

Electron Supplementary Information (ESI)

## The curious case of opossum prion: a physicochemical study on the copper (II) binding features of proteins N-terminal domain

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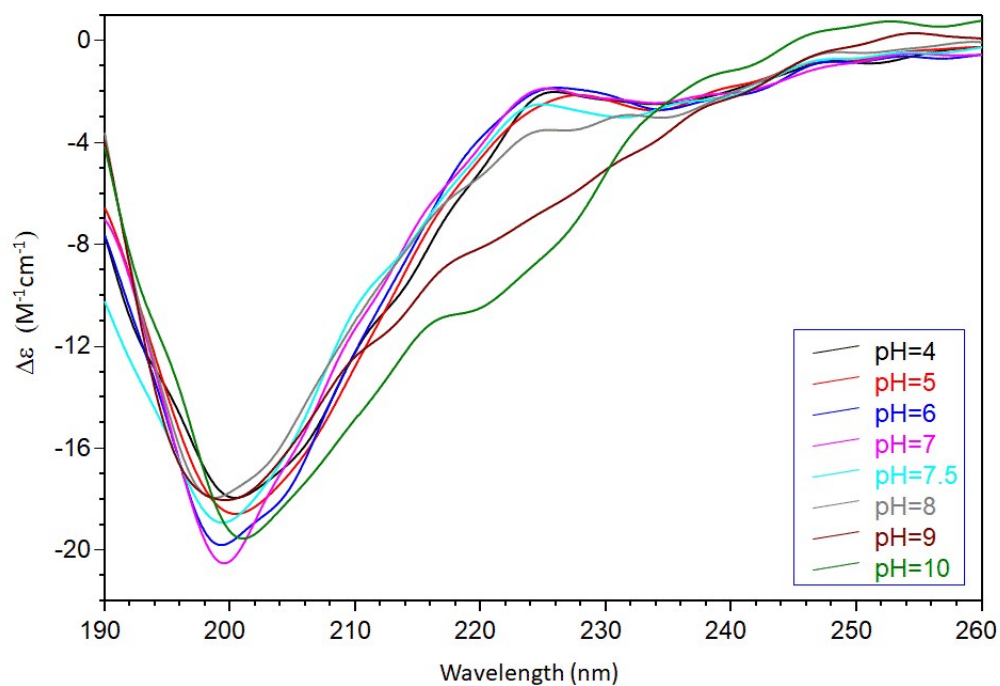
† These authors contribute equally.

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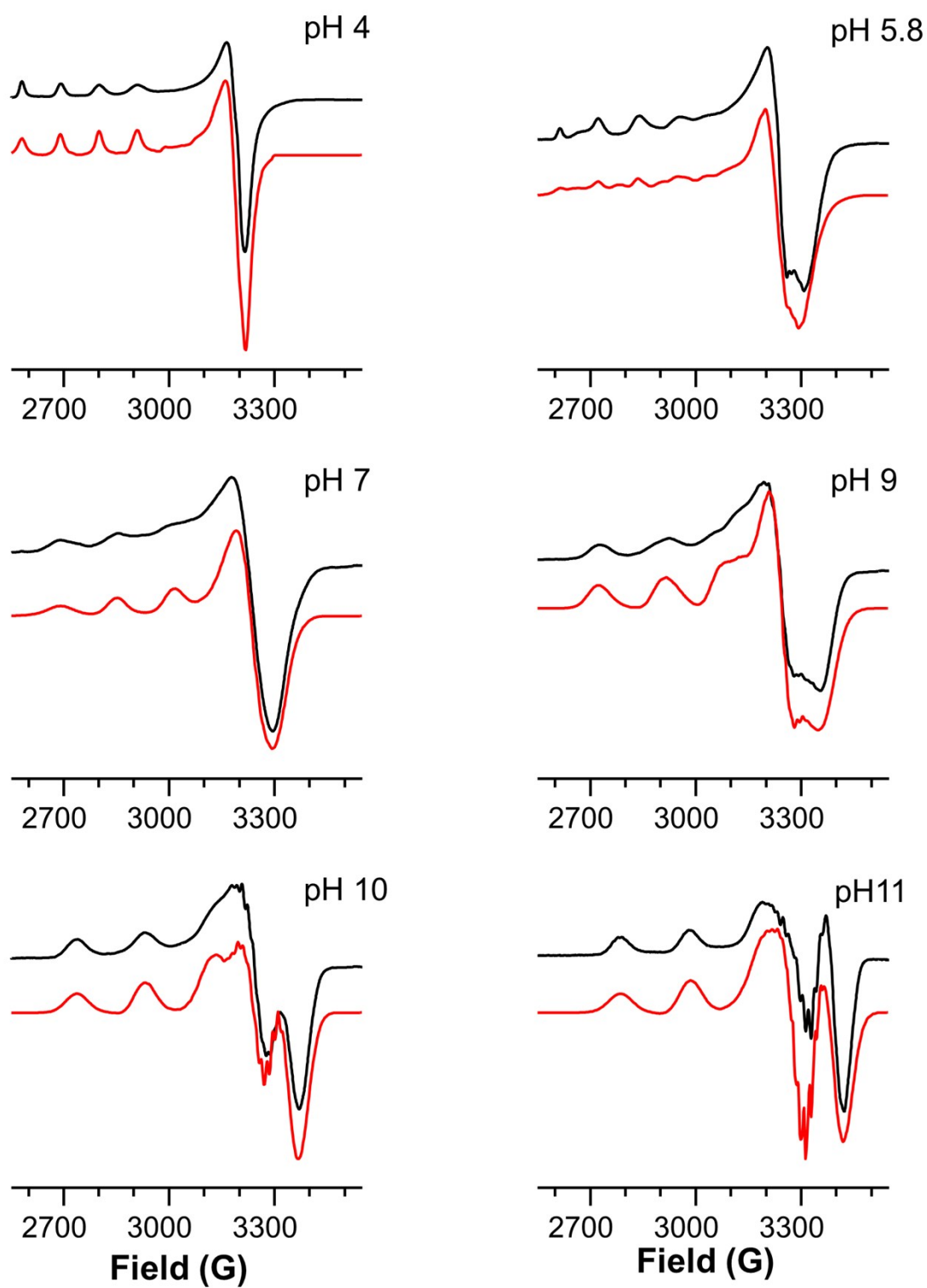
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**Table S1.** Proton Chemical shift of the Op\_bis-deca peptide in H<sub>2</sub>O, at pH 5.0.

