

Electronic Supplementary Information

The stability of group 10 metal POCOP pincer complexes: decomposition/reconstruction pathways of the pincer backbone

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Synthesis and characterization of [2,6-(*i*Pr₂PO)₂C₆H₃]PtI

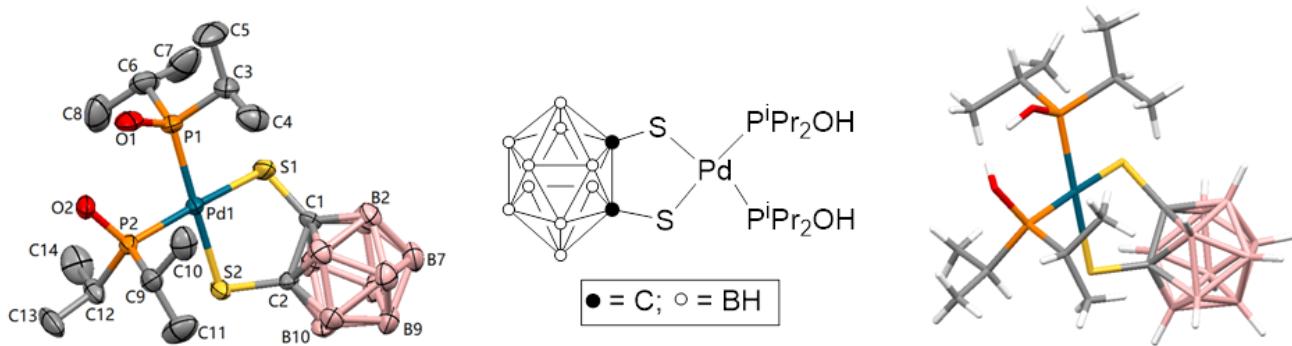
A mixture of [2,6-(*i*Pr₂PO)₂C₆H₃]PtCl (0.25 mmol), THF (5 mL), methanol (5 mL) and NaI (1 mmol) was stirred at room temperature in a flask for 6 h. Solvents in the reaction mixture were then removed under vacuum and the resulting residue was extracted with toluene. Removal of toluene from the combined extractions followed by washing with hexanes produced [2,6-(*i*Pr₂PO)₂C₆H₃]PtI as a light yellow solid (80%). ¹H NMR (600 MHz, CDCl₃, δ): 7.01 (t, 1H, ArH, J_{H-H} = 8.1 Hz), 6.58 (d, 2H, ArH, J_{H-H} = 8.1 Hz), 2.63–2.70 (m, 4H, CH(CH₃)₂), 1.12–1.40 (m, 24H, CH(CH₃)₂). ¹³C{¹H} NMR (151 MHz, CDCl₃, δ): 164.4 (t, ArC, J_{C-P} = 6.2 Hz), 126.9 (s, ArC), 122.7 (s, ArC), 105.5 (t, ArC, J_{C-P} = 6.2 Hz), 29.5–29.9 (m, CH(CH₃)₂), 17.7–17.8 (m, CH(CH₃)₂), 16.7 (s, Pt satellites, CH(CH₃)₂, J_{C-Pt} = 20.7 Hz). ³¹P{¹H} NMR (243 MHz, CDCl₃, δ): 177.3 (s, Pt satellites, J_{P-Pt} = 2977 Hz).

Table S1. Summary of crystal data and structure refinement for complexes **2a**, **2c** and **3b**

Complex	2a	2c	3b
Empirical formula	C ₂₄ H ₃₆ O ₂ P ₂ Ni	C ₂₄ H ₃₆ O ₂ P ₂ Pt	C ₂₀ H ₄₂ B ₁₀ O ₂ P ₂ Pd
Formula weight	477.18	613.56	590.97
Temp, K	170	295	294
Crystal system	Monoclinic	Monoclinic	Orthorhombic
Space group	P2 ₁ /m	P2 ₁ /m	Cmce
<i>a</i> , Å	8.3415(1)	8.4272(3)	13.2661(2)
<i>b</i> , Å	16.5817(3)	16.7958(4)	15.3835(2)
<i>c</i> , Å	9.2895(1)	9.4851(3)	28.9916(3)
α (°)	90	90	90
β (°)	112.390(2)	112.887(4)	90
γ (°)	90	90	90
Volume, Å ³	1188.02(3)	1236.84(7)	5916.58(13)
Z	2	2	8
<i>d</i> _{calc} , g cm ⁻³	1.334	1.647	1.327
λ, Å	1.54184	0.71073	1.54184
μ, mm ⁻¹	2.577	5.819	6.194
No. of data collected	9944	11328	11829
No. of unique data	2379	3043	2976
<i>R</i> _{int}	0.0355	0.0339	0.0432
Goodness-of-fit on <i>F</i> ²	1.104	1.061	1.071
<i>R</i> ₁ , w <i>R</i> ₂ (<i>I</i> > 2σ(<i>I</i>))	0.0397, 0.1024	0.0292, 0.0539	0.0354, 0.0877
<i>R</i> ₁ , w <i>R</i> ₂ (all data)	0.0408, 0.1037	0.0386, 0.0581	0.0377, 0.0889

Table S2. Summary of crystal data and structure refinement for complexes **3c**, **5a** and **6a**

Complex	3c	5a	6a
Empirical formula	C ₂₀ H ₄₂ B ₁₀ O ₂ P ₂ Pt	C ₃₈ H ₇₂ B ₁₀ O ₄ P ₄ S ₂ Ni ₂	C ₁₆ H ₄₈ B ₂₀ P ₂ S ₂ Ni
Formula weight	679.66	1006.47	641.51
Temp, K	170	295	150
Crystal system	Orthorhombic	Orthorhombic	Monoclinic
Space group	Cmce	Pbcn	P2 ₁ /c
<i>a</i> , Å	13.1407(2)	18.7762(6)	11.4737(2)
<i>b</i> , Å	15.2729(2)	15.5899(5)	9.9192(2)
<i>c</i> , Å	28.7118(3)	17.6800(6)	14.6100(3)
α (°)	90	90	90
β (°)	90	90	100.711(2)
γ (°)	90	90	90
Volume, Å ³	5762.36(13)	5175.3(3)	1633.79(6)
Z	8	4	2
<i>d</i> _{calc} , g cm ⁻³	1.567	1.292	1.304
λ , Å	1.54184	1.54184	1.54184
μ , mm ⁻¹	10.275	0.968	3.031
No. of data collected	12332	24457	9910
No. of unique data	2940	5072	3179
<i>R</i> _{int}	0.0304	0.0408	0.0238
Goodness-of-fit on <i>F</i> ²	1.042	1.133	1.043
<i>R</i> ₁ , w <i>R</i> ₂ (<i>I</i> > 2σ(<i>I</i>))	0.0230, 0.0531	0.0529, 0.1179	0.0257, 0.0631
<i>R</i> ₁ , w <i>R</i> ₂ (all data)	0.0266, 0.0553	0.0770, 0.1318	0.0310, 0.0659



Selected bond lengths (Å) and angles (°): Pd1-S1, 2.379(2); Pd1-S2, 2.3852(19); Pd1-P1, 2.276(2); Pd1-P2, 2.2863(19); P1-O1, 1.575(7); P2-O2, 1.553(6); C1-C2, 1.658(10); P1-Pd1-P2, 93.21(7); P1-Pd1-S1, 89.43(8); S1-Pd1-S2, 87.84(7); P2-Pd1-S2, 89.48(7); S1-Pd1-P2, 177.13(8); S2-Pd1-P1, 176.75(8).

