

**Supplementary Information for**

**The 12-ethynylmonocarba-*closو*-dodecaborate anion as a versatile ligand for Cu(I) alkyne and heterobimetallic Cu(I)/M(II) (M = Pd, Pt) alkynide complexes**

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## I General Information

### I. a) Chemicals, reaction conditions and characterization

If not otherwise specified, reagents and organic solvents were commercially available and used without further purification. Acetone-*d*<sub>6</sub> and CD<sub>2</sub>Cl<sub>2</sub> were purchased from Cambridge Isotope Laboratories and filtered through Al<sub>2</sub>O<sub>3</sub> prior to use. [Cs][12-C≡CH-CB<sub>11</sub>H<sub>11</sub>] was prepared according to the literature.<sup>[1a]</sup>

Glassware for air-sensitive reactions was dried at 200 °C for at least 6 h and allowed to cool in a vacuum.

Air-sensitive reactions were carried out in a glovebox under a nitrogen atmosphere with O<sub>2</sub>, H<sub>2</sub>O < 1 ppm.

NMR spectra were recorded on a Bruker AVANCE III 500 spectrometer (<sup>1</sup>H NMR 500.13 MHz, <sup>13</sup>C NMR 125.77 MHz, <sup>11</sup>B NMR 160.46 MHz) or a Bruker AVANCE III 400 spectrometer (<sup>1</sup>H NMR 400.13 MHz, <sup>13</sup>C NMR 100.62 MHz, <sup>11</sup>B NMR 128.38 MHz) at 23 °C. Data are reported as follows: Chemical shift in ppm, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, dd = doublet of doublet, etc.), coupling constant *J* in Hz, integration, and (where applicable) interpretation. Signals were referenced against solvent peaks (residual CHD<sub>2</sub>C(O)CD<sub>3</sub> = 2.05 ppm, residual CHDCl<sub>2</sub> = 5.32 ppm, residual CHD<sub>2</sub>S(O)CD<sub>3</sub> = 2.50 ppm, <sup>13</sup>C{<sup>1</sup>H}: CD<sub>3</sub>C(O)CD<sub>3</sub> = 29.84 ppm, CD<sub>2</sub>Cl<sub>2</sub> = 53.84 ppm). <sup>11</sup>B and <sup>11</sup>B{<sup>1</sup>H} NMR spectra were calibrated against external BF<sub>3</sub>\*Et<sub>2</sub>O = 0 ppm (BF<sub>3</sub>\*Et<sub>2</sub>O capillary in C<sub>6</sub>D<sub>6</sub>). <sup>31</sup>P{<sup>1</sup>H} NMR spectra were calibrated against external 85% H<sub>3</sub>PO<sub>4</sub> = 0 ppm.

IR spectra were recorded on a Nicolet iS10 FT-IR spectrometer as KBr pellets and are reported as wavenumbers (ν, cm<sup>-1</sup>). Cluster B–H and alkynyl C–H stretchings were assigned according to reference [1b].

Elemental analysis was carried out by the analytical facilities of the Department of Chemistry of Zhejiang University.

Single-crystal X-ray diffraction studies were performed on an Oxford Diffraction Gemini A Ultra diffractometer equipped with 135 mm Atlas CCD detector and using Mo or Cu X-ray sources or on a Bruker D8 Venture instrument with Ga wavelength.

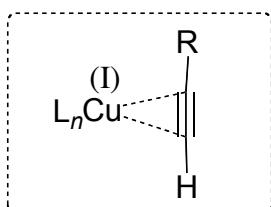
## II. b) Literature search on X-ray crystal structures involving terminal alkynes coordinated to Cu(I)

*General considerations:*

The latest version of The Cambridge Structural Database CSD ConQuest software (CSD System; version 2.0.0) was used for the X-ray structure search in the CCDC database.

URL: <https://www.ccdc.cam.ac.uk/solutions/csd-system>

The search was carried out for X-ray crystal structures of complexes including the motif Cu(I)( $\eta^2$ -RCCH)] (multiple copper centers in one complex allowed, but with no Cu-acetylene-Cu bridging; R = H or R = C)



*Results:*

**R = H.** For coordination compounds of Cu(I) and unsubstituted acetylene, only three hits were found that correspond to non-bridging complexes (*i.e.*, with one side-on coordination of the acetylene to Cu(I)); molecular drawings corresponding to each X-ray structure along with the corresponding CSD refcodes and the CCDC numbers are provided in **Table S1**.

**Table S1.** Molecular representations of reported X-ray crystal structures of Cu(I)-acetylene complexes.

Structure	CSD refcode	CCDC number
	CADZEM	1118977

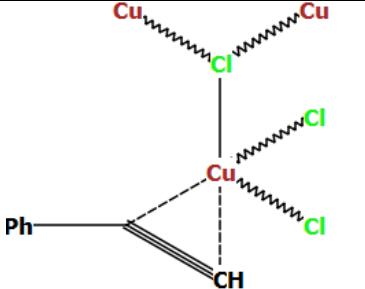
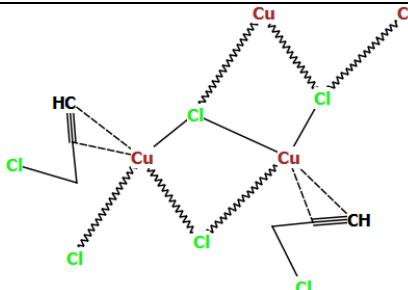
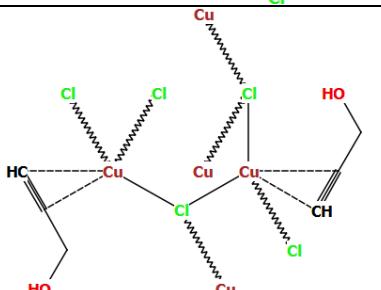
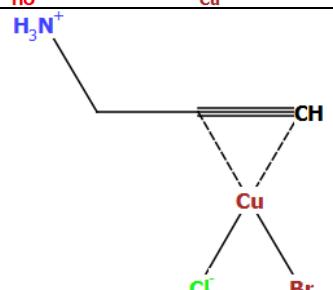
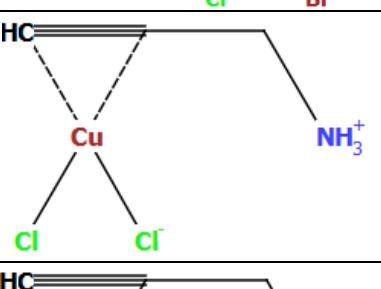
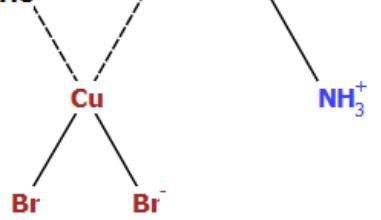
	CADZEM10	1118978
	YAFSON	--

**R = C.** For coordination compounds of copper and terminal alkyne ligands, 25 hits were found that correspond to complexes with non-bridging terminal alkynes (*i.e.*, with one side-on coordination of the alkyne to Cu(I)). Molecular drawings corresponding to each X-ray structure along with the corresponding CSD refcodes and the CCDC numbers are provided in **Table S2**.

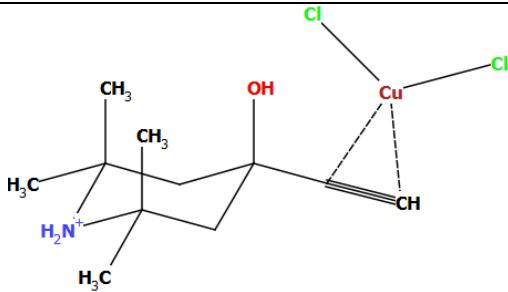
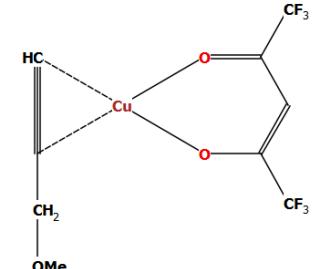
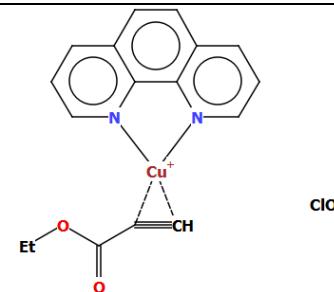
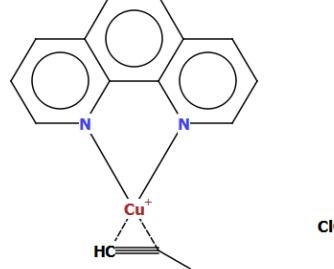
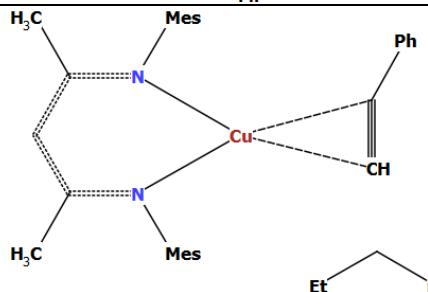
**Table S2.** Molecular representations of reported X-ray crystal structures of Cu(I)–(terminal alkyne) complexes.

Structure	CSD refcode	CCDC number
	CAQPAN	849107
	CAQPER	849108

	CAQPIV	867111
	HIPFUJ	808672
	HIPGAQ	808671
	IVUCUX	210585
	JIMSUW	1834245
	JIMTAD	1834246

	LEYGOL	1206364
	LEYGUR	1206365
	LEYHAY	1206366
	MIFMEU	--
	NOSZEA	176124
	NOSZIE	176125

	OFANEO	138842
	PETVOA	610419
	TASNOR	250204
	VEQXIA	818801
	VORTIF	--
	WONMUH	141001

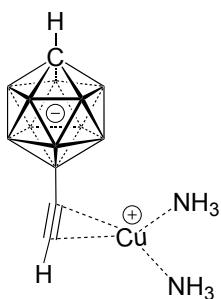
	XEMYAR	753618
	XUMMOH	190274
	YAFSIH	--
	YAFSUT	--
	YECWIN	279255

Four additional hits were found for X-ray structures of multi-metallic and bridging terminal alkyne, CSD refcodes/CCDC numbers: LITVIT/144077; GOXKOT; GOKXEJ/129611; CIKRIX/-- (structures not shown).

## II Experimental Section

### II. a) Preparation of Compounds 2–4

#### Compound 2:



A microwave tube (10 mL), equipped a magnetic stir bar, was charged with  $[\text{Cs}][\text{CB}_{11}\text{H}_{11}-12-\text{C}\equiv\text{CH}]$  (200 mg, 0.667 mmol, 1 equiv) and anhydrous EtOH (1 mL). The clear solution was stirred at 0 °C, and an aqueous solution of  $[\text{Cu}(\text{NH}_3)_n]\text{OH}$  (0.11 M in water, 6 mL, 0.66 mmol, 1.0 equiv, prepared from CuI + aqueous NH<sub>3</sub> solution) was added dropwise over 3 min using a syringe. The resulting mixture was stirred for 1 h at 0 °C. A white precipitate formed, which was separated from the supernatant by centrifugation and washed by three subsequent cycles of suspending it in water (5 mL) followed by centrifugation. Removal of water (containing also CsI) with a syringe and drying in a vacuum at 25 °C for 2 days gave compound **2** (140 mg, 80%).

Compound **2** was kept under nitrogen. Exposure to air over 2–3 days did not lead to significant decomposition; however, prolonged exposure to air (weeks) lead to oxidation as indicated by a color change to green.

$^1\text{H}\{^{11}\text{B}\}$  NMR (500 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 3.62 (broad signal, 1H, C≡CH), 3.40–2.61 (very broad signal, 6 H, NH), 2.28 (broad signal, 1H, cage CH), 1.69 (broad overlapping signals, 10H, BH).

$^{11}\text{B}$  NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -6.99 (s, B-12), -12.55 (d, *J* = 137.70 Hz, 5B), -16.66 (d, *J* = 151.61 Hz, 5B).

$^{11}\text{B}\{^1\text{H}\}$  NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.00 (1B), -12.55 (5B), -16.66 (5B).

$^{13}\text{C}\{^1\text{H}\}$  NMR (150 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 49.71 (cage C), 84.80 (≡CH). The B-C signal could not be detected unambiguously.

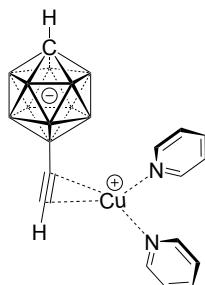
Elemental analysis: Calc. for  $\text{C}_3\text{H}_{18}\text{B}_{11}\text{CuN}_2$ : C 13.62%, H 6.86%, N 10.59; found: C 13.42%, H 7.02%, N 10.47%.

IR (KBr):  $\nu$  (cm<sup>-1</sup>) 3374, 3285, 3200 (C–H stretching of C≡C–H), 3055, 2564 (B–H), 1901 (C≡C), 1604, 1214, 1050, 1006, 662.

### General procedure for Compounds 3 and 4

In a glovebox, a glass vial (4 mL), equipped a magnetic stir bar, was charged with **2** (15 mg, 0.057 mmol, 1 equiv) pyridine ligand (0.125 mmol, 2.2 equiv) or phosphine ligand (0.228 mmol, 4.0 equiv). Anhydrous EtOH (0.5 mL) was added *via* a syringe, and the resulting mixture was stirred vigorously for 3 h at 25 °C. The resulting white precipitate was collected by filtration through a glass frit and dissolved in acetone (0.5 mL) in a 4 mL glass vial to give a clear solution, followed by addition of anhydrous ethanol (0.5 mL). Slow evaporation at 25 °C afforded colorless crystals within 2 days. The crystals were collected by filtration and dried in a vacuum at 25 °C for 24 h to give compounds **3a–d** and **4a–d**.

### **Cu(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)(Py)<sub>2</sub>** (**3a**)



**3a:** Prepared following the general procedure, using **2** and pyridine, compound **3a** was obtained as a colorless solid (10 mg, 45%).

<sup>1</sup>H{<sup>11</sup>B} NMR (500 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 8.72 (broad signal, 4H, Ar-H), 8.07 (d, 2H, Ar-H), 7.63 (broad signal, 4H, Ar-H), 4.37 (s, 1H, C≡CH), 2.24 (broad signal, 1H, cage CH), 1.59 (overlapping broad signals, 10H, BH).

<sup>11</sup>B NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.30 (s, B-12), -12.55 (d, *J* = 138.53 Hz, 5B), -16.71 (d, *J* = 152.64 Hz, 5B).

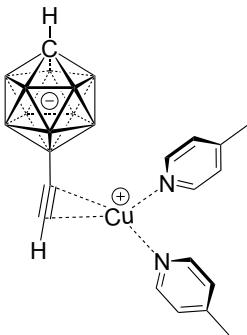
<sup>11</sup>B {<sup>1</sup>H} NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.28(1B), -12.55 (5B), -16.71 (5B).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 150.90, 139.68, 126.84 (aromatic signals), 85.08 (≡CH), 50.15 (cage C). The B-C signal could not be detected unambiguously.

High-resolution (+)-EI-MS: *m/z* calcd for [C<sub>13</sub>H<sub>22</sub>B<sub>11</sub>CuN<sub>2</sub>]<sup>+</sup>: 388.2175. Found: 388.2164.

IR (KBr):  $\nu$  (cm<sup>-1</sup>) 3176 (C–H stretching of C≡C–H), 3069, 2566 (B–H), 2538 (B–H), 1892 (C≡C), 1605, 1488, 1446, 1046, 754, 693.

**Cu(C<sub>B</sub><sub>11</sub>H<sub>11</sub>-12-C≡CH)(4-CH<sub>3</sub>Py)<sub>2</sub> (3b):**



**3b:** Prepared following the general procedure, using **2** and 4-methylpyridine, compound **3b** was obtained as a colorless solid (12 mg, 50%).

<sup>1</sup>H{<sup>11</sup>B} NMR (500 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 8.49 (broad signal, 4H, Ar-H), 7.43 (broad signal, 4H, Ar-H), 4.26 (s, 1H, C≡CH), 2.45 (s, 6H, Ar-CH<sub>3</sub>), 2.24 (broad signal, 1H, cage CH), 1.61 (overlapping broad signals, 10H, BH).

<sup>11</sup>B NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.25 (s, B-12), -12.53 (d, *J* = 138.97 Hz, 5B), -16.72 (d, *J* = 151.35 Hz, 5B).

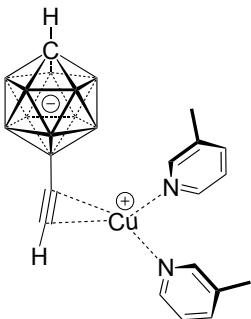
<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.24(1B), -12.53 (5B), -16.72 (5B).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 151.87, 150.36, 126.89 (aromatic signals), 85.22 (≡CH), 50.03 (cage C), 21.17 (Ar-CH<sub>3</sub>), The B-C signal could not be detected unambiguously.

High-resolution (+)-EI-MS: *m/z* calcd for [C<sub>15</sub>H<sub>26</sub>B<sub>11</sub>CuN<sub>2</sub>]<sup>+</sup>: 416.2488. Found: 416.2472.

IR (KBr): *v* (cm<sup>-1</sup>) 3204 (C–H stretching of C≡C–H), 3062, 2559 (B–H), 1901 (C≡C), 1623, 1430, 1047, 804, 722.

**Cu(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)(3-CH<sub>3</sub>Py)<sub>2</sub> (3c):**



**3c:** Prepared following the general procedure, using **2** and 3-methylpyridine, compound **3c** was obtained as a colorless solid (12 mg, 50%).

<sup>1</sup>H{<sup>11</sup>B} NMR (500 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 23 °C): δ 8.30 (s, 2H, Ar-H), 8.20 (broad signal, coupling not resolved, 2H, Ar-H), 7.75 (m, 2H, Ar-H), 7.39 (m, 2H, Ar-H), 4.17 (s, 1H, C≡CH), 2.38 (s, 6 H, Ar-CH<sub>3</sub>), 2.28 (broad signal, 1H, cage CH), 1.60-1.50 (overlapping broad signals, 10H, BH).

<sup>11</sup>B NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.25 (s, B-12), -12.49 (d, *J* = 138.14 Hz, 5B), -16.69 (d, *J* = 149.46 Hz, 5B).

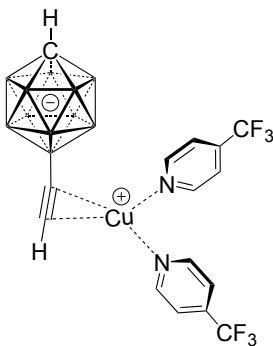
<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.24 (1B), -12.49 (5B), -16.69 (5B).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 151.25, 148.11, 139.69, 135.90, 125.35 (aromatic signals), 85.30 (≡CH), 49.92 (cage C), 18.34 (Ar-CH<sub>3</sub>). The B-C signal could not be detected unambiguously.

High-resolution (+)-EI-MS: *m/z* calcd for [C<sub>15</sub>H<sub>26</sub>B<sub>11</sub>CuN<sub>2</sub>]<sup>+</sup>: 416.2488. Found: 416.2473.

IR (KBr):  $\nu$  (cm<sup>-1</sup>) 3192 (C–H stretching of C≡C–H), 3073, 2918, 2560 (B–H), 2527 (B–H), 1909 (C≡C), 1609, 1484, 1049, 791, 701, 655.

**Cu(C<sub>B</sub><sub>11</sub>H<sub>11</sub>-12-C≡CH)(4-CF<sub>3</sub>Py)<sub>2</sub> (3d):**



**3d:** Prepared following the general procedure, using **2** and 4-trifluoromethylpyridine, compound **3d** was obtained as a colorless solid (14 mg, 53%).

<sup>1</sup>H{<sup>11</sup>B} NMR (500 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 9.05 (braod signal, coupling not resolved, 4H, Ar-H), 7.94 (d, 4H, *J* = 4.5 Hz, Ar-H), 4.34 (s, 1H, C≡CH), 2.27 (broad signal, 1H, cage CH), 1.61, 1.59 (overlapping broad signals, 10H, BH).

<sup>11</sup>B NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.62 (s, B-12), -12.57 (d, *J* = 138.43 Hz, 5B), -16.64 (d, *J* = 152.47 Hz, 5B).

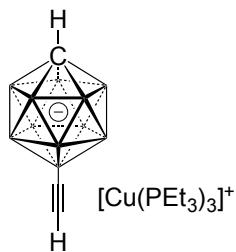
<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.60 (1B), -12.57 (5B), -16.64 (5B).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 152.59, 139.26 (weak, broad), 121.81 (broad) (aromatic signals), 50.34 (cage C). The CF<sub>3</sub>, B-C and ≡CH signals could not be detected unambiguously.

High-resolution (+)-EI-MS: *m/z* calcd for [C<sub>15</sub>H<sub>20</sub>B<sub>11</sub>CuF<sub>6</sub>N<sub>2</sub>]<sup>+</sup>: 524.1923. Found: 524.1909.

IR (KBr): *v* (cm<sup>-1</sup>) 3184 (C–H stretching of C≡C–H), 3053, 2562 (B–H), 2542 (B–H), 1895 (C≡C), 1617, 1479, 1330, 1134, 1080, 711.

**Cu(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)(PEt<sub>3</sub>)<sub>3</sub> (4a):**



**4a:** Prepared following the general procedure, using **2** and 4 equiv of PEt<sub>3</sub>, compound **4a** was obtained as a colorless solid (32 mg, 95%).

<sup>1</sup>H{<sup>11</sup>B} NMR (400 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 2.15 (broad signal, 1H, cage CH), 1.87 (m, 18H, CH<sub>2</sub> of PEt<sub>3</sub>), 1.85 (s, 1H, C≡CH), 1.73 (broad signal, 5H, BH), 1.64 (broad signal, 5H, BH), 1.16 (broad signal, 27H, CH<sub>3</sub> of PEt<sub>3</sub>).

<sup>11</sup>B NMR (128 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.43 (s, B-12), -12.23 (d, *J* = 140.12 Hz, 5B), -16.73 (d, *J* = 151.45 Hz, 5B).

<sup>11</sup>B{<sup>1</sup>H} NMR (128 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.41 (1B), -12.23 (5B), -16.73 (5B).

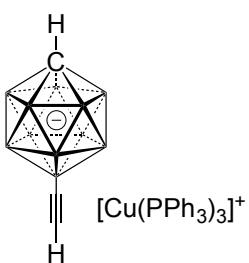
<sup>31</sup>P{<sup>1</sup>H} NMR (162 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -9.53.

<sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, acetone-*d*<sub>6</sub>, 23°C): δ 81.45 (≡CH), 48.76 (cage C), 16.85, 8.86. The B-C signal could not be detected unambiguously.

High-resolution (+)-ESI-MS: *m/z* calcd for [C<sub>18</sub>H<sub>45</sub>CuP<sub>3</sub>]<sup>+</sup>: 417.2030. Found: 417.2019.

IR (KBr):  $\nu$  (cm<sup>-1</sup>) 3292 (C–H stretching of C≡C–H), 2965, 2933, 2908, 2873, 2556 (B–H), 1935, 1620, 1450, 1052, 1035, 758, 719.

**Cu(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)(PPh<sub>3</sub>)<sub>3</sub> (4b):**



**4d:** Prepared following the general procedure, using **2** and 4 equiv of PPh<sub>3</sub>, compound **4d** was obtained as a colorless solid (50 mg, 86%).

<sup>1</sup>H{<sup>11</sup>B} NMR (500 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 7.47 (m, 9H, Ar-H), 7.29 (m, 18H, Ar-H), 7.18 (m, 18H, Ar-H), 2.16 (broad signal, 1H, cage CH), 1.85 (s, 1H, C≡CH), 1.74 (broad signal, 5H, BH), 1.64 (broad signal, 5H, BH).

<sup>11</sup>B NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.58 (s, B-12), -12.36 (d, *J* = 138.45 Hz, 5B), -16.85 (d, *J* = 148.64 Hz, 5B).

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.64 (1B), -12.36 (5B), -16.85 (5B).

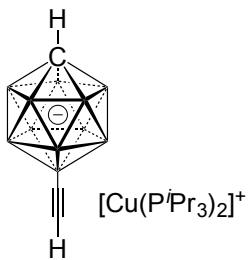
<sup>31</sup>P{<sup>1</sup>H} NMR (202 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 0.16.

<sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, acetone-*d*<sub>6</sub>, 23°C): δ 134.46 (d, *J*=13.90 Hz), 132.01 (d, *J* =30.74 Hz), 131.61, 129.96 (d, *J* = 7.5 Hz), 81.25 (≡CH), 48.78 (cage C). The B-C signal could not be detected unambiguously.

High-resolution (+)-ESI-MS: *m/z* calcd for [C<sub>54</sub>H<sub>45</sub>CuP<sub>3</sub>]<sup>+</sup>: 849.2030. Found: 849.2020.

IR (KBr):  $\nu$  (cm<sup>-1</sup>) 3269 (C–H stretching of C≡C–H), 3244, 3055, 2558 (B–H), 1648, 1435, 1095, 1051, 1005, 744, 695, 517.

**(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)Cu(PiPr<sub>3</sub>)<sub>2</sub> (4c):**



**4c:** Prepared following the general procedure, using **2** and 4 equiv of PiPr<sub>3</sub>, compound **4c** was obtained as a colorless solid (30 mg, 95%).

<sup>1</sup>H{<sup>11</sup>B} NMR (500 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 2.32 (m, 6H, CH of PiPr<sub>3</sub>), 2.16 (broad signal, 1H, cage CH), 1.86 (s, 1H, C≡CH), 1.73 (broad signal, 5H, BH), 1.64 (broad signal, 5H, BH) 1.33 (m, 36H, CH<sub>3</sub> of PiPr<sub>3</sub>).

<sup>11</sup>B NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.42 (s, B-12), -12.24 (d, *J* = 139.18 Hz, 5B), -16.73 (d, *J* = 151.46 Hz, 5B).

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.47 (1B), -12.24 (5B), -16.73 (5B).

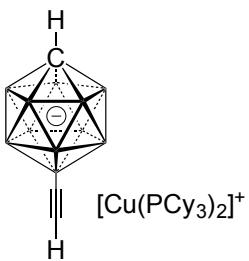
<sup>31</sup>P{<sup>1</sup>H} NMR (202 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 27.53 (only one signal was detected).

<sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 81.02 (≡CH), 48.76 (cage C), 23.14, 20.70. The B-C signal could not be detected unambiguously.

High-resolution (+)-ESI-MS: *m/z* calcd for [C<sub>18</sub>H<sub>42</sub>CuP<sub>2</sub>]<sup>+</sup>: 383.2058. Found: 383.2041.

IR (KBr): *v* (cm<sup>-1</sup>) 3430, 3283 (C–H stretching of C≡C–H), 2961, 2561 (B–H), 2142, 1463, 1387, 1051, 1006, 883.

**Cu(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)(PCy<sub>3</sub>)<sub>2</sub> (4d)**



**4d:** Prepared following the general procedure, using **2** and 4 equiv of PCy<sub>3</sub>, compound **4d** was obtained as a colorless solid (40 mg, 83%).

<sup>1</sup>H{<sup>11</sup>B} NMR (500 MHz, acetone-*d*<sub>6</sub>, 23 °C): 2.15 (broad signal, 1H, cage CH), 2.13-1.93 (overlapping m, 18H, cyclohexyl signals), 1.89-1.78 (m, 13H, cyclohexyl signals overlapping with C≡CH), 1.77-1.68 (m, 11H, cyclohexyl signals overlapping with 5 BH), 1.64 (broad signal, 5H, BH), 1.57-1.44 (m, 12 H, cyclohexyl signals), 1.44-1.21 (m, 18H, cyclohexyl signals).

<sup>11</sup>B NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.54 (s, B-12), -12.35 (d, *J* = 138.33 Hz, 5B), -16.83 (d, *J* = 150.02 Hz, 5B).

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.58 (1B), -12.35 (5B), -16.85 (5B).

<sup>31</sup>P{<sup>1</sup>H} NMR (202 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 13.17.

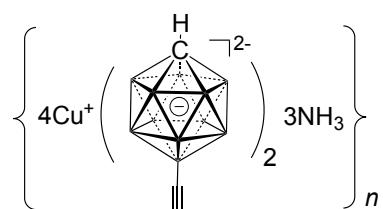
<sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, acetone-*d*<sub>6</sub>, 23°C): δ 81.39 (≡CH), 48.74 (cage C), 33.34 (cyclohexyl CH), 31.39 (cyclohexyl CH<sub>2</sub>), 28.02 (cyclohexyl CH<sub>2</sub>), 26.82 (cyclohexyl CH<sub>2</sub>). The B-C signal could not be detected unambiguously. The four cyclohexyl signals were broad and showed a relative integration of 1:2:2:1; the coupling to <sup>31</sup>P was not resolved. In a reference measurement of PCy<sub>3</sub> in acetone-*d*<sub>6</sub>, the <sup>13</sup>C{<sup>1</sup>H} resonances appeared in a 1:2:2:1 ratio at δ 31.47 (*d*, *J* = 17.8 Hz), 32.01 (*d*, *J* = 12.6 Hz), 28.32 (*d*, *J* = 9.2 Hz), 27.30 (s).

High-resolution (+)-ESI-MS: *m/z* calcd for [C<sub>36</sub>H<sub>66</sub>CuP<sub>2</sub>]<sup>+</sup>: 623.3936. Found: 623.3923.

IR (KBr):  $\nu$  (cm<sup>-1</sup>) 3305 (C–H stretching of C≡C–H), 2928, 2851, 2554 (B–H), 1447, 1052, 1005.

## **II. b) Preparation of Compounds 5–7**

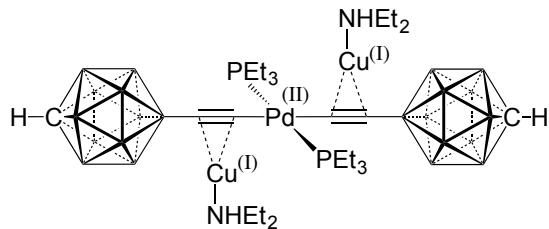
## Compound 5:



A microwave tube (25 mL), equipped with a magnetic stir bar, was charged with **2** (150 mg, 0.566 mmol, 1 equiv) and anhydrous EtOH (1 mL). During vigorous stirring at 25 °C, an aqueous solution of [Cu(NH<sub>3</sub>)<sub>2</sub>]OH (0.18 M in water, 3.8 mL, 0.68 mmol, 1.2 equiv, prepared from CuI (1.2 equiv) in aqueous NH<sub>3</sub>) was added. The resulting mixture was stirred for 2 h. The yellow precipitate was separated from the supernatant by centrifugation and three subsequent cycles of suspending in water (5 mL) followed by centrifugation. Removal of water with a syringe and drying in a vacuum at 25 °C for 2 days gave compound **5** as a yellow solid (170 mg, 94%).

The analytical data matched with the ones reported.<sup>[1c]</sup>

**Compound 6a:**



A vial (4 mL), equipped a magnetic stir bar, was charged with **5** (40 mg, 0.078 mmol, 1 equiv), *trans*-dichlorobis(triethylphosphine)palladium(II) (46.4 mg, 0.078 mmol, 1.0 equiv) and anhydrous HNEt<sub>2</sub> (2 mL). After stirring for 1 h at 25 °C, the white precipitate that formed was collected by filtration through a glass frit and dried in a vacuum at 25 °C to give Compound **6a** (66.8 mg, 90%).

This product exhibits low solubility in acetone, acetonitrile, dichloromethane, dimethyl formamide, dimethyl sulfoxide and tetrahydrofuran. We were not able to obtain satisfactory <sup>13</sup>C{<sup>1</sup>H} and <sup>31</sup>P{<sup>1</sup>H} NMR data.

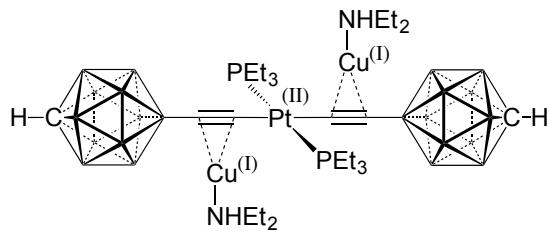
<sup>1</sup>H{<sup>11</sup>B} NMR (400 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ 3.00 (m, 8H, CH<sub>2</sub> of HNEt<sub>2</sub>), 2.27 (broad signal, 2H, cage CH), 2.02 (m, 12H), 1.76 (broad signal, 10H, BH), 1.69 (broad signal, 10H, BH) 1.37 (m, 12H), 1.18 (m, 18H).

<sup>11</sup>B NMR (128 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.45 (s, B-12), -12.41 (d, J = 148.55 Hz, 5B), -16.36 (d, J = 142.35 Hz, 5B).

<sup>11</sup>B{<sup>1</sup>H} NMR (128 MHz, acetone-*d*<sub>6</sub>, 23 °C): δ -7.49 (1B), -12.41 (5B), -16.36 (5B).

