Supplementary Information

Photofunctions of iridium(III) complexes in vesicles: long-lived excited states and visible-light sensitization for hydrogen evolution in aqueous solution

Shinogu Sebata, Shin-ya Takizawa,* Naoya Ikuta and Shigeru Murata*

Department of Basic Science, Graduate School of Arts and Sciences, The University of Tokyo, 3-8-1 Komaba, Meguro-ku, Tokyo 153-8902, Japan.

E-mail: ctaki@mail.ecc.u-tokyo.ac.jp, cmura@mail.ecc.u-tokyo.ac.jp

Contents

(1)	Synthetic scheme and ¹ H NMR spectrum of Ir3	2
(2)	UV-vis absorption and phosphorescence spectra of Ir2 and Ir3	3
	in CH ₃ CN and aqueous CH ₃ CN	
(3)	Phosphorescent properties of Ir2 and Ir3 at 77 K	3
(4)	Cyclic voltammograms of the blank solution and ferrocene	4
(5)	Photographs of the vesicle solution before and after ultrasonic process	4
(6)	Modified Stern-Volmer plots obtained from quenching experiments	5
(7)	Changes observed in absorption spectra during photochemical	5
	H_2 evolution using Ir2 and Ir3-embedded vesicles	
(8)	Changes observed in absorption spectra during irradiation of	6
	Ir3-embedded vesicles only with NiP or AscNa	
(9)	Reference	6



Fig. S1 Synthetic scheme and ¹H NMR spectrum of Ir3 in CDCl₃



Fig. S2 UV-vis absorption and phosphorescence spectra of **Ir2** and **Ir3**. (a) UV-vis absorption spectra in CH₃CN. (b) Phosphorescence spectra in CH₃CN (excited at 478 nm). (c) UV-vis absorption spectra in CH₃CN–pH 4.5 acetate/acetic acid buffer solution (1:1 v/v). (d) Phosphorescence spectra in CH₃CN–pH 4.5 acetate/acetic acid buffer solution (1:1 v/v) (excited at 477–478 nm).

Tabl	le S	51	Phos	phorescent	prop	perties	of l	[r2	and	Ir3	at 77	\mathbf{K}^{a}
------	------	----	------	------------	------	---------	------	-----	-----	-----	-------	------------------

Complex	$\lambda_{ m em}/{ m nm}^b$	$arPhi^c$	
Ir2	578, 630, 690	0.92	
Ir3	579, 630, 693	0.84	

^{*a*} In 2-MeTHF, excited at 481 nm. ^{*b*} Emission maxima. ^{*c*} Emission quantum yield.



Fig. S3 Cyclic voltammograms of (a) the blank solution and (b) ferrocene recorded at a 100 mV/s scan rate (in deaerated CH₃CN, *vs.* Ag/AgNO₃). In the blank solution, a weak reversible wave was observed at approximately -1.2 V most probably due to residual oxygen.¹



Fig. S4 Photographs of the DPPC vesicle solution containing Ir2 (a) before and (b) after the ultrasonic process



Fig. S5 (a) Modified Stern–Volmer plot for quenching of the phosphorescence from **Ir2** (excitation: 477 nm) with AscNa in vesicle solution ([**Ir2**] = 11 μ M). (b) Modified Stern–Volmer plot for quenching of the phosphorescence from **Ir3** (excitation: 475 nm) with AscNa in vesicle solution ([**Ir2**] = 11 μ M). The least-squares analyses of the plots yield quenching constants, K_{sv} , of 409 M⁻¹ and 7280 M⁻¹, respectively.



Fig. S6 Changes observed in the absorption spectra during the photochemical H₂ evolution using (a) the **Ir2**-embedded DPPC vesicles and (b) the **Ir3**-embedded DPPC vesicles. Conditions: AscNa (100 mM)/DPPC vesicles with **Ir2** or **Ir3** (11 μ M)/**NiP** (0.05 mM), $\lambda > 440$ nm.



Fig. S7 Changes observed in the absorption spectra during the irradiation (> 440 nm) of the **Ir3**-embedded vesicles only with (a) AscNa (100 mM) or (b) **NiP** (0.05 mM).

Reference

1. N. Elgrishi, K. J. Rountree, B. D. McCarthy, E. S. Rountree, T. T. Eisenhart and J. L. Dempsey, *J. Chem. Educ.*, 2018, **95**, 197.