Fig. S1 Molecular image of Pd(II)(*i*Pr₂POH)₂(1,2-S₂-*o*-carborane)

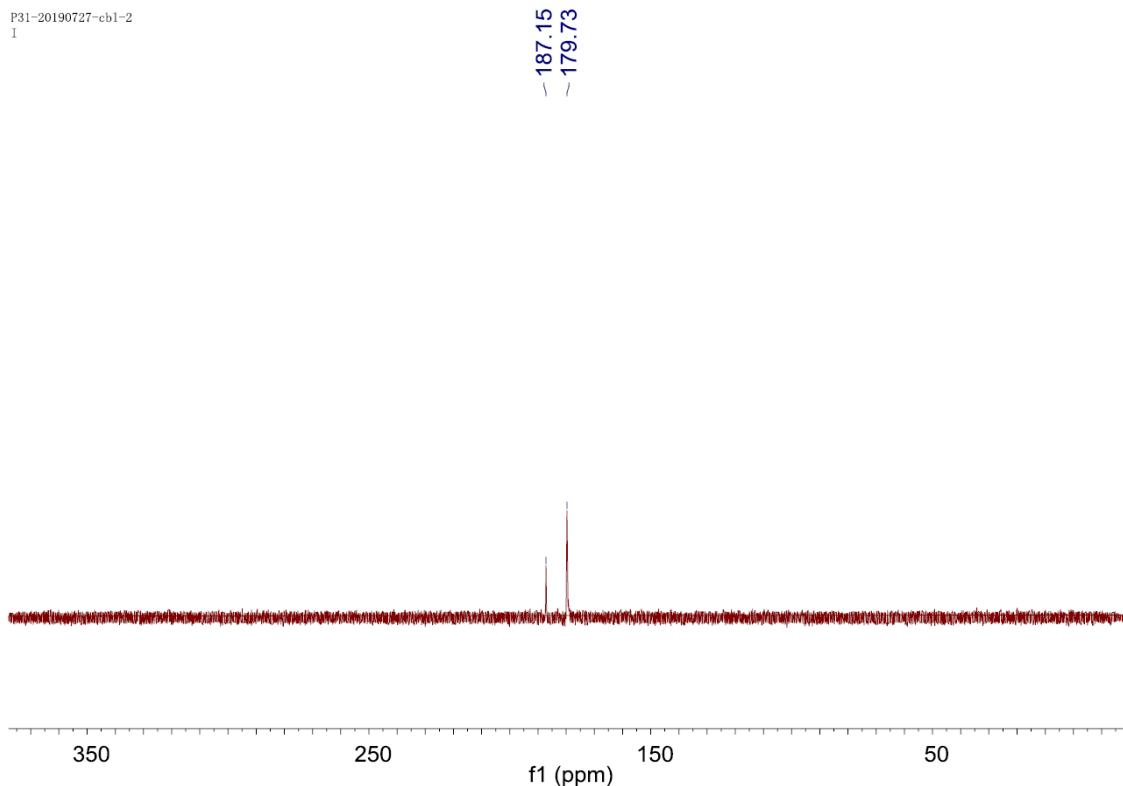


Fig. S2 $^{31}\text{P}\{\text{H}\}$ NMR spectra of the reaction of **4a** with 1 equiv of $^n\text{BuLi}$ in toluene- d_8 .

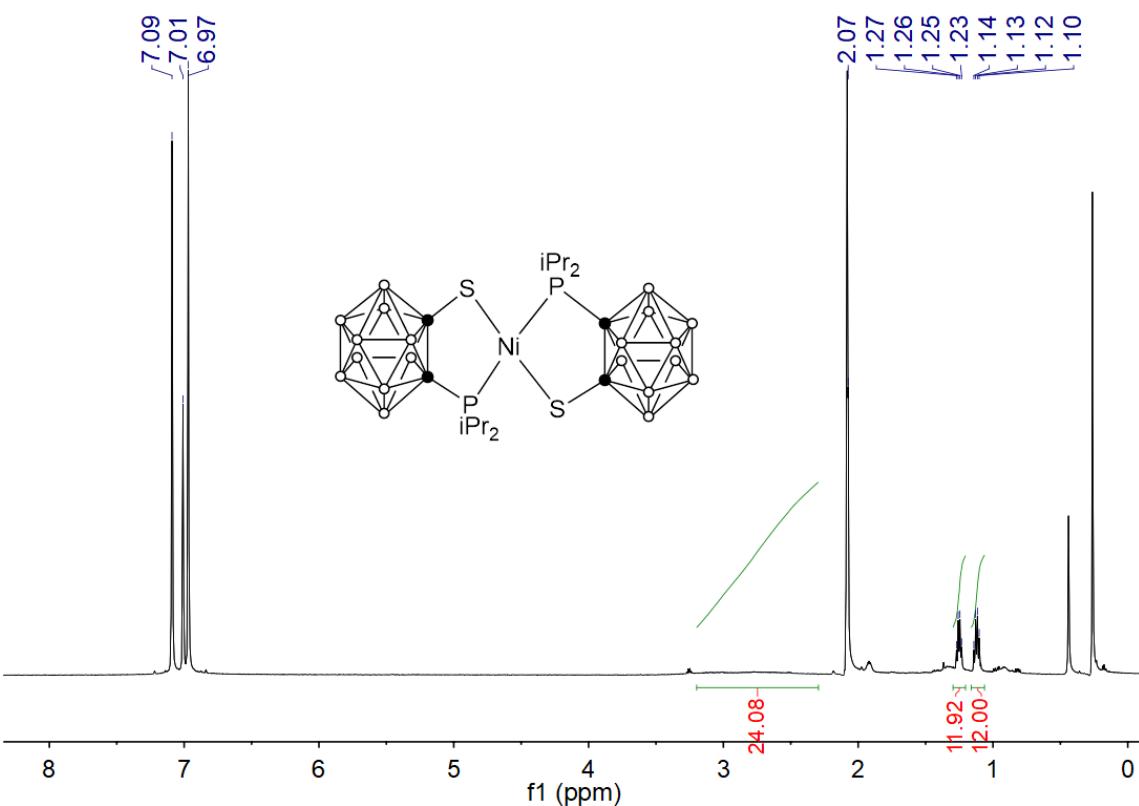


Fig. S3 ^1H NMR spectrum of complex **6a** (600 MHz, toluene- d_8)