Elemental analysis: Calc. for C<sub>26</sub>H<sub>74</sub>B<sub>22</sub>Cu<sub>2</sub>N<sub>2</sub>P<sub>2</sub>Pd: C 32.94%, H 7.87%, N 2.95%; found: C 32.86%, H 7.86%, N 2.79%.

**Compound 6b:**



A vial (4 mL), equipped a magnetic stir bar, was charged with **5** (40 mg, 0.078 mmol, 1 equiv), *trans*-diiodobis(triethylphosphine)platinum(II) (53 mg, 0.078 mmol, 1.0 equiv) and anhydrous HNEt<sub>2</sub> (2 mL). After stirring for 1 h at 25°C, the white precipitate that formed was collected by filtration through a glass frit and dried in a vacuum at 25 °C to give compound **6b** (74.3 mg, 92%).

This product exhibits low solubility in acetone, acetonitrile, dichloromethane, dimethyl formamide, dimethyl sulfoxide and tetrahydrofuran. We were not able to obtain satisfactory <sup>13</sup>C{<sup>1</sup>H} and <sup>31</sup>P{<sup>1</sup>H} NMR data.

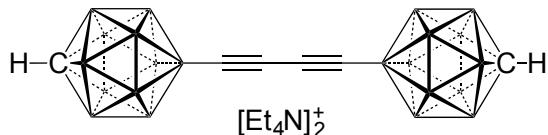
<sup>1</sup>H{<sup>11</sup>B} NMR (400 MHz, DMSO-*d*<sub>6</sub>, 23 °C): δ 2.72 (m, 8H), 2.36 (broad signal, 2H, cage CH), 1.96 (m, 12H), 1.62 (broad signal, 10H, BH), 1.55 (broad signal, 10H, BH), 1.17 (m, 12H), 1.05 (m, 18H).

<sup>11</sup>B NMR (128 MHz, DMSO-*d*<sub>6</sub>, 23 °C): δ -6.27 (s, B-12), -12.44 (d, *J* = 147.75 Hz, 5B), -16.73 (d, *J* = 144.55 Hz, 5B).

<sup>11</sup>B{<sup>1</sup>H} NMR (128 MHz, DMSO-*d*<sub>6</sub>, 23 °C): δ -6.31 (1B), -12.44 (5B), -16.73 (5B).

Elemental analysis: Calc. for C<sub>26</sub>H<sub>74</sub>B<sub>22</sub>Cu<sub>2</sub>N<sub>2</sub>P<sub>2</sub>Pt: C 30.12%, H 7.19%, N 2.70%; found: C 29.58%, H 6.96%, N 2.16%.

**Compound 7:**



A glass vial (4 mL), equipped a magnetic stir bar, was charged with **5** (40 mg, 0.078 mmol, 1 equiv). Anhydrous ethanol (1.0 mL) was added *via* a syringe, and the resulting mixture was stirred vigorously open to air for 2 h at 25 °C. Water (5.0 mL) was slowly added, and ethanol was removed using a rotary evaporator. HCl solution (1 M in H<sub>2</sub>O, 5 mL, 5 mmol) was added, and the mixture was extracted with diethyl ether (3 x 15 mL). The combined organic layers were dried over Cs<sub>2</sub>CO<sub>3</sub>, filtered into a 100 mL one-necked round-bottom flask. To the flask water (8 mL) was added. Diethyl ether was removed using a rotary evaporator, and the turbid water layer was filtered into a 25 mL flask. [Et<sub>4</sub>N]<sup>+</sup>Br<sup>-</sup> (65 mg, 2.21 equiv) was added to the filtrate, and the resulting white solid was collected by filtration through a glass frit and dried in a vacuum at 65 °C for 12 h to give compound **7** (43.9 mg, 95%).

<sup>1</sup>H{<sup>11</sup>B} NMR (400 MHz, CD<sub>3</sub>CN, 23 °C): δ 3.16 (q, *J* = 7.26 Hz, 16H, CH<sub>2</sub> of cation), 2.28 (s, 2H, cage CH), 1.57 (overlapping broad signals, 20H, BH), 1.21 (m, 24H, CH<sub>3</sub> of cation).

<sup>11</sup>B NMR (128 MHz, CD<sub>3</sub>CN, 23 °C): δ -7.82 (s, B-12), -12.45 (d, *J* = 139.34 Hz, 5B), -16.62 (d, *J* = 150.71 Hz, 5B).

<sup>11</sup>B{<sup>1</sup>H} NMR (128 MHz, CD<sub>3</sub>CN, 23 °C): δ -7.81 (2B), -12.45 (10B), -16.62 (10B).

<sup>13</sup>C{<sup>1</sup>H} NMR (101 MHz, CD<sub>3</sub>CN, 23 °C): δ 89.92 (broad q, B-C), 80.03 (broad signal, ≡CH), 53.06 (CH<sub>2</sub> of cation), 49.57 (cage C), 7.70 (CH<sub>3</sub> of cation).

High-resolution (-)-ESI-MS: *m/z* calcd for [C<sub>6</sub>H<sub>22</sub>B<sub>22</sub>]<sup>2-</sup>: 166.1957. Found: 166.1993.

### III X-ray Crystallography

#### General remarks

CCDC1910714–1910719, 1910721 and 1910723–1910725 contain the supplementary crystallographic data for this publication. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre *via* [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

The structures were solved with the dual-space algorithm using *SHELXT*<sup>[2]</sup> and were refined by full-matrix least-squares methods on  $F^2$  with *SHELXL-2014*<sup>[3]</sup> using the GUI *OLEX2*<sup>[4]</sup>. The graphical output was produced with the help of the program *Mercury*<sup>[5]</sup>.

The terminal hydrogen atoms of the copper-bound alkynes were always found in the difference electron density map and refined at a 20% bigger isotropic value than the corresponding adjacent mother carbon atom. If needed, the C–H distance was restrained to 0.96 Å. Hydrogen atoms of the carboranes bound to a carbon atom were refined with the AFIX 154 command, allowing the C–H distance to vary. If the distances became too short, they were fixed at a minimal distance of 0.96 Å.

**Table S3.** Summary of X-ray diffraction data.

Compound	2a	3a	3b	3c
Empirical formula	C <sub>9</sub> H <sub>26</sub> B <sub>11</sub> CuN <sub>2</sub>	C <sub>13</sub> H <sub>22</sub> B <sub>11</sub> CuN <sub>2</sub>	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> CuN <sub>2</sub>	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> CuN <sub>2</sub>
M (g/mol)	344.78	388.77	416.83	416.83
Wavelength(Å)	1.34139	0.71073	0.71073	0.71073
T (K)	170	150	296	293
Crystal system	monoclinic	triclinic	monoclinic	triclinic
Space group	C2/c	P $\bar{1}$	P2 <sub>1</sub> /c	P $\bar{1}$
a (Å)	27.1076(7)	8.2894(7)	12.8275(7)	8.3718(8)
b (Å)	7.1263(2)	11.3934(10)	10.7515(5)	11.5149(11)
c (Å)	19.4860(5)	11.7893(10)	17.3597(9)	13.4954(9)
$\alpha$ (°)	90	71.393(7)	90.00	102.473(7)
$\beta$ (°)	102.529(2)	79.489(7)	108.206(2)	106.926(7)
$\gamma$ (°)	90	70.380(8)	90.00	107.523(9)
Volume (Å <sup>3</sup> )	3674.61(17)	990.43(16)	2274.3(2)	1119.18(18)
Z	8	2	4	2
$D_{\text{calc}}$ (g/cm <sup>3</sup> )	1.246	1.304	1.217	1.237
$\mu$ (mm <sup>-1</sup> )	6.329	1.102	0.964	0.979
F (000)	1424	396	856	428
N (coll. refl.)	3494	3596	4462	4083
N (parameters/restraints)	227/0	248/0	268/0	268/0
R1	0.0485	0.0435	0.0707	0.0468
wR2	0.1006	0.1032	0.1107	0.1155
GOF	1.013	1.050	1.068	1.047
CCDC	1910714	1910715	1910716	1910717

**Table S3.** (continued).

<b>Compound</b>	<b>3d</b>	<b>4b</b>	<b>4c</b>
Empirical formula	C <sub>15</sub> H <sub>20</sub> B <sub>11</sub> CuF <sub>6</sub> N <sub>2</sub>	C <sub>57</sub> H <sub>57</sub> B <sub>11</sub> CuP <sub>3</sub>	C <sub>21</sub> H <sub>54</sub> B <sub>11</sub> CuP <sub>2</sub>
<i>M</i> (g/mol)	524.78	1017.38	551.03
Wavelength(Å)	0.71073	0.71073	0.71073
<i>T</i> (K)	170	150	293
Crystal system	triclinic	triclinic	monoclinic
Space group	P $\bar{1}$	P $\bar{1}$	P2 <sub>1</sub> /c
<i>a</i> (Å)	10.0994(8)	12.5248(8)	11.9970(7)
<i>b</i> (Å)	11.2360(9)	12.9185(8)	15.5376(7)
<i>c</i> (Å)	11.7320(9)	18.8957(9)	17.7707(8)
$\alpha$ (°)	66.212(7)	106.031(5)	90
$\beta$ (°)	75.011(7)	97.416(4)	92.697(5)
$\gamma$ (°)	89.796(7)	104.435(5)	90
Volume (Å <sup>3</sup> )	1169.09(17)	2780.6(3)	3308.9(3)
<i>Z</i>	2	2	4
<i>D</i> <sub>calc</sub> (g/cm <sup>3</sup> )	1.491	1.215	1.106
$\mu$ (mm <sup>-1</sup> )	0.990	0.516	0.767
<i>F</i> (000)	524	1056	1176
<i>N</i> (coll. refl.)	4264	10143	6266
<i>N</i> (parameters/restraints)	320/1	650/90	329/0
<i>R</i> 1	0.0504	0.0613	0.0424
<i>wR</i> 2	0.1302	0.1726	0.1107
GOF	1.046	1.034	1.018
CCDC	1910718	1910719	1910721

**Table S3.** (continued).

Compound	<b>6a</b>	<b>6b</b>	<b>7</b>
Empirical formula	C <sub>32</sub> H <sub>86</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> O <sub>2</sub> P <sub>2</sub> Pd	C <sub>32</sub> H <sub>86</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> O <sub>2</sub> P <sub>2</sub> Pt	C <sub>22</sub> H <sub>62</sub> B <sub>22</sub> N <sub>2</sub>
<i>M</i> (g/mol)	1064.26	1152.97	592.55
Wavelength(Å)	0.71073	0.71073	0.71073
<i>T</i> (K)	293	293	293
Crystal system	monoclinic	monoclinic	monoclinic
Space group	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	C2/c
<i>a</i> (Å)	15.7282(9)	15.7322(6)	33.205(4)
<i>b</i> (Å)	10.9616(6)	10.9370(5)	11.124(1)
<i>c</i> (Å)	17.0109(14)	16.9807(11)	22.362(3)
$\alpha$ (°)	90	90	90
$\beta$ (°)	97.662(6)	97.802(5)	110.903(16)
$\gamma$ (°)	90	90	90
Volume (Å <sup>3</sup> )	2906.9(3)	2474.7(2)	7716.4(18)
<i>Z</i>	2	2	8
<i>D</i> <sub>calc</sub> (g/cm <sup>3</sup> )	1.216	1.323	1.020
$\mu$ (mm <sup>-1</sup> )	1.115	3.223	0.049
<i>F</i> (000)	1104	1168	2544
<i>N</i> (coll. refl.)	5307	5279	7024
<i>N</i> (parameters/restraints)	294/1	294/0	490/14
<i>R</i> 1	0.0524	0.0464	0.0904
<i>wR</i> 2	0.1381	0.1130	0.3023
GOF	1.030	1.059	1.037
CCDC	1910723	1910724	1910725

## Crystal structure of 2' (CCDC1910714)

Compound **2** (25 mg) was dissolved in acetone (0.5 mL) in a 4 mL glass vial. The resulting solution was filtered into an 18 cm long glass NMR tube and layered with hexane (1.0 mL). Colorless crystals of the composition [Cu(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)(H<sub>2</sub>N-C(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>(CH<sub>3</sub>)C=NH)<sub>2</sub>] suitable for X-ray diffraction grew within 4 d at 25 °C.

---

Bond precision: C-C = 0.0050 Å Wavelength=1.34139

Cell: a=27.1076(7) b=7.1263(2) c=19.4860(5)  
alpha=90 beta=102.529(2) gamma=90  
Temperature: 170 K

	Calculated	Reported
Volume	3674.61(17)	3674.60(17)
Space group	C 2/c	C 1 2/c 1
Hall group	-C 2yc	-C 2yc
Moiety formula	C <sub>9</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>	C <sub>9</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>
Sum formula	C <sub>9</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>	C <sub>9</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>
Mr	344.78	344.77
Dx, g cm <sup>-3</sup>	1.246	1.246
Z	8	8
μ (mm <sup>-1</sup> )	6.335	6.329
F000	1424.0	1424.0
F000'	1403.55	
h,k,lmax	32,8,23	32,8,23
Nref	3503	3494
Tmin, Tmax	0.927, 0.969	0.535, 0.751
Tmin'	0.531	

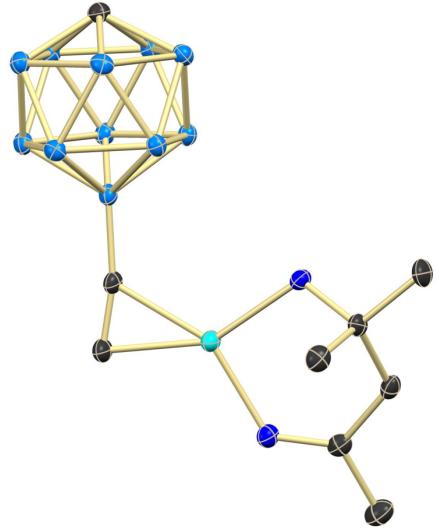
Correction method= # Reported T Limits: Tmin=0.535 Tmax=0.751  
AbsCorr = MULTI-SCAN

Data completeness= 0.997 Theta(max)= 54.938

R(reflections)= 0.0485( 2343) wR2(reflections)= 0.1006( 3494)

S = 1.013 Npar= 227

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**Figure S1.** ORTEP representation of **2'**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

### Crystal structure of 3a (CCDC1910715)

Compound **3a** (10 mg) was dissolved in anhydrous ethanol (2.0 mL) in a 4 mL glass vial to give a clear solution, then added 2 drops of acetone. Slow evaporation afforded colorless crystals of the composition [(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)Cu(Py)<sub>2</sub>] suitable for X-ray diffraction within 4 d at 25 °C.

---

Bond precision: C-C = 0.0050 Å                          Wavelength=0.71073

Cell:                a=8.2894(7)                b=11.3934(10)                c=11.7893(10)  
                      alpha=71.393(7)                beta=79.489(7)                gamma=70.380(8)  
Temperature:        150 K

	Calculated	Reported
Volume	990.43(16)	990.43(16)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C13 H22 B11 Cu N2	C13 H22 B11 Cu N2
Sum formula	C13 H22 B11 Cu N2	C13 H22 B11 Cu N2
Mr	388.79	388.77
Dx, g cm <sup>-3</sup>	1.304	1.304
Z	2	2
μ (mm <sup>-1</sup> )	1.102	1.102
F000	396.0	396.0
F000'	396.68	
h,k,lmax	9,13,14	9,13,14
Nref	3612	3596
Tmin,Tmax	0.627,0.718	0.737,1.000
Tmin'	0.577	

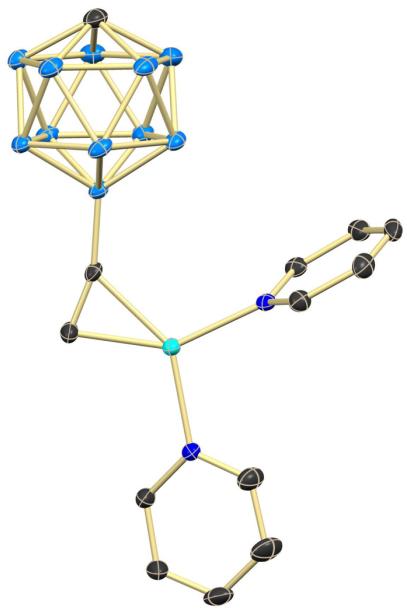
Correction method= # Reported T Limits: Tmin=0.737 Tmax=1.000  
AbsCorr = MULTI-SCAN

Data completeness= 0.996                          Theta(max)= 25.348

R(reflections)= 0.0435( 3067)                wR2(reflections)= 0.1032( 3596)

S = 1.050                                  Npar= 248

---



**Figure S2.** ORTEP representation of **3a**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

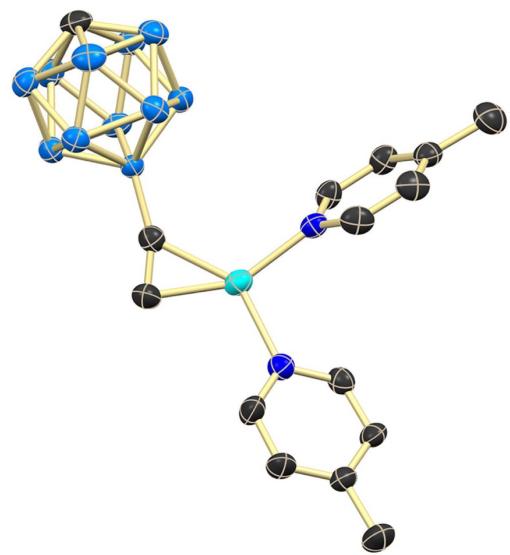
## Crystal structure of 3b (CCDC1910716)

Compound **3b** (10 mg) was dissolved in anhydrous ethanol (2.0 mL) in a 4 mL glass vial to give a clear solution. Slow evaporation afforded colorless crystals of the composition [Cu(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)(4-MePy)<sub>2</sub>] suitable for X-ray diffraction within 4 d at 25 °C.

---

Bond precision:	C-C = 0.0072 Å	Wavelength=0.71073
Cell:	a=12.8275(7)	b=10.7515(5)
	alpha=90	beta=108.206(2)
Temperature:	296 K	gamma=90
	Calculated	Reported
Volume	2274.3(2)	2274.3(2)
Space group	P 21/n	P 1 21/n 1
Hall group	-P 2yn	-P 2yn
Moiety formula	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>
Sum formula	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>
Mr	416.84	416.83
D <sub>x</sub> , g cm <sup>-3</sup>	1.217	1.217
Z	4	4
μ (mm <sup>-1</sup> )	0.964	0.964
F <sub>000</sub>	856.0	856.0
F <sub>000'</sub>	857.39	
h,k,lmax	15,13,21	15,13,21
Nref	4467	4462
Tmin,Tmax	0.636,0.714	0.634,0.719
Tmin'	0.611	
Correction method= # Reported T Limits: Tmin=0.634 Tmax=0.719		
AbsCorr = MULTI-SCAN		
Data completeness= 0.999	Theta(max)= 25.997	
R(reflections)= 0.0707( 2700)	wR2(reflections)= 0.1107( 4462)	
S = 1.068	Npar= 268	

---



**Figure S3.** ORTEP representation of **3b**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

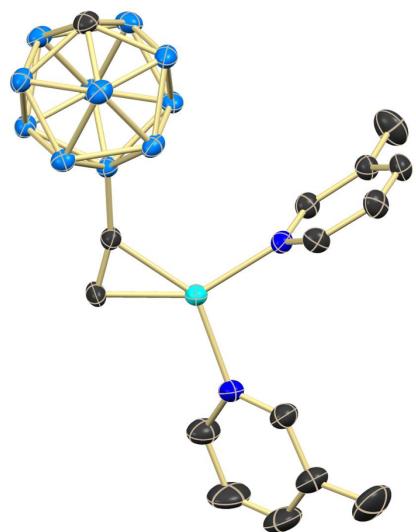
### Crystal structure of 3c (CCDC1910717)

Compound **3c** (10 mg) was dissolved in anhydrous ethanol (2.0 mL) in a 4 mL glass vial to give a clear solution. Slow evaporation afforded colorless crystals of the composition [Cu(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)(3-MePy)<sub>2</sub>] suitable for X-ray diffraction within 4 d at 25 °C.

---

Bond precision:	C-C = 0.0055 Å	Wavelength=0.71073	
Cell:	a=8.3718(8) alpha=102.473(7)	b=11.5149(11) beta=106.926(7)	c=13.4954(9) gamma=107.523(9)
Temperature:	293 K		
	Calculated	Reported	
Volume	1119.2(2)	1119.18(18)	
Space group	P -1	P -1	
Hall group	-P 1	-P 1	
Moiety formula	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>	
Sum formula	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>	C <sub>15</sub> H <sub>26</sub> B <sub>11</sub> Cu N <sub>2</sub>	
Mr	416.84	416.83	
D <sub>x</sub> , g cm <sup>-3</sup>	1.237	1.237	
Z	2	2	
μ (mm <sup>-1</sup> )	0.979	0.979	
F <sub>000</sub>	428.0	428.0	
F <sub>000'</sub>	428.69		
h,k,lmax	10,13,16	10,13,16	
Nref	4098	4083	
Tmin,Tmax	0.744,0.889	0.712,1.000	
Tmin'	0.650		
Correction method=	# Reported T Limits: Tmin=0.712 Tmax=1.000		
AbsCorr =	MULTI-SCAN		
Data completeness=	0.996	Theta(max)= 25.347	
R(reflections)=	0.0468( 3361)	wR2(reflections)= 0.1155( 4083)	
S =	1.047	Npar= 268	

---



**Figure S4.** ORTEP representation of **3c**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

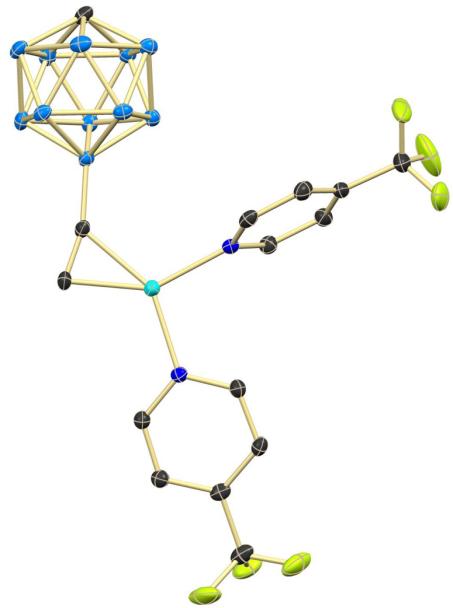
## Crystal structure of 3d (CCDC1910718)

Compound **3d** (10 mg) was dissolved in anhydrous ethanol (2.0 mL) in a 4 mL glass vial to give a clear solution. Slow evaporation afforded colorless crystals of the composition [(CB<sub>11</sub>H<sub>11</sub>-12-C≡CH)Cu(4-CF<sub>3</sub>Py)<sub>2</sub>] suitable for X-ray diffraction within 4 d at 25 °C.

---

Bond precision:	C-C = 0.0052 Å	Wavelength=0.71073	
Cell:	a=10.0994(8) alpha=66.212(7)	b=11.2360(9) beta=75.011(7)	c=11.7320(9) gamma=89.796(7)
Temperature:	168 K		
	Calculated	Reported	
Volume	1169.09(18)	1169.09(17)	
Space group	P -1	P -1	
Hall group	-P 1	-P 1	
Moiety formula	C <sub>15</sub> H <sub>20</sub> B <sub>11</sub> Cu F <sub>6</sub> N <sub>2</sub>	C <sub>15</sub> H <sub>20</sub> B <sub>11</sub> Cu F <sub>6</sub> N <sub>2</sub>	
Sum formula	C <sub>15</sub> H <sub>20</sub> B <sub>11</sub> Cu F <sub>6</sub> N <sub>2</sub>	C <sub>15</sub> H <sub>20</sub> B <sub>11</sub> Cu F <sub>6</sub> N <sub>2</sub>	
Mr	524.79	524.78	
Dx, g cm <sup>-3</sup>	1.491	1.491	
Z	2	2	
Mu (mm <sup>-1</sup> )	0.990	0.990	
F000	524.0	524.0	
F000'	524.89		
h,k,lmax	12,13,14	12,13,14	
Nref	4276	4264	
Tmin,Tmax	0.691,0.888	0.706,1.000	
Tmin'	0.616		
Correction method= #	Reported T Limits: Tmin=0.706 Tmax=1.000		
AbsCorr =	MULTI-SCAN		
Data completeness=	0.997	Theta(max)= 25.348	
R(reflections)=	0.0504( 3389)	wR2(reflections)= 0.1302( 4264)	
S =	1.046	Npar= 320	

---



**Figure S5.** ORTEP representation of **3d**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

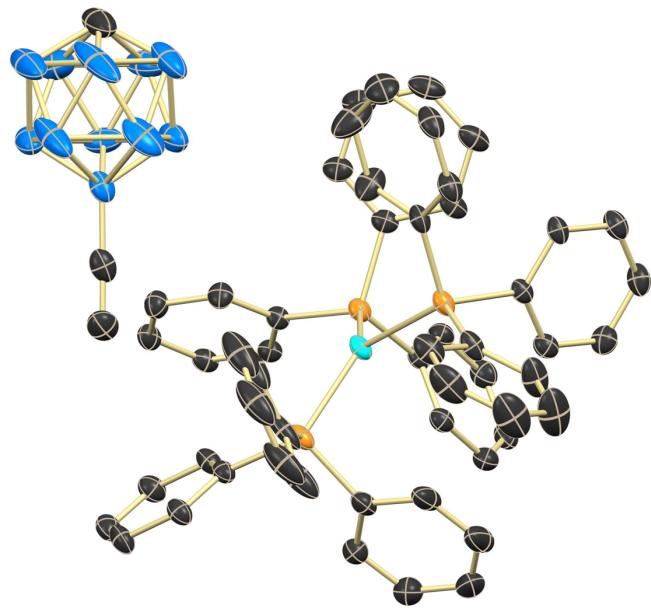
## Crystal structure of 4b (CCDC1910719)

Compound **4b** (10 mg) was dissolved in acetone (0.5 mL) in a 4 mL glass vial to give a clear solution, and 1.5 mL anhydrous ethanol was added to the vial. Slow evaporation afforded colorless crystals of the composition  $[\text{Cu}(\text{PPh}_3)_3][12\text{-C}\equiv\text{CH-CB}_{11}\text{H}_{11}]$  suitable for X-ray diffraction within 4 d at 25 °C.

---

Bond precision:	C-C = 0.0072 Å	Wavelength=0.71073	
Cell:	a=12.5248(8) alpha=106.031(5)	b=12.9185(8) beta=97.416(4)	c=18.8957(9) gamma=104.435(5)
Temperature:	150 K		
	Calculated	Reported	
Volume	2780.6(3)	2780.6(3)	
Space group	P -1	P -1	
Hall group	-P 1	-P 1	
Moiety formula	C54 H45 Cu P3, C3 H12 B11	C54 H45 Cu P3, C3 H12 B11	
Sum formula	C57 H57 B11 Cu P3	C57 H57 B11 Cu P3	
Mr	1017.40	1017.38	
Dx, g cm <sup>-3</sup>	1.215	1.215	
Z	2	2	
μ (mm <sup>-1</sup> )	0.516	0.516	
F000	1056.0	1056.0	
F000'	1057.50		
h,k,lmax	15,15,22	15,15,22	
Nref	10181	10143	
Tmin,Tmax	0.805,0.857	0.735,1.000	
Tmin'	0.805		
Correction method=	# Reported T Limits: Tmin=0.735 Tmax=1.000		
AbsCorr	= MULTI-SCAN		
Data completeness=	0.996	Theta(max)= 25.350	
R(reflections)=	0.0613( 7231)	wR2(reflections)= 0.1726( 10143)	
S =	1.034	Npar= 650	

---



**Figure S6.** ORTEP representation of **4b**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

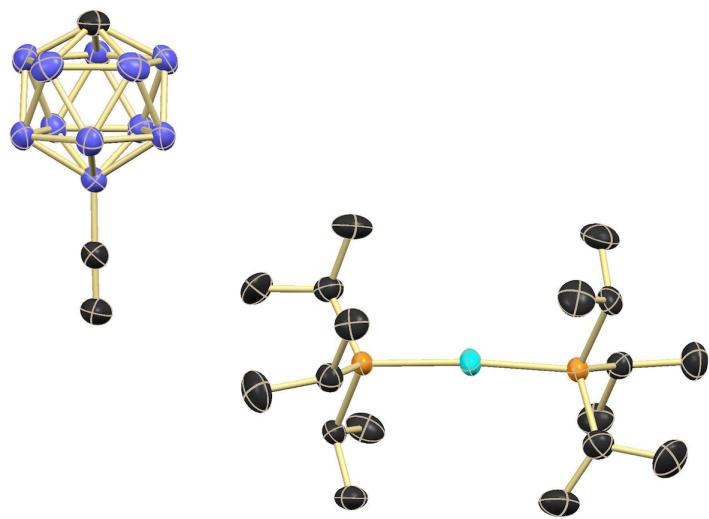
## Crystal structures of 4c (CCDC1910721)

Compound **4c** (10 mg) was dissolved in acetone (0.5 mL) in a 4 mL glass vial to give a clear solution, and 1.5 mL anhydrous ethanol was added to the vial. Slow evaporation afforded colorless crystals of the composition  $[\text{Cu}(\text{PiPr}_3)_2][12\text{-C}\equiv\text{CH-CB}_{11}\text{H}_{11}]$  suitable for X-ray diffraction within 4 d at 25 °C.

---

Bond precision:	C-C = 0.0045 Å	Wavelength=0.71073	
Cell:	a=11.9970(7) alpha=90	b=15.5376(7) beta=92.697(5)	c=17.7707(8) gamma=90
Temperature:	293 K		
	Calculated	Reported	
Volume	3308.9(3)	3308.9(3)	
Space group	P 21/n	P 1 21/n 1	
Hall group	-P 2yn	-P 2yn	
Moiety formula	C18 H42 Cu P2, C3 H12 B11	C18 H42 Cu P2, C3 H12 B11	
Sum formula	C21 H54 B11 Cu P2	C21 H54 B11 Cu P2	
Mr	551.04	551.03	
Dx, g cm <sup>-3</sup>	1.106	1.106	
Z	4	4	
Mu (mm <sup>-1</sup> )	0.767	0.767	
F000	1176.0	1176.0	
F000'	1178.25		
h,k,lmax	14,18,21	14,18,21	
Nref	6279	6266	
Tmin,Tmax	0.632,0.794	0.884,1.000	
Tmin'	0.601		
Correction method= # Reported T Limits: Tmin=0.884 Tmax=1.000			
AbsCorr = MULTI-SCAN			
Data completeness= 0.998	Theta(max)= 25.679		
R(reflections)= 0.0424( 4378)	wR2(reflections)= 0.1107( 6266)		
S = 1.018	Npar= 329		

---



**Figure S7.** ORTEP representation of **4c**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

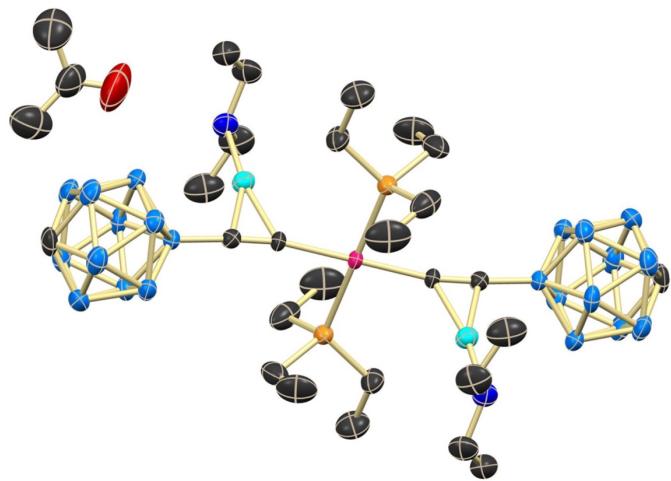
## Crystal structure of **6a** (CCDC1910723)

Compound **6a** (10 mg) was dissolved in acetone (0.7 mL) in a 4 mL glass vial, which was placed in a 20 mL glass vial containing hexane (4 mL). Vapor diffusion afforded colorless crystals of the composition *trans*-Pd(PEt<sub>3</sub>)<sub>2</sub>[(12-C≡C-CB<sub>11</sub>H<sub>11</sub>)(Cu(HNEt<sub>2</sub>))]<sub>2</sub> • 2Me<sub>2</sub>CO suitable for X-ray diffraction within 10 d at 25 °C.

