0846374	-1.6269266
C	4.2490221	0.2833606	1.7391006
C	3.9267677	-0.6752228	2.7218072
H	3.1510326	-1.4313021	2.5136556
C	4.5603745	-0.6542990	3.9729530
H	4.2954245	-1.3995512	4.7387288
C	5.5150621	0.3365098	4.2592434
C	5.8266684	1.3115580	3.2931723
H	6.5598171	2.1000130	3.5254129
C	5.1972455	1.2843456	2.0413274
H	5.4425758	2.0564526	1.2937302
C	6.2421539	0.3343583	5.5861389
H	3.4560211	-2.9200725	0.0028948
H	3.7364824	-5.3773175	-0.4086357
H	1.6484216	-6.8252885	-0.6111411

#Cu-1d singlet ground state
105

I	7.8510738	2.4160014	4.6575193
I	10.4026676	-0.8613287	6.4073086
Cu	7.9823723	-0.1928451	5.3729919
Cu	10.1512276	1.0270518	4.4956613
Cl	0.4096269	-0.5691754	5.0711256
Cl	7.1800052	-5.6566139	10.7540636
Cl	7.1175263	4.5464725	11.0600436
Cl	11.2574339	8.0307833	1.5317757
Cl	11.8847830	4.3375803	11.0060815
Cl	17.6187570	0.2900491	3.4047613
P	7.9817791	-1.4485054	3.4603701
P	6.4913985	-0.4424726	7.0980012
P	11.8005350	2.5766893	4.8495815
N	10.3174077	-0.1287693	2.7318207
C	9.5158973	-1.1750265	2.4114373
C	9.7834462	-2.0010023	1.3028479
H	9.1084336	-2.8399621	1.0724417
C	10.9035823	-1.7473249	0.5010927
C	11.7250431	-0.6592369	0.8306811
H	12.6180314	-0.4078797	0.2370181
C	11.3920097	0.1166712	1.9465266
H	12.0188909	0.9740802	2.2417348
C	6.6761582	-1.2558597	2.1502749
C	6.1166909	0.0299329	1.9858260
H	6.4453960	0.8562816	2.6417813

C	5.1550462	0.2638225	0.9896311
H	4.7303867	1.2747091	0.8717550
C	4.7335019	-0.7822293	0.1530800
H	3.9732885	-0.5994986	-0.6245260
C	5.2834194	-2.0652912	0.3128547
H	4.9562083	-2.8930452	-0.3383942
C	6.2515601	-2.3023993	1.3016401
H	6.6670715	-3.3168815	1.4167858
C	8.0276941	-3.2641786	3.7889396
C	6.8165183	-3.9542711	4.0327023
H	5.8510379	-3.4333995	3.9190258
C	6.8301427	-5.3003754	4.4285839
H	5.8763910	-5.8241399	4.6069539
C	8.0510738	-5.9747831	4.6044523
H	8.0605898	-7.0325046	4.9156696
C	9.2581841	-5.2901697	4.3887314
H	10.2216709	-5.8047818	4.5389595
C	9.2501670	-3.9442018	3.9867718
H	10.2066409	-3.4169430	3.8425266
C	4.7150132	-0.5023861	6.5815870
C	4.3144648	0.3255777	5.5082003
H	5.0538386	0.9878004	5.0238285
C	2.9914665	0.3168868	5.0456483
H	2.6842350	0.9634325	4.2094523
C	2.0560092	-0.5413178	5.6495360
C	2.4344519	-1.3815923	6.7096950
H	1.6916696	-2.0517646	7.1692249
C	3.7593833	-1.3557827	7.1719606
H	4.0476612	-2.0193061	8.0031199
C	6.6865989	-1.9178228	8.1915235
C	7.2701400	-3.0840024	7.6542559
H	7.6270092	-3.0899775	6.6115068
C	7.4182574	-4.2400585	8.4342956
H	7.8759609	-5.1463442	8.0095702
C	6.9918869	-4.2259103	9.7724373
C	6.4232887	-3.0694379	10.3349483
H	6.1042105	-3.0725419	11.3885342
C	6.2744277	-1.9222982	9.5421419
H	5.8360779	-1.0147775	9.9892531
C	6.5737428	0.9636915	8.2932203
C	7.8043505	1.1828163	8.9536051
H	8.6585636	0.5075204	8.7724927
C	7.9668791	2.2670489	9.8245417
H	8.9249366	2.4355034	10.3396866
C	6.8981973	3.1600625	10.0185593
C	5.6688118	2.9626935	9.3710982
H	4.8422800	3.6712101	9.5341257
C	5.5106658	1.8611409	8.5141989
H	4.5434478	1.7127444	8.0075498
C	11.6579852	4.1766408	3.9347760
C	10.9223988	4.1997612	2.7311648
H	10.4141180	3.2836613	2.3850402
C	10.8006651	5.3795045	1.9822248
H	10.2209561	5.3966540	1.0468017

C	11.4100335	6.5556486	2.4500233
C	12.1326347	6.5614418	3.6565672
H	12.5935321	7.4951559	4.0138759
C	12.2514454	5.3735125	4.3918472
H	12.8115064	5.3834425	5.3416763
C	11.9204150	3.1258673	6.6101106
C	13.0567859	2.9176643	7.4171030
H	13.9600806	2.4476392	6.9963144
C	13.0547076	3.2949053	8.7697817
H	13.9384161	3.1211188	9.4027216
C	11.9095130	3.8950450	9.3161117
C	10.7709636	4.1271965	8.5242947
H	9.8795650	4.5974006	8.9673057
C	10.7773457	3.7306046	7.1813185
H	9.8671780	3.8781565	6.5745219
C	13.5128846	1.9970483	4.4412861
C	14.4463722	2.7606262	3.7080462
H	14.1912477	3.7806934	3.3786590
C	15.7085840	2.2386424	3.3833854
H	16.4329955	2.8368438	2.8093391
C	16.0493869	0.9419195	3.8019947
C	15.1353731	0.1619459	4.5327085
H	15.4117221	-0.8554000	4.8498258
C	13.8725118	0.6871340	4.8385329
H	13.1459118	0.0635173	5.3905239
H	11.1279812	-2.3865596	-0.3686418

#Cu-2b singlet ground state
108

N	-1.3480693	-2.5623783	0.1128592
C	-2.5405055	-3.1977642	0.1834321
C	-2.6742905	-4.5659581	0.4355873
C	-1.5245412	-5.3568119	0.6319942
C	-1.6150663	-6.8277818	0.9431436
C	-0.2880234	-4.6909786	0.5406579
C	-0.2292329	-3.3105639	0.2759718
P	1.4280902	-2.4443167	0.1417762
C	2.3472199	-3.2872564	1.5196073
C	3.3029407	-4.3079479	1.3273712
C	3.9659227	-4.8757301	2.4291513
C	3.6789285	-4.4387474	3.7323795
C	2.7249976	-3.4252014	3.9312438
C	2.0668511	-2.8474342	2.8345949
C	2.1527493	-3.1789078	-1.3904657
C	3.5044763	-2.8797988	-1.6784849
C	4.1016032	-3.3433042	-2.8597776
C	3.3518524	-4.0912123	-3.7841884
C	2.0025741	-4.3704503	-3.5179588
C	1.4048508	-3.9201433	-2.3288066
H	0.6505823	-5.2533796	0.6754679
H	3.5359996	-4.6636369	0.3107878
H	4.7129357	-5.6701241	2.2636425
H	2.4913215	-3.0748785	4.9506240

H	4.0962821	-2.2698465	-0.9750224
H	5.1584296	-3.1044222	-3.0642991
H	3.8185155	-4.4476411	-4.7173766
H	1.4020101	-4.9429518	-4.2442452
H	0.3427939	-4.1430025	-2.1414510
Cu	1.2545904	-0.1652858	0.1006742
Cu	-1.3861414	-0.4497552	-0.0629380
I	-0.0176323	0.1833673	-2.2796651
I	-0.2966855	0.4347254	2.2372000
P	2.9709761	1.3544921	0.1389489
P	-3.4586005	0.4898846	-0.3104195
C	4.3304392	1.1318382	-1.1005293
C	3.8710104	1.5244779	1.7418039
C	2.3469711	3.0610203	-0.1946225
C	-3.5029531	2.1410193	-1.1380635
C	-4.4509221	0.7557375	1.2286943
C	-4.5870767	-0.5335395	-1.3671493
C	5.6764632	1.4677473	-0.8371176
C	2.5847491	3.7315524	-1.4120519
C	4.4365376	2.7400109	2.1858956
C	5.1316896	2.8056156	3.4027096
C	-2.4649115	3.0576572	-0.8561083
C	-5.4309729	1.7648610	1.3503597
C	-5.8781871	-0.9422976	-0.9696392
C	-6.1640497	1.9160307	2.5366487
C	-2.4655509	4.3367585	-1.4301991
C	-6.6566560	-1.7718421	-1.7946060
C	1.9707853	4.9645297	-1.6847425
C	6.6686216	1.2764062	-1.8104340
C	3.9975070	0.5907515	-2.3632514
C	4.9789142	0.4073296	-3.3485483
C	6.3052820	0.7519425	-3.0579966
H	2.9519792	0.3058784	-2.5782717
H	4.7278333	-0.0112624	-4.3357016
H	7.7227821	1.5290224	-1.6156948
H	5.9629432	1.8794923	0.1438997
C	-4.0850368	-0.9783506	-2.6143147
C	-4.8580669	-1.7945646	-3.4500946
C	-6.1374081	-2.1847023	-3.0272660
H	-3.0678855	-0.6856391	-2.9304330
H	-4.4795517	-2.1418797	-4.4242872
H	-7.6655003	-2.0975550	-1.4956999
H	1.3289071	-2.0405795	2.9966957
H	-3.4323987	-2.5701665	0.0216818
H	-3.6823470	-5.0103135	0.4774335
H	-1.6377826	2.7635786	-0.1876776
C	-3.5031596	4.6883699	-2.3037922
H	-1.6639740	5.0638245	-1.2276105
C	-4.5361517	3.7952944	-2.6157082
C	-4.5305589	2.5211970	-2.0278340
H	-5.3280526	4.1051518	-3.3155213
H	-5.3382349	1.8129240	-2.2749805
H	-6.2882279	-0.6100166	-0.0022367
C	-4.2055795	-0.0907549	2.3323060