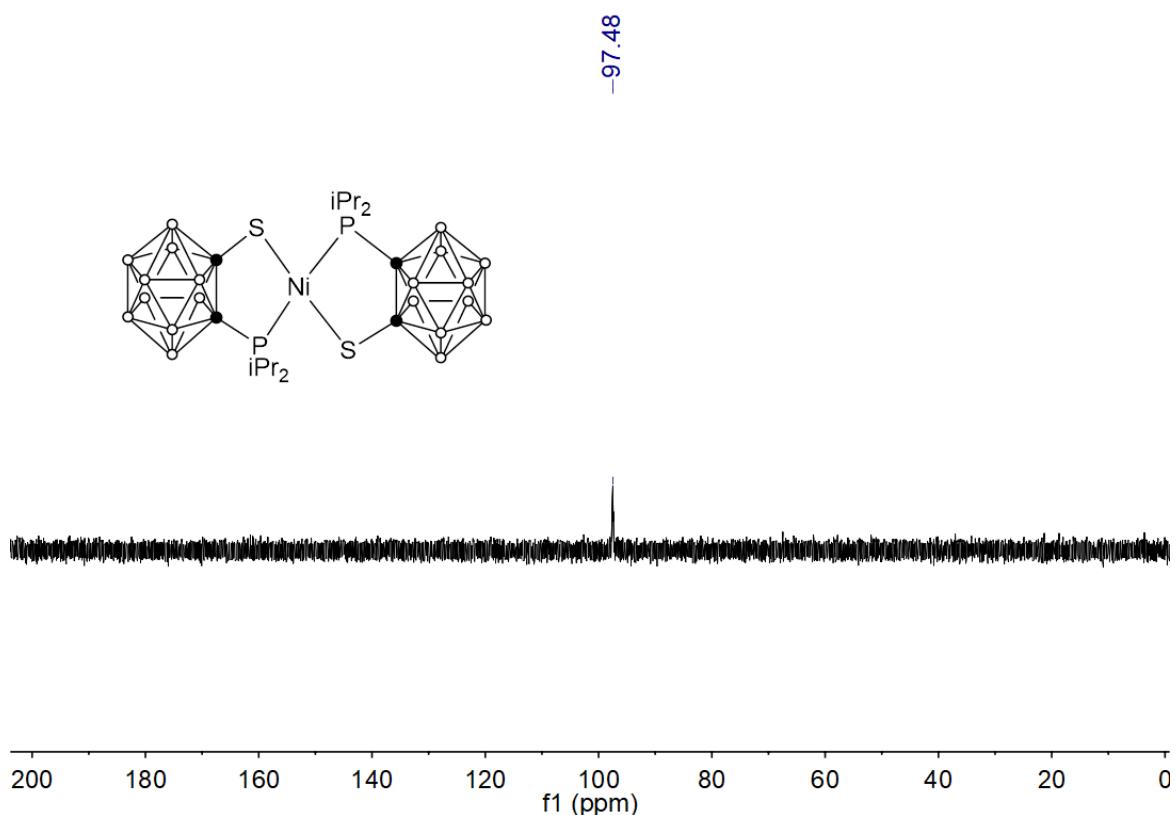


Fig. S4 $^{31}\text{P}\{\text{H}\}$ NMR spectrum of complex **6a** (243 MHz, toluene- d_8)

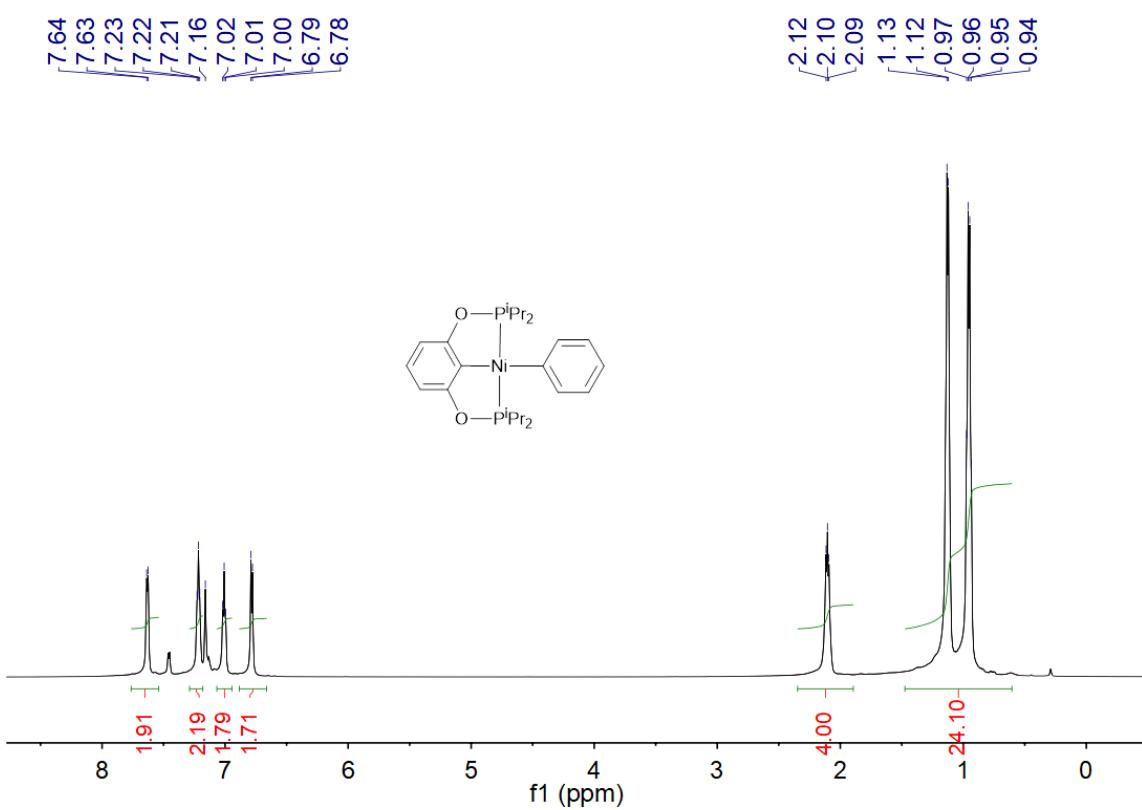


Fig. S5 ^1H NMR spectrum of complex **2a** (600 MHz, benzene- d_6)

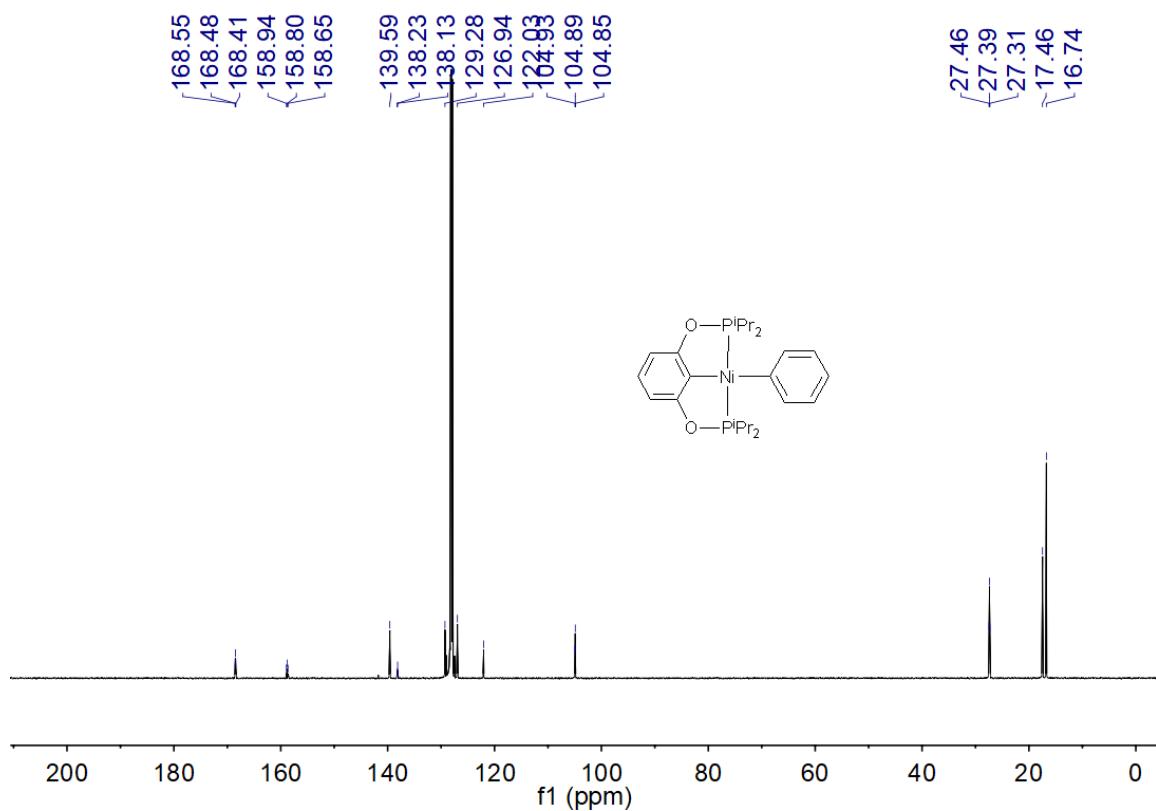


Fig. S6 $^{13}\text{C}\{\text{H}\}$ NMR spectrum of complex **2a** (151 MHz, benzene- d_6)

