

Supplementary information

New layered chromium chalcogenides CsLiCrSe₂, RbLiCrS₂ and CsLiCrS₂: structure and properties

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Table S1 Summary of Rietveld refinement parameters for CsLiCrSe₂, CsLiCrS₂ and RbLiCrS₂.

	CsLiCrSe ₂	CsLiCrS ₂	RbLiCrS ₂
Space group		I4/mmm (No. 139)	
<i>a</i> (Å)	4.4185(2)	4.2678(1)	4.1799(2)
<i>c</i> (Å)	13.8002(7)	13.2824(3)	12.9501(8)
<i>V</i> (Å ³)	269.42(2)	241.93(1)	226.26(2)
<i>R_p</i> (%)	5.84	6.24	4.36
<i>R_{wp}</i> (%)	7.38	8.32	5.59
<i>R_{exp}</i> (%)	4.52	4.93	3.73
χ^2	2.67	2.85	2.25
<i>Ch</i> -Cr/Li- <i>Ch</i> (°)	120.2(1) × 2	125.0(1) × 2	121.1(1) × 2
<i>d</i> (Cr/Li- <i>Ch</i>) (Å)	104.35(5) × 4	102.30(5) × 4	103.96(6) × 4
<i>d</i> (Cr/Li- <i>Ch</i>) (Å)	2.547(1) × 4	2.414(2) × 4	2.397(2) × 4
A= Cs, Rb (<i>x, y, z</i>)		2a (0, 0, 0)	
A= Cs, Rb occupancy		1	
A= Cs, Rb <i>U_{iso}</i> (Å ²)	0.025(1)	0.0263(6)	0.0252(8)
Cr/Li (<i>x, y, z</i>)		4d (0, 0.5, 0.25)	
Cr/Li occupancy		1	
Cr/Li <i>U_{iso}</i> (Å ²)	0.055(2)	0.028(1)	0.044(1)
<i>Ch</i> = Se, S (<i>x, y, z</i>)	4d (0, 0, 0.3419(2))	4d (0, 0, 0.3350(2))	4d (0, 0, 0.3407(3))
<i>Ch</i> = Se, S occupancy		1	
<i>Ch</i> =Se, S <i>U_{iso}</i> (Å ²)	0.021(1)	0.023(1)	0.029(1)

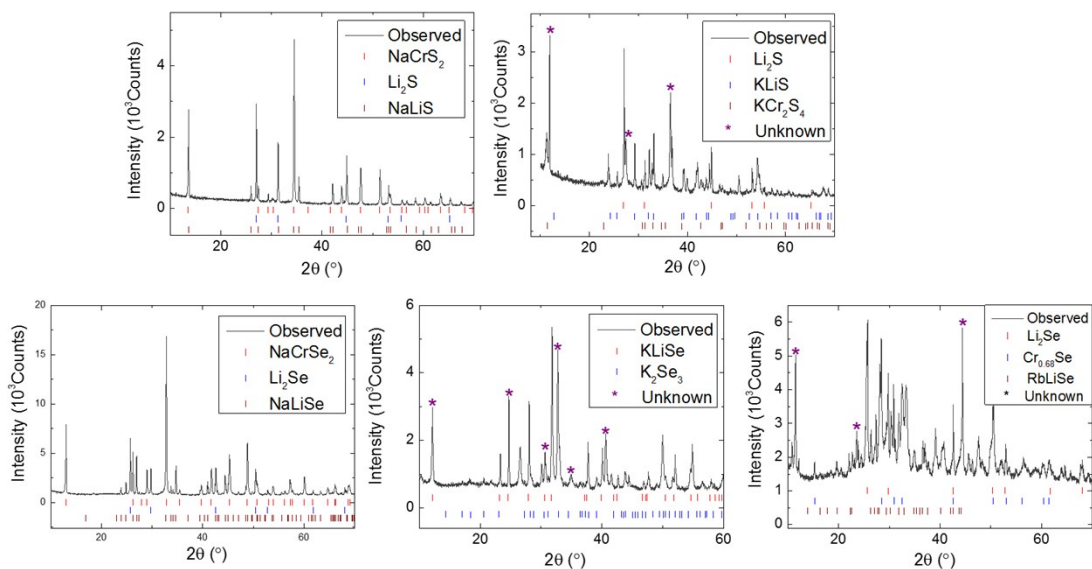


Fig S1 XRD patterns and the identified products for the failed reactions list in Table 1.

