

## Electronic Supplementary Information (ESI)

### A new synthesis of porphyrin via a putative *trans*-manganese(IV)-dihydroxide intermediate

Sruti Mondal, Kasturi Sahu, Bratati Patra, Subhrakant Jena, Himansu S. Biswal,\* and Sanjib Kar\*

School of Chemical Sciences, National Institute of Science Education and Research (NISER), Bhubaneswar, Khordha, 752050, India and Homi Bhabha National Institute, Training School Complex, Anushakti Nagar, Mumbai, 400 094, India

E-mail: [himansu@niser.ac.in](mailto:himansu@niser.ac.in), [sanjib@niser.ac.in](mailto:sanjib@niser.ac.in)

## Experimental section

### Synthesis of 5,15-bis(4-cyanophenyl)-10-(pentafluorophenyl)tetrapyrrole

The following synthesis was performed by following a protocol reported earlier.<sup>1</sup> 0.100 g (0.4 mmol) of 5-(4-cyanophenyl)dipyrromethane and 49.3  $\mu$ L (0.4 mmol) of 2,3,4,5,6-Pentafluorobenzaldehyde were dissolved in 50mL MeOH and 25 mL water (2:1) mixture. Subsequently, 5 mL of aqueous HCl (36%) was added to it. The reaction mixture was kept on stirring for 2 hours at room temperature. The crude product was washed and extracted several times with chloroform and water and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. The solvent was evaporated to dryness. The reaction mixture was purged through a silica gel (100-200 mesh) column. The final product was eluted using EtOAc/hexane mixture as eluent.

### For 5,15-bis(4-cyanophenyl)-10-(pentafluorophenyl)tetrapyrrole

Yield: 30% (41 mg). Anal. Calcd (found) for C<sub>39</sub>H<sub>25</sub>F<sub>5</sub>N<sub>6</sub>: C, 69.64 (69.77); H, 3.75 (3.84); N, 12.49 (12.37). <sup>1</sup>H NMR (400 MHz, Chloroform-*d*)  $\delta$  8.14 – 7.86 (br s, 4H), 7.53 (m, 5H), 6.79 – 6.60 (m, 2H), 6.15 (dt, *J* = 11.2, 3.0 Hz, 2H), 5.97 – 5.53 (m, 9H), 5.42 (d, *J* = 6.3 Hz, 2H), 5.31 (s, 1H) (Fig. S13); The electrospray mass spectrum in acetonitrile showed peaks centred at *m/z* = 695.1893 correspond to [M+Na<sup>+</sup>] (695.196 calcd for C<sub>39</sub>H<sub>25</sub>F<sub>5</sub>N<sub>6</sub>Na) (Fig. S14).

1. B. Koszarna and D. T. Gryko, *J. Org. Chem.* 2006, **71**, 3707-3717.

<b>Table S1</b>	UV-Vis. Data <sup>a</sup>
<b>Table S2</b>	Composition of selected molecular orbitals of <b>7-Mn</b> .
<b>Table S3</b>	Composition of selected molecular orbitals of <b>7-Mn</b> .
<b>Table S4</b>	TD-DFT transitions for <b>7-Mn</b> .
<b>Table S5</b>	TD-DFT calculated IR frequencies for <b>7-Mn</b> .
<b>Fig. S1</b>	ESI- MS spectrum in CH <sub>3</sub> CN shows the (a) measured spectrum, (b) isotopic distribution pattern (experimental) and (c) isotopic distribution pattern (simulated) of <b>7-Mn</b> .
<b>Fig. S2</b>	Cyclic voltammogram (—) of <b>7-Mn</b> in CH <sub>3</sub> CN. The potentials are vs. ferrocene/ferricinium.
<b>Fig. S3</b>	<sup>1</sup> H NMR spectrum of 5,15-Bis(pentafluorophenyl)-10,20-diphenylporphyrin, <b>1</b> in CDCl <sub>3</sub> .
<b>Fig. S4</b>	<sup>1</sup> H NMR spectrum of 5,15-Bis(4-cyanophenyl)-10,20-bis(4-methoxyphenyl)porphyrin, <b>2</b> in CDCl <sub>3</sub> .
<b>Fig. S5</b>	<sup>1</sup> H NMR spectrum of 5,15-Dimesityl-10,20-diphenylporphyrin, <b>3</b> in CDCl <sub>3</sub> .
<b>Fig. S6</b>	<sup>1</sup> H NMR spectrum of 5,15-Bis(4-methoxyphenyl)-10,20-diphenylporphyrin, <b>4</b> in CDCl <sub>3</sub> .
<b>Fig. S7</b>	<sup>1</sup> H NMR spectrum of 5,15-Bis(4-cyanophenyl)-10,20-diphenylporphyrin, <b>5</b> in CDCl <sub>3</sub> .
<b>Fig. S8</b>	<sup>1</sup> H NMR spectrum of 5,15-Bis(4-methoxyphenyl)-10,20-bis(pentafluorophenyl)porphyrin, <b>6</b> in CDCl <sub>3</sub> .
<b>Fig. S9</b>	<sup>1</sup> H NMR spectrum of 5,15-bis(4-cyanophenyl)-10,20-bis(pentafluorophenyl)porphyrin, <b>7</b> in CDCl <sub>3</sub> .
<b>Fig. S10</b>	<sup>1</sup> H NMR spectrum of 5,10,15,20-Tetraphenylporphin, <b>8</b> in CDCl <sub>3</sub> .
<b>Fig. S11</b>	<sup>1</sup> H NMR spectrum of 5,10,15,20-Tetrakis(4-methoxyphenyl)porphyrin, <b>9</b> in CDCl <sub>3</sub> .
<b>Fig. S12</b>	Chemical reduction of <b>7-Mn</b> via LiAlH <sub>4</sub> in DCM solution at RT. Black line indicates the absorption spectra before reduction and red line indicates the species obtained after reduction.