C	-4.9374636	0.0431513	3.5210852
C	-5.9097014	1.0478087	3.6071438
H	-3.4099554	-0.8521409	2.2690486
H	-4.7494540	-0.6056193	4.3906737
H	-5.6229148	2.4511811	0.5093968
H	3.2480328	3.2852922	-2.1698505
C	1.1196211	5.5229780	-0.7256428
H	2.1359344	5.4914431	-2.6373557
C	0.8761053	4.8892891	0.5002371
C	1.4850298	3.6542613	0.7573110
H	0.2015629	5.3598797	1.2327564
H	1.2676776	3.1364287	1.7062812
C	4.0058801	0.3751252	2.5499038
C	4.7056592	0.4243043	3.7634957
C	5.2602939	1.6430472	4.1750473
H	3.5449992	-0.5749240	2.2315355
H	4.8122700	-0.4671229	4.4011935
H	5.5738736	3.7470236	3.7648398
H	4.3275369	3.6536035	1.5787357
H	4.2000138	-4.8874485	4.5944324
F	7.2541846	0.5688430	-3.9943197
F	5.9252044	1.7024470	5.3424805
H	-6.9275641	2.7020069	2.6472493
F	-6.6090945	1.1898089	4.7471601
F	-3.5038960	5.9113465	-2.8639606
F	-6.8793208	-2.9774771	-3.8224685
F	0.4953029	6.6891187	-0.9933860
H	-0.6596651	-7.3520728	0.7301601
H	-2.4254425	-7.3182814	0.3613163
H	-1.8477225	-6.9854055	2.0218155

#Cu-2c singlet ground state
126

I	0.0020425	0.1481756	2.1991697
I	0.2370401	-0.3400653	-2.2954575
Cu	-1.2763591	-0.6047638	-0.0674294
Cu	1.3521163	-0.8580487	0.1241688
P	-1.4136814	-2.8790306	0.2279467
P	-3.0573065	0.8251711	-0.2774804
P	3.3899391	0.1740361	0.1424265
F	-6.8699454	-1.5813821	4.6856201
F	-6.9367274	0.5000228	5.3310885
F	-8.3949393	-0.2439951	3.8829216
F	-5.8696324	-0.7047871	-6.2882552
F	-6.4573244	1.3979137	-6.1751284
F	-7.6342516	-0.1224195	-5.1482602
F	6.8556456	-3.6311663	-4.2741685
F	8.5297470	-3.1123710	-2.9680056
F	7.9061008	-1.7429751	-4.5488739
F	-2.1127867	7.5515659	0.6030136
F	-0.0620795	6.8166024	0.4583619
F	-1.1839189	7.2745036	-1.3540111
F	3.8385069	6.4498277	-2.5600476

F	2.9438814	6.9194647	-0.6235796
F	1.6868603	6.2297029	-2.2658440
F	5.4927815	-0.1889348	6.6142291
F	7.3968756	-0.2836245	5.5557901
F	6.5174070	1.6525808	6.0332841
N	1.3564049	-2.9657431	0.1334205
C	0.2414541	-3.7341911	0.0458349
C	0.3102875	-5.1210990	-0.1775778
H	-0.6244374	-5.7008905	-0.2500356
C	1.5527304	-5.7714919	-0.3060586
C	2.6977435	-4.9585689	-0.1907670
H	3.7092728	-5.3885478	-0.2745897
C	2.5538383	-3.5850309	0.0243175
C	1.6509550	-7.2494265	-0.5775863
H	2.5141846	-7.7057371	-0.0466187
H	1.8029631	-7.4353931	-1.6661068
H	0.7258511	-7.7848603	-0.2767344
C	-2.4381551	-3.8799270	-0.9520980
C	-2.1660558	-3.7275241	-2.3326885
H	-1.3549533	-3.0530033	-2.6615819
C	-2.9271021	-4.4214656	-3.2856201
H	-2.6989737	-4.2986240	-4.3576920
C	-3.9763549	-5.2645784	-2.8775439
H	-4.5770205	-5.8050961	-3.6277211
C	-4.2553981	-5.4135283	-1.5097041
H	-5.0760189	-6.0728413	-1.1809366
C	-3.4901493	-4.7290467	-0.5491663
H	-3.7162084	-4.8593179	0.5214097
C	-1.9924043	-3.4268533	1.8956676
C	-3.2256632	-2.9044448	2.3477571
H	-3.7973256	-2.2103552	1.7086085
C	-3.7288406	-3.2513440	3.6105522
H	-4.6933612	-2.8357784	3.9452231
C	-2.9928279	-4.1046718	4.4504734
H	-3.3808733	-4.3672125	5.4484771
C	-1.7569934	-4.6104773	4.0185300
H	-1.1683589	-5.2706806	4.6770612
C	-1.2590561	-4.2785567	2.7472537
H	-0.2862962	-4.6839573	2.4272203
C	-4.3408401	0.6057607	1.0363304
C	-5.6619942	0.2001181	0.7518080
H	-5.9960153	0.0915952	-0.2921339
C	-6.5637261	-0.0739734	1.7917289
H	-7.5921291	-0.3909241	1.5585658
C	-6.1593878	0.0634429	3.1306374
C	-4.8478782	0.4837141	3.4246713
H	-4.5262385	0.5967075	4.4720298
C	-3.9438793	0.7430047	2.3880071
H	-2.9080803	1.0340329	2.6320825
C	-7.0967829	-0.3049095	4.2557985
C	-4.0330225	0.7121052	-1.8457101
C	-4.7618267	1.8008758	-2.3695854
H	-4.7488529	2.7751772	-1.8543819
C	-5.4996197	1.6562581	-3.5524943

H	-6.0582942	2.5123784	-3.9621740
C	-5.5207411	0.4192137	-4.2233656
C	-4.7893383	-0.6672316	-3.7130518
H	-4.7860985	-1.6313507	-4.2451578
C	-4.0440488	-0.5174241	-2.5350579
H	-3.4531509	-1.3667200	-2.1539791
C	-6.3669802	0.2486595	-5.4655953
C	-2.6170488	2.6208139	-0.1963082
C	-1.5619269	3.0584473	-1.0289649
H	-1.0371520	2.3380934	-1.6806125
C	-1.1617446	4.3992358	-1.0207495
H	-0.3398658	4.7314572	-1.6742837
C	-1.7930877	5.3181984	-0.1595836
C	-2.8422229	4.8914863	0.6702488
H	-3.3408506	5.6107901	1.3377589
C	-3.2574353	3.5500932	0.6469163
H	-4.0852126	3.2326739	1.3005988
C	-1.2968816	6.7455612	-0.1149341
C	4.6752681	-0.6090957	-0.9393994
C	4.2948115	-0.9502068	-2.2604252
H	3.2746274	-0.7178486	-2.6143207
C	5.1939861	-1.5972082	-3.1155583
H	4.8844983	-1.8572608	-4.1405988
C	6.4849552	-1.9333276	-2.6618403
C	6.8658428	-1.6145018	-1.3483930
H	7.8716103	-1.8821361	-0.9895176
C	5.9684558	-0.9539268	-0.4936714
H	6.2859785	-0.7075707	0.5319296
C	7.4498246	-2.6084966	-3.6096487
C	3.3407967	1.9347722	-0.4228705
C	4.0956822	2.4305785	-1.5037201
H	4.7906437	1.7712332	-2.0472962
C	3.9666226	3.7703558	-1.9042610
H	4.5555396	4.1502582	-2.7530726
C	3.0907002	4.6285994	-1.2206292
C	2.3429544	4.1433590	-0.1290919
H	1.6535401	4.8143877	0.4072405
C	2.4575873	2.8039919	0.2586380
H	1.8405831	2.4237071	1.0913343
C	2.9003172	6.0600717	-1.6665316
C	4.2371518	0.3006358	1.7836961
C	3.9093406	-0.6444917	2.7775500
H	3.1315242	-1.4004522	2.5767488
C	4.5401632	-0.6100487	4.0298055
H	4.2709603	-1.3447840	4.8042108
C	5.4975428	0.3810565	4.3059108
C	5.8146190	1.3429474	3.3284539
H	6.5498786	2.1317989	3.5525763
C	5.1880041	1.3021585	2.0755722
H	5.4376578	2.0640167	1.3189271
C	6.2218442	0.3931877	5.6341763
H	3.4425029	-2.9371503	0.1020992

#Cu-2d singlet ground state

I	7.8432185	2.4229043	4.6675738
I	10.3958736	-0.8587753	6.4084593
Cu	7.9760066	-0.1882464	5.3751648
Cu	10.1448068	1.0330692	4.4974844
Cl	0.4022371	-0.5604486	5.0857008
Cl	7.1800706	-5.6602555	10.7492464
Cl	7.1228700	4.5409602	11.0733710
Cl	11.2522501	8.0372698	1.5398634
Cl	11.8966977	4.3388402	11.0094699
Cl	17.6054441	0.2842198	3.3922234
P	7.9695735	-1.4378209	3.4579145
P	6.4877371	-0.4399256	7.1023839
P	11.7942198	2.5795860	4.8525467
N	10.3011072	-0.1145561	2.7313827
C	9.4980441	-1.1556474	2.4034437
C	9.7558390	-1.9709875	1.2864234
H	9.0707975	-2.8029875	1.0559503
C	10.8696380	-1.7349002	0.4607726
C	11.6868726	-0.6411728	0.8133986
H	12.5766557	-0.3819152	0.2163909
C	11.3691958	0.1272657	1.9352801
H	12.0046314	0.9797719	2.2265409
C	11.1816599	-2.6046337	-0.7286831
H	10.3452588	-3.2959399	-0.9630372
H	12.0893695	-3.2222328	-0.5387844
H	11.3927881	-1.9914519	-1.6324615
C	6.6545281	-1.2442370	2.1575055
C	6.0971145	0.0427844	1.9953862
H	6.4329128	0.8691080	2.6478245
C	5.1283709	0.2778128	1.0063915
H	4.7054733	1.2896734	0.8903996
C	4.6972369	-0.7683645	0.1748203
H	3.9311401	-0.5847951	-0.5968111
C	5.2449674	-2.0526154	0.3323239
H	4.9100918	-2.8805695	-0.3148251
C	6.2203839	-2.2907636	1.3137844
H	6.6344266	-3.3060167	1.4272989
C	8.0204099	-3.2541459	3.7824475
C	6.8124809	-3.9461469	4.0367085
H	5.8454021	-3.4259344	3.9339742
C	6.8312579	-5.2931165	4.4295175
H	5.8797978	-5.8182063	4.6161229
C	8.0545292	-5.9665338	4.5924299
H	8.0682067	-7.0247815	4.9017421
C	9.2587275	-5.2800105	4.3665670
H	10.2241559	-5.7937840	4.5071030
C	9.2453628	-3.9333832	3.9669969
H	10.1995110	-3.4044511	3.8138949
C	4.7102221	-0.4981570	6.5893497
C	4.3079947	0.3326965	5.5188091
H	5.0467377	0.9960082	5.0349302
C	2.9842386	0.3253910	5.0584054