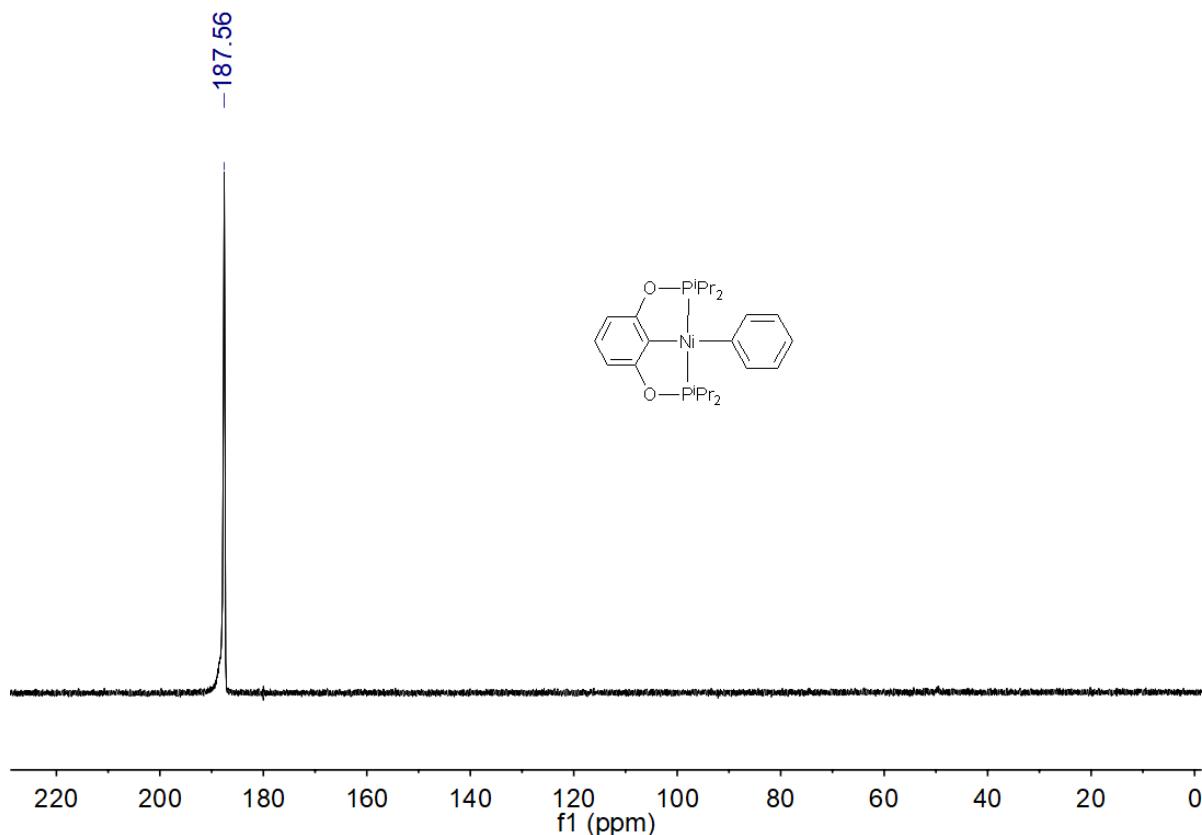


Fig. S7 $^{31}\text{P}\{\text{H}\}$ NMR spectrum of complex **2a** (243 MHz, benzene- d_6)

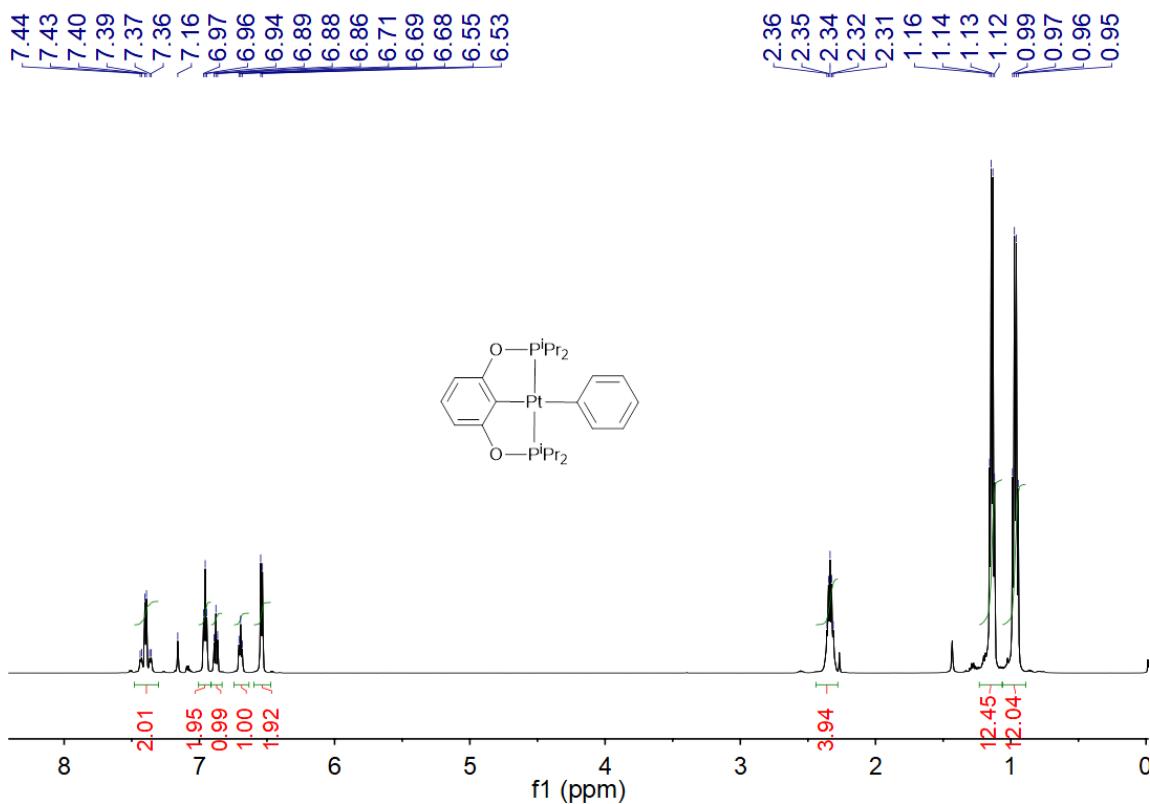


Fig. S8 ^1H NMR spectrum of complex **2c** (600 MHz, benzene- d_6)