Amino acid	NH	$\alpha$ CH	$\beta$ CH / $\gamma$ CH	Others
Pro <sup>1</sup>	---	4.26	2.14; 1.85; 1.71	2.01 Ac 3.51 $\delta\delta'$
His <sup>2</sup>	8.38	4.80	2.99 $\beta$ ; 2.91 $\beta'$	8.29 H $\delta$ 2; 7.08 H $\epsilon$ 1
Pro <sup>3</sup>	---	4.26	2.15; 1.86	3.60 $\delta$ ; 3.42 $\delta'$
Gly <sup>4</sup>	8.44	3.90 $\alpha\alpha'$		
Gly <sup>5</sup>	8.18	3.94 $\alpha\alpha'$		
Ser <sup>6</sup>	8.24	4.34	3.71 $\beta\beta'$	
Asn <sup>7</sup>	8.36	4.62	2.67; 2.60	7.41 H $\delta$ 21; 6.77 H $\delta$ 22
Trp <sup>8</sup>	7.97	4.53	3.18 $\beta\beta'$	7.14 H $\delta$ 1 10.05 H $\epsilon$ 1 7.39 H $\zeta$ 2 7.06 H $\eta$ 2 7.37 H $\zeta$ 3 7.48 H $\epsilon$ 3
Gly <sup>9</sup>	8.09	3.70 $\alpha\alpha'$		
Gln <sup>10</sup>	7.87	4.48	2.25; 1.96; 1.81	7.42 H $\epsilon$ 21; 6.79 H $\epsilon$ 22
Pro <sup>11</sup>	---	4.26	2.14; 1.85; 1.71	3.51 $\delta\delta'$
His <sup>12</sup>	8.38	4.80	2.99 $\beta$ ; 2.91 $\beta'$	8.31H $\delta$ 2; 7.08 H $\epsilon$ 1
Pro <sup>13</sup>	---	4.26	2.15; 1.87	3.60 $\delta$ ; 3.42 $\delta'$
Gly <sup>14</sup>	8.46	3.90 $\alpha\alpha'$		
Gly <sup>15</sup>	8.18	3.94 $\alpha\alpha'$		
Ser <sup>16</sup>	8.24	4.34	3.71 $\beta\beta'$	
Asn <sup>17</sup>	8.36	4.62	2.67; 2.60	7.41 H $\delta$ 21; 6.77 H $\delta$ 22
Trp <sup>18</sup>	7.99	4.53	3.18 $\beta\beta'$	7.14 H $\delta$ 1 10.05 H $\epsilon$ 1 7.39 H $\zeta$ 2 7.06 H $\eta$ 2 7.37 H $\zeta$ 3 7.48 H $\epsilon$ 3
Gly <sup>19</sup>	8.13	3.67-3.74 $\alpha\alpha'$		
Gln <sup>20</sup>	7.93	4.19	2.23-2.04-1.84	7.52; 7.03 NH <sub>2</sub> t 7.42 H $\epsilon$ 22; 6.79 H $\epsilon$ 21



**Figure S1.** Far UV CD spectra of Cu(II)-Op\_bis-deca in H<sub>2</sub>O 2:1 metal to ligand molar ratio, [L]= 1 x 10<sup>-5</sup> M at different pH values.