H	2.6757718	0.9741429	4.2243748
C	2.0496856	-0.5343175	5.6615182
C	2.4297305	-1.3774886	6.7187677
H	1.6876191	-2.0488123	7.1777106
C	3.7554332	-1.3530377	7.1789463
H	4.0449835	-2.0187943	8.0078723
C	6.6838208	-1.9171069	8.1934512
C	7.2676770	-3.0820537	7.6538988
H	7.6242000	-3.0858612	6.6110223
C	7.4166591	-4.2394451	8.4318189
H	7.8746165	-5.1447596	8.0052907
C	6.9909121	-4.2278057	9.7701586
C	6.4221468	-3.0725755	10.3350277
H	6.1036497	-3.0776667	11.3887841
C	6.2724203	-1.9241098	9.5442867
H	5.8339979	-1.0175100	9.9932078
C	6.5725664	0.9639522	8.3002078
C	7.8047440	1.1817517	8.9580969
H	8.6585910	0.5069688	8.7732805
C	7.9693364	2.2641781	9.8308881
H	8.9286812	2.4315993	10.3439711
C	6.9010985	3.1567152	10.0293722
C	5.6701852	2.9607059	9.3844303
H	4.8440515	3.6688985	9.5508656
C	5.5100279	1.8609406	8.5255929
H	4.5416355	1.7136574	8.0208622
C	11.6526662	4.1806788	3.9392109
C	10.9143412	4.2057379	2.7373151
H	10.4040025	3.2905193	2.3918725
C	10.7926105	5.3862433	1.9895510
H	10.2106898	5.4049795	1.0555262
C	11.4047881	6.5611222	2.4568026
C	12.1301311	6.5650370	3.6616757
H	12.5931606	7.4978373	4.0186289
C	12.2488513	5.3763553	4.3958026
H	12.8110340	5.3847457	5.3444004
C	11.9194905	3.1280206	6.6130251
C	13.0588493	2.9214460	7.4162204
H	13.9614449	2.4529115	6.9922590
C	13.0606872	3.2983037	8.7690232
H	13.9467482	3.1257383	9.3990096
C	11.9163818	3.8965179	9.3192889
C	10.7747799	4.1269229	8.5313676
H	9.8840627	4.5954746	8.9775026
C	10.7773155	3.7306376	7.1882803
H	9.8648268	3.8765969	6.5845384
C	13.5049807	1.9978532	4.4401133
C	14.4380585	2.7602943	3.7052028
H	14.1838959	3.7810384	3.3771613
C	15.6985863	2.2363489	3.3770462
H	16.4226591	2.8336781	2.8016543
C	16.0380364	0.9387787	3.7939595
C	15.1245413	0.1599542	4.5264823
H	15.3999118	-0.8580469	4.8423701

C	13.8633128	0.6870959	4.8356476
H	13.1367692	0.0643413	5.3887006

#Cu-3a singlet ground state
117

I	0.3562667	10.2178972	10.3005664
I	-3.0663185	7.6387532	9.0637494
Cu	-2.2493751	9.6374711	10.7011044
Cu	-1.2257319	9.4338771	8.2579912
P	-3.5047336	11.3995129	9.9554830
P	-2.4442928	8.9149607	12.8724126
P	0.1506132	8.3916565	6.7494322
N	-2.3932867	10.9980201	7.4463483
C	-3.3608538	11.6780975	8.1046660
C	-4.2043455	12.5995918	7.4553977
H	-4.9684669	13.1156660	8.0532520
C	-4.0789093	12.8601490	6.0787208
C	-3.0567847	12.1452425	5.4161420
H	-2.8699875	12.2735076	4.3381543
C	-2.2557521	11.2466279	6.1224978
H	-1.4653822	10.6809376	5.6029369
C	-4.9765078	13.8503645	5.3202008
C	-6.0061734	14.5265686	6.2462401
H	-6.6916573	13.7871559	6.7158258
H	-6.6311297	15.2336370	5.6577599
H	-5.5150024	15.1093305	7.0563743
C	-5.7375184	13.0870181	4.2078657
H	-5.0441559	12.6042948	3.4850393
H	-6.3856834	13.7891741	3.6369762
H	-6.3861158	12.2928982	4.6393918
C	-4.0904300	14.9488205	4.6822984
H	-3.5283517	15.5114056	5.4600977
H	-4.7237235	15.6731471	4.1226211
H	-3.3518379	14.5255767	3.9670807
C	-5.3187945	11.1527098	10.1896779
C	-6.0929705	10.4662215	9.2273420
H	-5.6397514	10.1582572	8.2714500
C	-7.4374002	10.1530792	9.4875616
H	-8.0260419	9.6162280	8.7251015
C	-8.0270636	10.5131061	10.7103995
H	-9.0836219	10.2692388	10.9106899
C	-7.2579921	11.1770957	11.6818128
H	-7.7057895	11.4507434	12.6516316
C	-5.9130360	11.4870679	11.4294852
H	-5.3168111	11.9877543	12.2107349
C	-3.2057082	13.1338110	10.5576620
C	-4.2071155	14.1283807	10.6225057
H	-5.2485953	13.8881491	10.3525319
C	-3.8908132	15.4278343	11.0496469
H	-4.6839757	16.1926428	11.1017519
C	-2.5721142	15.7527104	11.4108144
H	-2.3263415	16.7744050	11.7456036
C	-1.5721420	14.7691813	11.3482025

H	-0.5349191	15.0127159	11.6328691
C	-1.8853662	13.4656978	10.9297033
H	-1.0959610	12.6938020	10.8850219
C	-2.4719724	10.3059286	14.0927156
C	-1.4081652	11.2376833	14.0543128
H	-0.5799870	11.0928406	13.3385813
C	-1.4042599	12.3462411	14.9132992
H	-0.5661358	13.0619347	14.8739851
C	-2.4675596	12.5522967	15.8100229
C	-3.5324544	11.6384378	15.8451370
H	-4.3709954	11.7922811	16.5451473
C	-3.5355244	10.5186959	14.9949316
H	-4.3733612	9.8041395	15.0356709
C	-1.0545576	7.8390186	13.4487919
C	-0.4262366	7.9815328	14.7028210
H	-0.7418485	8.7831607	15.3904367
C	0.6078545	7.1073133	15.0793737
H	1.0956805	7.2302474	16.0609566
C	1.0168703	6.0816994	14.2121439
C	0.3944424	5.9377511	12.9600972
H	0.7117980	5.1374737	12.2706229
C	-0.6281625	6.8164321	12.5727336
H	-1.1002282	6.7127981	11.5798573
C	-3.9369271	7.9135146	13.3144578
C	-5.0866332	8.0200628	12.5048729
H	-5.0683149	8.6651625	11.6115622
C	-6.2469340	7.2936226	12.8206486
H	-7.1366815	7.3870795	12.1763325
C	-6.2649673	6.4445549	13.9384274
C	-5.1179578	6.3226324	14.7423160
H	-5.1230634	5.6497357	15.6162255
C	-3.9592581	7.0515910	14.4327877
H	-3.0612134	6.9441190	15.0635186
C	-0.6217376	8.1312127	5.0838825
C	-1.8998876	7.5239798	5.0475662
H	-2.3773119	7.1992509	5.9891819
C	-2.5667051	7.3457821	3.8266332
H	-3.5591425	6.8648646	3.8147752
C	-1.9802496	7.7864170	2.6265059
C	-0.7190258	8.4023564	2.6559460
H	-0.2510093	8.7509394	1.7198074
C	-0.0398392	8.5715179	3.8756146
H	0.9534589	9.0487523	3.8849043
C	0.7145300	6.6960510	7.2299494
C	1.1435398	6.5062239	8.5622144
H	1.1017772	7.3469264	9.2770040
C	1.6163105	5.2526820	8.9785025
H	1.9569911	5.1205025	10.0192112
C	1.6470191	4.1714690	8.0809629
C	1.2061252	4.3510382	6.7601384
H	1.2209890	3.5061219	6.0512853
C	0.7448375	5.6079894	6.3330959
H	0.4019507	5.7391748	5.2938416
C	1.7400767	9.2547404	6.3498907

C	1.8072006	10.6501164	6.5481671
H	0.9383151	11.1801775	6.9744980
C	2.9799232	11.3577389	6.2373604
H	3.0207835	12.4471923	6.4025584
C	4.1024573	10.6759987	5.7402420
C	4.0495663	9.2835016	5.5559030
H	4.9323090	8.7417789	5.1766799
C	2.8759878	8.5752844	5.8595053
H	2.8444148	7.4818450	5.7199970
H	2.0101843	3.1846950	8.4136639
H	5.0272563	11.2295591	5.5063739
H	1.8273735	5.3957535	14.5102655
H	-7.1727583	5.8667107	14.1805889
H	-2.4661826	13.4283793	16.4798113
H	-2.5087517	7.6504225	1.6681076

#Cu-3b singlet ground state

117

I	0.3607996	10.2107089	10.3048163
I	-3.0481729	7.6230527	9.0701871
Cu	-2.2481739	9.6375740	10.6980741
Cu	-1.2257097	9.4361091	8.2584329
P	-3.5068911	11.3971740	9.9529719
P	-2.4448024	8.9141621	12.8675665
P	0.1517085	8.3955100	6.7517435
F	-2.4188007	13.5967497	16.6182516
F	1.9689296	5.2323204	14.5640393
F	-7.3799851	5.7793022	14.1982367
F	-2.6316625	7.5712920	1.4955124
F	2.1237575	3.0097189	8.4968297
F	5.1899427	11.3785278	5.4517046
N	-2.3919195	10.9987212	7.4457209
C	-3.3606110	11.6781383	8.1033434
C	-4.2023850	12.6012593	7.4543851
H	-4.9677726	13.1162757	8.0515618
C	-4.0735890	12.8656107	6.0786451
C	-3.0498213	12.1520933	5.4168708
H	-2.8602385	12.2843020	4.3398752
C	-2.2513034	11.2510148	6.1227611
H	-1.4595774	10.6863741	5.6040099
C	-4.9687530	13.8580270	5.3205775
C	-5.9992221	14.5334422	6.2462698
H	-6.6868925	13.7940684	6.7127335
H	-6.6221094	15.2424273	5.6580811
H	-5.5089606	15.1142462	7.0583823
C	-5.7283782	13.0970863	4.2055292
H	-5.0345245	12.6155531	3.4823843
H	-6.3749073	13.8010009	3.6351636
H	-6.3786446	12.3027955	4.6341795
C	-4.0801951	14.9566780	4.6862720
H	-3.5185636	15.5172980	5.4657979
H	-4.7121533	15.6826703	4.1274618
H	-3.3414779	14.5344679	3.9705867