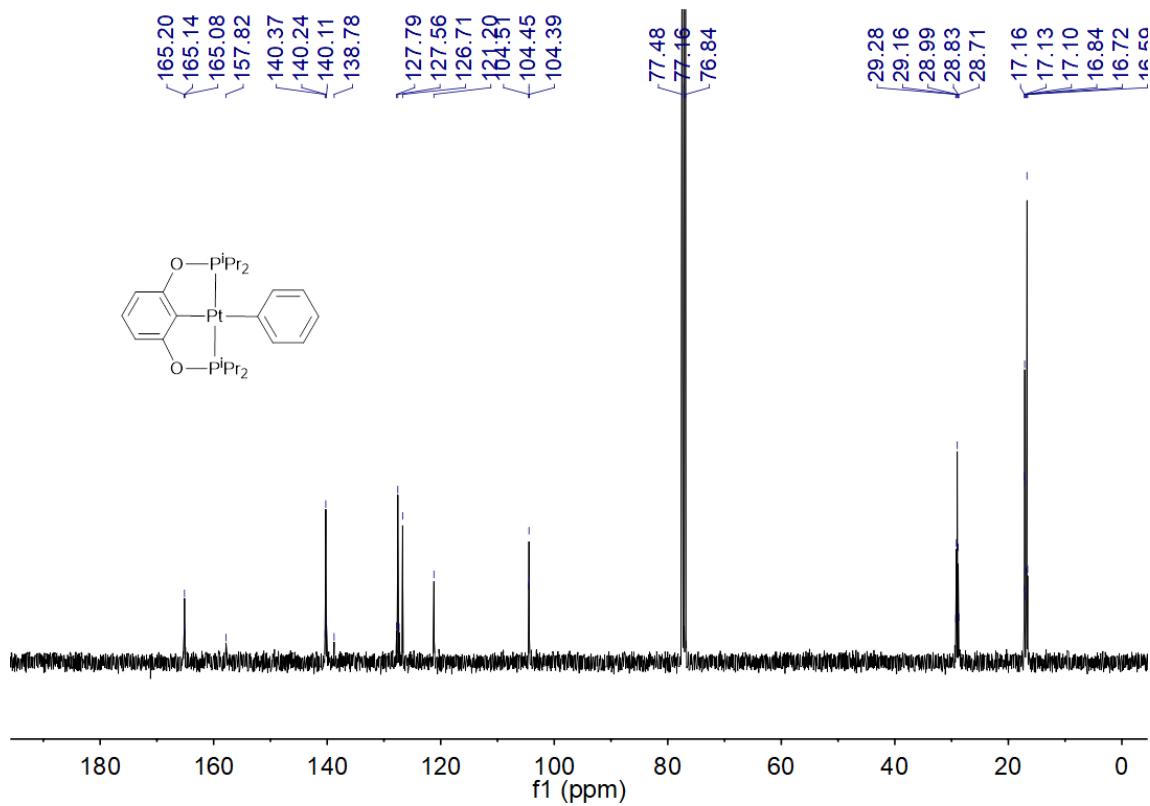


Fig. S9 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of complex **2c** (101 MHz, CDCl_3)

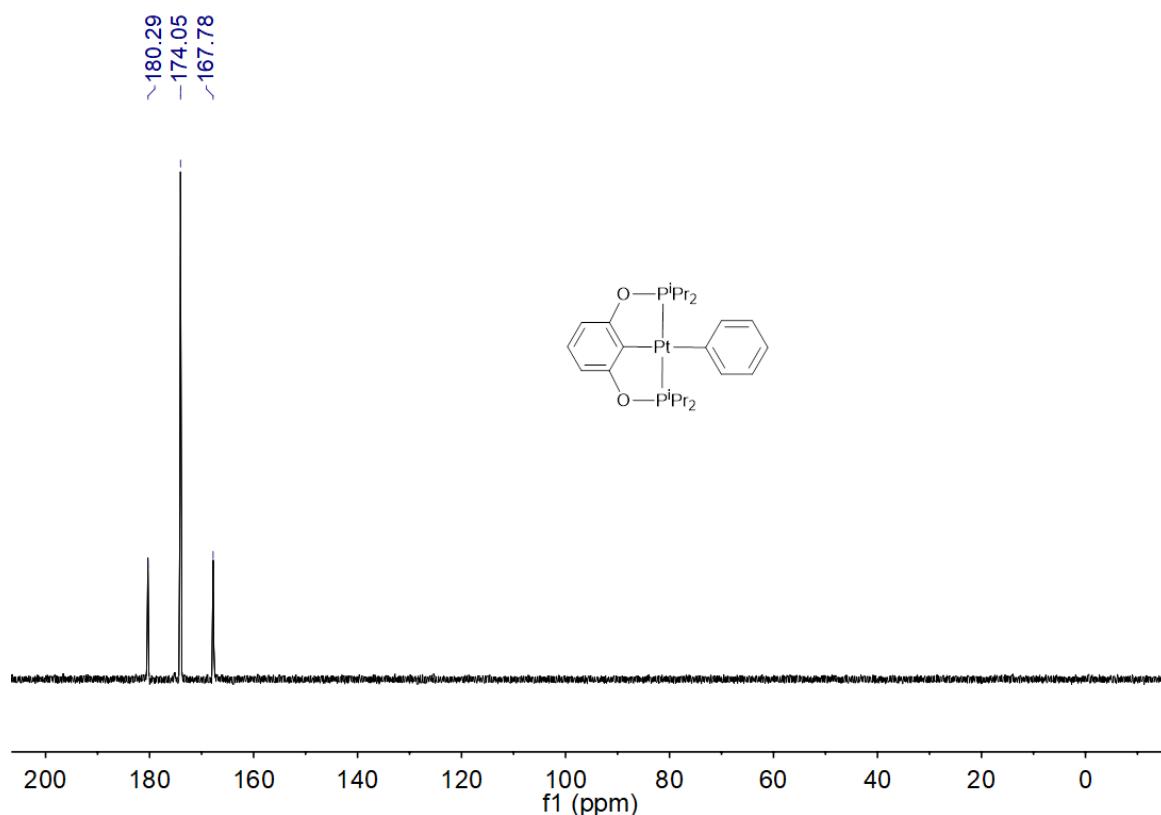


Fig. S10 $^{31}\text{P}\{\text{H}\}$ NMR spectrum of complex **2c** (243 MHz, benzene- d_6)

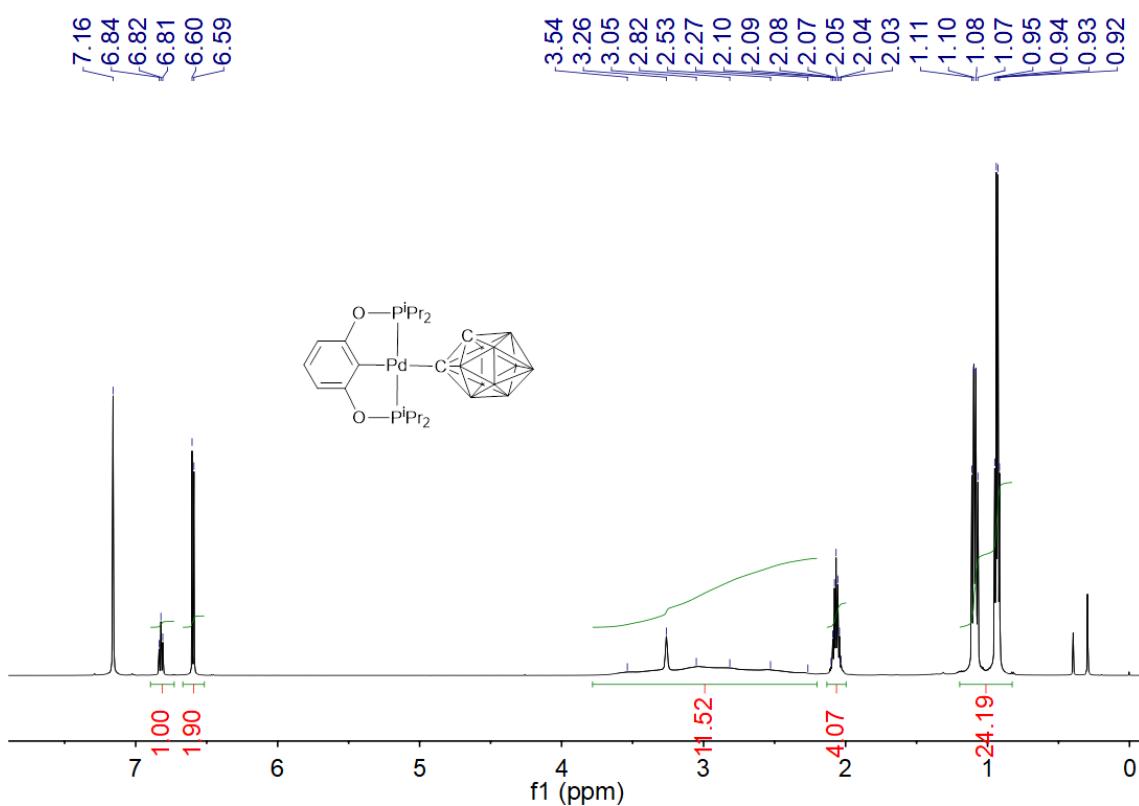


Fig. S11 ^1H NMR spectrum of complex **3b** (600 MHz, benzene- d_6)

