

Supporting Information

Regioselective conversions of H₄pdta (1,2-propanediaminetetraacetic acid) and H₄eed3a to their triacetates on peroxotitanates

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Figure S1. UV spectra of (NH₄)[Ti(O₂)(Hpdt)]·H₂O (**1**), (NH₄)₃[Ti(O₂)(pdta)H(pdta)(O₂)Ti]·7H₂O (**2**), (NH₄)[Ti(O₂)(pd3a)]·H₂O (**3**) and (NH₄)[Ti(O₂)(Heed3a)]·H₂O (**5**)

Figure S2. IR spectra of peroxotitanium(IV) complexes: (NH₄)[Ti(O₂)(Hpdt)]·H₂O (**1**), (NH₄)₃[Ti(O₂)(pdta)H(pdta)(O₂)Ti]·7H₂O (**2**) and (NH₄)[Ti(O₂)(pd3a)]·H₂O (**3**)

Figure S3. IR spectra of (NH₄)[Ti(O₂)(Heed3a)]·H₂O (**5**)

Figure S4. Raman spectra of (NH₄)[Ti(O₂)(Hpdt)]·H₂O (**1**) and (NH₄)₃[Ti(O₂)(pdta)H(pdta)(O₂)Ti]·7H₂O (**2**)

Figure S5. ¹³C NMR spectrum of (NH₄)[Ti(O₂)(ed3a)]

Figure S6. ¹³C NMR spectrum of reaction solution for (NH₄)[Ti(O₂)(pd3a)]·H₂O (**3**)

Figure S7. ¹³C NMR and ¹H NMR spectra of blank experiment of oxidation pyridine at 80°C. Reaction conditions: pyridine 1g, 30% H₂O₂ 3ml, reaction time 24h.

Figure S8. ¹³C NMR spectra for the oxidation of pyridine at 80 °C. Reaction conditions: pyridine (1.0 g), 30% H₂O₂ (3.0 mL), catalyst **2** (7.5 mol%).

Scheme S1. Catalytic reaction for the oxidation of pyridine.

Table S1. Crystallographic data and structural refinements for (NH₄)[Ti(O₂)(Hpdt)]·H₂O (**1**), (NH₄)₃[Ti(O₂)(pdta)H(pdta)(O₂)Ti]·7H₂O (**2**), (NH₄)[Ti(O₂)(pd3a)]·H₂O (**3**) and (NH₄)[Ti(O₂)(Heed3a)]·H₂O (**5**).

Table S2. Selected bond distances and angles for (NH₄)[Ti(O₂)(Hpdt)]·H₂O (**1**), (NH₄)₃[Ti(O₂)(pdta)H(pdta)(O₂)Ti]·7H₂O (**2**), (NH₄)[Ti(O₂)(pd3a)]·H₂O (**3**) and (NH₄)[Ti(O₂)(Heed3a)]·H₂O (**5**).

Table S3. Geometrical parameters of hydrogen bonds in (NH₄)[Ti(O₂)(Hpdt)]·H₂O (**1**), (NH₄)₃[Ti(O₂)(pdta)H(pdta)(O₂)Ti]·7H₂O (**2**) and (NH₄)[Ti(O₂)(Heed3a)]·H₂O (**5**)

Table S4. Relevant ¹³C NMR data of H₄edta, (NH₄)[Ti(O₂)(Hedta)]·2H₂O, (NH₄)₂[Ti(O₂)(edta)]·2H₂O and (NH₄)[Ti(O₂)(ed3a)]