C	-5.3201220	11.1424757	10.1839985
C	-6.0862451	10.4467989	9.2217330
H	-5.6278645	10.1383531	8.2684145
C	-7.4295430	10.1258629	9.4784003
H	-8.0121165	9.5824568	8.7159982
C	-8.0260720	10.4875027	10.6975067
H	-9.0818436	10.2381222	10.8947859
C	-7.2652733	11.1615571	11.6685095
H	-7.7193170	11.4381457	12.6345487
C	-5.9213335	11.4791960	11.4197980
H	-5.3319399	11.9890632	12.2002218
C	-3.2104253	13.1288490	10.5613035
C	-4.2130788	14.1221013	10.6278170
H	-5.2538009	13.8823291	10.3546202
C	-3.8987760	15.4200445	11.0609766
H	-4.6927598	16.1837207	11.1152384
C	-2.5810799	15.7448772	11.4257469
H	-2.3371925	16.7651289	11.7658915
C	-1.5795448	14.7631368	11.3598842
H	-0.5430139	15.0074611	11.6462704
C	-1.8907286	13.4609698	10.9357638
H	-1.1000727	12.6906587	10.8881257
C	-2.4576270	10.3009069	14.0911216
C	-1.3962428	11.2359201	14.0411059
H	-0.5769484	11.0968071	13.3141237
C	-1.3735690	12.3439703	14.8978146
H	-0.5518043	13.0764751	14.8690054
C	-2.4314888	12.5288104	15.7993858
C	-3.5011036	11.6283016	15.8640453
H	-4.3148945	11.8055367	16.5847521
C	-3.5072893	10.5131566	15.0098051
H	-4.3459550	9.8006707	15.0633969
C	-1.0644221	7.8291776	13.4431004
C	-0.4434242	7.9572012	14.7027794
H	-0.7600723	8.7519463	15.3976729
C	0.5866310	7.0836074	15.0864766
H	1.0850159	7.1734992	16.0644232
C	0.9854909	6.0754600	14.2004475
C	0.3869743	5.9268750	12.9415837
H	0.7303643	5.1247642	12.2694941
C	-0.6298692	6.8128747	12.5622565
H	-1.0931052	6.7154851	11.5646019
C	-3.9445586	7.9259972	13.3041366
C	-5.0875881	8.0285340	12.4830117
H	-5.0622307	8.6674389	11.5855914
C	-6.2554048	7.3114172	12.7833804
H	-7.1507555	7.3833505	12.1467996
C	-6.2668227	6.4764343	13.9068868
C	-5.1423614	6.3417563	14.7339559
H	-5.1868527	5.6657363	15.6022397
C	-3.9836384	7.0683276	14.4260791
H	-3.0943651	6.9574906	15.0683543
C	-0.6251522	8.1172485	5.0922364
C	-1.9147856	7.5325890	5.0662718

H	-2.4006656	7.2361595	6.0130414
C	-2.5910868	7.3368627	3.8550659
H	-3.5921615	6.8787818	3.8237858
C	-1.9812355	7.7469951	2.6605171
C	-0.7133138	8.3403845	2.6544357
H	-0.2653296	8.6506305	1.6972262
C	-0.0376775	8.5193414	3.8732558
H	0.9630865	8.9804233	3.8680253
C	0.7292826	6.7078573	7.2366194
C	1.1627003	6.5215800	8.5690031
H	1.1199567	7.3625856	9.2834920
C	1.6438090	5.2766878	8.9953919
H	1.9931074	5.1192587	10.0280187
C	1.6700630	4.2084674	8.0880872
C	1.2312237	4.3597555	6.7671422
H	1.2596174	3.4959265	6.0846203
C	0.7642799	5.6148661	6.3454201
H	0.4184349	5.7365104	5.3061328
C	1.7307684	9.2688164	6.3444663
C	1.7977590	10.6625769	6.5592965
H	0.9323787	11.1875795	6.9983440
C	2.9606145	11.3843451	6.2521029
H	3.0282791	12.4700608	6.4225383
C	4.0681491	10.6959376	5.7412397
C	4.0397579	9.3095682	5.5338240
H	4.9362848	8.8027676	5.1435209
C	2.8683696	8.6016909	5.8391003
H	2.8447770	7.5097828	5.6873179

#Cu-4a singlet ground state
121

I	3.6051093	11.1169097	6.8998190
I	4.8763476	10.4106122	2.6053267
Cu	4.1818184	12.2229816	4.4943445
Cu	5.3271831	9.9164518	5.2020057
P	6.2129595	13.2282316	4.8481537
P	2.2865208	13.2833916	3.7408091
P	5.1935351	7.6450344	5.4630393
N	7.1881932	10.7812403	5.7250214
C	7.4563302	12.1104528	5.6931100
C	8.6194085	12.6488585	6.2725894
H	8.7877449	13.7373610	6.2302817
C	9.5650091	11.8164664	6.9014992
C	9.2820236	10.4365227	6.8961209
H	9.9785741	9.7136997	7.3524014
C	8.1038069	9.9713231	6.3036094
H	7.8754483	8.8928076	6.2849852
C	10.7901485	12.3762836	7.5834734
H	11.0968474	13.3246230	7.0857402
H	11.6343459	11.6592366	7.4684798
C	10.5777659	12.6534031	9.0871129
H	10.2367835	11.7041779	9.5675462
C	9.5573179	13.7542393	9.4330540

H	9.7810450	14.6630825	8.8247948
H	8.5040896	13.4713710	9.2182296
C	9.8189080	14.0292649	10.9267431
H	9.4921279	15.0456536	11.2339924
H	9.2372419	13.3090607	11.5437263
C	11.3471580	13.7967583	11.1281868
H	11.5353239	13.1398546	12.0045241
H	11.8862822	14.7470966	11.3302914
C	11.8504301	13.1403187	9.8139720
H	12.5822768	12.3211767	9.9862358
H	12.3596217	13.8969247	9.1731618
C	6.3289175	14.7178443	5.9527014
C	6.9030510	15.9457682	5.5615922
H	7.3430751	16.0541346	4.5571274
C	6.9190816	17.0362390	6.4489330
H	7.3703062	17.9904770	6.1288701
C	6.3696793	16.9131518	7.7351920
H	6.3866055	17.7697189	8.4295505
C	5.7943295	15.6921720	8.1309349
H	5.3561121	15.5853420	9.1374818
C	5.7664406	14.6032147	7.2463387
H	5.2937048	13.6534071	7.5561102
C	7.0984612	13.7572057	3.3136690
C	6.3964259	14.6191713	2.4406606
H	5.3752866	14.9505097	2.6946721
C	6.9798129	15.0464807	1.2389880
H	6.4142973	15.7187006	0.5728357
C	8.2623091	14.5990211	0.8783930
H	8.7160015	14.9237246	-0.0727677
C	8.9563907	13.7241240	1.7284275
H	9.9568875	13.3555924	1.4465963
C	8.3816543	13.3069071	2.9408518
H	8.9398500	12.6156128	3.5915484
C	0.8898004	12.1561267	3.2831929
C	0.5940335	11.0964321	4.1690260
H	1.2089811	10.9532871	5.0753739
C	-0.4744813	10.2272229	3.9008468
H	-0.6995198	9.4096782	4.6063205
C	-1.2481682	10.3927078	2.7387444
H	-2.0836870	9.7049869	2.5255764
C	-0.9504013	11.4365690	1.8482306
H	-1.5499795	11.5713300	0.9323456
C	0.1103383	12.3179778	2.1190448
H	0.3323047	13.1355378	1.4142463
C	1.4923150	14.4575206	4.9311819
C	0.1069972	14.7266665	4.9281510
H	-0.5521321	14.2129646	4.2090530
C	-0.4381282	15.6418999	5.8429205
H	-1.5229105	15.8410574	5.8349218
C	0.3923723	16.2987725	6.7670986
H	-0.0394838	17.0145601	7.4865292
C	1.7712311	16.0321284	6.7783405
H	2.4312881	16.5327147	7.5063073
C	2.3169457	15.1102967	5.8708650

H	3.3962556	14.8861237	5.8994823
C	2.5230644	14.3166725	2.2217283
C	3.0766103	13.6968886	1.0766813
H	3.3345500	12.6239114	1.1042674
C	3.3162131	14.4422177	-0.0873831
H	3.7442619	13.9424583	-0.9723137
C	3.0255730	15.8177652	-0.1219969
H	3.2171442	16.4028835	-1.0371515
C	2.4900833	16.4425310	1.0160073
H	2.2597980	17.5211114	0.9992666
C	2.2350888	15.6976133	2.1807294
H	1.80666931	16.1976050	3.0641948
C	4.8273154	6.9996262	7.1601304
C	5.2455229	7.7720054	8.2652892
H	5.7320032	8.7473809	8.0947828
C	5.0179629	7.3223874	9.5758335
H	5.3456468	7.9389475	10.4293032
C	4.3537838	6.1048833	9.7976801
H	4.1642091	5.7571796	10.8270100
C	3.9179078	5.3387049	8.7034006
H	3.3857863	4.3872399	8.8709922
C	4.1526660	5.7819174	7.3916945
H	3.8012599	5.1771483	6.5396134
C	3.9334297	6.8064581	4.4017204
C	4.1352015	5.5315815	3.8318061
H	5.0856561	4.9977933	3.9969830
C	3.1326253	4.9421658	3.0437844
H	3.3015880	3.9467795	2.5996418
C	1.9233937	5.6201522	2.8172623
H	1.1404674	5.1591194	2.1920408
C	1.7204059	6.8930152	3.3763172
H	0.7840302	7.4432450	3.1853144
C	2.7211293	7.4873842	4.1600272
H	2.5703089	8.4980895	4.5751816
C	6.7665501	6.7716339	5.0116304
C	7.4341641	5.8695852	5.8672704
H	6.9954556	5.6118438	6.8450012
C	8.6562101	5.2928676	5.4774311
H	9.1665695	4.5884367	6.1558236
C	9.2204023	5.6036024	4.2300528
H	10.1767533	5.1468588	3.9250023
C	8.5605272	6.5023921	3.3724324
H	8.9978713	6.7555975	2.3921844
C	7.3480821	7.0901965	3.7611654
H	6.8451363	7.8119917	3.0932201