---

Bond precision:	C-C = 0.0116 Å	Wavelength=0.71073	
Cell:	a=15.7282(9) alpha=90	b=10.9616(6) beta=97.622(6)	c=17.0109(14) gamma=90
Temperature:	293 K		
	Calculated	Reported	
Volume	2906.9(3)	2906.9(3)	
Space group	P 21/n	P 1 21/n 1	
Hall group	-P 2yn	-P 2yn	
Moiety formula	C <sub>26</sub> H <sub>74</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> P <sub>2</sub> Pd, 2(C <sub>3</sub> H <sub>6</sub> O)	C <sub>26</sub> H <sub>74</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> P <sub>2</sub> Pd, 2(C <sub>3</sub> H <sub>6</sub> O)	
Sum formula	C <sub>32</sub> H <sub>86</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> O <sub>2</sub> P <sub>2</sub> Pd	C <sub>32</sub> H <sub>86</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> O <sub>2</sub> P <sub>2</sub> Pd	
Mr	1064.29	1064.26	
Dx, g cm <sup>-3</sup>	1.216	1.216	
Z	2	2	
Mu (mm <sup>-1</sup> )	1.115	1.115	
F000	1104.0	1104.0	
F000'	1103.85		
h,k,lmax	18,13,20	18,13,20	
Nref	5319	5307	
Tmin,Tmax	0.694,0.846	0.366,1.000	
Tmin'	0.634		
Correction method= # Reported T Limits: Tmin=0.366 Tmax=1.000			
AbsCorr = MULTI-SCAN			
Data completeness= 0.998	Theta(max)= 25.350		
R(reflections)= 0.0524( 3566)	wR2(reflections)= 0.1381( 5307)		
S = 1.030	Npar= 294		

---



**Figure S8.** ORTEP representation of **6a**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

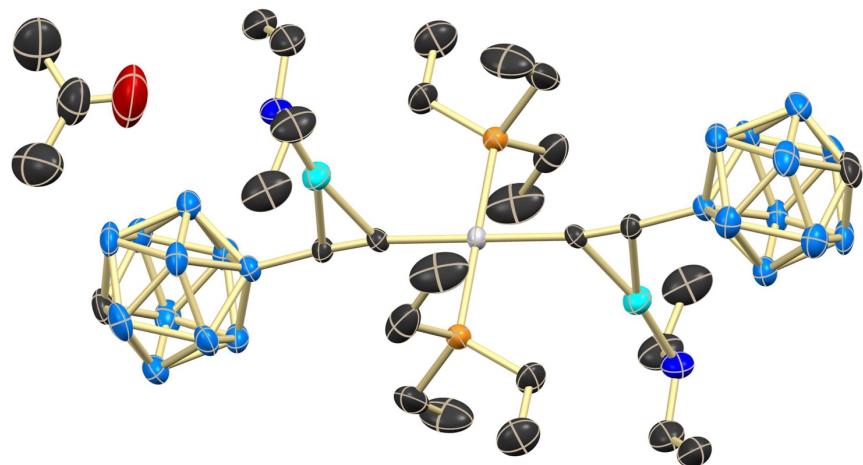
## Crystal structure of **6b** (CCDC1910724)

Compound **6b** (10 mg) was dissolved in acetone (0.7 mL) in a 4 mL glass vial, which was placed in a 20 mL glass vial containing hexane (4 mL). Vapor diffusion afforded colorless crystals of the composition *trans*-Pt(PEt<sub>3</sub>)<sub>2</sub>[(12-C≡C-CB<sub>11</sub>H<sub>11</sub>)(Cu(HNEt<sub>2</sub>))]<sub>2</sub> • 2Me<sub>2</sub>CO suitable for X-ray diffraction within 10 d at 25 °C.

---

Bond precision:	C-C = 0.0146 Å	Wavelength=0.71073	
Cell:	a=15.7322(6) alpha=90	b=10.9370(5) beta=97.802(5)	c=16.9807(11) gamma=90
Temperature:	293 K		
	Calculated	Reported	
Volume	2894.7(3)	2894.7(3)	
Space group	P 21/n	P 1 21/n 1	
Hall group	-P 2yn	-P 2yn	
Moiety formula	C <sub>26</sub> H <sub>74</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> P <sub>2</sub> Pt, 2(C <sub>3</sub> H <sub>6</sub> O)	C <sub>26</sub> H <sub>74</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> P <sub>2</sub> Pt, 2(C <sub>3</sub> H <sub>6</sub> O)	
Sum formula	C <sub>32</sub> H <sub>86</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> O <sub>2</sub> P <sub>2</sub> Pt	C <sub>32</sub> H <sub>86</sub> B <sub>22</sub> Cu <sub>2</sub> N <sub>2</sub> O <sub>2</sub> P <sub>2</sub> Pt	
Mr	1152.97	1152.95	
Dx, g cm <sup>-3</sup>	1.323	1.323	
Z	2	2	
Mu (mm <sup>-1</sup> )	3.223	3.223	
F000	1168.0	1168.0	
F000'	1166.54		
h,k,lmax	18,13,20	18,13,20	
Nref	5296	5279	
Tmin, Tmax	0.239, 0.476	0.418, 1.000	
Tmin'	0.198		
Correction method=	# Reported T Limits: Tmin=0.418 Tmax=1.000		
AbsCorr =	MULTI-SCAN		
Data completeness=	0.997	Theta(max)= 25.350	
R(reflections)=	0.0464( 4037)	wR2(reflections)= 0.1130( 5279)	
S =	1.059	Npar= 294	

---



**Figure S9.** ORTEP representation of **6b**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

## Crystal structure of 7 (CCDC1910725)

Compound 7 (20 mg) was dissolved in acetone (0.7 mL) in a 4 mL glass vial, which was placed in a 20 mL glass vial containing hexane (4 mL). Vapor diffusion afforded colorless crystals of the composition  $[CB_{11}H_{11}-12-(C\equiv C-C\equiv C)-12-CB_{11}H_{11}][NEt_4]_2$  suitable for X-ray diffraction within 10 d at 25 °C.

---

Bond precision: C-C = 0.0054 Å Wavelength=0.71073

Cell: a=33.205(4) b=11.124(1) c=22.362(3)  
alpha=90 beta=110.903(16) gamma=90  
Temperature: 293 K

	Calculated	Reported
Volume	7716.3(18)	7716.4(18)
Space group	C 2/c	C 1 2/c 1
Hall group	-C 2yc	-C 2yc
Moiety formula	C <sub>6</sub> H <sub>22</sub> B <sub>22</sub> , 2(C <sub>8</sub> H <sub>20</sub> N)	2(C <sub>8</sub> H <sub>20</sub> N), C <sub>6</sub> H <sub>22</sub> B <sub>22</sub>
Sum formula	C <sub>22</sub> H <sub>62</sub> B <sub>22</sub> N <sub>2</sub>	C <sub>22</sub> H <sub>62</sub> B <sub>22</sub> N <sub>2</sub>
Mr	592.56	592.55
D <sub>x</sub> , g cm <sup>-3</sup>	1.020	1.020
Z	8	8
Mu (mm <sup>-1</sup> )	0.049	0.049
F <sub>000</sub>	2544.0	2544.0
F <sub>000'</sub>	2544.41	
h,k,lmax	39,13,26	40,13,26
Nref	7057	7024
Tmin,Tmax	0.977,0.986	0.946,1.000
Tmin'	0.977	

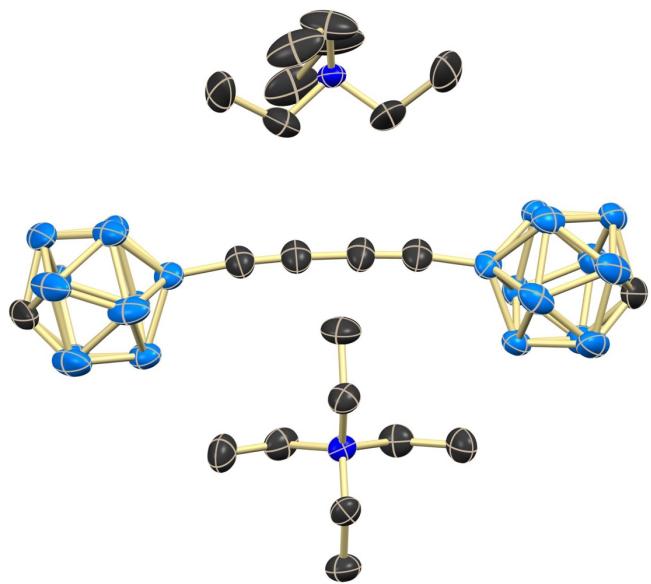
Correction method= # Reported T Limits: Tmin=0.946 Tmax=1.000  
AbsCorr = MULTI-SCAN

Data completeness= 0.995 Theta(max)= 25.349

R(reflections)= 0.0904( 3756) wR2(reflections)= 0.3023( 7024)

S = 1.037 Npar= 490

---



**Figure S10.** ORTEP representation of **7**. Hydrogen atoms are omitted for clarity; 30% displacement ellipsoids.

## IV References

- [1] a) Himmelsbach, A.; Reiss, G. J.; Finze, M. *Inorg. Chem.* **2012**, *51*, 2679;  
b) Himmelsbach, A.; Finze, M. *J. Organomet. Chem.* **2010**, *695*, 1337;  
c) Zhang, K.; Shen, Y.; Yang, X.; Liu, J.; Jiang, T.; Finney, N.; Spingler, B.;  
Duttwyler, S. *Chem. Eur. J.* **2019**, *25*, 8754.
- [2] Sheldrick, G. M. *Acta Cryst.* **2015**, *A71*, 3.
- [3] Sheldrick, G. M. *Acta Cryst.* **2015**, *C71*, 3.
- [4] Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. *J. Appl. Cryst.* **2009**, *42*, 339.
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Pidcock, E.; Rodriguez-Monge, L.; Taylor, R.; van de Streek, J.; Wood, P. A. *J. Appl. Cryst.* **2008**, *41*, 466.

<sup>1</sup>H{<sup>11</sup>B} NMR, 500MHz, acetone-d6\*

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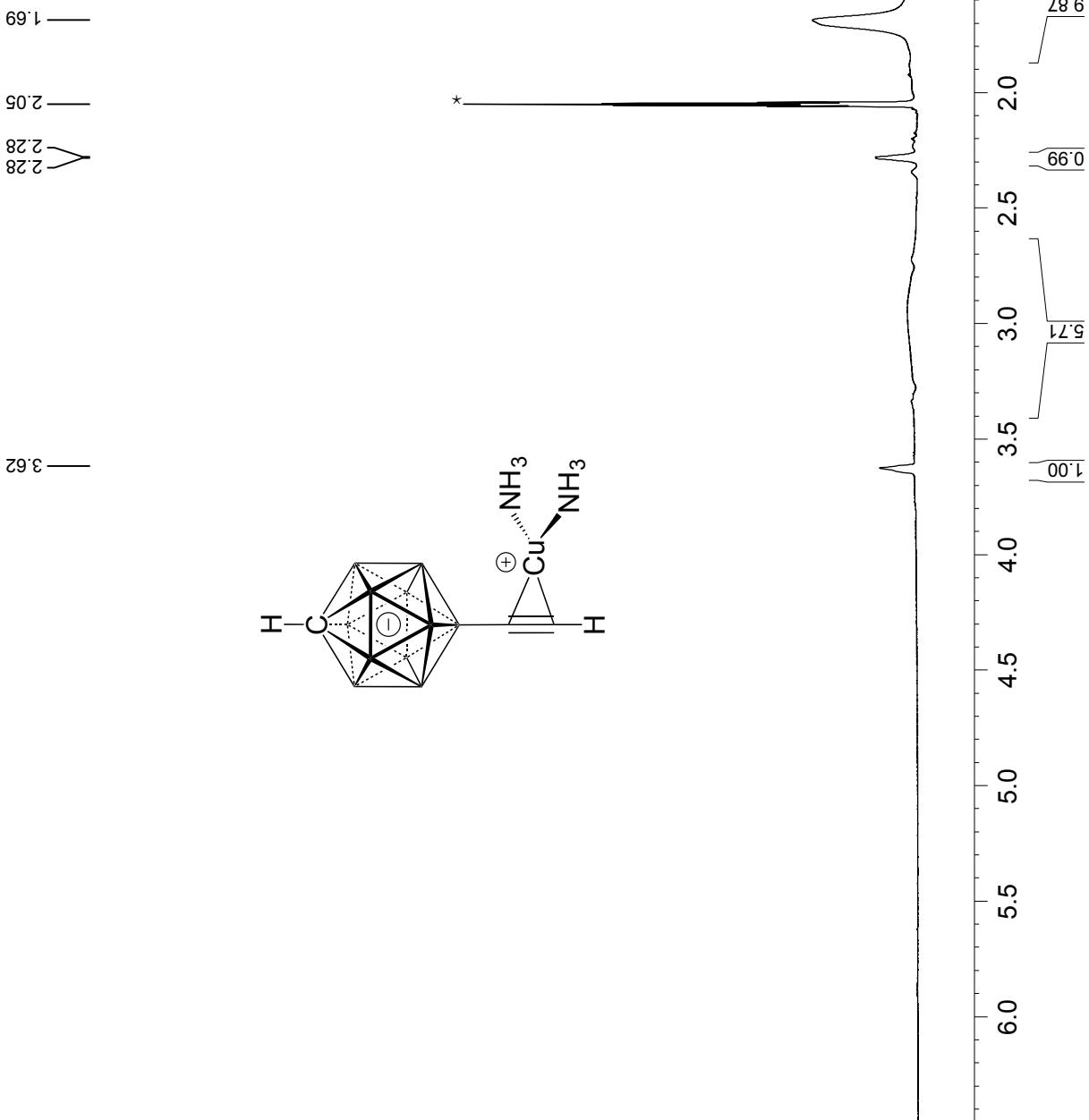
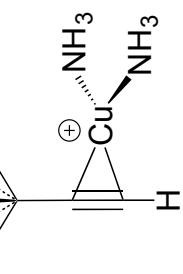
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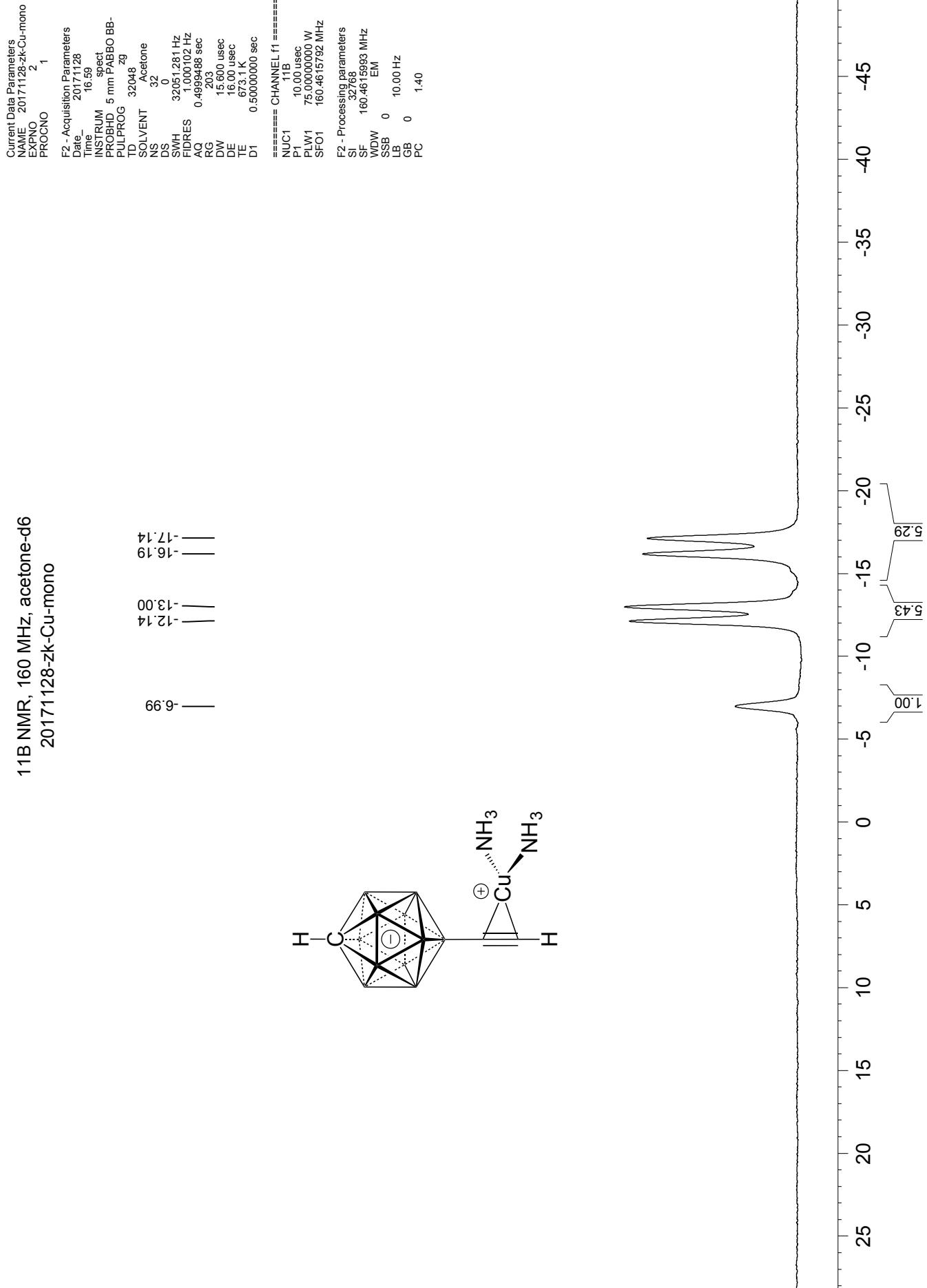
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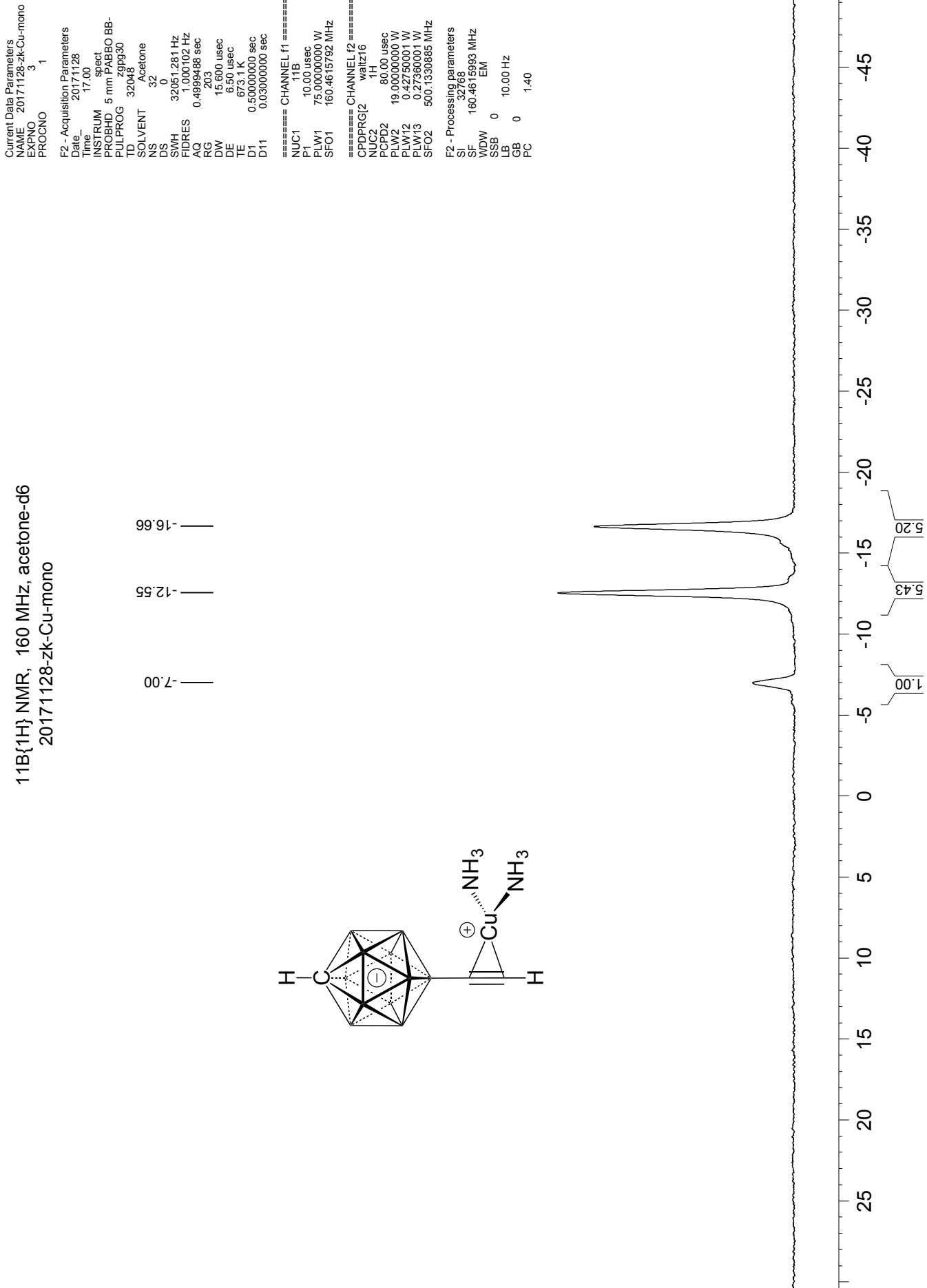
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20171128-zk-Cu-mono



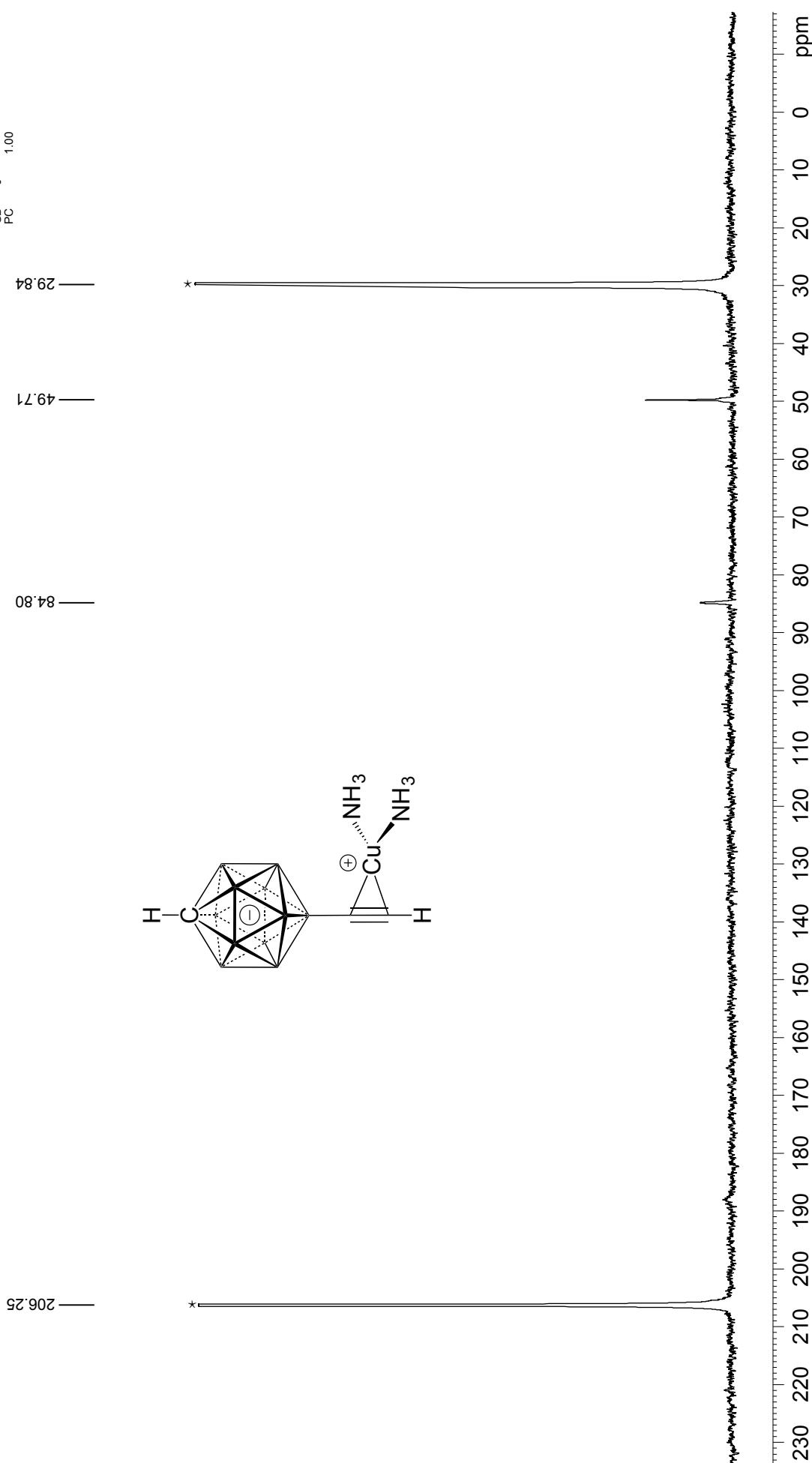
11B{<sup>1</sup>H} NMR, 160 MHz, acetone-d<sub>6</sub>  
20171128-zk-Cu-mono

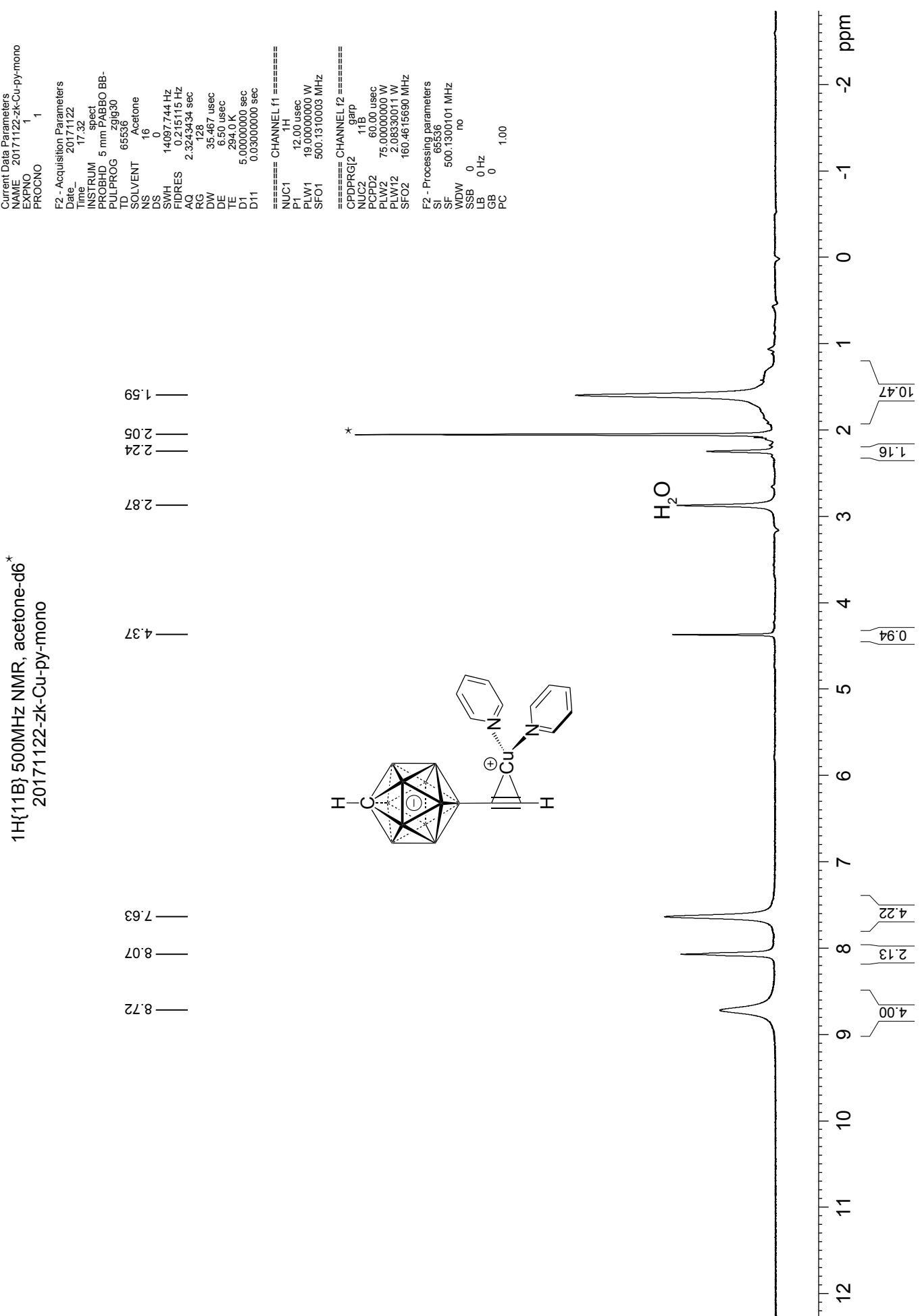


Current Data Parameters  
NAME 20171128-zk-Cu-mono-600MH-13C  
EXPNO 1  
PROCNO 1

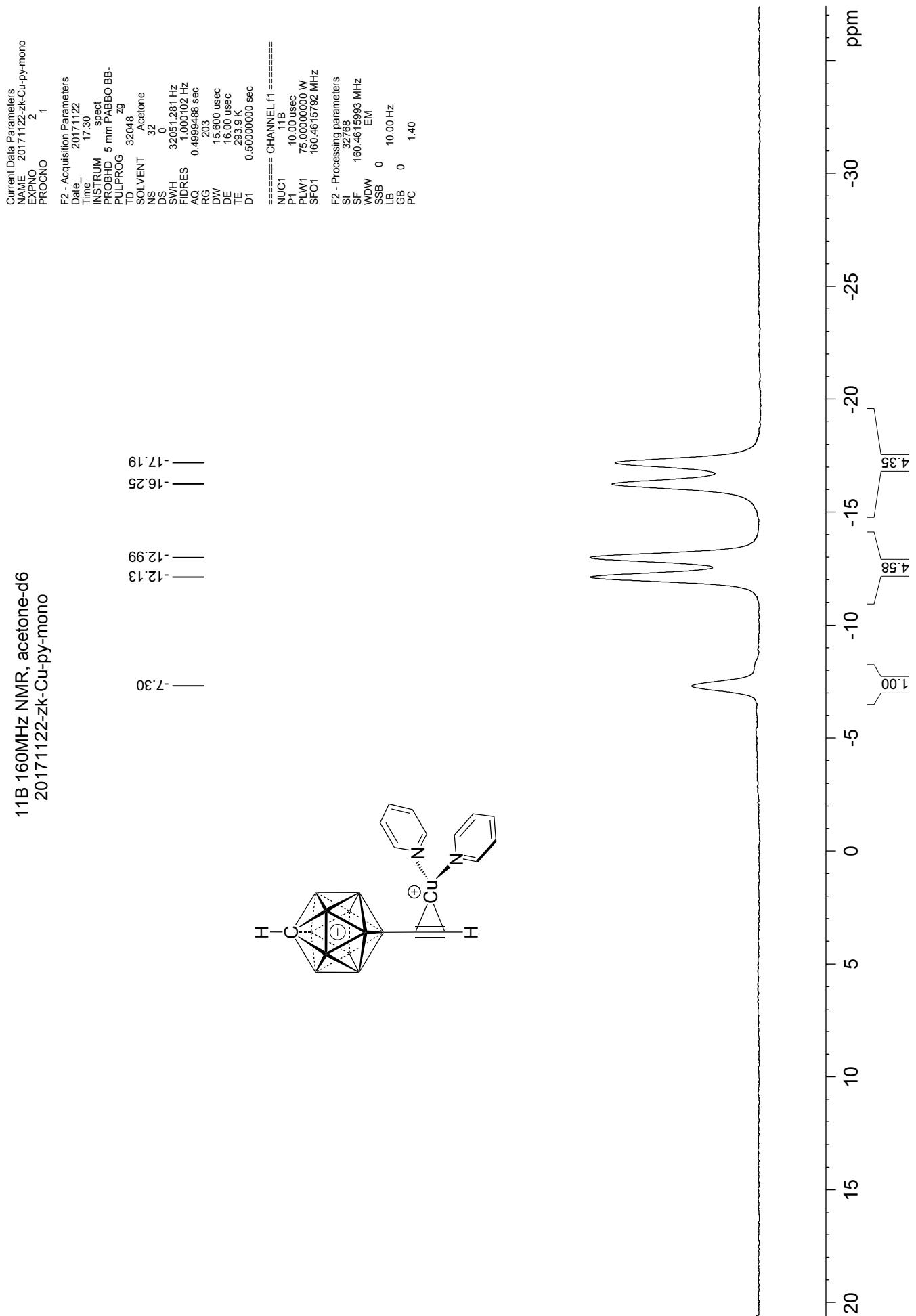
F2 - Processing parameters  
SI 68536  
SF 150.8139882 MHz  
WDW EM  
SSB 0  
LB 10.00 Hz  
GB 0  
PC 1.00

<sup>13</sup>C{<sup>1</sup>H} NMR, 151 MHz, acetone-d<sub>6</sub>  
20171128-zk-Cu-mono