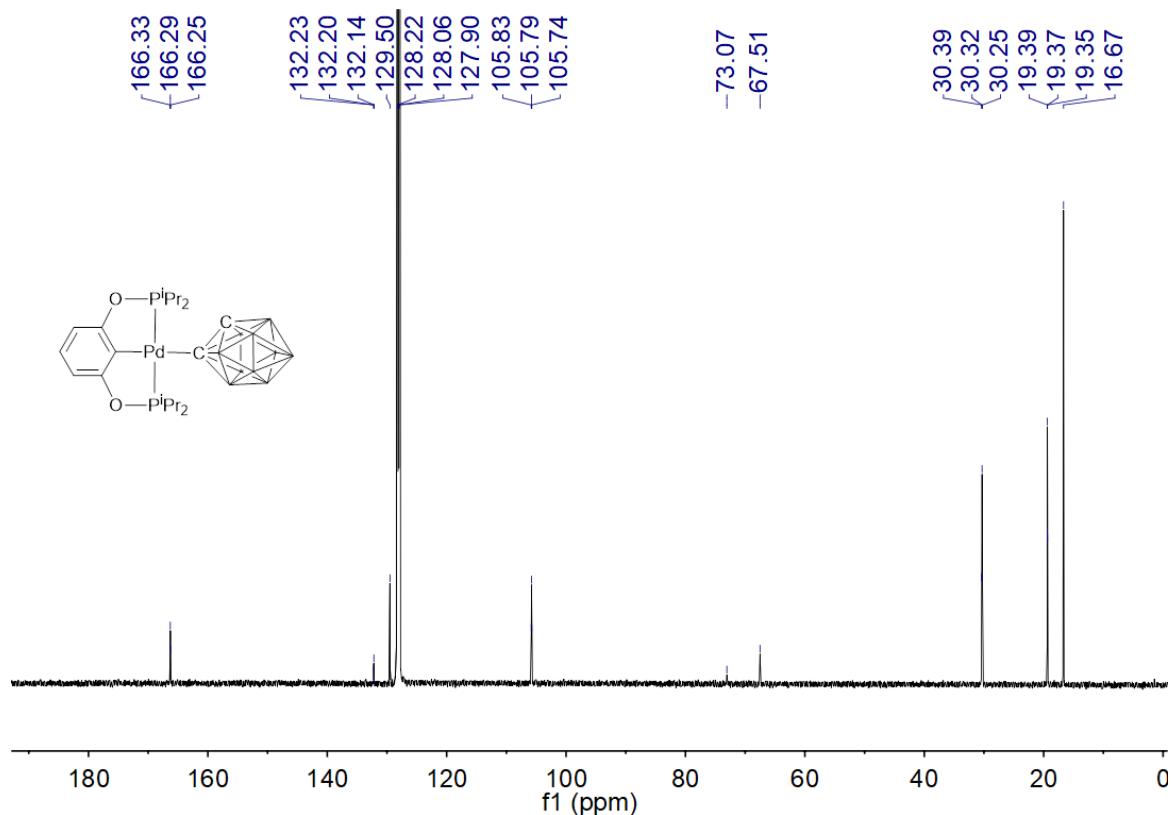


Fig. S12 $^{13}\text{C}\{\text{H}\}$ NMR spectrum of complex **3b** (151 MHz, benzene- d_6)

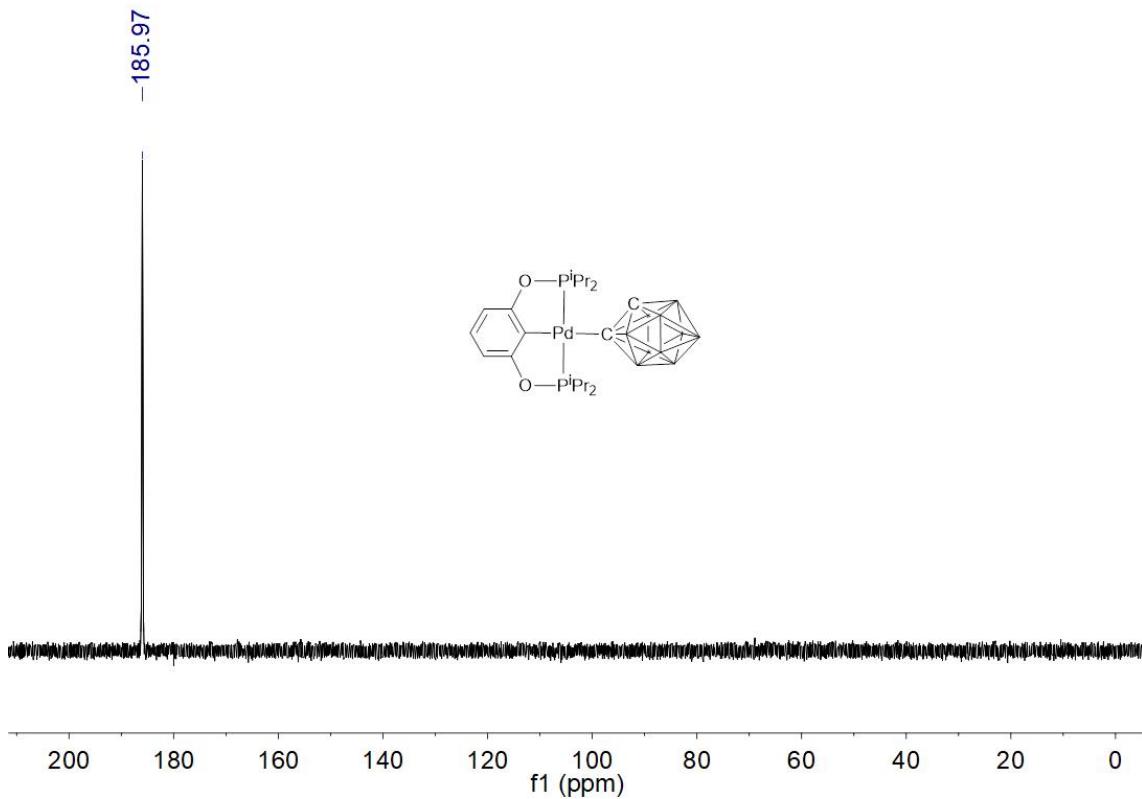


Fig. S13 $^{31}\text{P}\{\text{H}\}$ NMR spectrum of complex **3b** (243 MHz, benzene- d_6)