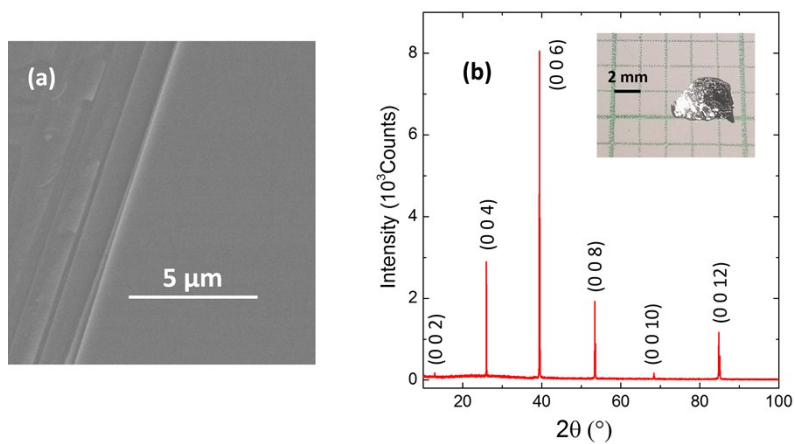


Fig S2 (a) Electron microscope image, (b) the XRD pattern and photograph of CsLiCrSe₂ single crystal (nominal).

Table S2 Chemical compositions, lattice parameters and spin-glass transition temperature in Cs(Li_{1-x}Cr_x)₂Se₂.

Nominal formula	Formula from refinements	Formula from EDS & ICP-AES	<i>a</i> (Å)	<i>c</i> (Å)	T _S (K)
Cs(Li _{0.55} Cr _{0.45}) ₂ Se ₂	Cs(Li _{0.552} Cr _{0.448}) ₂ Se ₂	Cs _{1.10(5)} (Li _{0.62(3)} Cr _{0.42(2)}) ₂ Se ₂	4.41603(18)	13.82173(65)	8
Cs(Li _{0.5} Cr _{0.5}) ₂ Se ₂	Cs(Li _{0.496} Cr _{0.504}) ₂ Se ₂	Cs _{1.08(5)} (Li _{0.55(3)} Cr _{0.50(3)}) ₂ Se ₂	4.41855(21)	13.80022(69)	15
Cs(Li _{0.45} Cr _{0.55}) ₂ Se ₂	Cs(Li _{0.424} Cr _{0.576}) ₂ Se ₂	Cs _{0.97(5)} (Li _{0.46(2)} Cr _{0.59(3)}) ₂ Se ₂	4.43787(20)	13.72584(71)	18

Table S3 Comparison of lattice parameters, bond length *d*(M'/M-Ch) and bond angle β in AM'MCh₂.

	<i>a</i> (Å)	<i>c</i> (Å)	<i>d</i> (M'/M-Ch) (Å)	β(°)
KLiFeSe ₂ ¹	4.1044(2)	13.7194(9)	2.5050 (1)	110.020(2)
RbLiFeSe ₂ ¹	4.1609(1)	14.0420(5)	2.5218 (2)	111.172(1)
KCoCuS ₂ ²	3.8831(9)	13.058(1)	2.5198(4)	100.8019
RbCoCuS ₂ ²	3.9314(5)	13.448(2)	2.5691(3)	99.8399
CsCoCuS ₂ ²	3.9667(9)	13.833(4)	2.6313(5)	97.8348
KLiMnS ₂ ³	4.070(3)	13.227(7)	2.439(4)	113.0(2)
RbLiMnS ₂ ³	4.106(2)	13.579(6)	2.4765(9)	111.9888
CsLiMnS ₂ ³	4.168(5)	14.006(9)	2.592(3)	107.0504
KLiZnS ₂ ³	3.998(2)	13.262(6)	2.411(4)	111.9(2)
RbLiZnS ₂ ³	4.046(2)	13.537(5)	2.4493(9)	111.37

1. D. D. Yuan, N. Liu, K. K. Li, S. F. Jin, J. G. Guo and X. L. Chen, *Inorg Chem*, 2017, **56**, 13187-13193.
2. M. Oledzka, J. G. Lee, K. V. Ramanujachary and M. Greenblatt, *J Solid State Chem*, 1996, **127**, 151-160.
3. D. Schmitz and W. Bronger, *Z Anorg Allg Chem*, 1987, **553**, 248-260.