**Figure S2.** Experimental (black traces) and simulated (red traces) EPR spectra of Cu(II)-Op\_bis-deca peptide at Cu(II): L 1:1 molar ratio at different pH values. T= 150 K, [L]=[Cu(II)] =  $1 \times 10^{-3}$  M.

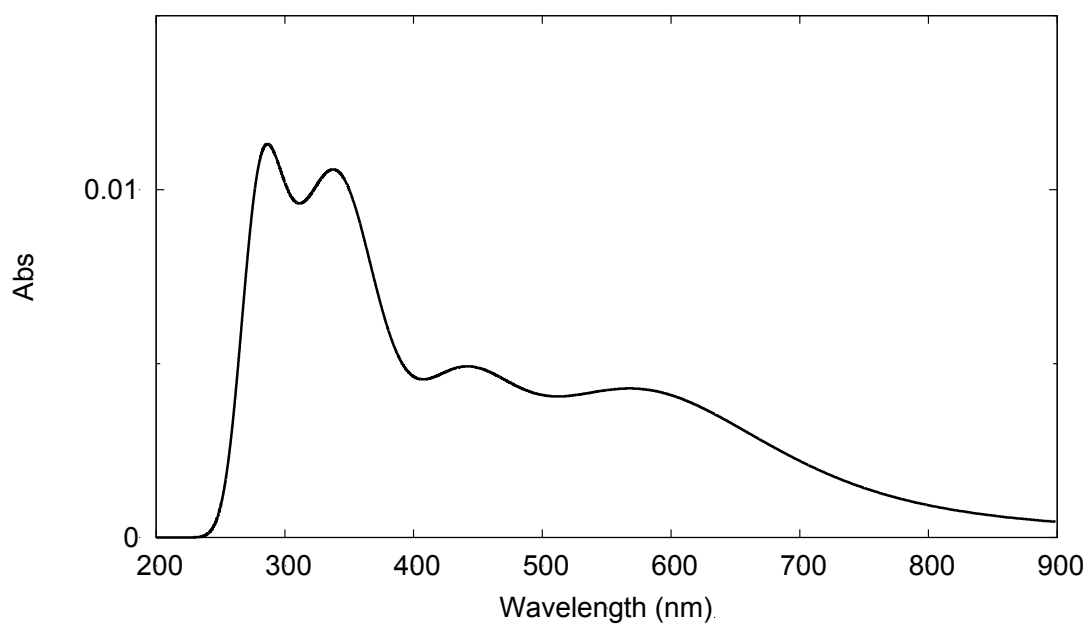
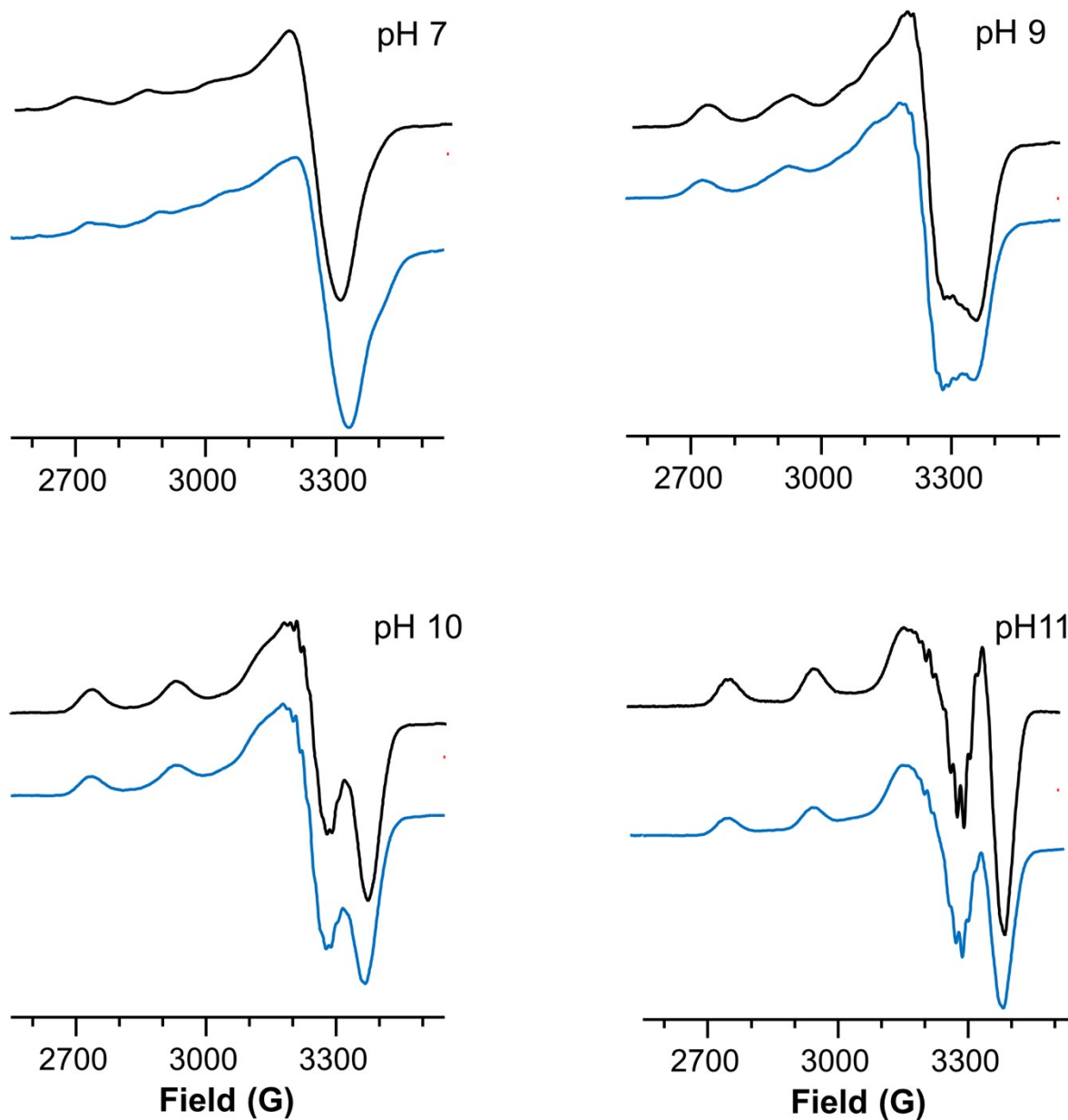