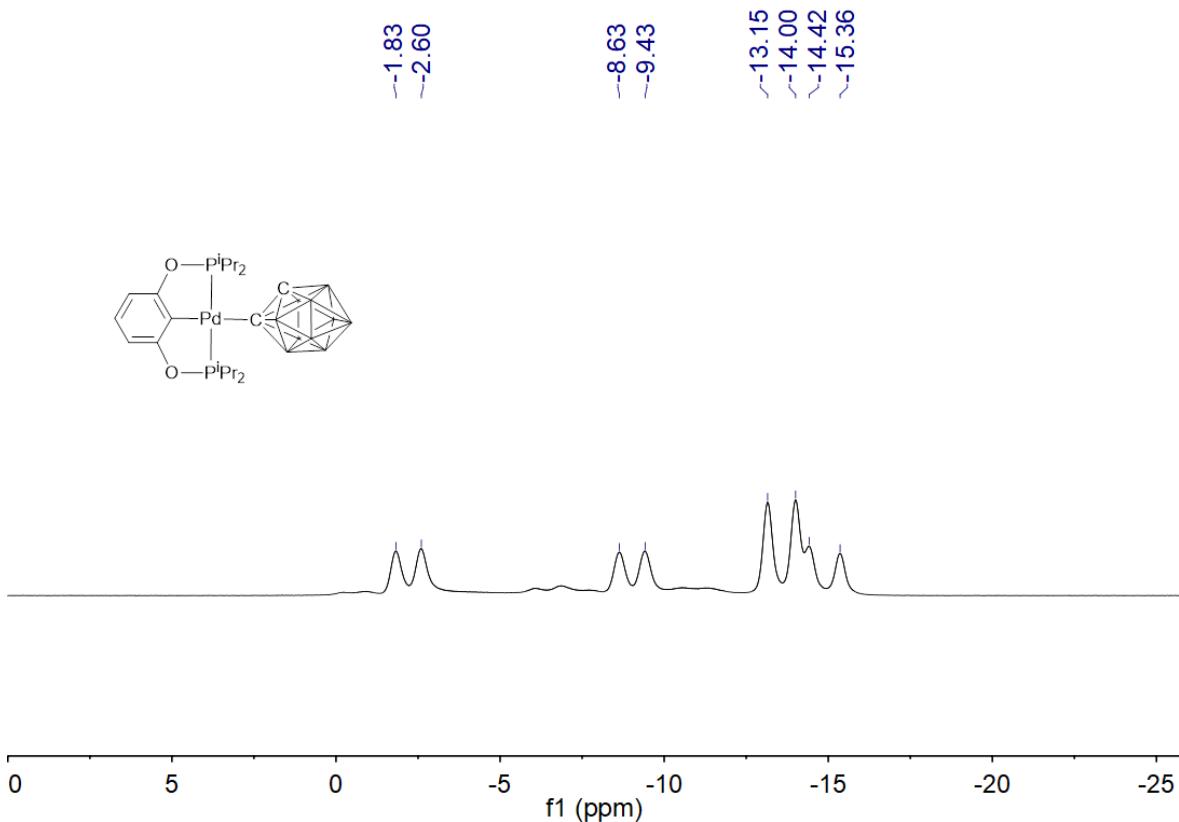


Fig. S14 ^{11}B NMR spectrum of complex **3b** (193 MHz, benzene- d_6)

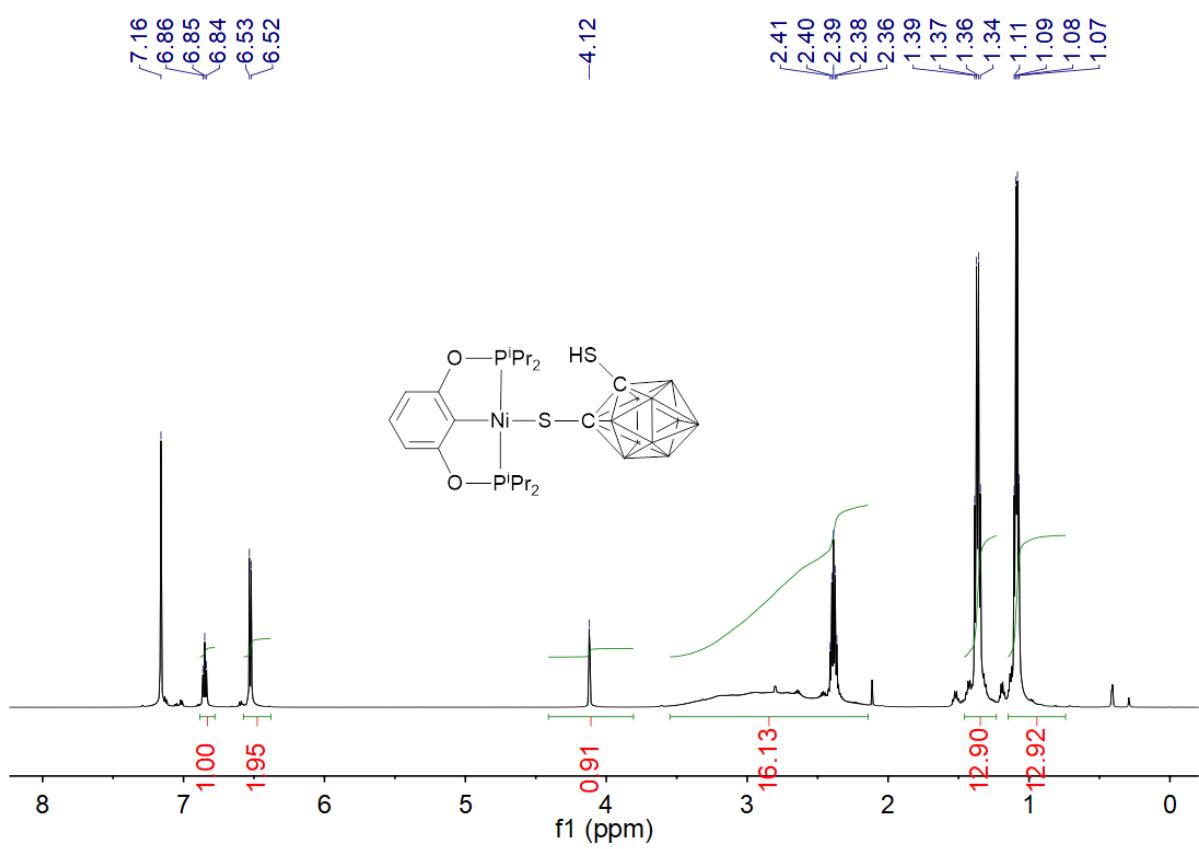


Fig. S15 ^1H NMR spectrum of complex **4a** (600 MHz, benzene- d_6)