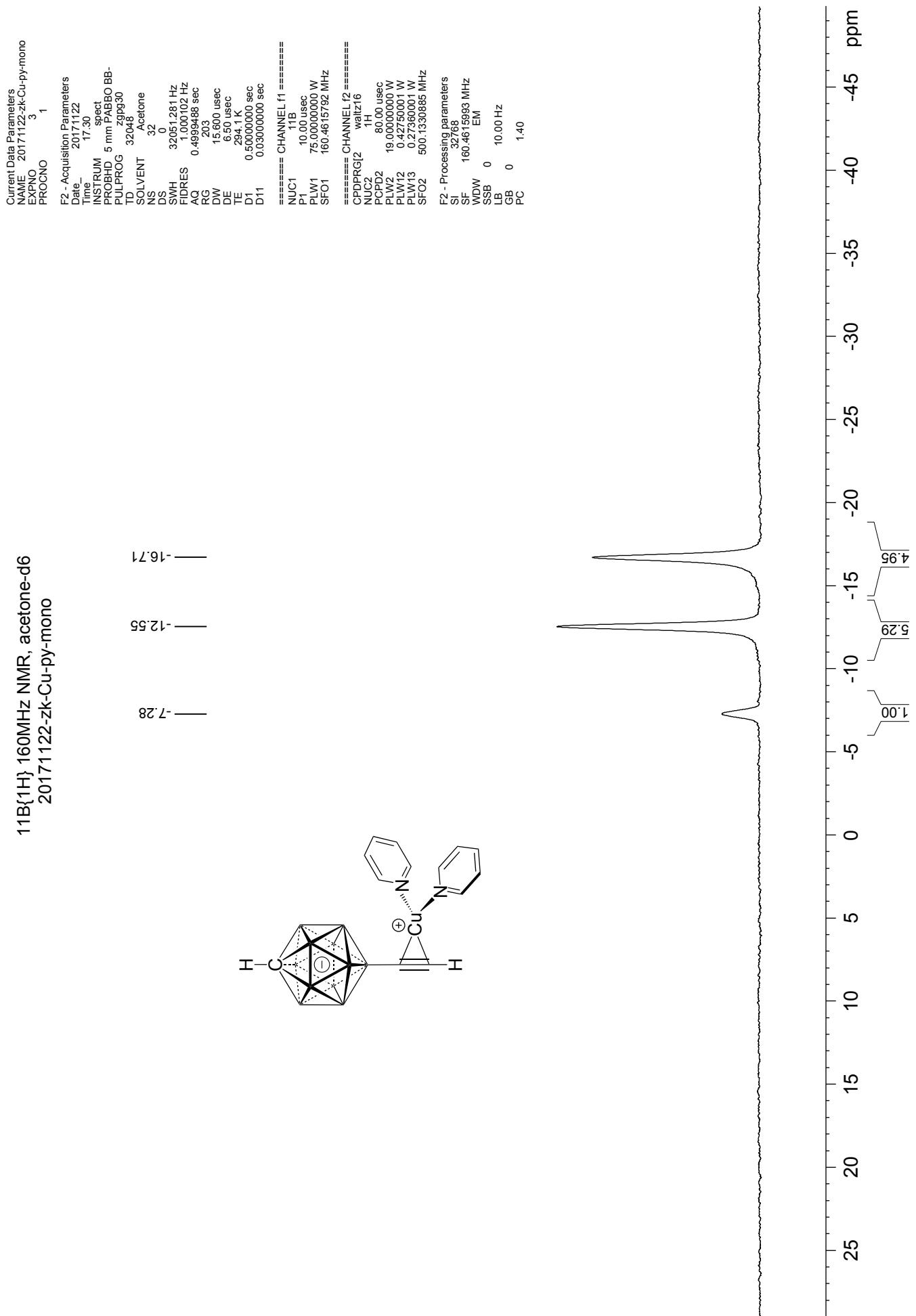




11B 160MHz NMR, acetone-d<sub>6</sub>  
20171122-zk-Cu-py-mono

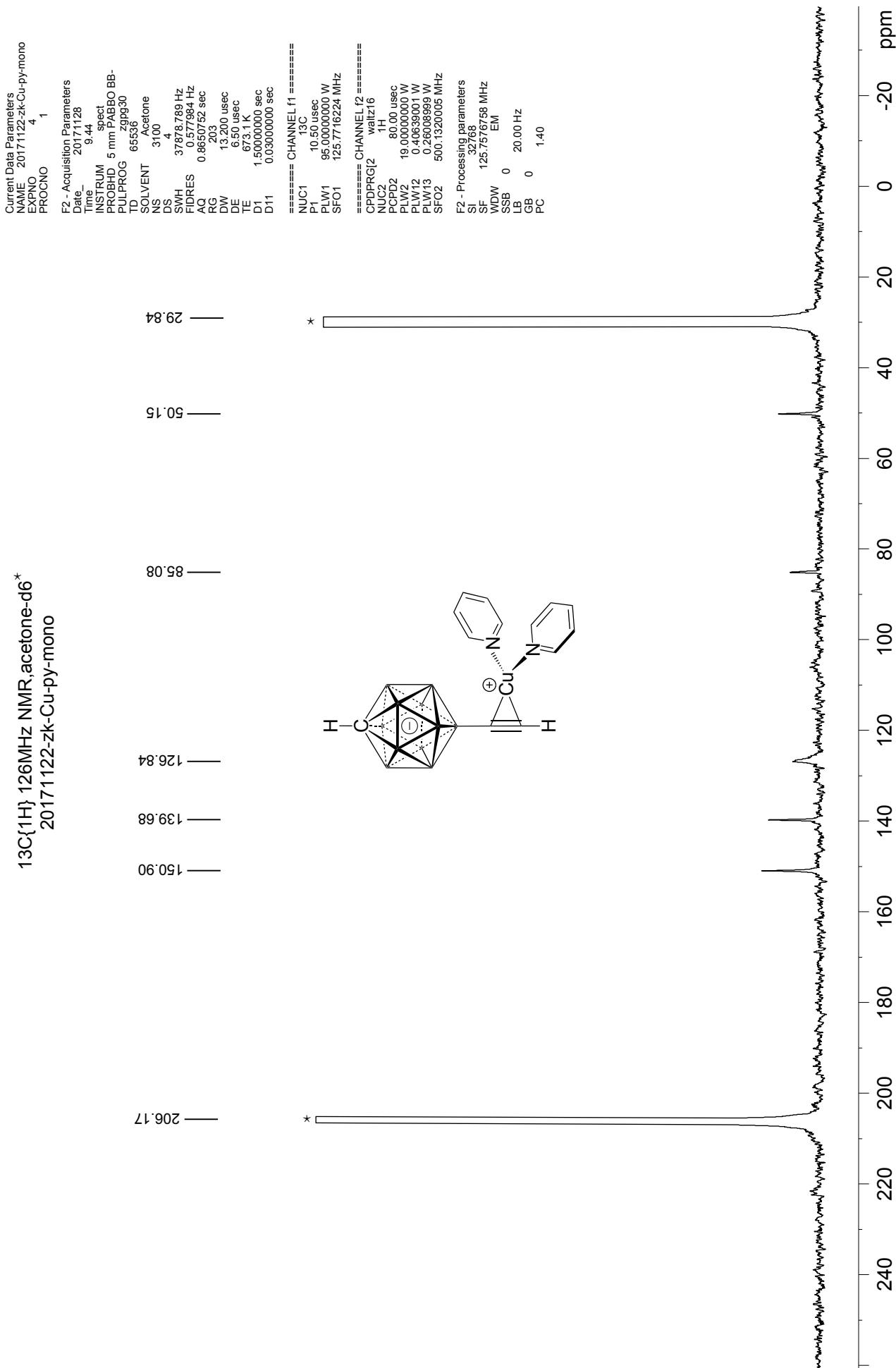


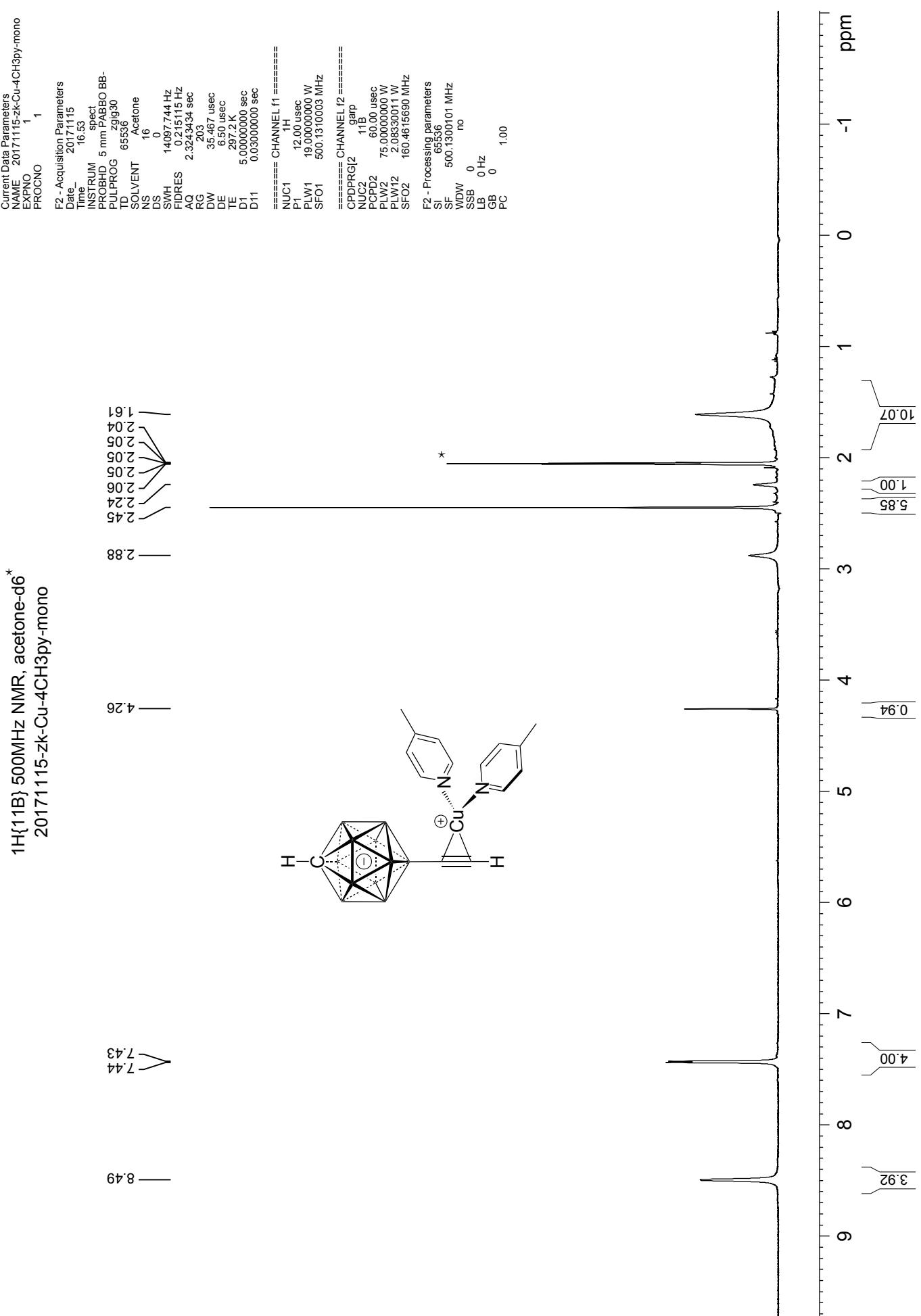
11B{<sup>1</sup>H} 160MHz NMR, acetone-d<sub>6</sub>  
20171122-zk-Cu-py-mono



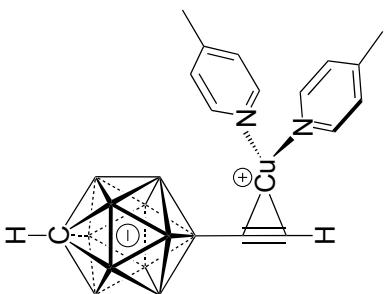
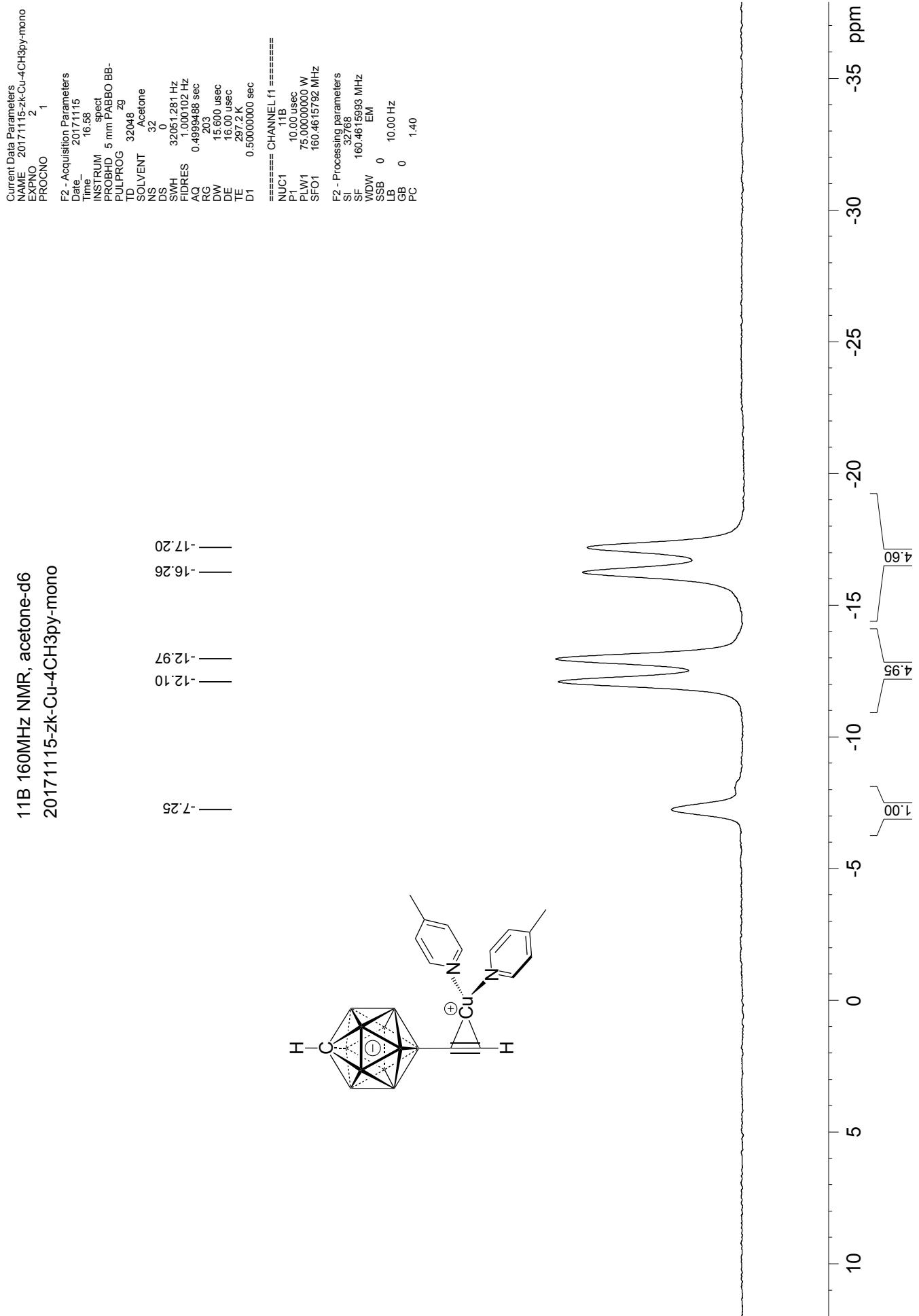
13C{1H} 126MHz NMR,acetone-d6\*

20171122-zk-Cu-py-mono





11B 160MHz NMR, acetone-d<sub>6</sub>  
 20171115-zk-Cu-4CH3py-mono



11B{<sup>1</sup>H} 160MHz NMR, acetone-d<sub>6</sub>  
20171115-zk-Cu-4CH<sub>3</sub>py-mono

11B{1H} 160MHz NMR, acetone-d<sub>6</sub>  
20171115-zk-Cu-4CH3py-mono

The figure shows the chemical structure of a Cu complex. It features a central Cu atom bonded to two 4-methylpyridine ligands. The Cu atom is also coordinated to the vertices of a cyclooctane cage, which is composed of eight carbon atoms. One methyl group from each pyridine ligand is in direct contact with one of the cage vertices. The entire complex is shown in a perspective view.

Current Data Parameters  
NAME 20171115-zk-Cu-4CH3py-mono  
EXPNO 3  
PROCNO 1

F2 - Acquisition Parameters

Date	20171115
Time	16:57
INSTRUM	Spect
PROBHD	5 mm PABBO BB-
PULPROG	zgpg30
TD	32048
SOLVENT	Acetone
NS	32
DS	0
SWH	32051.281 Hz
FIDRES	1.000102 Hz
AQ	0.4999488 sec
RG	203
DW	15.600 usec
DE	6.50 usec
TE	297.3 K
DI	0.50000000 sec
D11	0.03000000 sec

===== CHANNEL f1 =====

NUC1	<sup>11</sup> B
P1	10.00 usec
PLW1	75.00000000 W
SFO1	160.4615792 MHz

===== CHANNEL f2 =====

CPDPRGf2	waltz16
NUC2	<sup>1</sup> H
PCPD2	80.00 usec
PLW2	19.00000000 W
PLW12	0.12750001 W
PLW13	0.27380001 W
SFO2	500.1330885 MHz

F2 - Processing parameters

SI	32768
SF	160.4615993 MHz
WDW	EM
SSB	0
LB	10.00 Hz
GB	0
PC	1.40

-16.72

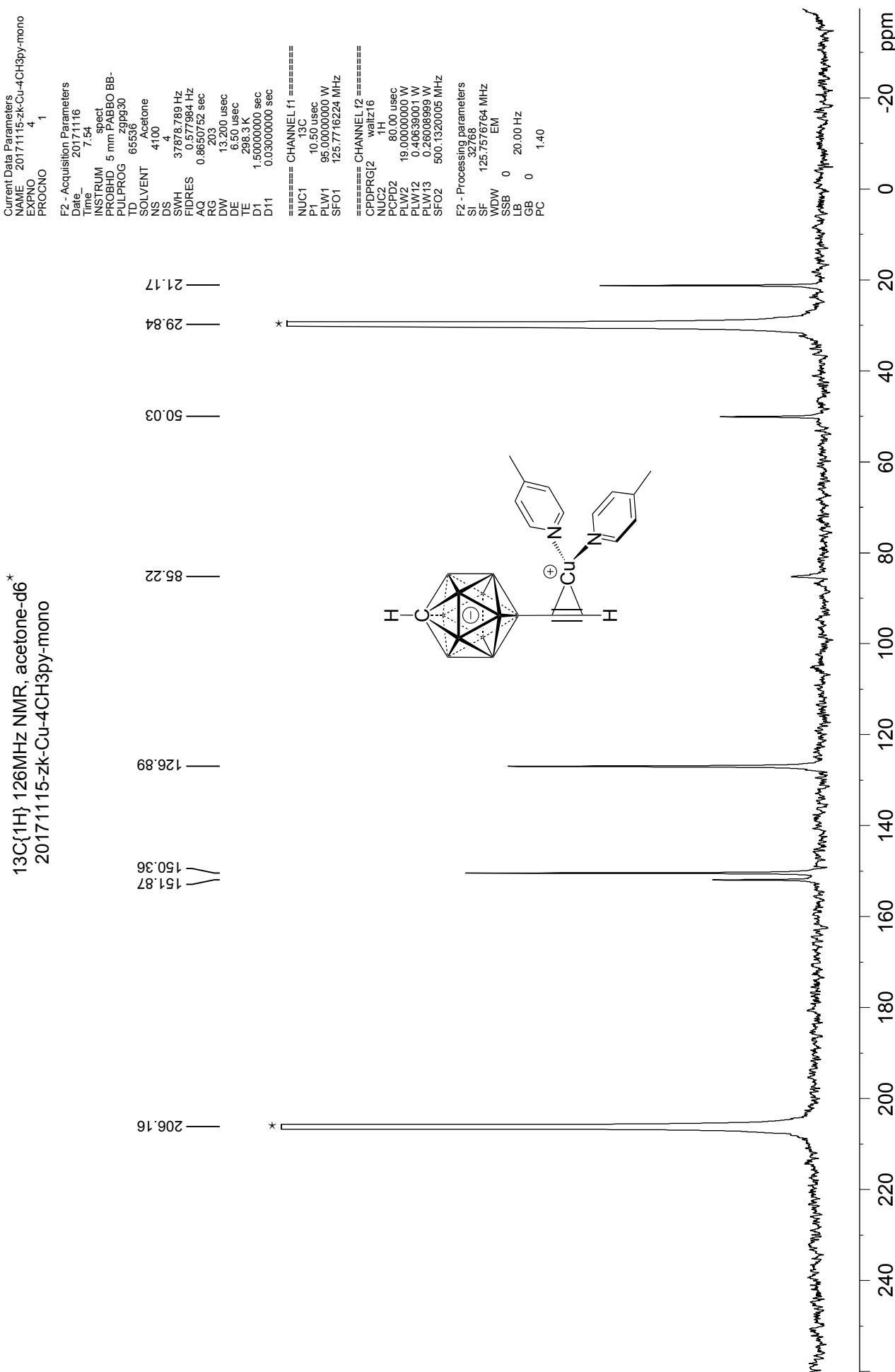
-12.53

-7.24

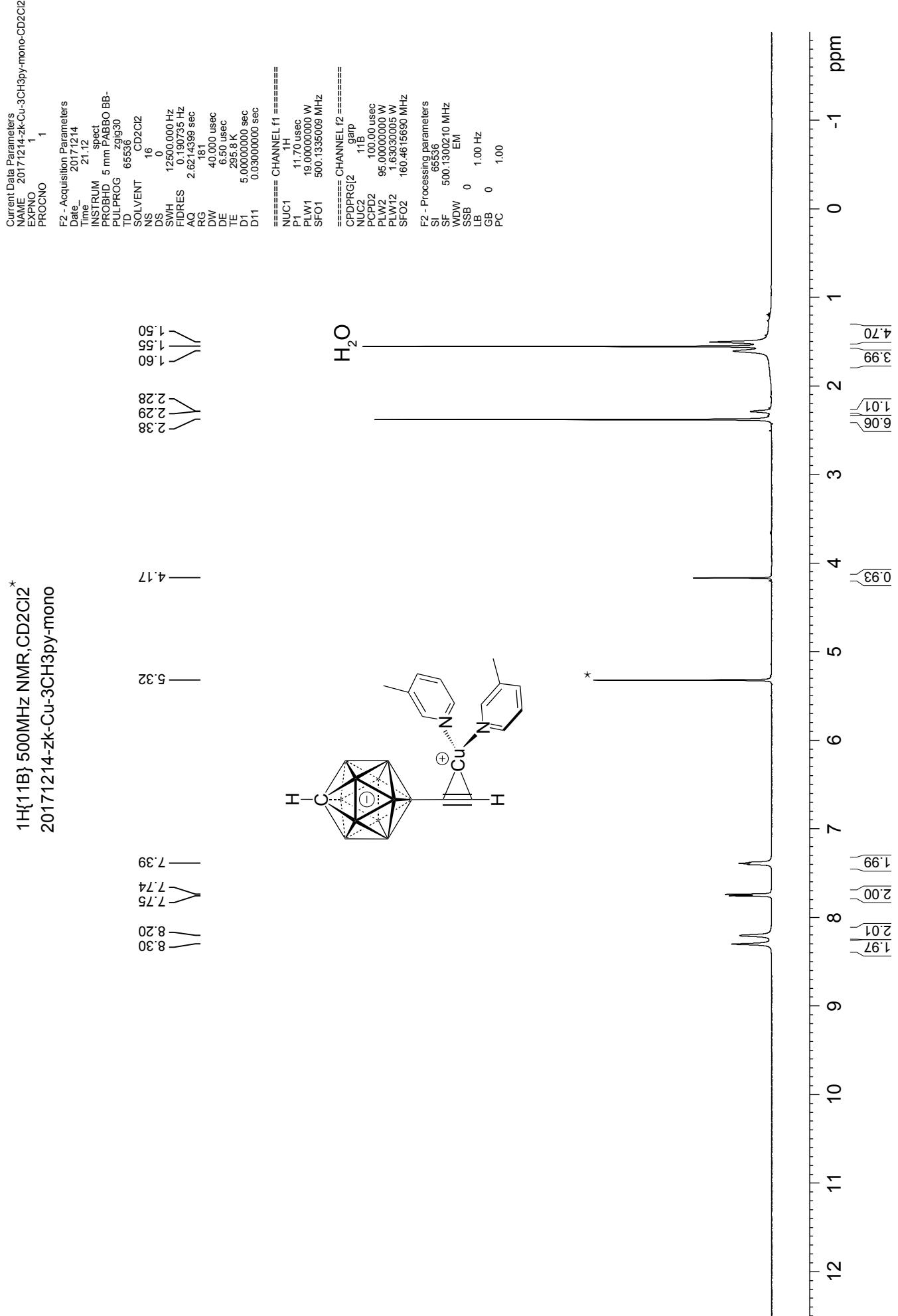
-5 0 5 10 15 20 25 -30 -25 -20 -15 -10

ppm

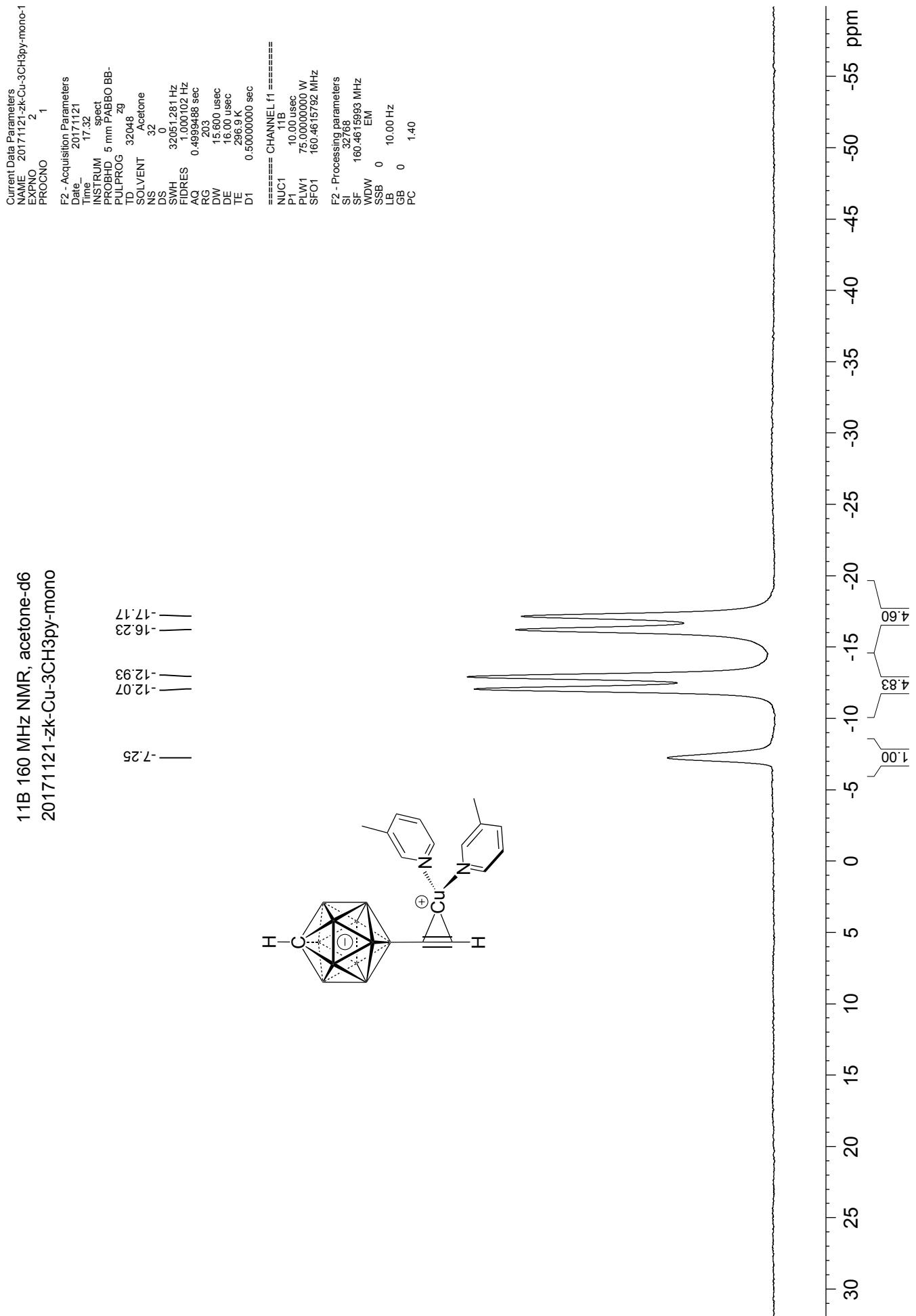
4.85 4.85 4.85 1.00



<sup>1</sup>H{<sup>13</sup>C} 500MHz NMR, CD2Cl<sub>2</sub>  
20171214-zk-Cu-3CH3py-mono\*



11B 160 MHz NMR, acetone-d<sub>6</sub>  
 20171121-zk-Cu-3CH3py-mono



Current Data Parameters  
NAME 20171121-zk-Cu-3CH3Py-mono-1  
EXPNO 3  
PROCNO 1

11B{<sup>1</sup>H} 160 MHz NMR, acetone-d<sub>6</sub>  
20171121-zk-Cu-3CH3py-mono

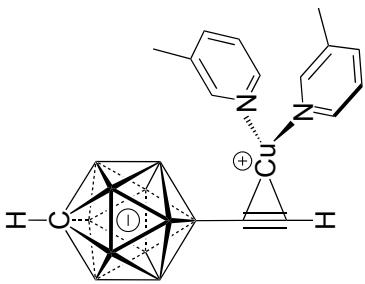
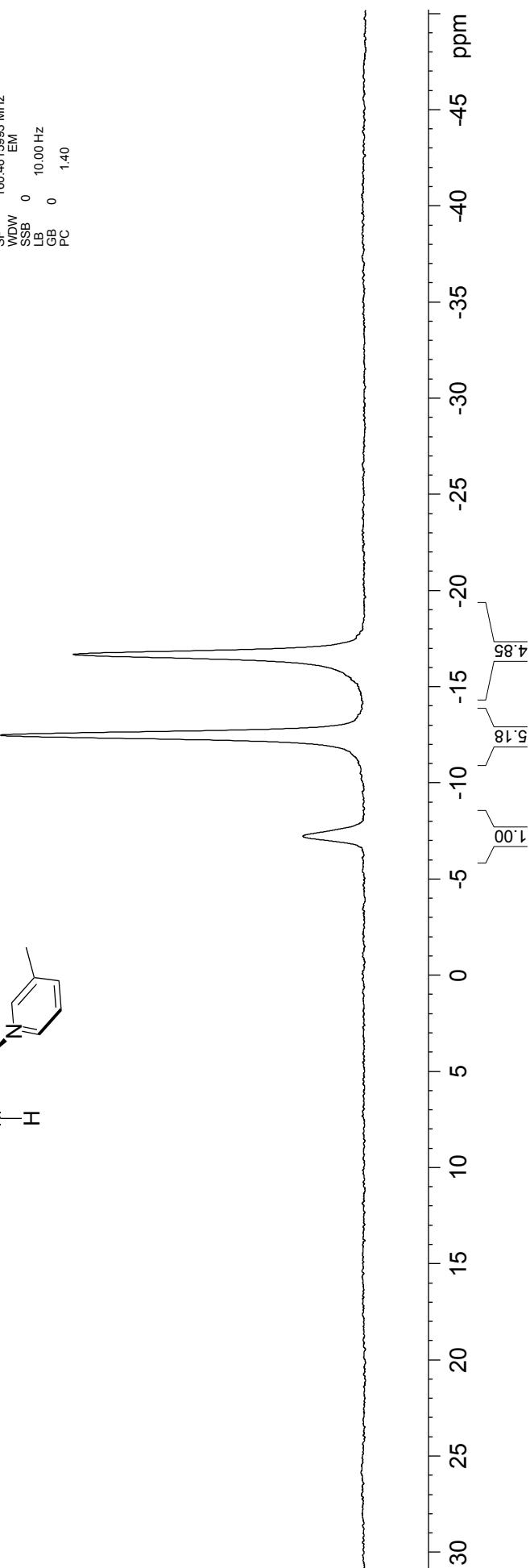
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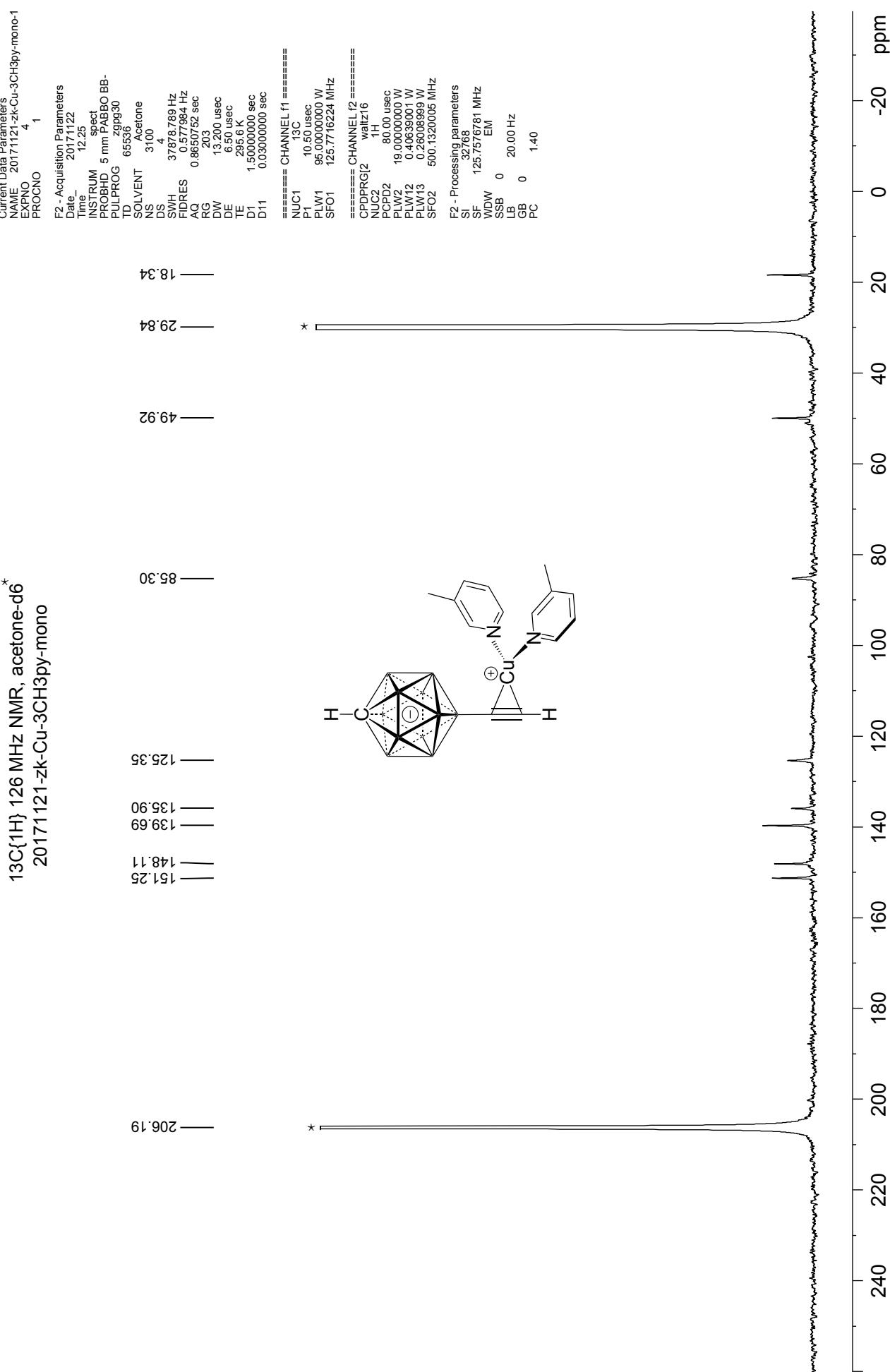
=====
===== CHANNEL 11 =====
NUC1      11B
P1        10.00 usec
PLW1     75.0000000 W
SF01    160.4615792 MHz