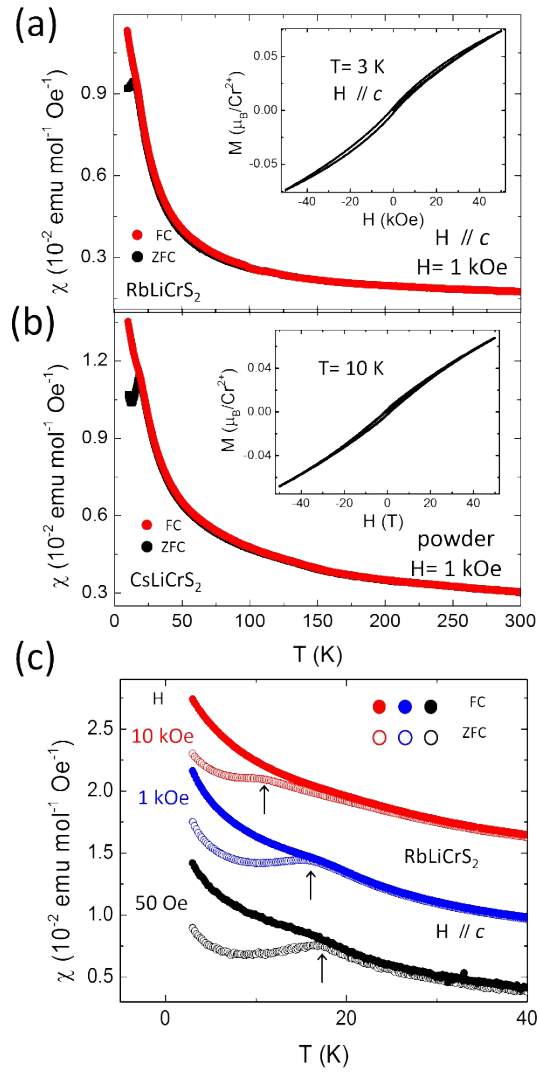


Fig S3 Zero-field-cooled (ZFC) and field-cooled (FC) susceptibilities of (a) RbLiCrS $_2$ single crystal for $H // c$ axis and (b) CsLiCrS $_2$ powder sample at $H = 1$ kOe. Insets show their M - H curves, respectively. (c) ZFC and FC curves of RbLiCrS $_2$ single crystal under different magnetic fields.