Figure S1

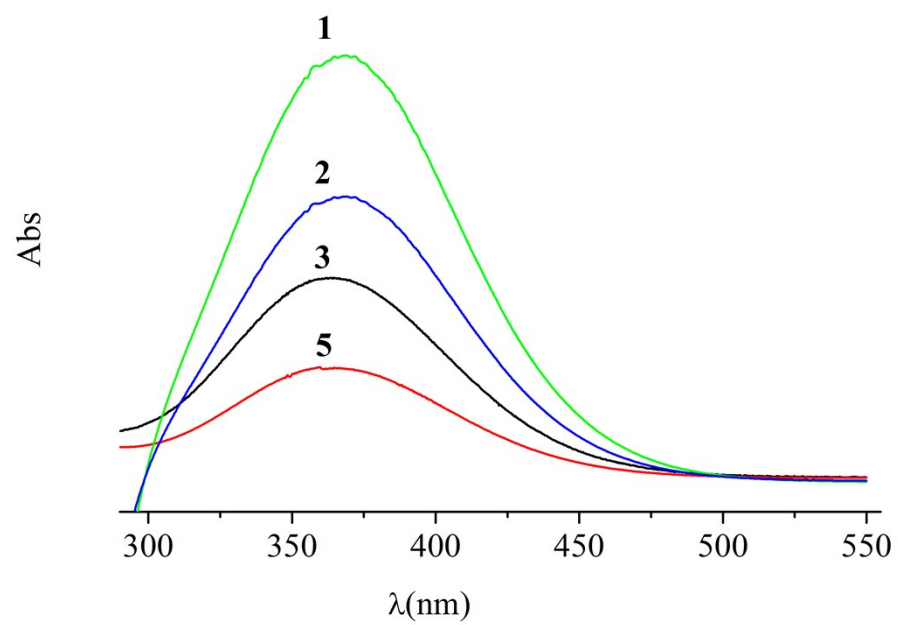
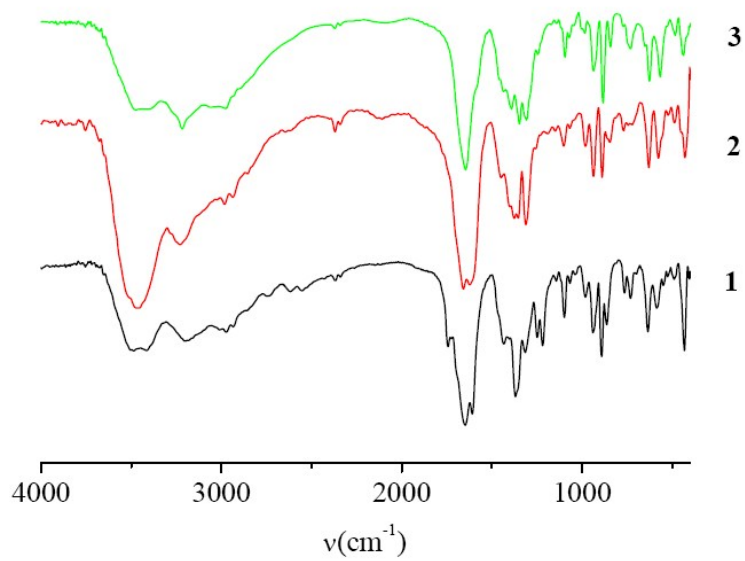
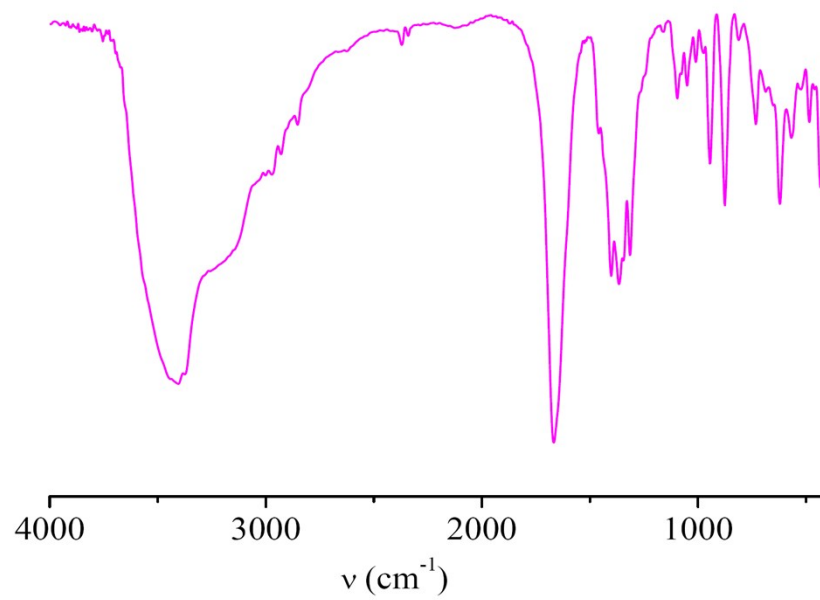
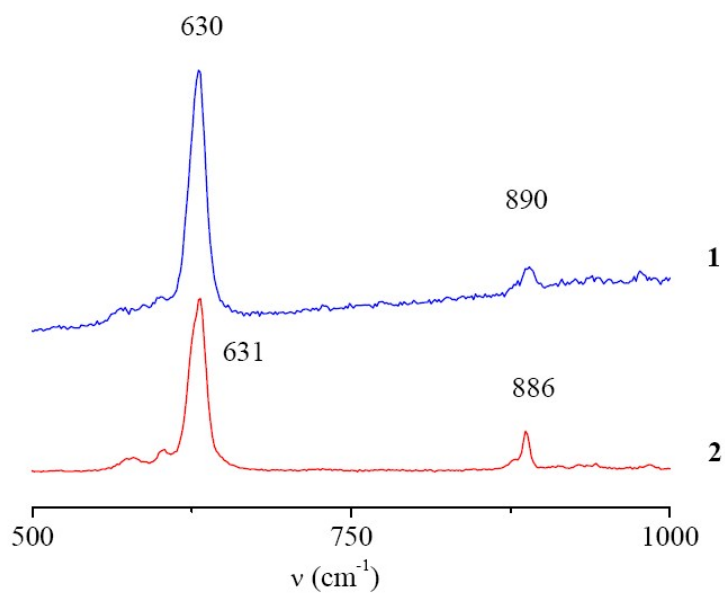