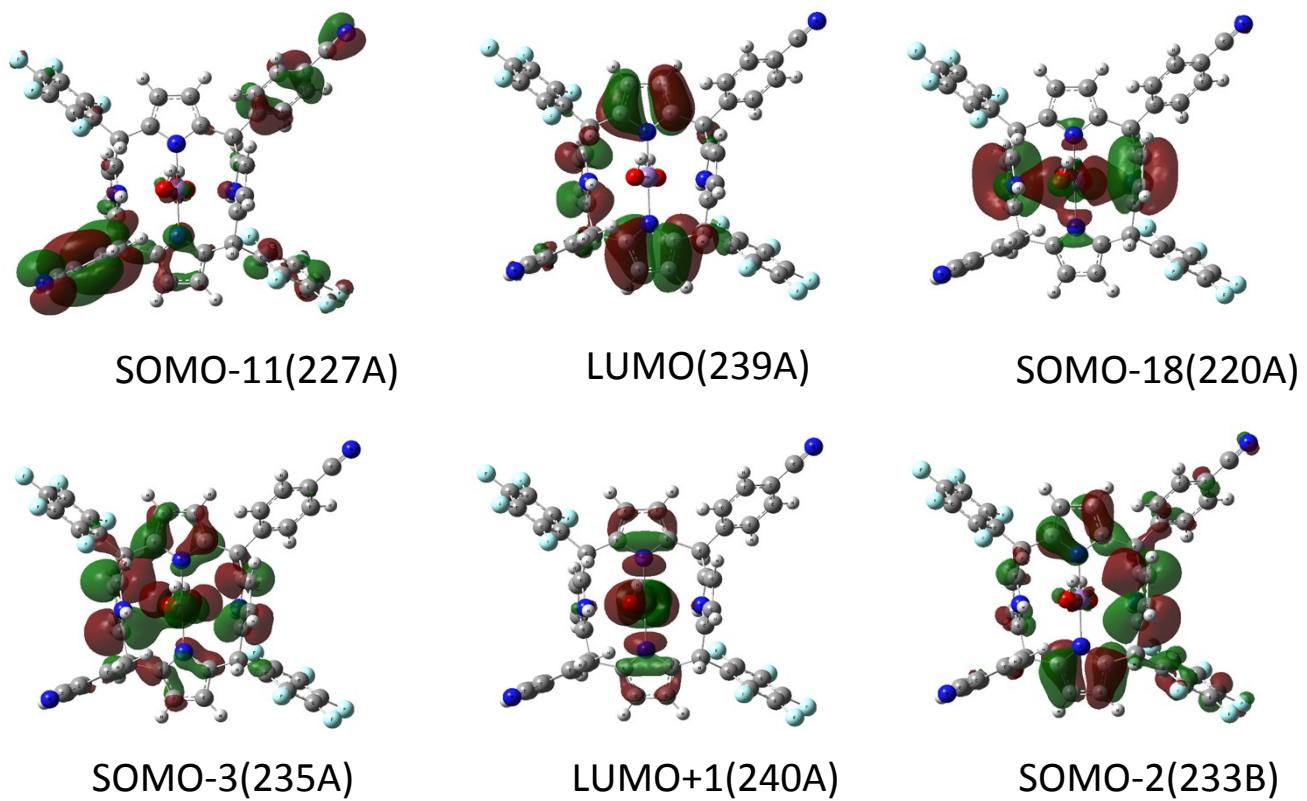
**Fig. S13**  $^1\text{H}$  NMR spectrum of 5,15-bis(4-cyanophenyl)-10-(pentafluorophenyl)tetrapyrrole in  $\text{CDCl}_3$ .

**Fig. S14** ESI- MS spectrum in  $\text{CH}_3\text{CN}$  shows the (a) measured spectrum, (b) isotopic distribution pattern of 5,15-bis(4-cyanophenyl)-10-(pentafluorophenyl)tetrapyrrole.

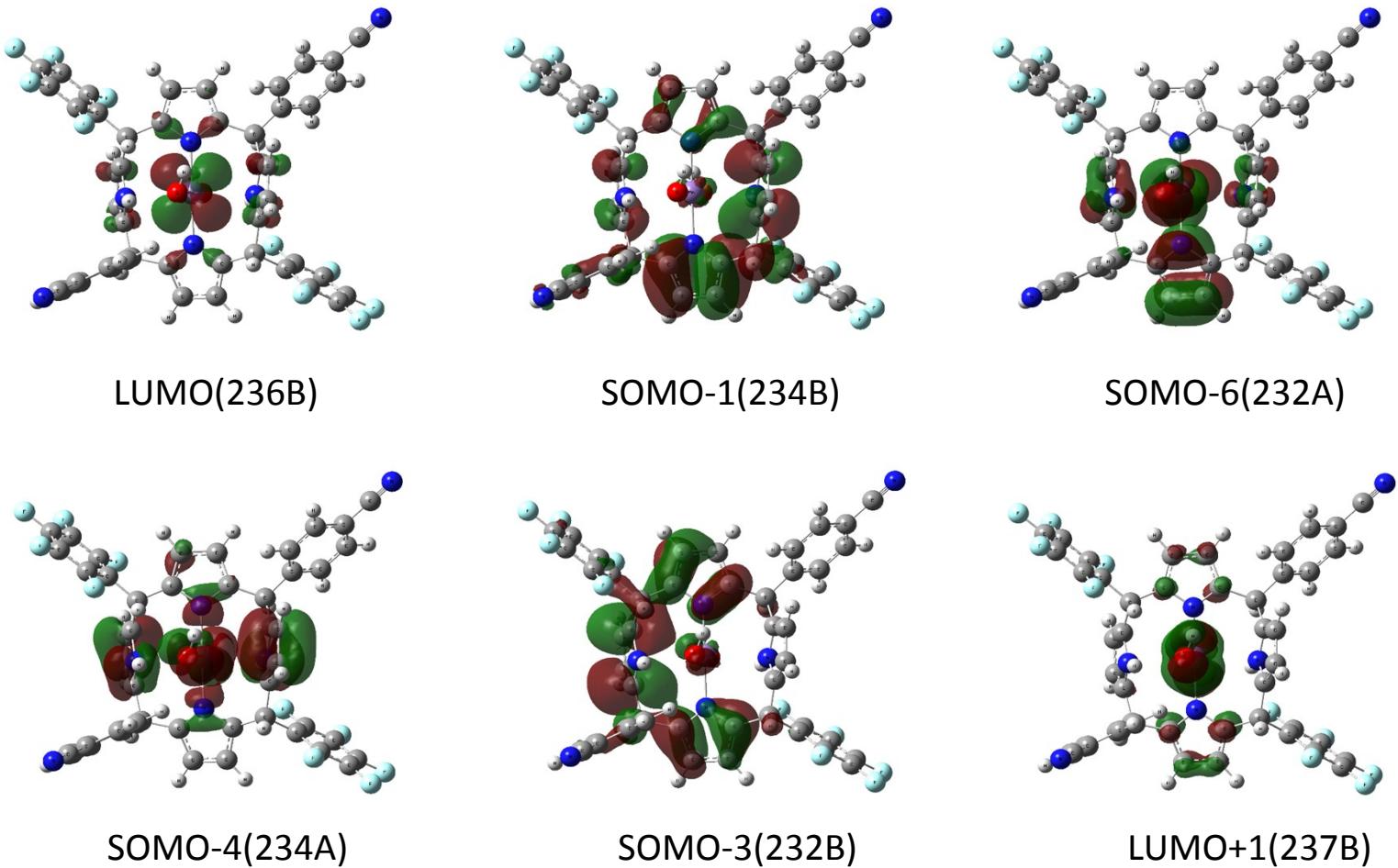
**Table S1** UV–Vis. Data <sup>a</sup>

Compound	UV–vis. Data <sup>a</sup> $\lambda_{\text{max}}$ / nm ( $\epsilon$ / M <sup>-1</sup> cm <sup>-1</sup> )
<b>7-Mn</b> <sup>a</sup>	416 (10,860), 470 (7970), 576 (4020), 621 (7140).

<sup>a</sup> In dichloromethane.