=====
===== CHANNEL 12 =====
CPDFRG12   wait16
NUC2      1H
CPDPD2    80.00 usec
PLW2     19.0000000 W
PLW12    0.4275001 W
PLW012   0.2738001 W
SF012    500.1330885 MHz

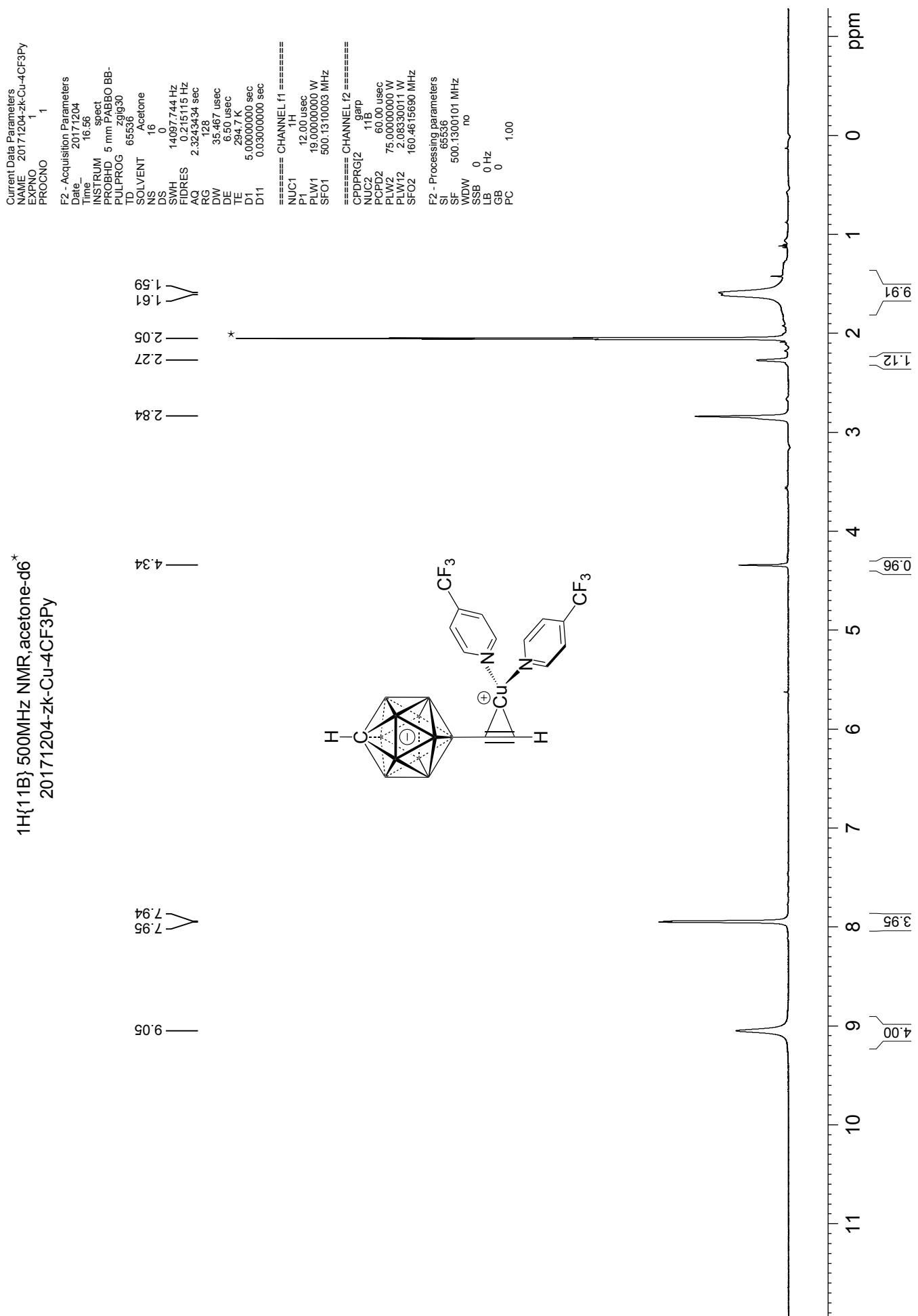
=====
F2 - Processing parameters
SI          32768
SSB         0
LB          10.00 Hz
GB          0
C          1.40

```

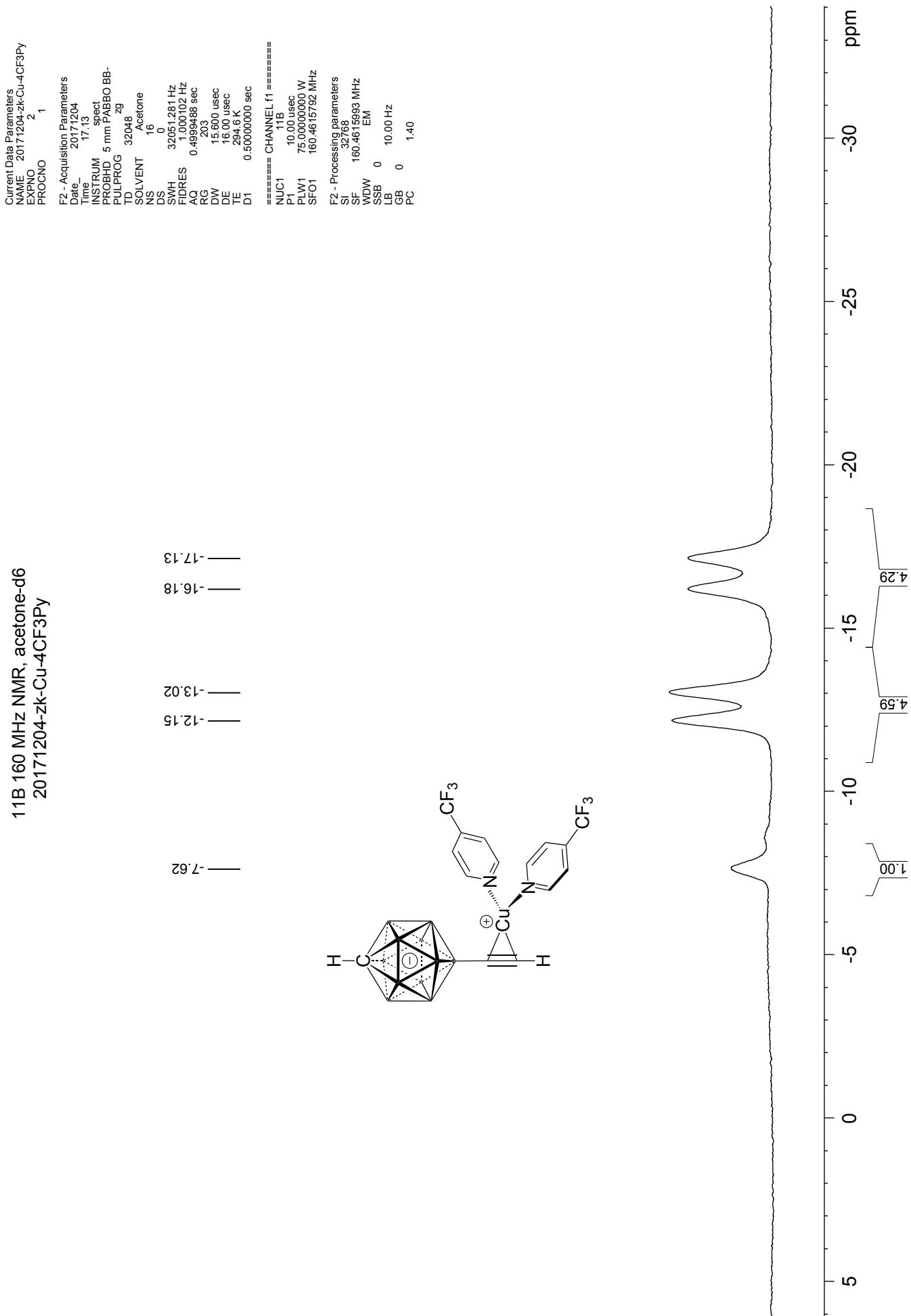




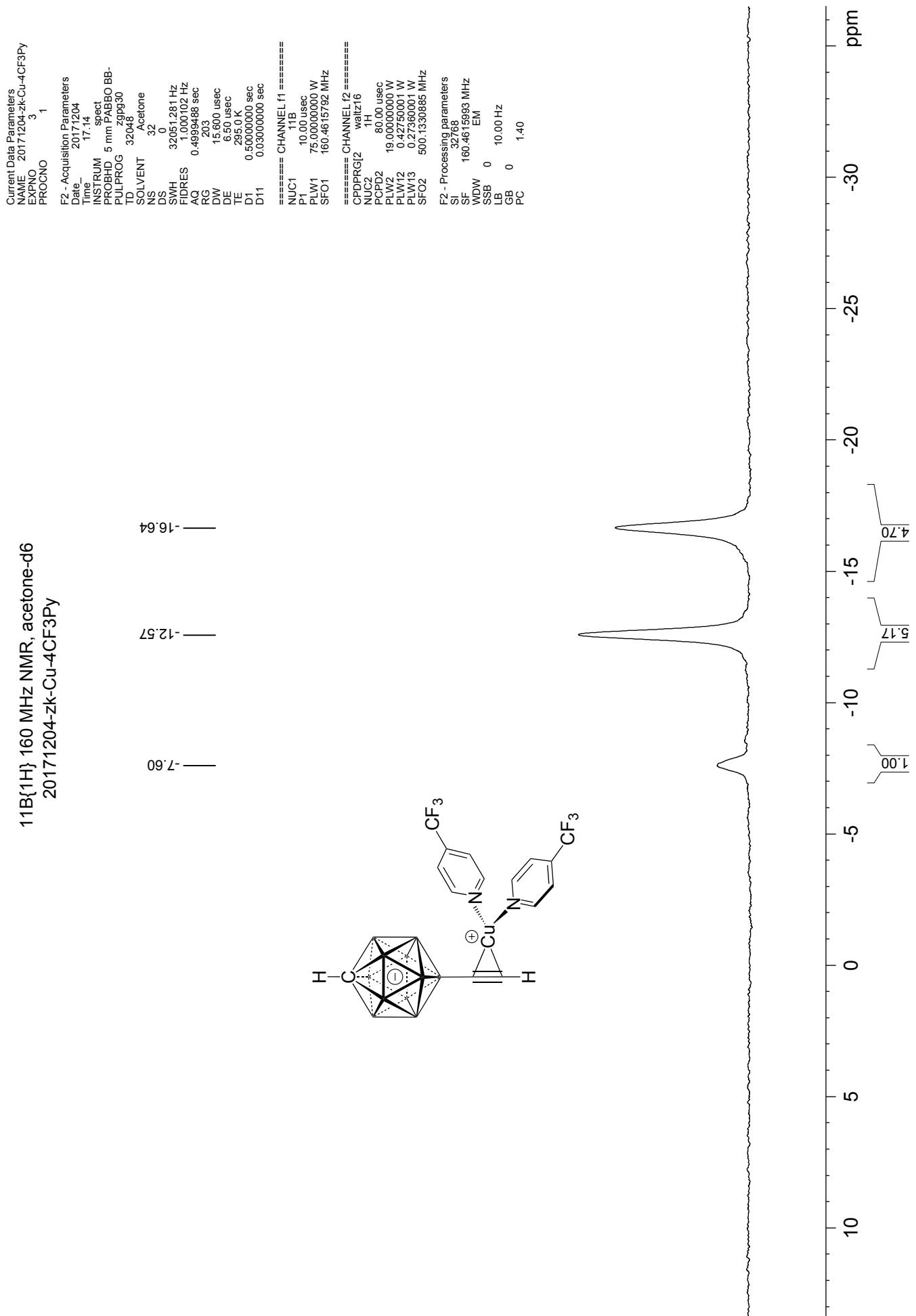
1H{11B} 500MHz NMR, acetone-d<sub>6</sub><sup>\*</sup>  
20171204-Zk-Cu-4CF3Py



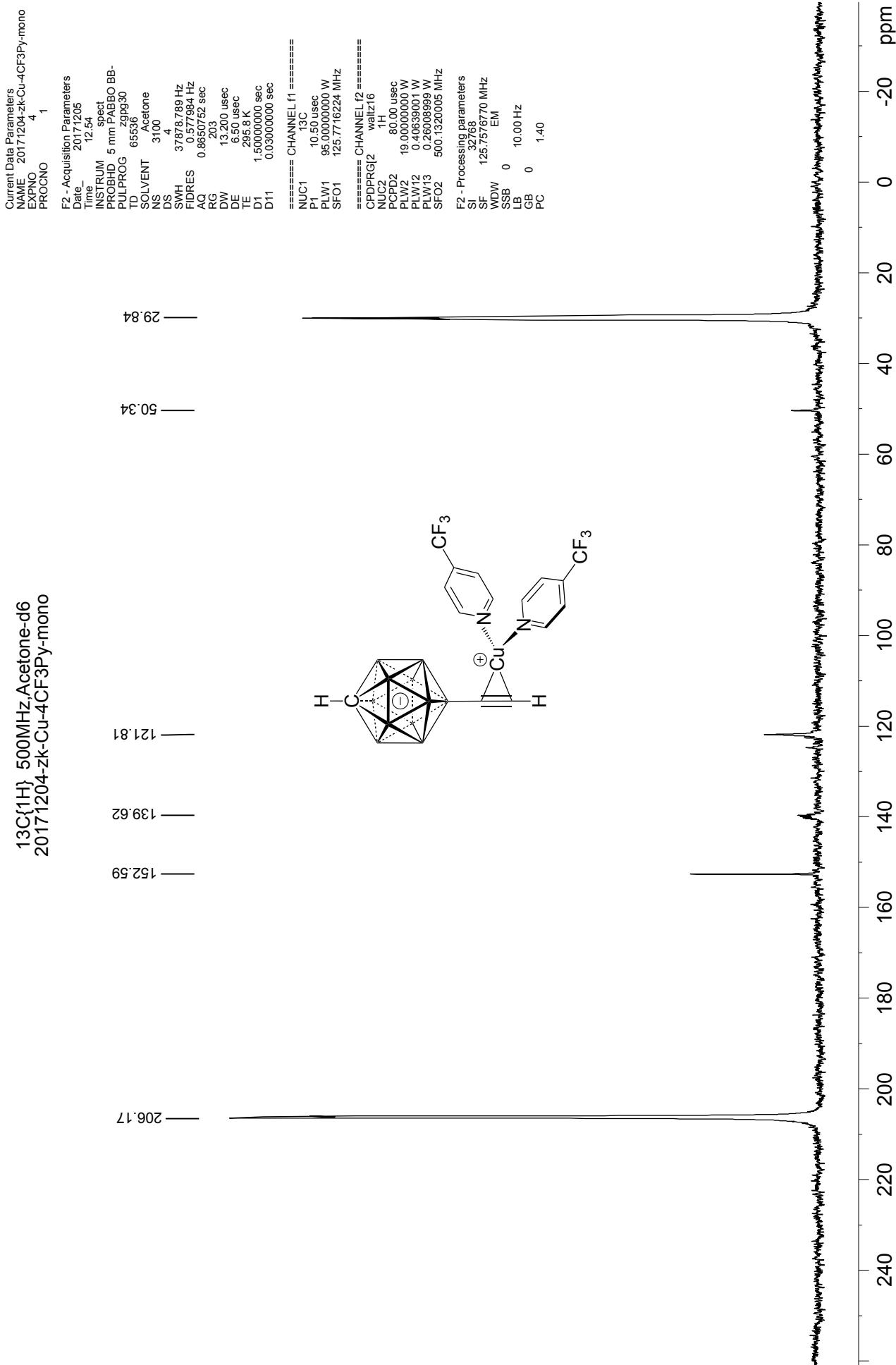
11B 160 MHz NMR, acetone-d<sub>6</sub>  
20171204-zk-Cu-4CF3Py

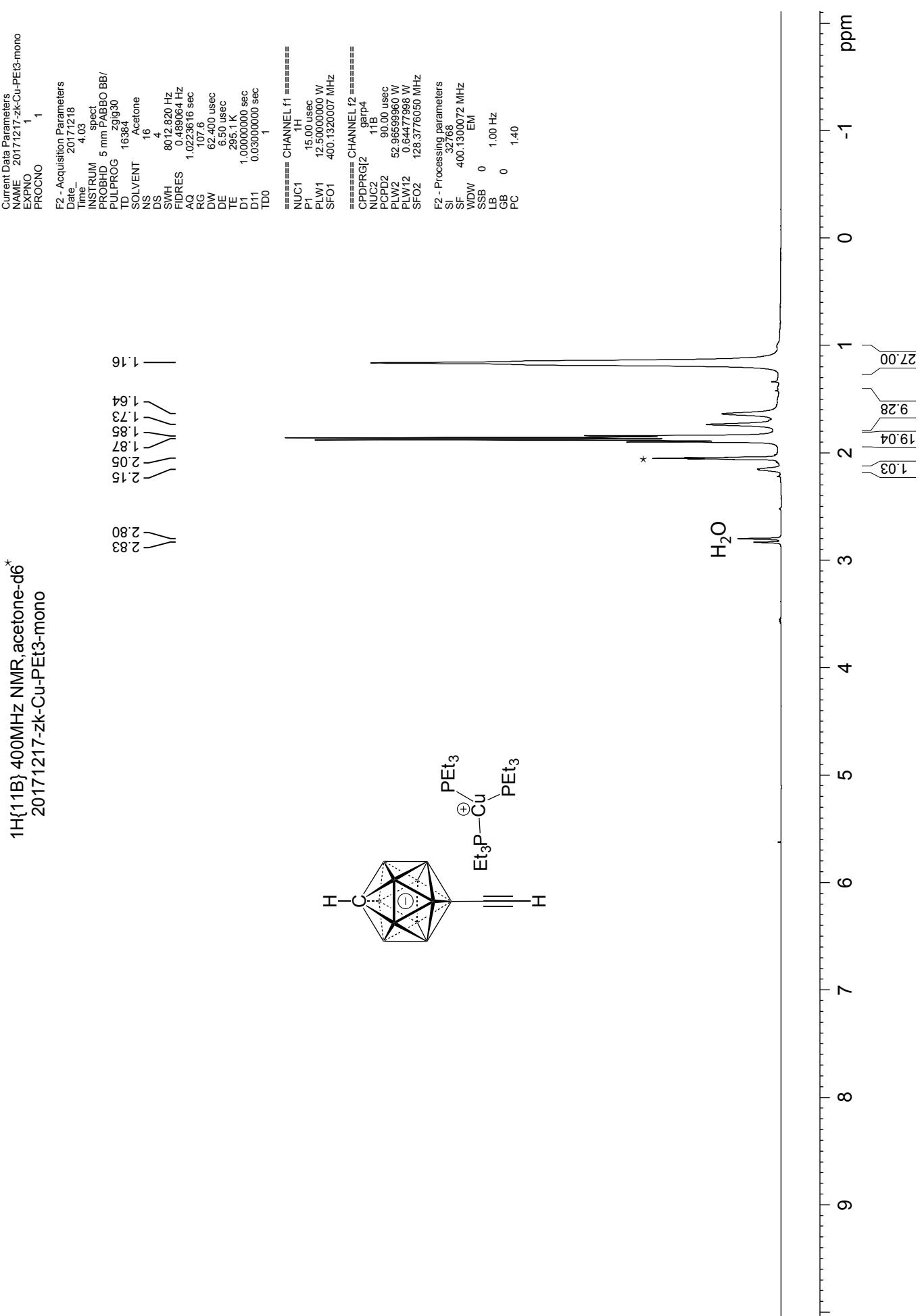


11B{1H} 160 MHz NMR, acetone-d<sub>6</sub>  
20171204-zk-Cu-4CF3Py

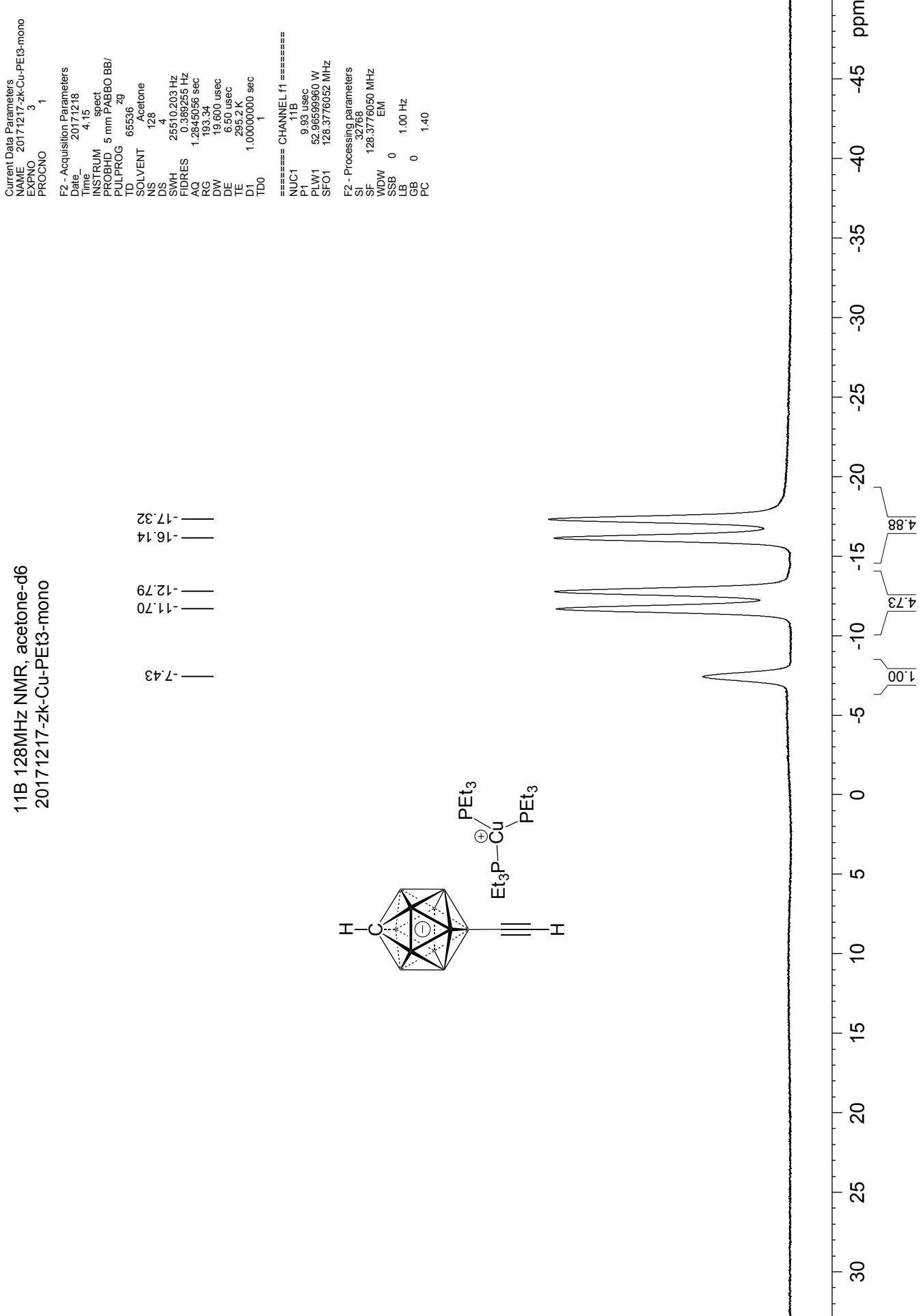


<sup>13</sup>C{<sup>1</sup>H} 500MHz,Acetone-d6  
20171204-2k-Cu-4CF3Py-mono

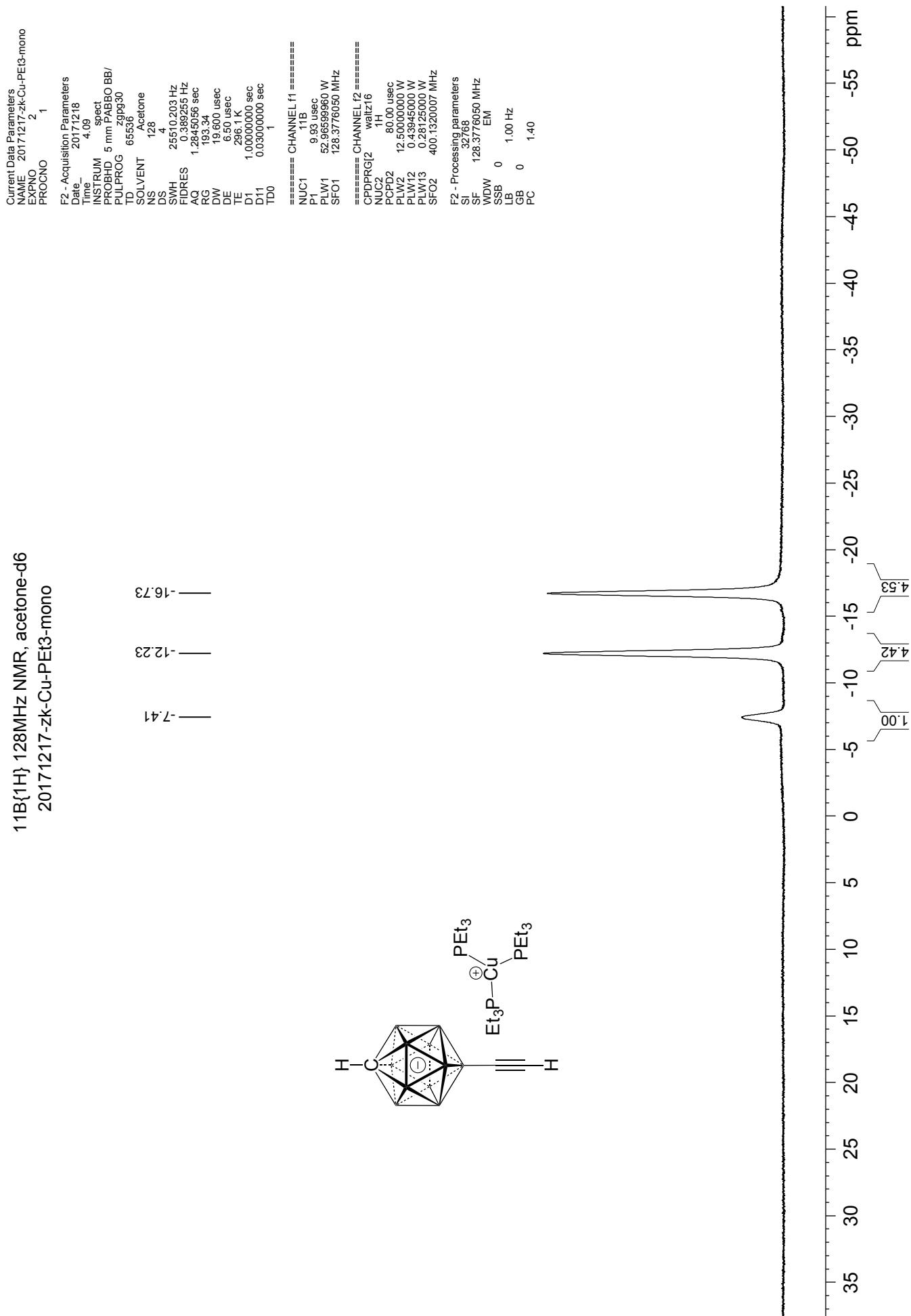


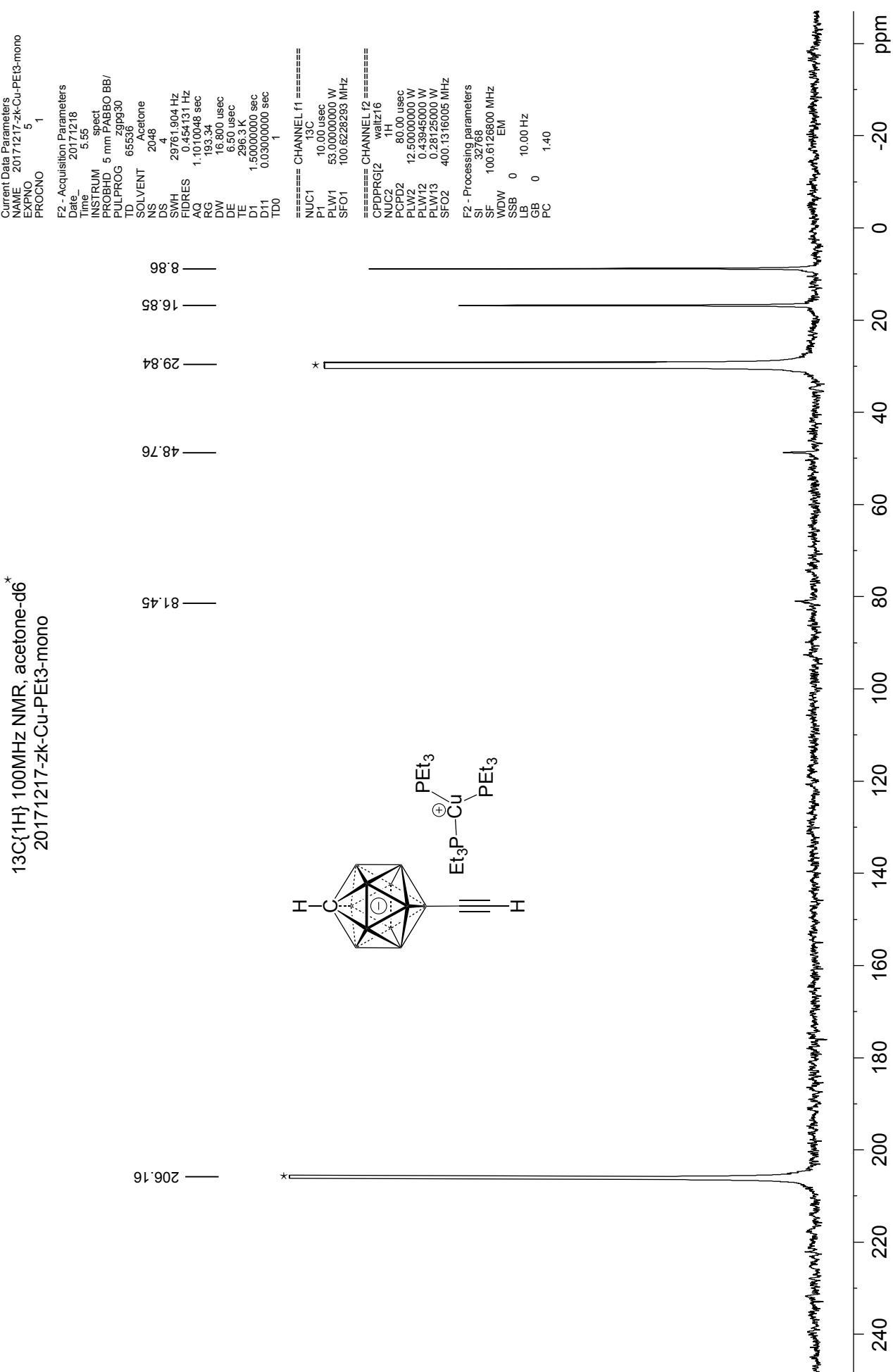


11B 128MHz NMR, acetone-d<sub>6</sub>  
20171217-zk-Cu-Pt3-mono

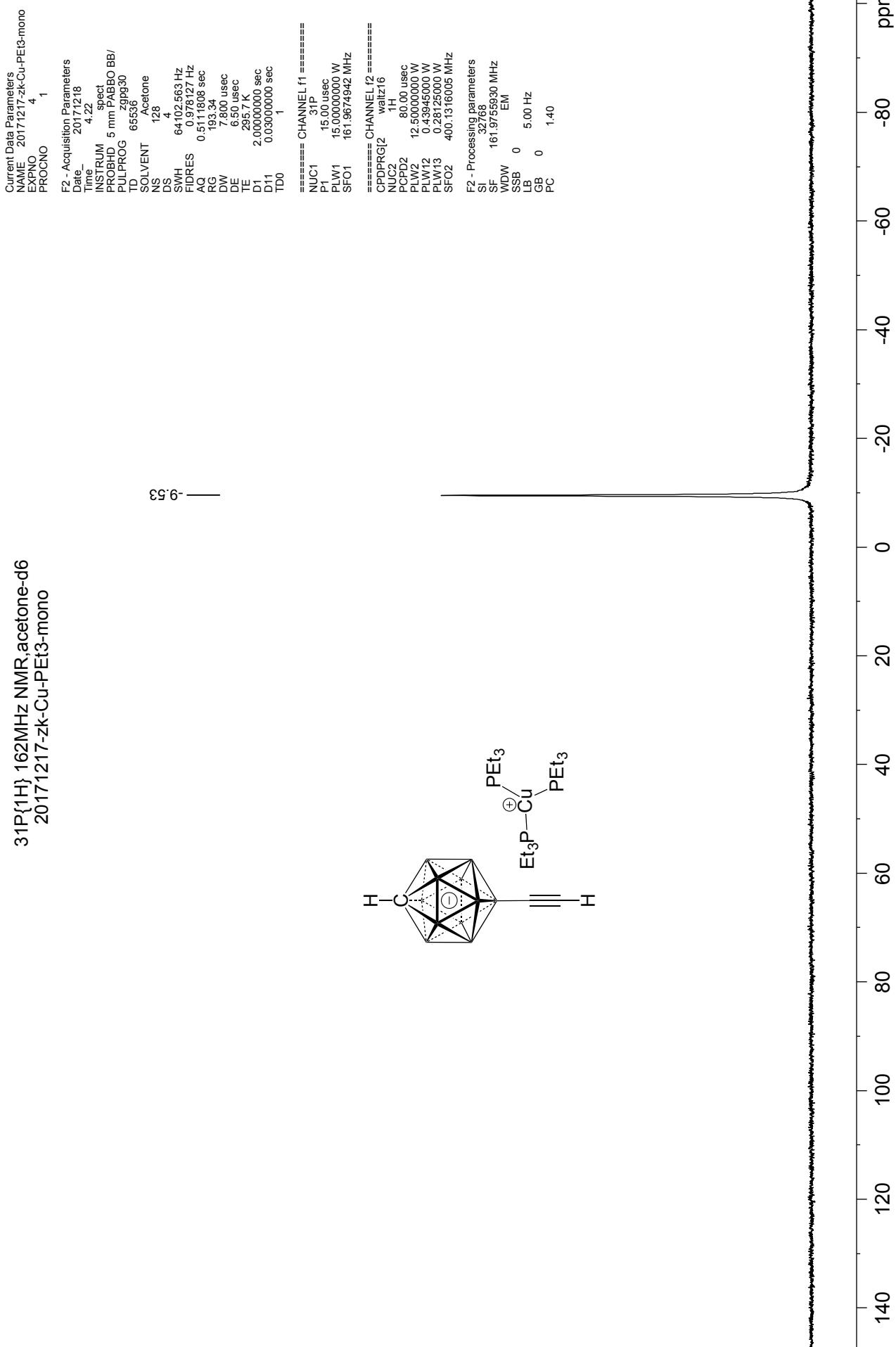


11B{1H} 128MHz NMR, acetone-d6  
20171217-zk-Cu-PEt3-mono

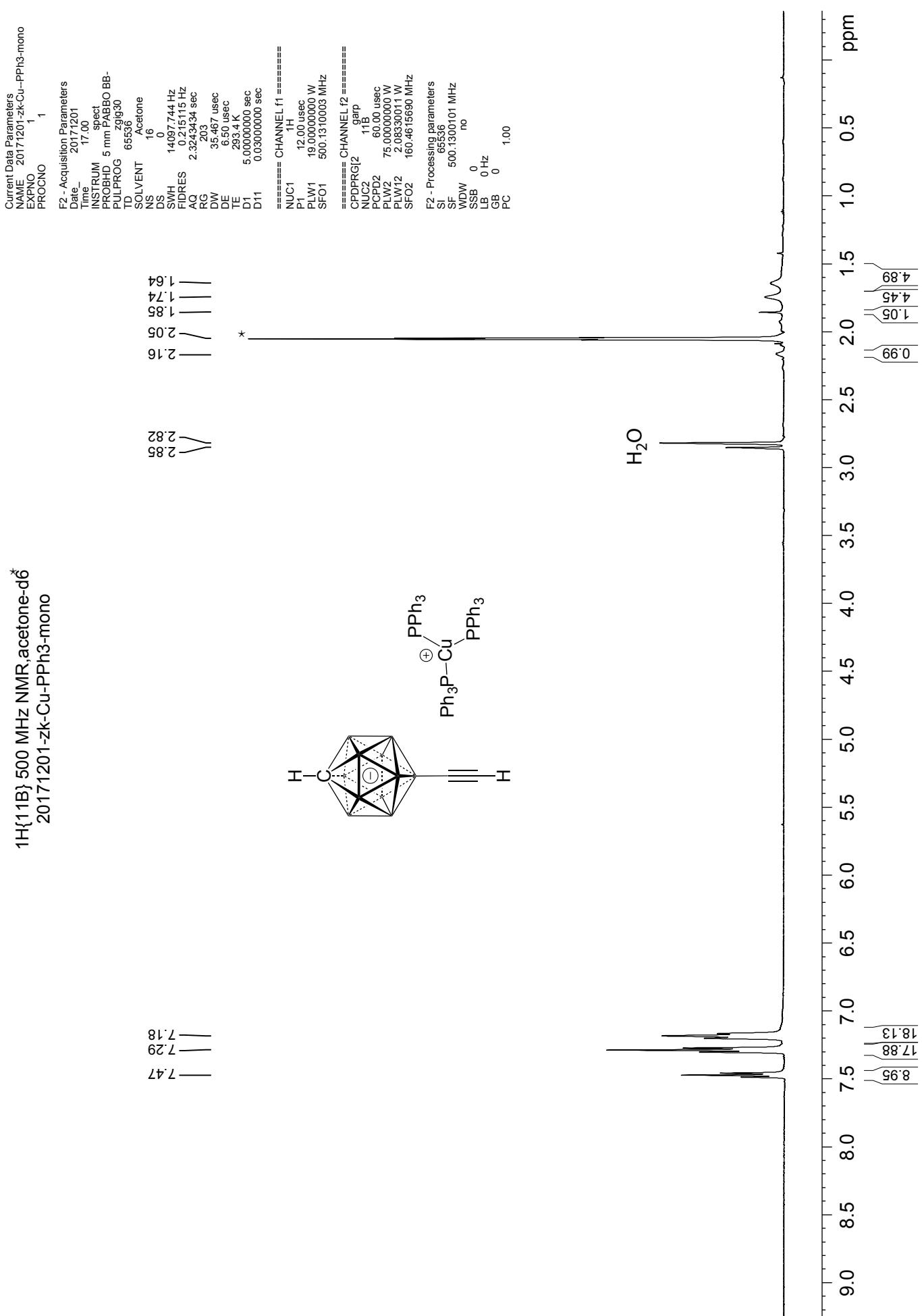




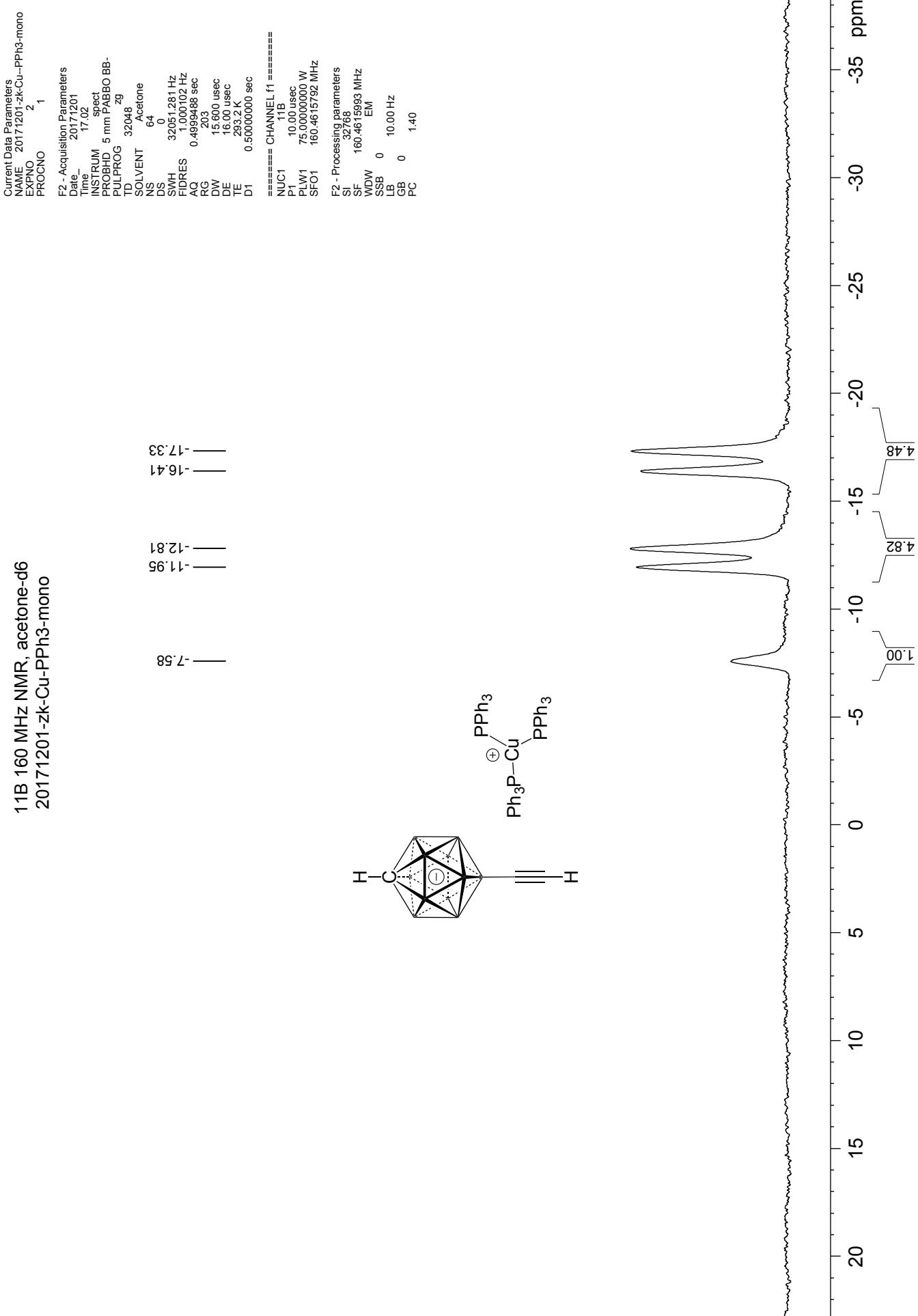
31P{1H} 162MHz NMR,acetone-d6  
20171217-zk-Cu-PEt<sub>3</sub>-mono



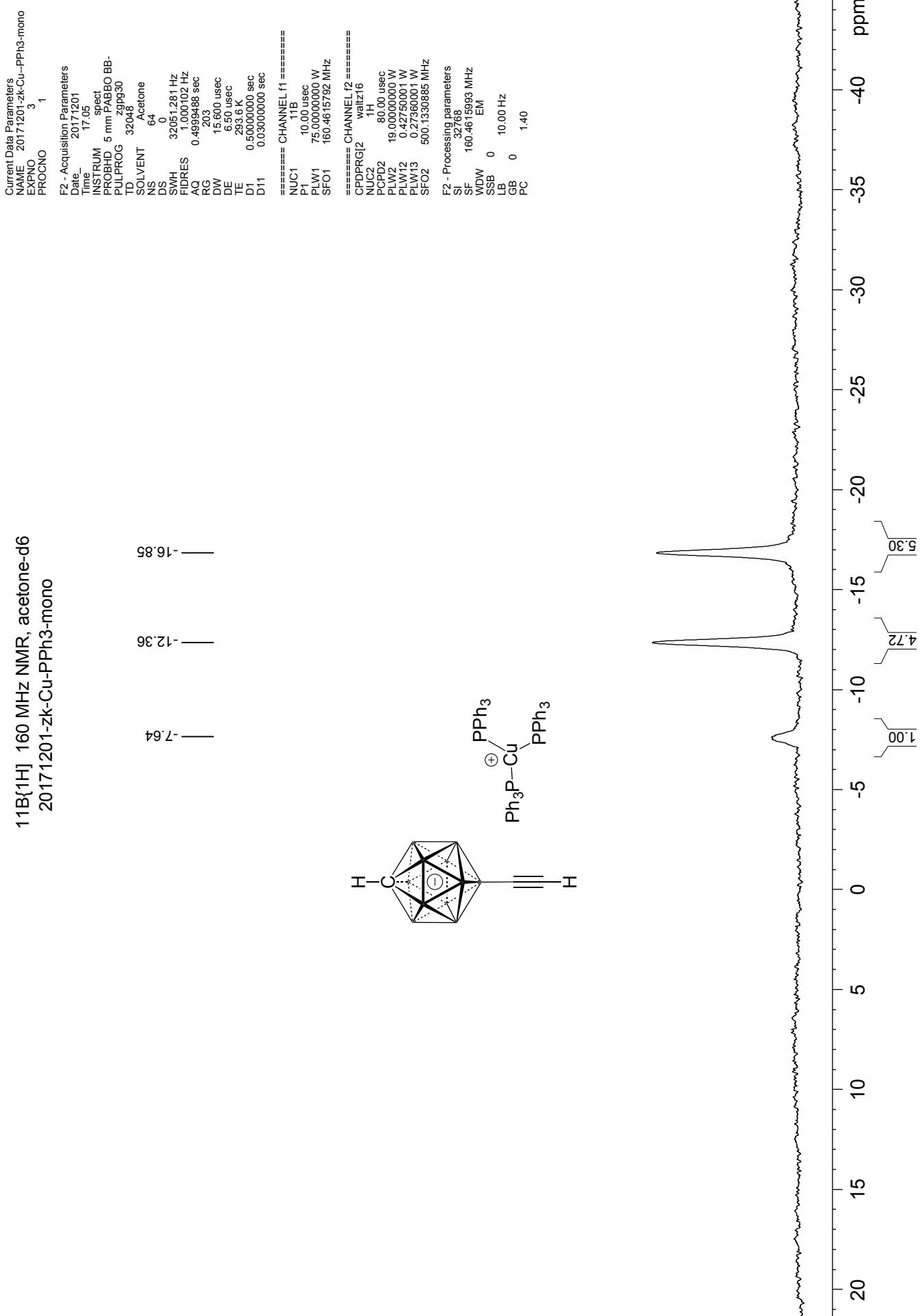
<sup>1</sup>H{<sup>11</sup>B} 500 MHz NMR, acetone-d<sub>6</sub>  
20171201-zk-Cu-PPh<sub>3</sub>-mono

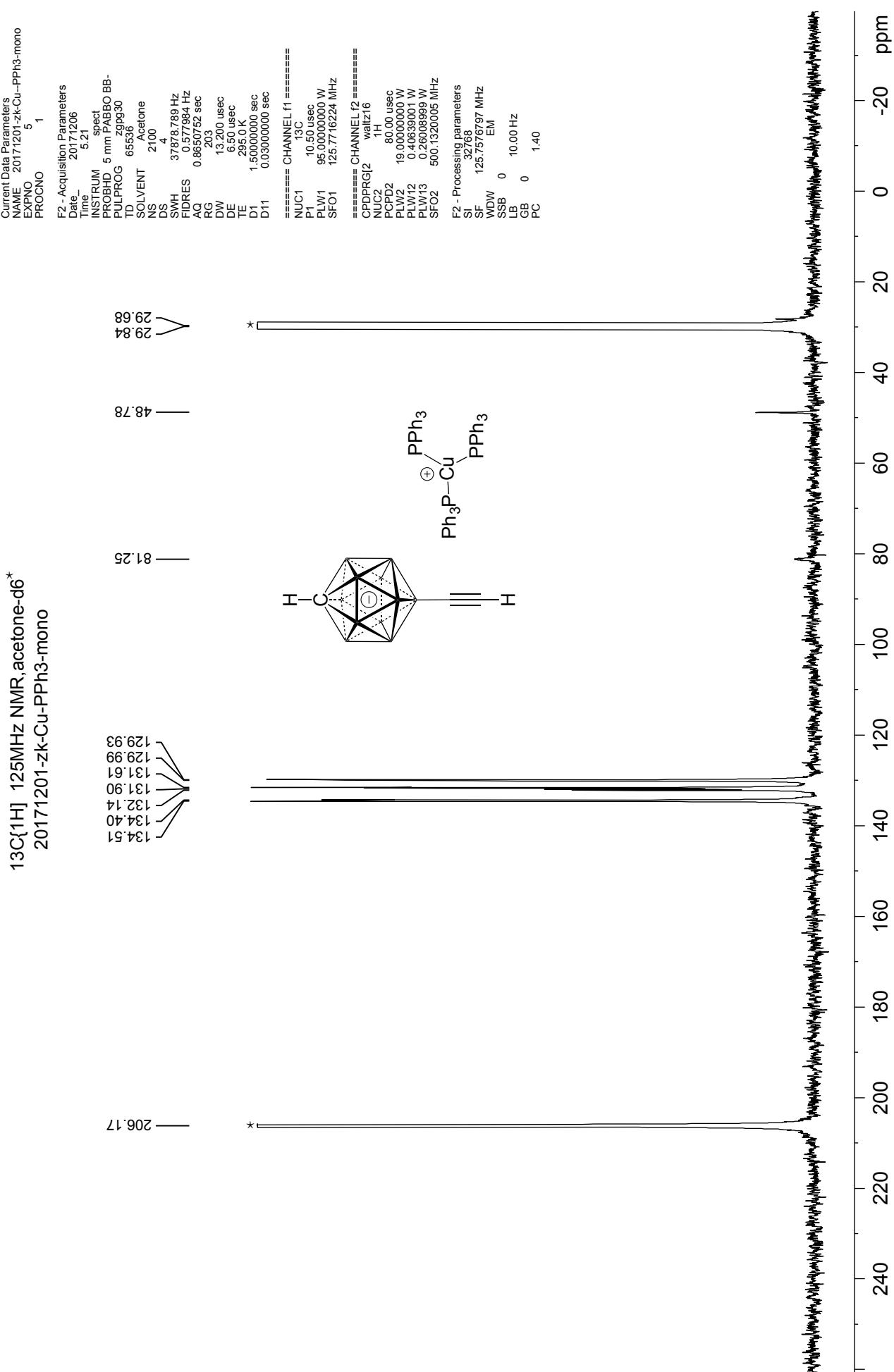


11B 160 MHz NMR, acetone-d<sub>6</sub>  
20171201-zk-Cu-PPh<sub>3</sub>-mono

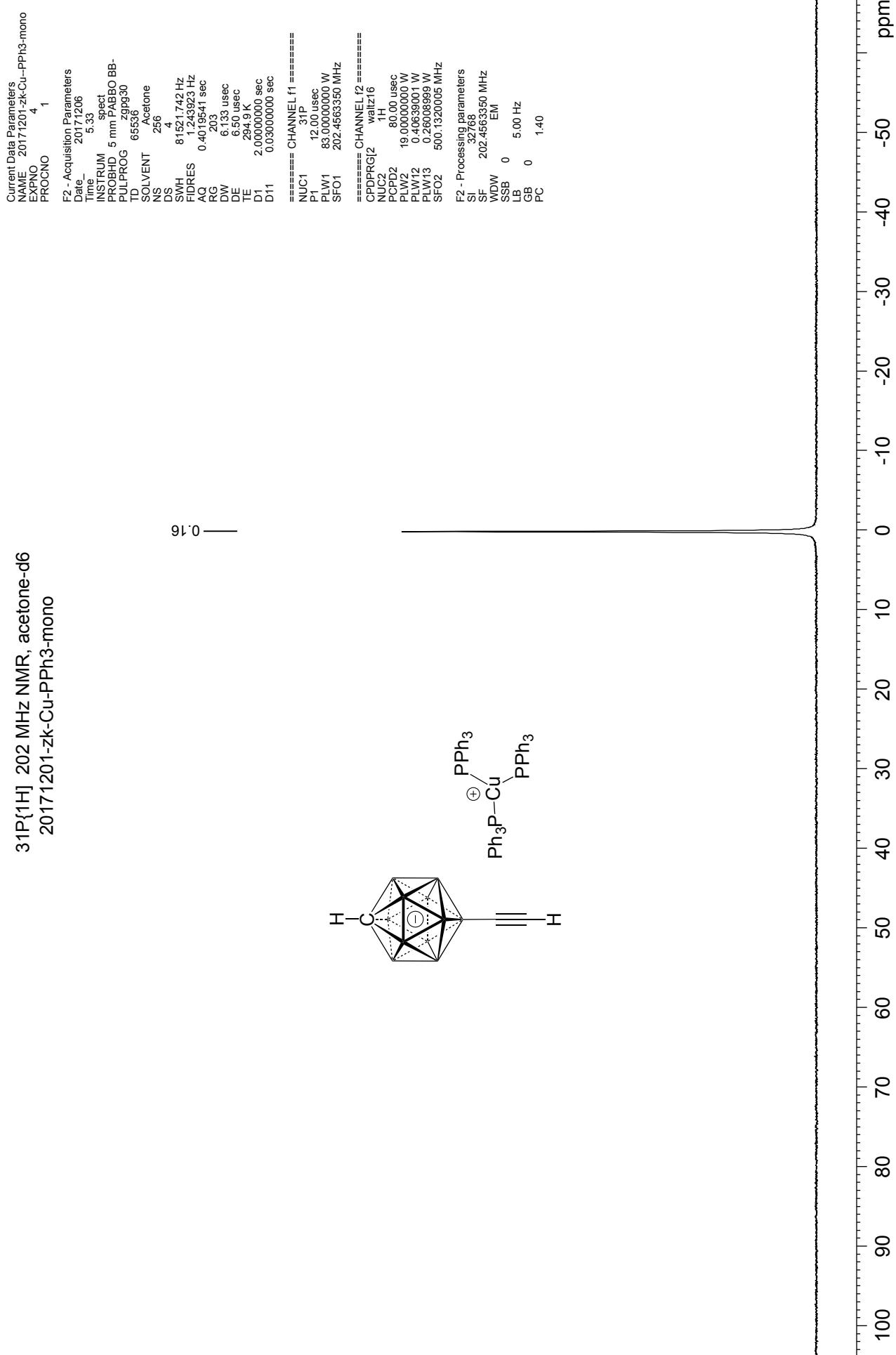


11B{1H} 160 MHz NMR, acetone-d<sub>6</sub>  
20171201-zk-Cu-PPh<sub>3</sub>-mono

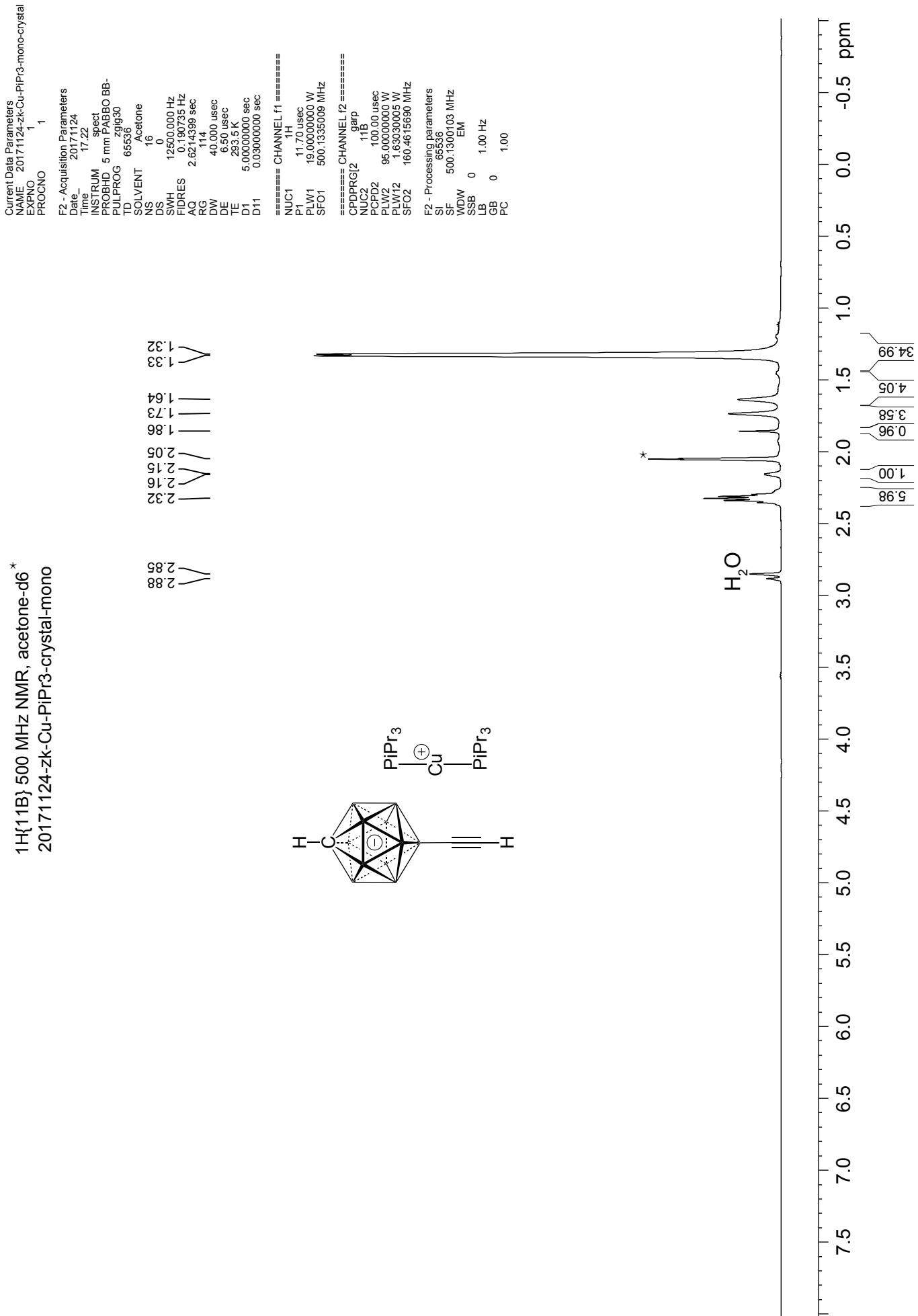




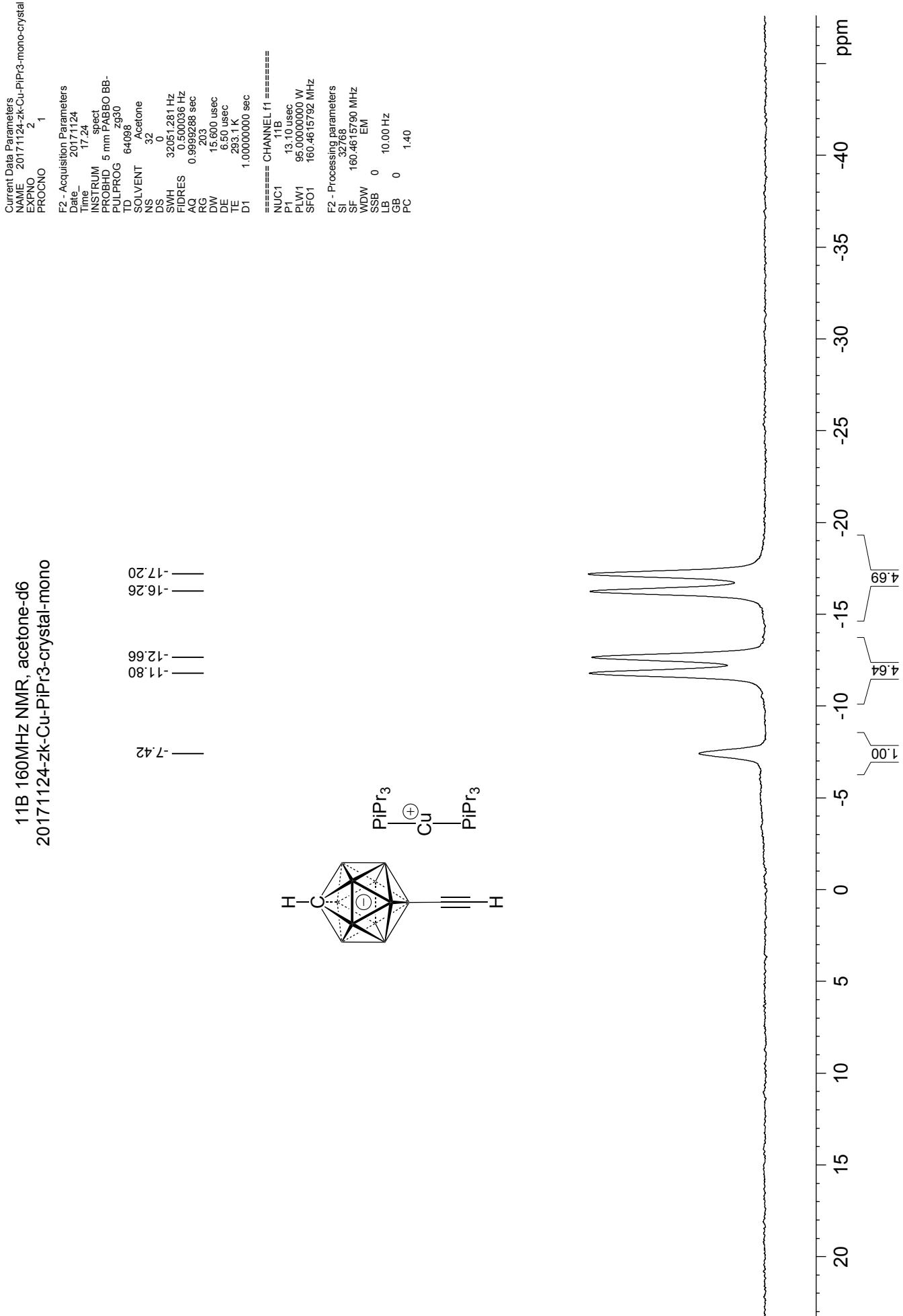
31P{1H] 202 MHz NMR, acetone-d6  
20171201-zk-Cu-PPh3-mono



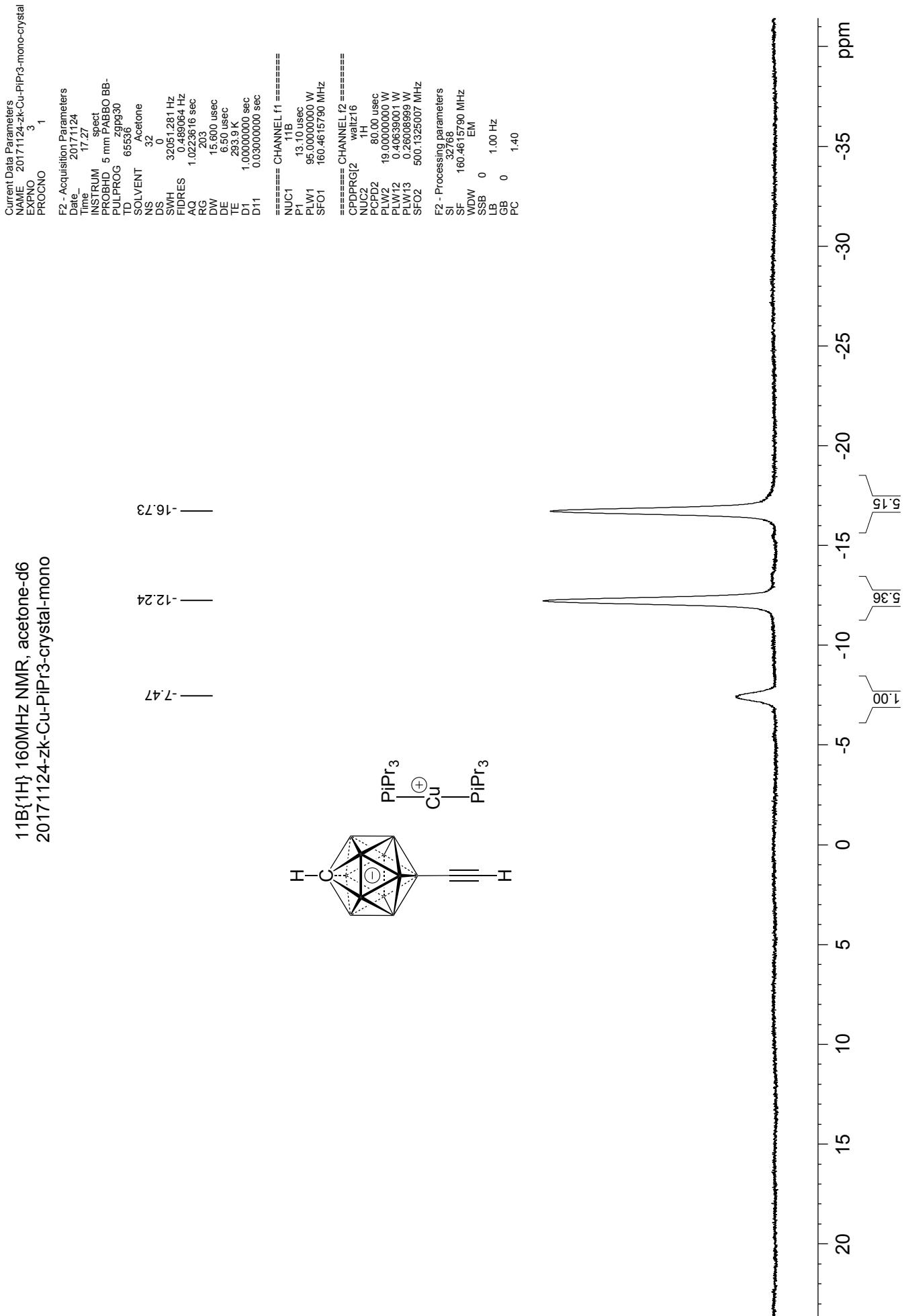