#Cu-4b singlet ground state
121

I	3.3153208	7.8722438	5.7173849
I	0.2167261	4.8003445	4.4882249
Cu	2.8587063	5.3940527	4.7205252
Cu	1.0903113	6.5007622	6.3710697
P	3.4427097	3.9699439	6.4094712

P	3.7034952	5.4374717	2.5895215
P	-0.6219001	7.9784984	6.7193585
F	9.3871003	7.2949878	2.1723291
F	3.5787297	0.2247778	-0.3864010
F	0.2224318	9.2069208	-0.4988465
F	-3.6120297	9.6165562	1.7910200
F	-4.7222757	5.5388193	10.3656669
F	0.9023796	13.1625758	9.3267490
N	1.6611126	5.3397319	8.0487597
C	2.6353802	4.3962126	8.0470132
C	3.0591646	3.7609447	9.2275081
H	3.8584069	3.0037371	9.1748296
C	2.4810655	4.0835079	10.4698952
C	1.4592544	5.0531100	10.4476517
H	0.9451037	5.3546868	11.3752733
C	1.0850127	5.6406291	9.2352030
H	0.2777855	6.3909335	9.2036143
C	2.9656545	3.4622682	11.7574643
H	2.1113734	3.3699571	12.4655021
H	3.3384008	2.4312461	11.5592250
C	4.0902413	4.2703286	12.4402325
H	3.7171105	5.3104495	12.6043946
C	4.5303589	3.6872715	13.8009476
H	3.7897216	3.8814571	14.6071421
H	4.6162730	2.5802305	13.7035448
C	5.9219217	4.3141997	14.0858309
H	5.8533399	5.1133225	14.8548921
H	6.6260899	3.5547478	14.4882080
C	6.4110721	4.8953817	12.7245690
H	7.4635130	4.6259787	12.4923350
H	6.3654498	6.0067540	12.7437999
C	5.4222058	4.3586333	11.6710974
H	5.7218284	3.3348785	11.3420050
H	5.3715077	4.9913303	10.7586179
C	5.2144877	3.8430576	6.9591214
C	5.9093170	5.0604138	7.1509747
H	5.3938556	6.0205004	6.9647601
C	7.2493632	5.0545130	7.5688583
H	7.7756477	6.0122457	7.7180299
C	7.9180922	3.8374582	7.7893683
H	8.9726863	3.8339851	8.1115817
C	7.2360386	2.6256465	7.5938170
H	7.7529503	1.6660742	7.7627595
C	5.8912290	2.6252061	7.1844726
H	5.3681763	1.6661792	7.0397223
C	2.9342759	2.2152093	6.1397897
C	1.8853261	1.5925547	6.8478847
H	1.3616431	2.1317382	7.6528145
C	1.4888832	0.2837865	6.5252962
H	0.6627791	-0.1862082	7.0843570
C	2.1341271	-0.4204481	5.4970196
H	1.8219432	-1.4481043	5.2477654
C	3.1726166	0.1967976	4.7782841
H	3.6774338	-0.3405469	3.9583870

C	3.5617508	1.5081046	5.0880944
H	4.3591020	1.9907106	4.4978091
C	5.4355001	6.0526983	2.4156500
C	6.3402207	5.7818936	3.4646682
H	5.9895744	5.2523616	4.3663545
C	7.6788530	6.1901320	3.3867205
H	8.3898481	5.9858981	4.2024920
C	8.1069027	6.8906467	2.2515901
C	7.2301947	7.1882715	1.1993739
H	7.6022159	7.7494300	0.3278153
C	5.8951656	6.7656305	1.2865578
H	5.2007136	7.0043168	0.4644010
C	3.7126296	3.8422468	1.6464040
C	4.6865751	3.5283090	0.6732468
H	5.4953939	4.2420826	0.4486029
C	4.6475445	2.3061856	-0.0143409
H	5.4041248	2.0466167	-0.7714446
C	3.6199503	1.3974145	0.2722374
C	2.6407813	1.6792530	1.2334366
H	1.8517864	0.9397196	1.4415258
C	2.6971990	2.8992525	1.9236707
H	1.9376164	3.1237777	2.6937229
C	2.7228319	6.5753315	1.5145103
C	1.8395993	6.1035772	0.5214604
H	1.7918981	5.0280519	0.2883436
C	0.9989499	6.9892022	-0.1714981
H	0.2942726	6.6324313	-0.9386509
C	1.0528159	8.3527102	0.1354791
C	1.9307193	8.8545164	1.1055458
H	1.9470175	9.9347776	1.3195495
C	2.7574486	7.9607331	1.7983713
H	3.4240703	8.3479653	2.5866643
C	-1.5898440	8.4610794	5.2216793
C	-0.8919292	8.6429150	4.0062672
H	0.1949669	8.4580680	3.9628554
C	-1.5696918	9.0424852	2.8458545
H	-1.0456299	9.1859166	1.8882471
C	-2.9554795	9.2417666	2.9037263
C	-3.6767694	9.0523740	4.0893778
H	-4.7669557	9.2084136	4.0916915
C	-2.9865683	8.6623391	5.2472001
H	-3.5508987	8.5065184	6.1811946
C	-1.9232313	7.3067957	7.8566328
C	-2.4212593	6.0099027	7.5822148
H	-2.0506786	5.4567151	6.7008107
C	-3.3723688	5.4134541	8.4200823
H	-3.7704676	4.4076511	8.2131352
C	-3.8158819	6.1121670	9.5526270
C	-3.3328994	7.3900717	9.8578899
H	-3.7041376	7.9088796	10.7557874
C	-2.3889270	7.9848106	9.0035765
H	-2.0133594	8.9939164	9.2382589
C	-0.1900099	9.6044416	7.4900221
C	-0.9430240	10.7807354	7.2825480

H	-1.8249212	10.7615851	6.6214489
C	-0.5792711	11.9859878	7.9010778
H	-1.1543452	12.9120238	7.7439052
C	0.5487672	12.0088603	8.7326174
C	1.3219115	10.8614397	8.9507353
H	2.2146558	10.9229191	9.5923622
C	0.9519719	9.6653205	8.3187900
H	1.5777733	8.7667456	8.4519066

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Cu	-1.0399834	-0.1743719	-0.0317738
Cu	1.4727399	-0.7439028	-0.0551604
I	0.3247855	0.2044831	-2.2936110
I	0.4755508	0.4580353	2.0828468
P	-1.4235968	-2.4959667	0.2302449
P	-2.7289980	1.3938506	-0.0967787
P	3.5890906	0.1091971	-0.1348915
F	-1.0076321	7.1159217	0.3487123
F	-5.9686773	1.2132790	-5.1308761
F	-6.6546527	0.5043374	4.3412177
F	3.1345044	5.9210884	-1.4602872
F	6.6563111	-0.5311175	4.9659196
F	7.0242096	-2.3536557	-4.3843761
N	1.3178074	-2.8493783	-0.0387032
C	0.0675363	-3.4993995	0.0277310
C	-0.0138908	-4.9146946	-0.0439104
H	-1.0090242	-5.3904170	-0.0143124
C	1.1277943	-5.6981668	-0.1589110
H	1.0565107	-6.7947186	-0.2276586
C	2.4023140	-5.0260154	-0.2047614
H	3.3485563	-5.5781401	-0.2997372
C	2.4103069	-3.6322429	-0.1497629
H	3.3748355	-3.0970786	-0.2089455
C	-2.1447002	-2.9217154	1.8875858
C	-1.4369359	-3.7265161	2.8036697
H	-0.4601582	-4.1469215	2.5150885
C	-1.9680977	-3.9929393	4.0768669
H	-1.4012068	-4.6242031	4.7815136
C	-3.2122991	-3.4618172	4.4515590
H	-3.6291068	-3.6744483	5.4501442
C	-3.9201640	-2.6491546	3.5496197
H	-4.8923703	-2.2161894	3.8389734
C	-3.3854358	-2.3719495	2.2816490
H	-3.9440501	-1.7157050	1.5929375
C	-2.6168585	-3.2482718	-0.9818612
C	-3.7860629	-3.9494569	-0.6217878
H	-4.0461864	-4.0818281	0.4404119
C	-4.6223722	-4.4967761	-1.6117603
H	-5.5292611	-5.0483532	-1.3117421
C	-4.3032753	-4.3534661	-2.9704644
H	-4.9583222	-4.7880257	-3.7438491
C	-3.1351612	-3.6602146	-3.3381992

H	-2.8672842	-3.5498057	-4.4024869
C	-2.2993340	-3.1127322	-2.3538929
H	-1.3806960	-2.5775668	-2.6507992
C	-2.2039697	3.1544780	0.0949968
C	-0.9763680	3.5687793	-0.4681139
H	-0.3251986	2.8342685	-0.9723064
C	-0.5715823	4.9090445	-0.3923841
H	0.3817452	5.2490362	-0.8260395
C	-1.3953989	5.8319043	0.2655047
C	-2.6115177	5.4476410	0.8455928
H	-3.2286057	6.2003752	1.3606100
C	-3.0118133	4.1063514	0.7558886
H	-3.9670075	3.8009760	1.2127327
C	-3.7207414	1.3891596	-1.6524321
C	-4.3139841	2.5562651	-2.1804217
H	-4.1766426	3.5258906	-1.6751667
C	-5.0745691	2.5009825	-3.3573295
H	-5.5375352	3.4031320	-3.7868002
C	-5.2410256	1.2677761	-4.0027802
C	-4.6611253	0.0943814	-3.5034221
H	-4.7986619	-0.8583534	-4.0379733
C	-3.8953388	0.1640907	-2.3316358
H	-3.4210395	-0.7552624	-1.9502644
C	-3.9888734	1.1808998	1.2373372
C	-5.3459001	0.9078184	0.9662252
H	-5.7045739	0.8640874	-0.0744691
C	-6.2538611	0.6816593	2.0135921
H	-7.3144075	0.4593043	1.8175725
C	-5.7944392	0.7404315	3.3346698
C	-4.4541582	1.0229216	3.6341834
H	-4.1263082	1.0513877	4.6847793
C	-3.5543417	1.2306203	2.5821490
H	-2.4922588	1.4197001	2.8132251
C	3.6037541	1.9083635	-0.5272761
C	3.8943730	2.3772408	-1.8258504
H	4.2247955	1.6746537	-2.6073869
C	3.7510269	3.7359717	-2.1451643
H	3.9611512	4.1142788	-3.1575521
C	3.3156226	4.6220448	-1.1527856
C	3.0395754	4.1889416	0.1509285
H	2.7033267	4.9152798	0.9069887
C	3.1748526	2.8287901	0.4560904
H	2.9230896	2.4800558	1.4707229
C	4.5438150	-0.0321371	1.4350808
C	4.2476751	-1.1077672	2.3013116
H	3.4355736	-1.8095572	2.0480697
C	4.9631430	-1.2861478	3.4931310
H	4.7394911	-2.1174933	4.1792643
C	5.9725565	-0.3713881	3.8214082
C	6.2796301	0.7138068	2.9891661
H	7.0717317	1.4179364	3.2881331
C	5.5614515	0.8788697	1.7959519
H	5.7902596	1.7399795	1.1469356
C	4.6799581	-0.6536120	-1.4181486