**Table S2** Composition of selected molecular orbitals of **7-Mn**.



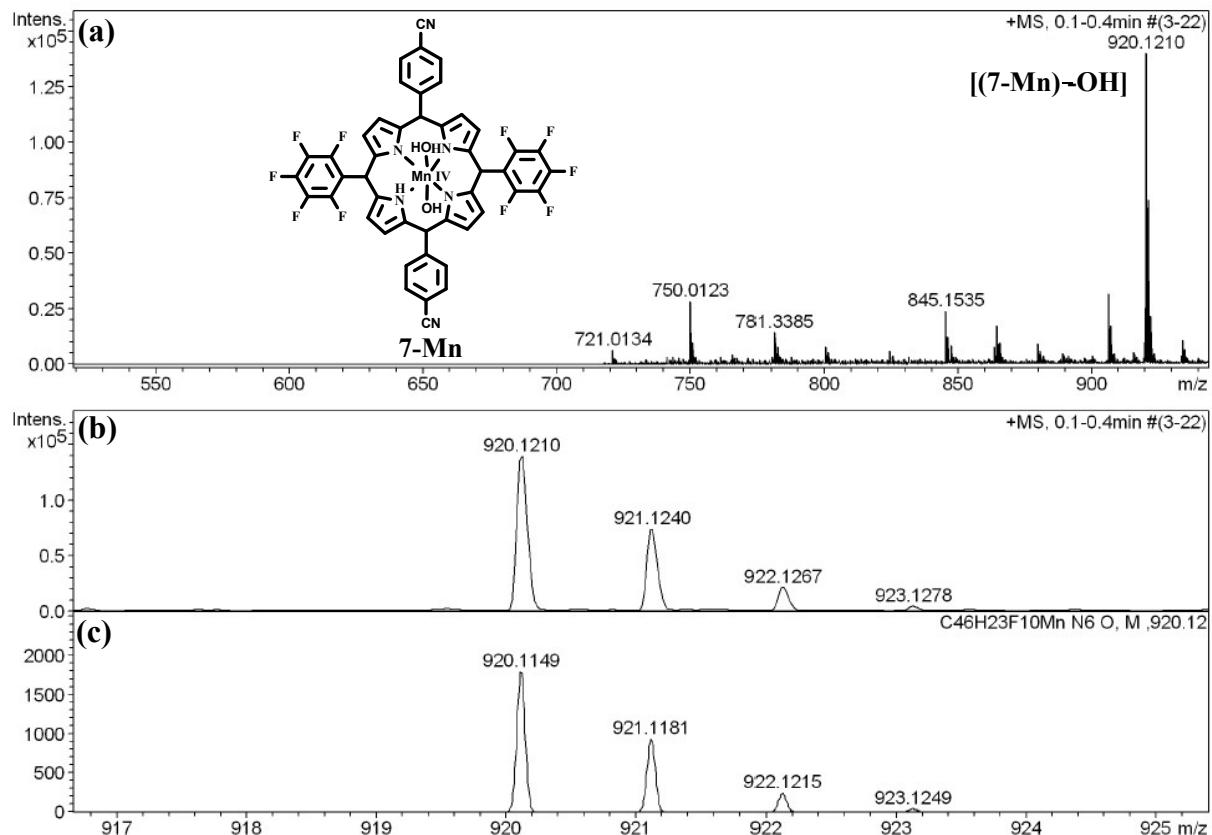
**Table S3** Composition of selected molecular orbitals of **7-Mn**.

State	E(eV)	Transition Orbitals	Contribution	Character
<b>S<sub>13</sub></b>	1.76	SOMO-11(227A) to LUMO(239A)	0.99(100%)	ILCT
<b>S<sub>26</sub></b>	2.31	SOMO-18(220A) to LUMO239A)	0.11(1.19%)	LMCT
		SOMO-3(235A) to LUMO+1(240A)	0.11(1.30%)	LMCT
		SOMO-2(233B) to LUMO(236B)	0.98(96.30%)	LMCT
		SOMO-1(234B) to LUMO(236B)	0.11(1.20%)	LMCT
<b>S<sub>29</sub></b>	2.52	SOMO-6(232A) to LUMO+1(240A)	0.22(4.87%)	LMCT
		SOMO-4(234A) to LUMO+1(240A)	0.66(44.37%)	LMCT
		SOMO-3(232B) to LUMO(236B)	0.69(48.37%)	LMCT
		SOMO-1(234B) to LUMO+1(237B)	0.14(1.92%)	LMCT

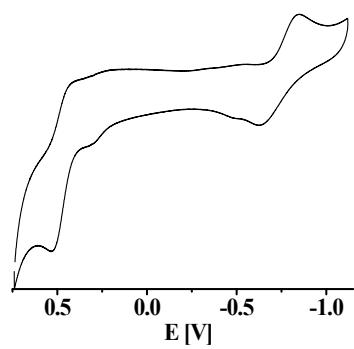
**Table S4** TD-DFT transitions for **7-Mn**.

**Table S5** TD-DFT calculated IR frequencies for 7-Mn.

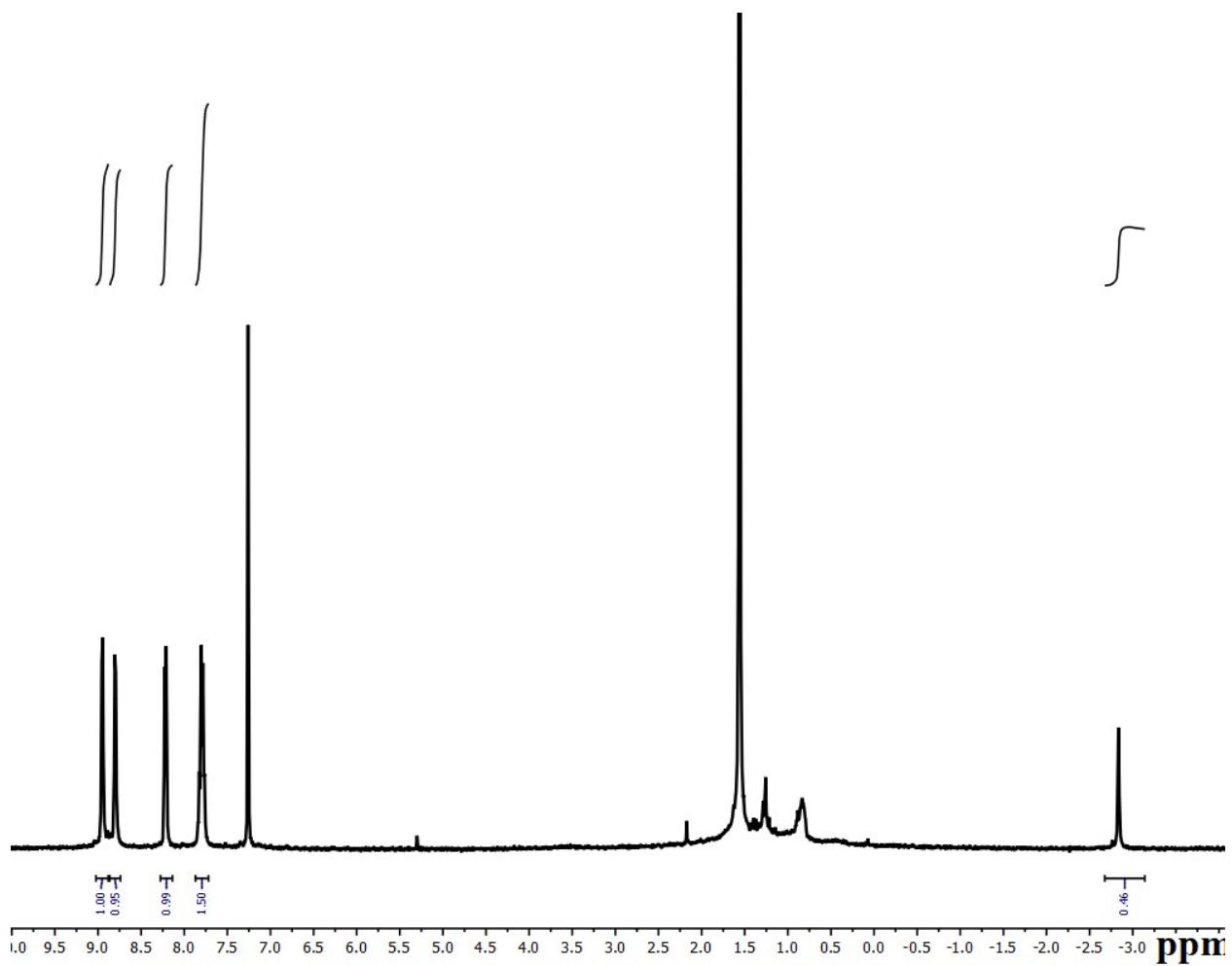
<b>Functional Group</b>	<b>IR Frequency(cm<sup>-1</sup>)</b>
<b>Water Stretch</b>	3667.10
	3664.82
<b>Pyrrole NH</b>	3466.32
	3437.08
<b>Pyrrole CH</b>	3235.57 -3176.80
<b>Benzonitrile CH</b>	3153.95-3119.51
<b>PFB</b>	2993-2924
<b>CN Stretch</b>	2257.96
	2257.83
<b>Aromatics C=C bending</b>	1627.74-900
<b>OH bending</b>	700-900