Figure S2







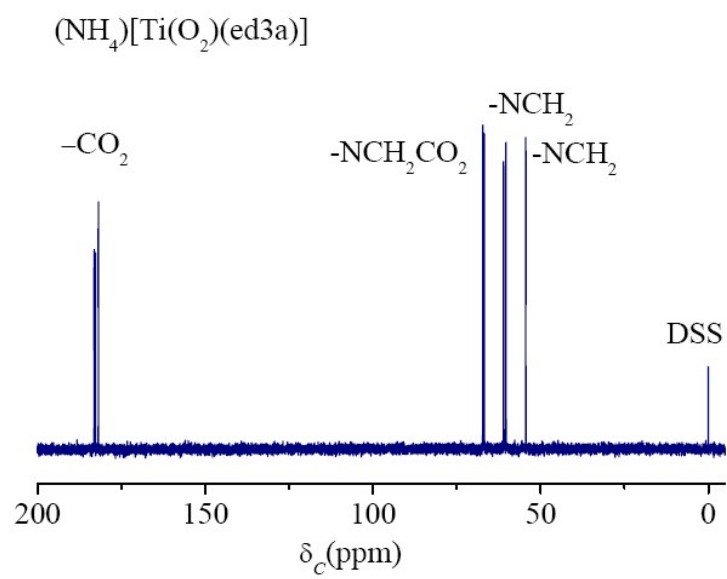


Figure S6

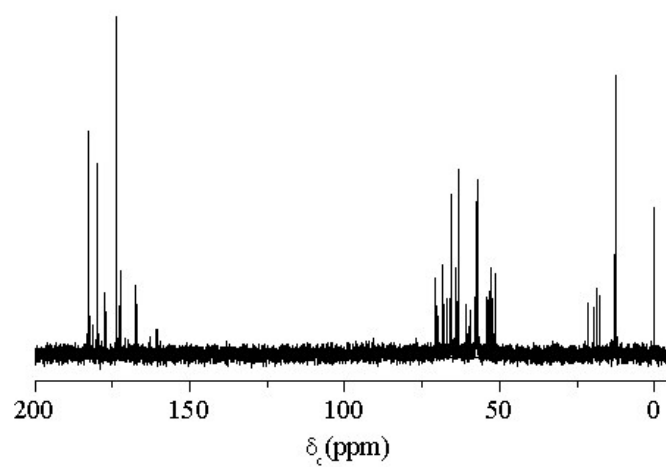


Figure S7