<sup>1</sup>H{<sup>113</sup>B} 500 MHz NMR, acetone-d<sub>6</sub>  
20171124-zk-Cu-PiPr<sub>3</sub>-mono-crystal-mono



11B 160MHz NMR, acetone-d<sub>6</sub>  
20171124-zk-Cu-PIPr<sub>3</sub>-crystal-mono



11B{1H} 160MHz NMR, acetone-d<sub>6</sub>  
20171124-zk-Cu-PiPr<sub>3</sub>-mono



<sup>13</sup>C{<sup>1</sup>H} 125 MHz NMR,acetone-d<sub>6</sub>  
20171124-zk-Cu-PiPr<sub>3</sub>-mono-crystal-mono\*

Current Data Parameters  
NAME 20171124-zk-Cu-PiPr<sub>3</sub>-mono-crystal  
EXPN0 5  
PROCNO 1

F2 - Acquisition Parameters

Date 20171130  
Time 11:36

INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zgpg30  
TD 65536

SOLVENT Acetone  
NS 3100  
DS 4

SWH 3787.789 Hz  
FIDRES 0.577984 Hz

AQ 0.865052 sec  
RG 203

DW 13.200 usec  
DE 6.500 usec

TE 297.1 K  
D1 1.5000000 sec

D11 0.0300000 sec

===== CHANNEL f1 =====

NUC1 13C  
P1 10.50usec  
PLW1 95.0000000 W

SFO1 125.7716224 MHz

===== CHANNEL f2 =====

CPDFRG12 waltz16  
NUC2 1H

PCPD2 80.00 usec  
PLW2 19.0000000 W

PLW12 0.40633001 W

PLW13 0.26008999 W

SFO2 500.1320005 MHz

F2 - Processing parameters

S1 32768  
SF 125.7576770 MHz

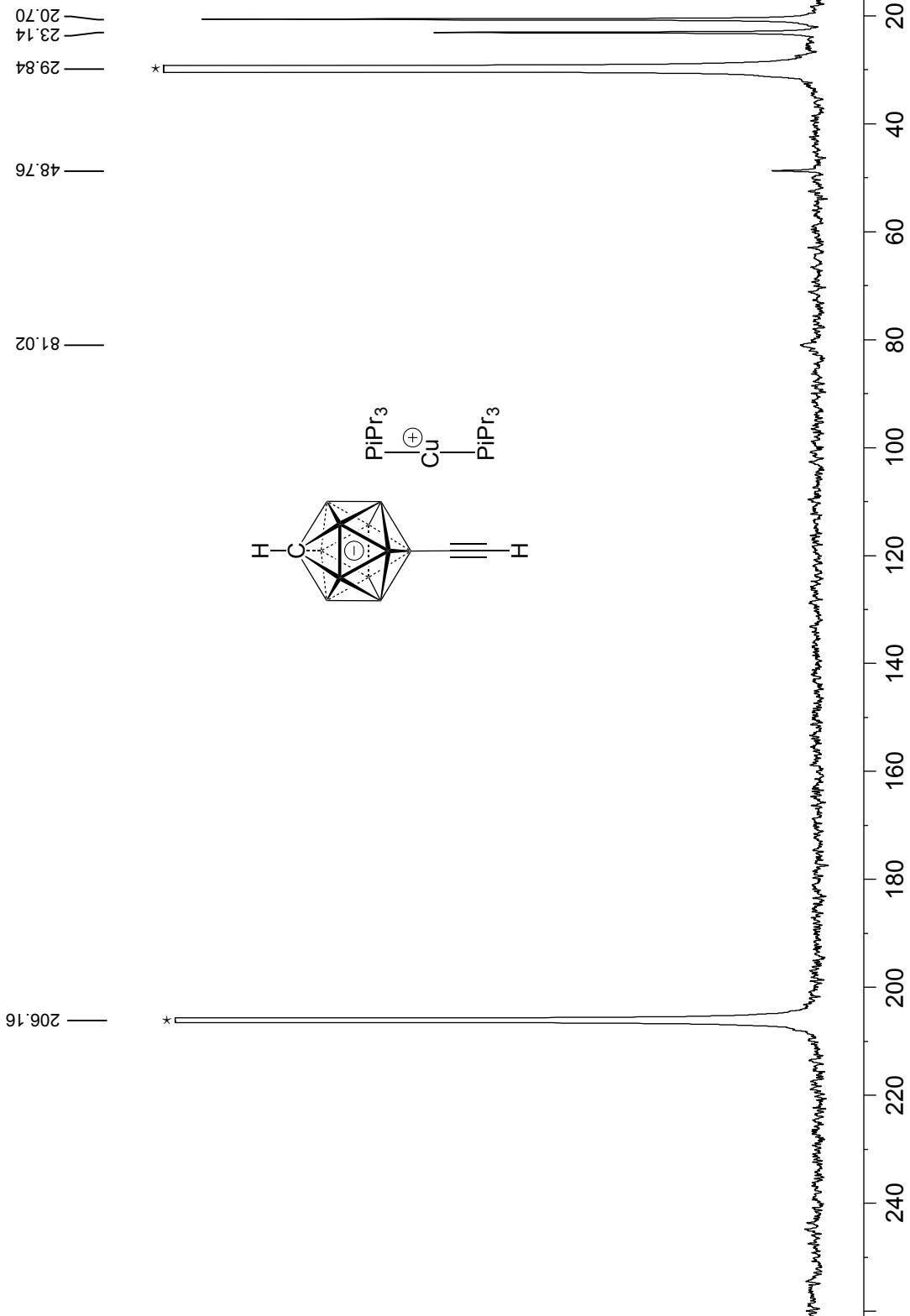
WDW EM

SSB 0

LB 20.00 Hz

GB 0

PC 1.40



$^{31}\text{P}\{\text{1H}\}$  202 MHz NMR, acetone-d<sub>6</sub>  
20171124-Zk-Cu-PiPr<sub>3</sub>-mono-crystal-mono

Current Data Parameters  
NAME 20171124-Zk-Cu-PiPr3-mono-crystal  
EXPT 4  
PROCNO 1

F2 - Acquisition Parameters

Date 20171124  
Time 17:49  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zgprg30  
TD 65536  
SOLVENT Acetone  
NS 512  
DS 4  
SWH 81.621742 Hz  
FIDRES 1.243923 Hz  
AQ 0.4019841 sec  
RG 203  
DW 6.133 usec  
DE 6.50 usec  
TE 293.4 K  
D1 2.000000 sec  
D11 0.0300000 sec

===== CHANNEL f1 =====

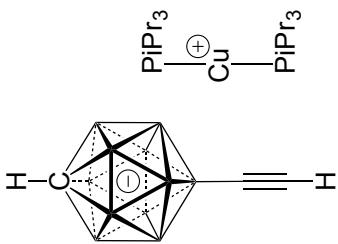
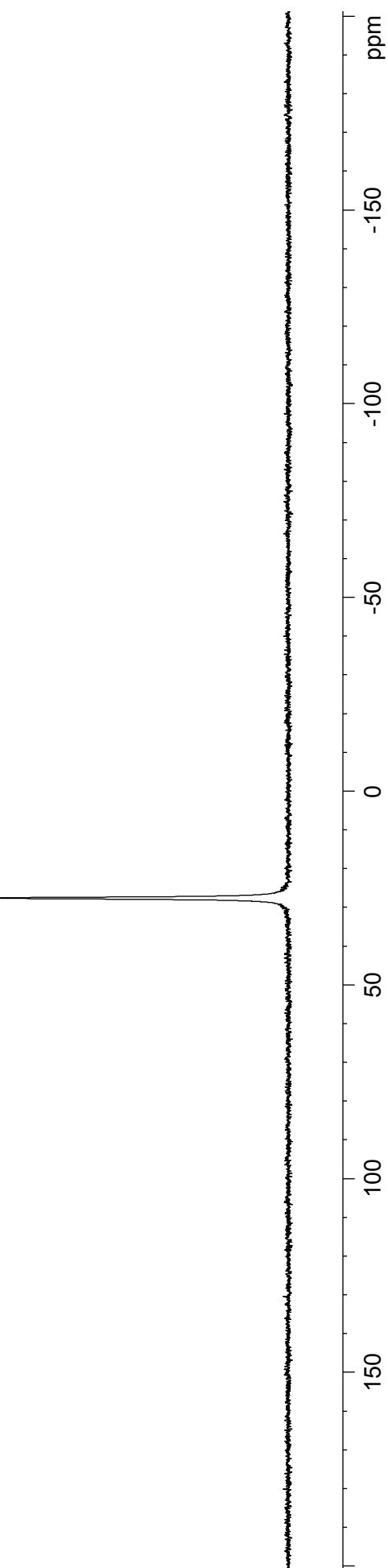
NUC1 31P  
P1 12.00usec  