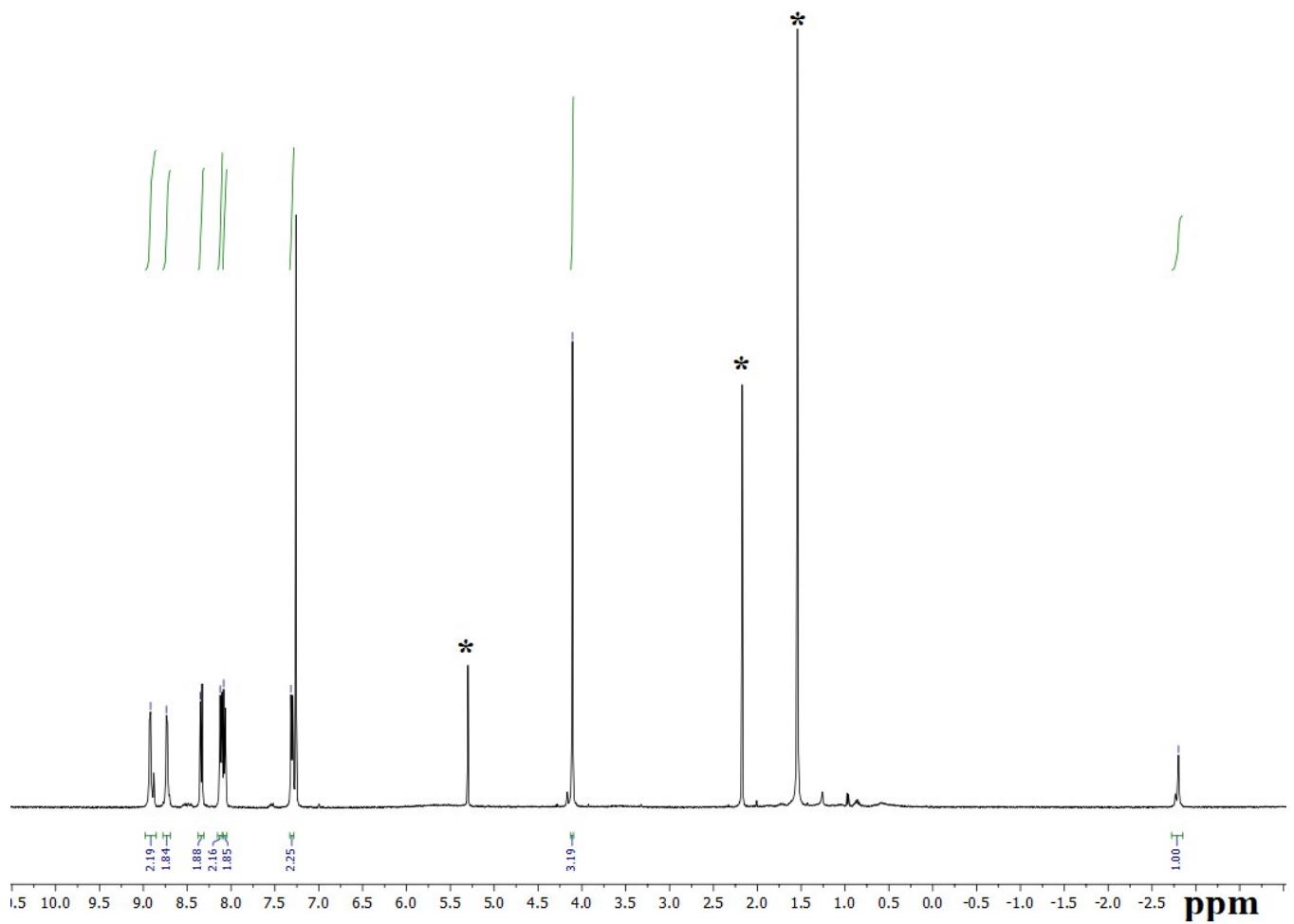
**Fig. S1** ESI- MS spectrum in CH<sub>3</sub>CN shows the (a) measured spectrum, (b) isotopic distribution pattern (experimental) and (c) isotopic distribution pattern (simulated) of 7-Mn.



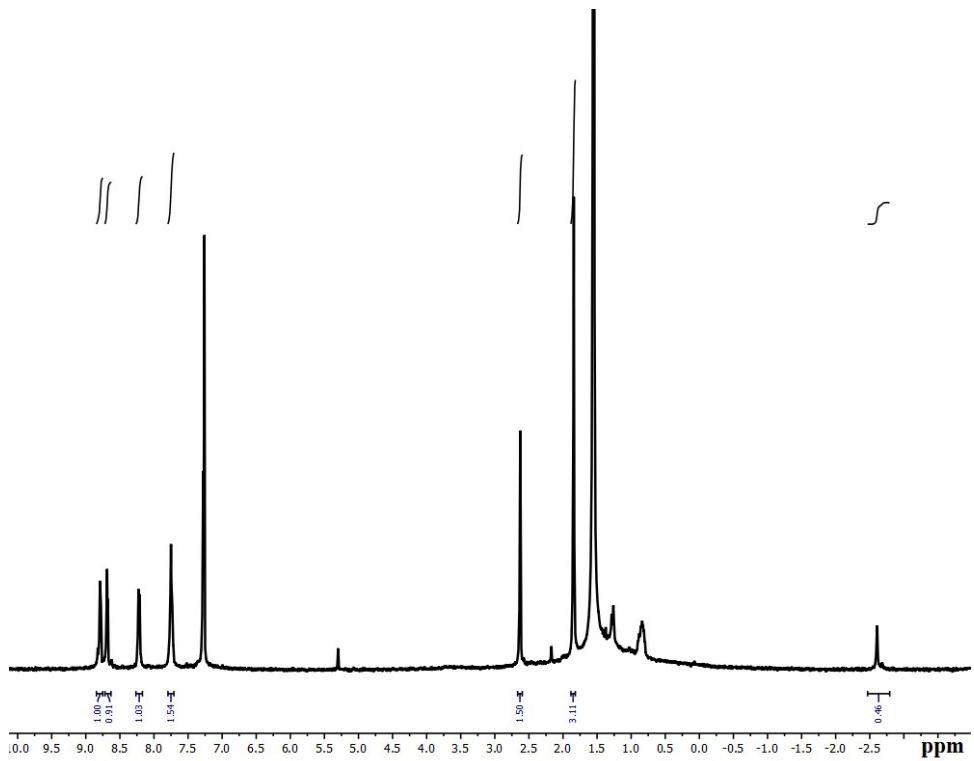
**Fig. S2** Cyclic voltammogram (——) of **7-Mn** in CH<sub>3</sub>CN. The potentials are vs. ferrocene/ferricinium.