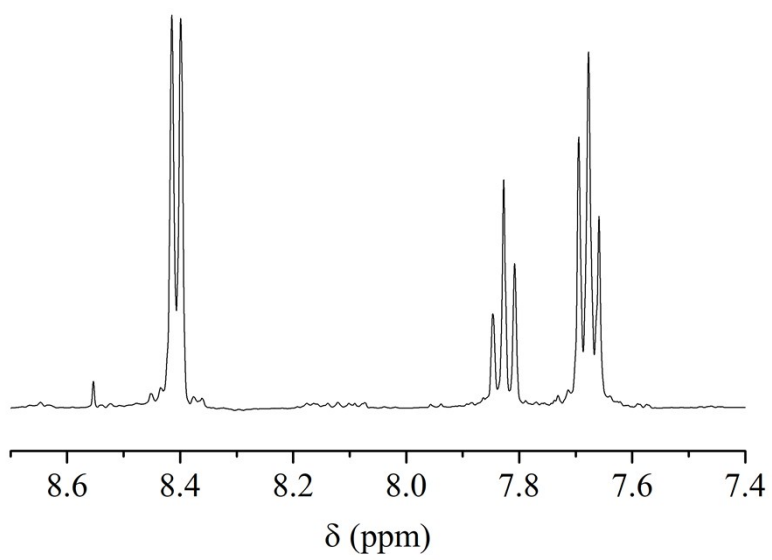
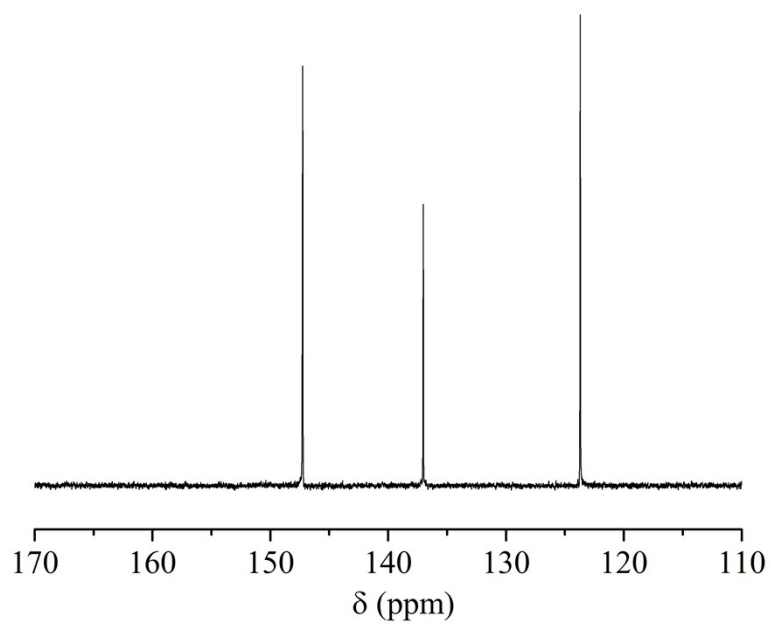
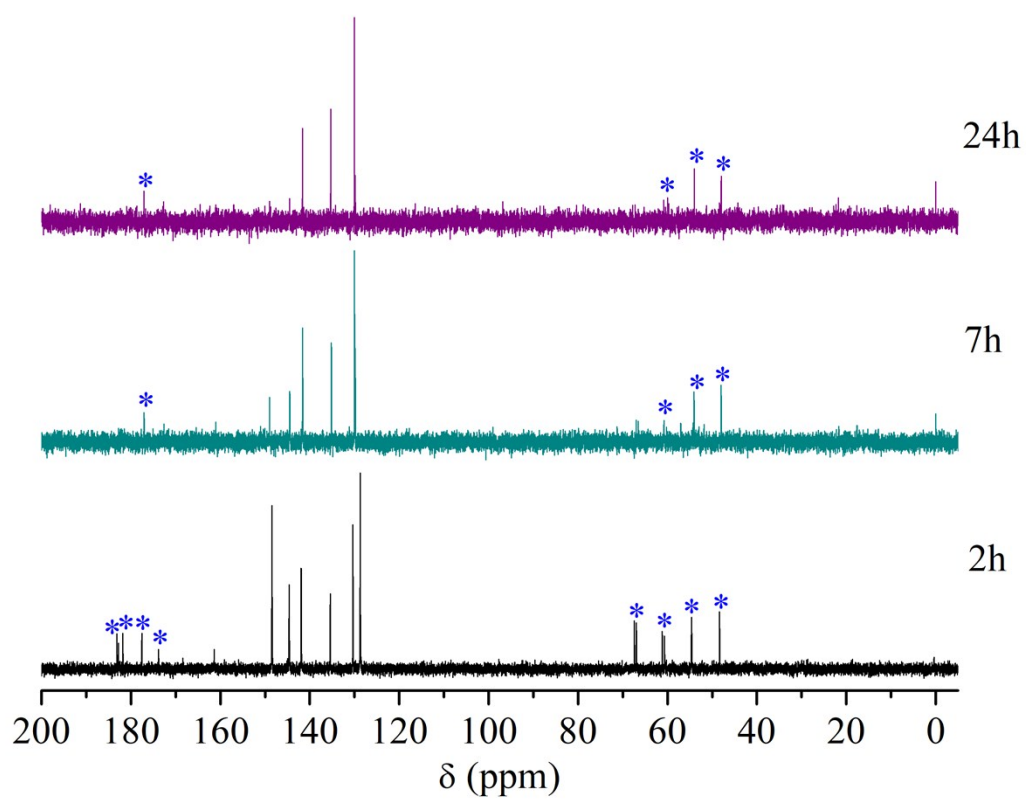


