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Electronic Supplementary Information (ESI)

Implications of including a magnetic ion (Cr³⁺ and Fe³⁺) at the vanadium site in

geometrically frustrated spinel MgV₂O₄: Magnetic and Catalytic properties



Fig. S1 Thermogravimetric (TG) and Differential Scanning Calorimetric (DSC) traces of xerogel obtained from the reaction of propylene oxide with (a) MgCl₂.6H₂O₂ VCl₃, CrCl₃.6H₂O and (b) MgCl₂.6H₂O, VCl₃, Fe(NO₃)₃.9H₂O.



Fig. S2 Temporal changes in the absorbance characteristics of 50 mL MB dye solution (10 μ M) in the presence of 5 mL H₂O₂ and 50 mg of (a) MgV₂O₄, (b) MgVCrO₄ and (c) MgVFeO₄ as catalyst.

Formula	MgV ₂ O ₄	MgVCrO ₄	MgVFeO ₄	
Crystal system	Cubic	Cubic	Cubic	
Space group	$Fd^{3}m$ (#227)	$Fd^{3}m$ (#227)	$Fd^{3}m$ (#227)	
<i>a</i> (Å)	8.3990 (4)	8.3497 (5)	8.3957 (27)	
Cell volume (Å ³)	592.49 (8)	582.12 (10)	591.81 (6)	
Formula weight (g/mol)	187.64	184.04	189.12	
Z	8	8	8	
ρ calc (g/cm ³)	4.207	4.200	4.246	
Temperature (°C)	25	25	25	
No. of data points	2250	2250	1875	
2θ range	10-100°	10-100°	25-100°	
R_{p} (%)	7.38	6.54	3.10	
R_{wp} (%)	9.57	8.22	3.99	
χ^2_{-}	1.461	1.343	1.318	

 Table S1 Summary of the crystallographic details from the Rietveld refinement of PXRD

 patterns

Atoms	Wyck	x/a	у/b	z/c	SOF	U(iso)Å ²
V	16с	0	0	0	0.971 (8)	0.0264 (8)
Mg	8b	0.375	0.375	0.375	1.015 (9)	0.0286 (14)
Ο	32e	0.2411 (4)	0.2411 (4)	0.2411 (4)	1	0.025
V	16с	0	0	0	0.437 (8)	0.0197 (10)
Cr	16c	0	0	0	0.496 (7)	0.0197 (10)
Mg	<i>8b</i>	0.375	0.375	0.375	0.986 (8)	0.0344 (21)
Ο	32e	0.2409 (6)	0.2409 (6)	0.2409 (6)	1.0	0.025
V	16с	0	0	0	0.520 (27)	0.022(4)
Fe	16c	0	0	0	0.397 (19)	0.0322 (32)
Mg	8b	0.375	0.375	0.375	0.992 (18)	0.0209 (30)
0	32e	0.2402 (8)	0.2402 (8)	0.2402 (8)	1.0	0.025

Table S2: Atomic parameters after the final cycle of refinement of MgV_2O_4 , $MgCrVO_4$, and $MgVFeO_4$.