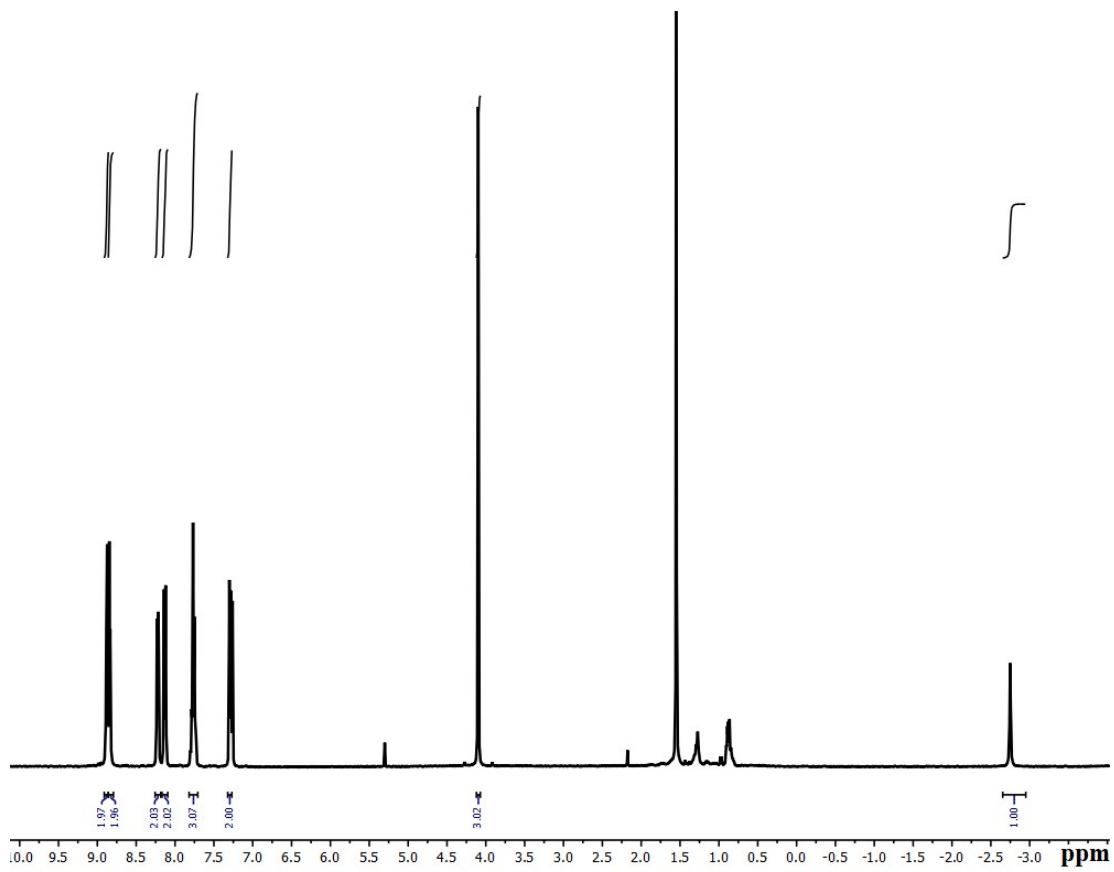
**Fig. S3** <sup>1</sup>H NMR spectrum of 5,15-Bis(pentafluorophenyl)-10,20-diphenylporphyrin, **1** in CDCl<sub>3</sub>.



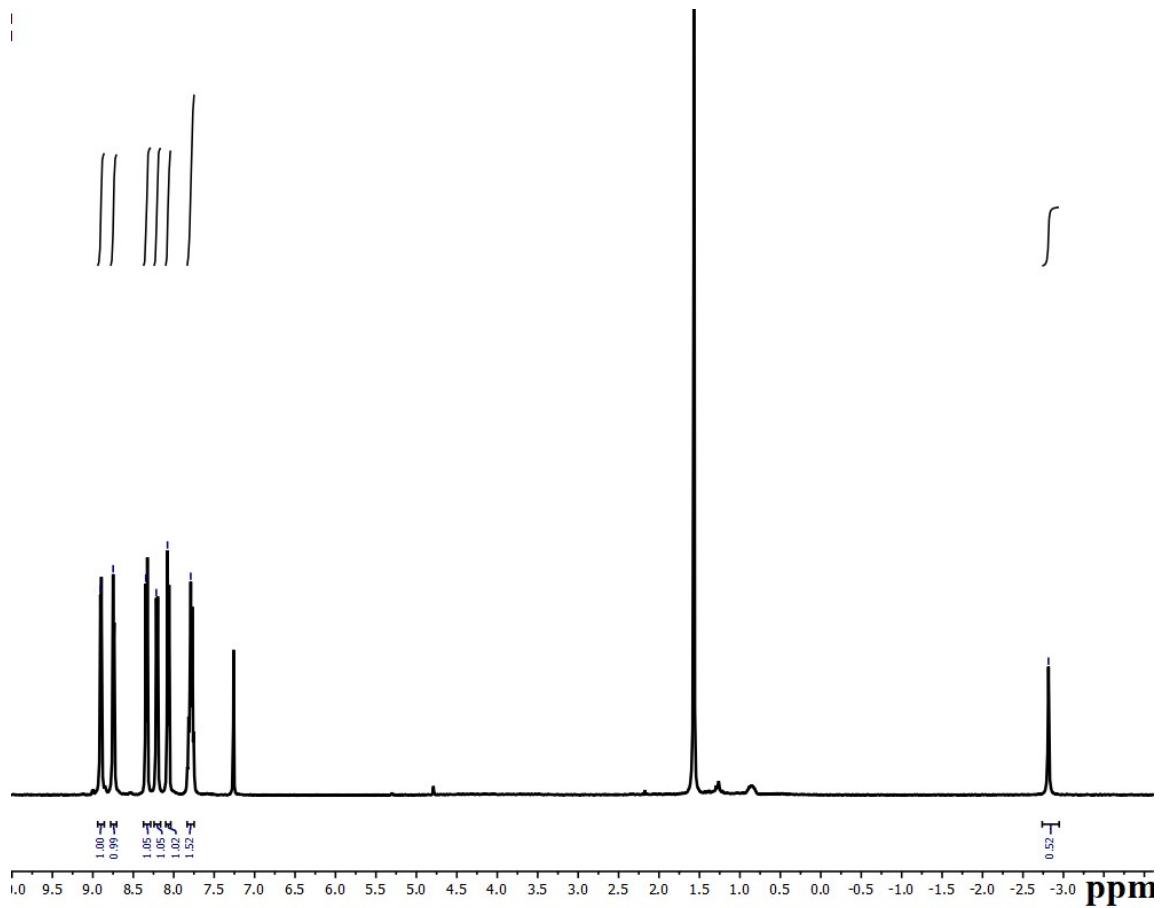
**Fig. S4**  $^1\text{H}$  NMR spectrum of 5,15-Bis(4-cyanophenyl)-10,20-bis(4-methoxyphenyl)porphyrin, **2** in  $\text{CDCl}_3$ .