C	4.0813862	-1.3019649	-2.5208714
H	2.9818603	-1.3642724	-2.5894599
C	4.8683175	-1.8739889	-3.5311898
H	4.4131171	-2.3843777	-4.3939213
C	6.2628318	-1.8029694	-3.4253919
C	6.8856896	-1.1763912	-2.3361596
H	7.9852908	-1.1492953	-2.2838386
C	6.0892422	-0.6039351	-1.3354863
H	6.5791943	-0.1206352	-0.4749891

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N	-1.4300264	-2.5833798	0.0799744
C	-2.5874324	-3.2653536	0.0881261
C	-2.7105822	-4.6500736	0.2346080
C	-1.5109422	-5.4389558	0.4063520
C	-1.5947235	-6.9279771	0.5934090
C	-0.3069780	-4.7474317	0.3940077
C	-0.2474841	-3.3326396	0.2188673
P	1.3369012	-2.4723573	0.2279423
C	2.2140287	-3.1359124	1.7276260
C	3.1167846	-4.2215397	1.6937714
C	3.6756938	-4.7228044	2.8823067
C	3.3427912	-4.1509163	4.1202191
C	2.4386756	-3.0736458	4.1646894
C	1.8783224	-2.5719682	2.9805167
C	2.3321882	-3.1560773	-1.1821008
C	3.7231544	-2.9188055	-1.2546692
C	4.4698540	-3.3768049	-2.3509494
C	3.8337553	-4.0604816	-3.4019460
C	2.4480256	-4.2809141	-3.3479964
C	1.7009388	-3.8339899	-2.2448524
H	0.6368091	-5.3043397	0.5284423
H	3.3892656	-4.6834498	0.7311694
H	4.3767663	-5.5732377	2.8365486
H	2.1633189	-2.6202038	5.1316970
H	4.2328257	-2.3721094	-0.4427684
H	5.5565111	-3.1920132	-2.3871647
H	4.4200556	-4.4193843	-4.2643329
H	1.9399238	-4.8112144	-4.1707800
H	0.6143505	-4.0141023	-2.2009540
Cu	1.1546211	-0.1362231	0.0556699
Cu	-1.3974096	-0.4740225	-0.1033754
I	-0.1110092	0.3259367	-2.2786278
I	-0.4084078	0.7581053	2.0425884
P	2.9323957	1.3308344	0.0185125
P	-3.4655126	0.4853401	-0.3137827
C	4.2100336	1.0744395	-1.2944731
C	3.9060812	1.4487479	1.5798072
C	2.3344340	3.0509137	-0.2823205
C	-3.5131209	2.1840910	-1.0289756
C	-4.3913359	0.6225019	1.2753046
C	-4.5896097	-0.4850477	-1.4120557

C	5.5105655	1.6170377	-1.1897681
C	2.4053687	3.6378018	-1.5634596
C	4.5342696	2.6384379	2.0114551
C	5.2923971	2.6597518	3.1912867
C	-2.4841926	3.0965339	-0.7029729
C	-5.2702029	1.6919032	1.5495755
C	-5.9332455	-0.7597662	-1.0792656
C	-5.9641850	1.7527648	2.7674035
C	-2.5020633	4.4017686	-1.2134875
C	-6.7486023	-1.4988445	-1.9501479
C	1.8060847	4.8807771	-1.8178634
C	6.4474876	1.4357380	-2.2155645
C	3.8703704	0.3330213	-2.4463609
C	4.7973523	0.1511693	-3.4837788
C	6.0756814	0.7063196	-3.3545738
H	2.8661793	-0.1140213	-2.5368290
H	4.5421005	-0.4301722	-4.3830245
H	7.4657227	1.8493207	-2.1461573
H	5.8056361	2.1848269	-0.2928236
C	-4.0686051	-0.9760911	-2.6310157
C	-4.8762115	-1.7078264	-3.5106897
C	-6.2093167	-1.9606816	-3.1576968
H	-3.0120998	-0.7919446	-2.8917829
H	-4.4832655	-2.0984429	-4.4620805
H	-7.7980407	-1.7243839	-1.7044700
H	1.1621360	-1.7333424	3.0262130
H	-3.5007548	-2.6581078	-0.0399941
H	-3.7066258	-5.1172650	0.2236663
H	-1.6525025	2.7812388	-0.0501988
C	-3.5496532	4.7863110	-2.0614496
H	-1.7050213	5.1253515	-0.9827702
C	-4.5763674	3.8996147	-2.4104882
C	-4.5519426	2.5967065	-1.8916484
H	-5.3771798	4.2358385	-3.0873642
H	-5.3525624	1.8930454	-2.1712334
H	-6.3532821	-0.3967830	-0.1273827
C	-4.2078869	-0.3841910	2.2498952
C	-4.9020616	-0.3375862	3.4660051
C	-5.7726442	0.7337441	3.7095405
H	-3.5034069	-1.2120441	2.0646823
H	-4.7660547	-1.1128922	4.2356648
H	-5.4120284	2.4959225	0.8092990
H	2.9240834	3.1160967	-2.3833708
C	1.1351215	5.5335609	-0.7777482
H	1.8429317	5.3436782	-2.8160890
C	1.0610344	4.9847159	0.5090970
C	1.6554150	3.7388655	0.7492910
H	0.5286998	5.5297297	1.3042640
H	1.5756754	3.2942413	1.7540765
C	4.0437967	0.2816040	2.3617029
C	4.8026738	0.2884891	3.5399118
C	5.4197626	1.4809638	3.9393923
H	3.5388778	-0.6503078	2.0574863
H	4.9064526	-0.6174772	4.1570317

H	5.7842130	3.5797764	3.5439605
H	4.4236710	3.5674707	1.4286894
H	3.7819185	-4.5469080	5.0510174
F	6.9731624	0.5310223	-4.3390401
F	6.1436834	1.5003734	5.0706469
H	-6.6482082	2.5837589	3.0000290
F	-6.4336615	0.7892296	4.8773513
F	-3.5660038	6.0341394	-2.5588737
F	-6.9860287	-2.6683676	-3.9938819
F	0.5226631	6.7094817	-1.0213737
H	-0.5894421	-7.3883560	0.7049688
H	-2.1038803	-7.4190210	-0.2702844
H	-2.1939780	-7.1915442	1.4976470

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I	0.1524143	0.2485098	2.1169989
I	0.2409663	-0.1123686	-2.2470870
Cu	-1.1930323	-0.6128960	-0.0417684
Cu	1.3556367	-0.9047793	0.0555911
P	-1.3499931	-2.9477886	0.2426990
P	-3.0075081	0.7976378	-0.2050330
P	3.3731639	0.1718258	0.0282100
F	-7.3063034	-1.6169241	4.3584833
F	-6.9117112	0.2796412	5.3609101
F	-8.4835647	0.1469476	3.8480625
F	-5.6723700	-0.6657376	-6.2913183
F	-6.2025668	1.4530915	-6.1991785
F	-7.4529831	-0.0306334	-5.2052313
F	7.7551331	-3.6950807	-3.3806433
F	8.6235147	-1.7571178	-3.8719020
F	6.9749799	-2.5530545	-5.0668713
F	-1.9733803	7.4285604	1.1474460
F	0.0468806	6.7327000	0.6999076
F	-1.2770144	7.3431253	-0.9209133
F	3.6210902	6.3753715	-2.8315431
F	2.9742539	6.9125197	-0.8156206
F	1.5154578	6.2176136	-2.2790680
F	5.5550947	-0.4173949	6.4444021
F	7.4498911	-0.4215656	5.3662863
F	6.5322842	1.4716727	5.9387640
N	1.4237377	-2.9985909	0.1656682
C	0.2494718	-3.7774017	0.1998062
C	0.3282270	-5.2009300	0.1977929
H	-0.6111824	-5.7812527	0.1947659
C	1.5432918	-5.8738818	0.1909028
C	2.7354838	-5.0516511	0.1829255
H	3.7396611	-5.5010239	0.1723172
C	2.5935407	-3.6633311	0.1593261
C	1.6481659	-7.3728383	0.1767758
H	2.2222058	-7.7499503	1.0568705
H	2.1937525	-7.7348168	-0.7274178
H	0.6482610	-7.8575542	0.1876296