Figure S8



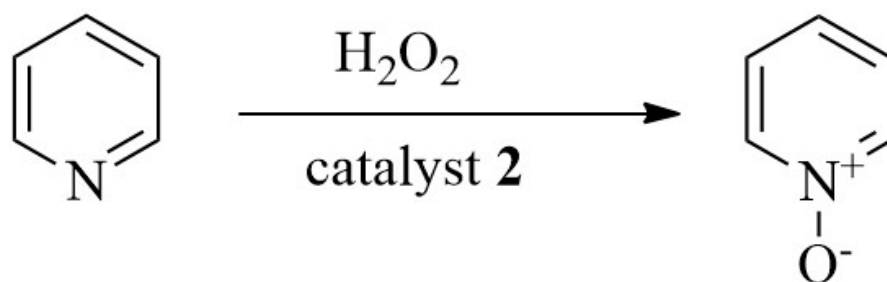


Table S1. Crystal data summaries of intensity data collections and structure refinements for (NH₄)[Ti(O₂)(Hpdtta)]·H₂O (**1**), (NH₄)₃[Ti(O₂)(pdta)H(pdta)(O₂)Ti]·7H₂O (**2**), (NH₄)[Ti(O₂)(pd3a)]·H₂O (**3**) and (NH₄)[Ti(O₂)(Heed3a)]·H₂O (**5**)