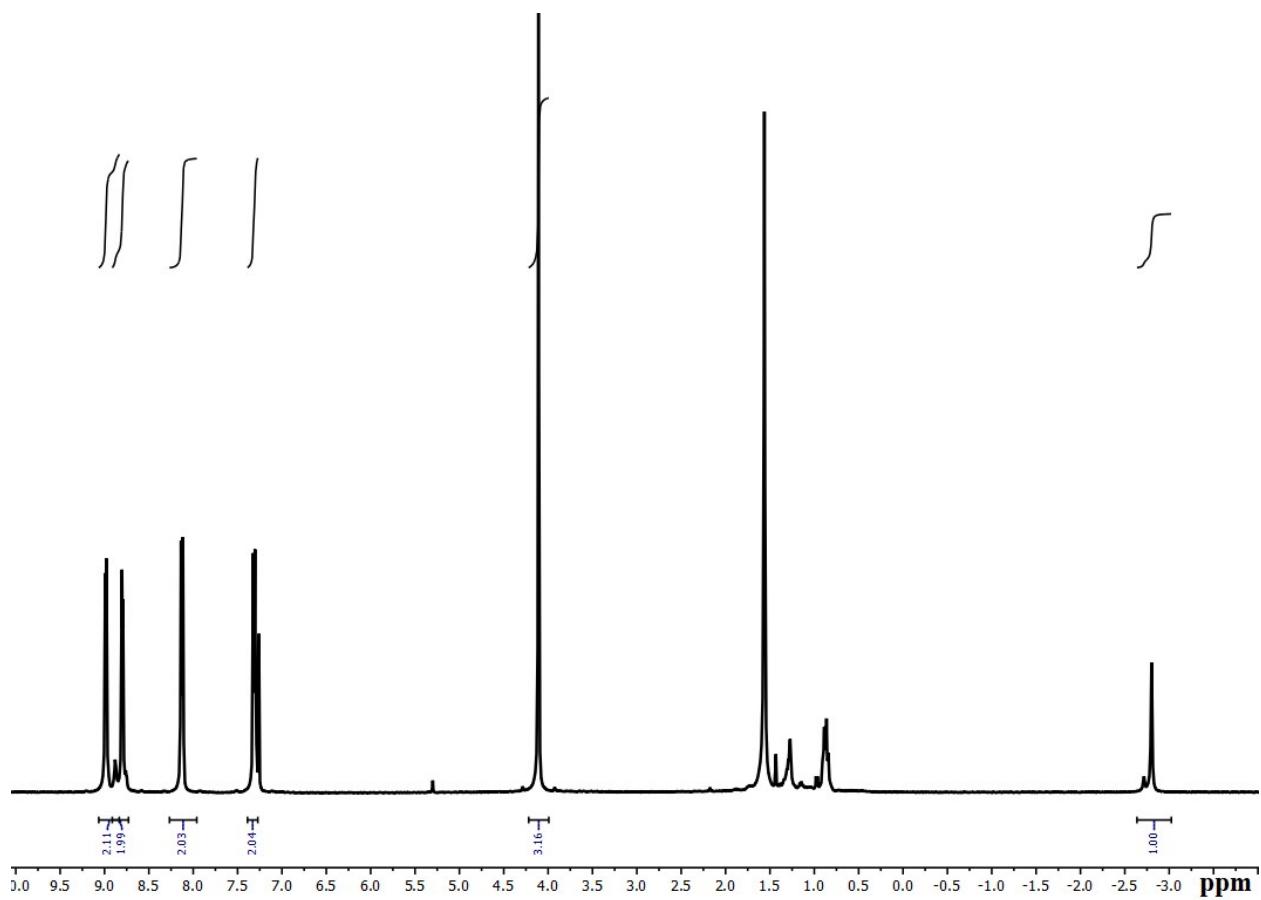
**Fig. S5** <sup>1</sup>H NMR spectrum of 5,15-Dimesityl-10,20-diphenylporphyrin, **3** in CDCl<sub>3</sub>.



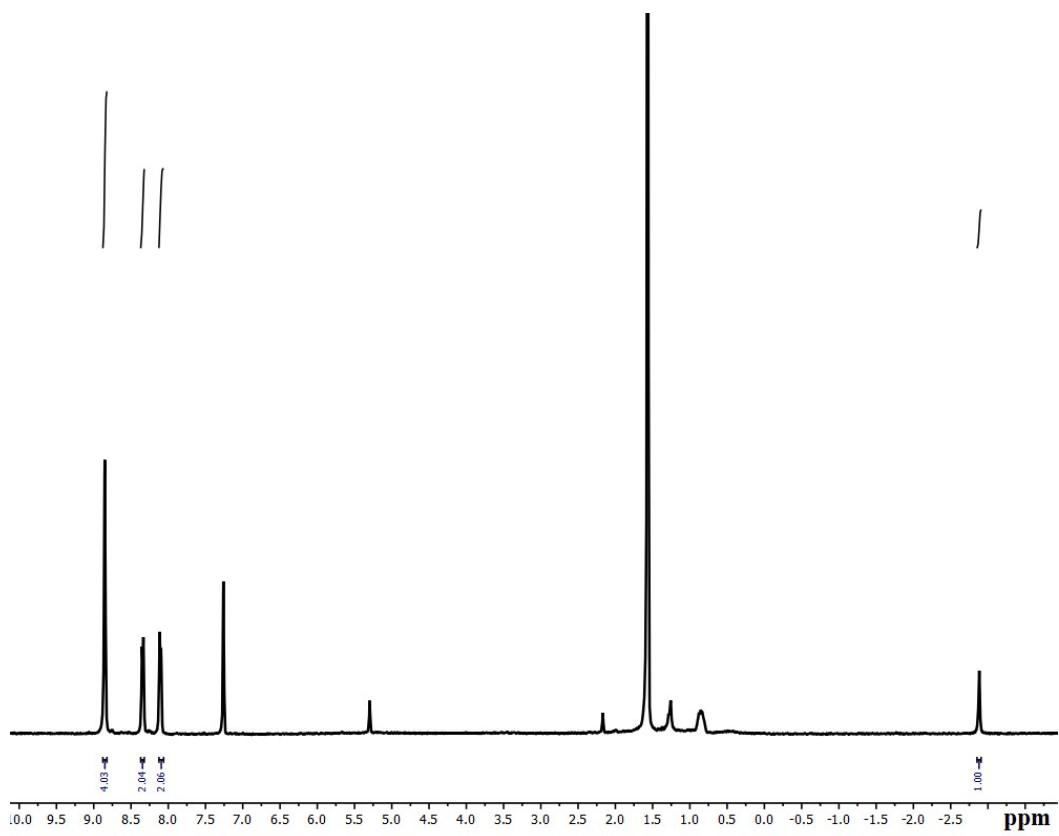
**Fig. S6**  $^1\text{H}$  NMR spectrum of 5,15-Bis(4-methoxyphenyl)-10,20-diphenylporphyrin, **4** in  $\text{CDCl}_3$ .