C	-2.3627009	-3.8410195	-1.0369389
C	-2.0090297	-3.6323686	-2.3906865
H	-1.1760475	-2.9517295	-2.6382456
C	-2.6976296	-4.2916322	-3.4196618
H	-2.4039974	-4.1197014	-4.4688544
C	-3.7506171	-5.1740198	-3.1151958
H	-4.2901234	-5.6950284	-3.9234826
C	-4.1013518	-5.3943413	-1.7746374
H	-4.9165986	-6.0936592	-1.5238931
C	-3.4132625	-4.7343774	-0.7404578
H	-3.6966090	-4.9273505	0.3064029
C	-2.1657597	-3.3989676	1.8477957
C	-3.5101976	-3.0383217	2.0908882
H	-4.0950228	-2.5263167	1.3076150
C	-4.1170242	-3.3260305	3.3237593
H	-5.1664731	-3.0374430	3.4995398
C	-3.3813217	-3.9621802	4.3386635
H	-3.8561533	-4.1843001	5.3088123
C	-2.0390593	-4.3060937	4.1129275
H	-1.4532609	-4.7992152	4.9066184
C	-1.4336869	-4.0271269	2.8760422
H	-0.3802147	-4.2996892	2.7017692
C	-4.3178912	0.5425724	1.0755434
C	-5.6855730	0.4275365	0.7507139
H	-6.0184413	0.5195608	-0.2949626
C	-6.6355196	0.1891325	1.7551888
H	-7.7010287	0.0962373	1.4935188
C	-6.2296744	0.0739401	3.0957877
C	-4.8692003	0.2018545	3.4298962
H	-4.5469018	0.1086668	4.4784754
C	-3.9188847	0.4211500	2.4251291
H	-2.8504775	0.4812424	2.6937814
C	-7.2379891	-0.2693855	4.1691105
C	-3.9280736	0.7069386	-1.8027553
C	-4.5905392	1.8217252	-2.3591222
H	-4.5492743	2.8010475	-1.8555445
C	-5.2978804	1.6944716	-3.5629137
H	-5.8039038	2.5695395	-3.9997815
C	-5.3544523	0.4513634	-4.2190818
C	-4.6903561	-0.6612410	-3.6735454
H	-4.7132969	-1.6321893	-4.1924653
C	-3.9763293	-0.5318235	-2.4744519
H	-3.4415833	-1.4069182	-2.0690939
C	-6.1680131	0.3025578	-5.4870092
C	-2.5714776	2.5896436	-0.0505503
C	-1.6215320	3.1162978	-0.9551867
H	-1.1765433	2.4709751	-1.7310215
C	-1.2189534	4.4529351	-0.8637874
H	-0.4766247	4.8521749	-1.5724828
C	-1.7445992	5.2795724	0.1483963
C	-2.6911332	4.7656377	1.0485839
H	-3.1098388	5.4130525	1.8340566
C	-3.1068304	3.4280377	0.9470540
H	-3.8528260	3.0427218	1.6595061

C	-1.2467672	6.7024894	0.2686906
C	4.6279126	-0.5993906	-1.0943204
C	4.1733877	-1.2624616	-2.2539671
H	3.0916664	-1.3332733	-2.4596081
C	5.0886511	-1.8389367	-3.1464397
H	4.7259757	-2.3492724	-4.0518544
C	6.4670134	-1.7705660	-2.8816709
C	6.9274592	-1.1189567	-1.7229902
H	8.0078116	-1.0619978	-1.5168776
C	6.0134915	-0.5373054	-0.8339435
H	6.3880928	-0.0334600	0.0713713
C	7.4577432	-2.4439481	-3.8080840
C	3.2919404	1.9308090	-0.5269850
C	3.9166033	2.3791818	-1.7074607
H	4.5228319	1.6888334	-2.3147045
C	3.7676443	3.7110345	-2.1257854
H	4.2561738	4.0547837	-3.0501881
C	2.9984143	4.6063399	-1.3656142
C	2.3794931	4.1677116	-0.1785716
H	1.7723438	4.86668544	0.4175700
C	2.5171888	2.8373152	0.2322068
H	2.0008409	2.4964178	1.1452102
C	2.7880037	6.0322484	-1.8239265
C	4.2264723	0.2518183	1.6644086
C	4.0290942	-0.8122226	2.5701899
H	3.3488285	-1.6404316	2.3108462
C	4.6781178	-0.8142500	3.8120210
H	4.5120843	-1.6430220	4.5174901
C	5.5230903	0.2527488	4.1664500
C	5.7107626	1.3248374	3.2767268
H	6.3571359	2.1691609	3.5626501
C	5.0668130	1.3241839	2.0307772
H	5.2133658	2.1747704	1.3455759
C	6.2631385	0.2245516	5.4870527
H	3.5005236	-3.0336544	0.1245832

#Cu-2d first triplet extcied state

108

I	8.0312492	2.4471819	4.7914637
I	10.4401338	-0.7310891	6.5457738
Cu	8.1019184	-0.1685614	5.3797020
Cu	10.2194138	0.9648607	4.4927776
Cl	0.5589169	-1.1343214	5.0521993
Cl	7.6038168	-5.4712371	10.9553131
Cl	6.8279463	4.7970242	10.7100085
Cl	11.3951588	7.7787025	1.2937318
Cl	11.5677727	4.2052914	11.0733849
Cl	17.7781336	0.3182758	3.9372654
P	8.0304969	-1.3815203	3.3648817
P	6.5919544	-0.4706908	7.0902337
P	11.8428722	2.5208720	4.9118494
N	10.4039988	-0.1209225	2.6889729
C	9.4710228	-1.1058992	2.3161116

C	9.6115703	-1.7967489	1.0751289
H	8.8481755	-2.5437559	0.7953923
C	10.6735160	-1.5565064	0.2138322
C	11.6306926	-0.5558629	0.6300039
H	12.5034645	-0.3047897	0.0090230
C	11.4248570	0.1038003	1.8452656
H	12.1493203	0.8736023	2.1661658
C	10.8426429	-2.2784384	-1.0936754
H	10.0288205	-3.0146848	-1.2684397
H	11.8152675	-2.8248206	-1.1341216
H	10.8505231	-1.5661866	-1.9530301
C	6.5709808	-1.1479001	2.2367664
C	6.1172566	0.1735141	2.0187245
H	6.5931977	1.0094225	2.5601505
C	5.0810875	0.4346048	1.1097094
H	4.7422685	1.4726756	0.9538651
C	4.4809518	-0.6195258	0.3970646
H	3.6669226	-0.4149516	-0.3180546
C	4.9331249	-1.9333783	0.5969146
H	4.4767543	-2.7665263	0.0361886
C	5.9700012	-2.1980135	1.5079518
H	6.3103896	-3.2363403	1.6509516
C	7.9793449	-3.1929829	3.7592093
C	6.7854542	-3.7878813	4.2287983
H	5.8585790	-3.1917887	4.2870640
C	6.7658576	-5.1352482	4.6218282
H	5.8241906	-5.5871331	4.9760042
C	7.9423055	-5.9037921	4.5692712
H	7.9267532	-6.9625576	4.8774734
C	9.1361768	-5.3148630	4.1215010
H	10.0644754	-5.9089121	4.0809229
C	9.1559288	-3.9690733	3.7171501
H	10.0928589	-3.5099552	3.3614283
C	4.8381230	-0.6825688	6.5527201
C	4.4629580	-0.1760829	5.2902275
H	5.2125420	0.3200361	4.6514724
C	3.1475505	-0.3064504	4.8219832
H	2.8669240	0.0797874	3.8302679
C	2.1953350	-0.9562656	5.6248916
C	2.5489644	-1.4746868	6.8838442
H	1.7917658	-1.9880629	7.4961147
C	3.8670550	-1.3368684	7.3409230
H	4.1386495	-1.7567845	8.3231063
C	6.8914973	-1.8807475	8.2414282
C	7.4800801	-3.0576342	7.7329320
H	7.7821835	-3.1138577	6.6746737
C	7.6950788	-4.1682332	8.5606591
H	8.1547614	-5.0834175	8.1579669
C	7.3305167	-4.0985617	9.9158855
C	6.7553052	-2.9308848	10.4467199
H	6.4843467	-2.8892437	11.5128048
C	6.5382339	-1.8282852	9.6075508
H	6.0972754	-0.9115393	10.0321397
C	6.5688343	0.9998770	8.2038826

C	7.7458821	1.3079595	8.9231256
H	8.6285289	0.6497055	8.8529461
C	7.8203536	2.4603681	9.7147393
H	8.7362455	2.6983674	10.2770545
C	6.7171497	3.3305096	9.7686296
C	5.5387222	3.0434194	9.0616242
H	4.6834372	3.7341012	9.1162477
C	5.4683047	1.8761347	8.2846258
H	4.5423961	1.6584466	7.7282639
C	11.7101865	4.0678395	3.9177326
C	11.1296253	3.9904368	2.6333743
H	10.7307966	3.0303577	2.2650836
C	11.0358551	5.1269882	1.8186803
H	10.5803567	5.0650138	0.8188480
C	11.5159882	6.3587740	2.2960389
C	12.0843433	6.4610823	3.5776297
H	12.4454396	7.4356060	3.9403651
C	12.1791531	5.3160077	4.3818274
H	12.6142018	5.4054144	5.3907787
C	11.8629815	3.0968117	6.6624366
C	12.8413764	2.6551780	7.5768411
H	13.6745701	2.0199503	7.2361947
C	12.7639872	3.0052095	8.9332371
H	13.5227523	2.6523092	9.6481299
C	11.6987100	3.8040970	9.3790127
C	10.7212129	4.2681677	8.4818258
H	9.8891347	4.8898745	8.8467323
C	10.8036353	3.9060425	7.1312528
H	10.0152647	4.2453614	6.4397179
C	13.5657290	1.9216479	4.6060870
C	14.6304452	2.8009933	4.3127494
H	14.4515546	3.8855219	4.2340459
C	15.9276043	2.3123086	4.1067612
H	16.7571007	2.9978562	3.8752592
C	16.1668721	0.9290403	4.1939728
C	15.1201828	0.0360938	4.4772674
H	15.3185477	-1.0450141	4.5314586
C	13.8247838	0.5358020	4.6760468
H	13.0004920	-0.1674983	4.8855244

#Cu-4a first triplet extcied state

121

I	3.2020313	10.9096827	6.6734429
I	4.8202180	10.2715853	2.6809603
Cu	4.1241883	12.1403571	4.4790649
Cu	5.2735192	9.9685554	5.2778284
P	6.1268408	13.2567900	5.0290148
P	2.3167004	13.2223781	3.5327285
P	5.1185572	7.7052515	5.6001241
N	7.1277761	10.8058279	5.8603553
C	7.4041304	12.1757875	5.7025765
C	8.6493156	12.7143208	6.1442118
H	8.8270681	13.7983051	6.0328826