	1	2	3	5
Empirical formula	C ₁₁ H ₂₁ N ₃ O ₁₁ Ti	C ₂₂ H ₅₅ N ₇ O ₂₇ Ti	C ₉ H ₁₉ N ₃ O ₉ Ti	C ₁₀ H ₂₁ N ₃ O ₁₀ Ti
Formula weight	419.21	945.47	361.17	391.20
Crystal system	Triclinic	Monoclinic	Monoclinic	Monoclinic
<i>a</i> /Å	6.6948(3)	23.00(1)	16.1839(6)	7.9665(1)
<i>b</i> /Å	7.5422(4)	12.077(1)	11.1970(4)	30.1390(3)
<i>c</i> /Å	16.7297(8)	16.442(9)	15.8291(6)	13.2150(1)
<i>α</i> /°	83.398(4)			
<i>β</i> /°	85.937(4)	124.99(4)	103.231(4)	104.765(1)
<i>γ</i> /°	77.802(4)			
Unit cell volume/Å ³	819.25(7)	3741(3)	2792.3(2)	3068.18(6)
Space group	<i>P</i> $\bar{1}$	<i>C</i> 2/ <i>c</i>	<i>C</i> 2/ <i>c</i>	<i>P</i> 2 ₁ / <i>a</i>
No. of units per unit cell, <i>Z</i>	2	4	8	8
Temperature, K			173(2)	
Radiation type			Mo K α	
<i>D</i> _{calc} (g·cm ⁻³)	1.699	1.679	1.713	1.694
Absorption coefficient, μ /mm ⁻¹	0.590	0.537	0.666	0.618
No. of reflections measured	10186	8458	10553	23075
No. of independent reflections	5294	4114	3195	4575
<i>R</i> _{int}	0.0425	0.0975	0.0443	0.0203
Final <i>R</i> _{<i>I</i>} values [<i>I</i> > 2 σ (<i>I</i>)]	0.0348	0.0849	0.0346	0.0129
Final <i>wR</i> (<i>F</i> ²) values [<i>I</i> > 2 σ (<i>I</i>)]	0.0599	0.1097	0.0720	0.0341
Final <i>R</i> _{<i>I</i>} values (all data)	0.0816	0.1521	0.0565	0.0881
Final <i>wR</i> (<i>F</i> ²) values (all data)	0.0651	0.1326	0.0759	0.0369
Goodness of fit on <i>F</i> ²	0.787	1.049	0.958	0.0896
Largest diff. peak and hole (e·Å ⁻³)	0.33/-0.36	0.51/-0.59	0.76/-0.27	0.78/-0.52

Table S2. Selected bond distances (Å) and angles (°) for (NH₄)[Ti(O₂)(HpdtA)]·H₂O (**1**), (NH₄)₃[Ti(O₂)(pdta)H(pdta)(O₂)Ti]·7H₂O (**2**), (NH₄)[Ti(O₂)(pd3a)]·H₂O (**3**) and (NH₄)[Ti(O₂)(Heed3a)]·H₂O (**5**).

Compd.	1	2	3	5
Bond lengths (Å)				
Ti (1)—O(1)	2.069(1)	1.954(4)	1.962(1)	2.048(2)
Ti (1)—O(3)	1.996(1)	2.066(3)	2.015(1)	1.979(2)
Ti (1)—O(5)	1.944(1)	1.982(4)	1.976(1)	1.962(2)
Ti (1)—O(7)			1.881(2)	
Ti (1)—O(8)			1.854(2)	1.865(2)
Ti (1)—O(9)	1.844(1)	1.835(3)		1.850(2)
Ti (1)—O(10)	1.862(1)	1.856(3)		
Ti (1)—O(1w)				
Ti (1)—N(1)	2.280(1)	2.318(4)	2.301(2)	2.286(2)
Ti (1)—N(2)	2.284(1)	2.285(4)	2.204(2)	2.272(2)
Bond angles (°)				
O(1)—Ti(1)—O(3)	92.19(4)	96.3(1)	98.62(6)	94.00(7)
O(1)—Ti(1)—O(5)	93.40(5)	157.2(1)	160.88(6)	90.90(7)
O(1)—Ti(1)—O(7)			93.49(6)	
O(1)—Ti(1)—O(8)			98.97(6)	81.03(8)
O(1)—Ti(1)—O(9)	127.00(5)	100.7(2)		127.51(8)
O(1)—Ti(1)—O(10)	80.48(5)	94.5(2)		
O(1)—Ti(1)—O(1w)				
O(1)—Ti(1)—N(1)	73.66(4)	76.2(1)	77.48(6)	73.72(7)
O(1)—Ti(1)—N(2)	151.00(5)	89.1(2)	86.89(6)	150.37(7)
O(3)—Ti(1)—O(5)	158.06(4)	86.2(1)	88.59(6)	156.95(7)
O(3)—Ti(1)—O(7)			128.79(6)	
O(3)—Ti(1)—O(8)			82.50(6)	101.20(8)
O(3)—Ti(1)—O(9)	93.84(5)	81.6(2)		94.84(8)
O(3)—Ti(1)—O(10)	99.56(5)	128.1(1)		
O(3)—Ti(1)—O(1w)				
O(3)—Ti(1)—N(1)	77.51(5)	73.4(1)	74.81(6)	77.04(8)
O(3)—Ti(1)—N(2)	87.42(4)	148.4(1)	149.09(6)	86.96(7)
O(5)—Ti(1)—O(7)			95.71(6)	
O(5)—Ti(1)—O(8)			99.52(6)	101.81(7)
O(5)—Ti(1)—O(9)	99.64(5)	102.0(2)		99.99(8)
O(5)—Ti(1)—O(10)	102.29(5)	101.7(2)		
O(5)—Ti(1)—O(1w)				
O(5)—Ti(1)—N(1)	83.72(5)	82.9(1)	87.49(6)	82.77(7)
O(5)—Ti(1)—N(2)	77.51(5)	77.6(1)	78.22(6)	77.90(7)
O(7)—Ti(1)—O(8)			46.42(6)	
O(7)—Ti(1)—O(9)				
O(7)—Ti(1)—O(10)				
O(7)—Ti(1)—O(1w)				
O(7)—Ti(1)—N(1)			156.14(6)	
O(7)—Ti(1)—N(2)			80.74(6)	
O(1w)—Ti(1)—N(1)				

