

## ***Supporting Information for***

### **Enhanced photocatalytic activity of g-C<sub>3</sub>N<sub>4</sub>/MnO composites for hydrogen evolution under visible light**

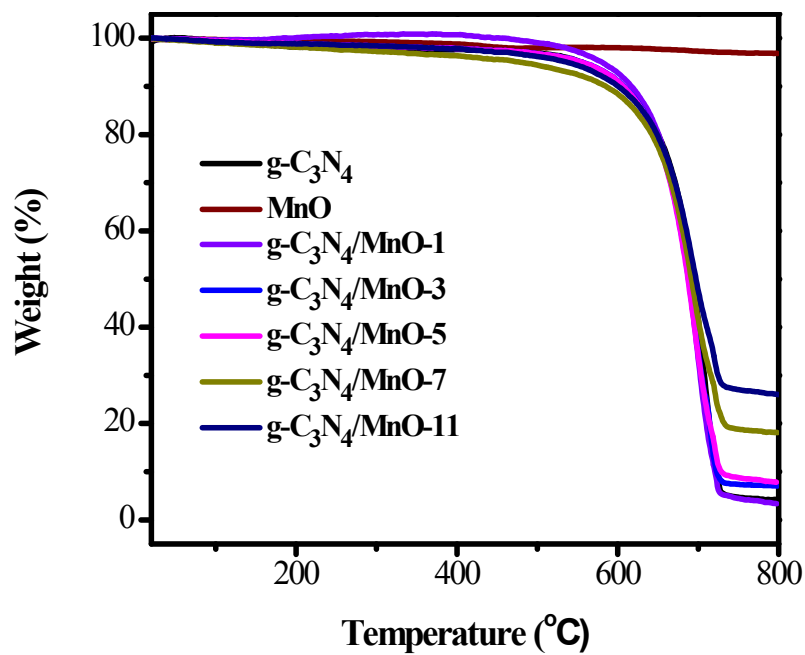
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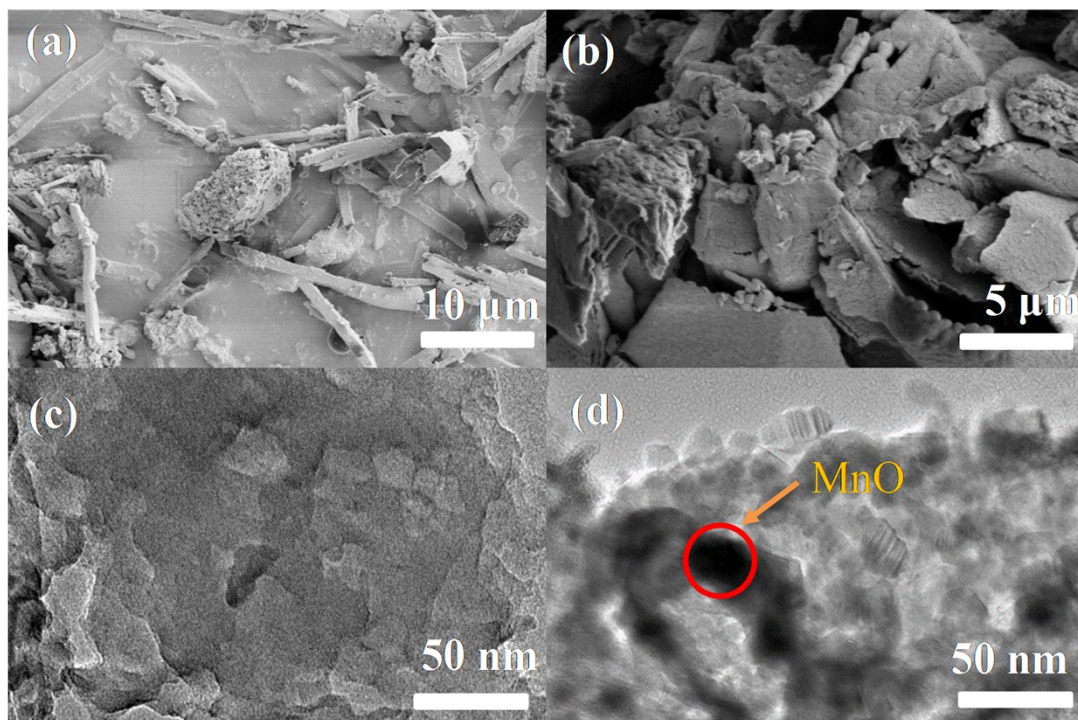