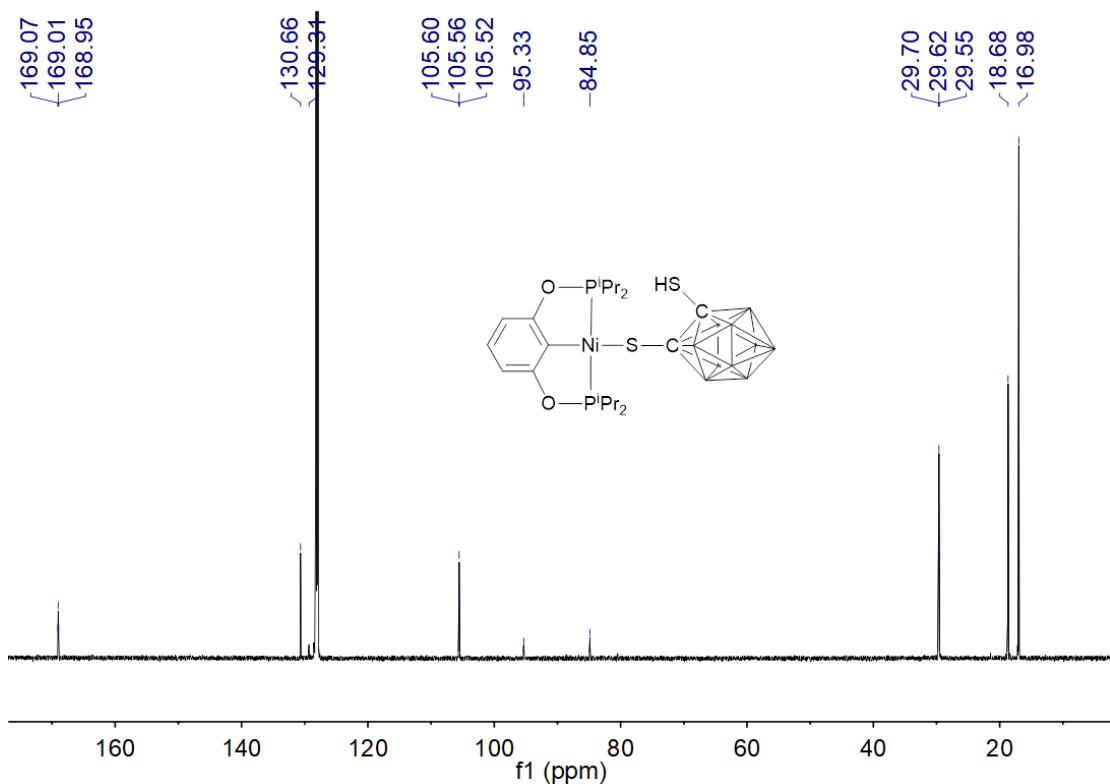


Fig. S16 $^{13}\text{C}\{\text{H}\}$ NMR spectrum of complex **4a** (151 MHz, benzene- d_6)

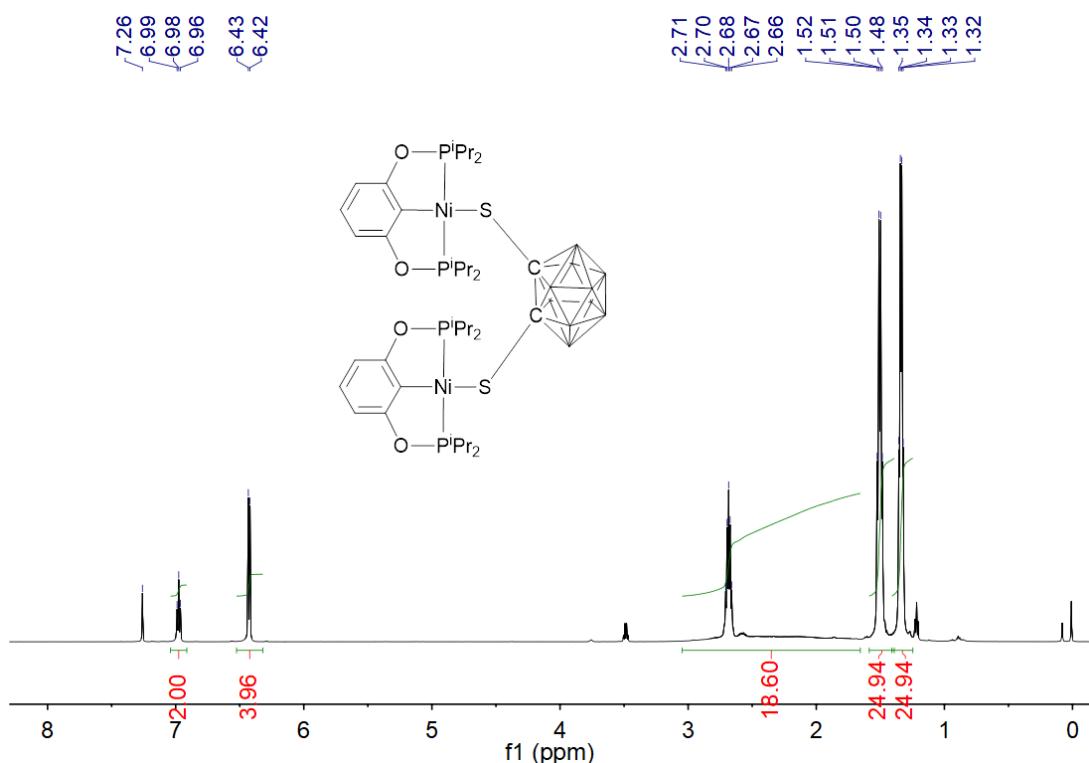


Fig. S17 ^1H NMR spectrum of complex **5a** (600 MHz, CDCl_3)

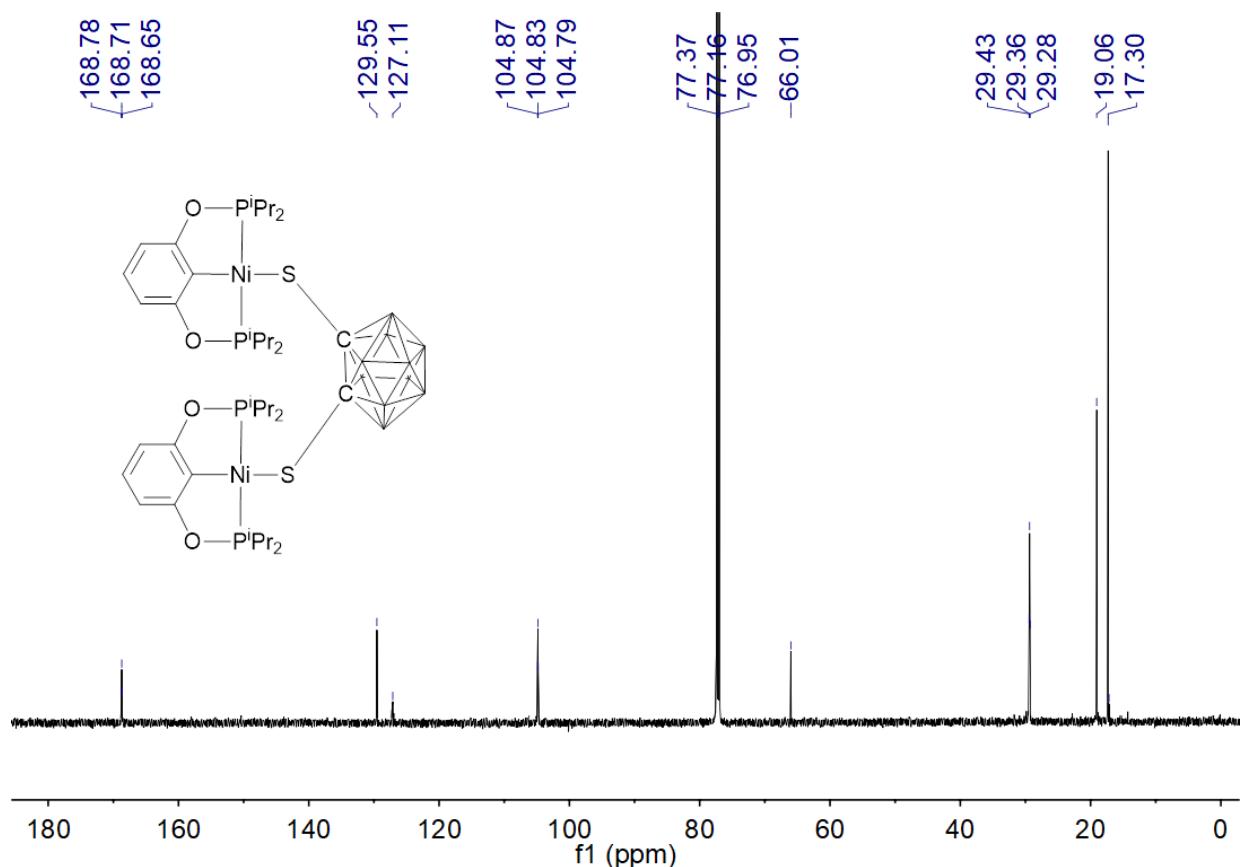


Fig. S18 $^{13}\text{C}\{\text{H}\}$ NMR spectrum of complex **5a** (151 MHz, CDCl_3)