PLW1 83.0000000 W  
SF01 202.4563350 MHz

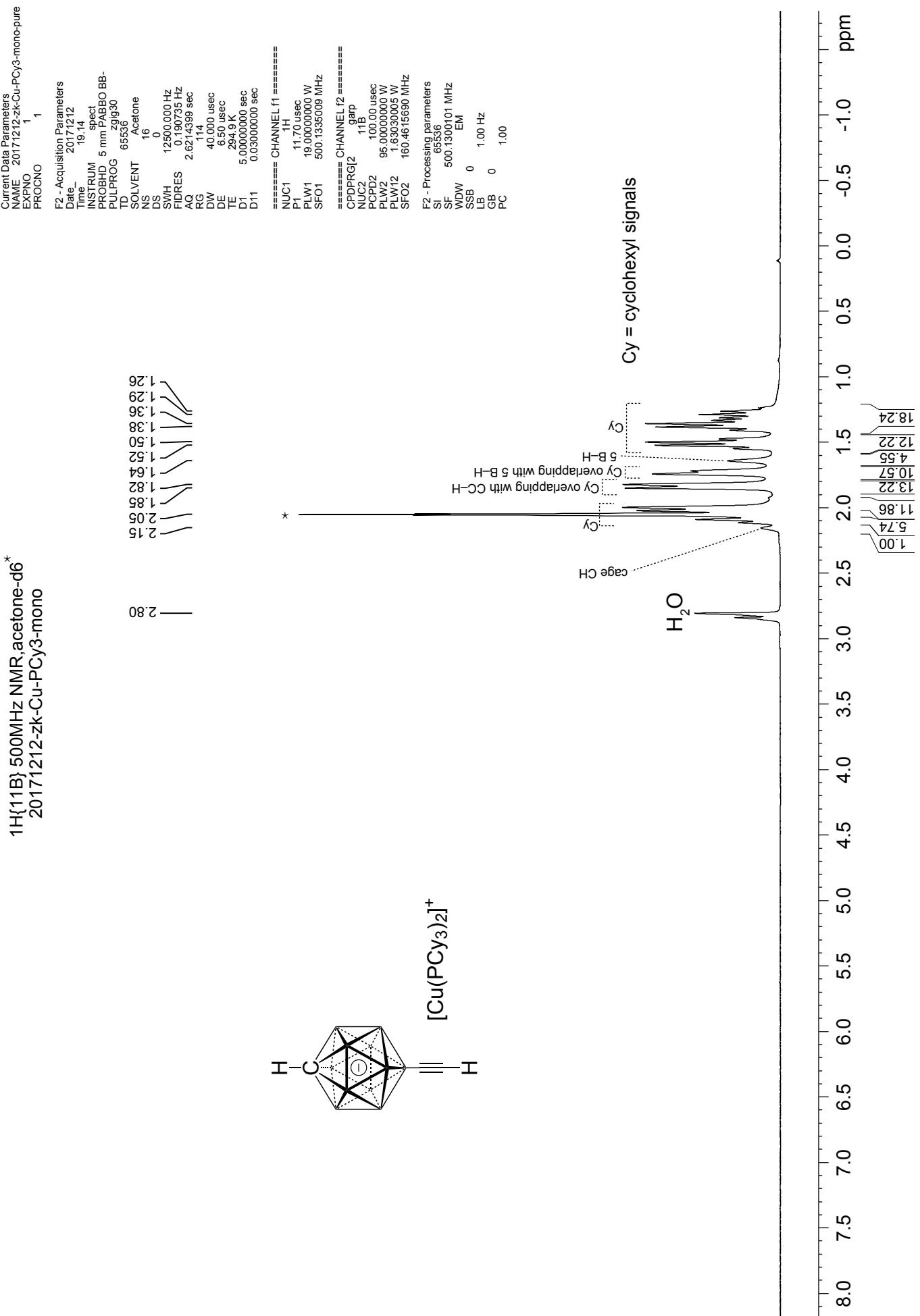
===== CHANNEL f2 =====

CPDFRG12 wait16  
NUC2 1H  
PCPND2 80.00 usec  
PLW2 19.0000000 W  
PLW12 0.40633001 W  
PLW13 0.26008999 W  
SF02 500.1320005 MHz

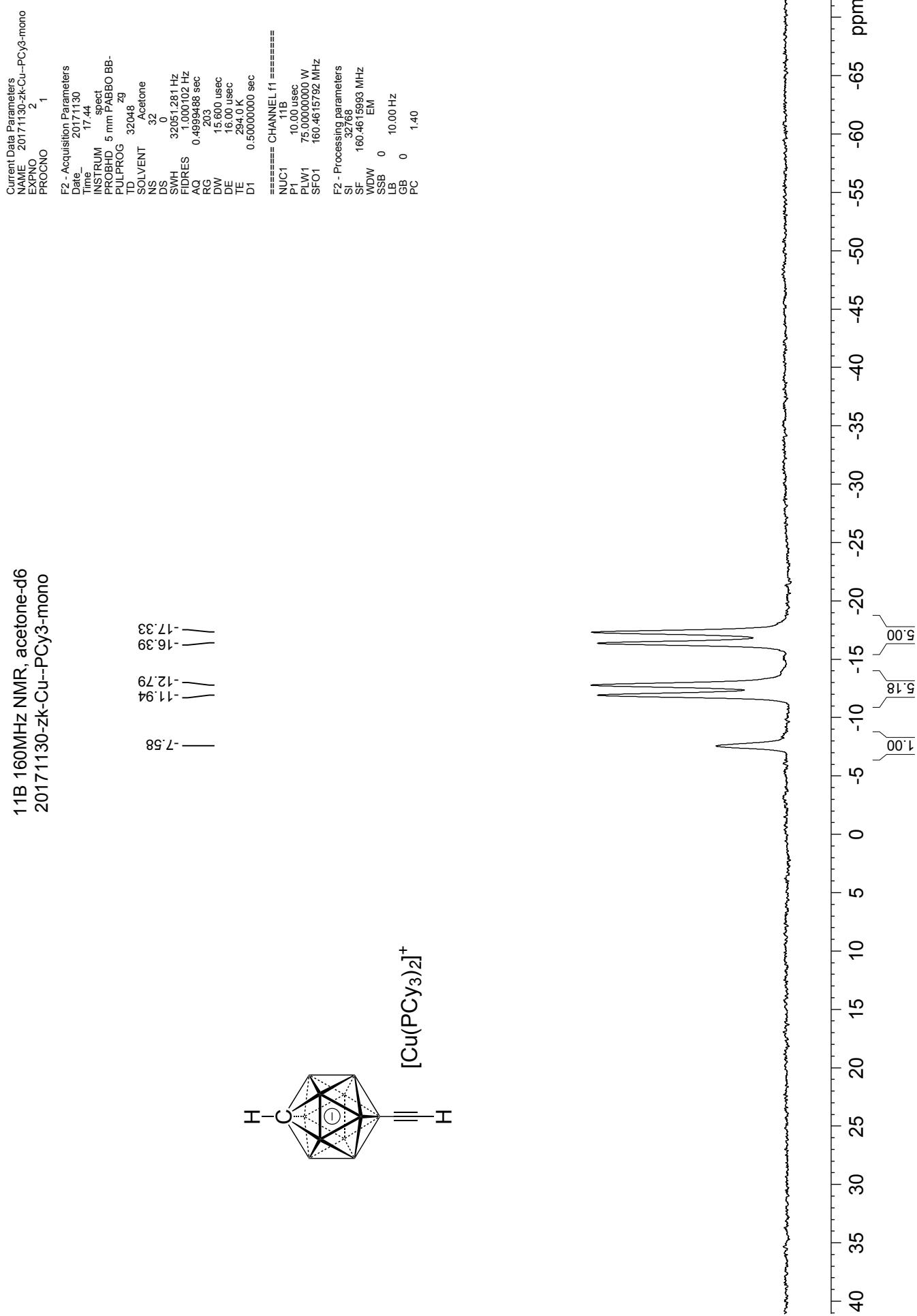
F2 - Processing parameters

S1 32768  
SF 202.4563350 MHz  
WDW EM  
SSB 0  
LB 10.00 Hz  
GB 0  
PC 1.40

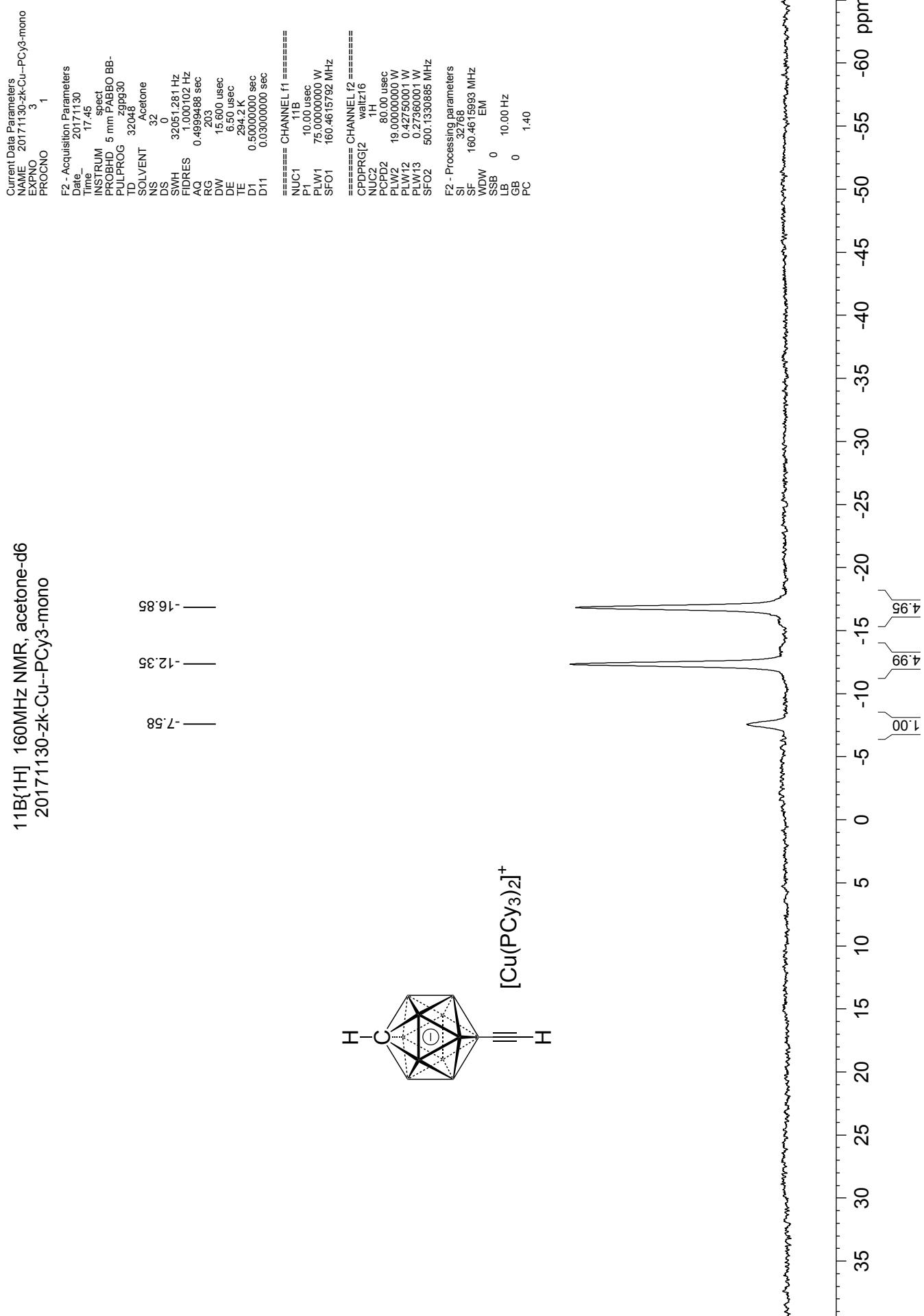


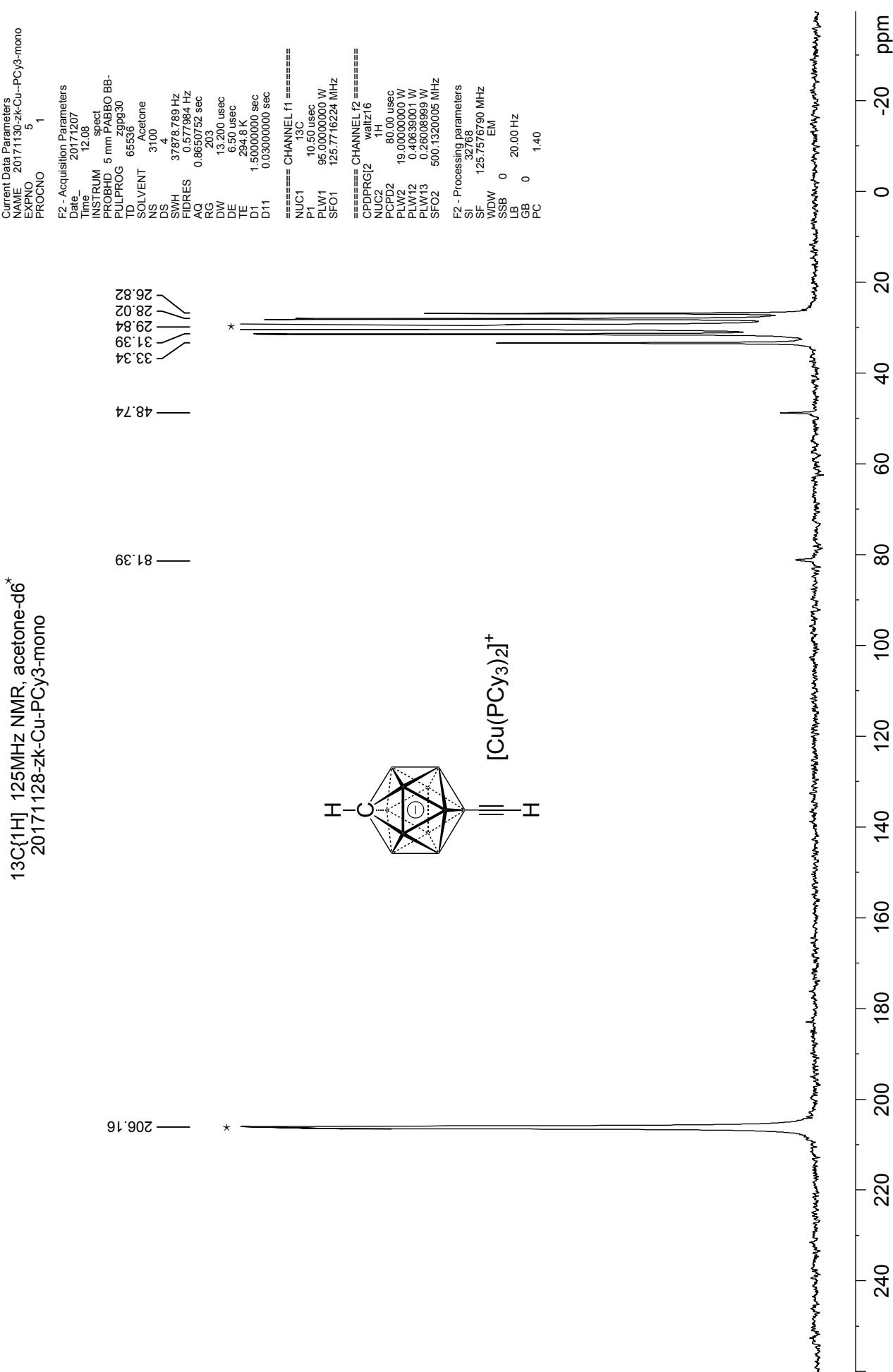


11B 160MHz NMR, acetone-d<sub>6</sub>  
20171130-Zk-Cu-PCy<sub>3</sub>-mono

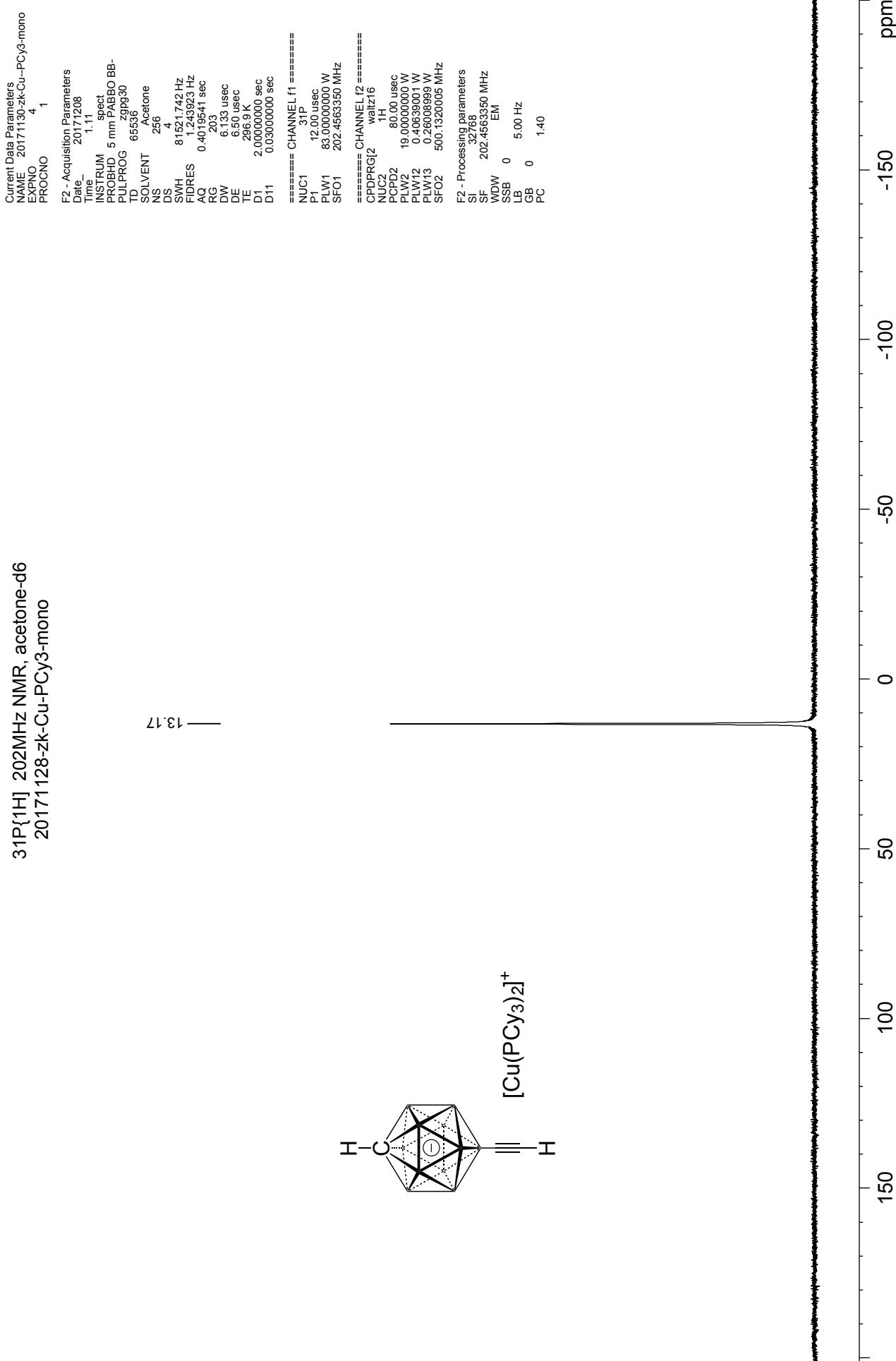


11B{<sup>1</sup>H} 160MHz NMR, acetone-d<sub>6</sub>  
20171130-Zk-Cu--PCy<sub>3</sub>-mono





31P{1H} 202MHz NMR, acetone-d6  
20171128-zk-Cu-PCy3-mono



<sup>\*</sup>  $^1\text{H}$ {11B} 400 MHz NMR, acetone-d<sub>6</sub>  
20180530-zk-Pd-PEt<sub>3</sub>-HNEt<sub>2</sub>

Current Data Parameters  
 NAME 20180530-zk-Pd-PEt<sub>3</sub>-HNEt<sub>2</sub>  
 EXPNO 1  
 PROCN0

F2 - Acquisition Parameters  
 Date 20180531  
 Time 1.23  
 INSTRUM spect  
 PROBHD 5 mm PABBO BB/  
 PULPROG zgig30  
 TD 16384  
 SOLVENT Acetone  
 NS 32  
 DS 4  
 SWH 8012.820 Hz  
 FIDRES 0.489064 Hz  
 AQ 1.0223616 sec  
 RG 193.34  
 DW 62.400 usec  
 DE 6.50 usec  
 TE 293.4 K  
 D1 1.0000000 sec  
 D11 0.0300000 sec  
 TDO 1

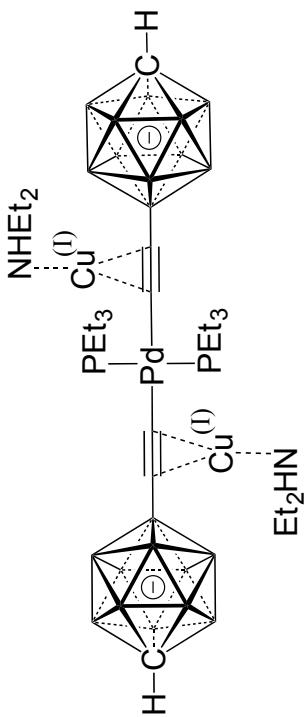
===== CHANNEL f1 ======  
 NUC1 1H  
 P1 15.00 usec  
 PLW1 12.5000000 W  
 SFO1 400.1320007 MHz

===== CHANNEL f2 ======  
 CPDPRG12 garp4  
 NUC2 11B  
 PCPD2 90.00 usec  
 PLW2 52.9659960 W  
 PLW12 0.6447798 W  
 SFO2 128.3776056 MHz

F2 - Processing parameters  
 SF 32768  
 WWDW 400.130074 MHz  
 SSB 0 EM  
 LB 1.00 Hz  
 GB 0  
 PC 1.40

\* Because of the overlap with the solvent residual signal, half of the signal was chosen for integration.