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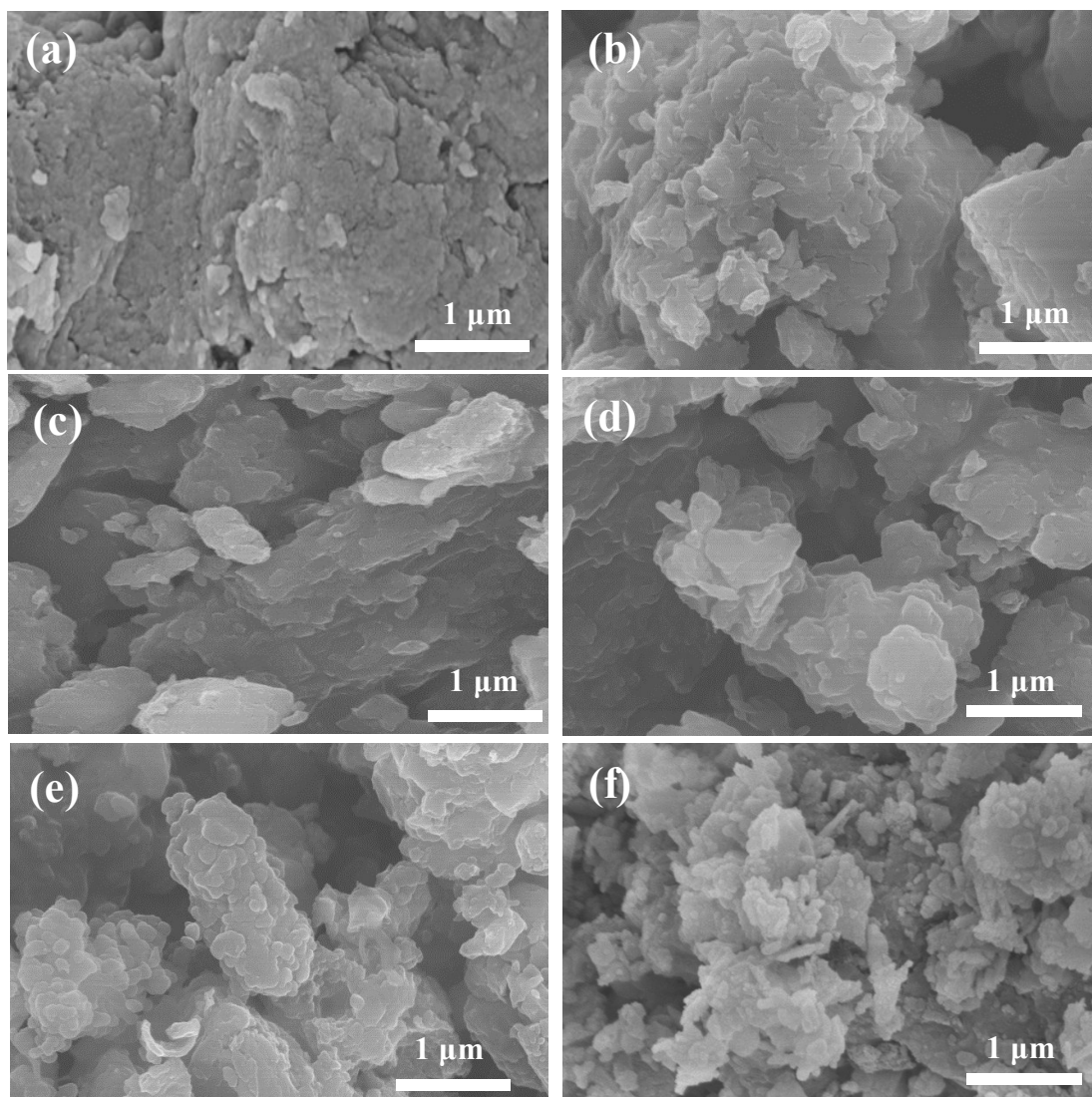
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**Fig. S1.** Thermogravimetric analysis curves of g-C<sub>3</sub>N<sub>4</sub> and the g-C<sub>3</sub>N<sub>4</sub>/MnO composites under N<sub>2</sub> atmosphere.