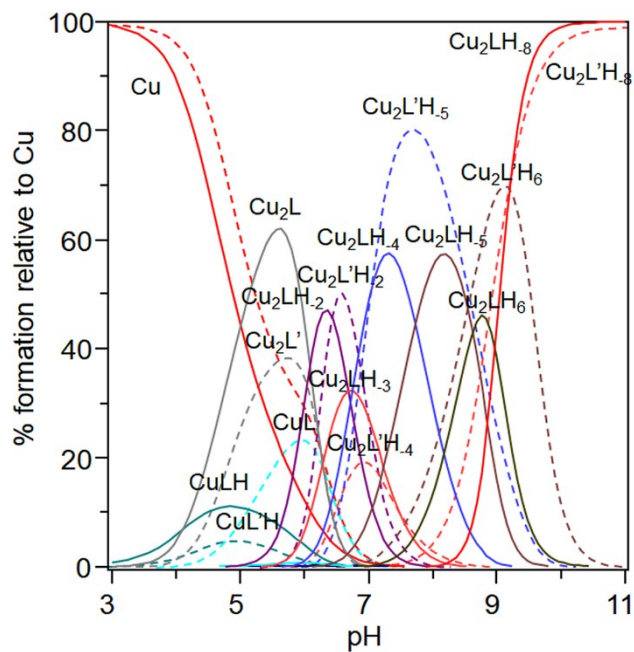
Figure S3. Predicted UV spectrum at the PBE/6-31G\* level including the PCM model for the copper(II) coordinated within the <sup>2</sup>HPGG<sup>5</sup> fragment of Op\_bis-deca peptide.



**Figure S4.** Comparison of EPR spectra of Cu(II)-Op\_bis-deca peptide at Cu(II): L 1:1 molar ratio (black traces) with those obtained at Cu(II): L 2:1 molar ratio (blue traces), at different pH values.  $T = 150 \text{ K}$ ,  $[L] = 1 \times 10^{-3} \text{ M}$ .