O(1w)—Ti(1)—N(2)				
O(8)—Ti(1)—O(9)				46.50(8)
O(8)—Ti(1)—N(1)			156.10(6)	154.44(8)
O(8)—Ti(1)—N(2)			126.94(7)	127.87(8)
O(9)—Ti(1)—O(10)	46.59(5)	46.5(1)		
O(9)—Ti(1)—N(1)	158.38(5)	154.3(2)		158.22(8)
O(9)—Ti(1)—N(2)	81.92(5)	128.0(1)		81.77(7)
O(10)—Ti(1)—N(1)	153.79(5)	157.9(2)		
O(10)—Ti(1)—N(2)	128.18(5)	82.2(1)		
N(1)—Ti(1)—N(2)	77.96(5)	77.7(1)	76.77(6)	77.68(7)

Table S3. Geometrical parameters of hydrogen bonds (Å, degree) in (NH₄)[Ti(O₂)(HpdtA)]·H₂O (**1**), (NH₄)₃[Ti(O₂)(pdta)H(pdta)(O₂)Ti]·7H₂O (**2**) and (NH₄)[Ti(O₂)(Heed3a)]·H₂O (**5**)

Complexs.	D—H···A	D—H (Å)	H···A (Å)	D···A (Å)	D—H···A(°)
1	O7—H···O1w	0.756(2)	1.798(2)	2.548(2)	171(3)
2	O7—H···O7a	1.225(3)	1.225(3)	2.450(6)	180(1)
5	O7—H···O16a	0.839(2)	2.004(8)	2.838(3)	172(8)
	O17—H···O6b	0.826(3)	2.032(2)	2.790(3)	152(8)

Symmetric transformation for **2**: $a, \frac{1}{2} - x, -\frac{1}{2} - y, 1 - z$; **5**: $a, 1 + x, y, z$; **b**: $\frac{1}{2} + x, \frac{1}{2} - y, 1 + z$

Table S4. Relevant ¹³C NMR data, δ/ppm of H₄edta, (NH₄)[Ti(O₂)(Hedta)]·2H₂O, (NH₄)₂[Ti(O₂)(edta)]·2H₂O and (NH₄)[Ti(O₂)(ed3a)]

Compounds.	—CH ₂ N	—NCH ₂ CO ₂	—COOH / CH ₂ CO ₂
H ₄ edta	51.3	54.6	172.0
(NH ₄)[Ti(O ₂)(Hedta)]·2H ₂ O ¹¹	59.7	66.1(11.5)	179.9(7.9)
(NH ₄) ₂ [Ti(O ₂)(edta)]·2H ₂ O ¹¹	60.3	66.7(12.1)	180.4(8.4)
(NH ₄)[Ti(O ₂)(ed3a)] ¹¹	54.4, 60.3, 61.0	66.8, 67.2	181.8, 182.8, 183.1

Δδ values are given in parentheses.