**Fig. S2.** (a) and (b) Scanning electron microscopy images of pure g-C<sub>3</sub>N<sub>4</sub> from melamine, (c) Transmission electron microscopy images of g-C<sub>3</sub>N<sub>4</sub> and (d) the g-C<sub>3</sub>N<sub>4</sub>/MnO-5 composite.



**Fig. S3.** Scanning electron microscopy images of (a) pure g-C<sub>3</sub>N<sub>4</sub>; (b) g-C<sub>3</sub>N<sub>4</sub>/MnO-1 composite; (c) g-C<sub>3</sub>N<sub>4</sub>/MnO-3 composite; (d) g-C<sub>3</sub>N<sub>4</sub>/MnO-5 composite; (e) g-C<sub>3</sub>N<sub>4</sub>/MnO-7 composite; (f) g-C<sub>3</sub>N<sub>4</sub>/MnO-11 composite;

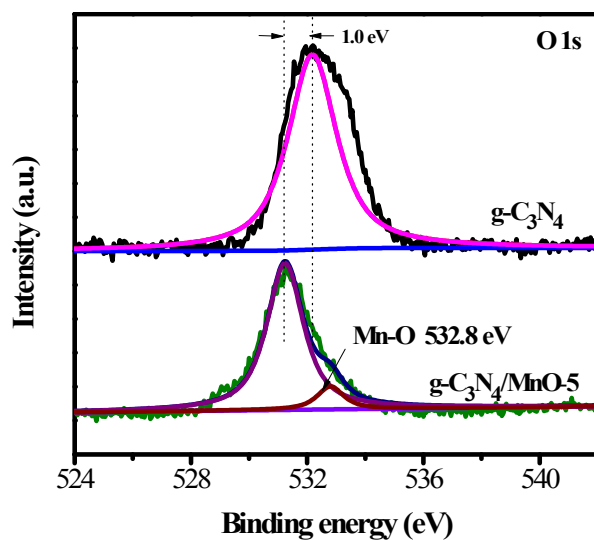


Fig. S4. XPS spectra of the bare  $g\text{-C}_3\text{N}_4$  and the  $g\text{-C}_3\text{N}_4/\text{MnO-5}$  composite for O 1s.

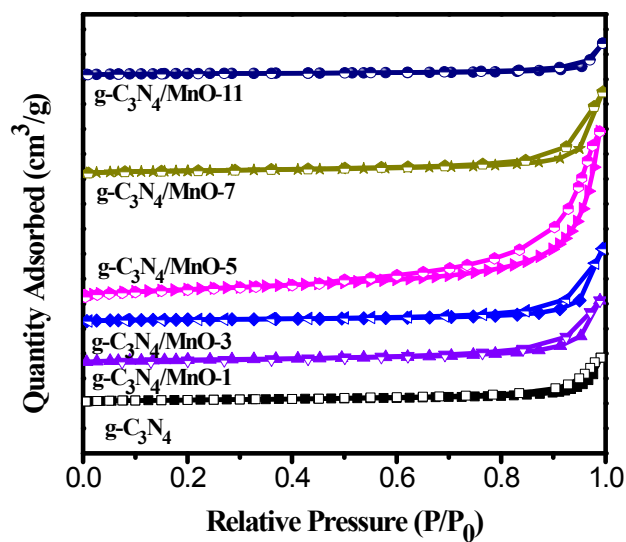


Fig. S5. Nitrogen adsorption (filled symbols) / desorption (empty symbols) isotherms for the bare  $g\text{-C}_3\text{N}_4$  and the  $g\text{-C}_3\text{N}_4/\text{MnO}$  composites collected at 77.3 K.