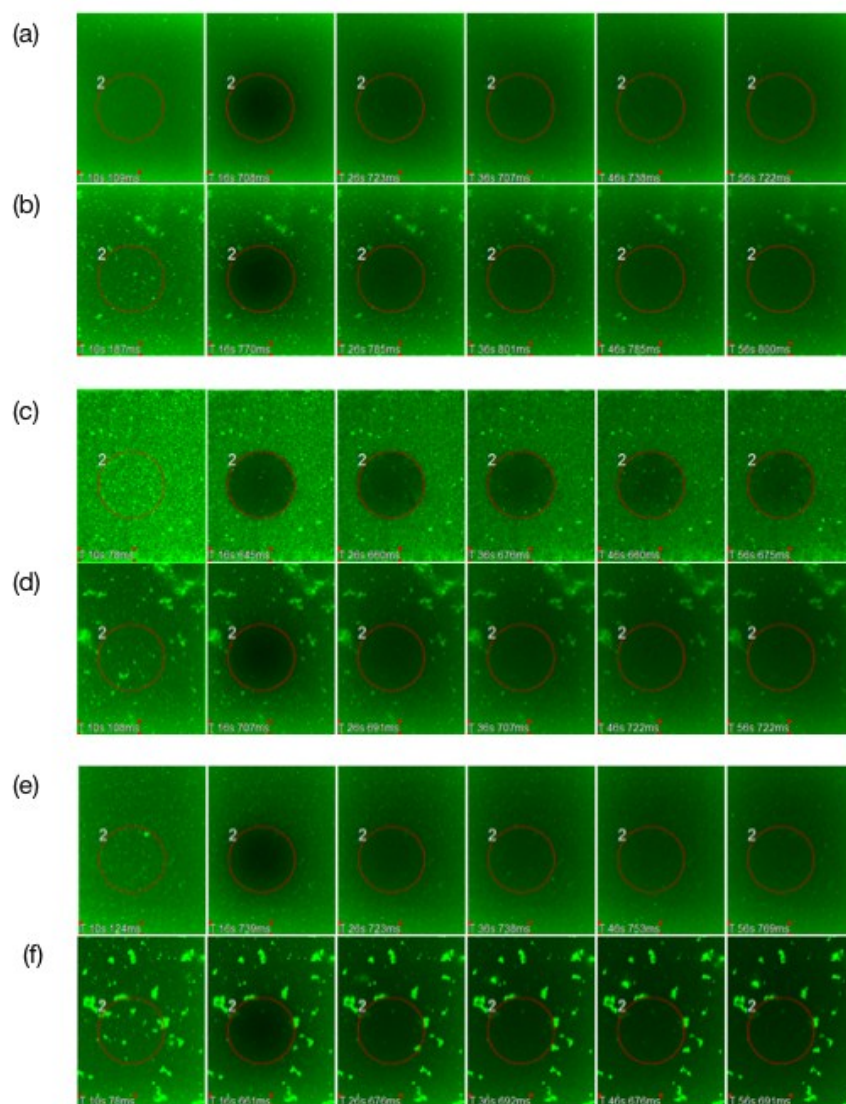
**Table S2.** Coordination parameters for Cu(II) coordinated to Op\_bis-deca peptide. Root mean square displacement during geometry optimization is  $10^{-3}$  Å.

Bond	Bond length (Å)	Angle type	Angle (degrees)
Cu <sup>2+</sup> -O	2.06	N <sub>im</sub> -Cu <sup>2+</sup> -N <sup>-</sup>	97.63
Cu <sup>2+</sup> -N <sub>im</sub>	1.97	N <sub>im</sub> -Cu <sup>2+</sup> -O	84.12
Cu <sup>2+</sup> -N <sup>-</sup>	1.91	O-Cu <sup>2+</sup> -N <sup>-</sup>	96.05
Cu <sup>2+</sup> -N <sup>-</sup>	1.93	N <sup>-</sup> -Cu <sup>2+</sup> -N <sup>-</sup>	86.44



**Figure S4.** Species distribution diagrams for Cu(II) complexes with L= Op\_bis-deca (solid trace) and L'= Hu\_bis-octa (dashed trace), 2:1 metal to ligand molar ratio.  $[L]=[L']= 1 \times 10^{-3}$  M.





**Figure S5.** Representative LSM images for the FRAP experiment of SLB made of POPC–NBD formed on glass surfaces and exposed for 30 min to 1 mM peptide solutions. Micrographs recorded before and after bleach at intervals of 5 s. (a) Control bare SLBs; (b) SLB+ 1 mM CuSO<sub>4</sub>; (b) SLB+ 1 mM CuSO<sub>4</sub>; (c) SLB+ Op\_bis-deca; (d) SLB+ Hu\_bis-deca: CuSO<sub>4</sub> (1:1); (e) SLB+ Hu\_bis-octa; (f) SLB+ Op\_bis-deca: CuSO<sub>4</sub> (1:1).