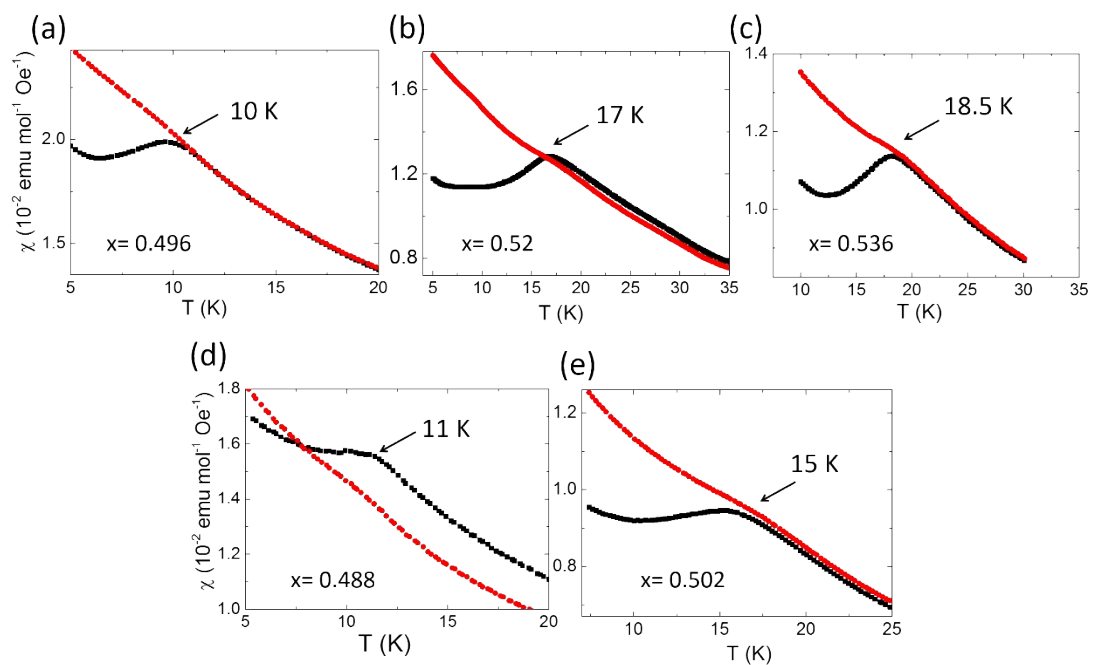


Fig S4 ZFC and FC curves of (a-c) Cs(Li $_{1-x}$ Cr $_x$) $_2$ S $_2$ and (d-e) Rb(Li $_{1-x}$ Cr $_x$) $_2$ S $_2$ with different x . x is obtained from Rietveld refinements.

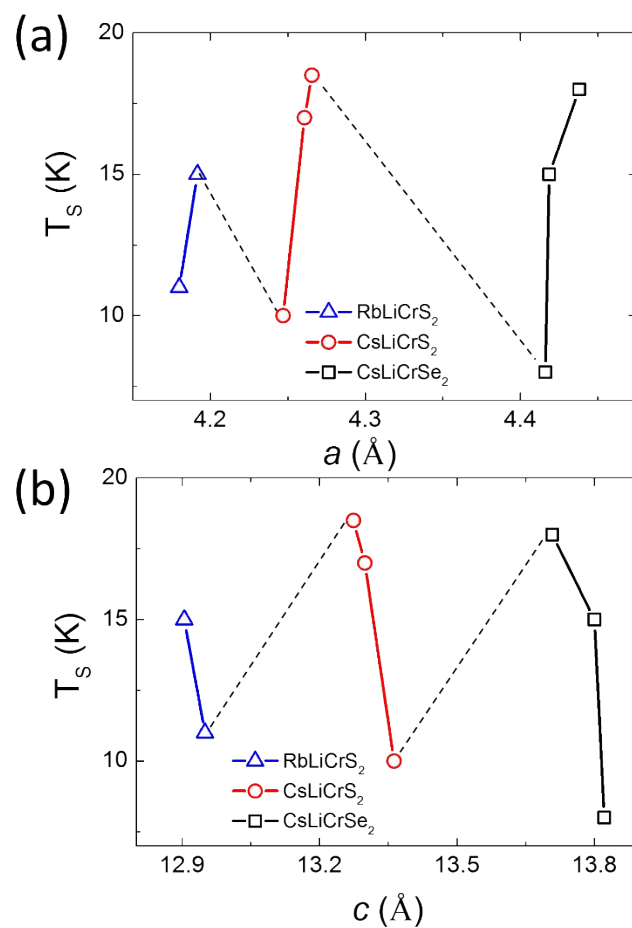


Fig S5 T_s in $A(\text{Li}_{1-x}\text{Cr}_x)_2\text{Ch}_2$ ($A = \text{Cs, Rb}$, $\text{Ch} = \text{S, Se}$) drawn against lattice parameter (a) a and (b) c .

Since the Cr concentration x affects both lattice parameters and the spin-glass temperature T_s , the relationship between lattice parameters and T_s can be established via x in $\text{Cs}(\text{Li}_{1-x}\text{Cr}_x)_2\text{Se}_2$, $\text{Cs}(\text{Li}_{1-x}\text{Cr}_x)_2\text{S}_2$ and $\text{Rb}(\text{Li}_{1-x}\text{Cr}_x)_2\text{S}_2$. As shown in Fig S6, higher T_s corresponds to larger a and smaller c . However, when the $A(\text{Li}_{1-x}\text{Cr}_x)_2\text{Ch}_2$ system is considered as a whole, T_s shows no clear dependence on a or c .