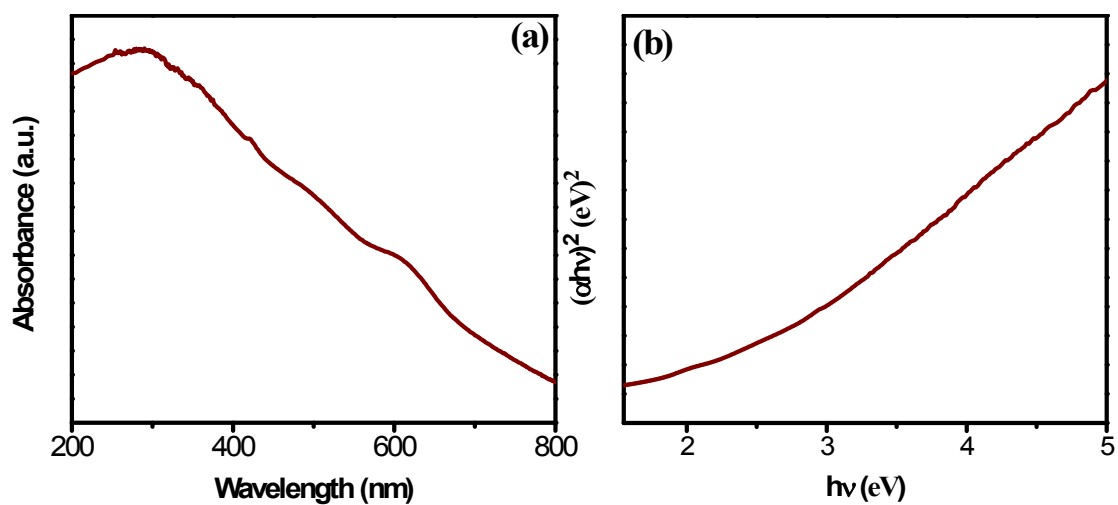


Fig. S6. (a) The UV-Vis reflection spectrum of MnO; (b) Band-gap plot for MnO.

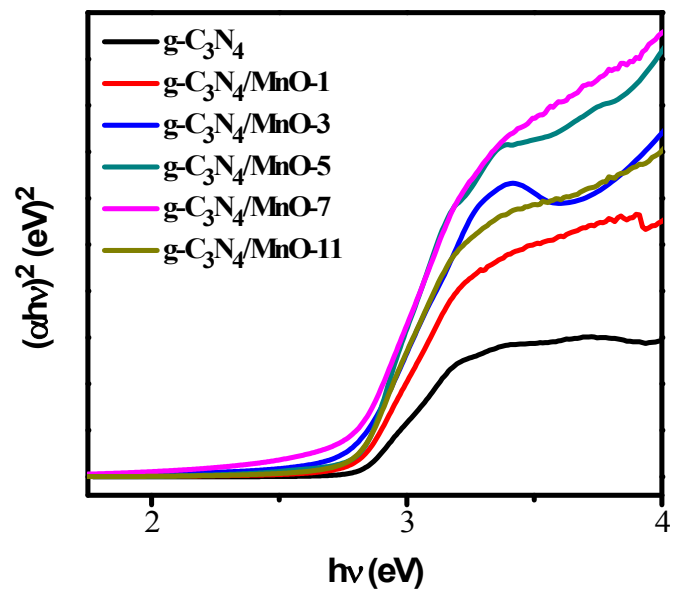


Fig. S7. Band-gap plots of all of the samples.