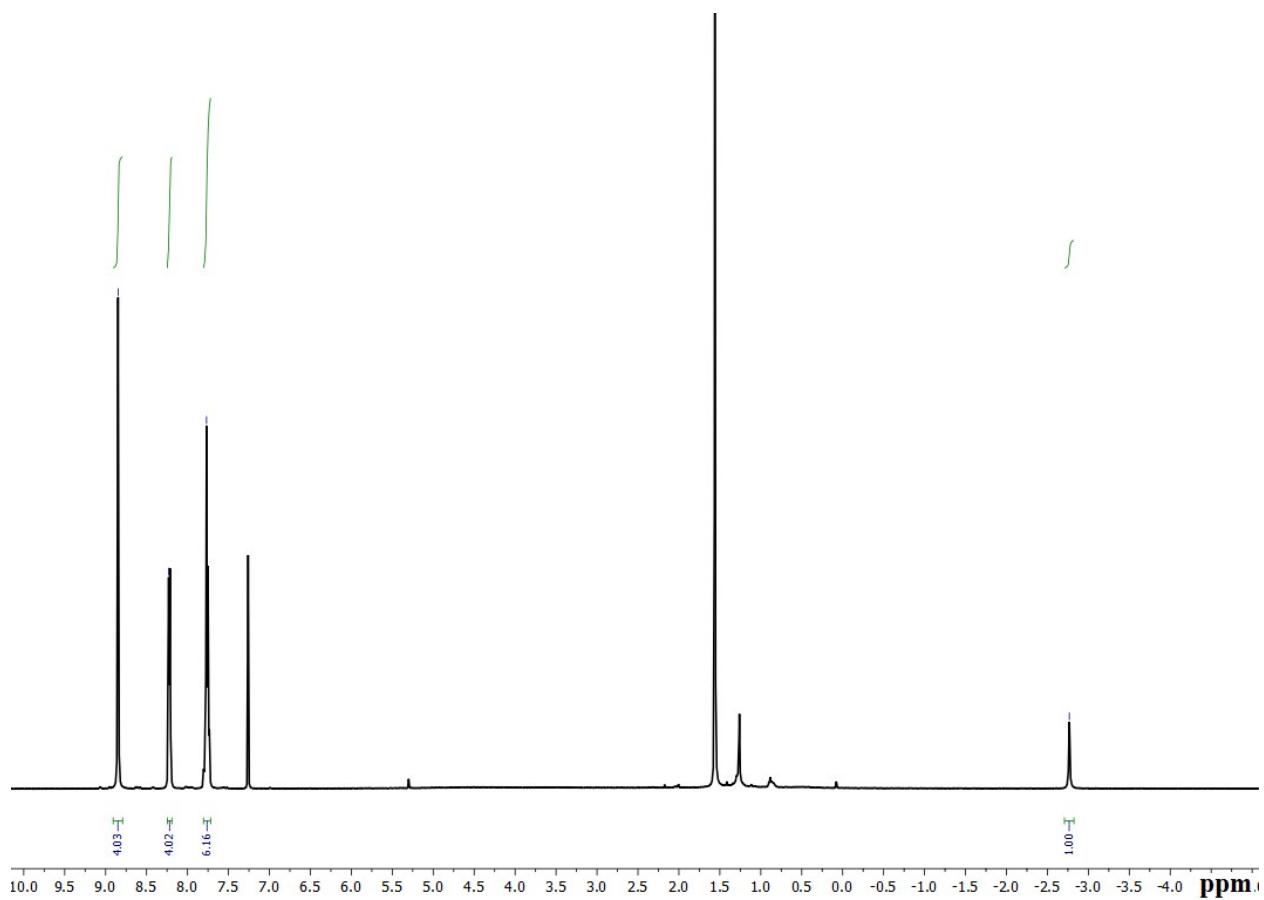
**Fig. S7**  ${}^1\text{H}$  NMR spectrum of 5,15-Bis(4-cyanophenyl)-10,20-diphenylporphyrin, **5** in  $\text{CDCl}_3$ .



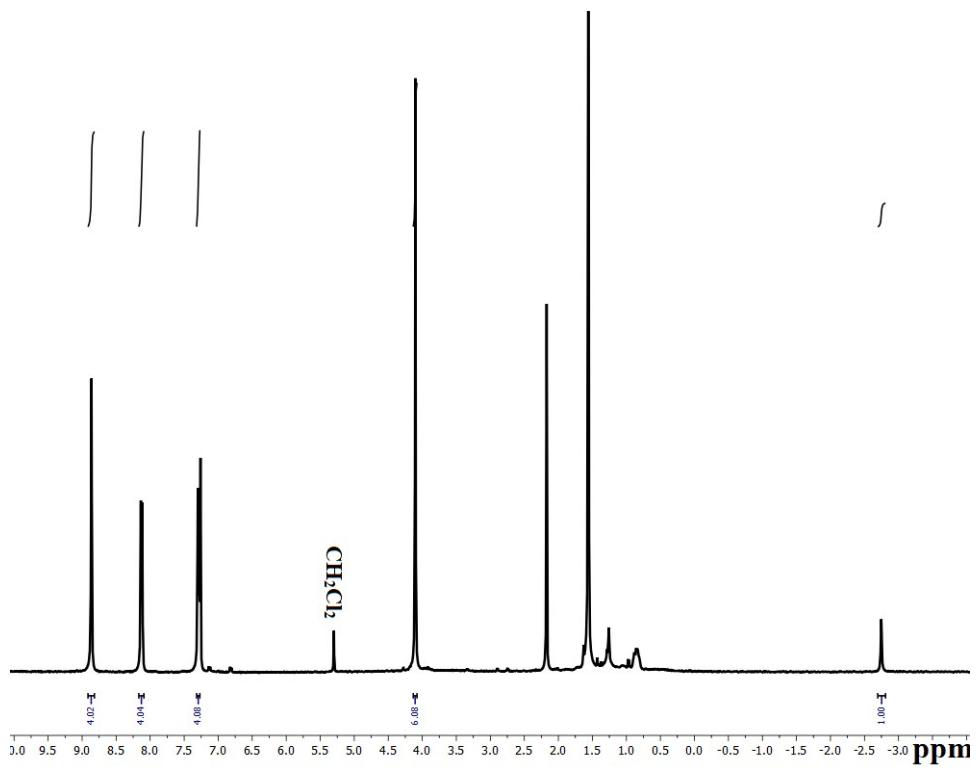
**Fig. S8**  $^1\text{H}$  NMR spectrum of 5,15-Bis(4-methoxyphenyl)-10,20-bis(pentafluorophenyl)porphyrin, **6** in  $\text{CDCl}_3$ .



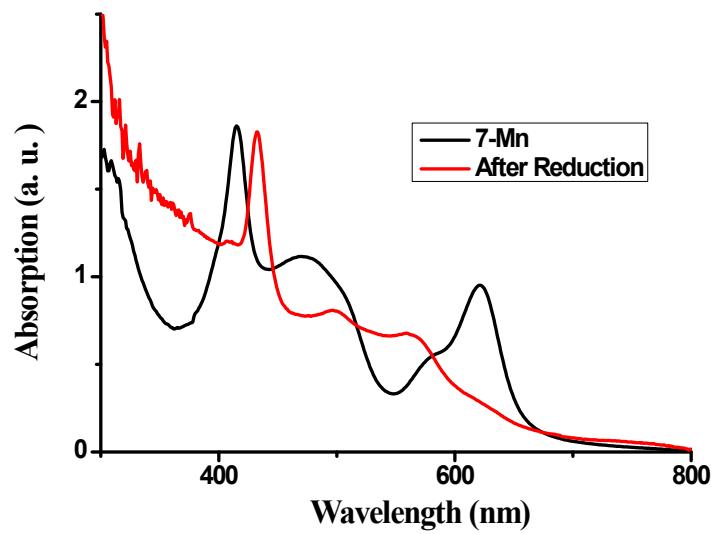
**Fig. S9** <sup>1</sup>H NMR spectrum of 5,15-bis(4-cyanophenyl)-10,20-bis(pentafluorophenyl)porphyrin, **7** in CDCl<sub>3</sub>.