**Chemical Structure:**



11B NMR, 128 MHz NMR, acetone-d<sub>6</sub>  
 20180530-zk-Pd-PEt<sub>3</sub>-HNEt<sub>2</sub>

Current Data Parameters  
 NAME 20180530-zk-Pd-PEt<sub>3</sub>-HNEt<sub>2</sub>  
 EXPNO 3  
 PROCN0 1

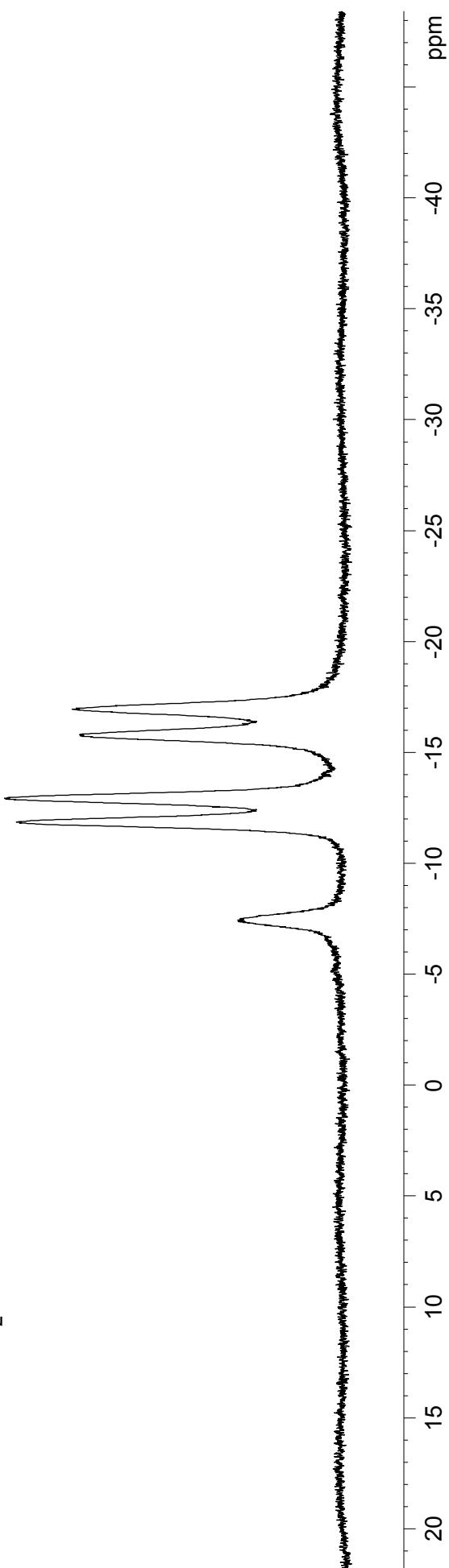
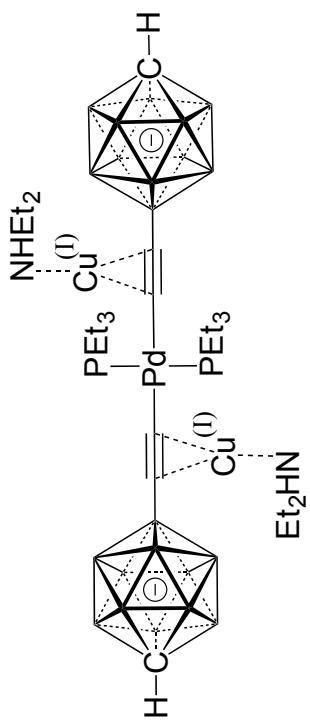
F2 - Acquisition Parameters  
 Date 20180531  
 Time 1.30  
 INSTRUM spect  
 PROBHD 5 mm PABBO BB/  
 PULPROG 2g  
 TD 65536  
 SOLVENT Acetone  
 NS 64  
 DS 4  
 SWH 25510.203 Hz  
 FIDRES 0.389255 Hz  
 AQ 1.2845056 sec  
 RG 193.34  
 DW 19.600 usec  
 DE 6.500 usec  
 TE 293.4 K  
 D1 1.0000000 sec  
 TDO 1

===== CHANNEL f1 =====

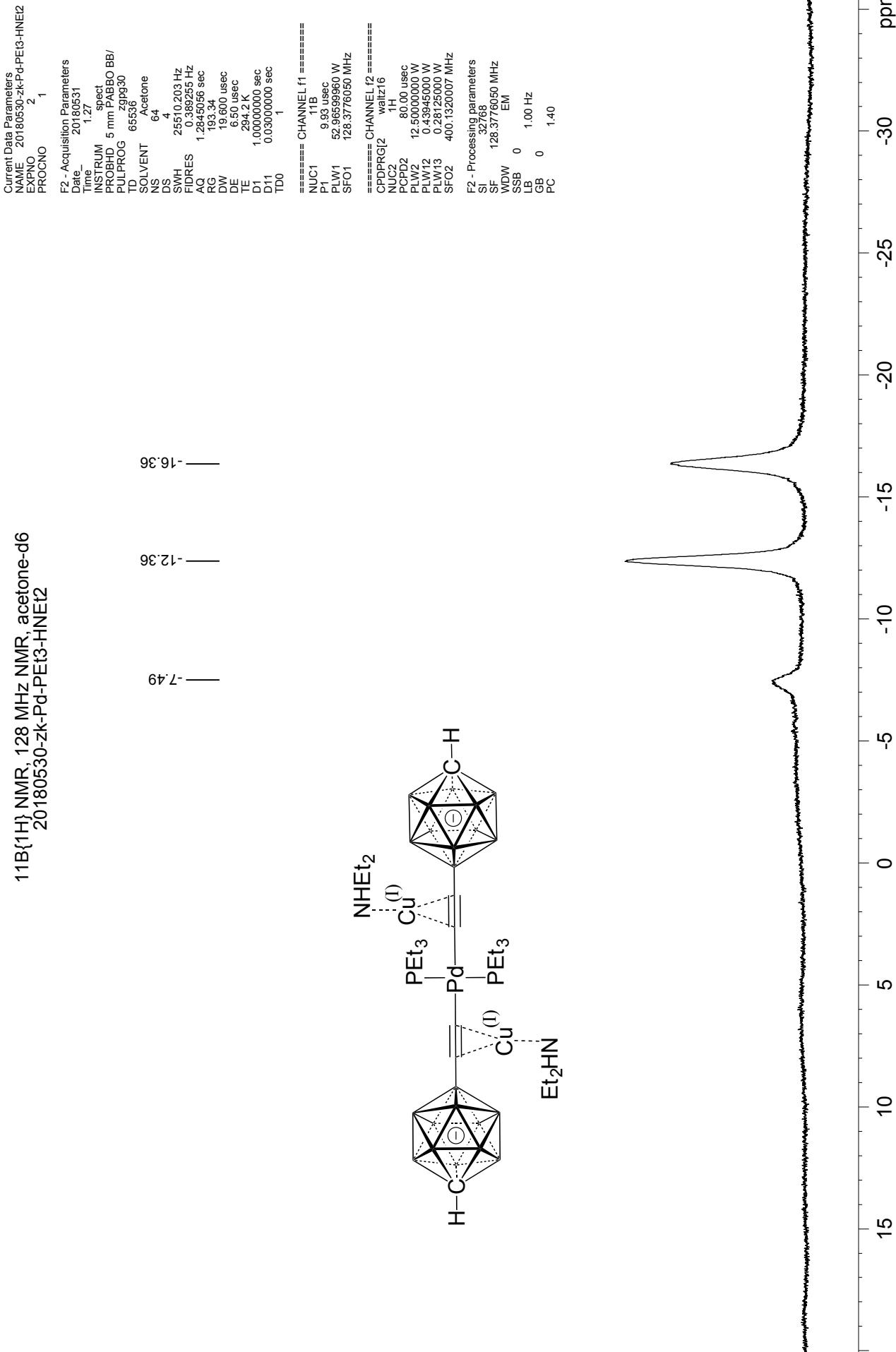
NUC1 11B  
 P1 9.93 usec  
 PLW1 52.98599960 W  
 SFO1 128.3776052 MHz

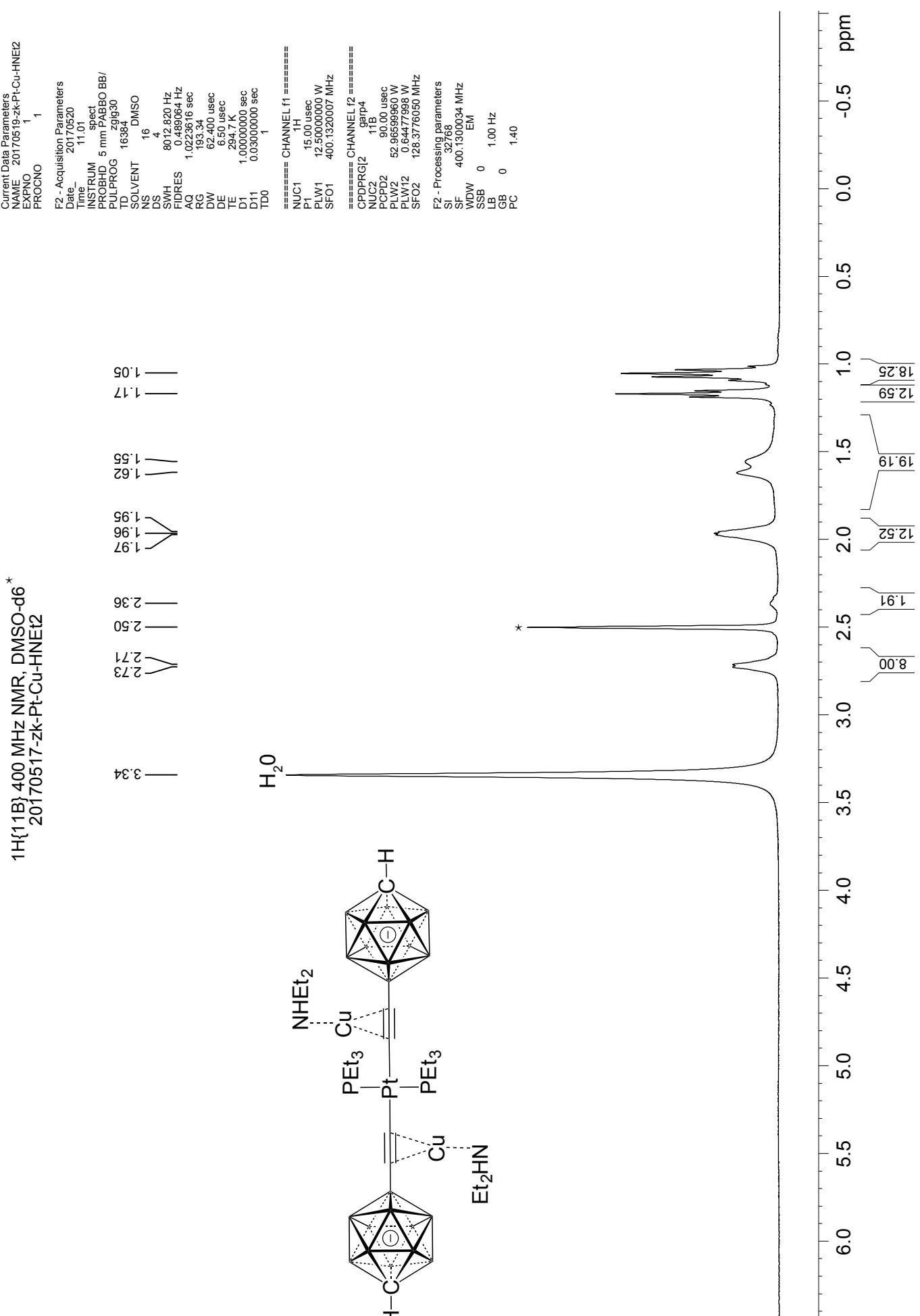
F2 - Processing parameters  
 SI 32768  
 SF 128.3776050 MHz  
 WDW EM  
 SSB 0  
 LB 1.00 Hz  
 GB 0  
 PC 1.40

— -16.96  
 — -15.83  
 — -12.94  
 — -11.88  
 — -7.45

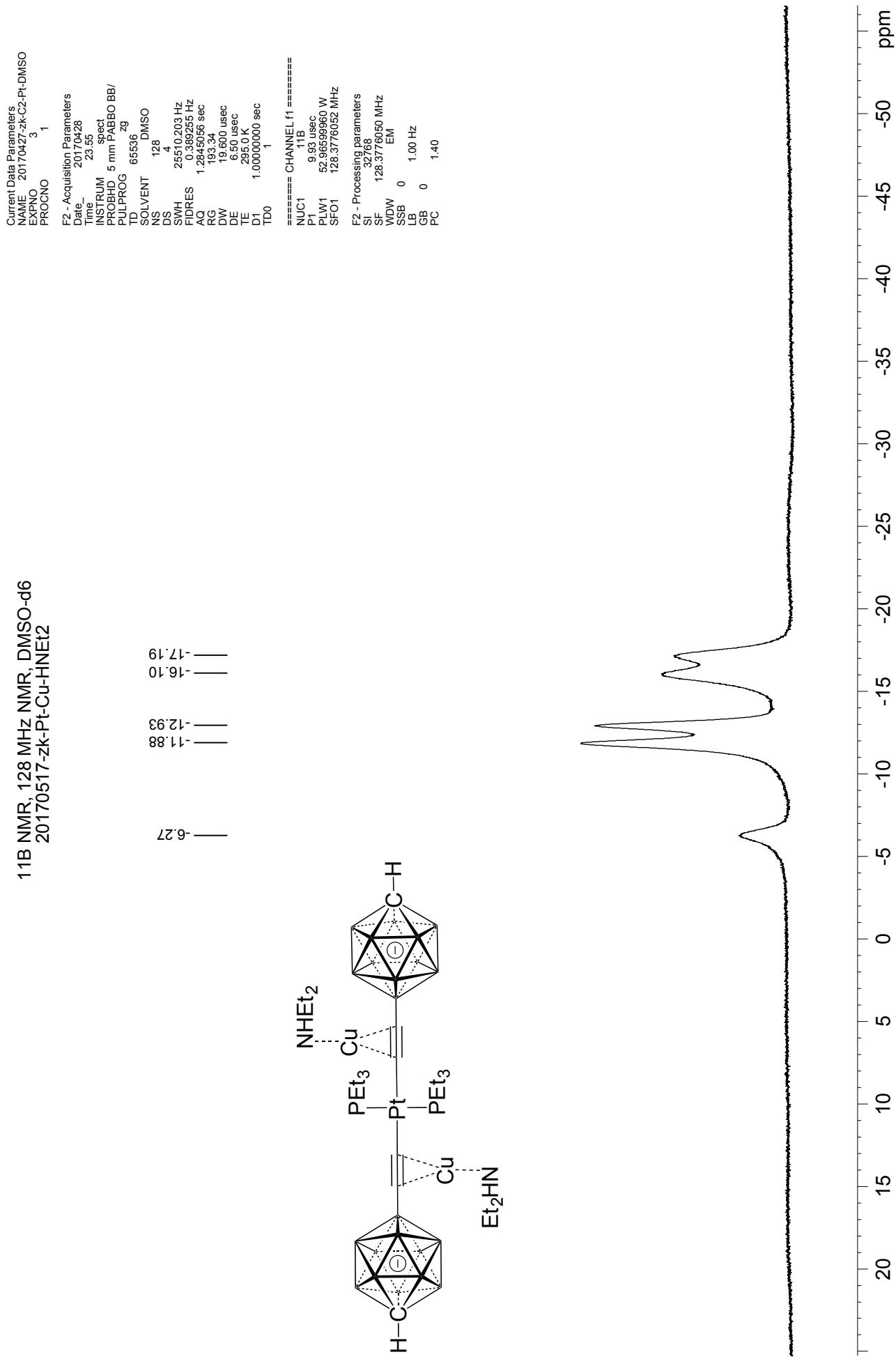


11B{<sup>1</sup>H} NMR, 128 MHz NMR, acetone-d<sub>6</sub>  
20180530-zk-Pd-PEt<sub>3</sub>-HNEt<sub>2</sub>

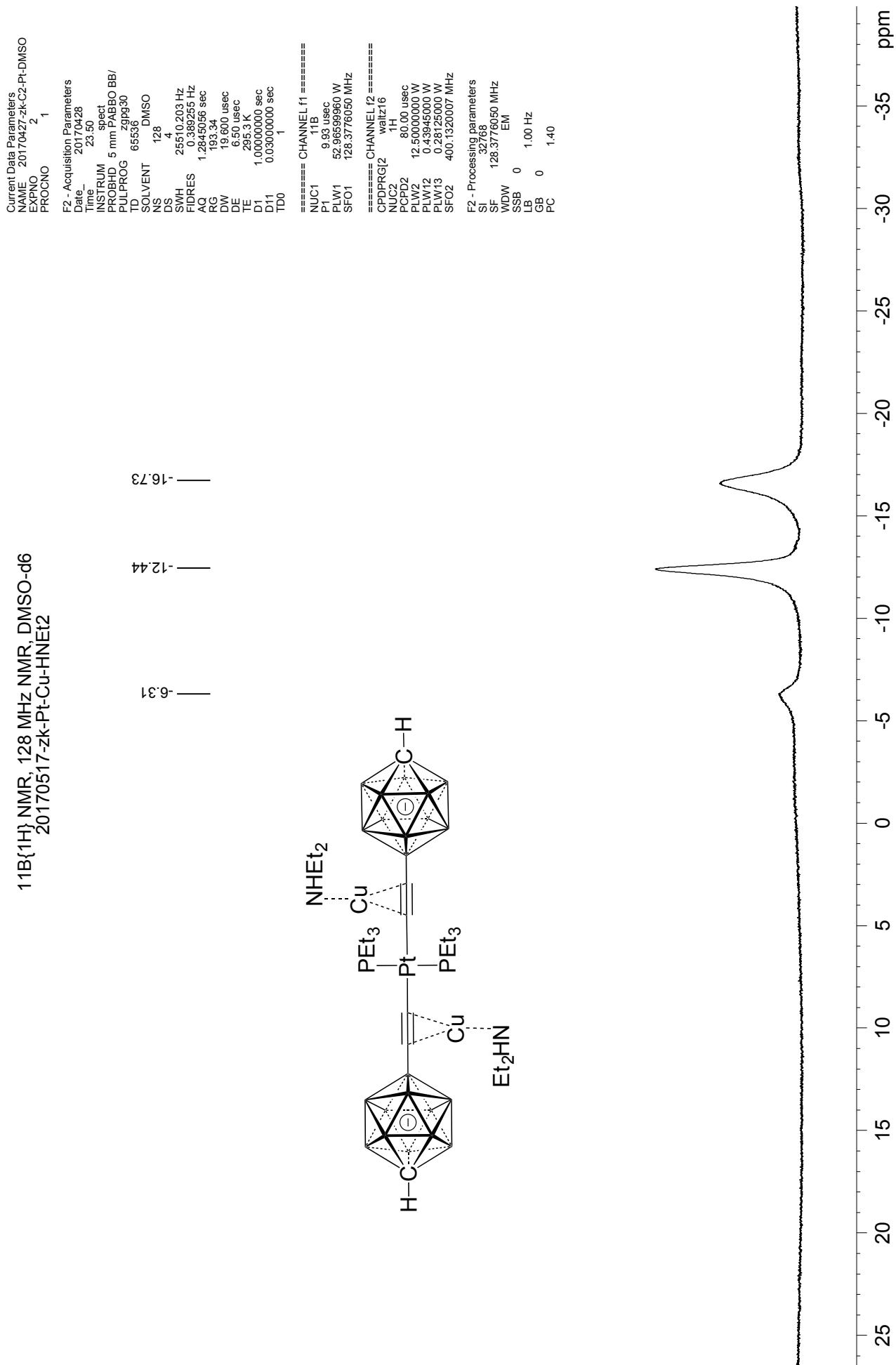




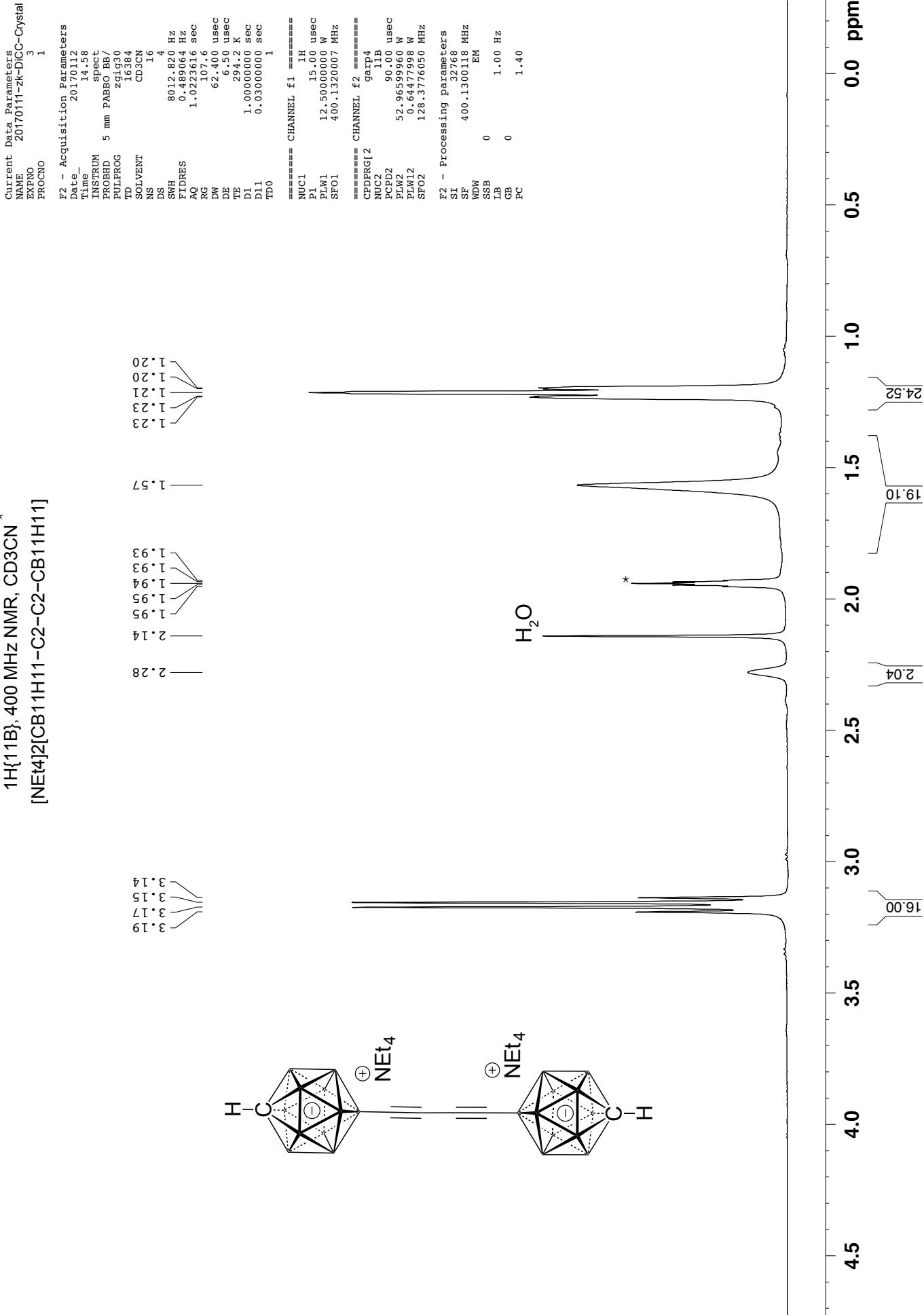
11B NMR, 128 MHz NMR, DMSO-d6  
20170517-zk-Pt-Cu-HNEt<sub>2</sub>



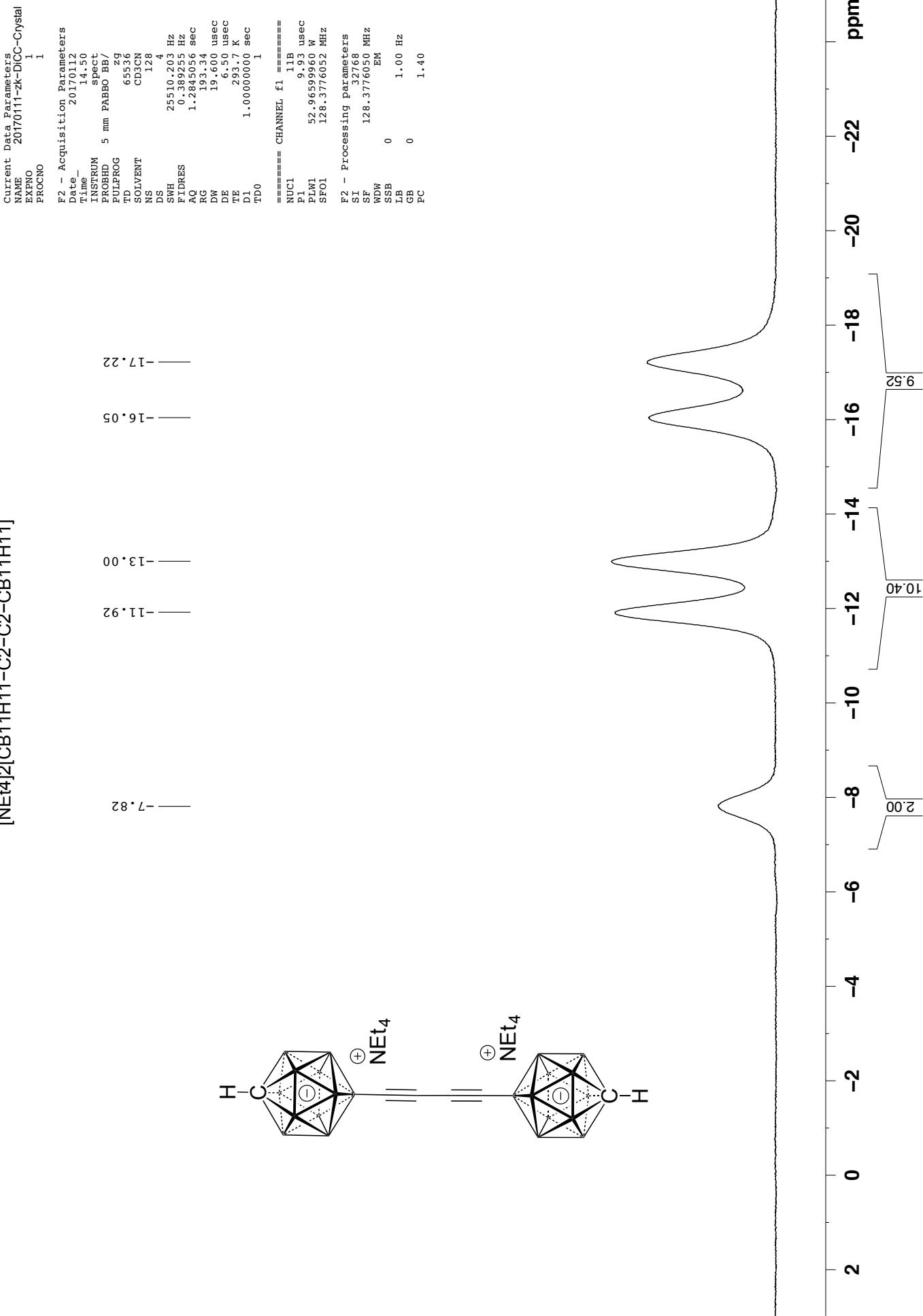
11B{<sup>1</sup>H} NMR, 128 MHz NMR, DMSO-d<sub>6</sub>  
20170517-zk-Pt-Cu-HNEt<sub>2</sub>



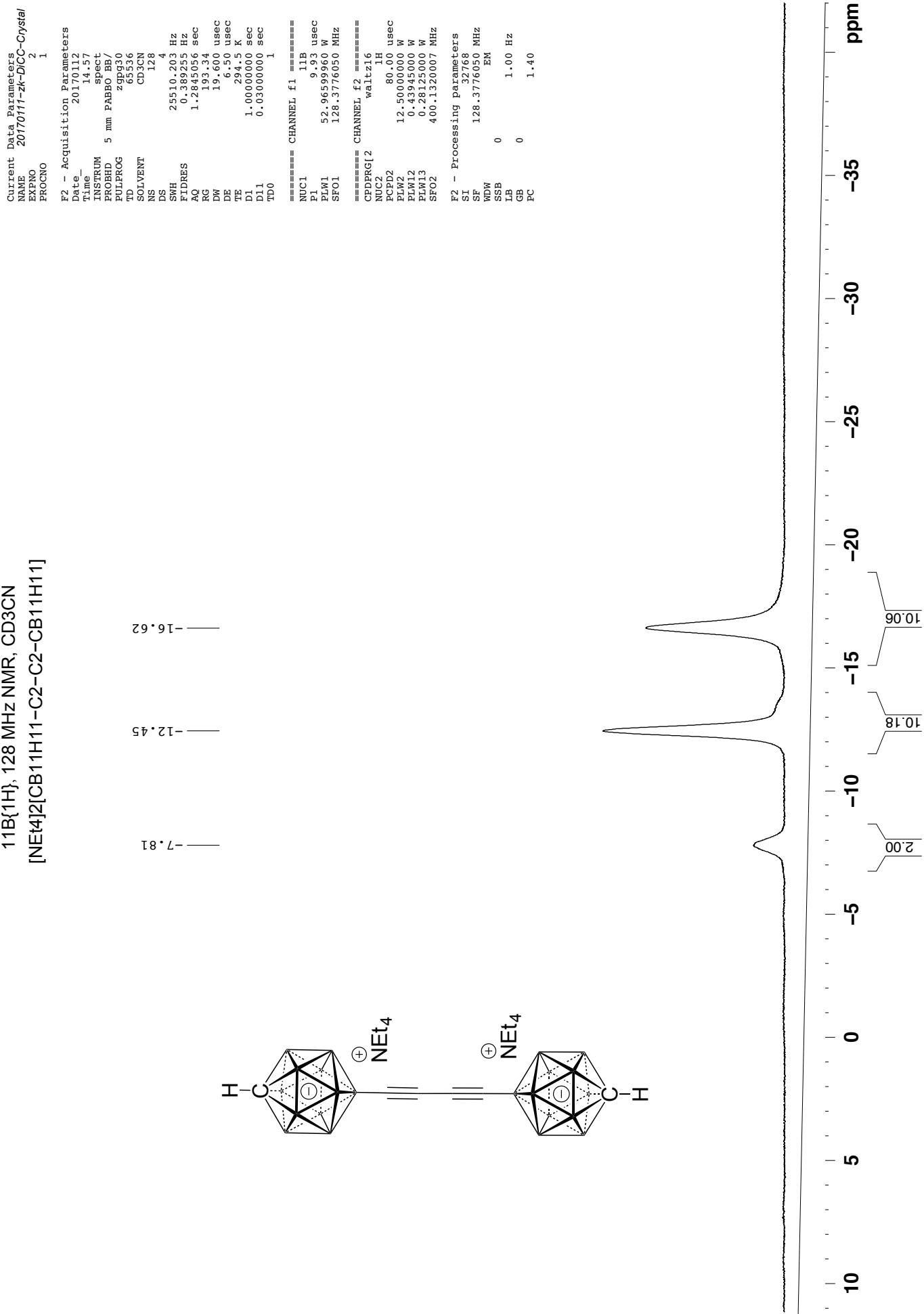
$^1\text{H}\{^{11}\text{B}\}$ , 400 MHz NMR, CD<sub>3</sub>CN  
 \* [NEt<sub>4</sub>]<sub>2</sub>[CB'1H11-C2-C2-CB11H11]



11B{1H}, 128 MHz NMR, CD3CN  
[NEt<sub>4</sub>]<sub>2</sub>[CB11H11-C2-C2-CB11H11]



11B{1H}, 128 MHz NMR, CD3CN  
[NEt<sub>4</sub>]2[CB11H11-C2-C2-CB11H11]



$^{13}\text{C}\{^1\text{H}\}$ , 100 MHz NMR, CD<sub>3</sub>CN \*  
 [NEt<sub>4</sub>]<sub>2</sub>[CB11H11-C2-C2-CB11H11]