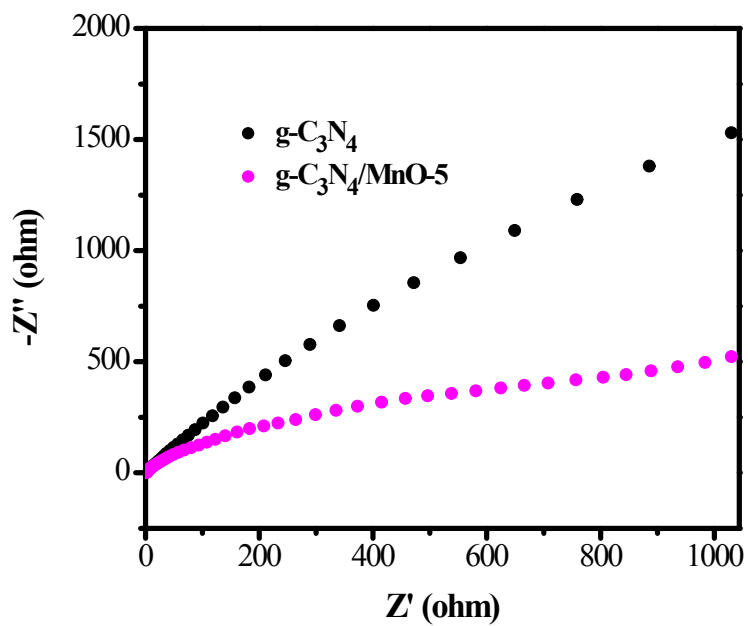
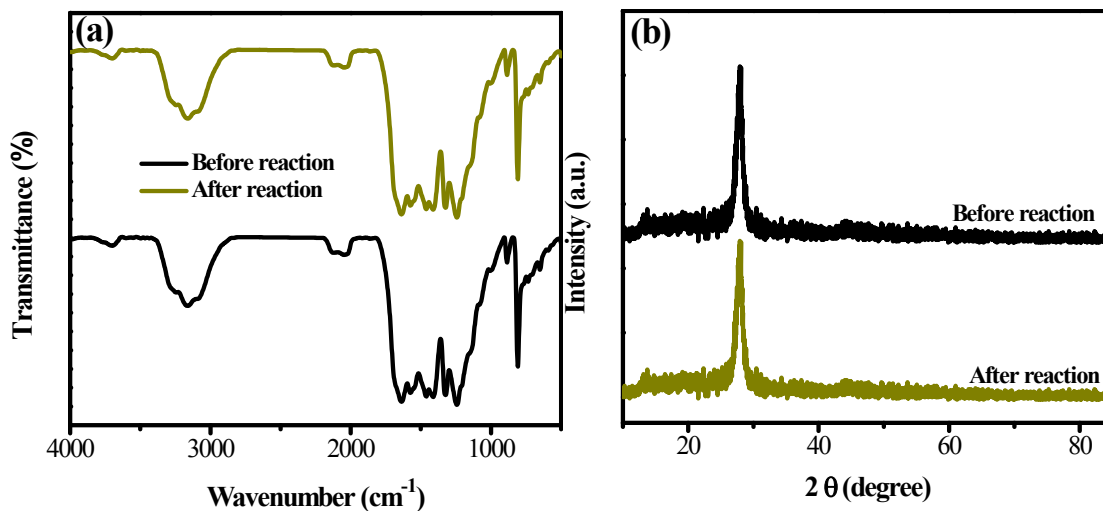
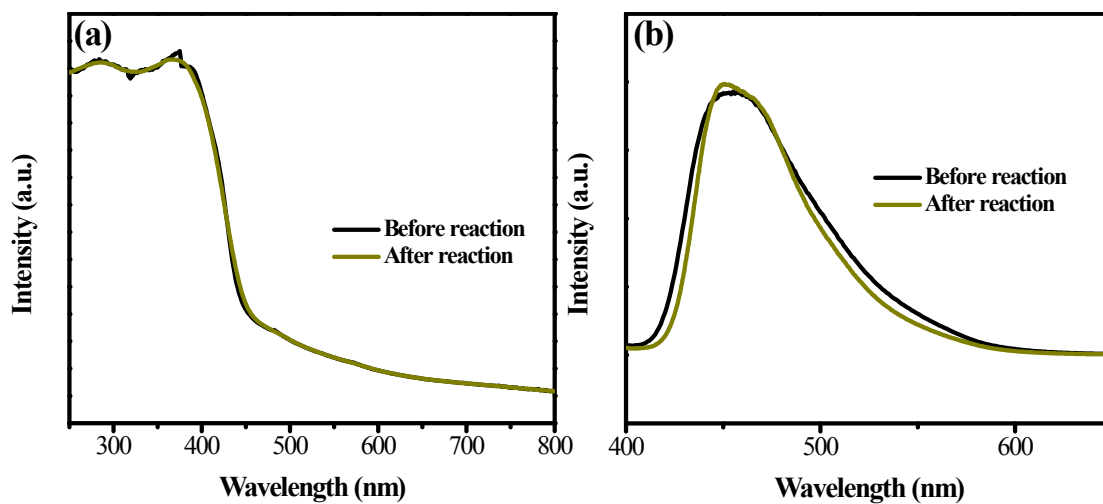


Fig. S8. EIS plots of the bare g-C<sub>3</sub>N<sub>4</sub> and the g-C<sub>3</sub>N<sub>4</sub>/MnO-5 composite.



**Fig. S9.** (a) FT-IR spectra, and (b) Powder XRD patterns of the g-C<sub>3</sub>N<sub>4</sub>/MnO-5 composite before and after irradiation under visible light ( $\lambda > 400$  nm) for 15 h in a triethanolamine/water mixture.



**Fig. S10.** (a) UV-Vis spectra of the g-C<sub>3</sub>N<sub>4</sub>/MnO-5 composite, and (b) Photoluminescence spectra ( $\lambda_{\text{excitation}} = 365$  nm) of the g-C<sub>3</sub>N<sub>4</sub>/MnO-5 composite before and after irradiation under visible light ( $\lambda > 400$  nm) for 15 h in a triethanolamine/water mixture.