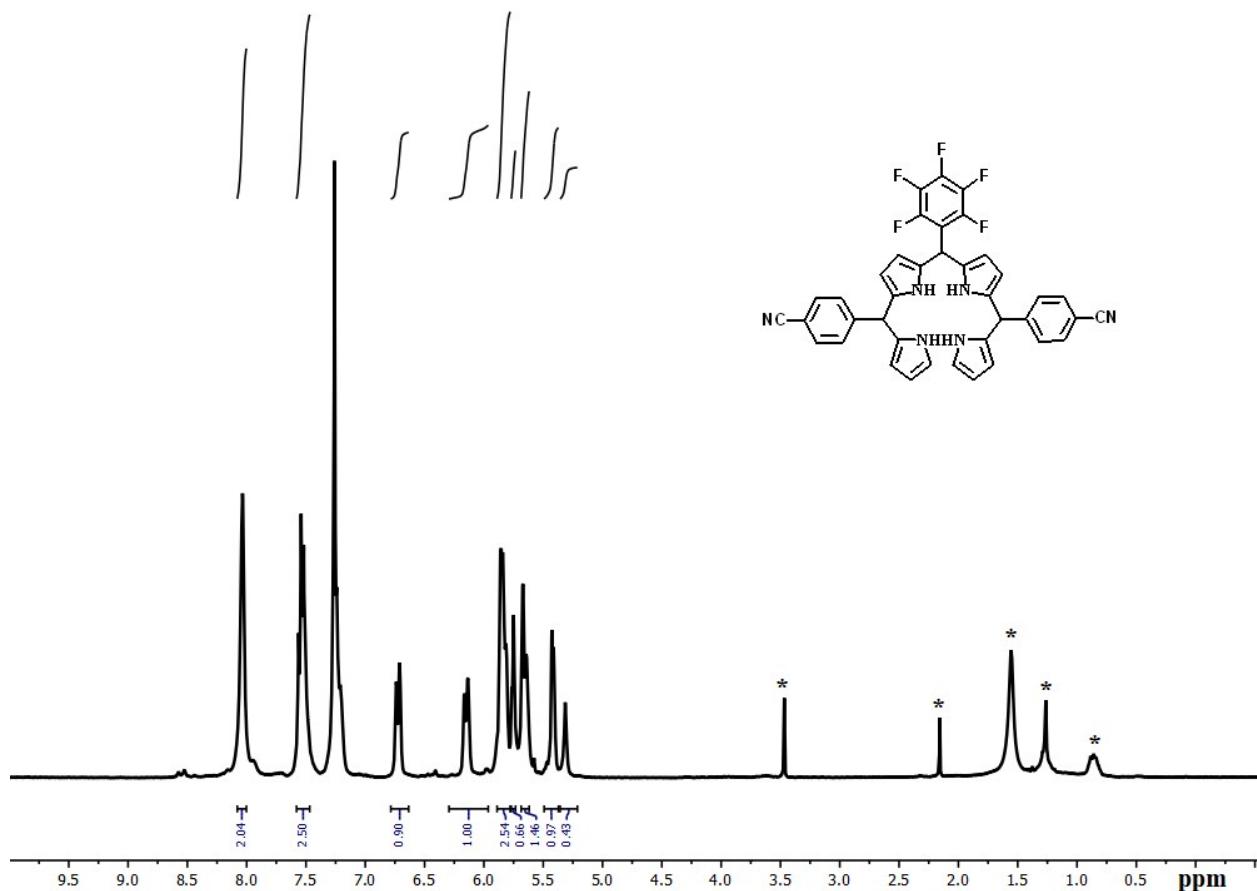
**Fig. S10** <sup>1</sup>H NMR spectrum of 5,10,15,20-Tetraphenylporphin, **8** in CDCl<sub>3</sub>.



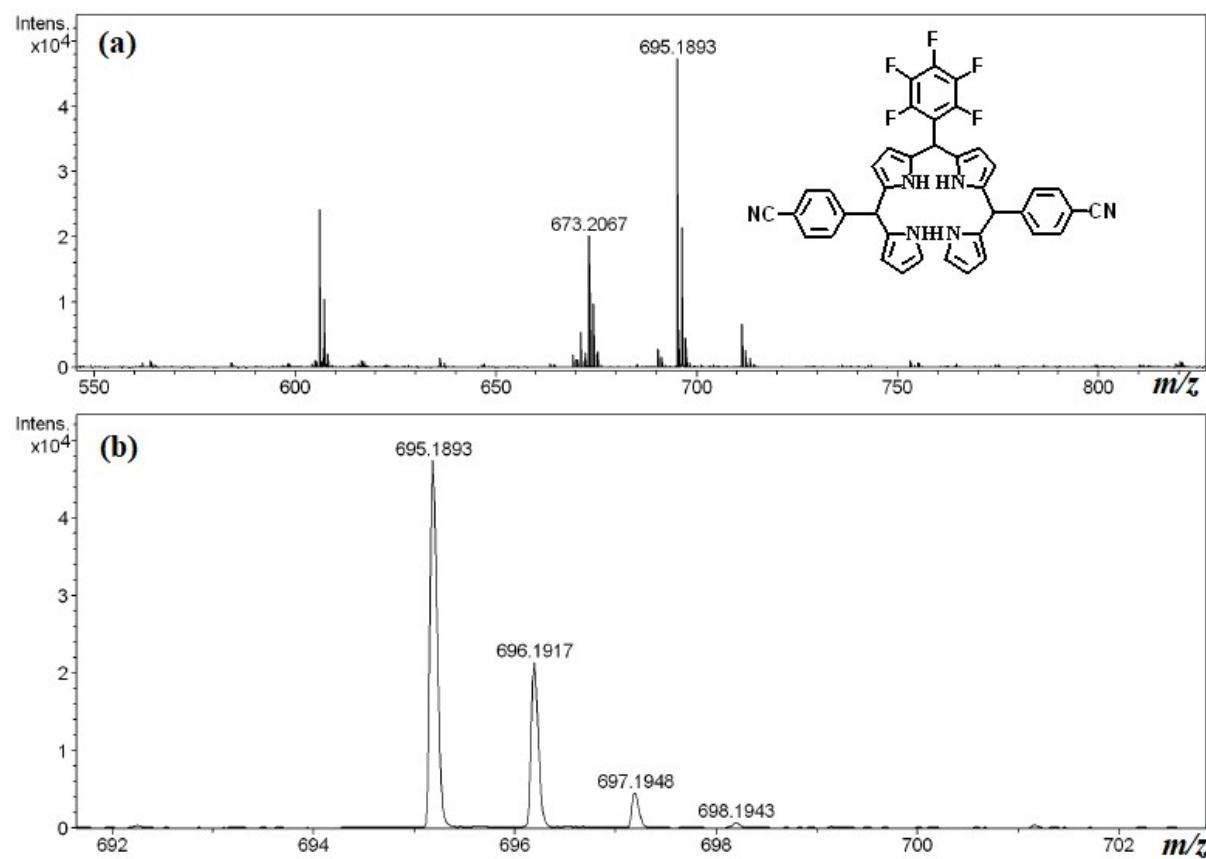
**Fig. S11**  $^1\text{H}$  NMR spectrum of 5,10,15,20-Tetrakis(4-methoxyphenyl)porphyrin, **9** in  $\text{CDCl}_3$ .



**Fig. S12** Chemical reduction of **7-Mn** via LiAlH<sub>4</sub> in DCM solution at RT. Black line indicates the absorption spectra before reduction and red line indicates the species obtained after reduction.



**Fig. S13** <sup>1</sup>H NMR spectrum of 5,15-bis(4-cyanophenyl)-10-(pentafluorophenyl)tetrapyrrole in CDCl<sub>3</sub>.



**Fig. S14** ESI- MS spectrum in  $\text{CH}_3\text{CN}$  shows the (a) measured spectrum, (b) isotopic distribution pattern of 5,15-bis(4-cyanophenyl)-10-(pentafluorophenyl)tetrapyrrole.