C	9.6357840	11.9211291	6.7151123
C	9.3385106	10.5119369	6.8408974
H	10.0705218	9.8056351	7.2607507
C	8.0934607	10.0474361	6.4050092
H	7.8620794	8.9716070	6.4959028
C	10.9388804	12.4859290	7.2229143
H	11.0834342	13.5170271	6.8237193
H	11.7859599	11.8703363	6.8328344
C	11.0494265	12.5351428	8.7599436
H	10.8889355	11.5021026	9.1545379
C	10.0514913	13.4699768	9.4674349
H	10.0695541	14.4664953	8.9643106
H	9.0026614	13.1060180	9.4243097
C	10.6047677	13.5795461	10.9021419
H	10.2913441	14.5174826	11.4096126
H	10.2115815	12.7402634	11.5180938
C	12.1534497	13.4581614	10.7590188
H	12.5555573	12.6958432	11.4612956
H	12.6688222	14.4112569	11.0074400
C	12.4072912	13.0500752	9.2829983
H	13.2226535	12.3020094	9.1710111
H	12.7044661	13.9408131	8.6814911
C	6.0249732	14.6454919	6.2625579
C	6.7464622	15.8544656	6.1472929
H	7.3521706	16.0521801	5.2482725
C	6.7010050	16.8143563	7.1733901
H	7.2736959	17.7511894	7.0664731
C	5.9372599	16.5844346	8.3283749
H	5.9044443	17.3386694	9.1322144
C	5.2199005	15.3803332	8.4551524
H	4.6211718	15.1839832	9.3604965
C	5.2647522	14.4196761	7.4339772
H	4.7085172	13.4732211	7.5488146
C	6.8558707	14.0715950	3.5295764
C	6.2450155	15.2148659	2.9679187
H	5.3579317	15.6607075	3.4492462
C	6.7591142	15.7974916	1.7995056
H	6.2738088	16.6939399	1.3789427
C	7.8784185	15.2338131	1.1628508
H	8.2814322	15.6906561	0.2434029
C	8.4770467	14.0836349	1.7016248
H	9.3515510	13.6305876	1.2051183
C	7.9710718	13.5060341	2.8782405
H	8.4461302	12.6071834	3.3040105
C	0.9944890	12.0745466	2.9294024
C	0.3925530	11.1996506	3.8617295
H	0.7143261	11.2117083	4.9168342
C	-0.6111506	10.3096627	3.4515352
H	-1.0797501	9.6410461	4.1931665
C	-1.0144555	10.2674555	2.1052570
H	-1.8004060	9.5641694	1.7831284
C	-0.4106150	11.1251526	1.1729413
H	-0.7185435	11.0981859	0.1144674
C	0.5863157	12.0278475	1.5807208

H	1.0468148	12.7001028	0.8396681
C	1.4111980	14.3581702	4.6732965
C	0.0175464	14.5667751	4.5902531
H	-0.5790491	14.0208991	3.8413234
C	-0.6143756	15.4646390	5.4659813
H	-1.7042732	15.6178522	5.3965119
C	0.1358044	16.1615363	6.4283067
H	-0.3648924	16.8622456	7.1170589
C	1.5224872	15.9558144	6.5165294
H	2.1212168	16.4871607	7.2746489
C	2.1566801	15.0551346	5.6468487
H	3.2423757	14.8850494	5.7378163
C	2.6817791	14.2740391	2.0540267
C	3.6041495	13.7945498	1.0984109
H	4.1210675	12.8341029	1.2659093
C	3.8763660	14.5390691	-0.0596390
H	4.5994850	14.1515526	-0.7957659
C	3.2488339	15.7785947	-0.2666318
H	3.4697593	16.3669665	-1.1728417
C	2.3458874	16.2710391	0.6902634
H	1.8551378	17.2469406	0.5387228
C	2.0588718	15.5230169	1.8439373
H	1.3446012	15.9164649	2.5848536
C	4.7704283	7.2117570	7.3456546
C	5.3265006	7.9899911	8.3843432
H	5.9234666	8.8856034	8.1439665
C	5.1085418	7.6378928	9.7249619
H	5.5460166	8.2541253	10.5274652
C	4.3229533	6.5167769	10.0418966
H	4.1442393	6.2474038	11.0961432
C	3.7567674	5.7468340	9.0130201
H	3.1314459	4.8715954	9.2558133
C	3.9786005	6.0901975	7.6689964
H	3.5254090	5.4849487	6.8670273
C	3.8230063	6.8439038	4.6074614
C	4.0232137	5.5514020	4.0777259
H	4.9818883	5.0311668	4.2360324
C	3.0050873	4.9269368	3.3385252
H	3.1716827	3.9180305	2.9257252
C	1.7849270	5.5871324	3.1199011
H	0.9901584	5.0979261	2.5324091
C	1.5848440	6.8768490	3.6400057
H	0.6384511	7.4116744	3.4557539
C	2.5984986	7.5069697	4.3773547
H	2.4440762	8.5272517	4.7672778
C	6.6739177	6.7998860	5.1749943
C	7.2135947	5.7873044	5.9955279
H	6.7101734	5.5122623	6.9364694
C	8.3962265	5.1290917	5.6179287
H	8.8119506	4.3406581	6.2671869
C	9.0464566	5.4740638	4.4222336
H	9.9764458	4.9585455	4.1302756
C	8.5149300	6.4855833	3.6040495
H	9.0263485	6.7692465	2.6695598

C	7.3390575	7.1517646	3.9795086
H	6.9372178	7.9603290	3.3446496

#Cu-4b first triplet extctied state
121

I	3.0639271	8.0548637	5.3415549
I	0.1685860	4.9357863	4.4599416
Cu	2.7999923	5.4705405	4.6715604
Cu	1.1532988	6.5048595	6.3604909
P	3.5871067	4.1251708	6.4272119
P	3.5500402	5.3449650	2.4956678
P	-0.5295941	8.0126510	6.7445514
F	9.3079029	6.8004529	1.7126072
F	2.9643988	0.0289254	-0.2113883
F	0.0553494	9.1111450	-0.5716517
F	-3.5231581	9.8862374	1.9107689
F	-4.7526278	5.7337387	10.3294466
F	1.3875577	12.9786751	9.4842002
N	1.7322278	5.3845546	8.0599091
C	2.7085738	4.3755534	7.9821445
C	3.0797356	3.6436761	9.1505476
H	3.8630386	2.8706927	9.0657579
C	2.4903958	3.8768524	10.3856140
C	1.4776813	4.9057337	10.4347350
H	0.9442823	5.1422970	11.3676752
C	1.1660428	5.5965349	9.2591921
H	0.3855803	6.3771389	9.2886475
C	2.8984668	3.1334599	11.6332679
H	1.9840923	2.7646159	12.1584864
H	3.4939242	2.2323045	11.3567679
C	3.7179690	3.9778995	12.6292715
H	3.1240923	4.8890578	12.8859134
C	4.0748109	3.2464405	13.9400237
H	3.2019896	3.1274996	14.6190350
H	4.4287879	2.2191769	13.6885246
C	5.2265910	4.0736581	14.5681012
H	4.8479814	4.7257099	15.3851139
H	5.9947575	3.4145990	15.0274709
C	5.8113395	4.9334357	13.4041310
H	6.9175501	4.8577506	13.3277935
H	5.5799098	6.0090647	13.5695750
C	5.0991644	4.4385053	12.1283363
H	5.6369725	3.5605864	11.6966712
H	5.0422615	5.2068270	11.3278368
C	5.3865665	4.2660953	6.8705964
C	5.9724945	5.5535400	6.8516947
H	5.3802481	6.4165531	6.5020831
C	7.2930830	5.7467199	7.2843194
H	7.7307152	6.7588573	7.2602122
C	8.0537442	4.6580360	7.7489100
H	9.0910020	4.8098229	8.0908898
C	7.4793111	3.3769249	7.7790236
H	8.0647789	2.5169917	8.1460188

C	6.1580080	3.1795605	7.3442104
H	5.7270133	2.1654824	7.3697016
C	3.3989749	2.3293988	5.9993546
C	2.2959543	1.5945240	6.4809342
H	1.5872899	2.0764386	7.1744274
C	2.1039034	0.2590371	6.0881096
H	1.2383173	-0.3034350	6.4763965
C	3.0116828	-0.3602593	5.2134354
H	2.8654334	-1.4116144	4.9141246
C	4.1084884	0.3683022	4.7208021
H	4.8233479	-0.1069142	4.0284360
C	4.2954044	1.7054727	5.1029077
H	5.1549805	2.2688379	4.7014046
C	5.3031724	5.8447700	2.2122483
C	6.2429602	5.5547949	3.2249739
H	5.9130806	5.0873048	4.1675682
C	7.5989790	5.8691385	3.0625811
H	8.3340586	5.6529054	3.8533263
C	8.0117455	6.4880221	1.8755179
C	7.1018240	6.7991039	0.8555753
H	7.4633343	7.2949345	-0.0588437
C	5.7481437	6.4752483	1.0288410
H	5.0303613	6.7317194	0.2326480
C	3.4076234	3.7046611	1.6547967
C	4.0807916	3.4284480	0.4431868
H	4.7369365	4.1866102	-0.0139978
C	3.9337137	2.1884335	-0.1912559
H	4.4537834	1.9571071	-1.1339507
C	3.1054336	1.2195450	0.3952518
C	2.4315679	1.4617322	1.5974764
H	1.8017719	0.6748445	2.0398900
C	2.5910332	2.7061378	2.2260588
H	2.0710920	2.8992760	3.1790055
C	2.5668048	6.4772125	1.4208628
C	1.5061410	5.9916402	0.6267984
H	1.3239322	4.9079766	0.5498072
C	0.6605700	6.8757682	-0.0601610
H	-0.1782665	6.5105797	-0.6725947
C	0.8854822	8.2522647	0.0533497
C	1.9386824	8.7651794	0.8216374
H	2.0874218	9.8544928	0.8843009
C	2.7711831	7.8731651	1.5105722
H	3.5835838	8.2764579	2.1358681
C	-1.4774563	8.5764523	5.2671530
C	-0.7938918	8.7751123	4.0461408
H	0.2875342	8.5675310	3.9783400
C	-1.4813627	9.2240609	2.9102395
H	-0.9717885	9.3743677	1.9459388
C	-2.8595463	9.4629365	2.9993836
C	-3.5646186	9.2637356	4.1932910
H	-4.6489964	9.4530236	4.2213361
C	-2.8673507	8.8179323	5.3257732
H	-3.4201453	8.6508991	6.2643115
C	-1.8301273	7.3540278	7.8783561

C	-2.2916957	6.0345695	7.6675799
H	-1.8607162	5.4241440	6.8553362
C	-3.2847137	5.4849084	8.4881411
H	-3.6519130	4.4579794	8.3374686
C	-3.8060456	6.2586092	9.5347143
C	-3.3579634	7.5637739	9.7757858
H	-3.7862914	8.1374730	10.6124126
C	-2.3682784	8.1080964	8.9429506
H	-2.0116570	9.1336331	9.1309991
C	0.0360744	9.5677633	7.5579733
C	-0.5550588	10.8207926	7.2905537
H	-1.3755316	10.9005848	6.5591403
C	-0.1013151	11.9785798	7.9401951
H	-0.5467299	12.9650999	7.7378877
C	0.9489665	11.8726842	8.8613005
C	1.5582823	10.6418699	9.1423955
H	2.3907882	10.5995892	9.8615037
C	1.1027319	9.4934735	8.4817247
H	1.5958240	8.5273036	8.6804172