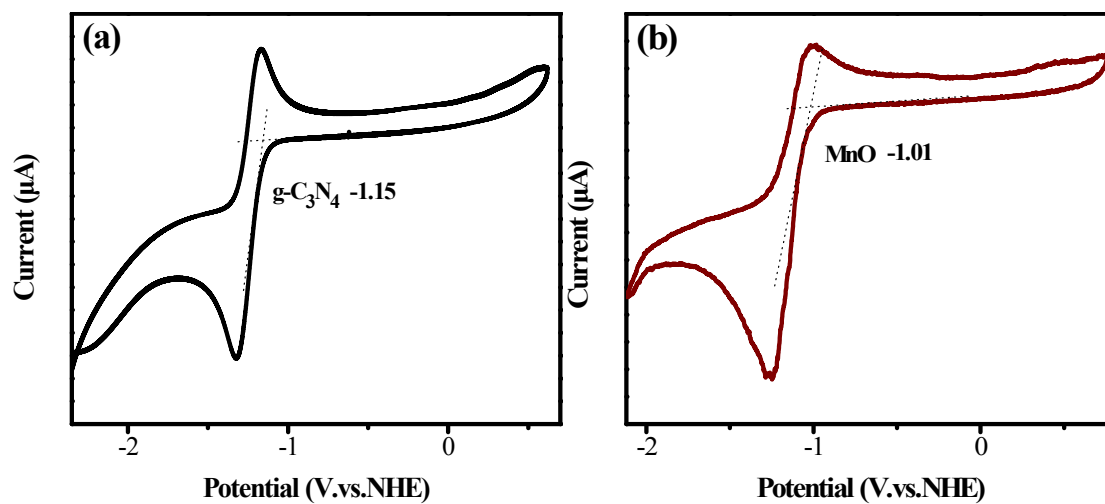


Fig. S11. Cyclic voltammetry measurement for (a)  $\text{g-C}_3\text{N}_4$  and (b)  $\text{MnO}$ .

**Table S1.** Weight percentage of C, N, O, Mn in the g-C<sub>3</sub>N<sub>4</sub>/MnO-5 composite and g-C<sub>3</sub>N<sub>4</sub>

samples	Weight (%)			Atomic Ratios	
	C <sup>(a)</sup>	N <sup>(a)</sup>	Mn <sup>(b)</sup>	N/C	C/Mn
g-C <sub>3</sub> N <sub>4</sub>	34.66	59.25	--	1.54	-
g-C <sub>3</sub> N <sub>4</sub> /MnO-5	32.51	57.64	2.89	1.56	54

(a) Data obtained by EA, (b) Data obtained by ICP-MS

**Table S2.** The BET surface area, pore volume and pore size of g-C<sub>3</sub>N<sub>4</sub>, g-C<sub>3</sub>N<sub>4</sub>/MnO-5 composite

Photocatalyst	<sup>a</sup> S <sub>BET</sub> / m <sup>2</sup> ·g <sup>-1</sup>	<sup>b</sup> Pore volume/cm <sup>3</sup> ·g <sup>-1</sup>	Pore size/nm
g-C <sub>3</sub> N <sub>4</sub>	12	0.06	20.8
g-C <sub>3</sub> N <sub>4</sub> /MnO-1	13	0.10	29.4
g-C <sub>3</sub> N <sub>4</sub> /MnO-3	14	0.01	33.3
g-C <sub>3</sub> N <sub>4</sub> /MnO-5	48	0.25	18.8
g-C <sub>3</sub> N <sub>4</sub> /MnO-7	18	0.13	28.8
g-C <sub>3</sub> N <sub>4</sub> /MnO-11	6	0.03	23.5
MnO	9	0.05	0

a. BET Surface Area; b. *t*-plot micropore volume.**Table S3.** Radiative fluorescence lifetimes and their relative percentages of photoexcited charge carriers in the g-C<sub>3</sub>N<sub>4</sub> and g-C<sub>3</sub>N<sub>4</sub>/MnO-5 composite.

Sample	$\tau_1$ (ns)(Rel %)	$\tau_2$ (ns)(Rel %)	$\tau_3$ (ns)(Rel %)	$t_{av}$ (ns) <sup>a</sup>
g-C <sub>3</sub> N <sub>4</sub>	1.13-38.00	4.41-47.55	26.62-14.45	17.62
g-C <sub>3</sub> N <sub>4</sub> /MnO-5	0.90-37.56	3.49-52.91	19.73-9.53	10.79